

## Supplementary material

### New Cyclometalated Ru(II) Polypyridyl Photosensitizers Trigger Oncosis in Cancer Cells by Inducing Damage of Cellular Membranes

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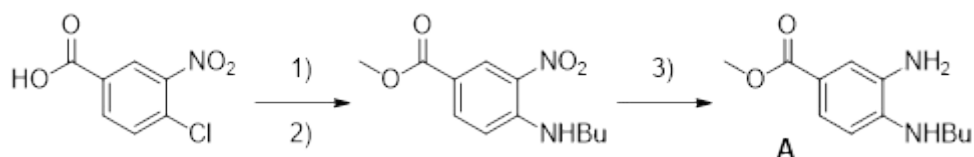
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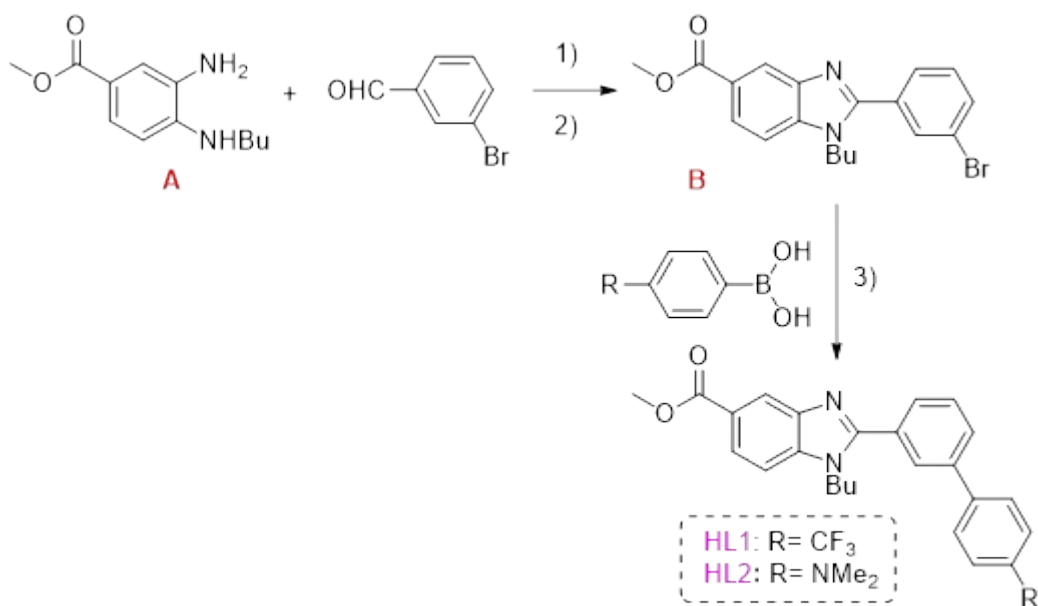
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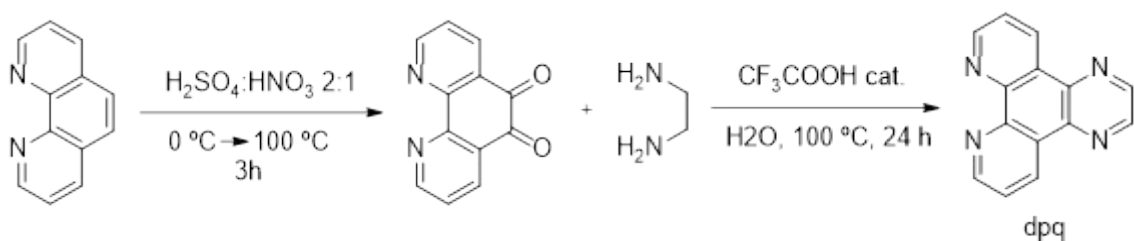
## Synthetic procedures



**Scheme S1.** Synthetic procedure for A. Reaction conditions: 1)  $\text{H}_2\text{SO}_4$  (cat), MeOH, 65 °C, overnight; 2)  $\text{BuNH}_2$ ,  $\text{Et}_3\text{N}$ ,  $\text{CH}_2\text{Cl}_2$ , 72 h,  $\text{N}_2$  atm., r.t.; 3) Zn (powder)  $\text{HCOONH}_4$ , AcOEt/MeOH  $\text{N}_2$  atm., 24 h, 50 °C.

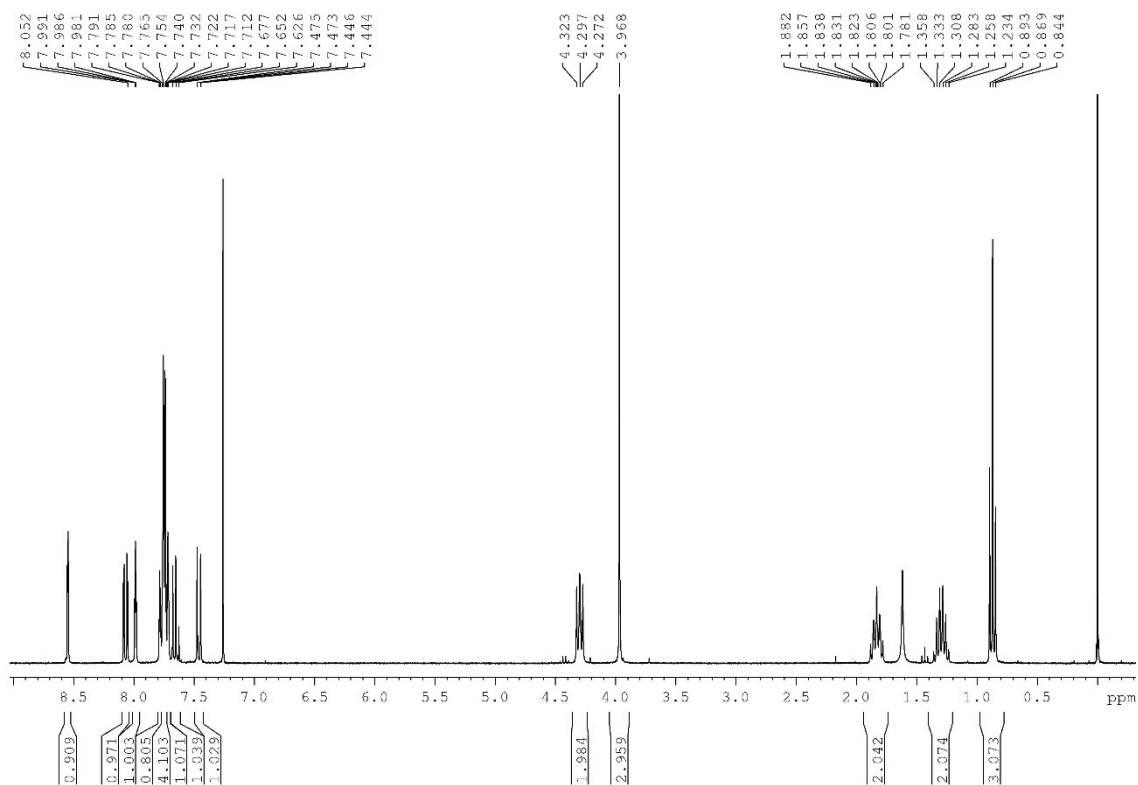


**Scheme S2.** Synthetic procedure for **HL1** and **HL2**. Reaction conditions: 1)  $\text{NaHSO}_3$ ,  $\text{H}_2\text{O}$ , 90 °C, 1 h; 2) EtOH:  $\text{H}_2\text{O}$  1:1, 90 °C overnight; 3)  $\text{Pd}(\text{PPh}_3)_4$ ,  $\text{K}_2\text{CO}_3$ , Toluene: $\text{H}_2\text{O}$  4:2, MW, 180 °C, 1 h.<sup>1,2</sup>

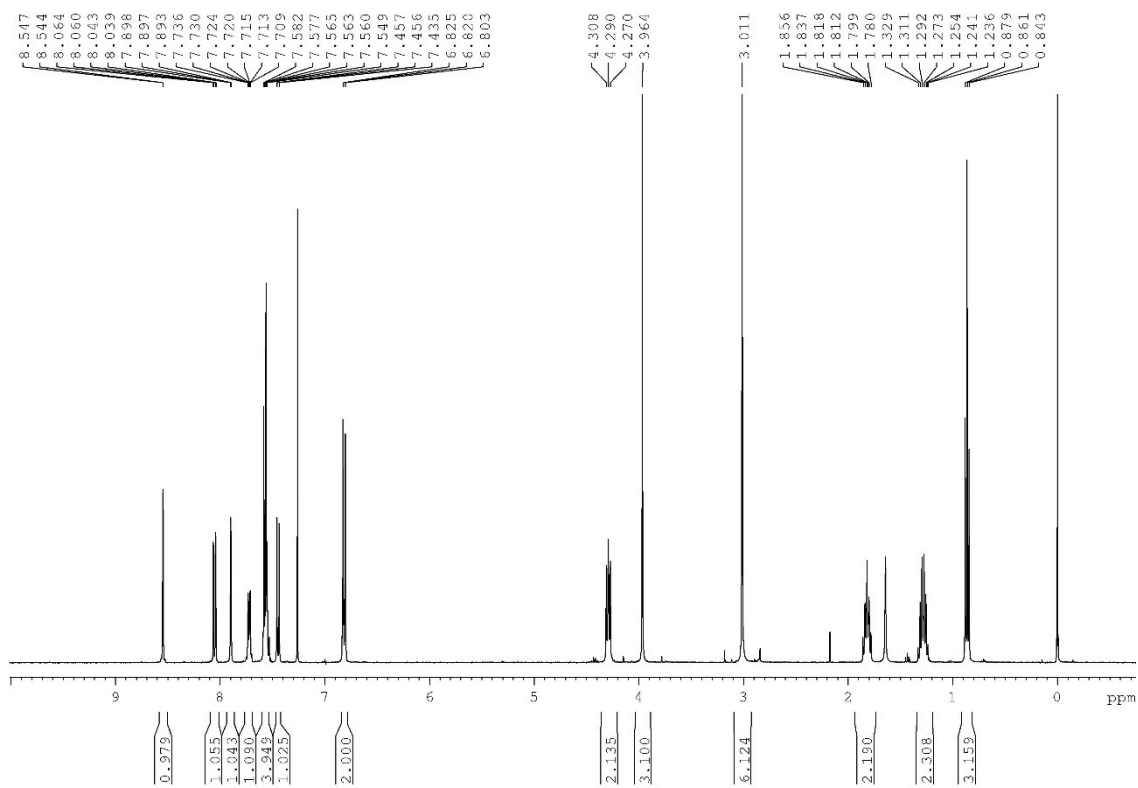


**Scheme S2.** Synthesis procedure of **dpq**.<sup>3</sup>

## Characterization of ligands and complexes



**Figure S1.  $^1\text{H}$  NMR of HL1 in  $\text{CDCl}_3$**



**Figure S2.  $^1\text{H}$  NMR of HL2 in  $\text{CDCl}_3$**

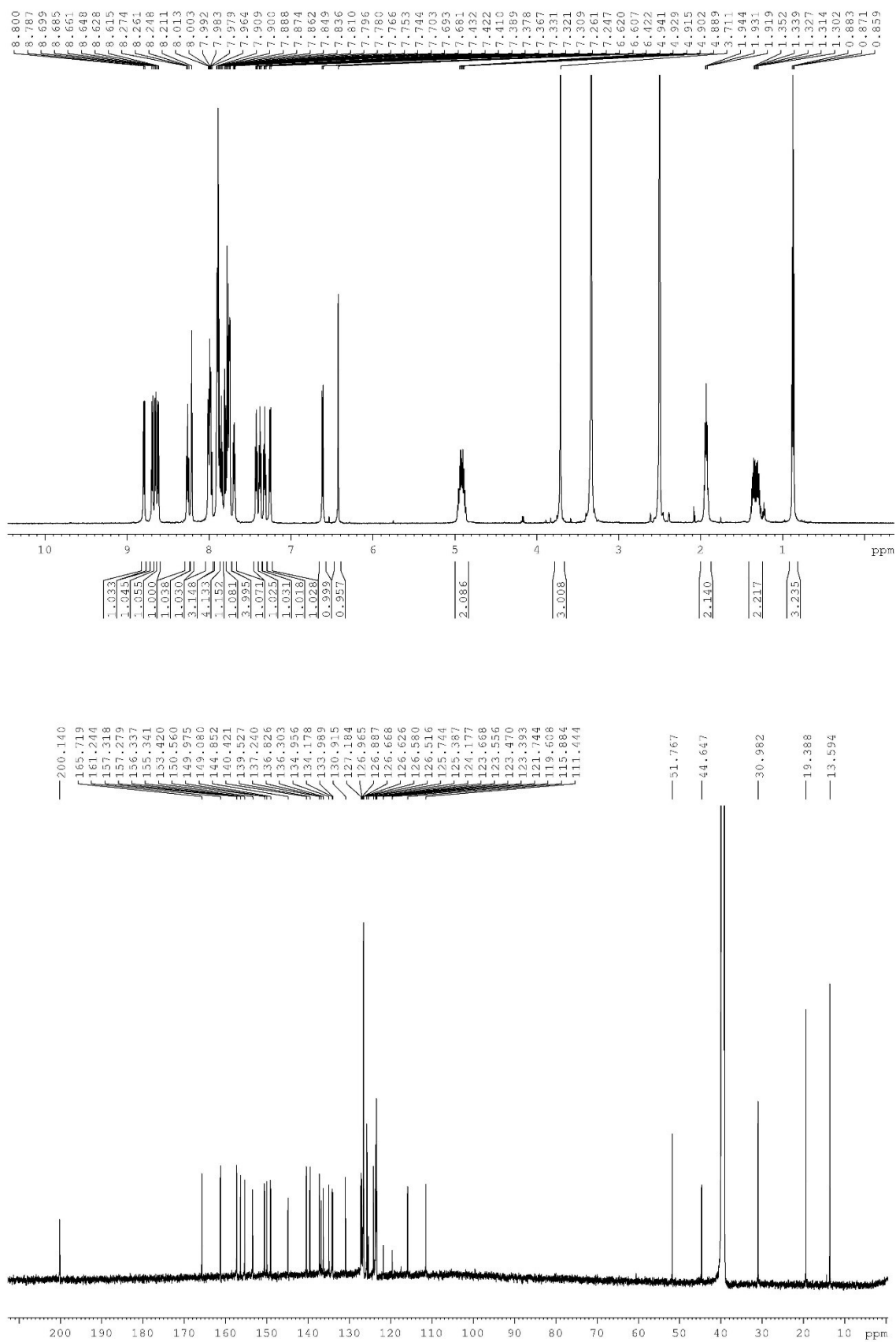
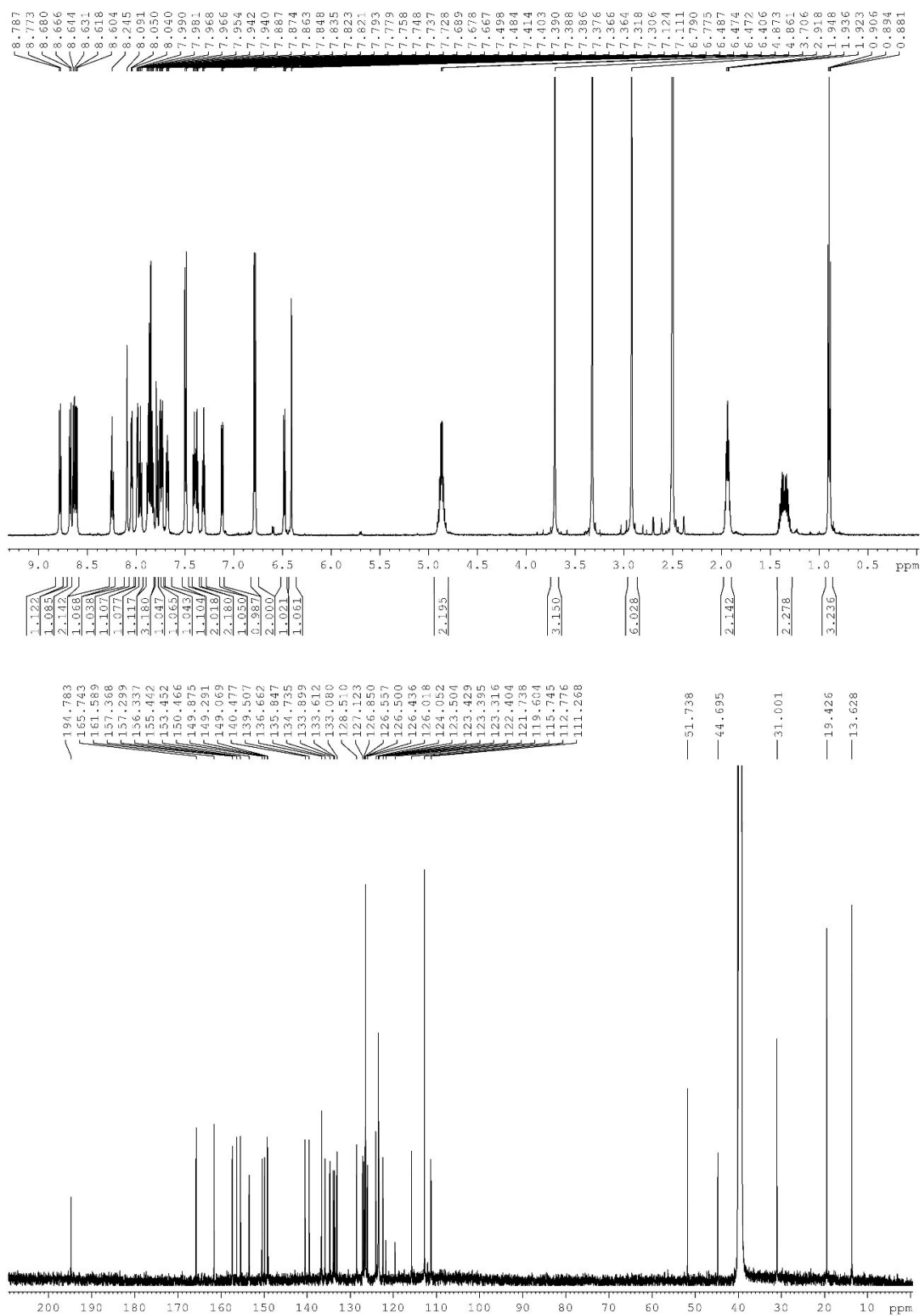
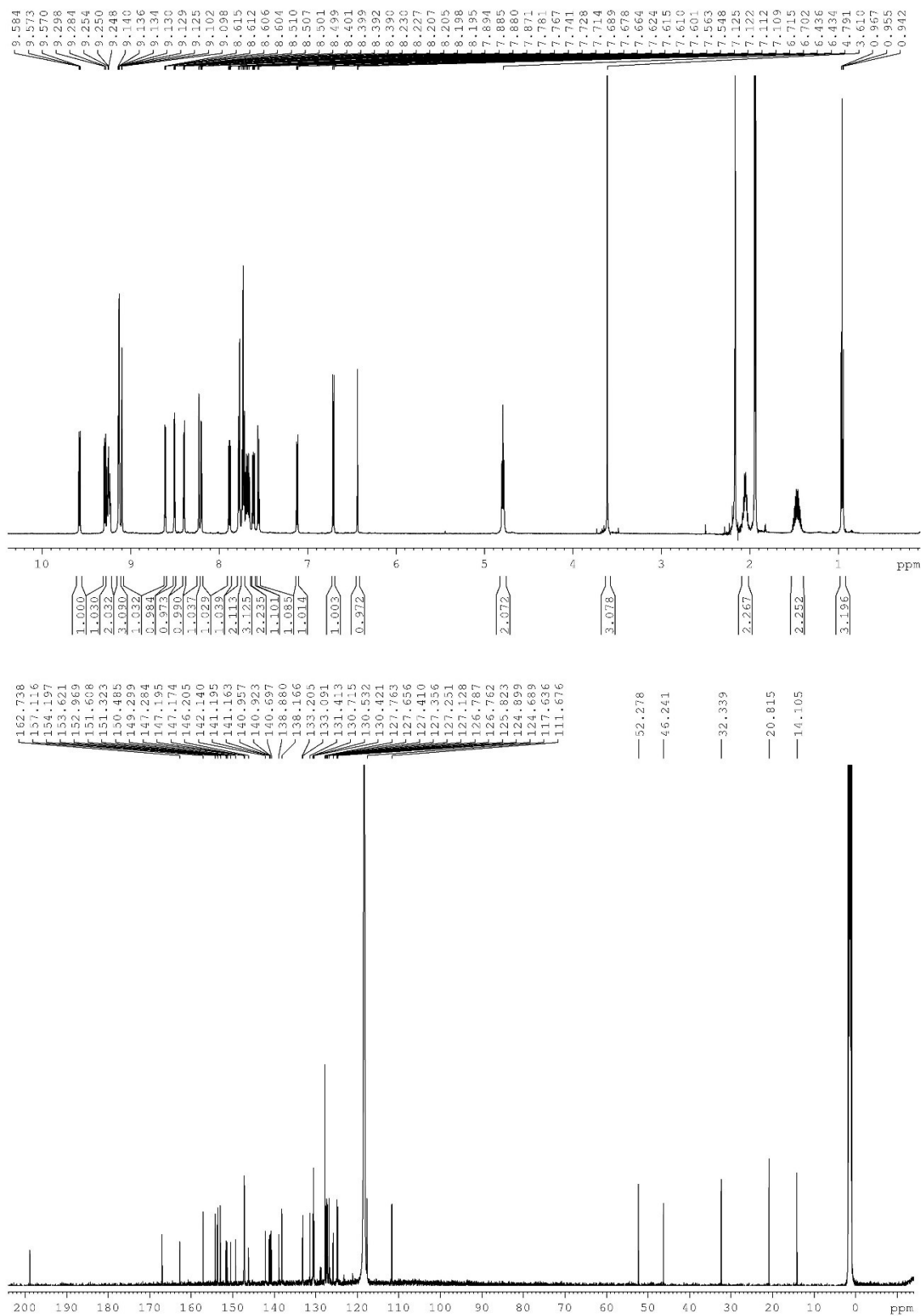


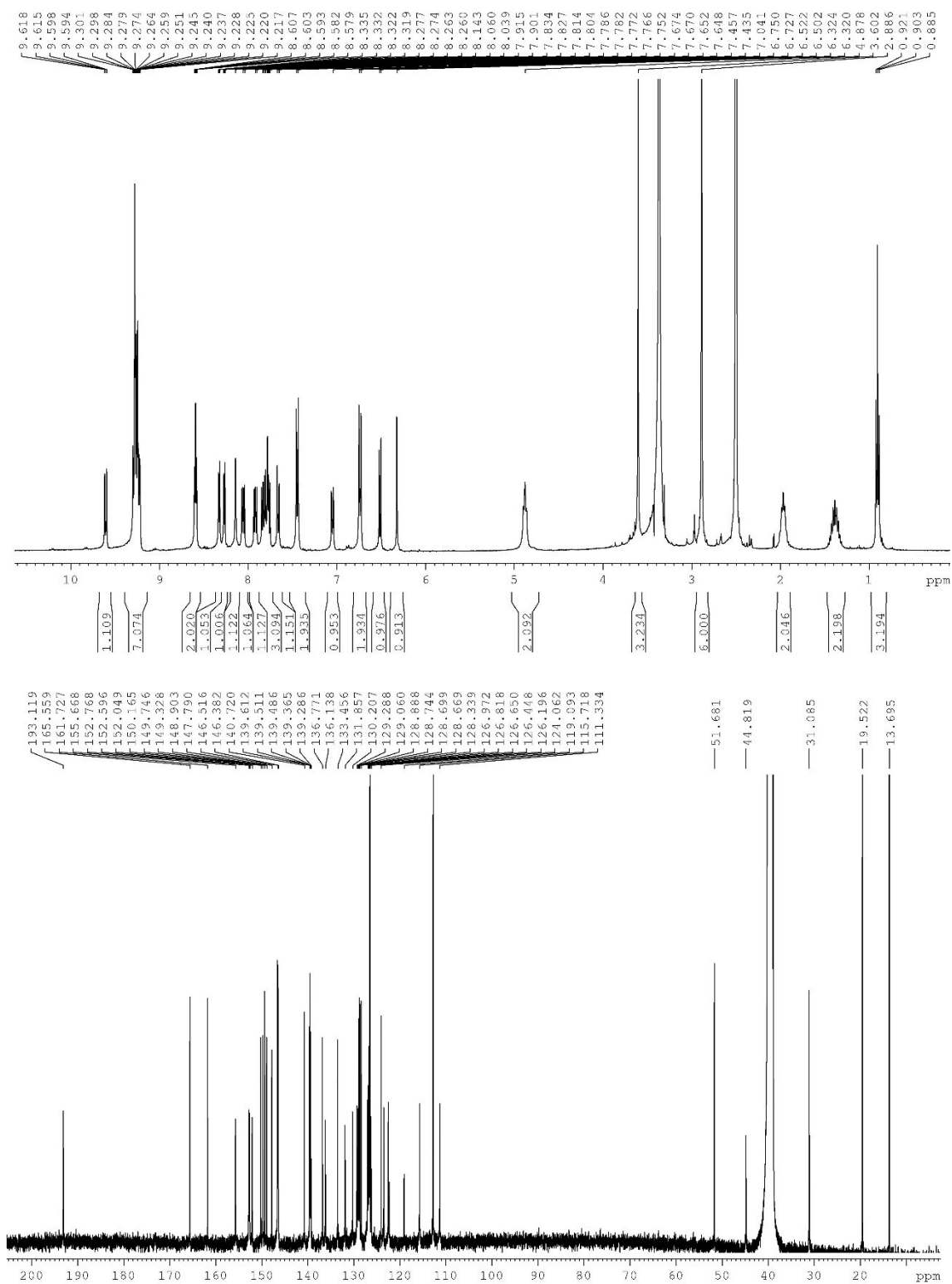
Figure S3.  $^1\text{H}$  (top) and  $^{13}\text{C}$  NMR (bottom) spectra of compound **Ru1** in DMSO- $d_6$



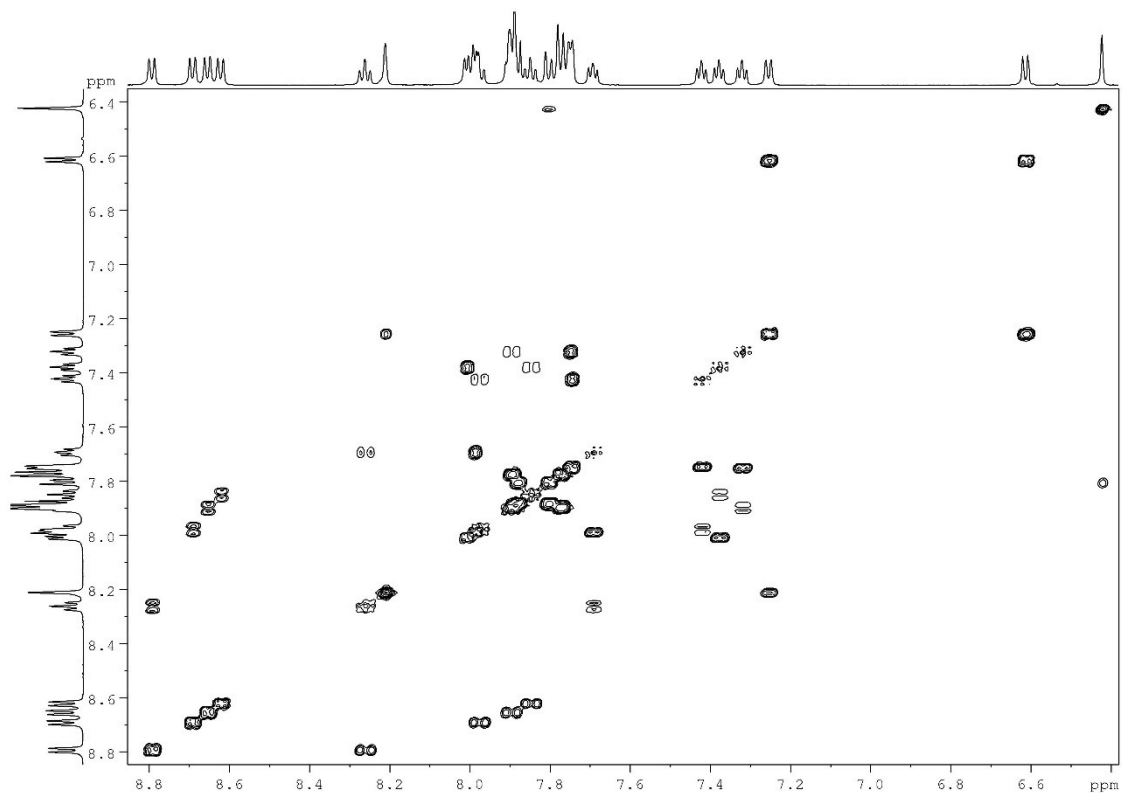
**Figure S4.** <sup>1</sup>H (top) and <sup>13</sup>C NMR (bottom) spectra of compound **Ru2** in DMSO-*d*<sub>6</sub>.



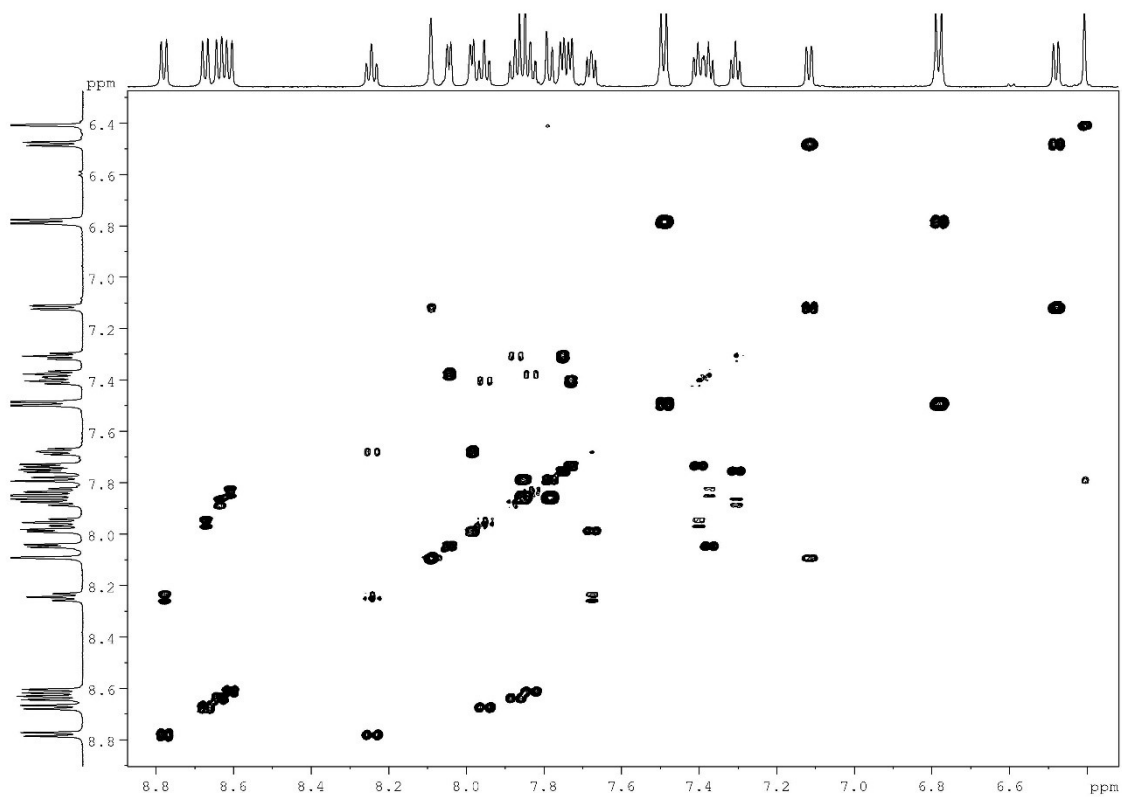
**Figure S5.** <sup>1</sup>H (top) and <sup>13</sup>C-NMR (bottom) spectra of compound **Ru3** in CD<sub>3</sub>CN



**Figure S6.** <sup>1</sup>H (top) and <sup>13</sup>C NMR (bottom) spectra of compound **Ru4** in DMSO-*d*<sub>6</sub>

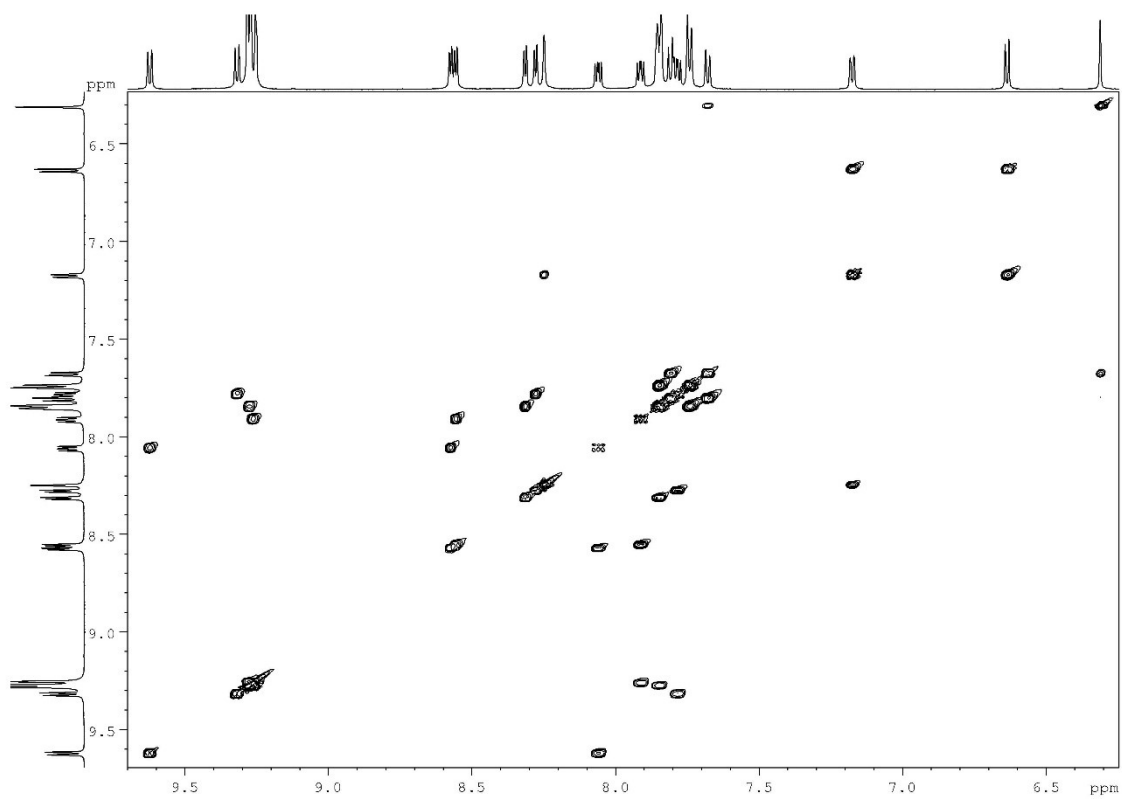


**Figure S7.**  $\{^1\text{H}-^1\text{H}\}$ -COSY NMR spectrum of compound **Ru1** in  $\text{DMSO}-d_6$

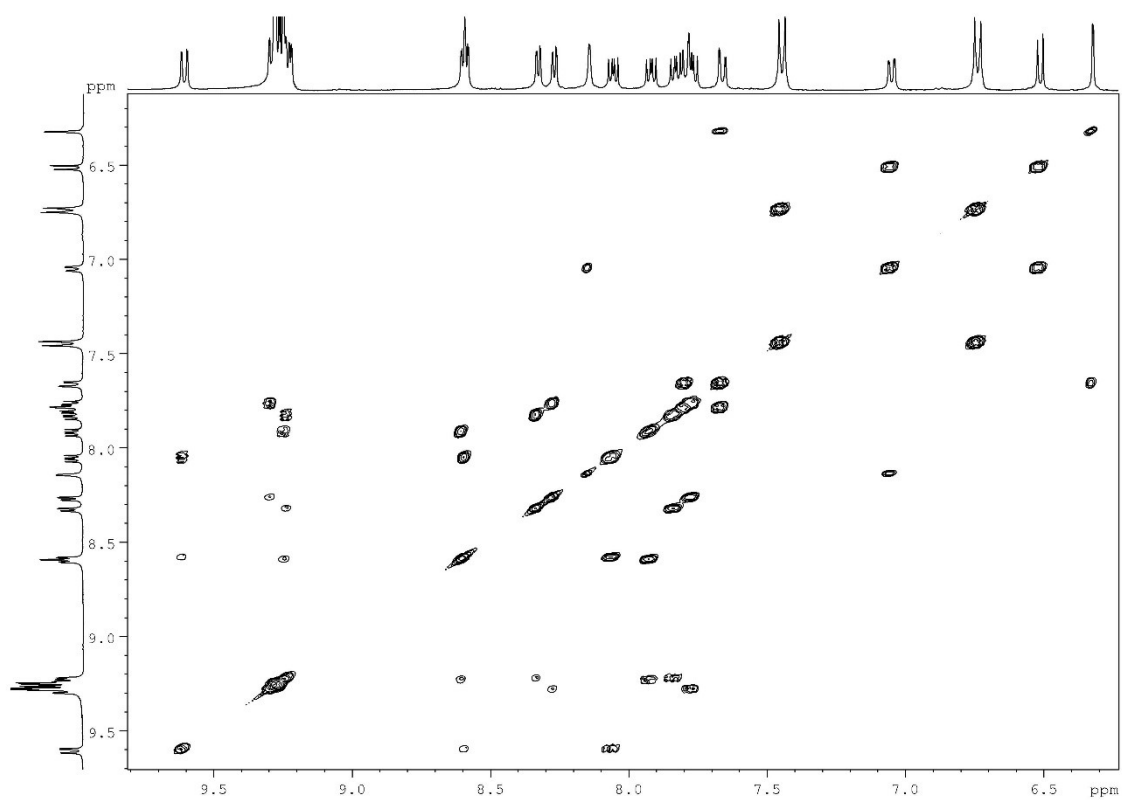


**Figure S8.**  $\{^1\text{H}-^1\text{H}\}$ -COSY NMR spectrum of compound **Ru2** in  $\text{DMSO}-d_6$





**Figure S9.**  $\{^1\text{H}-^1\text{H}\}$ -COSY NMR spectrum of compound **Ru3** in  $\text{DMSO}-d_6$



**Figure S10.**  $\{^1\text{H}-^1\text{H}\}$ -COSY NMR spectrum of compound **Ru4** in  $\text{DMSO}-d_6$

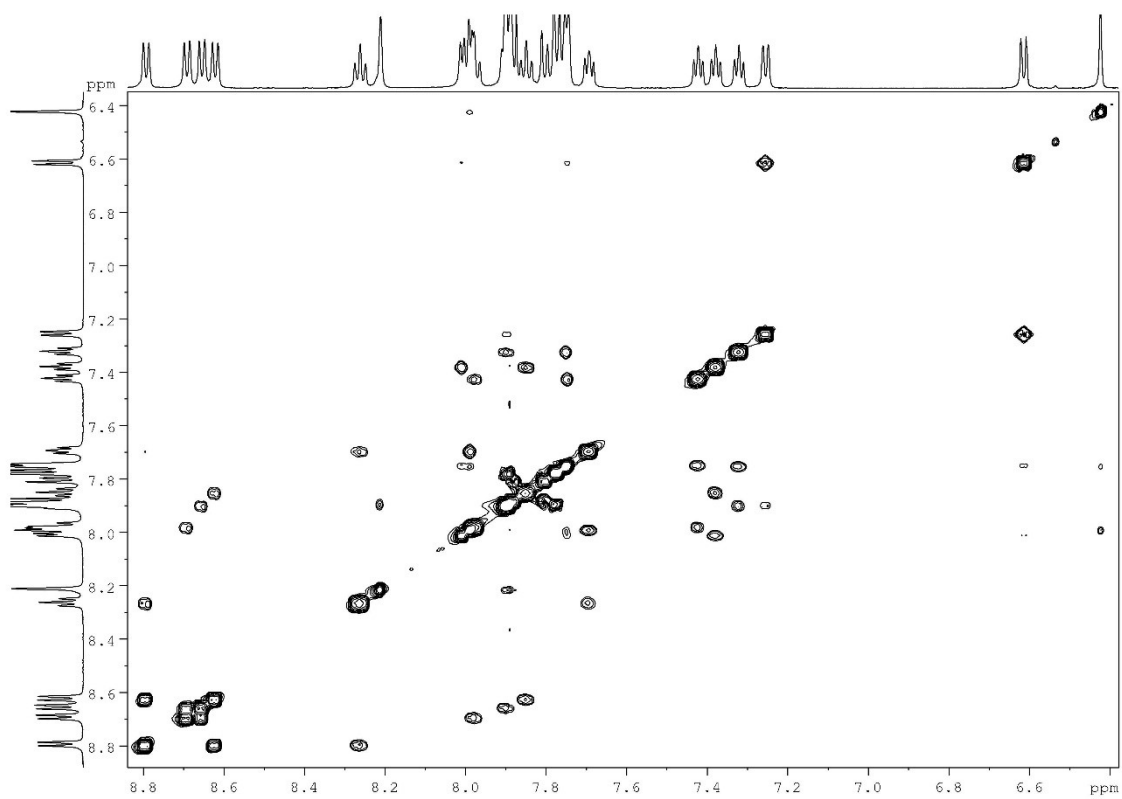


Figure S11.  $\{^1\text{H}-^1\text{H}\}$ -NOESY NMR spectrum of compound **Ru1** in  $\text{DMSO}-d_6$

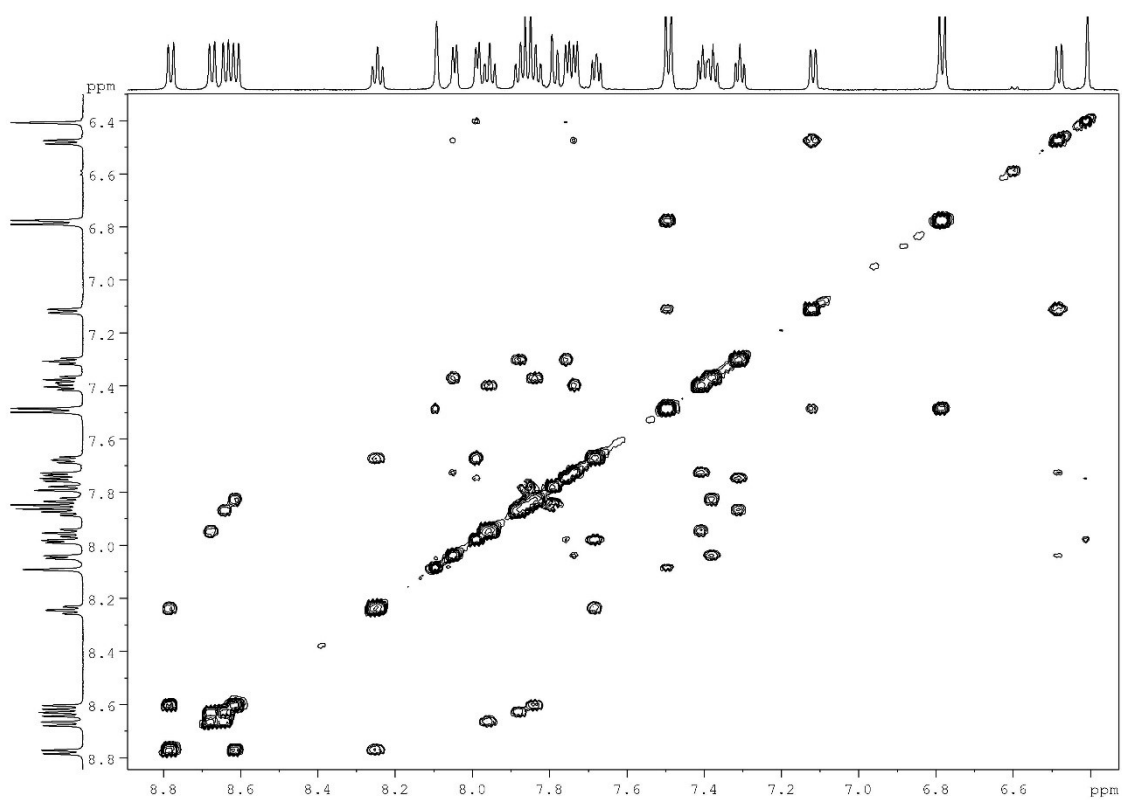
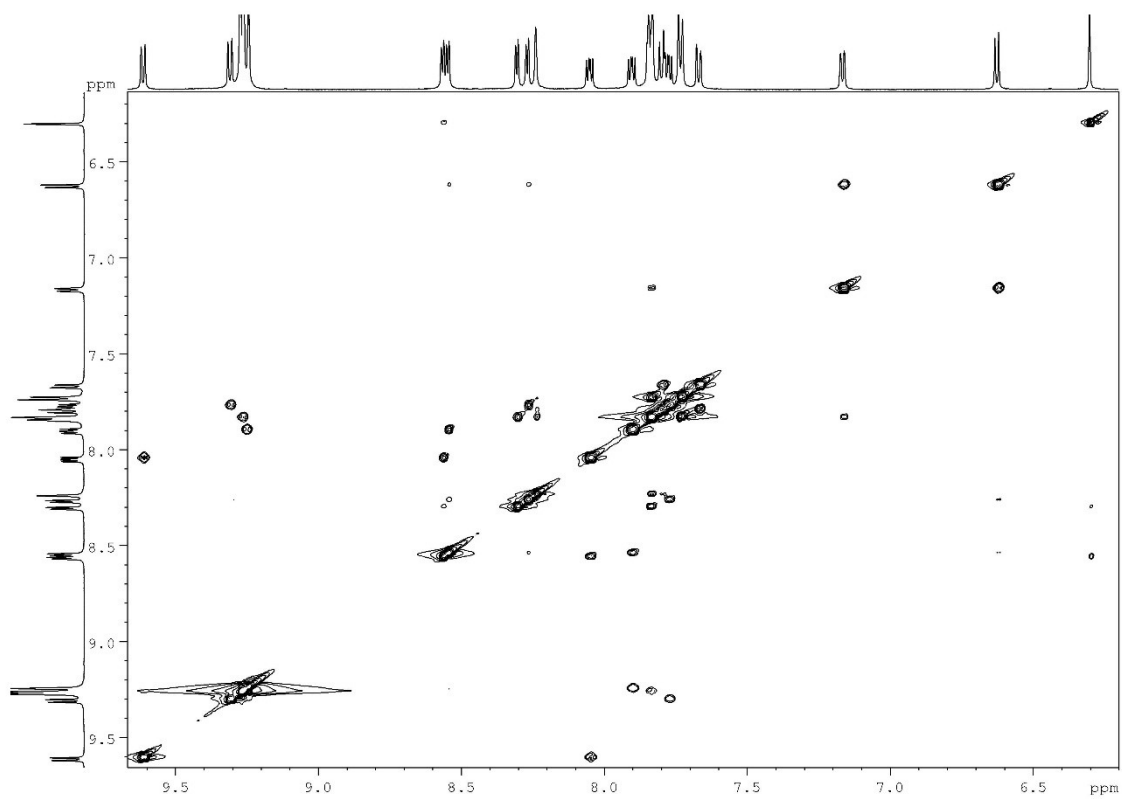
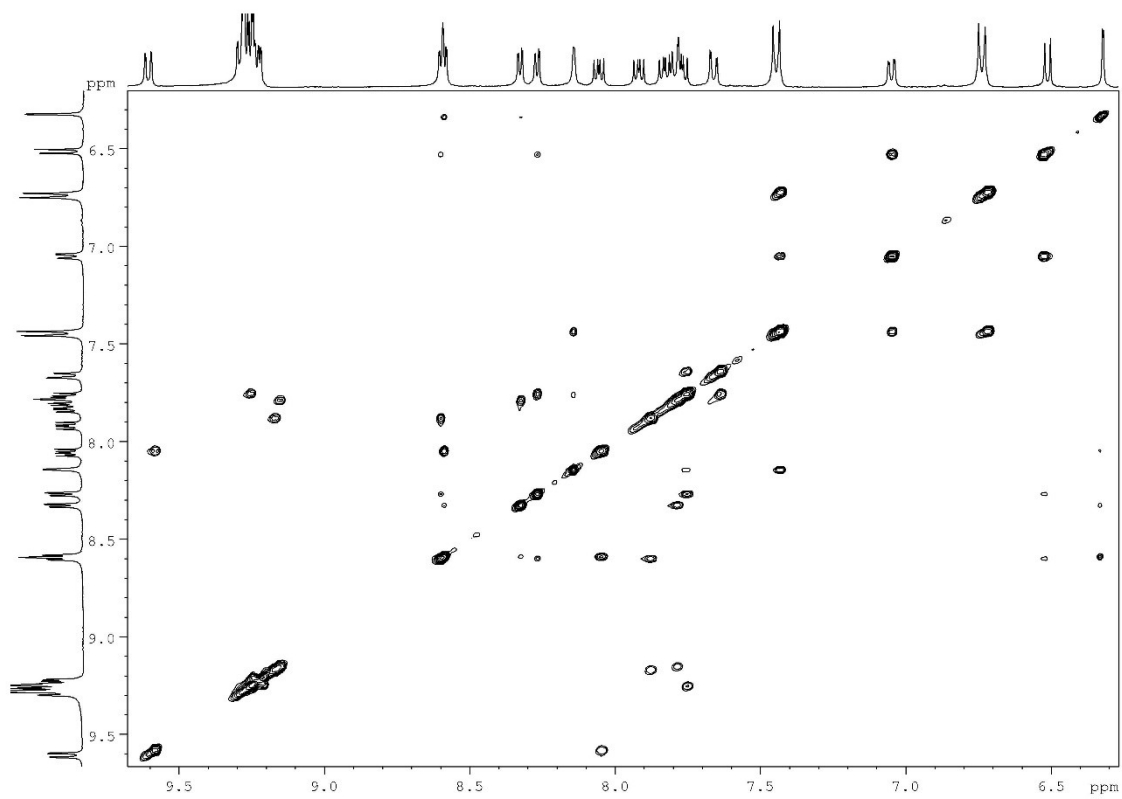


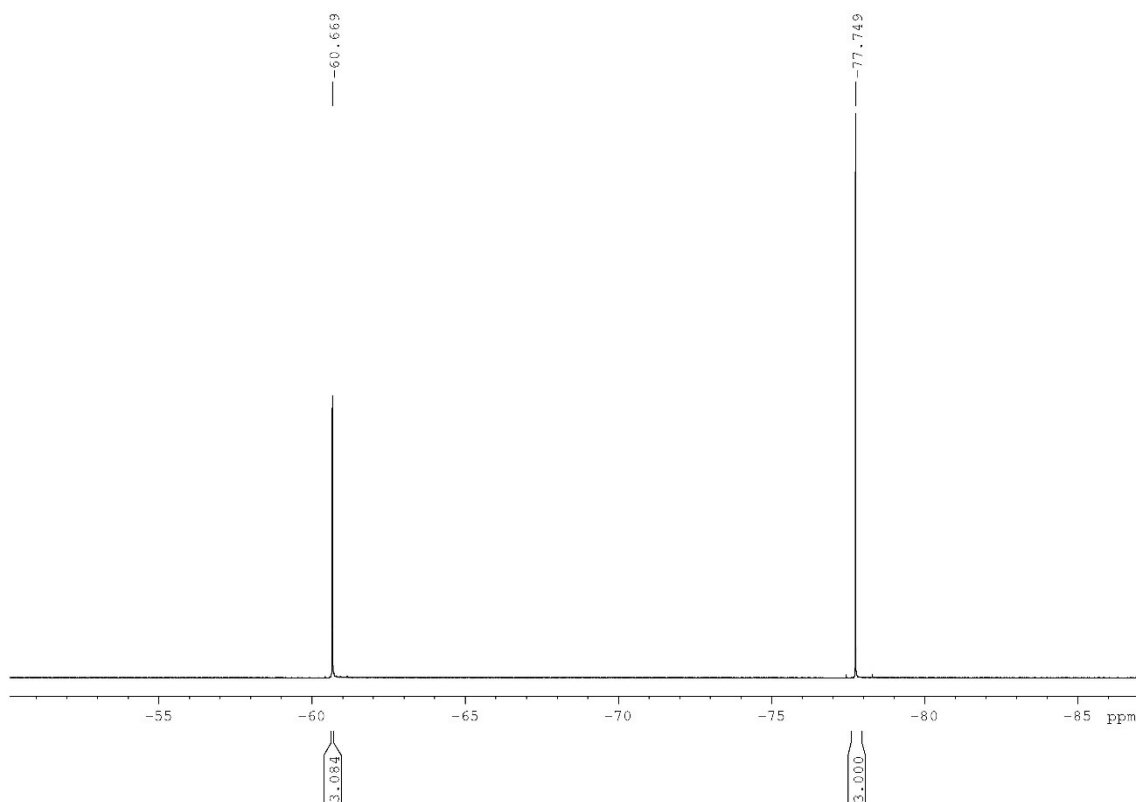
Figure S12.  $\{^1\text{H}-^1\text{H}\}$ -NOESY NMR spectrum of compound **Ru2** in  $\text{DMSO}-d_6$



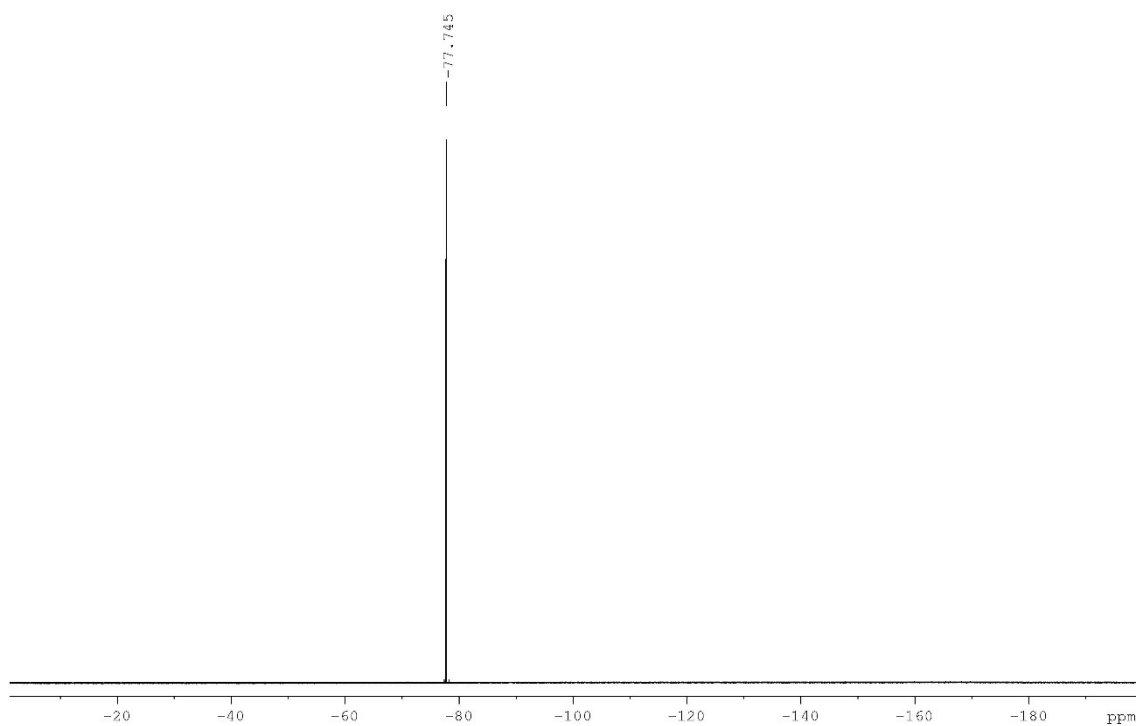
**Figure S13.**  $\{^1\text{H}-^1\text{H}\}$ -NOESY NMR spectrum of compound **Ru3** in  $\text{DMSO}-d_6$



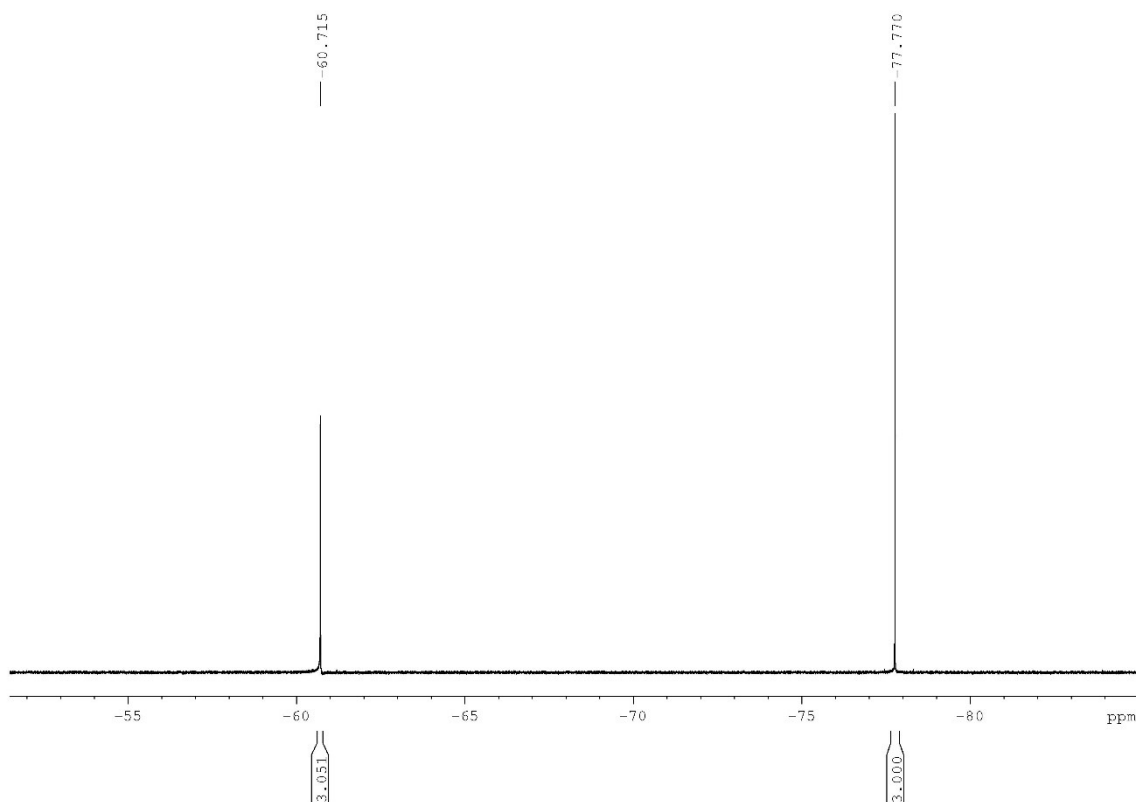
**Figure S14.**  $\{^1\text{H}-^1\text{H}\}$ -NOESY NMR spectrum of compound **Ru4** in  $\text{DMSO}-d_6$



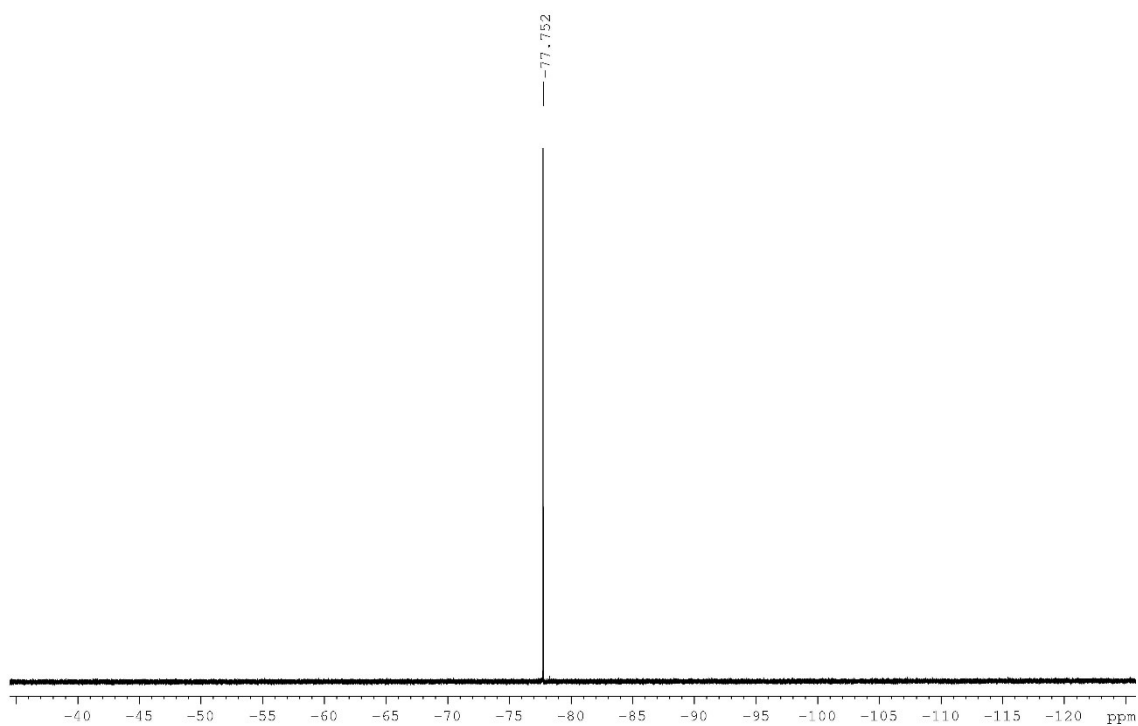
**Figure S15.**  $^{19}\text{F}$  NMR spectrum of compound **Ru1** in  $\text{DMSO-}d_6$



**Figure S16.**  $^{19}\text{F}$  NMR spectrum of compound **Ru2** in  $\text{DMSO-}d_6$



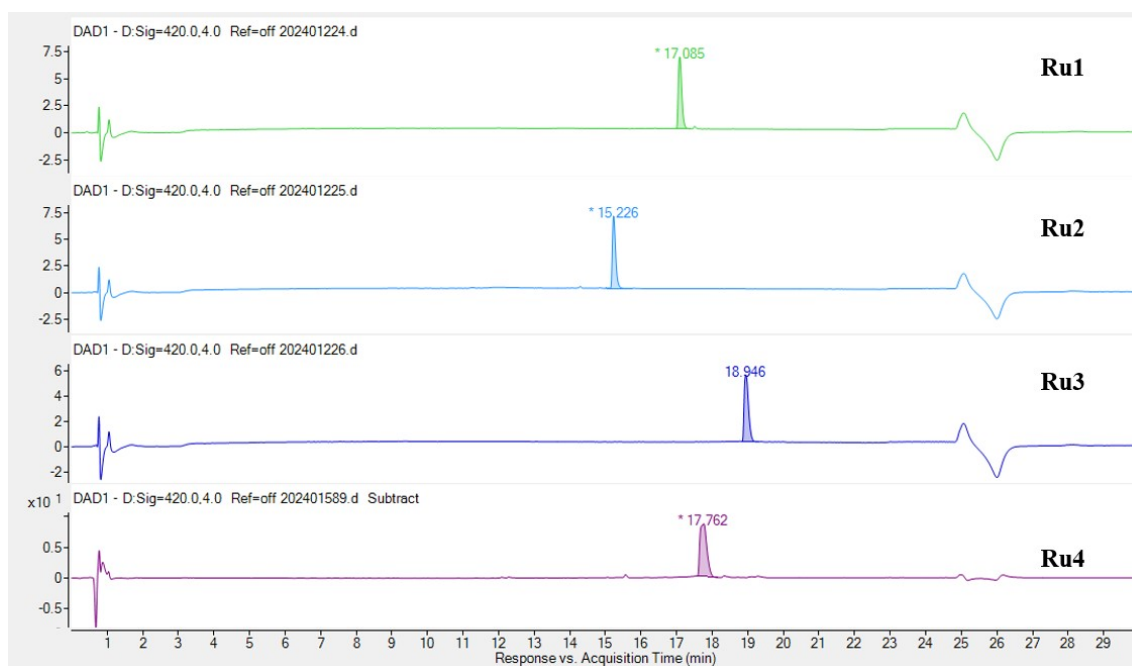
**Figure S17.**  $^{19}\text{F}$  NMR spectrum of compound **Ru3** in  $\text{DMSO-}d_6$



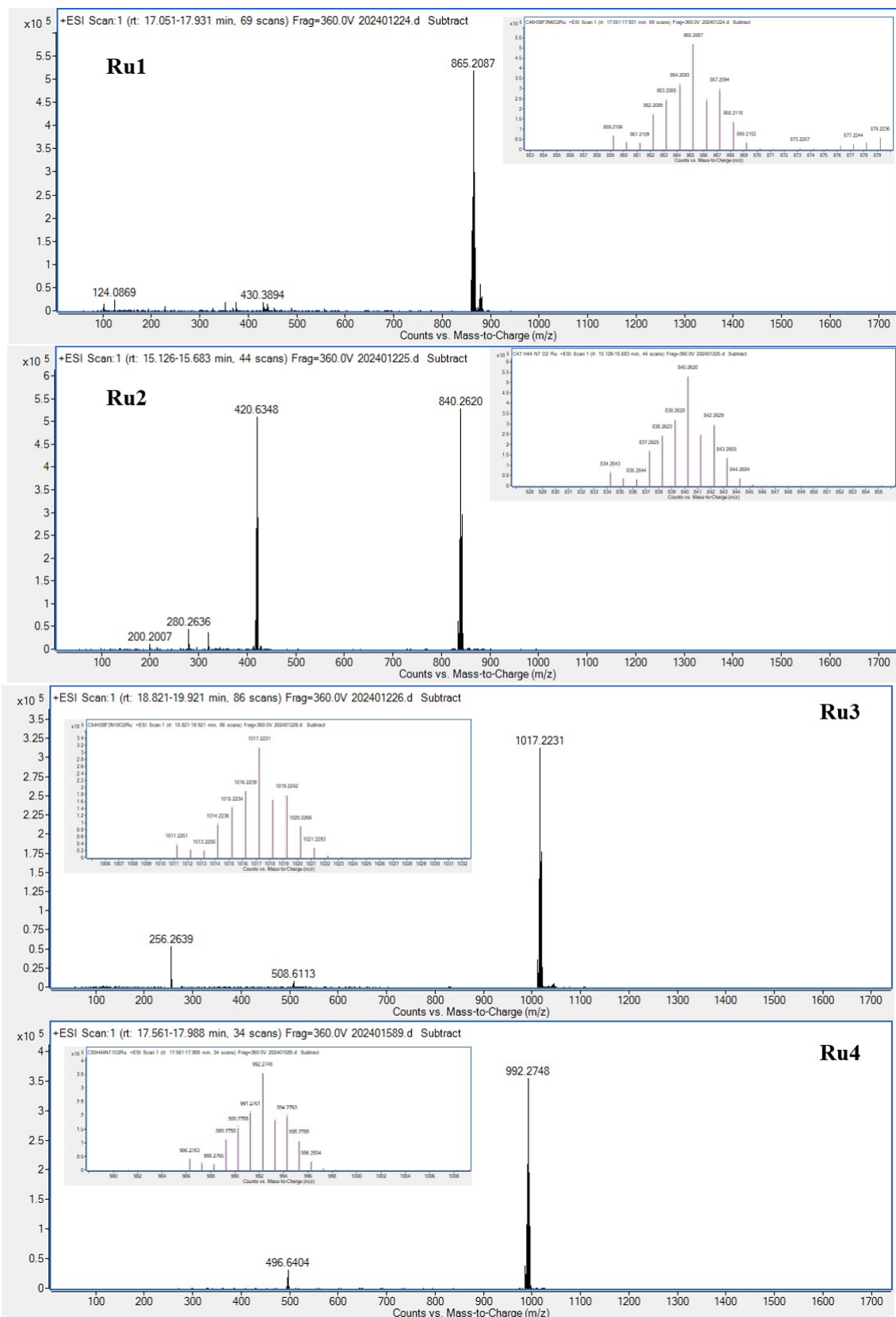
**Figure S18.**  $^{19}\text{F}$  NMR spectrum of compound **Ru4** in  $\text{DMSO-}d_6$

**Table S1.** HPLC method

Time (min)	0.1 % formic acid in H <sub>2</sub> O	0.1 % formic acid in CH <sub>3</sub> CN	Flow (mL/min)
0	80	20	0.4
25	0	100	



**Figure S19.** HPLC chromatograms with UV detection at 420 nm of complexes **Ru1-**  
**Ru4**



**Figure S20.** Mass spectra of the 15-18 min peak of chromatograms of **Figure S19** corresponding to the complexes.

**Table S2.** Crystal data and structure refinement of complex **Ru2**

Empirical formula	$C_{47}H_{44}N_7O_2Ru.CF_3O_3S.1(CH_2Cl_2).0.5(C_4H_{10}O).1.75[H_2O]$	
Formula weight	1142.54	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	P-1	
Unit cell dimensions	$a = 13.5090(6) \text{ \AA}$	$\alpha = 84.3410(10)^\circ$
	$b = 14.0322(7) \text{ \AA}$	$\beta = 73.105(2)^\circ$
	$c = 15.9709(6) \text{ \AA}$	$\gamma = 62.2960(10)^\circ$
Volume	2562.3(2) Å <sup>3</sup>	
Z	2	
Density (calculated)	1.481 Mg/m <sup>3</sup>	
Absorption coefficient	0.521 mm <sup>-1</sup>	
F(000)	1177	
Crystal size	0.290 x 0.230 x 0.120 mm <sup>3</sup>	
Theta range for data collection	1.641 to 27.103°	
Index ranges	-17<=h<=17, -17<=k<=17, -20<=l<=20	
Reflections collected	184861	
Independent reflections	11300 [R(int) = 0.0273]	
Completeness to theta = 26.000°	100.0 %	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	11300 / 32 / 691	
Goodness-of-fit on F <sup>2</sup>	1.058	
Final R indices [I>2sigma(I)]	R1 = 0.0610, wR2 = 0.1684	
R indices (all data)	R1 = 0.0629, wR2 = 0.1704	
Largest diff. peak and hole	1.844 and -2.047 e.Å <sup>-3</sup>	



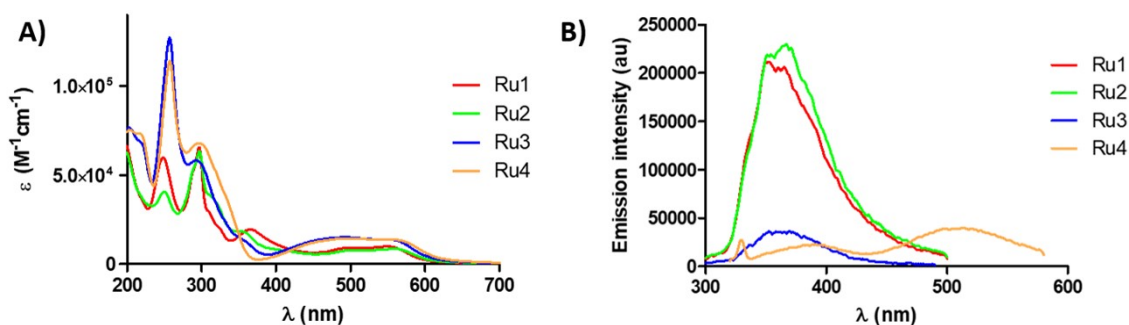
**Table S3.** Hydrogen bonds for **Ru2** [ $\text{\AA}$  and  $^\circ$ ]

	D-H...A	d(D-H)	d(H...A)	d(D...A)<(DHA)
C(11)-H(11A)...O(98)#1	0.99	2.52	3.395(7)	147.3
C(34)-H(34)...O(98)#2	0.95	2.53	3.407(6)	153.4
C(41)-H(41)...O(96)	0.95	2.38	3.273(6)	157.3
C(81)-H(81A)...O(96)#4	0.99	2.57	3.549(10)	170.9

Symmetry transformations used to generate equivalent atoms:

#1  $-x+1,-y+2,-z+1$  #2  $x+1,y-1,z$  #3  $x,y-1,z$  #4  $-x,-y+2,-z+1$

### Photophysical properties



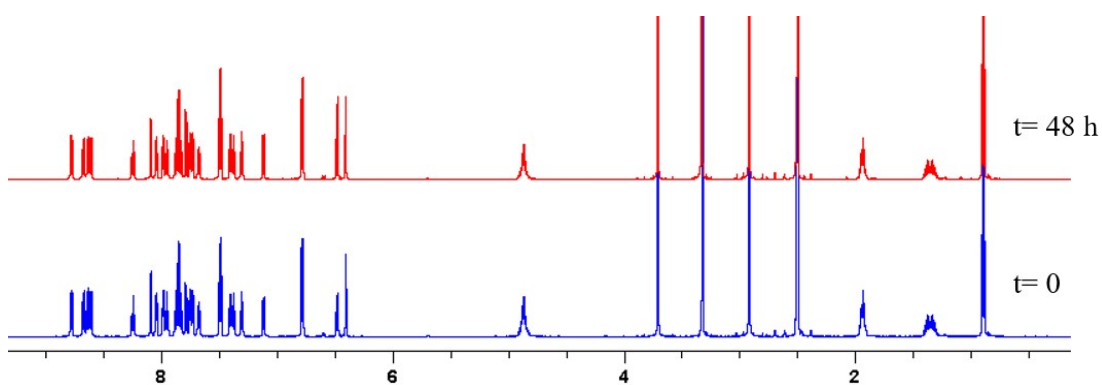
**Figure S21.** A) UV/vis spectra of compounds **Ru1-Ru4** ( $10 \mu\text{M}$ ) in acetonitrile. B) Emission spectra of **Ru1-Ru4** ( $10 \mu\text{M}$ ) in acetonitrile. **Ru1-Ru3**:  $\lambda_{ex} = 255 \text{ nm}$ ; **Ru4**:  $\lambda_{ex} = 300 \text{ nm}$ .

**Table S4.** Absorption wavelengths ( $\lambda_{abs}$ ) and molar extinction coefficient ( $\epsilon$ ) of absorption spectra of the complexes **Ru1-Ru4** in acetonitrile and water (1% DMSO)

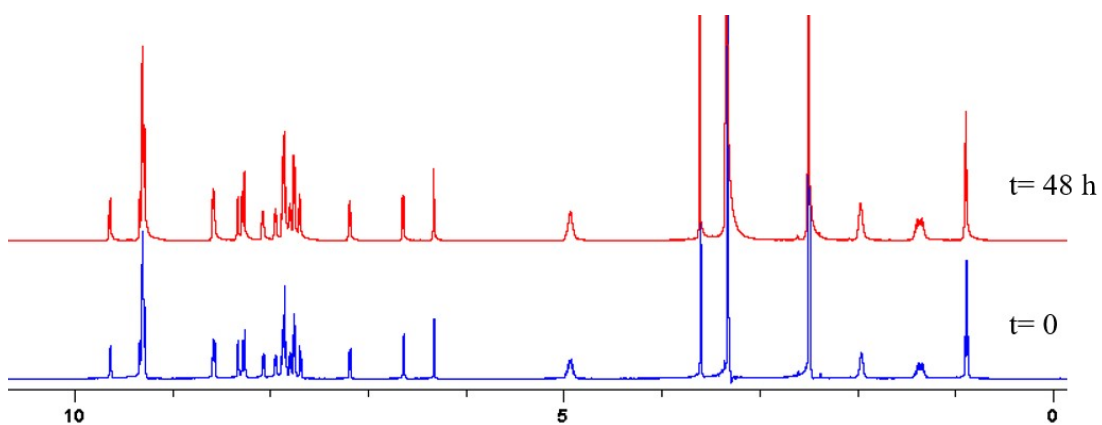
Complex	Solvent	$\lambda$ (nm) ( $\epsilon$ , $\text{M}^{-1}\text{cm}^{-1}$ )
<b>Ru1</b>	H <sub>2</sub> O (1% DMSO)	298 (68710), 373 (21290), 494 (11060), 549 (11960)
	CH <sub>3</sub> CN	297 (65610), 370 (19150), 500 (9130), 550 (9940)
<b>Ru2</b>	H <sub>2</sub> O (1% DMSO)	299 (73960), 351 (24340), 498 (10240), 560 (11020)

	CH <sub>3</sub> CN	297 (63470), 351 (18520), 504 (7810), 558 (8630)
<b>Ru3</b>	H <sub>2</sub> O (1% DMSO)	258 (117870), 295 (59550), 494 (16480), 548 (16590)
	CH <sub>3</sub> CN	257 (127210), 293 (58210), 498 (15080), 551 (14000)
<b>Ru4</b>	H <sub>2</sub> O (1% DMSO)	259 (10538), 296 (65380), 493 (15020), 562 (14760)
	CH <sub>3</sub> CN	257 (114280), 302 (66860), 500 (14570), 554 (13810)

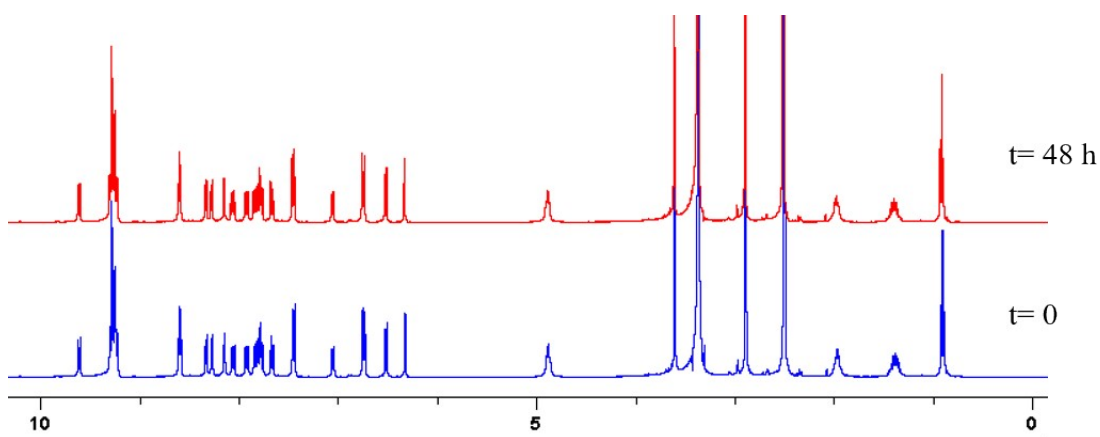
### Stability studies



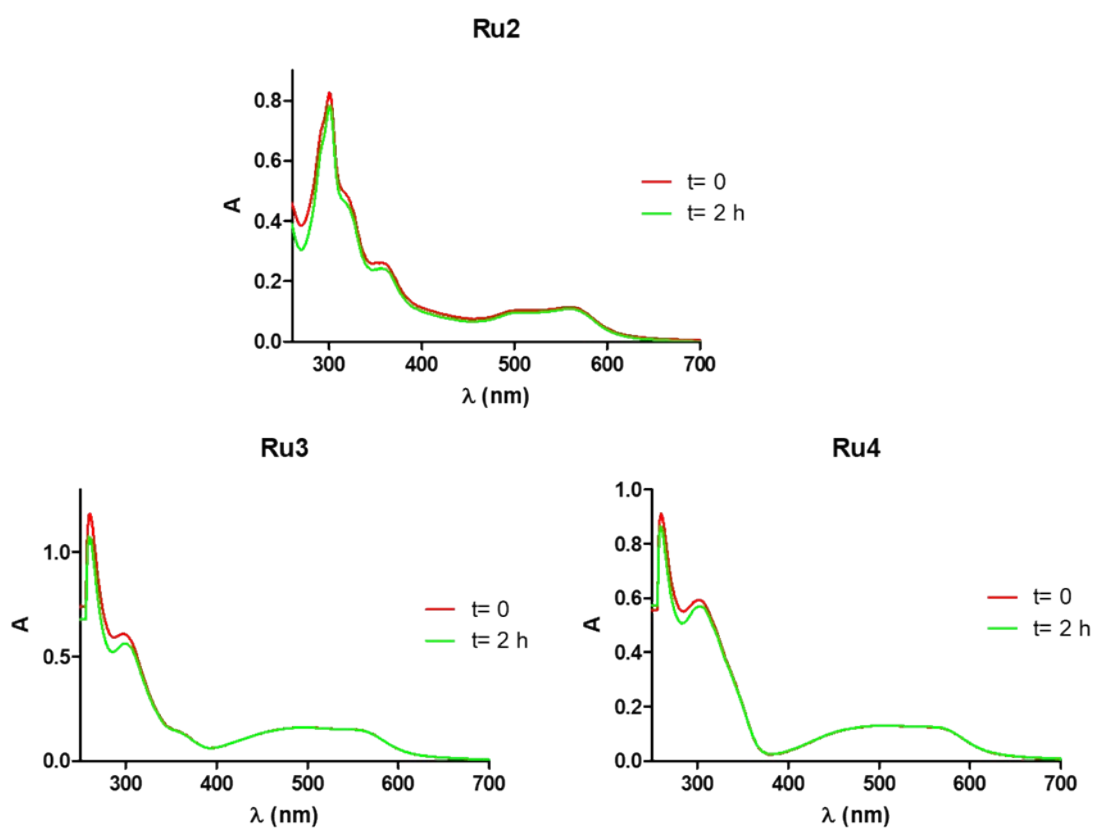
**Figure S22.** <sup>1</sup>H NMR spectra of compound **Ru2** in DMSO at t = 0 and after 48 h.



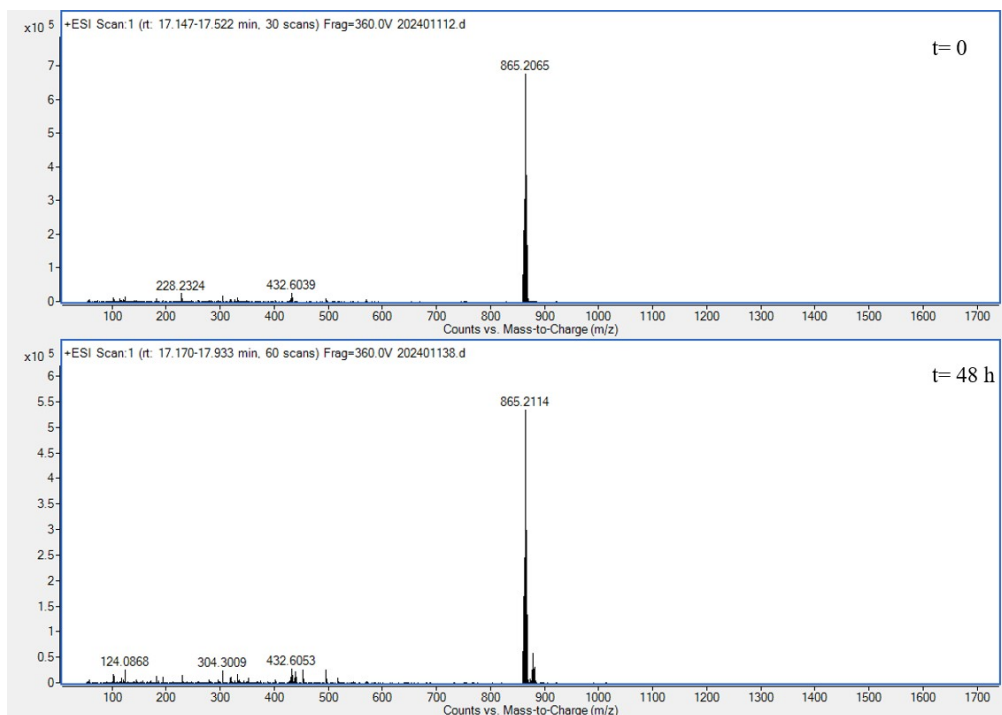
**Figure S23.** <sup>1</sup>H NMR spectra of compound **Ru3** in DMSO at t = 0 and after 48 h.



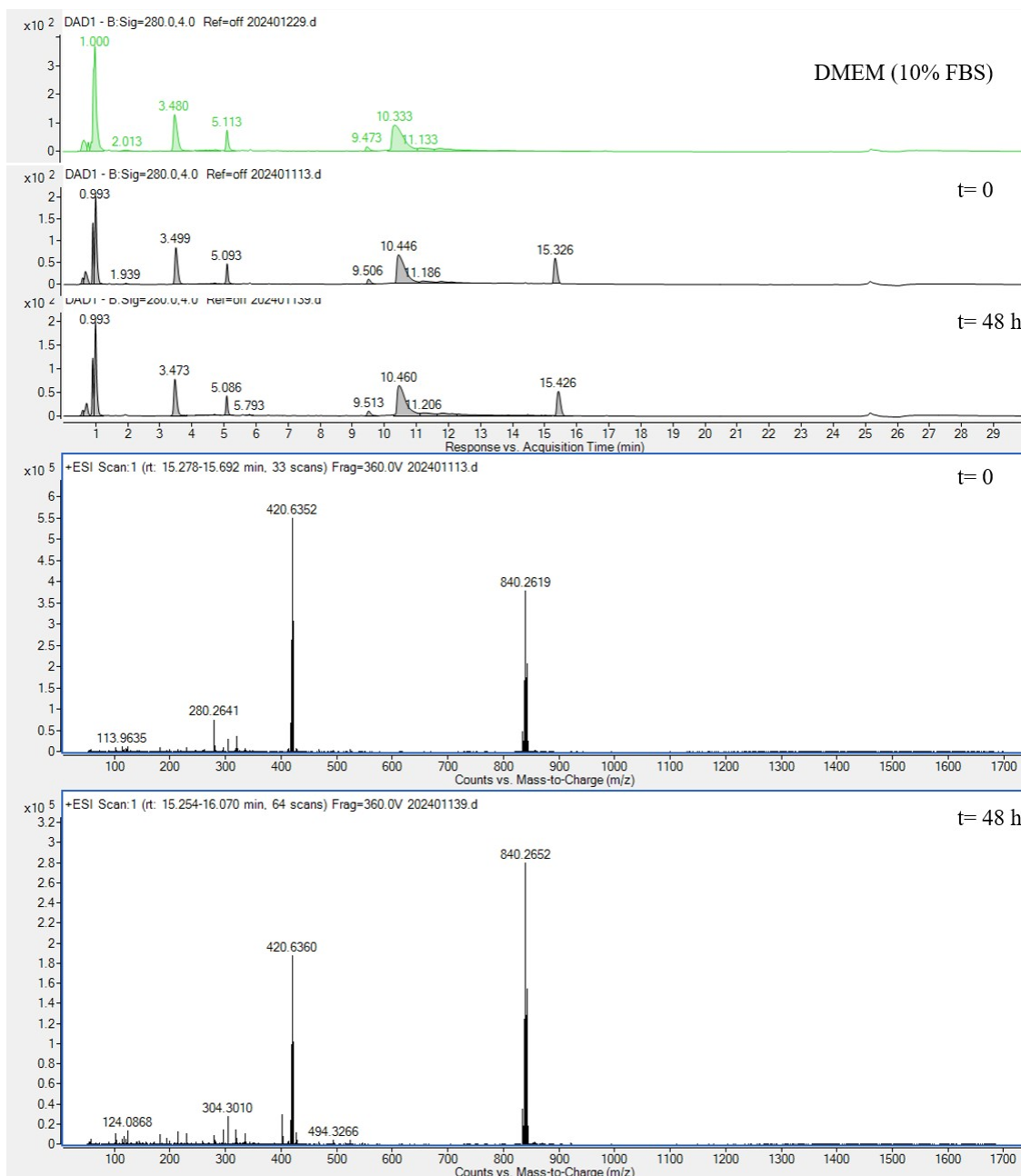
**Figure S24.**  $^1\text{H}$  NMR spectra of compound **Ru4** in DMSO at  $t = 0$  and after 48 h.



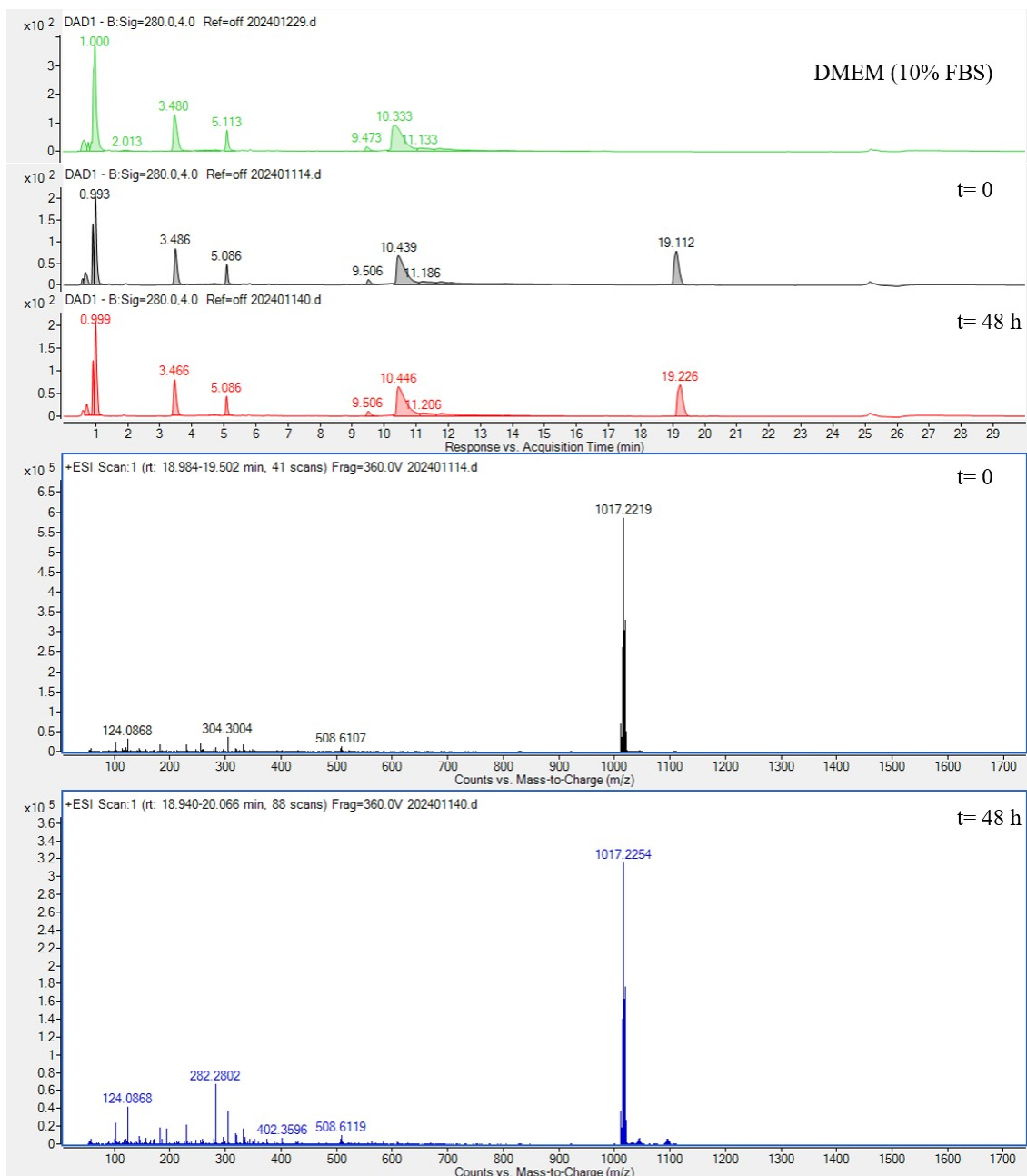
**Figure S25.** Time evolution of the absorption spectra of **Ru2** –**Ru4** ( $10\ \mu\text{M}$ ) in DMSO solution upon exposure to green light ( $1.8\ \text{mW}/\text{cm}^2$ ) during 2 h.



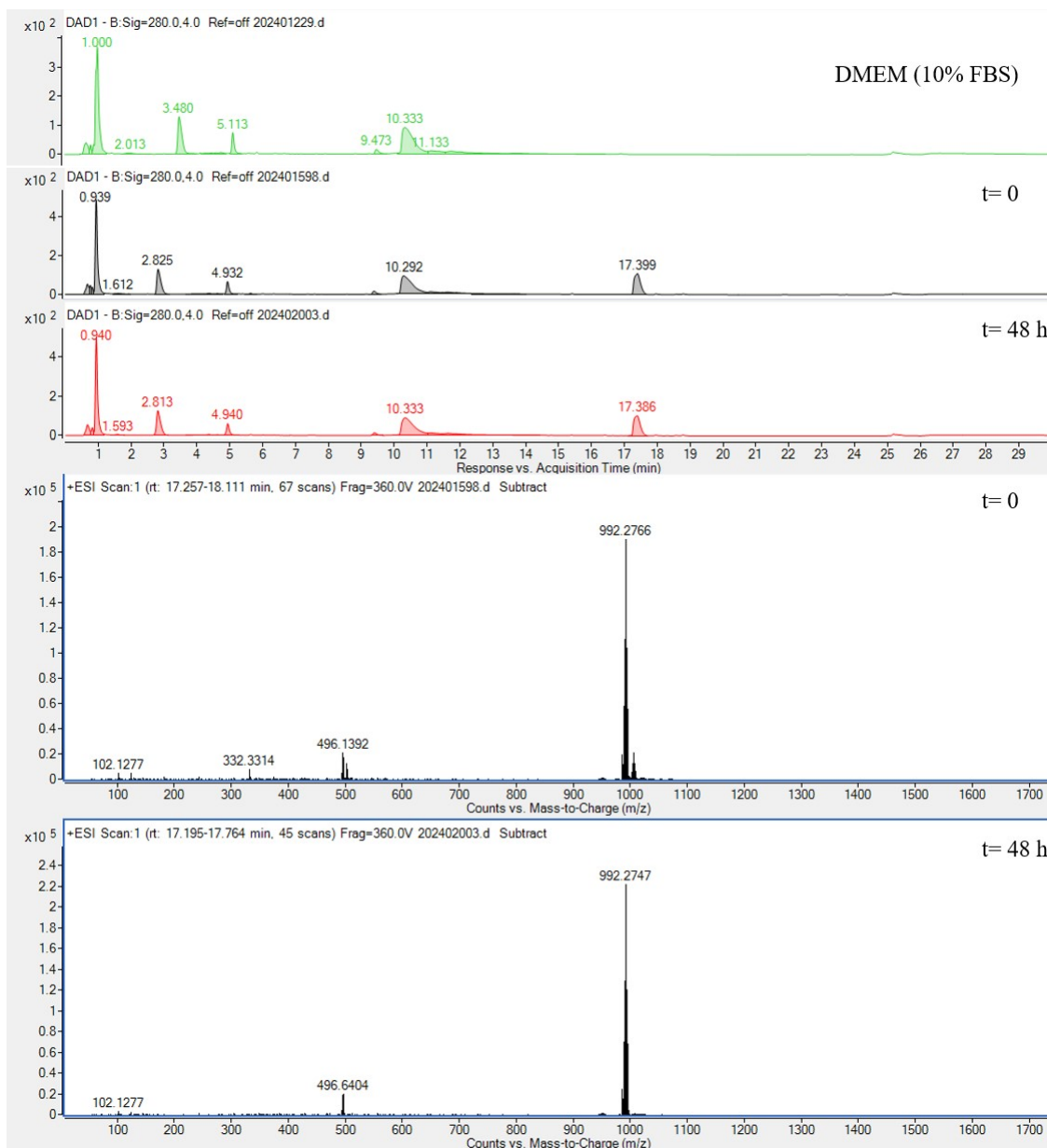
**Figure S26.** ESI-MS from HPLC of Figure 4C (**Ru1** (10  $\mu$ M)) in DMEM (5% DMSO) with 10 % FBS at t = 0 and after 48 hours and incubated at 37  $^{\circ}$ C.



**Figure S27.** HPLC-MS of Ru2 (10  $\mu$ M) in DMEM (5% DMSO) with 10 % FBS at t = 0 and after 48 hours and incubated at 37  $^{\circ}$ C.

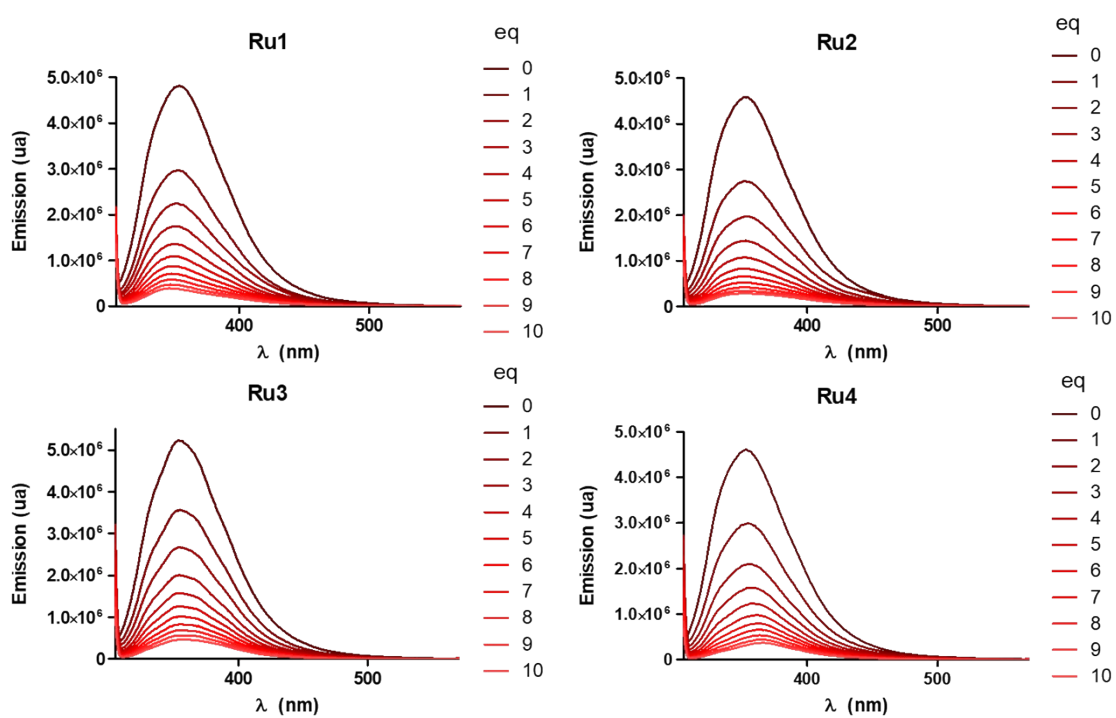


**Figure S28.** HPLC-MS of **Ru3** (10  $\mu$ M) in DMEM (5% DMSO) with 10 % FBS at t = 0 and after 48 hours and incubated at 37  $^{\circ}$ C



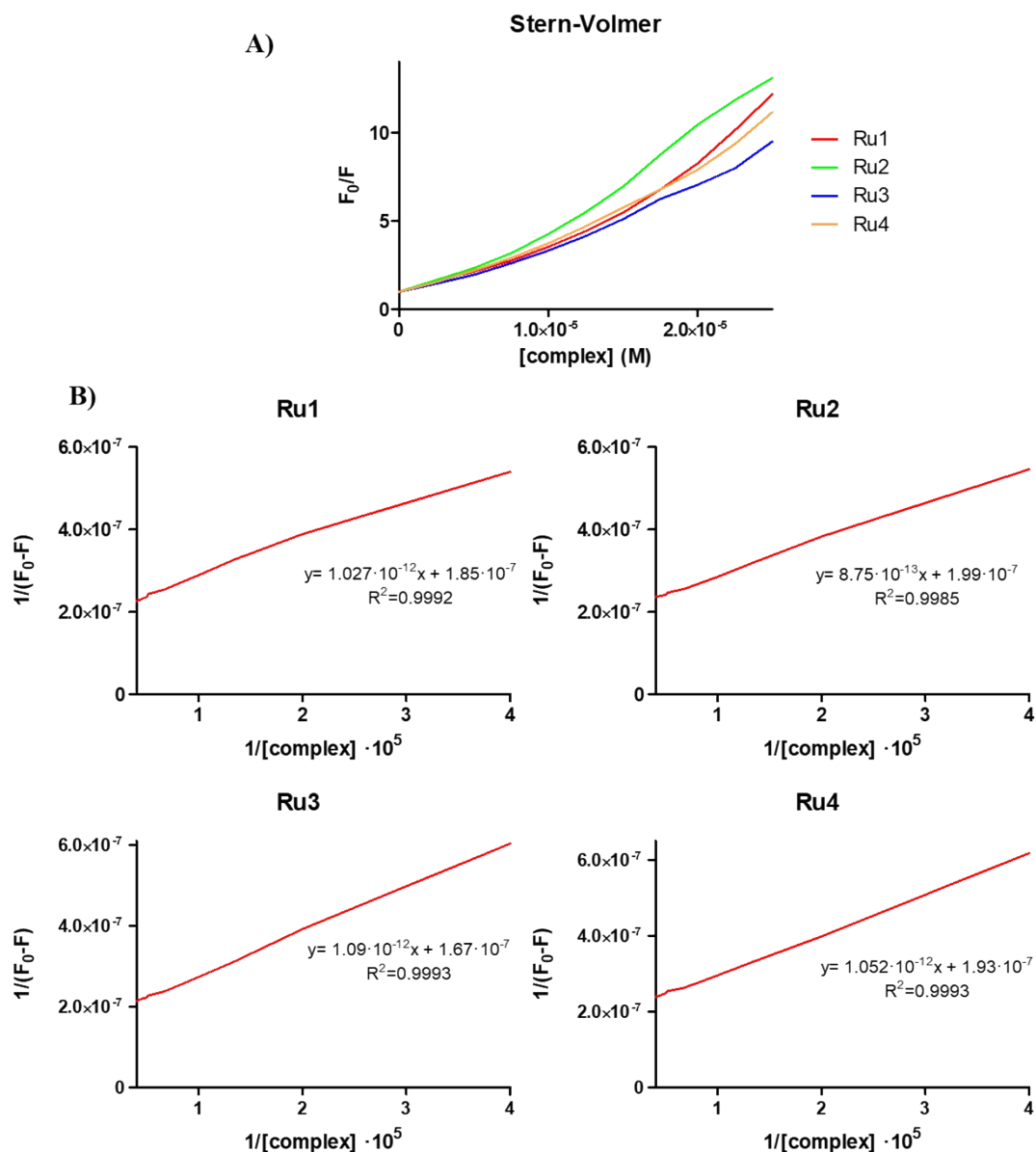
**Figure S29.** HPLC-MS of Ru4 (10 μM) in DMEM (5% DMSO) with 10 % FBS at t = 0 and after 48 hours and incubated at 37 °C.

### Reaction with Human Serum Albumin (HSA).

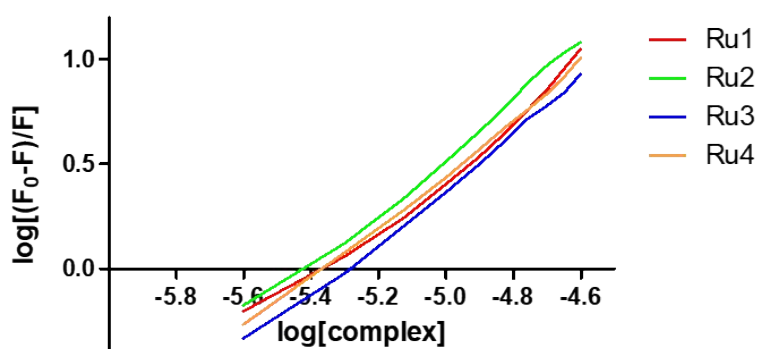


**Figure S30.** Emission spectra of HSA (2.5 μM) in the presence of increasing amounts of Ru1-Ru4 complexes (0-25 μM)



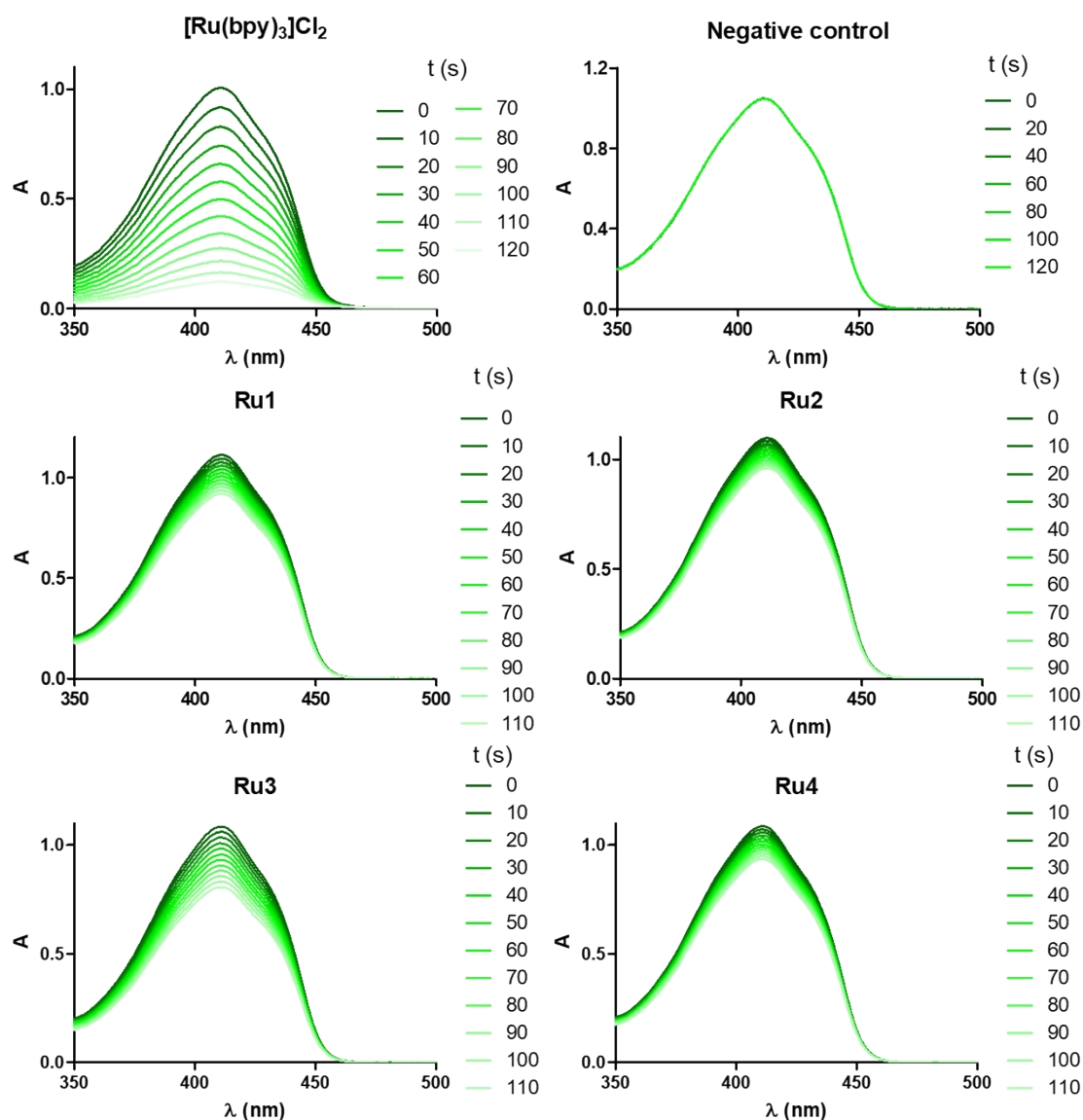


**Figure S31.** A) Stern-Volmer and B) Lineweaver-Burk curves of the quenching of HSA fluorescence in the presence of **Ru1-Ru4** complexes.



**Figure S32.** Scatchard curves for **Ru1-Ru4** complexes

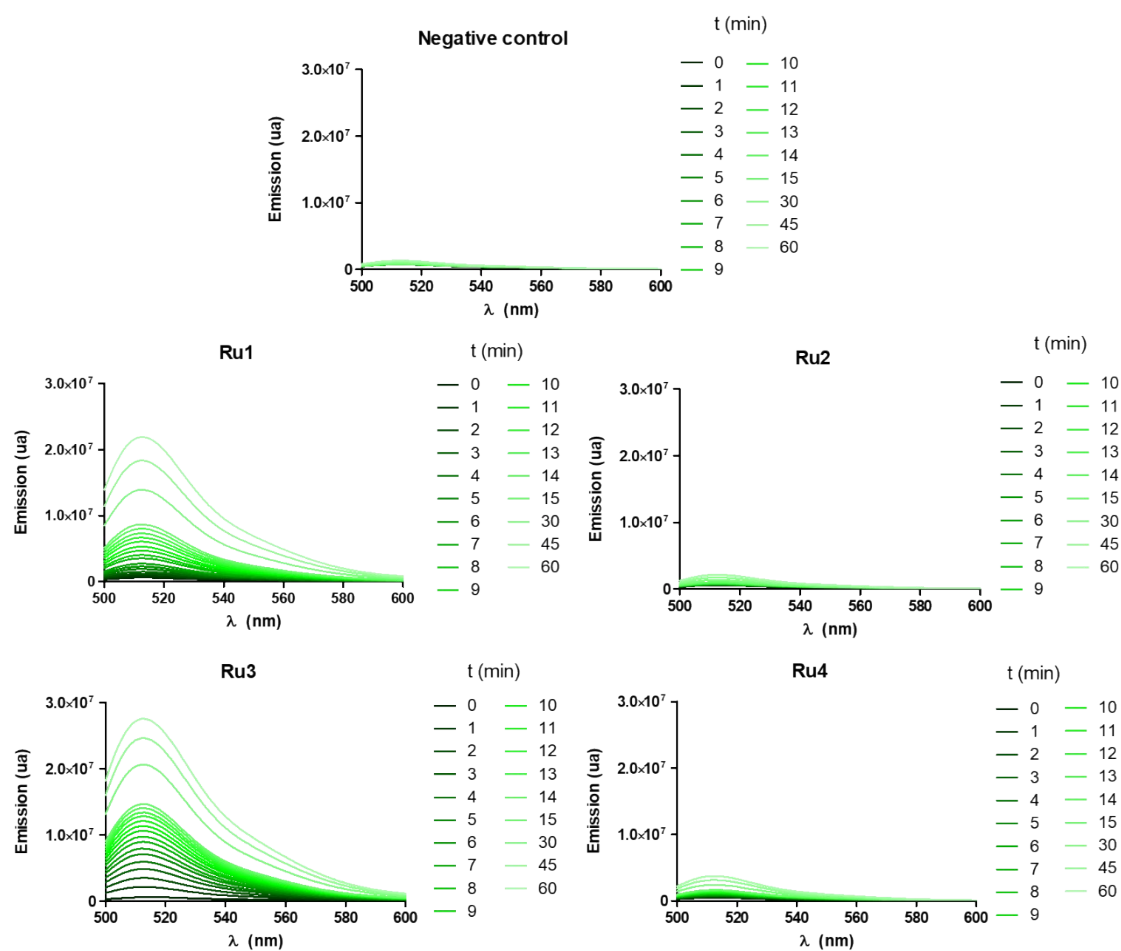
## 6. Evaluation for $^1\text{O}_2$ and/or $\cdot\text{OH}$ generation in cell free media.



**Figure S33.** Absorbance decrease of DPBF (50  $\mu\text{M}$ ) in presence of complexes **Ru1-Ru4** (4-7  $\mu\text{M}$ ) and the reference  $[\text{Ru}(\text{bpy})_3]\text{Cl}_2$  (50  $\mu\text{M}$ ) in aerated acetonitrile when irradiating with green light (520 nm, 0.5 mW/cm<sup>2</sup>). Acetonitrile was used as a negative control.

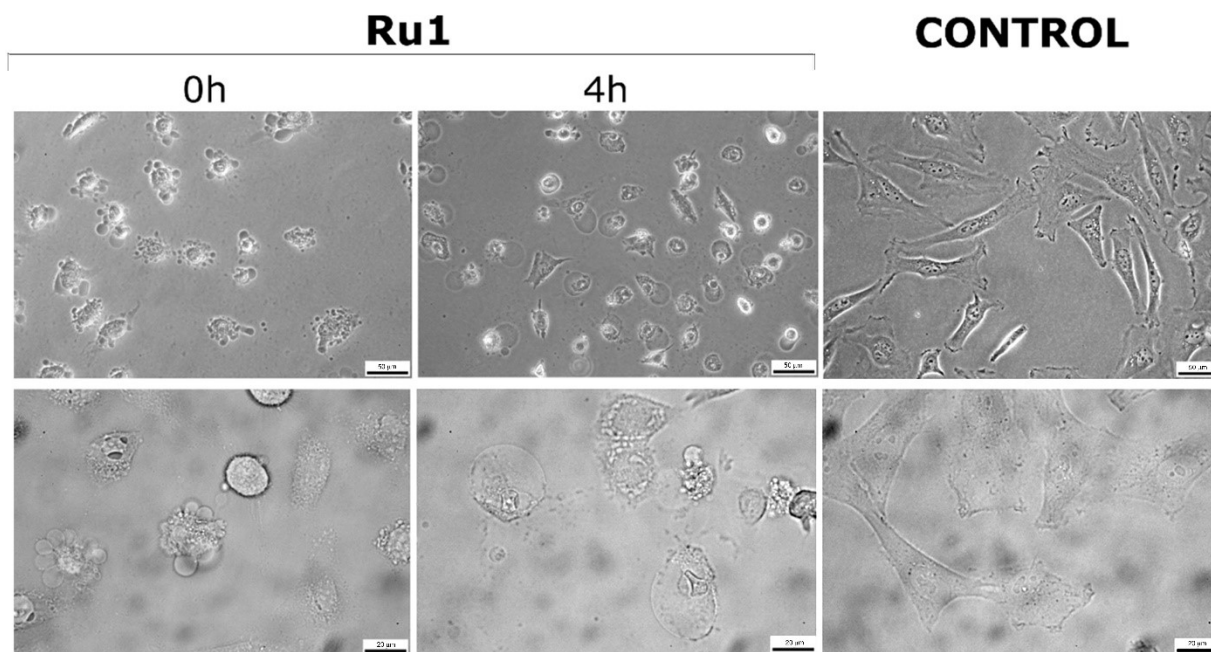
**Table S5.** Singlet oxygen quantum yields

Complex	$\phi$
<b>Ru1</b>	0.13
<b>Ru2</b>	0.078
<b>Ru3</b>	0.15
<b>Ru4</b>	0.093

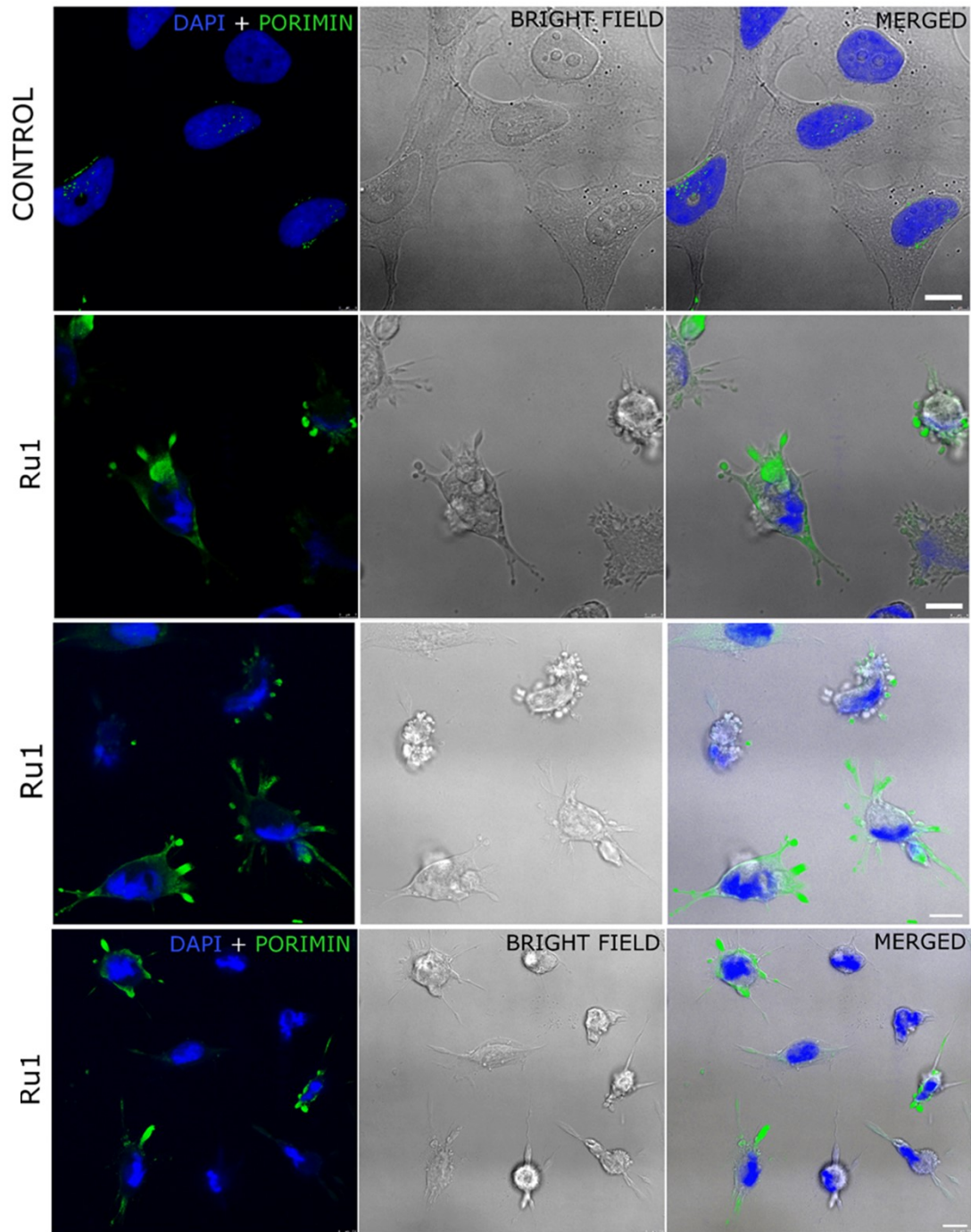


**Figure S34.** Increase of the fluorescence spectra emission of HPF upon photoirradiation of Ru(II) complexes Ru1-Ru4 at 520 nm (2 mW/cm<sup>2</sup>) in PBS (5 % DMF). HPF fluorescence was excited at 490 nm. PBS (5 % DMF) was used as a negative control.

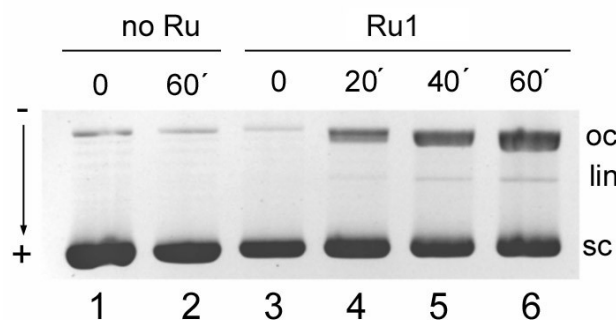
## 7. Biological assays



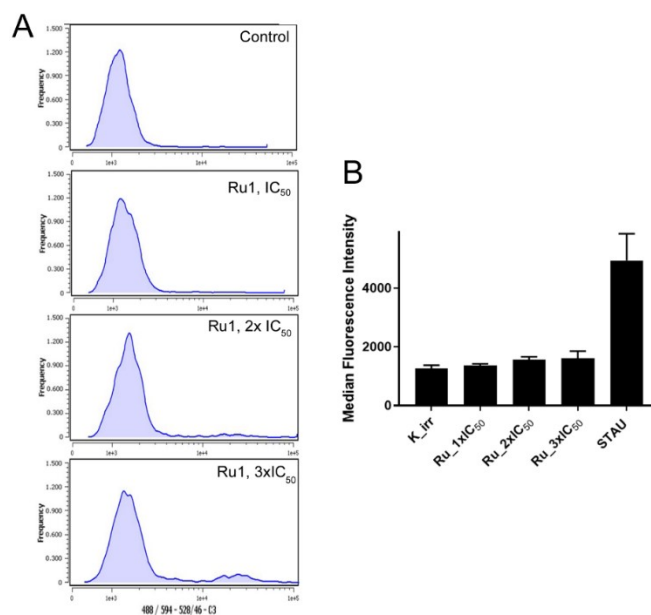
**Figure S35.** Microscopic images of HeLa cell morphology, vacuolization of cytoplasm, and cell swelling, as revealed by an inverted optical microscope. Cells were treated with **Ru1** (concentration corresponding to  $IC_{50}$ , Table 2) or untreated (control), irradiated and photographed immediately after irradiation (0h) or after 4 h recovery in cell free media. Scale bars represent 50  $\mu\text{m}$  (top panels) or 20  $\mu\text{m}$  (bottom panels).



**Figure S36.** Immunofluorescence staining of HeLa cells showing membrane localization of Porimin (green fluorescence signal); for better clarity, nuclei were stained with DAPI (blue signal). Top row: Control, untreated cells 4h after irradiation. Ru1: Cells treated with **Ru1** in a concentration corresponding  $IC_{50}$  (Table 2) 4 h after irradiation. Three representative images are shown. Scale bars represent 10  $\mu$ m.

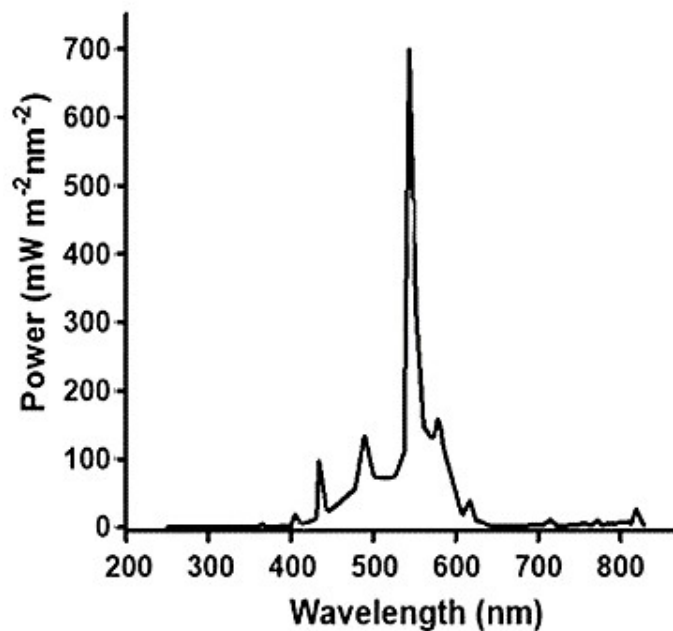


**Figure S37.** DNA photo-cleavage by **Ru1**. Plasmid pBR322 was mixed with the Ru complex **Ru1** (concentration ratio  $[Ru]/[DNA] = 1/10$ ). Samples were irradiated with a green light for the indicated time. Control, untreated DNA irradiated for 0 and 60 min was also included in the experiment (lanes 1 and 2). In both panels, the top DNA bands correspond to the nicked plasmid (OC), and the bottom bands correspond to the negatively supercoiled plasmid (SC). Linear DNA (lin) also appears in the samples treated with **Ru1** as a result of the formation of single-strand breaks in the close vicinity.

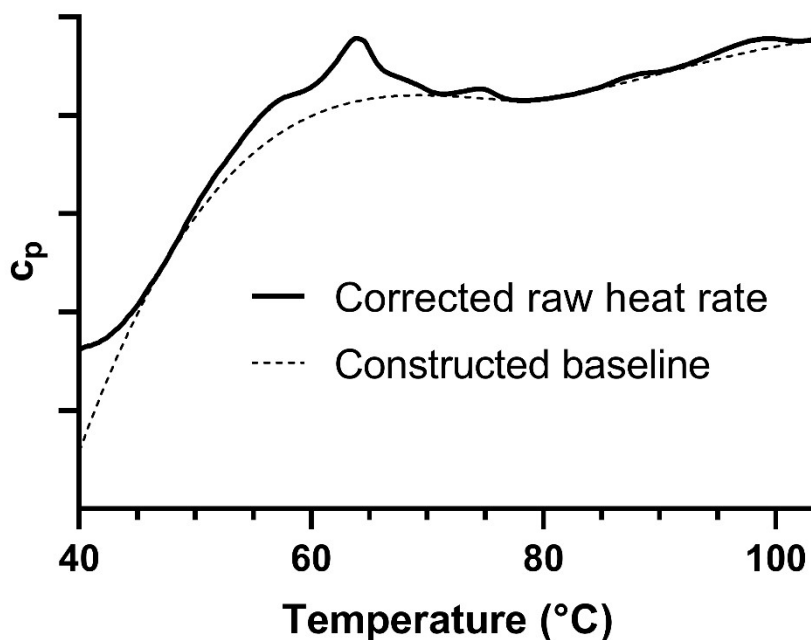


**Figure S38.** Activation of caspase 3/7 detected by CellEvent®Caspase3/7 Green Detection Reagent. HeLa cells were treated with **Ru1** (or incubated untreated) for 1 h in the dark and irradiated with green light. After 24 h of recovery in compound-free media, cell samples were stained with the CellEvent®Caspase 3/7 Green Detection Reagent, and fluorescence was analyzed by flow-cytometry; 30 000 cells were analyzed in each sample. A. Representative histograms. B. Quantitative evaluation. In the highest concentration (corresponding to 3x  $IC_{50}$ , a negligible amount of apoptotic cells are

detected; however, the apoptotic population is not significantly different from that in control, untreated irradiated cells. As a positive control, staurosporine (STAU, 1 $\mu$ M) was included in the experiment.



**Figure S39.** Spectral characteristic of green light used in the biological experiments.



**Figure S40.** The graph illustrates the construction of a baseline used to subtract background heat capacity in the evaluation of experiments conducted with differential scanning calorimetry.

## References

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