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## **Supporting information**

Constructing Vertically Aligned ZIF-67 Microrod Arrays on Carbon

Cloth with Commercial-level Mass-loading for High-performance

Supercapacitors

Chenchen Bai, a Bingjun Li, a Dejun Li, a Yan Han a and Caiyun Wang b

<sup>a</sup> College of Physics and Materials Science, Tianjin Normal University, Tianjin 300387, China

<sup>b</sup> Faculty of Engineering and Information Sciences, University of Wollongong, New South Wales 2500, Australia

**Fig. S1** (a, b) SEM images and (c, d) capacitive performances of Co/CC immersed in a 2-MeIM aqueous solution for 24 h followed by adding Co(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O aqueous solution for another 24 h (a, c) and 72 h (b, d).

**Fig. S2** (a-d) SEM images of bare CC (a, b) and Co/CC (c, d). (e) EDS mappings of Co/CC.

Fig. S3 XRD pattern of Co/CC.

Fig. S4 (a) SEM image and (b) EDS mappings of ZIF-67/CC-24.

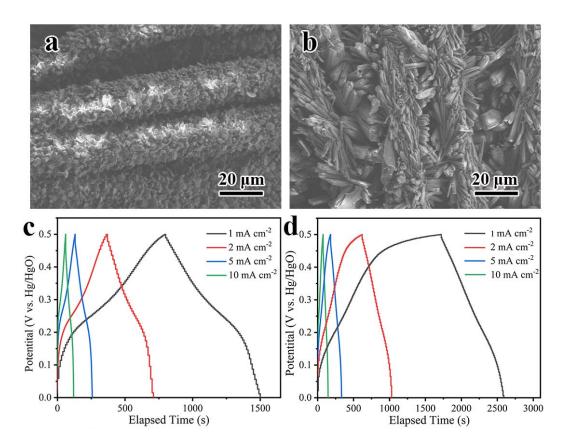
Fig. S5 TEM image of ZIF-67/CC.

**Fig. S6** Digital photographs of ZIF- 67 grown on bare CC and Co-covered CC: (a) bare CC immersed in an aqueous solution containing 2-MeIM and Co(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O for 72 h; (b) Co/CC immersed in a 2-MeIM aqueous solution for 72 h; (c) Co/CC immersed in a 2-MeIM aqueous solution for 24 h followed by adding Co(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O aqueous solution for another 48 h.

Fig. S7 (a) SEM image and (b) EDS mappings of ZIF-67/CC-NC.

Fig. S8 Full scan survey XPS spectrum for ZIF-67/CC-24.

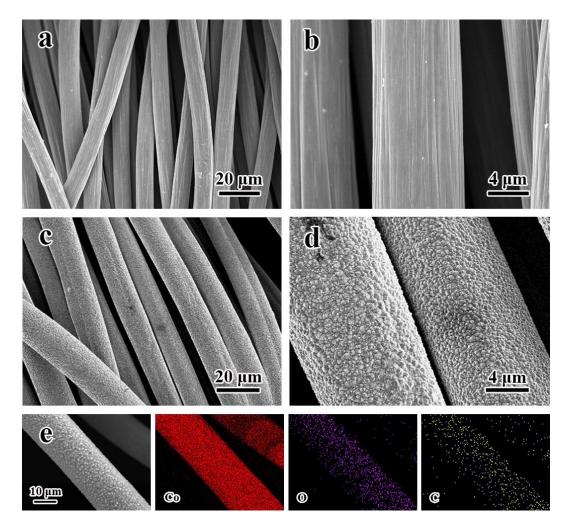
- Fig. S9 High-resolution XPS C1s spectrum of ZIF-67/CC.
- Fig. S10 TGA curves of ZIF-67/CC, CC and ZIF-67 powders.
- **Fig. S11** (a) XPS spectra of Co 2p and (b) XRD pattern of ZIF-67/CC electrode after cycling.
- **Fig. S12** (a) CV curves of AC/CC electrode at different scan rates. (b) Galvanostatic charge-discharge curves of AC/CC electrode at different current densities.
- **Fig. S13** CV curves of ZIF-67/CC//AC/CC ASSC device collected within different potential windows at 10 mV s<sup>-1</sup>.
- Fig. S14 Cycling performance of ZIF-67/CC//AC/CC ASSC device at 10 mA cm<sup>-2</sup>.
- Fig. S15 Nyquist plot of the ASSC device.
- **Table S1** Capacitive performance comparison of reported MOF materials.
- **Table S2** Comparison of areal energy and power densities for some reported MOF-based devices.



**Fig. S1** (a, b) SEM images and (c, d) capacitive performances of Co/CC immersed in a 2-MeIM aqueous solution for 24 h followed by adding Co(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O aqueous solution for another 24 h (a, c) and 72 h (b, d).

The immersion time of Co/CC was optimized to regulate the ZIF-67 mass-loading on CC in preliminary experiments. It was observed from SEM images that the morphology of samples obtained at different immersion time was different. A shorter immersion time (ZIF-67/CC-24, Fig. S1a) was not helpful to preferred orientation growth of ZIF-67, whereas a longer time (ZIF-67/CC-72, Fig. S1b) caused the bundled formation of ZIF-67 rods. The ZIF-67 mass-loading of ZIF-67/CC-24 and ZIF-67/CC-72 was about 6 and 20 mg cm<sup>-2</sup>, respectively. The areal specific capacitances of ZIF-67/CC-24 were about 1400, 1360, 1270 and 1180 mF cm<sup>-2</sup> at 1, 2, 5 and 10 mA cm<sup>-2</sup>. The corresponding gravimetric specific capacitances were 233, 227, 212 and 197 F g<sup>-1</sup>, respectively. For ZIF-67/CC-72 with higher mass-loading, the areal specific capacitances were 1740, 1640, 1530 and 1380 mF cm<sup>-2</sup> at 1, 2, 5 and 10 mA cm<sup>-2</sup>.

While the gravimetric specific capacitances dropped to 87, 82, 77 and 69 F g<sup>-1</sup>, respectively. Therefore, the change of capacitive performances was not only due to the variation of mass-loading but also the different morphology.



**Fig. S2** (a-d) SEM images of bare CC (a, b) and Co/CC (c, d). (e) EDS mappings of Co/CC.

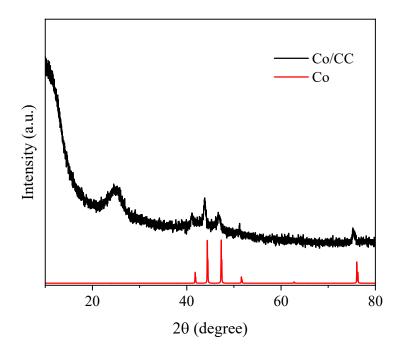


Fig. S3 XRD pattern of Co/CC.

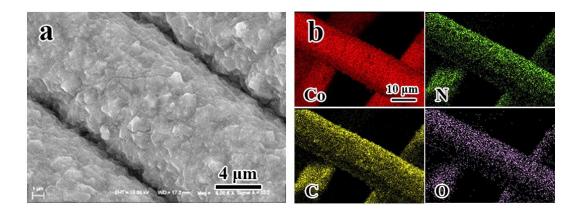


Fig. S4 (a) SEM image and (b) EDS mappings of ZIF-67/CC-24.

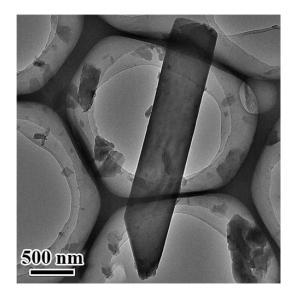
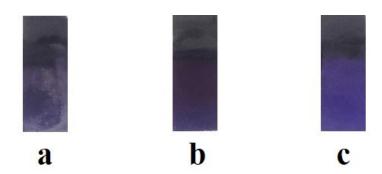


Fig. S5 TEM image of ZIF-67/CC.



**Fig. S6** Digital photographs of ZIF- 67 grown on bare CC and Co-covered CC: (a) bare CC immersed in an aqueous solution containing 2-MeIM and Co(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O for 72 h; (b) Co/CC immersed in a 2-MeIM aqueous solution for 72 h; (c) Co/CC immersed in a 2-MeIM aqueous solution for 24 h followed by adding Co(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O aqueous solution for another 48 h.

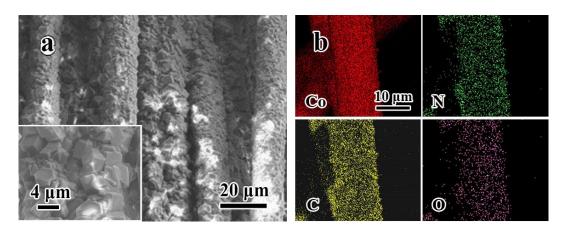


Fig. S7 (a) SEM image and (b) EDS mappings of ZIF-67/CC-NC.

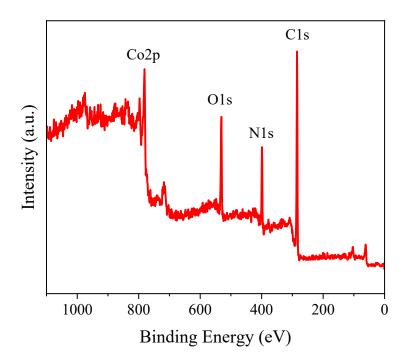


Fig. \$8 Full scan survey XPS spectrum for ZIF-67/CC-24.

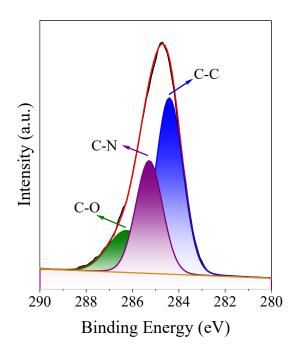


Fig. S9 High-resolution XPS C1s spectrum of ZIF-67/CC.

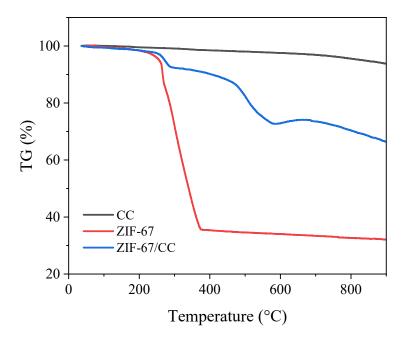
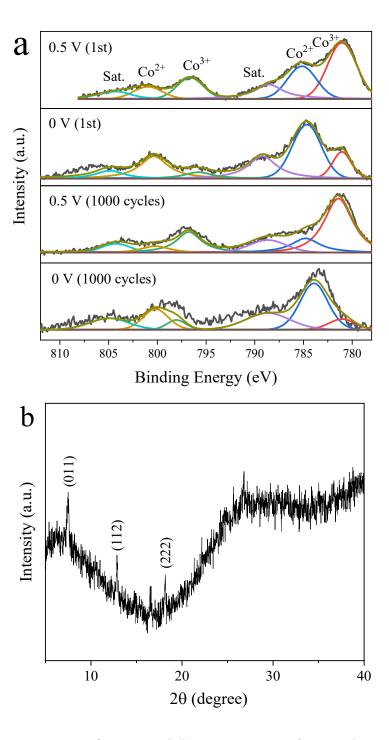
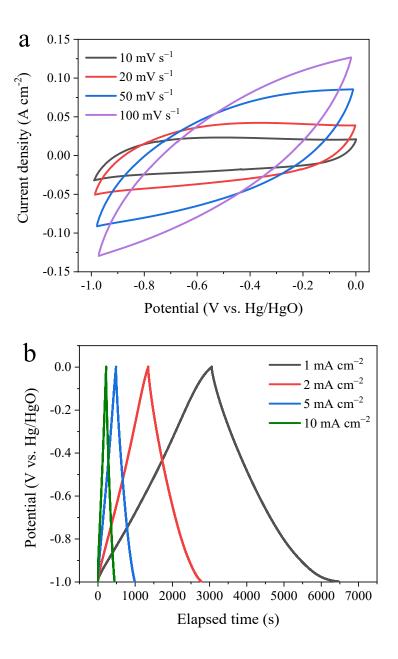


Fig. S10 TGA curves of ZIF-67/CC, CC and ZIF-67 powders.



**Fig. S11** (a) XPS spectra of Co 2p and (b) XRD pattern of ZIF-67/CC electrode after cycling.



**Fig. S12** (a) CV curves of AC/CC electrode at different scan rates. (b) Galvanostatic charge-discharge curves of AC/CC electrode at different current densities.

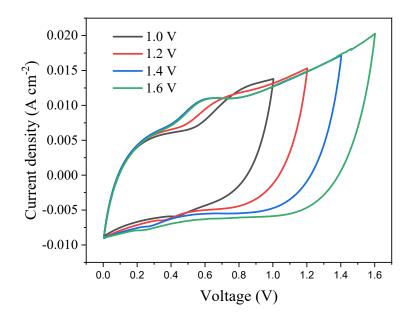


Fig. S13 CV curves of ZIF-67/CC//AC/CC ASSC device collected within different potential windows at  $10~\text{mV}~\text{s}^{-1}$ .

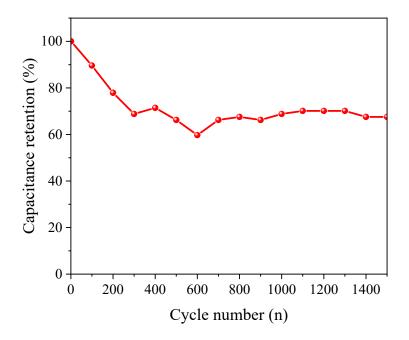


Fig. S14 Cycling performance of ZIF-67/CC//AC/CC ASSC device at 10 mA cm<sup>-2</sup>.

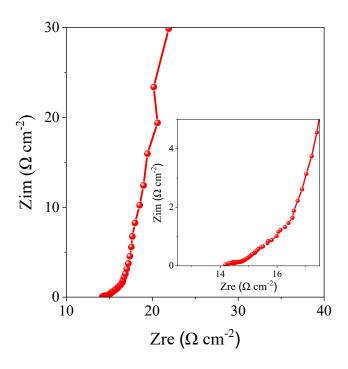


Fig. S15 Nyquist plot of the ASSC device.

Table S1 Capacitive performance comparison of reported MOF materials.

Materials	Electrolyte	Current	Capacitances	Rate capability	Cyclic stability	Ref.
		$(A g^{-1})$	$(F g^{-1})$	(%)	(%/cycles)	
Cu-CAT NWAs	3 M KCl	0.5	202	66 (0.5-10 A g <sup>-1</sup> )	80/5000	[11]
ZIF-67/PEDOT	1 M H <sub>2</sub> SO <sub>4</sub>	1	107	$55 (0.5-10 \text{ A g}^{-1})$	85/2000	[18]
ZIF-67 microflowers	1 M KOH	1	188.7	$73 (0.5-8 \text{ A g}^{-1})$	105/3000	[35]
PPy/Ni-CAT-NWs	3 M KCl	0.1	284	$70 (0.1 \text{-} 0.6 \text{ A g}^{-1})$	-	[36]
rGO/Cu-MOF fiber	1 M KCl	0.005 (mA cm <sup>-1</sup> )	2.17 (mF cm <sup>-1</sup> )	-	69/1000	[37]
ZIF-67/CC	1M LiOH	1	257	91 (1-10 mA cm <sup>-2</sup> )	99/10000	This work

**Table S2** Comparison of areal energy and power densities for some reported MOF-based devices.

Devices	Mass loading	Electrolysts	Energy density	Power density	Ref.
Devices	$(mg cm^{-2})$	Electrolyte	(mWh cm <sup>-2</sup> )	(mW cm <sup>-2</sup> )	
Ni@Z67//AC	-	3М КОН	$3.8 \times 10^{-2}$	0.7	[38]
$Ti_3C_2T_x/ZIF$ -	1-1.5	3М КОН	$3.6 \times 10^{-2}$	0.75	[26]
67/CoV <sub>2</sub> O <sub>6</sub> //AC	1-1.3				
ZIF-PPy*	10	PVA/Na <sub>2</sub> SO <sub>4</sub> gel	$1.1 \times 10^{-2}$	1.44	[39]
PANI-ZIF67-CC*	4	PVA/H <sub>2</sub> SO <sub>4</sub> gel	$4.4 \times 10^{-3}$	0.48	[4]
Co-MOF/NF//AC	6.656	2M KOH	1.7	4.0	[40]
Co-MOF/graphenen//AC	2.6	6M KOH	$2.1 \times 10^{-2}$	2.21	[41]
CoNi-MOF//AC	1	1М КОН	$2.9 \times 10^{-2}$	1.5	[12]
CC/CoNi-MOF//g-CNT	1.48	1M KOH	0.264	0.704	[42]
Ni-CAT/NiCo-	1.5	1М КОН	9.3×10 <sup>-2</sup>	18.3	[9]
LDH/NF//AC	1.3				
Cu <sub>3</sub> (HHTP) <sub>2</sub> //PPy	_	PVA/KCl	8.8×10 <sup>-5</sup>	4.5×10 <sup>-3</sup>	[43]
Cu <sub>3</sub> (111111 ) <sub>21</sub> /11 y		hydrogel			
Cu–CAT NWAs*	0.5	PVA/KCl gel	$5.2 \times 10^{-3}$	$10^{-4}$	[11]
Cu-CAT-NWAs/PPy	2.01	LiCl-PVA gel	$2.2 \times 10^{-2}$	0.47	[10]
ZIF-67/CC//AC	12	LiOH/PVA gel	380	1600	This work

<sup>\*</sup> Symmetric supercapacitor