

Supporting Information

Figure Captions

Fig. S1. Digital photographs of CSO:xMn ($x = 0.01-0.05$) phosphor samples irradiated with 254 nm and 365 nm UV lamps, respectively, as well as afterglow photographs after stopping irradiation with the 254 nm UV lamp.

Fig. S2. Emission spectra of the CaSb₂O₆:xMn samples under 310 nm excitation.

Fig. S3. XRD patterns of phosphor samples CSO:0.01Mn (different Mn sources), CSO:0.04Mn and CSO:0.04Mn,0.06La³⁺.

Fig. S4. XPS spectrum of CSO:0.04Mn,0.06La³⁺ phosphor sample.

Fig. S5. EDS spectrum of CSO:0.01Mn,0.01La³⁺ samples.

Fig. S6. Sa values of CSO:0.01Mn,0.02La³⁺ samples excited at different wavelengths.

Fig. S7. Sr values of CSO:0.01Mn,0.02La³⁺ samples excited at different wavelengths.

Fig. S8. Photographs of CSO under daylight, 254 nm and 365 nm UV irradiation, and at different times after stopping 254 nm UV irradiation.

Fig. S9. Afterglow decay curves of CSO matrix.

Fig. S10. Fitted TL spectra for CSO:0.01Mn,xLa³⁺ ($x = 0.02-0.06$). The red, green, blue and cyan lines correspond to peak 1 (trap 1), peak (trap 2), peak (trap 3) and peak (trap 4), respectively.

Fig. S11. Fitted TL spectra of the CSO. The red and blue lines correspond to peak 1 (trap 1) and peak 2 (trap 2), respectively.

Fig. S12. Pictures of CSO:0.01Mn phosphor under continuous irradiation with a 365 nm UV lamp (5 W) for different times. The phosphor was first irradiated with a 254 nm UV lamp for 3 minutes and then placed in a dark environment for different periods of time (1 day, 10 days, 30 days).

Fig S13. Thermogravimetric curves of CSO:0.01Mn,0.02La³⁺ samples.

Table. S1. J and d_0 values of $\text{Ca}^{2+}\text{-O}^{2-}$, $\text{Sb}^{5+}\text{-O}^{2-}$, $\text{La}^{3+}\text{-O}^{2-}$, $\text{Mn}^{4+}\text{-O}^{2-}$, $\text{Mn}^{2+}\text{-O}^{2-}$.

Ion pair	$J(\text{Kcal/mol})$	d_0 (Å)
$\text{Ca}^{2+}\text{-O}^{2-}$	126.6	1.967
$\text{Sb}^{5+}\text{-O}^{2-}$	65.64	1.912
$\text{La}^{3+}\text{-O}^{2-}$	135.56	2.172
$\text{Mn}^{4+}\text{-O}^{2-}$	77.6	1.753
$\text{Mn}^{2+}\text{-O}^{2-}$	109.3	1.790

Table. S2. $\text{Ca}^{2+}\text{-O}^{2-}$, $\text{Sb}^{5+}\text{-O}^{2-}$, $\text{Mn}^{4+}\text{-O}^{2-}$, $\text{Mn}^{2+}\text{-O}^{2-}$ Key parameters and bond energy values (E_{m-o} , $E_{\text{Mn}^{4+}\text{-O}^{2-}}$, $E_{\text{Mn}^{2+}\text{-O}^{2-}}$), Bond energy deviation ($\Delta E_{\text{Mn}^{4+}\text{-O}^{2-}}^m$, $\Delta E_{\text{Mn}^{2+}\text{-O}^{2-}}^m$), m represents the Ca, Sb bond energy is the unit of kcal/mol.

Central atom	Coordination atom	Count	d_{m-o} (Å)	E_{m-o}	$E_{\text{Mn}^{4+}\text{-O}^{2-}}$	$E_{\text{Mn}^{2+}\text{-O}^{2-}}$	$\Delta E_{\text{Mn}^{4+}\text{-O}^{2-}}^m$	$\Delta E_{\text{Mn}^{2+}\text{-O}^{2-}}^m$
Ca1	O1	3x	2.437	35.524	6.106	19.009		
	O1	3x	2.437	35.515	6.104	19.004		
Averag e				35.520	6.105	19.006	29.415	16.513
Central atom	Coordination atom	Count	d_{m-o} (Å)	E_{m-o}	$E_{\text{Mn}^{4+}\text{-O}^{2-}}$	$E_{\text{Mn}^{2+}\text{-O}^{2-}}$	$\Delta E_{\text{Mn}^{4+}\text{-O}^{2-}}^m$	$\Delta E_{\text{Mn}^{2+}\text{-O}^{2-}}^m$
Sb1	O1	2x	1.961	57.576	55.362	172.358		
	O1	1x	1.961	57.545	55.332	172.265		
	O1	1x	1.961	57.529	55.318	172.219		
	O1	2x	1.961	57.467	55.258	172.033		
Averag e				57.529	55.318	172.219	2.212	114.690

Table. S3. $\text{Ca}^{2+}\text{-O}^{2-}$, $\text{Sb}^{5+}\text{-O}^{2-}$, $\text{La}^{3+}\text{-O}^{2-}$ Key parameters and bond energy values (E_{m-o} , $E_{\text{La}^{3+}\text{-O}^{2-}}$), Bond energy deviation ($\Delta E_{\text{La}^{3+}\text{-O}^{2-}}^m$), m represents the Ca, Sb bond energy is the unit of kcal/mol.

Central atom	Coordination atom	Count	$d_{m-o}(\text{\AA})$	E_{m-o}	$E_{\text{La}^{3+}\text{-O}^{2-}}$	$\Delta E_{\text{La}^{3+}\text{-O}^{2-}}^m$
Ca1	O1	3x	2.437	35.524	44.132	
	O1	3x	2.437	35.515	44.120	
Average				35.520	44.126	8.607
Central atom	Coordination atom	Count	$d_{m-o}(\text{\AA})$	E_{m-o}	$E_{\text{La}^{3+}\text{-O}^{2-}}$	$\Delta E_{\text{La}^{3+}\text{-O}^{2-}}^m$
Sb1	O1	2x	1.961	57.576	400.159	
	O1	1x	1.961	57.545	399.943	
	O1	1x	1.961	57.529	399.835	
	O1	2x	1.961	57.467	399.403	
Average				57.529	399.835	342.305

Table. S4. Rietveld refinement lattice parameters of $\text{CSO:0.01Mn,0.01La}^{3+}$ phosphor as well as the standard CSO (PDF#46-1496).

Parameters	CSO	$\text{CSO:0.01Mn,0.01La}^{3+}$
Crystal system	trigonal	trigonal
Space-group	P-31m	P-31m

a (Å)	5.2405	5.2399
b (Å)	5.2405	5.2399
c (Å)	5.0221	5.0309
$V(\text{Å}^3)$	119.44	119.62
Z	1	6
R_{wp}		10.10%
R_p		7.64%
χ^2		1.32

Table. S5. The refined positions of all atoms for CSO:0.01Mn,0.01La³⁺.

Atomic parameters					
Atom	Wyck.	x/a	y/b	z/c	U [Å ²]
Ca1	1a	0	0	0	0.0232(19)
Sb1	2d	1/3	2/3	1/2	0.0156(15)
O1	6k	0.3644(10)	0	0.2692(9)	0.01

Table. S6. Relative parameters for temperature measurements of partial phosphors based on double light emission.

Hosts	Doped ions	Temperature range(K)	Sr(%K ⁻¹)	Sa(K ⁻¹)	Ref.
Ca ₂ LaSbO ₆	Eu ³⁺ ,Mn ⁴⁺	273-473	2.60	0.098	¹
La ₂ LiSbO ₆	Eu ³⁺ ,Mn ⁴⁺	303-523	0.89	0.000264	²

$\text{La}_2\text{Mg}_{1.33}\text{Ta}_{0.67}\text{O}$	$\text{Eu}^{3+}, \text{Mn}^{4+}$	298-443	2.72	0.057	3
6					
NaLaMgWO_6	$\text{Eu}^{3+}, \text{Mn}^{4+}$	303-523	0.86	0.0302	4
$\text{Ca}_2\text{LuSbO}_6$	$\text{Eu}^{3+}, \text{Mn}^{4+}$	303-523	0.79	0.0028	5
ZnTiO_3	$\text{Eu}^{3+}, \text{Mn}^{4+}$	80-310	2.70	0.028	6
$\text{Mg}_3\text{Eu}_2\text{Ge}_3\text{O}_{12}$	$\text{Eu}^{3+}, \text{Mn}^{4+}$	7-300	0.019	0.015	7
$\text{Ba}_3\text{MgSb}_2\text{O}_9$	$\text{Mn}^{2+}, \text{Mn}^{4+}$	10-300	1.51	1.32	8
$\text{ZnGa}_{2-y}\text{Al}_y\text{O}_4$	$\text{Mn}^{2+}, \text{Mn}^{4+}$	100-475	4.34	-	9
$\text{Ba}_{0.75}\text{Al}_{11}\text{O}_{17.25}$	$\text{Mn}^{2+}, \text{Mn}^{4+}$	303-453	1.63	-	10
$\text{LaM}_{1-x}\text{Al}_{1-y}\text{O}_{19}$	$\text{Mn}^{2+}, \text{Mn}^{4+}$	225-475	3.22	-	11
$\text{Ca}_2\text{Sb}_2\text{O}_7$	$\text{Eu}^{3+}, \text{host}$	303-523	3.36	0.3031	12
CaSb_2O_6	$\text{Mn}^{2+}, \text{Mn}^{4+}$	300-440	3.06	0.092	This work

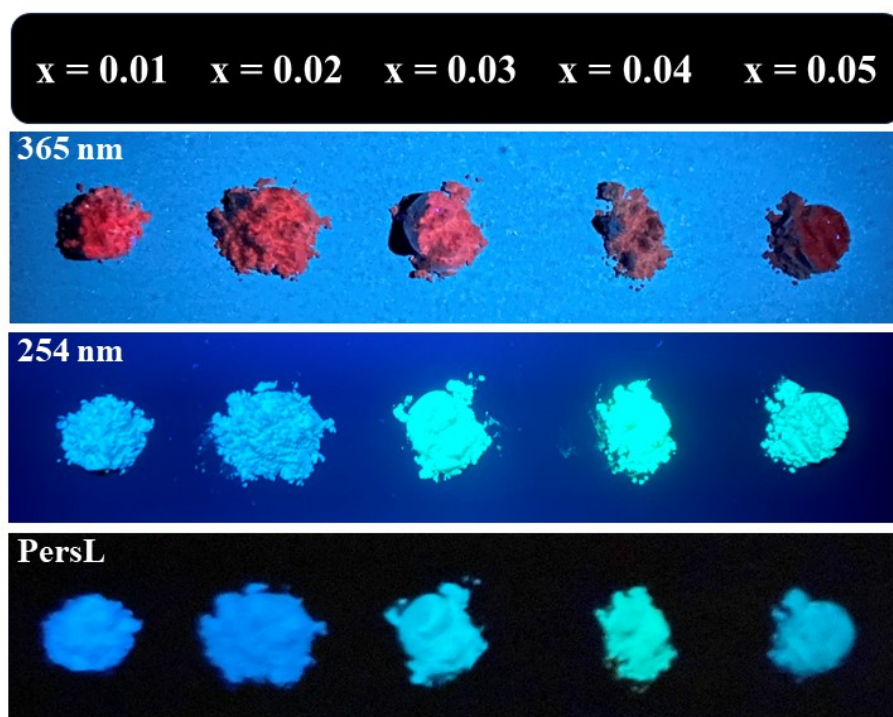


Fig. S1

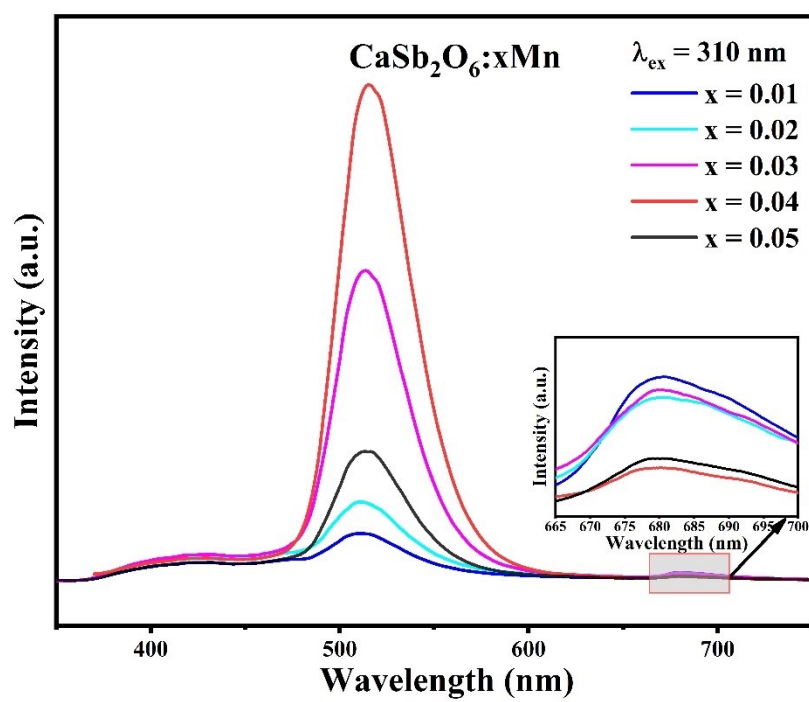


Fig. S2

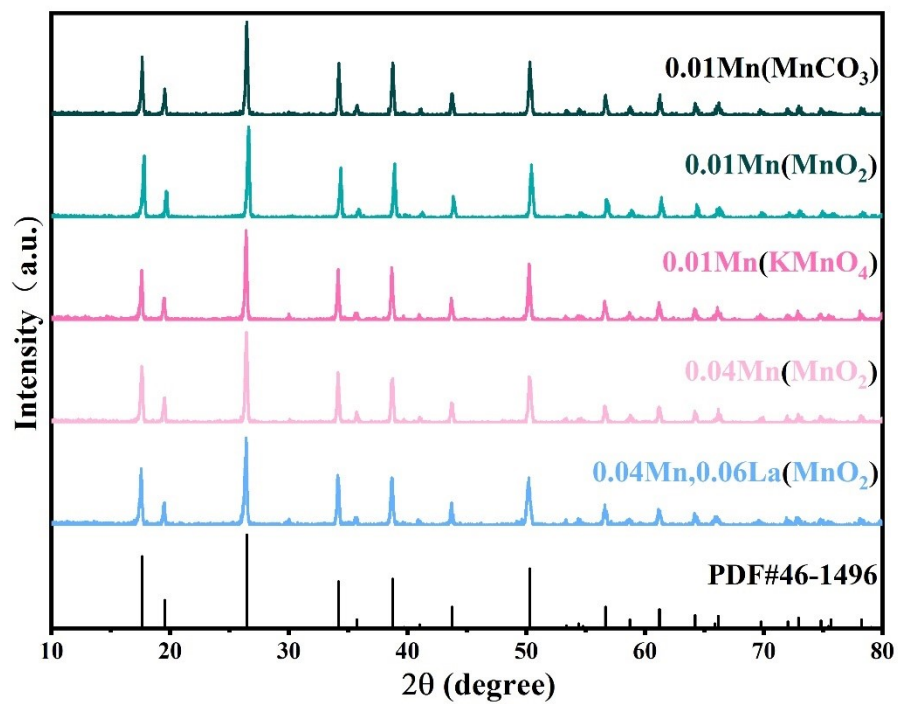


Fig. S3

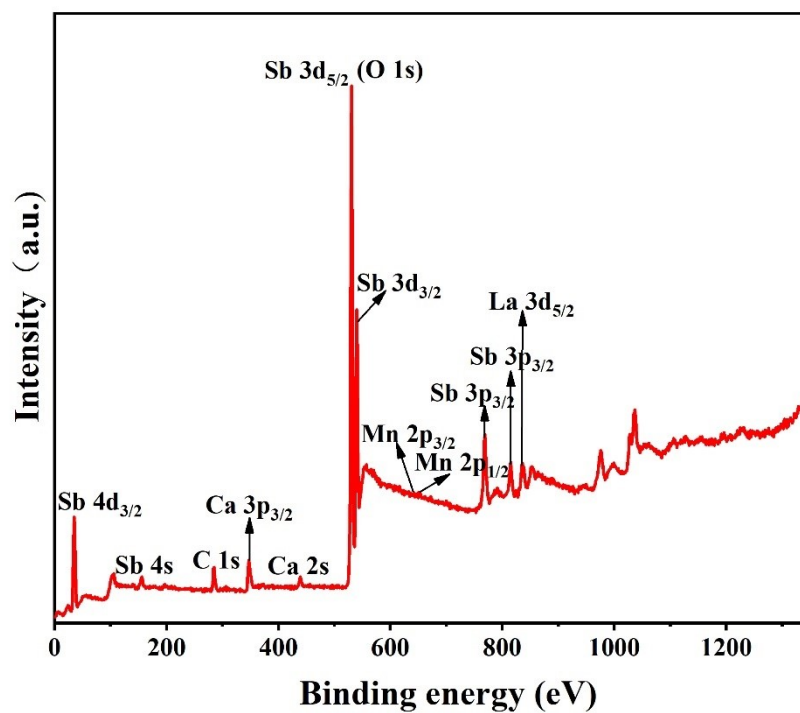


Fig. S4

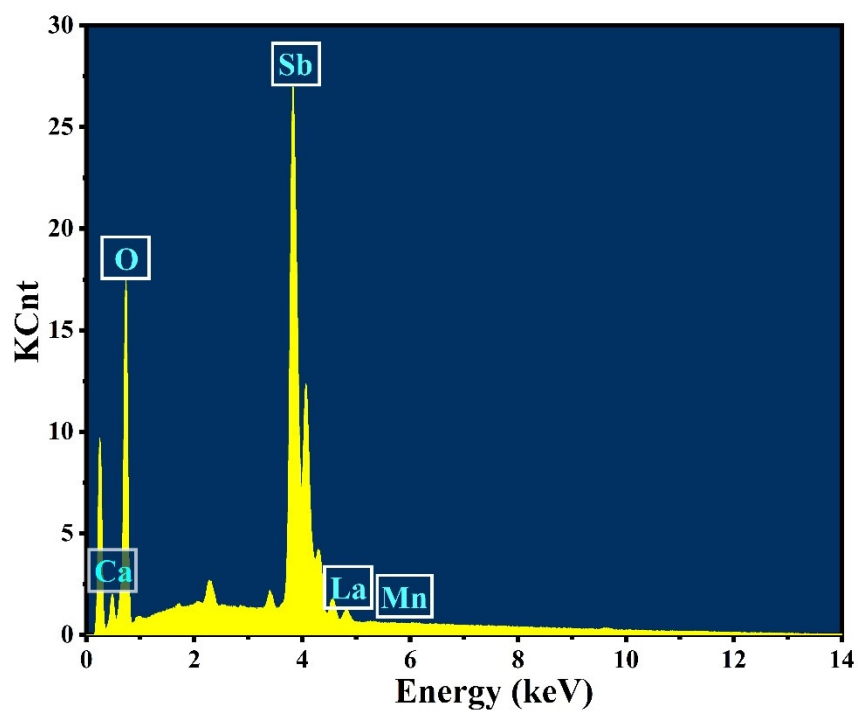


Fig. S5

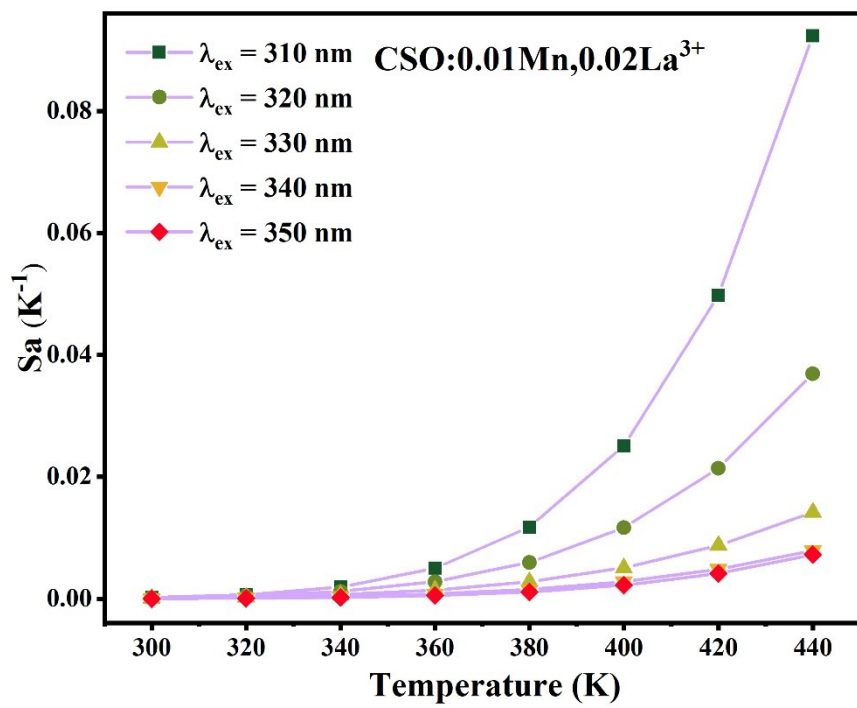


Fig. S6

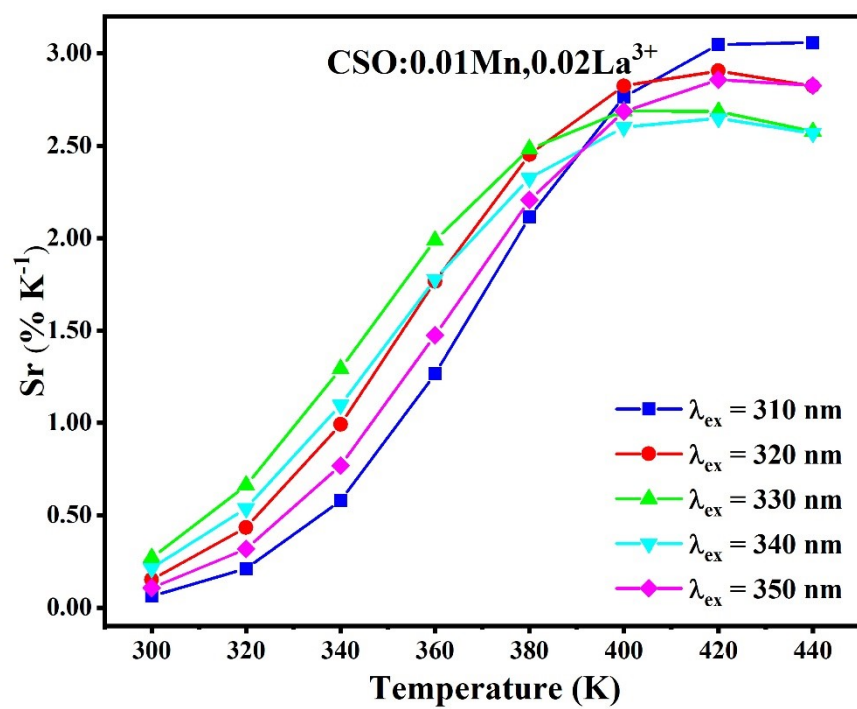


Fig. S7

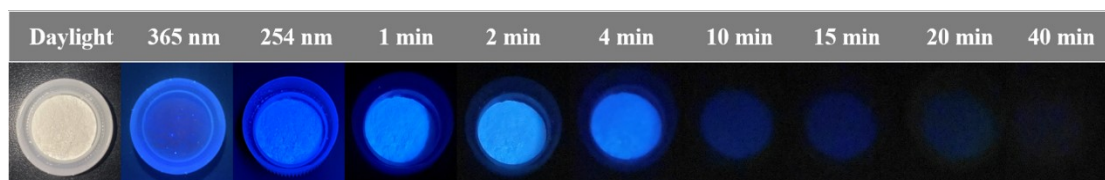


Fig. S8

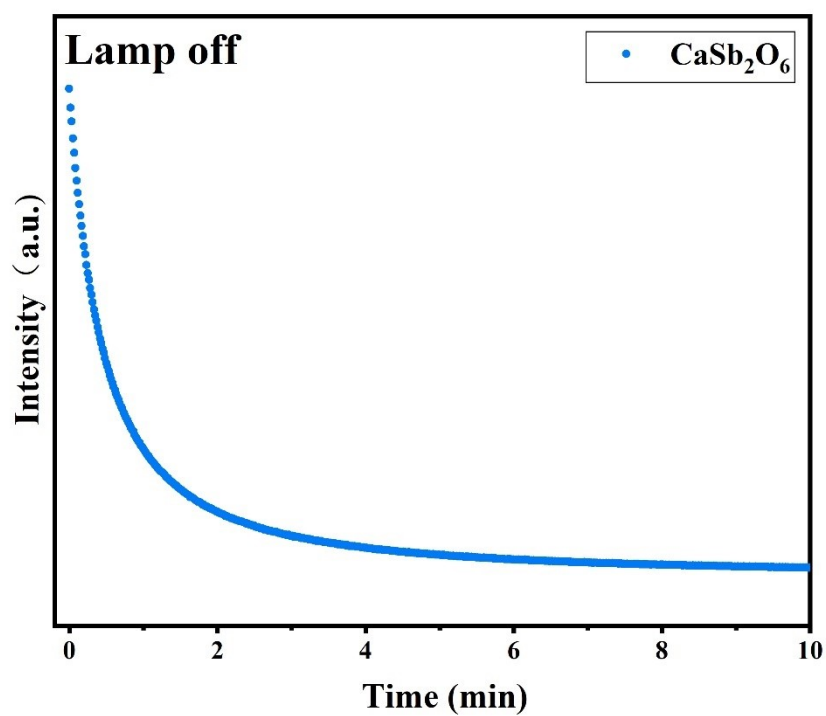


Fig. S9

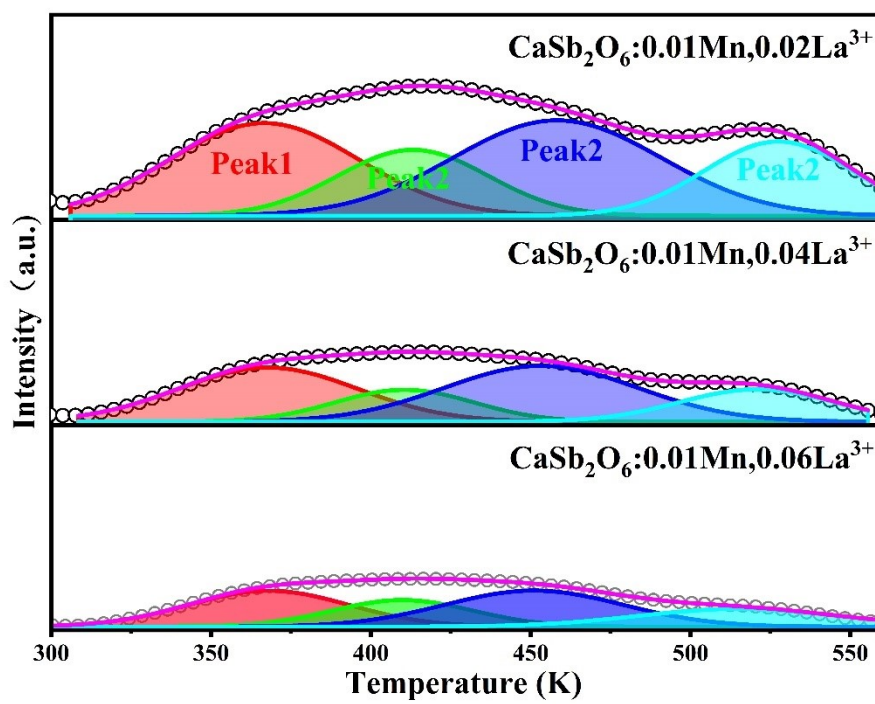


Fig. S10

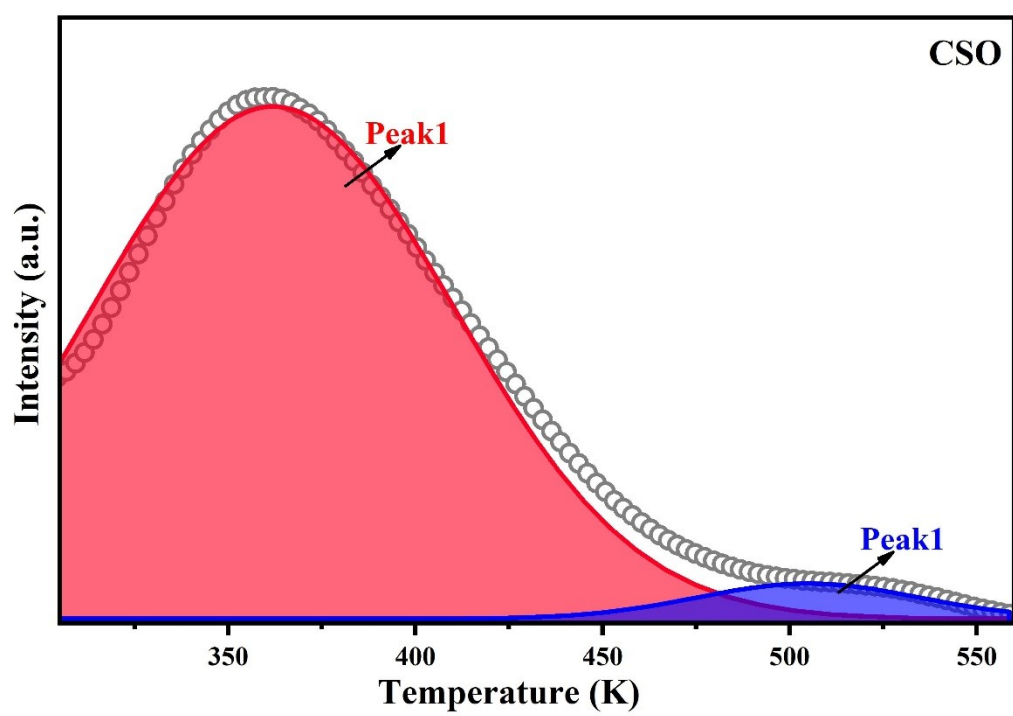


Fig. S11

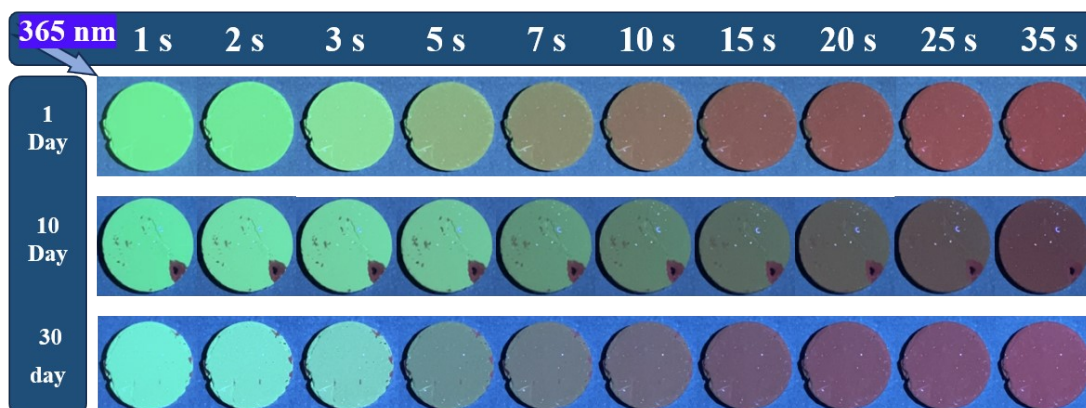


Fig. S12

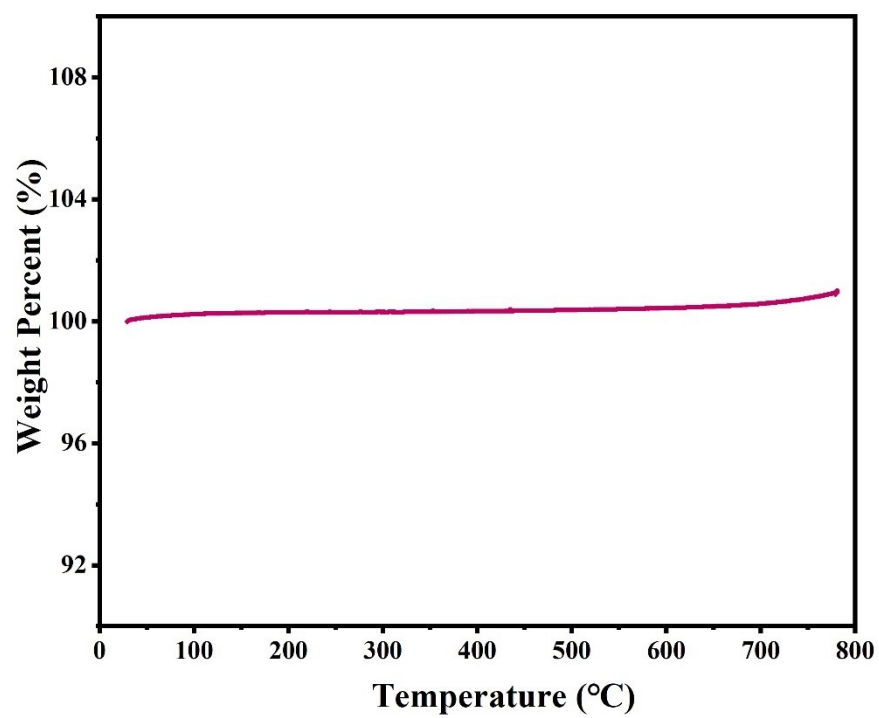


Fig. S13

Reference

1. H. Fan, Z. Lu, Y. Meng, P. Chen, L. Zhou, J. Zhao and X. He, Optical temperature sensor with superior sensitivity based on $\text{Ca}_2\text{LaSbO}_6\text{:Mn}^{4+}, \text{Eu}^{3+}$ phosphor, *Opt. Laser. Technol.*, 2022, **148**, 107804.
2. Y. Song, N. Guo, J. Li, R. Ouyang, Y. Miao and B. Shao, Photoluminescence and temperature sensing of lanthanide Eu^{3+} and transition metal Mn^{4+} dual-doped antimoniate phosphor through site-beneficial occupation, *Ceram. Int.*, 2020, **46**, 22164-22170.
3. M. Song, L. Wang, J. Wang and P. Du, Constructing double perovskite $\text{Eu}^{3+}/\text{Mn}^{4+}$ -codoped $\text{La}_2\text{Mg}_{1.33}\text{Ta}_{0.67}\text{O}_6$ phosphors for high sensitive dual-mode optical thermometers, *J. Luminescence*, 2022, **252**, 119347.
4. H. Zhou, N. Guo, M. Zhu, J. Li, Y. Miao and B. Shao, Photoluminescence and ratiometric optical thermometry in $\text{Mn}^{4+}/\text{Eu}^{3+}$ dual-doped phosphor via site-favorable occupation, *J. Luminescence*, 2020, **224**, 117311.
5. X. He, P. Fan, Y. Chen, Y. Guo, X. Liu and L. Li, Versatile $\text{Ca}_2\text{LnSbO}_6\text{:Eu}^{3+}/\text{Mn}^{4+}$ ($\text{Ln} = \text{La}$, Y, Gd, and Lu) phosphors for temperature sensing and plant growth LED, *J. Am. Ceram. Soc.*, 2023, **106**, 3568-3583.
6. B. Zhu, Q. Yang, W. Zhang, S. Cui, B. Yang, Q. Wang, S. Li and D. Zhang, A high sensitivity dual-mode optical thermometry based on charge compensation in $\text{ZnTiO}_{(3)}\text{:M}$ ($\text{M} = \text{Eu}^{(3+)}, \text{Mn}^{(4+)}$) hexagonal prisms, *Spectrochim. Acta. A. Mol. Biomol. Spectrosc.*, 2022, **274**, 121101.
7. Y. Wei, H. Yang, Z. Gao, Y. Liu, G. Xing, P. Dang, A. A. A. Kheraif, G. Li, J. Lin and R. S. Liu, Strategies for Designing Antithermal-Quenching Red Phosphors, *Adv Sci (Weinh)*,

2020, **7**, 1903060.

8. L. Y. Shi, D. Zhao, R. J. Zhang, Q. X. Yao and W. Liu, A new optical temperature sensor based on the fluorescence intensity ratio of Mn^{2+} and Mn^{4+} , *J. Am. Ceram. Soc.*, 2022, **105**, 7479-7491.
9. L. Dong, L. Zhang, Y. Jia, Y. Xu, S. Yin and H. You, $\text{ZnGa}_{(2-y)}\text{Al}_{(y)}\text{O}_{(4)}:\text{Mn}^{(2+)},\text{Mn}^{(4+)}$ Thermochromic Phosphors: Valence State Control and Optical Temperature Sensing, *Inorg Chem*, 2020, **59**, 15969-15976.
10. J. Hu, E. Song, Y. Zhou, S. Zhang, S. Ye, Z. Xia and Q. Zhang, Non-stoichiometric defect-controlled reduction toward mixed-valence Mn-doped hexaaluminates and their optical applications, *J. Mater. Chem. C*, 2019, **7**, 5716-5723.
11. D. Huang, Q. Ouyang, B. Liu, B. Chen, Y. Wang, C. Yuan, H. Xiao, H. Lian and J. Lin, $\text{Mn}^{(2+)}/\text{Mn}^{(4+)}$ co-doped $\text{LaM}_{(1-x)}\text{Al}_{(11-y)}\text{O}_{(19)}$ ($\text{M} = \text{Mg}, \text{Zn}$) luminescent materials: electronic structure, energy transfer and optical thermometric properties, *Dalton T*, 2021, **50**, 4651-4662.
12. X. Shi, Y. Xue, Q. Mao, L. Pei, X. Li, M. Liu, Q. Zhang and J. Zhong, Eu^{3+} Single-Doped Phosphor with Antithermal Quenching Behavior and Multicolor-Tunable Properties for Luminescence Thermometry, *Inorg. Chem*, 2023, **62**, 893-903.