Supporting information

Potential-driven constructing interface and tensile strain of derived-Cu catalyst for boosting CO₂ electrocatalytic reduction

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Fig. S1. (a) TEM, (b) HR-TEM, and (c) EDS-elemental mapping images of Cu₂O NCs.



Fig. S3. Potential-resolved operando XRD patterns of the Cu.





Fig. S5. Inverse FFT interplanar spacing profiles of 23.1% Cu/Cu₂O, 46.4% Cu/Cu₂O and 70.8% Cu/Cu₂O.



Fig. S6. Faradaic efficiencies for C₂H₄, CO, CH₄ and H₂ on (a) 23.1% Cu/Cu₂O and (b) 70.8% Cu/Cu₂O at different applied potentials, respectively.



Fig. 57. ECSA measurement. (a-c) CVs with various scan rates between 0.49 - 0.59 V vs. RHE in CO₂-saturated 0.1 M KHCO₃ solution for determining Cdl for (a) 23.1% Cu/Cu₂O, (b) 46.4% Cu/Cu₂O and (c) 70.8% Cu/Cu₂O, respectively; double layer capacitance of (d) 23.1% Cu/Cu₂O, (e) 46.4% Cu/Cu₂O and (f) 70.8% Cu/Cu₂O, respectively.



Fig. S8. Electrocatalytic CO₂RR performances of 46.4 % Cu/Cu₂O in a flow-cell with 1 M KOH electrolyte. (a) LSV curves were measured in 1 M KOH solution saturated with N₂ or CO₂. (b) Current-dependent FEs of various reduction products







Fig. S10. The CO and C_2H_4 adsorption configurations on (a and e) Cu, (b and f) 23% Cu/Cu₂O, (c and g) 46% Cu/Cu₂O, and (d and h) 71% Cu/Cu₂O.

Table S1.	Diffraction peak shifts and the corresponding lattice parameters and strain values observed during the operando XRI
	on Cu ₂ O nanocrystals at different potentials. Cu foil was tested on the same XRD instrument as a reference.

			Cu(111)			Cu(200)	
Cu precursor	CO ₂ RR potential (VRHE)	Average XRD peak position (2θ)	Lattice parameter (Å)	Strain(%)	Average XRD peak Position (2θ)	Lattice parameter (Å)	Strain(%)
Cu foil		43.388	3.6123	0	50.519	3.6133	0
Cu/Cu₂O	-1.1	43.342	3.6128	0.01	50.475	3.6161	0.07
	-1.2	43.287	3.6172	0.22	50.318	3.6236	0.29
	-1.3	43.232	3.6216	0.34	50.263	3.6274	0.39

Table S2. The content of Cu ⁺ and Cu ⁰ on Cu/Cu ₂ O samples obtained by XP

	Copper species (wt %)			
Catalysts	Cu+	Cu ^o	Cu ⁺ /Cu ⁰	
23.1% Cu/Cu ₂ O	76.9	23.1	3.33	
46.4% Cu/Cu ₂ O	53.6	46.4	1.16	
70.8% Cu/Cu₂O	29.2	70.8	0.41	

Table S3. The content of O_L , O_V and O_W on the surface of the different Cu/Cu ₂ O samples obtained by XPS.					
Oxygen species (wt %)					
Catalysts	OL	Ov	Ow		
23.1% Cu/Cu ₂ O	33.4	21.5	45.1		
46.4% Cu/Cu ₂ O	29.9	54.6	15.4		
70.8% Cu/Cu₂O	10.5	74.4	15.1		

Table S4. Catalytic performances of Cu-based catalysts

Table 54. Catalytic performances of Cu-based catalysts							
Catalyst	Electrolyte	FE _(C2H4)	E vs. RHE	Ref.			
CuO _x	0.5 M KHCO ₃	54%	-1.4V	1			
Cu-Pd bimetallic	0.1 M KCl	45.2%	-1.2 V	2			
Ag ₁ -Cu _{1.1}	0.1 M KHCO ₃	40%	-1.1 V	3			
CuNi-2	0.5 M NaHCO ₃	24%	-0.97V	4			
Cu@Ag Core-Shell	1M KOH	32.2%	-1.1 V	5			
Cu@Cu NS-12	0.1M KHCO ₃	40.7%	-1.357V	6			
Cu _y /CeO ₂	0.1 M KHCO ₃	42%	-1.3V	7			
GB-Cu	1M KOH	38%	-1.2V	8			
Cu _v -Cu ₂ O	0.1 M KHCO ₃	51%	-0.76V	9			
Cu@nanosilica	0.1 M KHCO ₃	46%	-1.4V	10			
CuO microboxes	0.1 M K ₂ SO ₄	49.6%	-1.05V	11			
46.4% Cu/Cu₂O NCs	0.1 M KHCO ₃	52.5%	-1.1V	This work			