Supplementary Information

Surface plasmon resonance and structure defects synergetic effect of ZnCdS₂/NiMoO₄@Cu Z-scheme heterojunction for enhanced photocatalytic CO₂ reduction to CH₄

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Fig. S1. a-b The SEM images of ZCS and ZCS/NMO.



Fig. S2. a-b The SEM images of NiMoO₄ and Cu.



Fig. S3. The XRD patterns and magnified patterns of ZCS and ZCS/NMO-X



Fig. S4. The XRD patterns and magnified patterns of ZCS and ZCS/NMO@Cu-X



Fig. S5. High resolution XPS spectra of Cu 2p of ZCS/NMO@Cu



Fig. S6. Photothermal maps of ZCS, ZCS/NMO, and ZCS/NMO@Cu under near-infrared light irradiation at different times.



Fig. S7. a Pore size distribution of ZCS, ZCS/NMO and ZCS/NMO@Cu. **b** Nitrogen sorption isotherms of ZCS, ZCS/NMO and ZCS/NMO@Cu. **c** Pore size distribution of ZCS, ZCS/NMO-X and ZCS/NMO@Cu-X. **d** Nitrogen sorption isotherms of ZCS, ZCS/NMO-X and ZCS/NMO@Cu-X.



Fig. S8. CO₂ TPD profiles of ZCS/NMO and ZCS/NMO@Cu.



Fig. S9. a UV-vis absorption spectra of ZCS and ZCS/NMO-X. b UV-vis absorption spectra of ZCS/NMO@Cu-X



Fig S10. CO_2 reduction performance under near-infrared light irradiation.



Fig S11. Mass spectra of ${}^{13}CH_4$ generated from the isotopic ${}^{13}CO_2$ photoreduction.



Fig S12. Band gap spectra of $ZnCdS_2$ and $NiMoO_4$.



Fig S13. Fluorescence spectra of ZCS/NMO@Cu under visible irradiation.



Fig S14. Static experiment of ZCS and NMO.

Sample	Surface area	Mean pore diameter	Pore volume
	(m ² /g)	(nm)	(cm ³ /g)
ZCS	68.43	17.10	0.29
ZCS/NMO-1	68.6	18.82	0.32
ZCS/NMO-2	85.45	17.10	0.37
ZCS/NMO-3	95.05	15.71	0.37
ZCS/NMO-4	128.97	12.48	0.40
ZCS/NMO-5	97.7	25.15	0.61
ZCS/NMO@Cu-1	56.19	28.71	0.40
ZCS/NMO@Cu-2	49.11	34.56	0.42
ZCS/NMO@Cu-3	51.29	34.05	0.44
ZCS/NMO@Cu-4	38.46	32.79	0.32

 Table S1 Specific surface areas and pore volume of ZCS, ZCS/NMO-X and ZCS/NMO@Cu-X.

Catalyst	Light Source	CH ₄ yield	Selectivity	Ref
UiO-66/Co ₉ S ₈	300W Xe lamp,	$240.9 \text{ umol} \cdot \text{g}^{-1} \cdot \text{h}^{-1}$	~100%	1
	AM 1.5	240.9 µmor g m	10070	1
CuSnInS ₄	300W Xe lamp,	5.83 umol·g-1·h-1	67.3%	2
	>420 nm	5.65 µmor g n ,		L
Vs-CuIn ₅ S ₈	300W Xe lamp,		- 100%	3
	>420 nm	8.7 μ mol·g ⁻¹ ·h ⁻¹ ,	10070	5
Bi ₂ WO ₆ /ZnIn ₂ S ₄	300W Xe lamp,		94.5%	4
	>420 nm	5.12 μ mol·g ⁻¹ ·h ⁻¹ ,		4
C ₆₀ /CuS@ZnIn ₂ S ₄	300W Xe lamp	$13.6 \text{ umal} (a^{-1} \cdot b^{-1})$	~100%	5
	(>400 nm), H ₂ O	45.0 µmor g *n ,		-
ZnIn ₂ S ₄ /N-doped	300W Xe lamp,	1.01 umol· a^{-1} ·h ⁻¹	42.4%	6
graphene	>420 nm	1.01 µmor g m		
TiO₂-Cu₂ZnSnS₄	400 W Xe lamp,	1.01 umol·g ⁻¹ ·h ⁻¹	0.05%	7
110 ₂ -Cu ₂ Zii3ii3 ₄	≥420 nm	101 pinor 8 -		
Cu-ZnIn ₂ S ₄ /ZIF-67	300W Xe lamp,	22 27 umol·g-1·h-1	94.7%	8
	400-780 nm	22.27 µmor g m		Ū
NH ₂ -B-TiO ₂ -	300W Xe lamp,	$3.34 \text{ umol} \cdot \text{g}^{-1} \cdot \text{h}^{-1}$	16%	9
Cu _x S	>420 nm	5.54 µmor g n		,
BiOBr/CdIn ₂ S ₄	300W Xe lamp,	1.50 umplug-lub-l	16 7%	10
	>420 nm	1.50 µmor g m	10.770	10
SnS_2	150W halogen	0.13 µmol cm ⁻²	92%	11
	lamp, AM 1.5	5.15 pinor em	<u>, , , ,</u>	

Table S2 Comparison of photocatalytic CO_2 conversion to CH_4 performance of our photocatalysts with previously reported system of similar catalytic materials.

CuS@SnS ₂	300W Xe lamp, ≥400 nm	77.5 μmol·g ⁻¹ ·h ⁻¹	35.6%	12
MoO ₃ /MoS ₂ /CuS	300W Xe lamp, >420 nm	44.64 μmol·g ⁻¹ ·h ⁻¹	85.5%	13
MoS ₂ /SnS ₂ /r-GO	8W mercury lamp, H ₂ O	50.548 μmol·g ⁻¹ ·h ⁻¹	42.4%	14
In_2S_3/WS_2	400W Xe lamp, >420 nm	16 μmol·g ⁻¹ ·h ⁻¹	47%	15
Co-ZnS/MoS ₂ /graphene	350W Xe lamp, >420 nm	23.4 µmol∙g ⁻¹ ·h ⁻¹	~100%	16
MS/In ₂ S ₃	300W Xe lamp, >420 nm	68.41 μmol·g ⁻¹ ·h ⁻¹	80.3%	17
BiOI/MoS ₂ /CdS	300W Xe lamp, >420 nm	46.22 μmol·g ⁻¹ ·h ⁻¹	55.6%	18
ZnFe ₂ O ₄ /ZnO/CdS	300W Xe lamp, >420 nm	105.9∙µmol∙g ⁻¹ ∙h ⁻¹	28.5%	19
ZnCdS ₂ /NiMoO ₄ @Cu	300W Xe lamp, >420 nm	92.17 μmol·g ⁻¹ ·h ⁻¹	~100%	This work

Samples	τ ₁ (ns)	A ₁	$\tau_2(ns)$	A_2	τ(ns)
ZCS	1.87	49.48	6.33	50.52	4.12
ZCS/NMO	0.85	18.83	4.17	81.17	3.55
ZCS/NMO@Cu	1.91	43.91	6.20	56.09	4.32

Table S3 Time-resolved fluorescence spectra of ZCS, ZCS/NMO-X and ZCS/NMO@Cu-X

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