

Viologen Doping Induced Charge Storage of Carbon Nitride for Enhanced Photocatalytic Hydrogen Production

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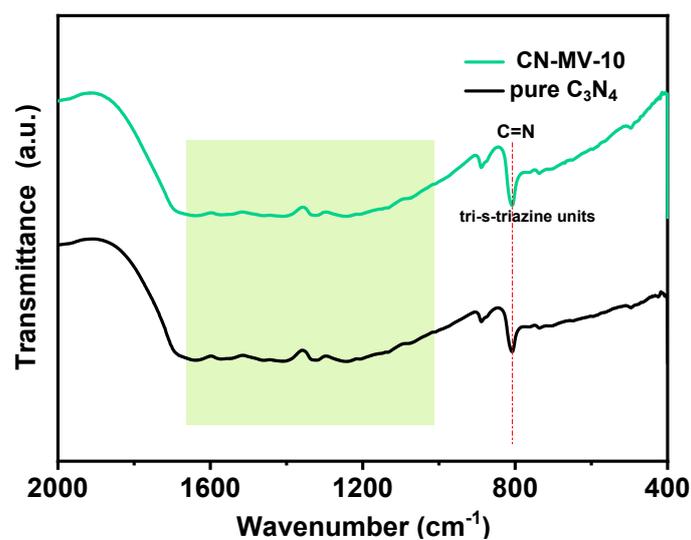


Fig. S1 FTIR spectra of g-C₃N₄ and the CN-MV-10 sample.

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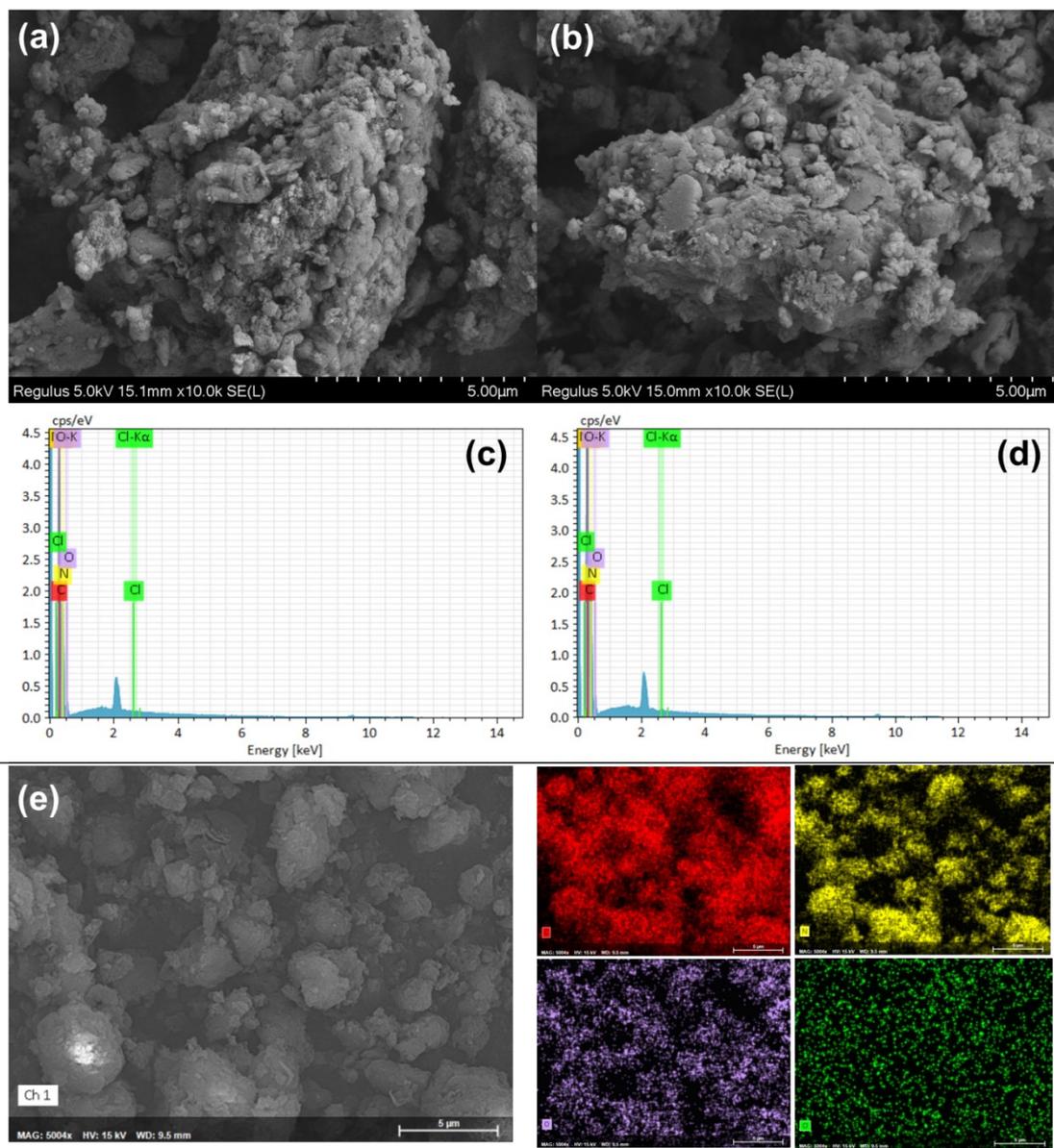


Fig. S2 SEM images and EDS spectrum of (a,c) g-C₃N₄ and (b,d) the CN-MV-10 sample. (e) Elemental mappings of the CN-MV-10 sample.

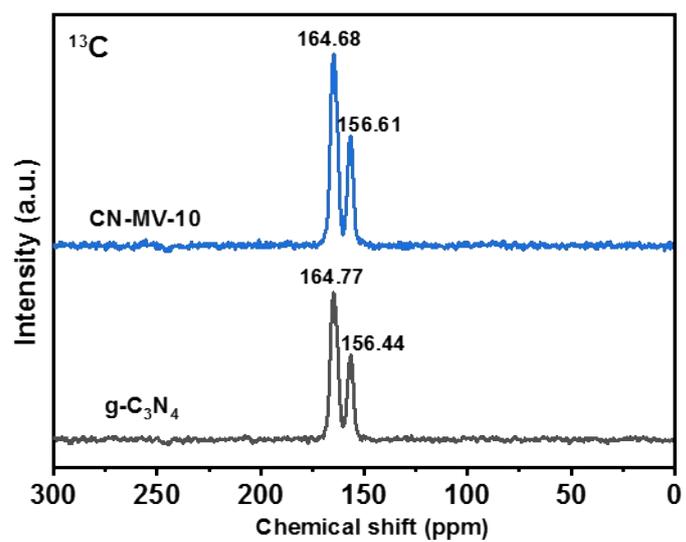


Fig. S3 Solid state ^{13}C cross-polarization magic angle spinning (CP-MAS) nuclear magnetic resonance (NMR) of g-C₃N₄ and CN-MV-10.

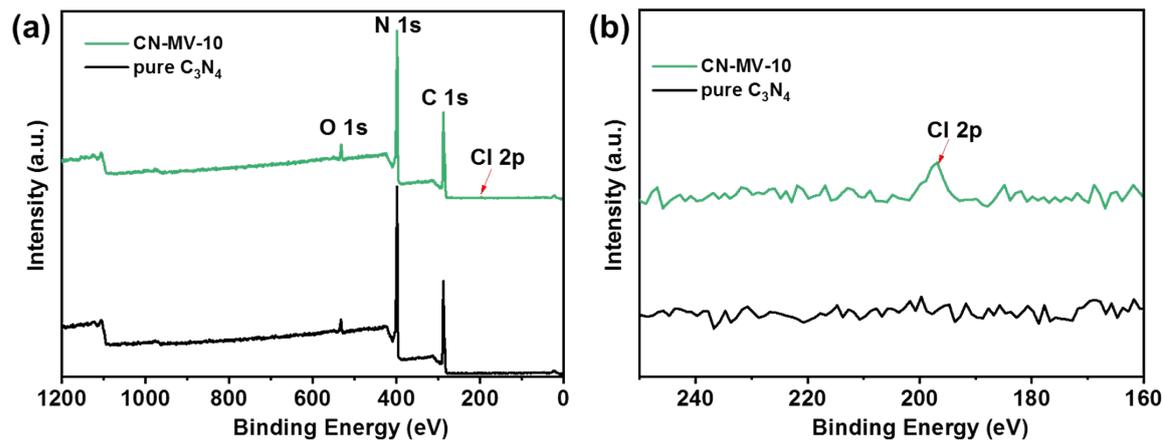


Fig. S4 (a) The XPS survey spectrum of pristine g-C₃N₄ and CN-MV-10. (b) Enlarged view of a portion of the XPS survey spectrum.

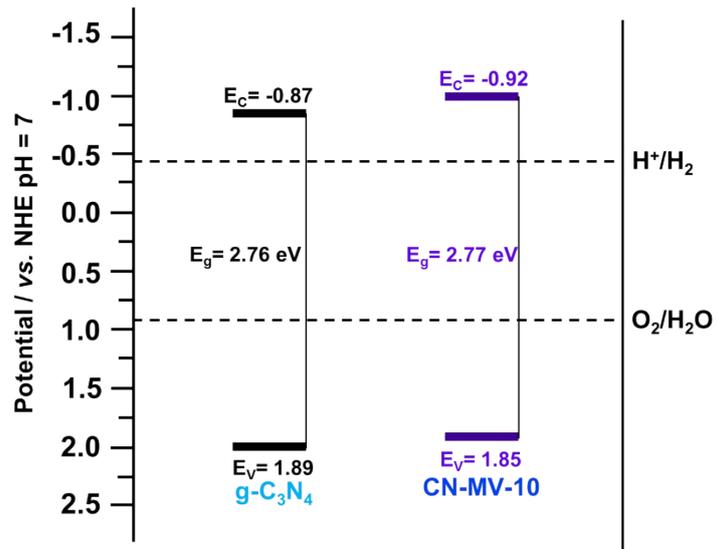


Fig. S5 Schematic illustration of the band structure of pure g-C₃N₄ and CN-MV-10.

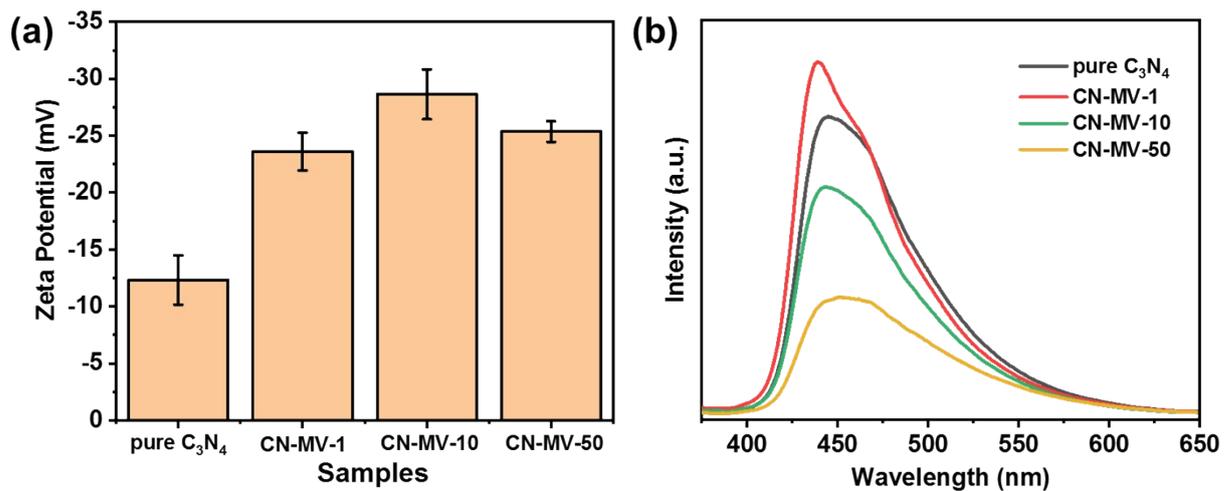


Fig. S6 (a) Zeta potentials and (b) PL spectra of g- C_3N_4 and CN-MV- x .

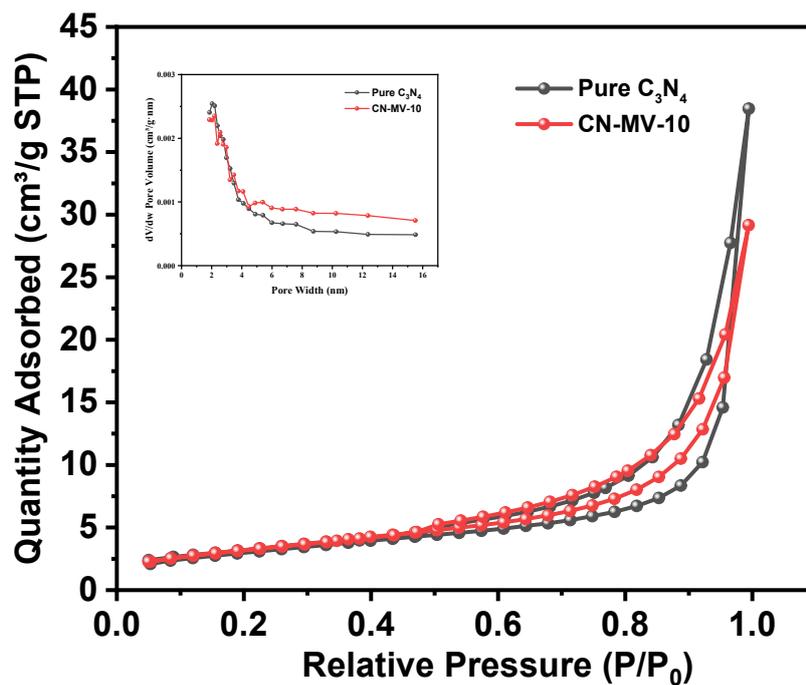


Fig. S7 N₂ adsorption/desorption curves and pore size distribution of pure C₃N₄ and CN-MV-10 samples.

Table S1 Comparison of specific surface area and pore size between Pure C₃N₄ and CN-MV-10

	BET Surface Area (m ² /g)	BJH Desorption average pore width (4V/A) (nm)
Pure C ₃ N ₄	11.0077	6.5605
CN-MV-10	11.7936	5.8332

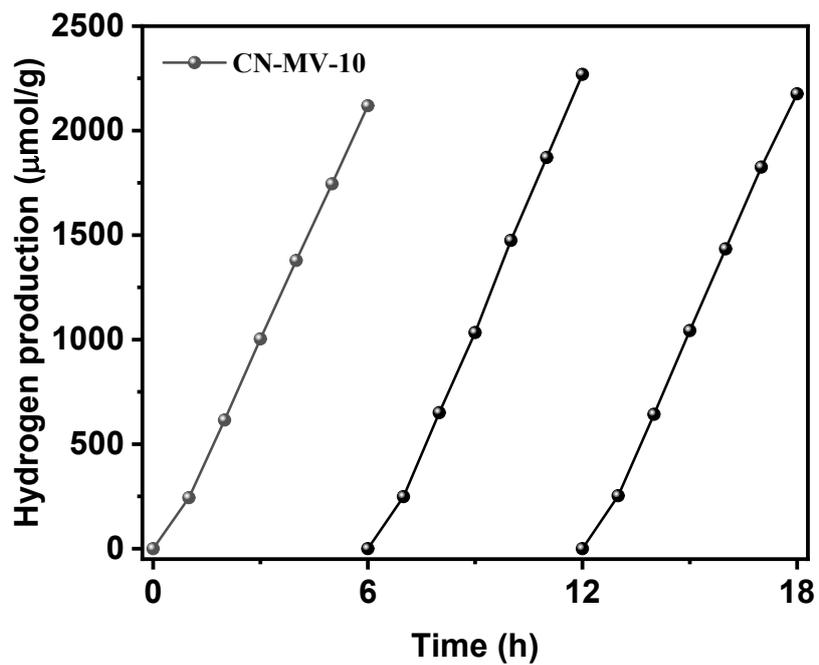


Fig. S8 Stability tests of photocatalytic hydrogen production for CN-MV-10

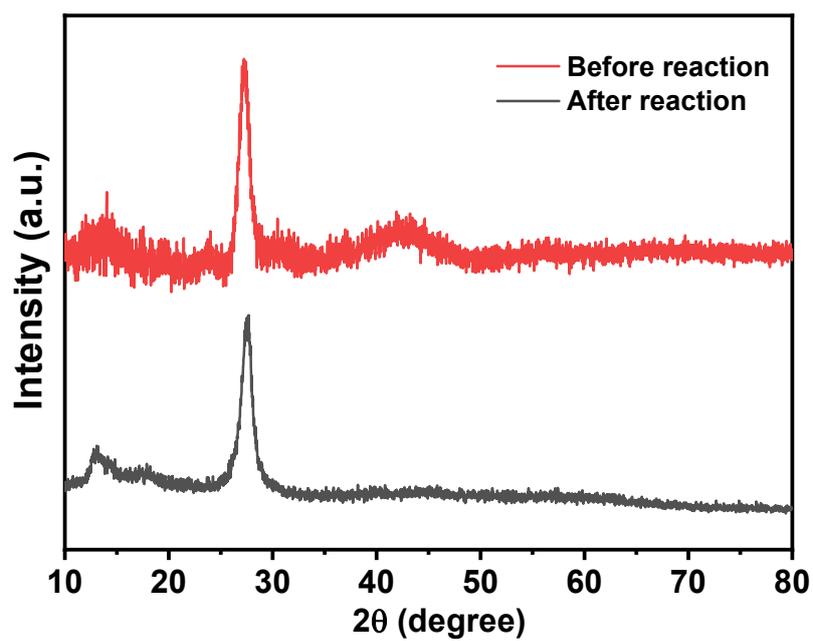


Fig.S9 Comparison of XRD before and after photocatalytic reaction

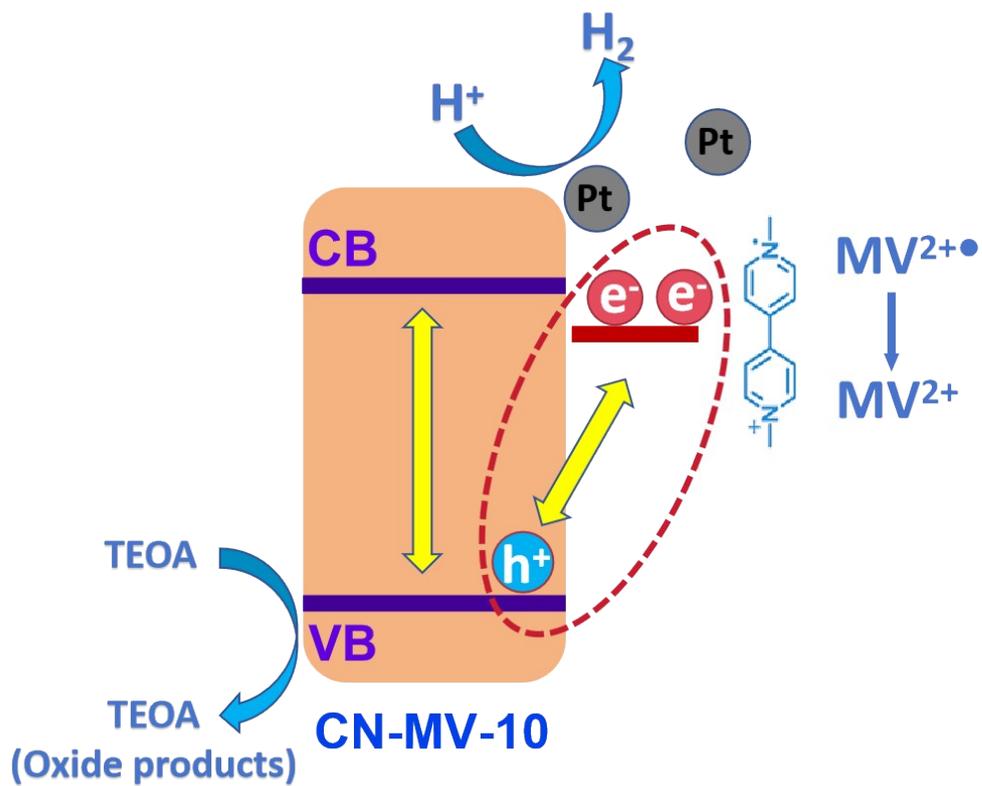


Fig.S10 proposed reaction mechanism of dark reaction hydrogen generation after illumination for the CN-MV-10 sample.

Table S2 Comparison of different research reports

Photocatalyst	Hydrogen production rate (μ mol/g/h)	Co-catalyst	Sacrificial agent	Reference
0.1HCCN	683.54	3wt%Pt	10vol%TEOA	1
Zn-Ni-P@g-C ₃ N ₄	531.2	\	15vol%TEOA	2
Ni _{0.4} Mo _{0.6} /g-C ₃ N ₄	1785	\	10vol%TEOA	3
CNK	919.5	3wt%Pt	10vol%TEOA	4
B-CN/P-CN(0.5:0.5)	655	3wt%Pt	10vol%TEOA	5
CN-M/CNU/Pt-TiO ₂	1735	1wt%Pt	10vol%TEOA	6
CNC-0.1	212.8	1wt%Pt	15vol%TEOA	7
CN-40	1210.3	3wt%Pt	10vol%TEOA	8
CN-MV-10	1650	1wt%Pt	5vol%TEOA	This work

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