

Synergistically enhancing stable interface with soft gel and garnet-type $\text{Li}_{6.4}\text{La}_3\text{Zr}_{1.4}\text{Ta}_{0.6}\text{O}_{12}$ bi-functional composite electrolyte of lithium metal batteries

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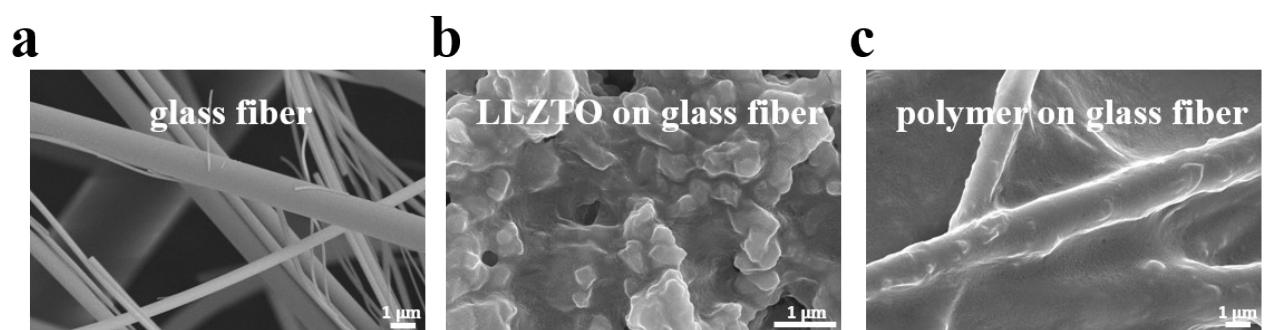


Fig. S1. SEM images of (a) glass fiber; (b) ceramic-rich layer; (c) gel-rich layer.

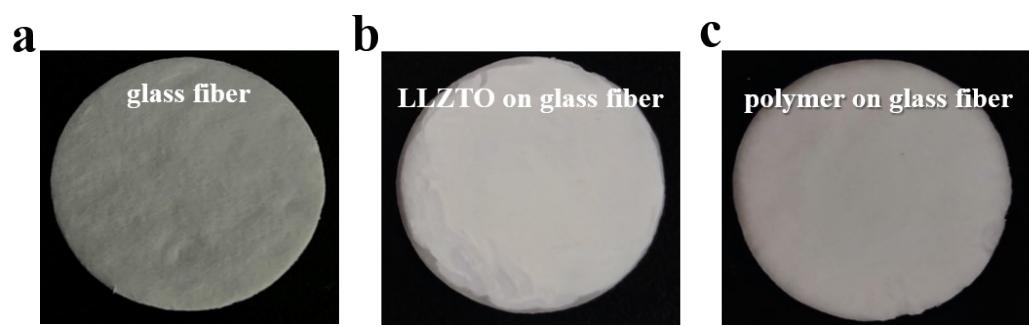


Fig. S2. Digital photographs of (a) glass fiber; (b) ceramic-rich layer; (c) gel-rich layer.

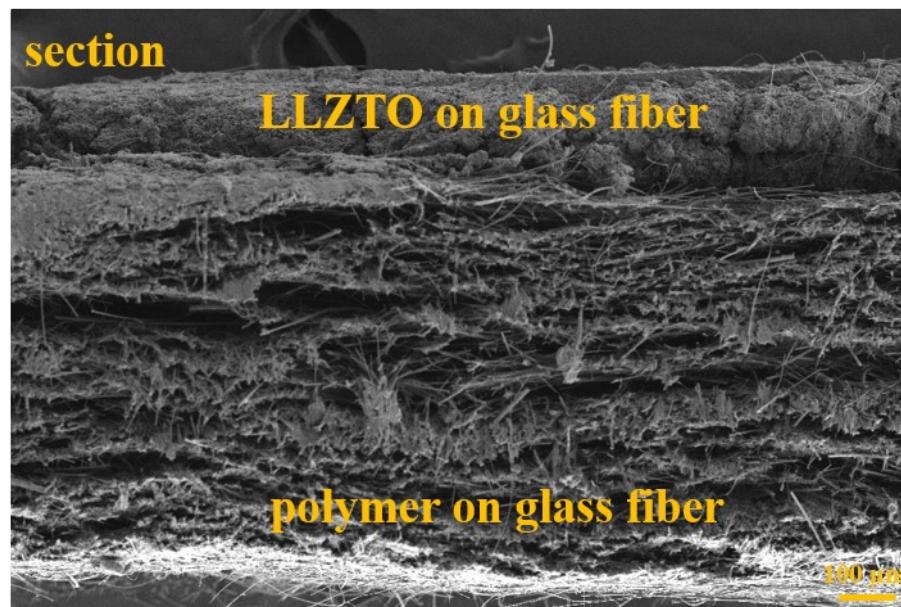


Fig. S3. Cross-section of As-THCE-10%LLZTO.

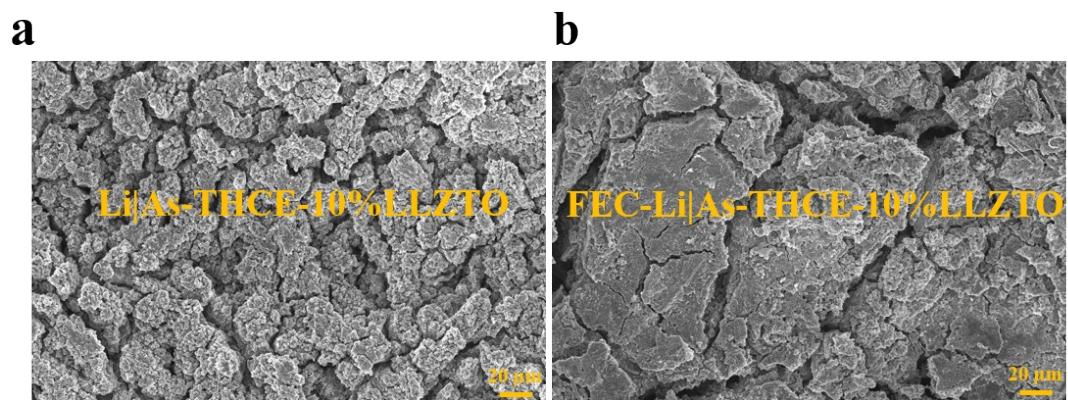


Fig. S4. SEM of images Li metal surface obtained from (a) Li||As-THCE-10%LLZTO||LEP; and (b) FEC-Li||As-THCE-10%LLZTO||LEP cells after 100 cycles.

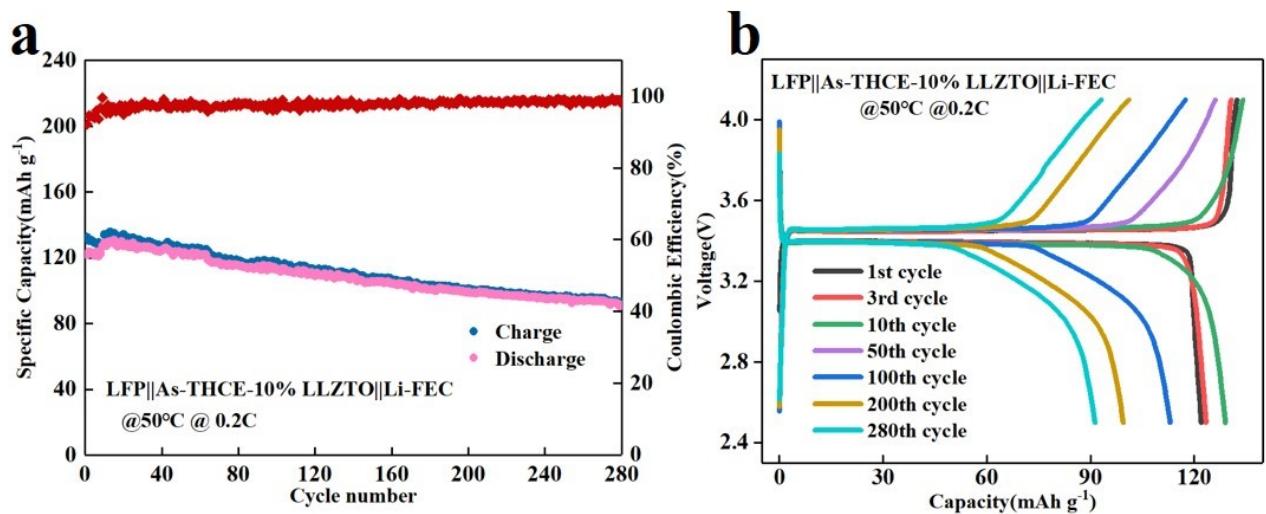


Fig. S5. (a) The cycle performance of LFP||As-THCE-10%LLZTO||Li-FEC cell at 0.2 C at 50 °C; (b) Charge-discharge curves at 0.2 C of LFP||As-THCE-10%LLZTO||Li-FEC cell at 50 °C

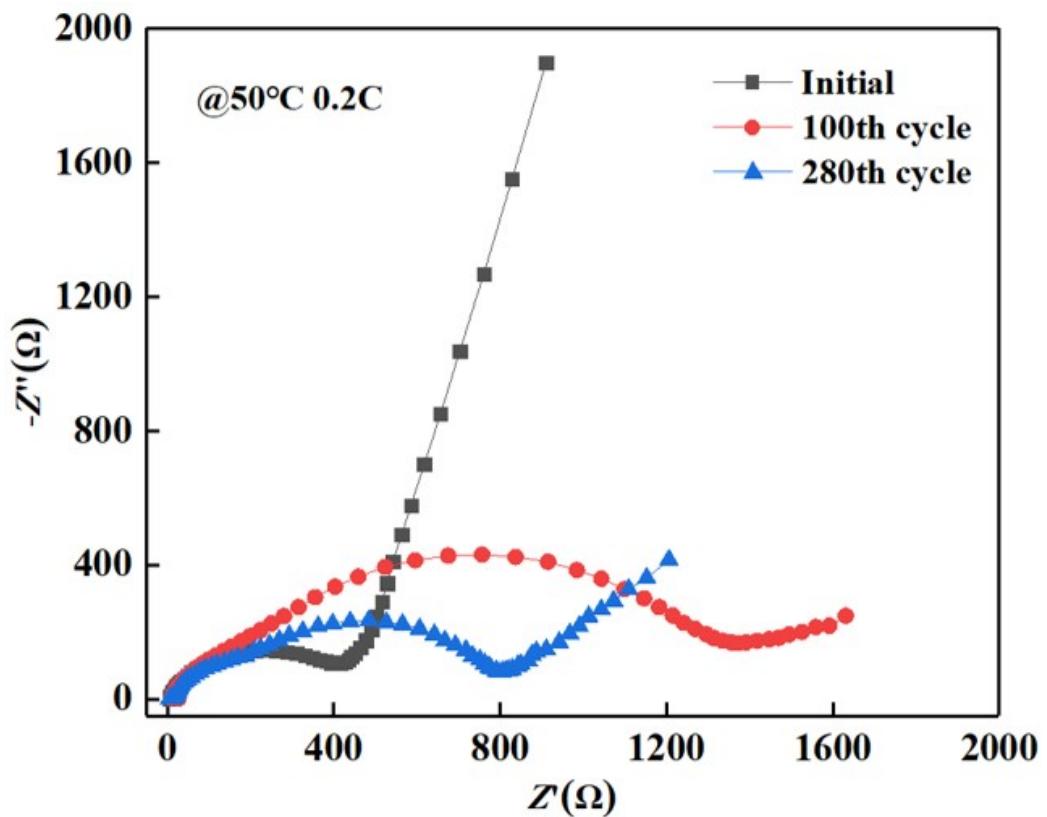


Fig. S6. Nyquist plots of LFP||As-THCE-10%LLZTO||Li-FEC before cycling, after 100 cycles, 280cycles.

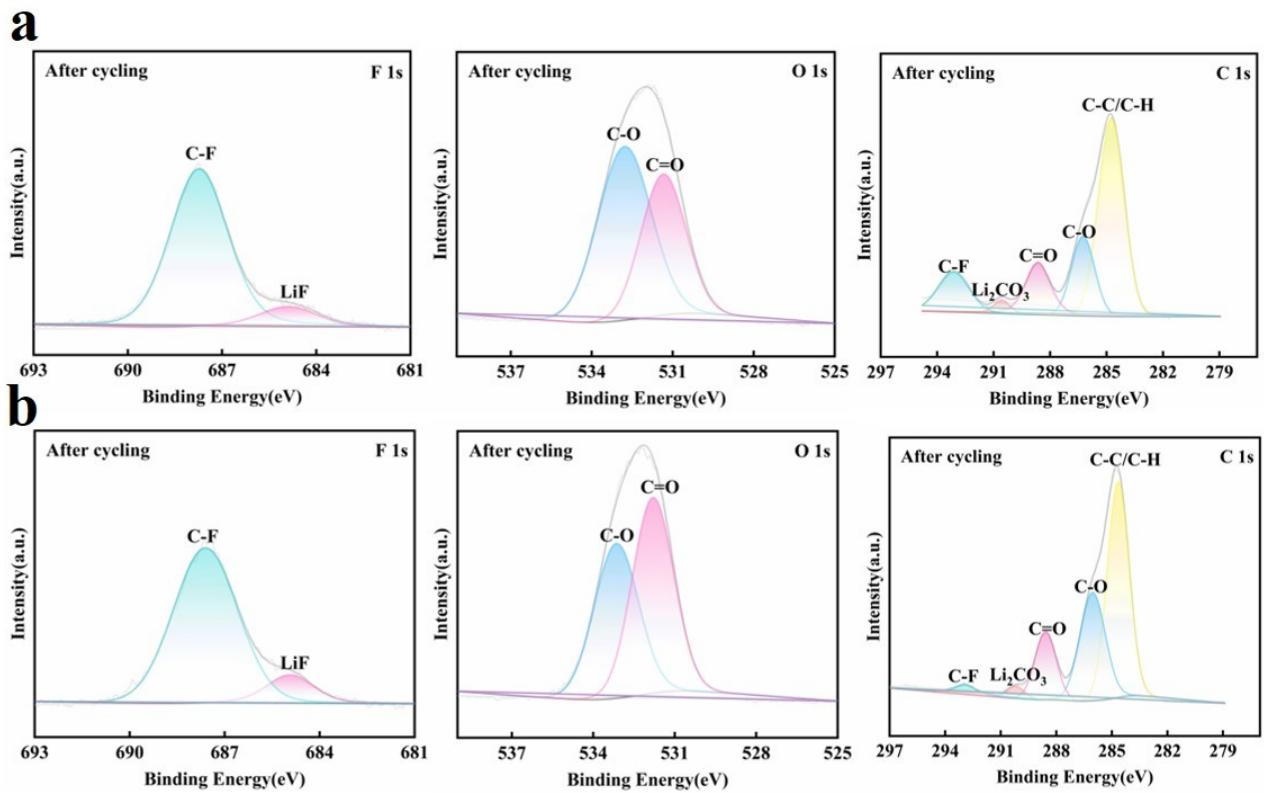


Fig. S7. XPS profiles of LFP obtained from (a) LFP||As-THCE-10%LLZTO||Li and (b) LFP||As-THCE-10%LLZTO||Li-FEC cells.

Tab. S1. Comparison of Ea of different samples

Sample	As-THCE- 5%LLZTO	As-THCE- 10%LLZTO	As-THCE- 15%LLZTO	As-THCE- 20%LLZTO
Ea (kJ·mol ⁻¹)	7.8	6.978	7.976	8.239
Ea (eV)	0.081	0.072	0.083	0.086

Tab. S2. Electrochemical performance comparison of the asymmetric electrolytes.

CPEs	Cell	Rate performance	Cycling stability	Ref.
PEO-LiTFSI-ACN-LLZTO	Li LEP	0.1 C, 99.1 mAh g ⁻¹	82.4%/200 th @25 °C	¹
TPGDA-LiPF ₆ -EC-DMC- EMC-LLZAO	Li NCM	149.1 mAh g ⁻¹	83.5%/100 th @25 °C	²
PEO-PAN-LiTFSI-LLZTO	Li LEP	0.2 C, 159.9 mAh g ⁻¹	96%/90 th	³
TPGDA- VEImNTF ₂ - LiPF ₆ -EC- DMC-EMC-LLZTO	Li LEP	0.5 C	300 th @25 °C	⁴
AHPAA-PEO-LiTFSI	Li LEP	0.5 C	84%/2000 th	⁵
TMPTMA-HDDA-N_{1,4,4,4}TFSI- LiTFSI-DMC-LLZTO	Li LEP	2 C, 111.6 mAh g⁻¹	84.73%/500th@25 °C	This work

References

- 1 H. Huo, Y. Chen, J. Luo, X. Yang, X. Guo and X. Sun, Rational design of hierarchical “Ceramic-in-Polymer” and “Polymer-in-Ceramic” electrolytes for dendrite-free solid-state batteries, *Adv. Energy Mater.*, 2019, **9**, 1804004.
- 2 N. Zhang, G. Wang, M. Feng and L. Z. Fan, In situ generation of a soft-tough asymmetric composite electrolyte for dendrite-free lithium metal batteries, *J. Mater. Chem. A*, 2021, **9**, 4018–4025.
- 3 Y. Li, L. Yang, R. Dong, T. Zhang, J. Yuan, Y. Liu, Y. Liu, Y. Sun, B. Zhong, Y. Chen, Z. Wu and X. Guo, A high strength asymmetric polymer–inorganic composite solid electrolyte for solid-state Li-ion batteries, *Electrochim. Acta*, 2022, **404**, 139701.
- 4 J. Li, Y. Cai, Y. Cui, H. Wu, H. Da, Y. Yang, H. Zhang and S. Zhang, Fabrication of asymmetric bilayer solid-state electrolyte with boosted ion transport enabled by charge-rich space charge layer for -20~70°C lithium metal battery, *Nano Energy*, 2022, **95**, 107027.
- 5 W. Liu, G. Li, W. Yu, L. Gao, D. Shi, J. Ju, N. Deng and W. Kang, Asymmetric organic-inorganic bi-functional composite solid-state electrolyte for long stable cycling of high-voltage lithium battery, *Energy Storage Mater.*, 2023, **63**, 103005.