

**Supporting information for:**

**Self-assembled manganese acetate@tin dioxide colloidal quantum dots as electron transport layer for efficient and stable perovskite solar cells**

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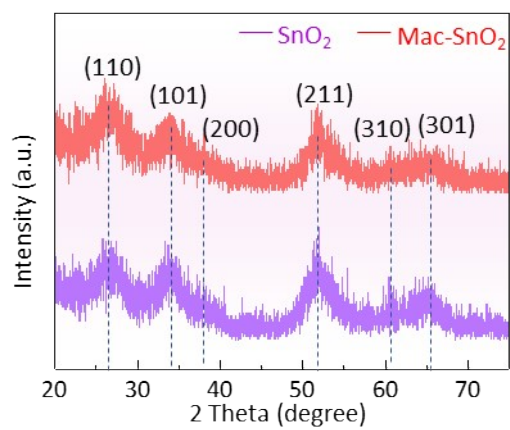
## ***Experimental section & Materials***

### ***Supplementary note 1***

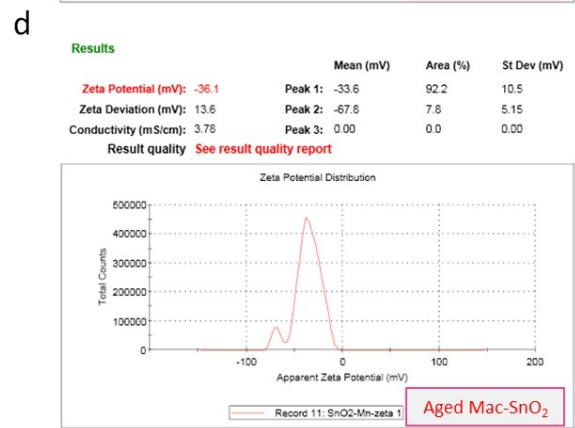
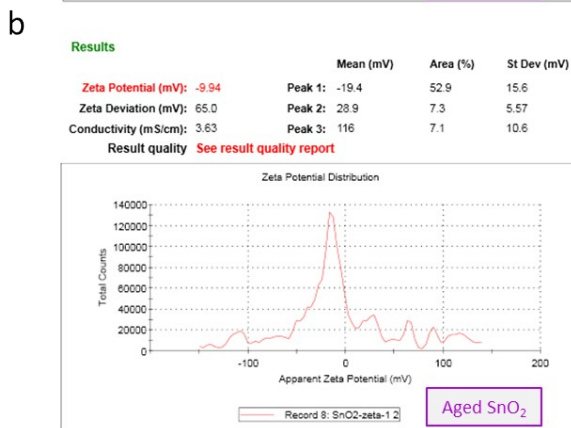
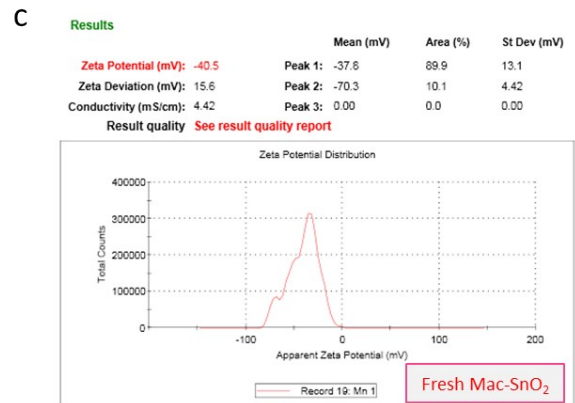
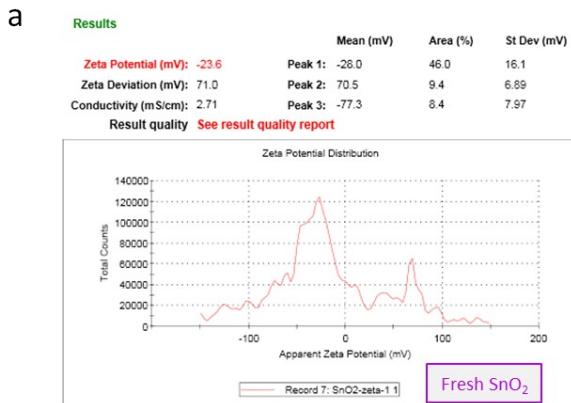
**Materials:** Mn(Ac)<sub>2</sub> was purchased from Aladdin. Lead iodide (PbI<sub>2</sub>), formamidinium iodide (FAI), lead bromide (PbBr<sub>2</sub>), Cesium iodide (CsI), Methylammonium chloride (MACl), methylammonium bromide (MABr), 4-tert-butylpyridine and lithium bis (trifluoromethanesulfonyl) imide (Li-TFSI), tris(2-(1H-pyrazol-1-yl)-4-tertbutylpyridine)-cobalt(III)tris(bis(trifluoromethylsulfonyl)imide) (FK209), tert-butylpyridine (TBP), SnO<sub>2</sub> aqueous colloidal dispersion (15wt%), 2,2',7,7'-tetrakis (N,N-dip-methoxyphenylamine) and 9,9'-Spirobifluorene (Spiro-OMeTAD) were purchased from Xi'an Yuri Solar Co., Ltd. Indium Tin Oxides (ITO) glass substrates, Dimethyl sulfoxide (DMSO) and N,N-Dimethylformamide (DMF) were purchased from YOUXUAN Technology Co. Ltd. (China). Chlorobenzene (CB) were purchased from Beijing InnoChem Science & Technology Co., Ltd. All chemicals and reagents were used as received from chemical companies without any further purification.

## ***Supplementary note 2***

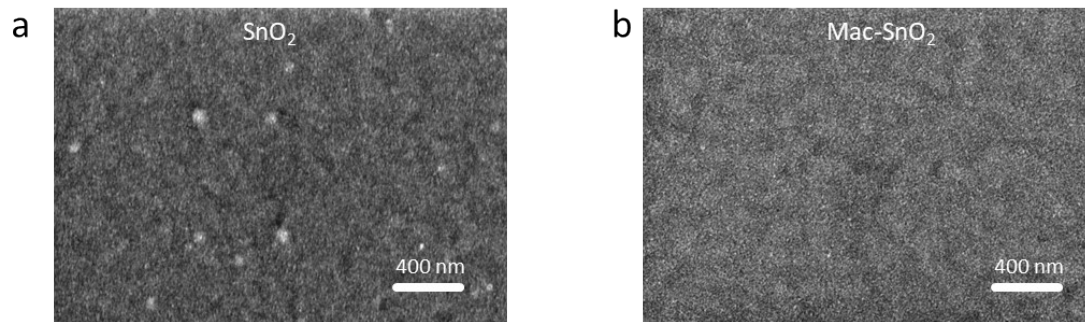
***Characterization:*** The scanning electron microscope (SEM) images was taken using a Hitachi S-4800. AFM was measured by Park NX20, Korea. The transmission electron microscopy (TEM) images was taken using JEOL 2100F. The optical properties of the films were analyzed using a UV-Vis spectrophotometer (Shimadzu, UV-3600Plus, Japan). Fourier transform infrared spectroscopy (FTIR) was measured by Nicolet iS10, America. The scanning electron microscope (SEM) images was taken using a Hitachi S-4800. AFM was measured by Park NX20, Korea. The crystal structure of the SnO<sub>2</sub> and the perovskite samples were carried out by X-ray diffraction (XRD) (PANalytical X-ray Spectrometer Empyrean DY01660 ID206161). PL spectra were obtained using a PL microscopic spectrometer (FLS1000, China) with a 405 nm CW laser excitation source. The TRPL (FLS1000, China) was measured by using an excitation wavelength of 405nm. The main corresponding setup consisted of perovskite, and mica-flakes. The transient photovoltage spectrum (TPV) was measured by CEL-TPV2000 (CEAULIGHT, China). X-ray photoelectron spectroscopy (XPS) was used to obtain the chemical states information (measured by Escalab250Xi, Germany). *J-V* characterizations was carried out under AM 1.5 G simulated sunlight illuminations (100 mW cm<sup>-2</sup>, Model 94043A, Oriel, American). Electrical impedance spectroscopy (EIS) was performed by using the electrochemical workstation (CHI660C, Chen Hua, China) with a frequency range from 1 Hz to 0.1MHz in the dark. The spectral responses were obtained from an EQE measurement system (Newport PV measurement, American).



**Fig. S1.** XRD patterns of SnO<sub>2</sub> and Mac-SnO<sub>2</sub> films on glass substrates.



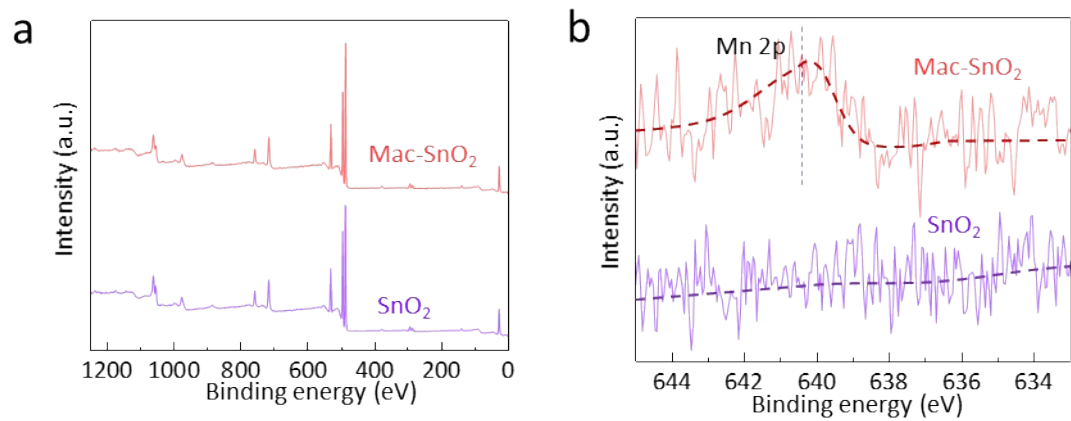
**Fig. S2.** Surface zeta potential of fresh (a) SnO<sub>2</sub> and (b) Mac-SnO<sub>2</sub> solutions. Surface zeta potential of 72 hours aged (c) SnO<sub>2</sub> and (d) Mac-SnO<sub>2</sub> solutions.



**Fig. S3.** The SEM images of (a) SnO<sub>2</sub> and (b) Mac-SnO<sub>2</sub> films.

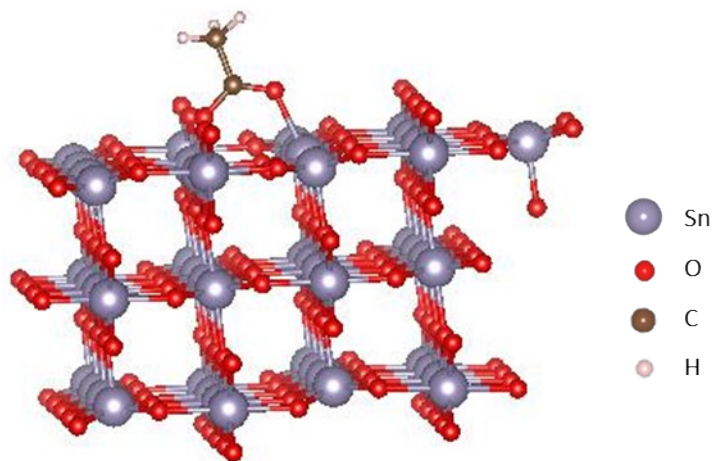


**Fig. S4.** Photograph of the SnO<sub>2</sub> and Mac-SnO<sub>2</sub> solutions.

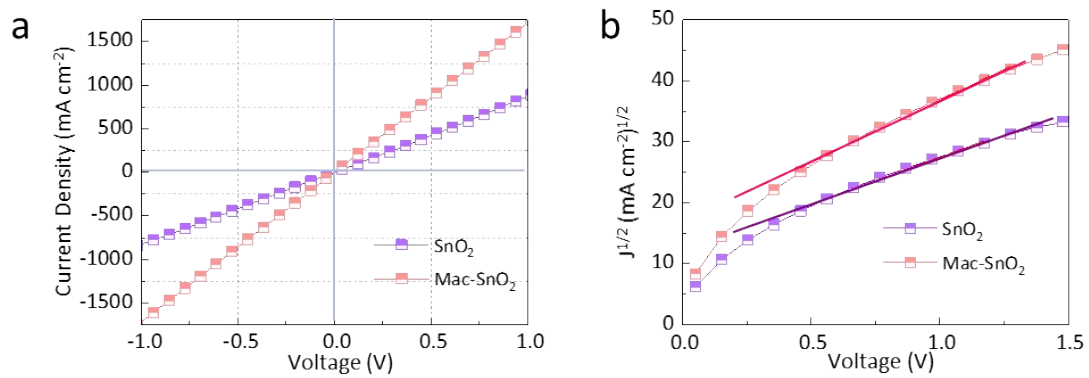


**Fig. S5.** XPS results of the SnO<sub>2</sub> and Mac-SnO<sub>2</sub>. (a) full spectra; (b) the amplified spectra of Mn 2p signals.

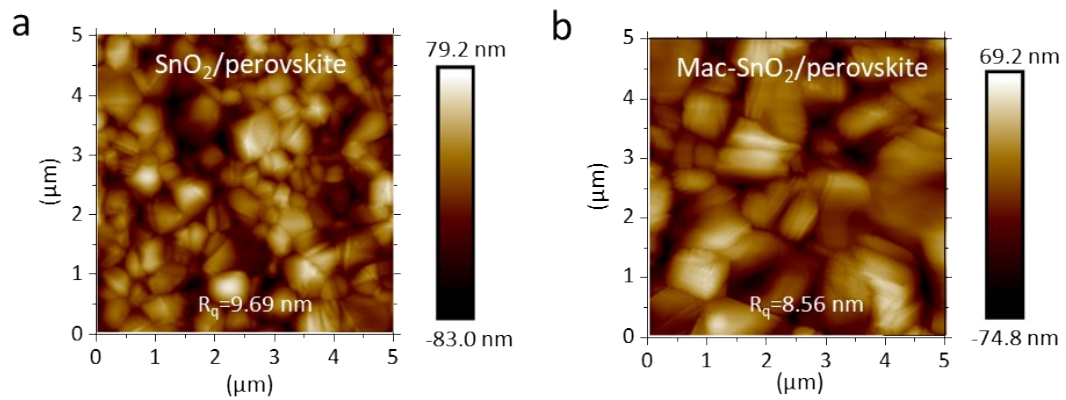




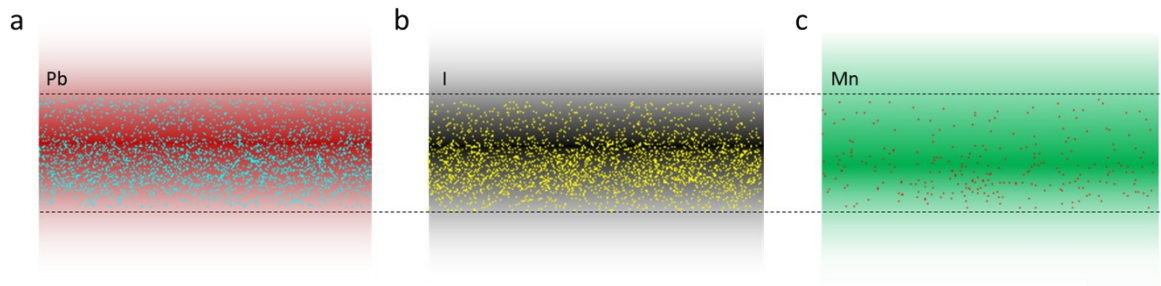
**Fig. S6.** Side view of the optimized  $\text{SnO}_2\text{-O}_v/\text{Ac}^-$ .



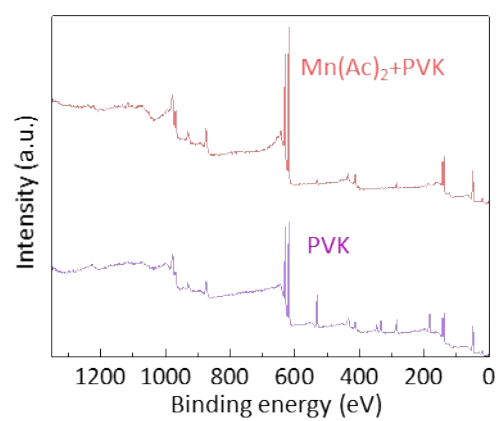
**Fig. S7.** The (a) conductivity and (b) electronic mobility of SnO<sub>2</sub> and Mac-SnO<sub>2</sub> films.



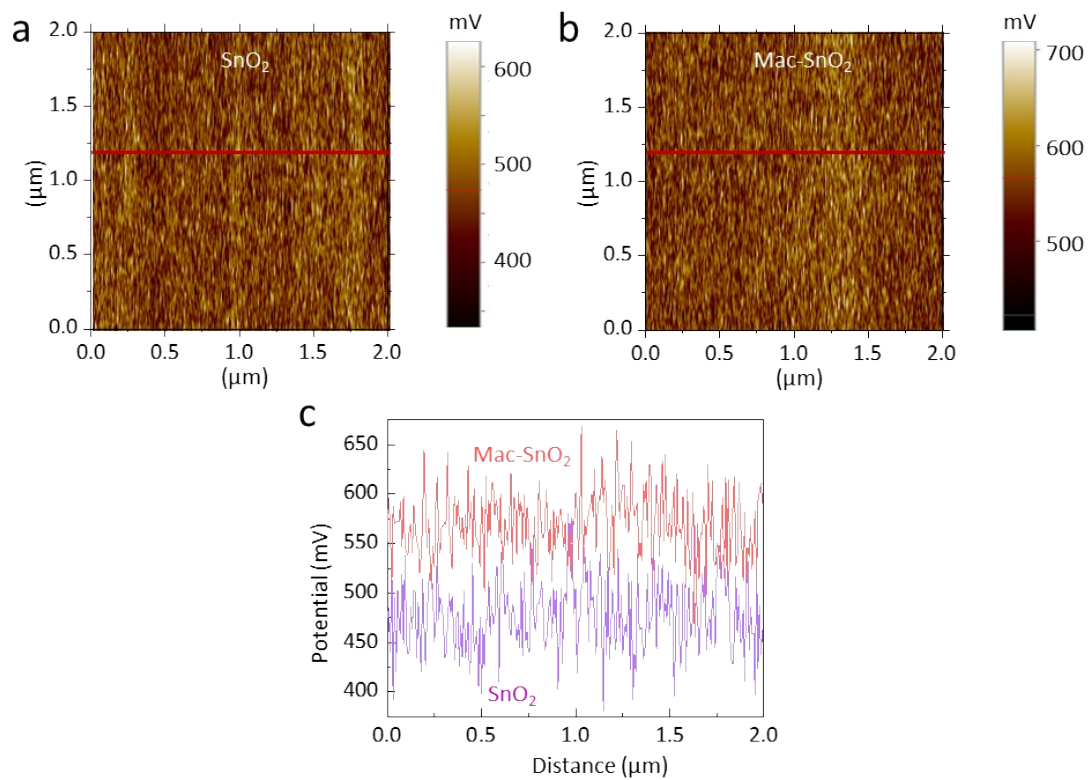
**Fig. S8.** AFM images of  $\text{SnO}_2/\text{perovskite}$  and  $\text{Mac-SnO}_2/\text{perovskite}$  films.



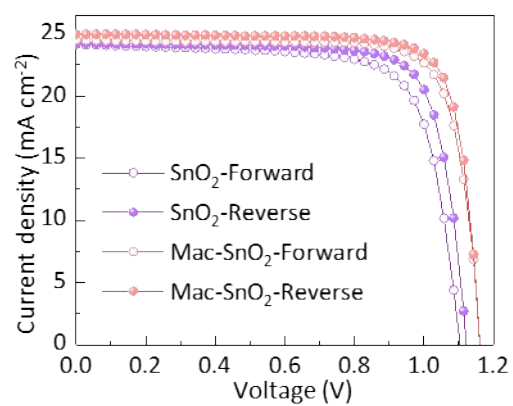
**Fig. S9.** The SEM element mapping analysis of (a) Pb, (b) I, and (c) Mn of the Mac-SnO<sub>2</sub>/perovskite films.



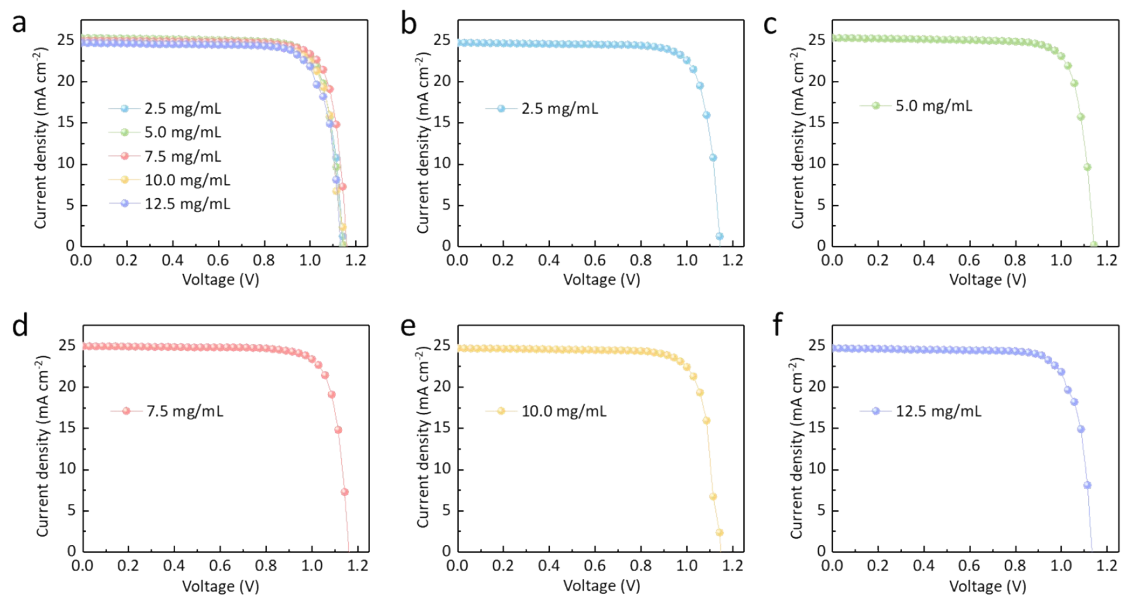
**Fig. S10.** XPS full spectra of the perovskite and Mn(Ac)<sub>2</sub>+perovskite.



**Fig. S11.** KPFM images of (a) SnO<sub>2</sub> and (b) Mac-SnO<sub>2</sub> films. (c) Surface potential data of SnO<sub>2</sub> and Mac-SnO<sub>2</sub> films extracted from KPFM.

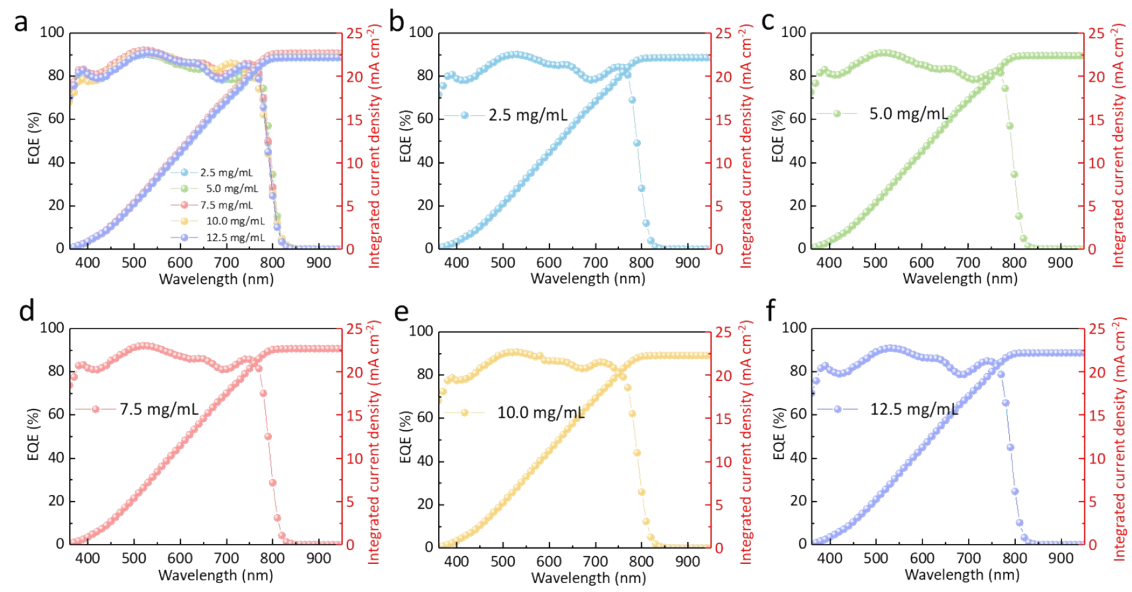


**Fig. S12.**  $J$ - $V$  curves of the SnO<sub>2</sub> and Mac-SnO<sub>2</sub> based PSCs at forward and reverse scans.

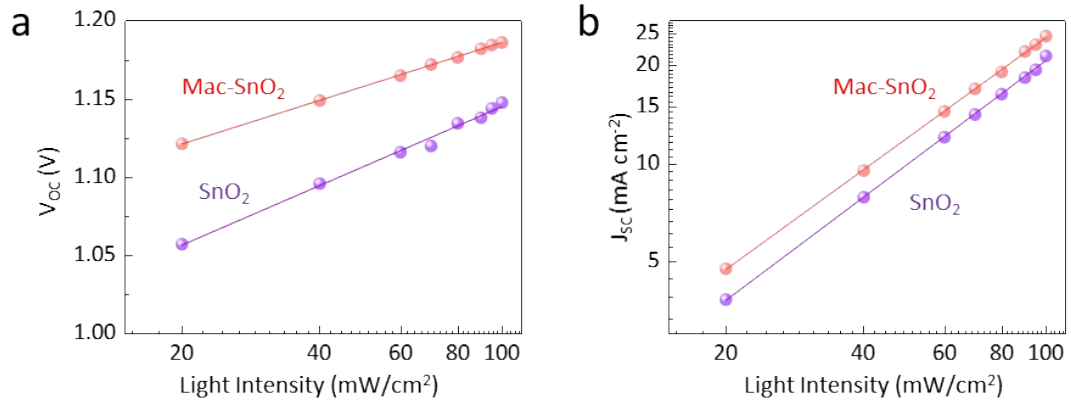


**Fig. S13.**  $J$ - $V$  curves in reverse scan of the PSCs based on different Mac- $\text{SnO}_2$  concentrations.

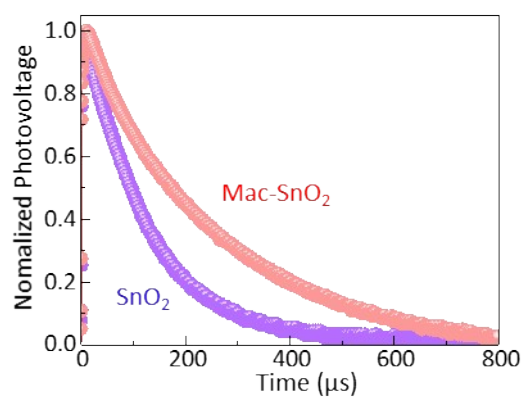




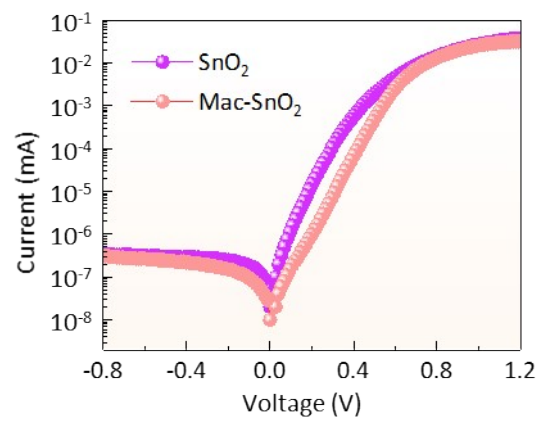
**Fig. S14.** EQE spectra of the PSCs based on different Mac-SnO<sub>2</sub> concentrations.



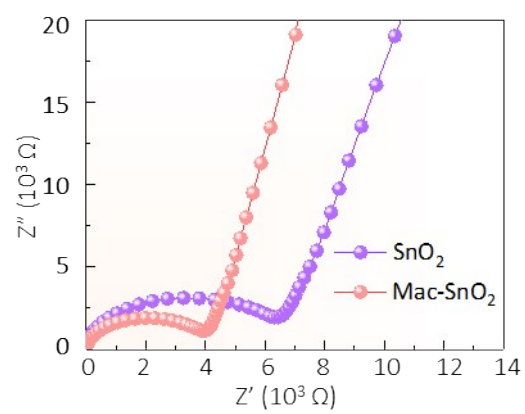
**Fig. S15.** (a) The dependence of  $V_{oc}$  on light intensity for SnO<sub>2</sub> and Mac-SnO<sub>2</sub> based devices. (b) The dependence of  $J_{sc}$  on light intensity for SnO<sub>2</sub> and Mac-SnO<sub>2</sub> based devices.



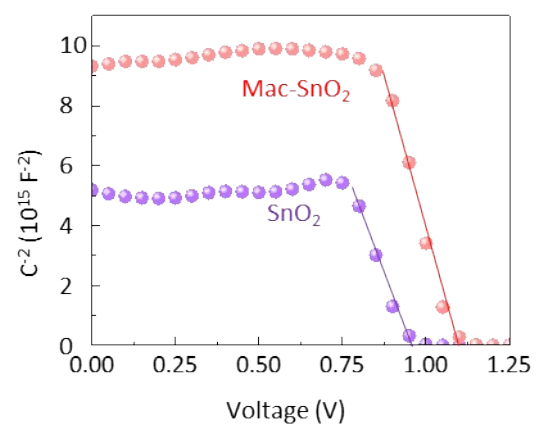
**Fig. S16.** TPV curves of SnO<sub>2</sub> and Mac-SnO<sub>2</sub> based devices.



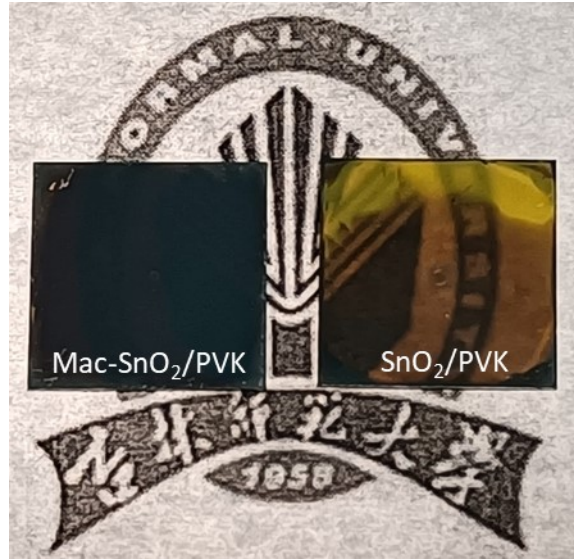
**Fig. S17.** Dark  $I$ - $V$  curves of the PSCs based on SnO<sub>2</sub> and Mac-SnO<sub>2</sub> ETL.



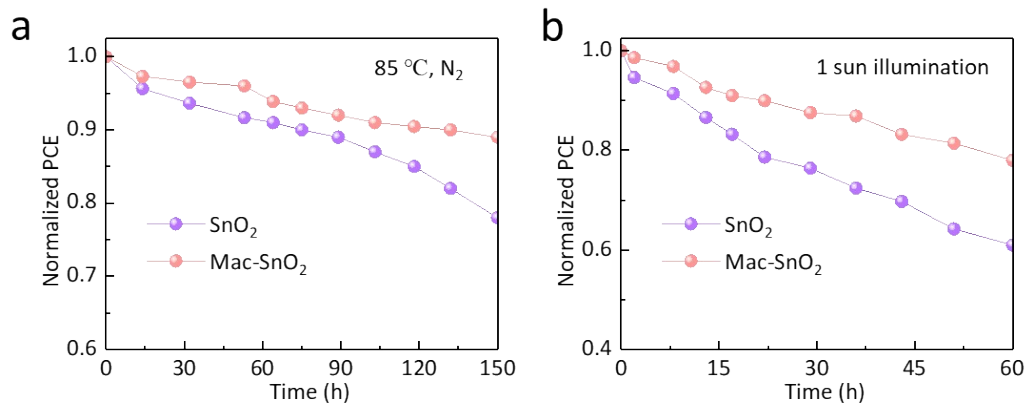
**Fig. S18.** Nyquist plots of the  $\text{SnO}_2$  and  $\text{Mac-SnO}_2$  based devices.



**Fig. S19.** Mott-Schottky plots of SnO<sub>2</sub> and Mac-SnO<sub>2</sub> based devices.



**Fig. S20.** Photograph of SnO<sub>2</sub>/perovskite and Mac-SnO<sub>2</sub>/perovskite exposed to air environment with 30%–40% humidity for 20 days.



**Fig. S21.** Normalized *PCEs* of SnO<sub>2</sub> and Mac-SnO<sub>2</sub> based devices stored (a) at 85 °C in N<sub>2</sub> atmosphere and (b) under 1 sun illumination.



Table S1. Detail values of  $V_{oc}$ ,  $J_{sc}$ ,  $FF$  and  $PCE$  for Mac-SnO<sub>2</sub> based and SnO<sub>2</sub> based devices.

	$V_{oc}$ (V)	$J_{sc}$ (mA cm <sup>-2</sup> )	$FF$ (%)	$PCE$ (%)
Mac-SnO <sub>2</sub> -Reverse	1.16	24.96	80.69	23.36
Mac-SnO <sub>2</sub> -Forward	1.15	24.49	80.47	22.56
SnO <sub>2</sub> -Reverse	1.12	24.25	77.66	21.11
SnO <sub>2</sub> -Forward	1.10	24.12	74.29	19.75

Table S2. Photovoltaic parameters derived from the champion devices with different Mac-SnO<sub>2</sub> concentrations.

	$V_{oc}$ (V)	$J_{sc}$ (mA cm <sup>-2</sup> )	$FF$ (%)	$PCE$ (%)
2.5 mg/mL	1.14	24.73	79.86	22.58
5.0 mg/mL	1.14	25.32	80.00	23.08
7.5 mg/mL	1.16	24.96	80.69	23.36
10.0 mg/mL	1.14	24.72	79.38	22.44
12.5 mg/mL	1.13	24.72	78.05	21.81

Table S3. The output parameters of the SnO<sub>2</sub> based devices.

	$V_{oc}$ (V)	$J_{sc}$ (mA cm <sup>-2</sup> )	$FF$ (%)	$PCE$ (%)
1	1.09	23.93	79.04	20.64
2	1.08	24.21	77.66	20.29
3	1.09	22.71	79.74	19.82
4	1.09	22.60	78.27	19.32
5	1.08	22.65	77.44	18.91
6	1.12	22.77	79.09	20.08
7	1.11	23.20	78.47	20.21
8	1.13	23.29	77.84	20.53
9	1.12	23.68	78.14	20.79
10	1.12	23.55	77.17	20.33
11	1.13	23.01	73.98	19.22
12	1.14	22.95	79.51	20.95
13	1.12	24.25	77.66	21.11
14	1.15	21.73	79.44	19.85
15	1.15	23.00	79.49	20.98
16	1.14	22.07	76.30	19.16
17	1.16	23.37	77.58	21.07
18	1.15	22.38	80.05	20.51
19	1.13	24.56	75.72	21.01
20	1.10	23.55	78.32	20.35

Table S4. The output parameters of the Mac-SnO<sub>2</sub> based devices.

	$V_{oc}$ (V)	$J_{sc}$ (mA cm <sup>-2</sup> )	$FF$ (%)	$PCE$ (%)
1	1.13	24.10	78.20	21.33
2	1.16	22.44	80.73	21.06
3	1.09	23.91	82.08	21.34
4	1.16	24.96	80.69	23.36
5	1.12	24.80	81.32	22.66
6	1.17	24.05	76.30	21.53
7	1.14	24.81	75.05	21.25
8	1.16	24.10	77.30	21.65
9	1.14	25.31	79.73	23.01
10	1.15	24.50	79.78	22.55
11	1.15	24.64	79.01	22.34
12	1.15	22.24	80.77	21.56
13	1.12	24.06	80.12	21.65
14	1.15	23.37	80.33	21.52
15	1.14	25.33	79.98	23.05
16	1.13	24.40	78.86	21.83
17	1.17	23.24	80.08	21.68
18	1.12	24.66	80.63	22.25
19	1.16	23.37	79.37	21.56
20	1.13	24.24	77.99	21.43