

Electronic Supplementary Information

Iridium(III)-Catalyzed One-Pot Synthesis of Planar Chiral Emissive Materials through C–H Activation

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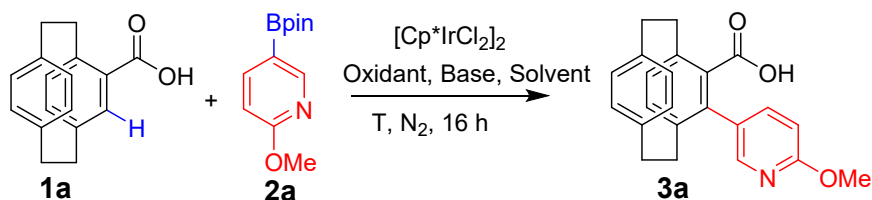
1. General Information

All solvents were used as received from commercial sources without further purification. Column flash chromatography was carried out on silica gel (200 – 400 mesh). Thin-layer chromatography (TLC) was performed on silica gel GF254. Reagents used to prepare the substrates and heteroaryl boron esters were purchased from Bidepharm and Energy Chemical without further purification. ^1H NMR and ^{13}C NMR spectra were recorded on Bruker-DRX (500 MHz and 126 MHz, respectively) instruments internally referenced to SiMe_4 , chloroform signals. HRMS spectra were recorded on an Agilent 100 ABI-API4000 spectrometer. X-ray data were collected on Bruker Smart APEX II CCD diffractometer. The optical rotation measurements were recorded on an SGW_®-2 automatic digital polarimeter (MA, China) at 589 nm wavelengths and at 28 °C by using DCM as the solvent (1 mg/mL). Chiral HPLC analysis of (*R*_p)/(*S*_p)-**1a**, **3a**, **3c**, **3h**, **3o**, and **3p** were performed using the Waters e2695 HPLC system with 2998PDA detector and CHIRALPAK IA column (250 × 4.6 mm, 5 μm) or CHIRALPAK IC column (250 × 4.6 mm, 5 μm); Mobile phase: mixed solvents of hexane and *iso*-propanol with the ratio of 95% : 5% as an eluent for **1a**, **3a**, and **3h**, 75% : 25% as an eluent for **3c**, 93% : 7% as an eluent for **3o**, and 98% : 2% as an eluent for **3p**; Flow rate: 1.0 mL/min for **1a**, **3a**, **3c**, **3h**, **3o**, and **3p**; Column temperature: 20 °C for **1a**; 35 °C for **3a**, **3c**, and **3h**; 30 °C for **3o** and **3p**. The absorbance spectra measurement was performed on a T9CS UV–vis spectrophotometer (Persee Instrument Co., Ltd. Beijing, China). The fluorescence spectra were measured on F-7100 (Hitachi, Japan) fluorescence spectrofluorometer (the pathlength of the quartz cell is 1 cm) with a xenon arc lamp as the light source. Circular dichroism (CD) spectra of (*R*_p)/(*S*_p)-**3a**, (*R*_p)/(*S*_p)-**3c**, (*R*_p)/(*S*_p)-**3h**, (*R*_p)/(*S*_p)-**3o**, and (*R*_p)/(*S*_p)-**3p** in THF solution (5.0×10^{-5} mol/L) were measured using a J-810-150s spectropolarimeter (JASCO J-1500·spectrophotometer, Japan), at room temperature (cell length: 10 mm, bandwidth: 1 nm, scanning speed: 100 nm/min, data pitch: 1 nm, accumulations: 2). The CD spectra were approximated using the simple moving average (SMA) method. Circularly polarized luminescence (CPL) spectra in THF solution (5.0×10^{-5} mol/L) and for (*R*_p)/(*S*_p)-**3a** and (*R*_p)/(*S*_p)-**3h** were recorded with a JASCO CPL-200 spectrofluoropolarimeter at room temperature. The CPL spectra were approximated using the SMA method. ($\lambda_{\text{ex}} = 310$ nm, cell length: 5 mm, E_x & E_m slit width: 3000 μm, scanning speed: 200 nm/min, data pitch: 1 nm, accumulations: 8).

2. Experimental Section

2.1 Optimization of Reaction Conditions

Table S1: Optimization of Reaction Conditions^a



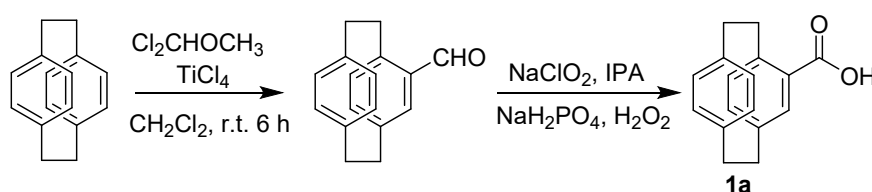
Entry	Oxidant	Base	Solvent	T (°C)	Yield (%)
1	Ag_2O	KH_2PO_4	THF	90	nr
2	Ag_2O	K_3PO_4	THF	90	nr
3	Ag_2O	KOAc	THF	90	68
4	Ag_2O	KF	THF	90	12
5	Ag_2O	K_2CO_3	THF	90	14
6	Ag_2O	KHCO_3	THF	90	10
7	Ag_2O	NaOAc	THF	90	45
8	Ag_2O	NaHCO_3	THF	90	14
9	Ag_2O	NaF	THF	90	13
10	Ag_2O	Li_2CO_3	THF	90	10
11	Ag_2O	LiOAc	THF	90	47
12	Ag_2O	LiF	THF	90	15
13	Ag_2O	Cs_2CO_3	THF	90	nr
14	Ag_2O	CsOAc	THF	90	50
15	PhCOOAg	KOAc	THF	90	nr
16	Ag_2CO_3	KOAc	THF	90	nr
17	AgOPiv	KOAc	THF	90	nr
18	AgOAc	KOAc	THF	90	nr
19	AgNO_3	KOAc	THF	90	nr
20	Ag_2SO_4	KOAc	THF	90	nr
21	AgF	KOAc	THF	90	nr
22	AgNTf_2	KOAc	THF	90	nr
23	Ag_2O	KOAc	Tol	90	35
24	Ag_2O	KOAc	1,4-Diox	90	57
25	Ag_2O	KOAc	DMSO	90	nr
26	Ag_2O	KOAc	DMF	90	nr
27	Ag_2O	KOAc	THF	130	67
28 ^b	Ag_2O	KOAc	THF	90	41
29 ^c	Ag_2O	KOAc	THF	90	nr
30 ^d	-	KOAc	THF	90	nr

31 ^e	Ag ₂ O	-	THF	90	nr
32 ^f	Ag ₂ O	KOAc	THF	90	nr

^aReaction Conditions: **1a** (25.2 mg, 0.1 mmol), **2a** (47.0 mg, 0.2 mmol), [Cp*IrCl₂]₂ (4.0 mg, 0.005 mmol), Oxidant (0.2 mmol), Base (0.2 mmol), Solvent (2 mL), 90 °C, 16 h, Nitrogen atmosphere. Yields were analyzed by ¹H NMR spectroscopy with 1,3,5-trimethoxybenzene as an internal standard; ^b[Cp*Ir(MeCN)₃](SbF₆)₂ (9.2 mg, 0.01 mmol) as the catalyst; ^c[Cp*RhCl₂]₂ (6.2 mg, 0.01 mmol) as the catalyst; ^dWithout Ag₂O; ^eWithout KOAc; ^fWithout the catalyst.

2.2 General Procedure for the Preparation of Substrates

I. The synthesis of the substrate 4-formyl[2.2]paracyclophane **1a**.



Scheme S1. Synthetic routes for **1a**.

The synthesis of the substrate 4-formyl[2.2]paracyclophane.¹

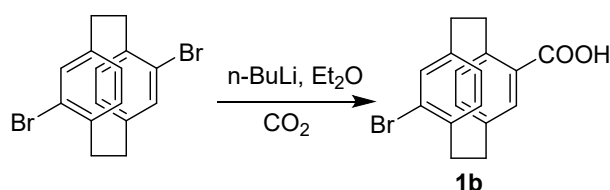
TiCl₄ (0.21 mL, 1.92 mmol) was added to a solution of [2.2]paracyclophane (0.20 g, 0.96 mmol) in dry DCM (20 mL). After the mixture was stirred at room temperature for 5 minutes, 1,1-dichlorodimethyl ether (93 μL, 1.06 mmol) was added, the resulting mixture was allowed to warm to room temperature. After stirring at r.t. for 6 h, the black solution was quenched by distilled water and stirred for 1 h until it became blue. The organic phase was separated and the aqueous phase was extracted with DCM (3 × 50 mL). The combined organic layers were washed with brine, dried over MgSO₄ and gravity-filtered. The solvent was removed under reduced pressure and the crude product was purified via silica gel column chromatography using dichloromethane/petroleum ether (1:2) as the eluent. 4-Formyl[2.2]paracyclophane was isolated as amorphous white solid (0.22 g, 95% yield).

The synthesis of the substrate: 4-carboxy[2.2]paracyclophane **1a**.²

4-Formyl[2.2]paracyclophane (0.17 g, 0.72 mmol) was dissolved in propan-2-ol (IPA) (10 mL). After the pH value of the mixed solution was adjusted to 4.5 by adding

sodium dihydrogen phosphate solution (8%), H₂O₂ (35%, 89 μ L, 1.1 mmol) was added dropwise. Then, sodium chlorite solution (0.09 g, 2 mol/L) was added dropwise to the mixture over 30 min and the reaction mixture was stirred at room temperature for 12 h. The sodium sulfite was added to destroy the oxidant, then the organic solvent was removed under reduced pressure and the residuum was acidified with dilute sulfuric acid to pH = 3~4. The organic phase was separated, and the aqueous phase was extracted with ethyl acetate (3 \times 10 mL). The organic solvent was removed under reduced pressure and the crude product was purified by column chromatography on silica gel (dichloromethane) to give the product **1a** (0.13 g, 72%). ¹H NMR (500 MHz, CDCl₃) δ 11.48 (s, 1H), 7.29 (d, *J* = 2.0 Hz, 1H), 6.72 (dd, *J* = 7.8, 2.0 Hz, 1H), 6.62 – 6.56 (m, 3H), 6.51 (ddt, *J* = 6.4, 4.6, 1.9 Hz, 2H), 4.21 (ddd, *J* = 12.9, 9.4, 2.3 Hz, 1H), 3.26 – 3.13 (m, 4H), 3.11 – 3.00 (m, 2H), 2.90 (ddd, *J* = 13.0, 10.0, 7.2 Hz, 1H). ¹³C{¹H} NMR (126 MHz, CDCl₃): δ 172.3, 143.8, 140.1, 140.0, 139.4, 137.4, 136.4, 136.2, 133.1, 132.8, 132.3, 131.8, 129.6, 36.3, 35.2, 35.1, 34.9. HRMS (ESI) *m/z*: Calcd for C₁₇H₁₆O₂: [M+Na]⁺ 275.1043; Found: [M+Na]⁺ 275.1038.

II. The synthesis of the substrate: 4-bromo-16-carboxy[2.2]paracyclophane **1b**.³

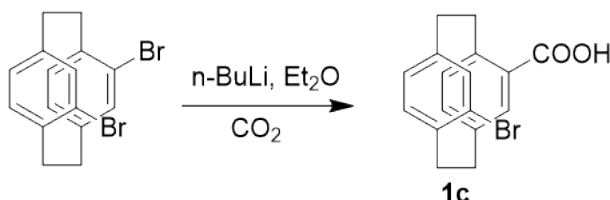


Scheme S2. Synthetic route for **1b**.

4,16-Dibromo-[2.2]paracyclophane (2.0 g, 5.5 mmol) was dissolved in Et₂O (30 mL), then at -78 °C, n-butyllithium (3.3 mL, 2.8 M solution in hexane, 8.3 mmol) was added dropwise to the above-mixed solution under argon. The reaction mixture was stirred at room temperature for 2 h, then an excess of dry ice (10 g) was added. The resulting mixture was allowed to warm to room temperature, the organic solvent was removed under reduced pressure and the crude product was dissolved in H₂O (200 mL). The insoluble 4,16-dibromo-[2.2]paracyclophane was isolated by filtration and the aqueous phase was thoroughly washed with ether (3 \times 50 mL) and CH₂Cl₂ (3 \times 50 mL). The organic solvent was removed under reduced pressure to obtain final product **1b** (1.11 g, 61%). ¹H NMR (500 MHz, DMSO-*d*₆) δ 12.60 (s, 1H), 7.21 (d, *J* = 7.6 Hz, 1H), 7.08 (d, *J* = 2.0 Hz, 1H), 6.63 (s, 1H), 6.51 (m, 3H), 4.02 (t, *J* = 11.2 Hz, 1H), 3.16 – 3.00 (m, 4H), 2.99 – 2.91 (m, 2H), 2.85 (m, 1H). ¹³C{¹H} NMR (126 MHz, DMSO-

d_6): δ 168.7, 157.0, 142.4, 142.2, 139.3, 138.9, 137.0, 136.1, 135.2, 134.6, 132.0, 130.6, 127.1, 35.3, 34.8, 33.9, 33.0. HRMS (ESI) m/z : Calcd for $C_{17}H_{15}BrO_2$: $[M+K]^+$ 368.9887; Found: $[M+K]^+$ 368.9885.

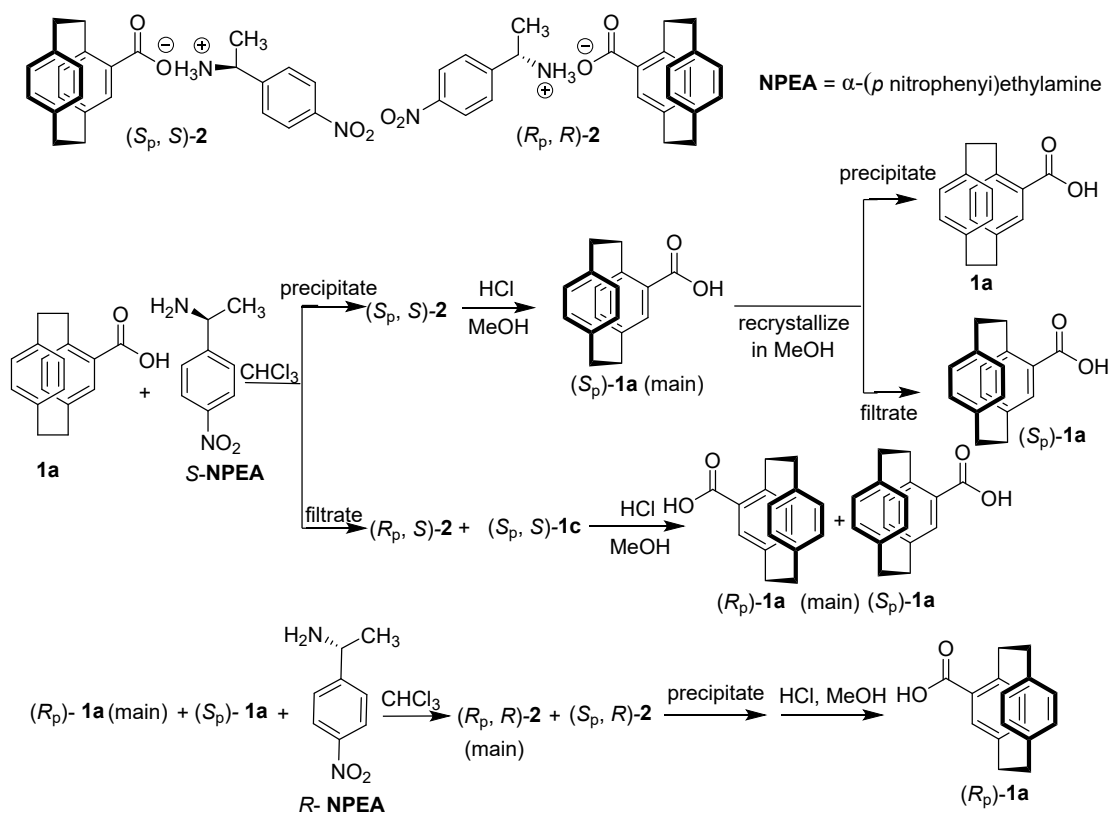
III. The synthesis of the substrate: 4-bromo-12-carboxy[2.2]paracyclophane **1c**.³



Scheme S3. Synthetic route for **1c**.

White solid (0.93 g, 51%). The product **1c** was obtained by flash column chromatography on silica gel using petroleum ether/dichloromethane = 1:2 as the eluent. ¹H NMR (500 MHz, CDCl₃): δ 11.60 (s, 1H), 7.93 (d, J = 1.9 Hz, 1H), 6.71 (dd, J = 7.9, 1.9 Hz, 1H), 6.69 – 6.63 (m, 2H), 6.61 (dd, J = 7.7, 1.8 Hz, 1H), 6.51 (d, J = 7.8 Hz, 1H), 4.23 – 4.15 (m, 1H), 3.54 – 3.47 (m, 1H), 3.27 (ddd, J = 13.2, 10.1, 6.8 Hz, 1H), 3.20 – 3.12 (m, 2H), 3.08 (m, 1H), 2.85 (dddd, J = 28.6, 13.4, 10.3, 7.1 Hz, 2H). ¹³C {¹H} NMR (126 MHz, CDCl₃): δ 171.5, 143.1, 142.1, 139.7, 139.0, 137.7, 136.2, 135.8, 135.0, 131.8, 131.1, 129.2, 126.8, 36.2, 35.8, 34.1, 32.5. HRMS (ESI) m/z : Calcd for $C_{17}H_{15}BrO_2$: $[M+H]^+$ 331.0328; Found: $[M+H]^+$ 331.0332.

IV. The synthesis of the substrate: (*R*_p)/(*S*_p)-4-carboxy[2.2]paracyclophane ((*R*_p)/(*S*_p)-**1a**).⁴



Scheme S4. Synthetic routes for (*S_p*)-**1a** and (*R_p*)-**1a**.

(*S*)-1-(4-Nitrophenyl)ethylamine hydrochloride ((*S*)-NPEA·HCl) (0.4 g, 1.9 mmol) was dissolved in ethanol and reacted with 30% sodium hydroxide solution and the reaction was monitored to be completed by TLC. The ethanol was removed under reduced pressure, and the aqueous phase was extracted with DCM (3 × 20 mL) and the combined organics was concentrated under reduced pressure to give (–)-(*S*)-NPEA (0.3 g, 80%). The (+)-(*R*)-NPEA (0.15g, 80%) was also synthesized following the above procedure.

The racemic **1a** (0.3 g, 1.2 mmol) and (–)-(*S*)-NPEA (0.2 g, 1.3 mmol) were dissolved in CHCl₃ (15 mL) and stirred at room temperature for 1 h. Then the reaction mixture was stirred at 50 °C for 2 h until the white solid precipitated from the solution. To complete sedimentation, the reaction mixture was stored overnight at –5 °C. The precipitate was isolated by filtration and dried to obtain the main compound (*S_p*, *S*)-**2**. Then the main compound (*S_p*, *S*)-**2** was dissolved in methanol and hydrolyzed with 2 mol/L HCl. The precipitated solid was washed twice with H₂O (2 × 30 mL) to obtain the crude product and the crude product was recrystallized in methanol. The solid was

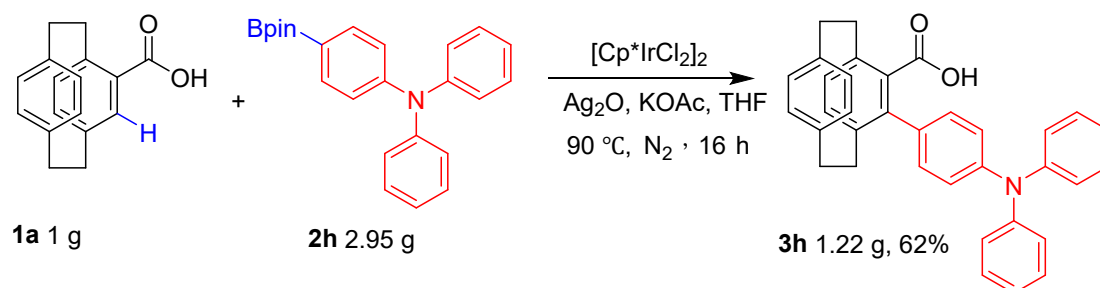
removed by suction filtration, and the filtrate was concentrated under reduced pressure to give pure (*S_p*)-**1a** (0.06 g, 20%), $[\alpha]_D^{25} = +147$ ($c = 0.1$, CH₂Cl₂).

The CHCl₃ filtrate was evaporated and hydrolyzed to obtain a partially resolved (*R_p*)-**1a**. The mixture of (*R_p*)-**1a** (main product) and (*S_p*)-**1a** (0.2 g, 0.8 mmol) and (+)-(*R*)-NPEA (0.15 g, 0.9 mmol) in CHCl₃ was stirred at room temperature for 1 h, then stirred at 50 °C for 2 h until the white optically pure diastereomers precipitated from the solution and stayed overnight at -5 °C to ensure enough diastereomer precipitation (*R_p*, *R*)-**2**. After recrystallization of (*S_p*, *S*)-**2** from ethanol and then hydrolysis, optically pure (*R_p*)-**1a** was obtained (0.09 g, 30%). $[\alpha]_D^{25} = -159$ ($c = 0.1$, CH₂Cl₂).

2.3 General Procedure for Ir(III)-Catalyzed C–H Arylation.⁵

To a 50 mL Schlenk-type sealed tube equipped with a magnetic stirring bar was added the substrate **1a** (0.1 mmol), [Cp*IrCl₂]₂ (4.0 mg, 0.005 mmol), 2-methoxypyridine boronic acid pinacol ester (0.2 mmol, 2 equiv), Ag₂O (46.0 mg, 0.2 mmol, 2 equiv), KOAc (19.6 mg, 0.2 mmol, 2 equiv) and dry THF (2.0 mL) under N₂ atmosphere. The tube was capped and subjected to a 90 °C preheated oil bath for 16 h. After cooling to room temperature, the reaction mixture was acidified with diluted hydrochloric acid (2 mol/L) to pH = 4~5. The filtrate was concentrated in vacuo to afford crude products, which was purified by column chromatography on silica gel (petroleum ether/ethyl acetate = 5:1) to give the pure product **3a**.

2.4 Large-Scale Synthesis.

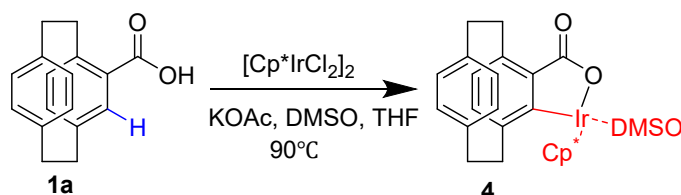


Scheme S5. Synthetic route for compound **3h**.

To a 50 mL three-necked flask with a magnetic stirring bar was added the substrate **1a** (1.0 g, 3.97 mmol), [Cp*IrCl₂]₂ (0.16 g, 0.19 mmol, 0.05 equiv), **2h** (2.95 g, 7.94 mmol), Ag₂O (1.84 g, 7.94 mmol, 2 equiv), KOAc (0.78 g, 7.94 mmol, 2 equiv) and dry THF (40 mL) under N₂ atmosphere and subjected to a 90 °C preheated oil bath for

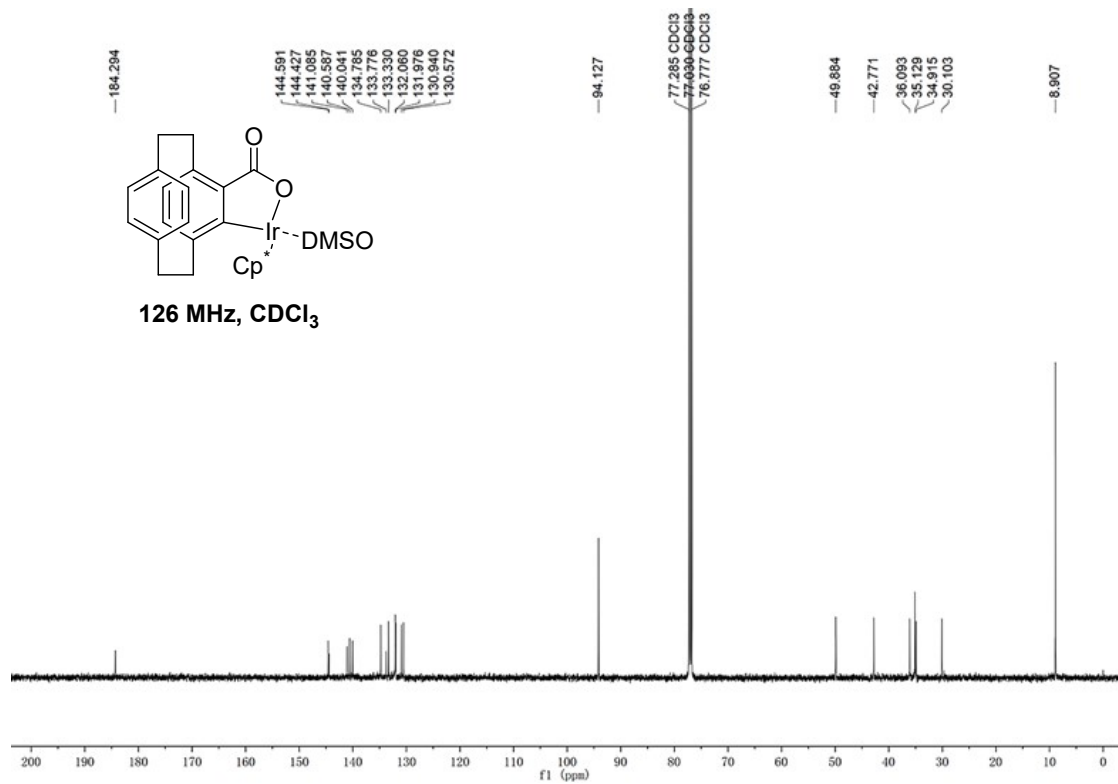
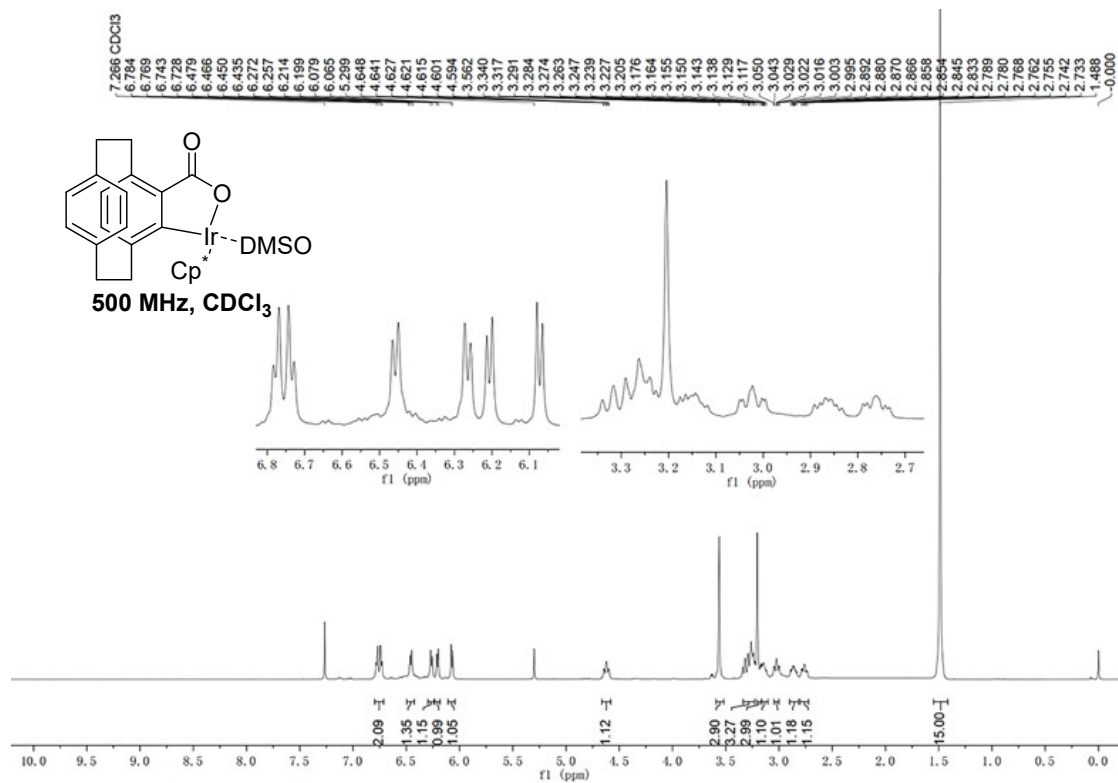
16 h. After cooling to room temperature, the reaction mixture was acidified with diluted hydrochloric acid (2 mol/L) to pH = 4~5, and then was filtered through a pad of Celite. The filtrate was concentrated in vacuo to afford crude products, which was purified by flash column chromatography on silica gel using dichloromethane/petroleum ether = 10:1 as the eluent to give the pure product **3h** (1.22 g, 62%).

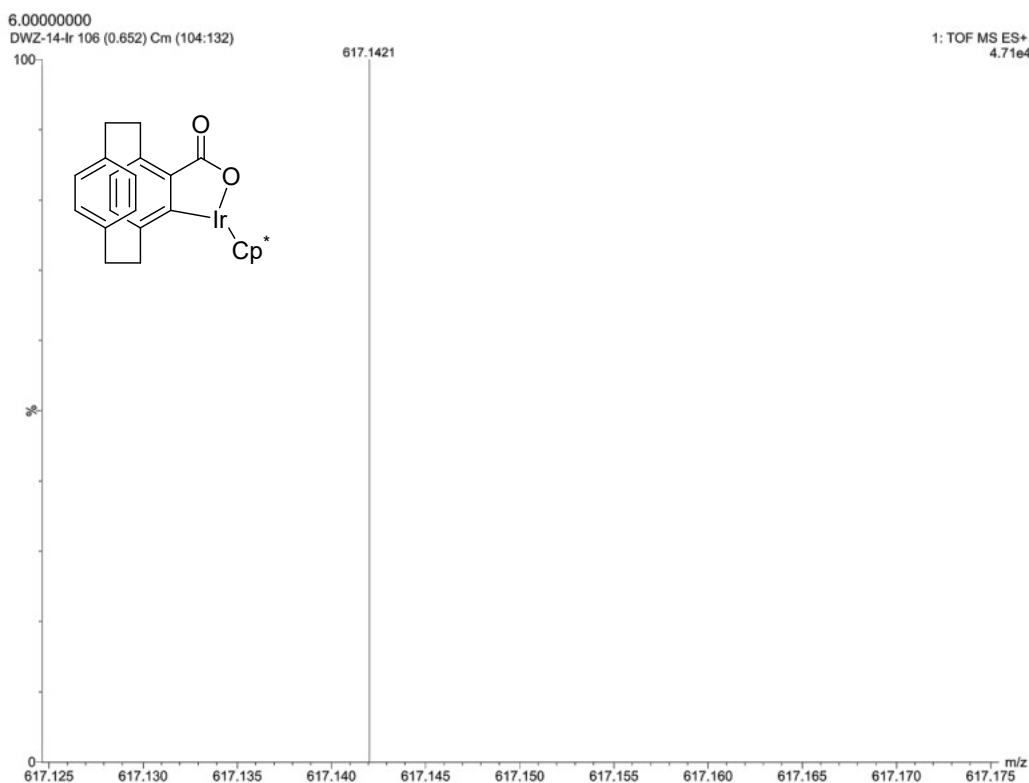
3. Catalytic Activity of Intermediate **4**



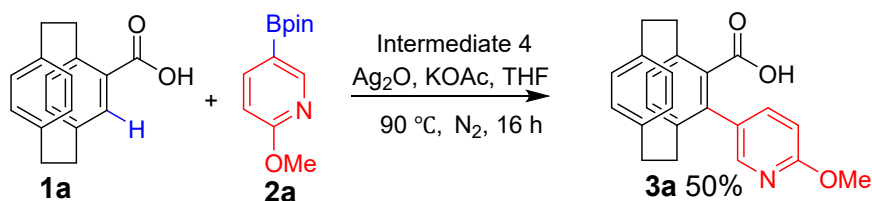
Scheme S6. Synthetic route for ntermediate **4**.

An oven-dried 25 mL Schlenk tube equipped with magnetic stirring bar was sequentially charged with **1a** (25.2 mg, 0.1 mmol), $[\text{Cp}^*\text{IrCl}_2]_2$ (40.0 mg, 0.05 mmol, 0.5 equiv), KOAc (19.6 mg, 0.2 mmol, 2 equiv), DMSO (14.2 μL , 0.2 mmol, 2 equiv) and THF (2 mL) in the air. The reaction mixture was stirred at 90 °C in oil bath for 2 h. A light yellow solid precipitate was formed, and the precipitate was isolated by filtration. The crude product was recrystallized in dichloromethane/petroleum ether and the intermediate **4** was collected as yellow solid (27 mg, 41%). ^1H NMR (500 MHz, CDCl_3) δ 6.80 – 6.71 (m, 2H), 6.46 (d, $J = 7.9$ Hz, 1H), 6.26 (d, $J = 7.8$ Hz, 1H), 6.21 (d, $J = 7.3$ Hz, 1H), 6.07 (d, $J = 7.3$ Hz, 1H), 4.62 (m, 1H), 3.56 (s, 3H), 3.36 – 3.23 (m, 3H), 3.20 (s, 3H), 3.14 (m, 1H), 3.05 – 3.00 (m, 1H), 2.86 (ddd, $J = 12.7, 10.5, 5.9$ Hz, 1H), 2.80 – 2.72 (m, 1H), 1.49 (s, 15H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 184.3, 144.6, 144.4, 141.1, 140.6, 140.0, 134.8, 133.8, 133.3, 132.1, 132.0, 130.9, 130.6, 94.1, 49.9, 42.8, 36.1, 35.1, 34.9, 30.1, 8.9. HRMS (ESI) m/z : Calcd for $\text{C}_{27}\text{H}_{29}\text{IrO}_2$: $[\text{M}+\text{K}]^+$ 617.1428; Found: $[\text{M}+\text{K}]^+$ 617.1421.





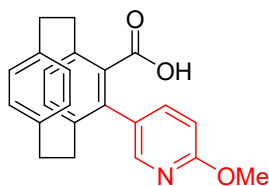
Intermediate 4 instead of $[\text{Cp}^*\text{IrCl}_2]_2$ was used in model reaction



Scheme S7. Synthetic route for **3a**.

To a 25 mL Schlenk-type sealed tube equipped with a magnetic stirring bar was added the substrate **1a** (25.2 mg, 0.1 mmol), intermediate **4** (6.6 mg, 0.01 mmol, 0.1 equiv), **2a** (47.0 mg, 0.2 mmol, 2 equiv), Ag_2O (46.4 mg, 0.2 mmol, 2 equiv), KOAc (19.6 mg, 0.2 mmol, 2 equiv), and dry THF (2 mL) under N_2 atmosphere. The tube was capped and subjected to a 90 °C preheated oil bath for 16 h. After cooling to room temperature, the reaction mixture was acidified with diluted hydrochloric acid (2 mol/L) to pH = 4~5, and then was filtered through a pad of Celite. The filtrate was concentrated in vacuo to afford crude products, which was purified by flash column chromatography on silica gel (petroleum ether/ethyl acetate = 3:1) to obtain the desired product **3a** (18 mg, 50%).

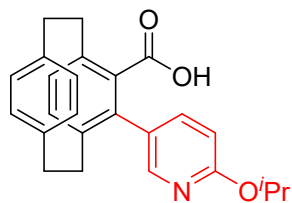
4. Characterization of New Compounds.



(*R*_p)/(*S*_p)-**3a**:

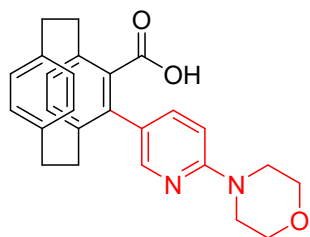
White solid (24.4 mg, 68%). The product **3a** was obtained by flash column chromatography on silica gel using petroleum ether/ethyl acetate = 3:1 as the eluent. *R*_p [α]_D²⁵ = +20.9 (*c* = 0.1, CH₂Cl₂), *S*_p [α]_D²⁵ = -22.1 (*c* = 0.1, CH₂Cl₂). ¹H NMR (500 MHz, CDCl₃) δ 12.34 (s, 1H), 9.03 – 8.86 (m, 1H), 7.49 (d, *J* = 19.8 Hz, 2H), 6.83 (d, *J* = 8.6 Hz, 2H), 6.76 – 6.70 (m, 2H), 6.67 (dd, *J* = 7.8, 1.8 Hz, 1H), 6.59 (d, *J* = 7.8 Hz, 1H), 3.80 (s, 3H), 3.57 – 3.38 (m, 2H), 3.14 (t, *J* = 11.1 Hz, 1H), 3.05 – 2.80 (m, 5H). ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 171.2, 162.9, 145.7, 142.9, 140.4, 140.3, 139.2, 138.5, 136.9, 136.1, 134.9, 132.9, 132.1, 131.6, 130.6, 129.7, 129.4, 111.1, 54.2, 35.3, 34.8, 33.0, 29.7. HRMS (ESI) *m/z*: Calcd for C₂₃H₂₁NO₃: [M+H]⁺ 360.1594; Found [M+H]⁺ 360.1592.

The yields of the products (**3a**: 68%) was determined by ¹H NMR analysis of the crude product using 1,3,5-trimethoxybenzene as the internal standard.



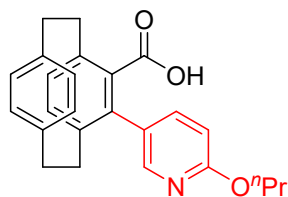
3b:

White solid (20.9 mg, 54%). The product **3b** was obtained by flash column chromatography on silica gel using petroleum ether/ethyl acetate = 3:1 as the eluent. ¹H NMR (500 MHz, CDCl₃) δ 11.97 (s, 1H), 8.95 (s, 1H), 7.55 – 7.34 (m, 2H), 6.76 (s, 1H), 6.64 (dd, *J* = 15.1, 7.8 Hz, 3H), 6.60 (d, *J* = 7.8 Hz, 1H), 6.51 (d, *J* = 7.9 Hz, 1H), 4.81 (s, 1H), 3.41 (s, 2H), 3.13 – 3.02 (m, 1H), 2.97 – 2.88 (m, 2H), 2.81 (m, 3H), 1.26 – 1.16 (m, 6H). ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 171.1, 162.0, 145.9, 142.8, 140.9, 140.4, 139.2, 138.5, 136.9, 136.6, 134.8, 133.1, 132.9, 132.0, 131.7, 130.4, 129.9, 111.9, 68.9, 35.4, 34.8, 33.0, 29.7, 22.1. HRMS (ESI) *m/z*: Calcd for C₂₅H₂₅NO₃: [M+H]⁺ 388.1907; Found: [M+H]⁺ 388.1906.



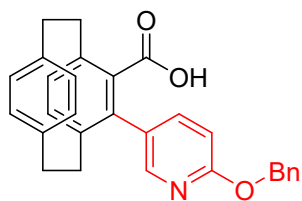
(R_p)/(S_p)-3c:

White solid (26.1 mg, 63%). The product **3c** was obtained by flash column chromatography on silica gel using petroleum ether/ethyl acetate = 3:1 as the eluent. *R_p* $[\alpha]_D^{25} = 64.0$ ($c = 0.1$, CH₂Cl₂), *S_p* $[\alpha]_D^{25} = -64.6$ ($c = 0.1$, CH₂Cl₂). ¹H NMR (500 MHz, CDCl₃) δ 8.90 (s, 1H), 7.40 (s, 2H), 6.82 – 6.67 (m, 4H), 6.65 (d, $J = 7.8$, 1H), 6.55 (d, $J = 7.8$ Hz, 1H), 3.74 (t, $J = 5.0$ Hz, 4H), 3.42 (m, 6H), 2.97 (t, $J = 7.2$ Hz, 2H), 2.87 (m, 4H). ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 170.5, 156.7, 145.8, 140.8, 139.2, 138.9, 138.2, 137.6, 135.7, 133.6, 131.8, 131.0, 130.6, 129.9, 128.9, 128.8, 125.2, 106.7, 65.5, 44.9, 41.0, 34.3, 33.8, 32.1, 28.7, 28.3. HRMS (ESI) m/z : Calcd for C₂₆H₂₆N₂O₃: [M+H]⁺ 415.2016; Found: [M+H]⁺ 415.2014.



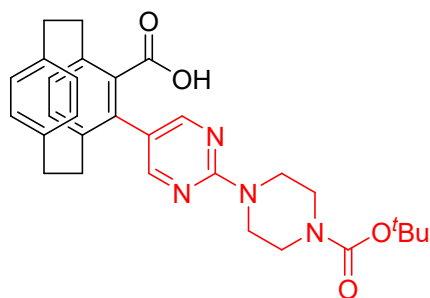
3d:

White solid (18.6 mg, 48%). The product **3d** was obtained by flash column chromatography on silica gel using dichloromethane/ethyl acetate = 10:1 as the eluent. ¹H NMR (500 MHz, CDCl₃) δ 11.77 (s, 1H), 8.92 (s, 1H), 7.47 (s, 2H), 6.81 (d, $J = 8.6$ Hz, 2H), 6.76 – 6.69 (m, 2H), 6.66 (dd, $J = 7.8, 1.9$ Hz, 1H), 6.57 (d, $J = 7.8$ Hz, 1H), 4.12 (dt, $J = 9.9, 6.5$ Hz, 1H), 3.97 (s, 1H), 3.46 (t, $J = 12.5$ Hz, 2H), 3.13 (t, $J = 11.0$ Hz, 1H), 3.05 – 2.95 (m, 2H), 2.89 (m, 3H), 1.34 – 1.23 (m, 2H), 0.93 (m, 3H). ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 171.4, 162.8, 145.8, 142.8, 140.3, 140.3, 139.2, 138.5, 136.8, 136.3, 134.8, 132.9, 132.0, 131.6, 130.7, 129.8, 129.1, 111.2, 68.3, 35.4, 34.8, 33.0, 29.7, 22.2, 10.5. HRMS (ESI) m/z : Calcd for C₂₅H₂₅NO₃: [M+H]⁺ 388.1907; Found: [M+H]⁺ 388.1908.



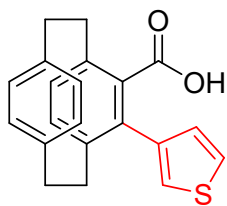
3e:

White solid (27.8 mg, 64%). The product **3e** was obtained by flash column chromatography on silica gel using petroleum ether/acetone = 8:1 as the eluent. ^1H NMR (500 MHz, CDCl_3) δ 11.65 (s, 1H), 8.93 (s, 1H), 7.51 (s, 1H), 7.35 (m, 6H), 6.89 (d, $J = 8.5$ Hz, 1H), 6.81 (s, 1H), 6.75 – 6.69 (m, 2H), 6.66 (d, $J = 7.8$ Hz, 1H), 6.57 (d, $J = 7.7$ Hz, 1H), 5.29 (d, $J = 11.2$ Hz, 1H), 5.08 (s, 1H), 3.57 – 3.34 (m, 2H), 3.14 (t, $J = 11.2$ Hz, 1H), 2.94 (m, 5H). ^{13}C $\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 171.7, 162.6, 145.7, 143.0, 140.1, 140.1, 139.3, 138.7, 137.0, 136.4, 136.1, 134.9, 133.0, 132.1, 131.6, 130.7, 130.6, 129.8, 128.5, 128.2, 128.1, 111.5, 68.8, 35.4, 34.9, 34.7, 33.1. HRMS (ESI) m/z : Calcd for $\text{C}_{29}\text{H}_{25}\text{NO}_3$: $[\text{M}+\text{H}]^+$ 436.1907; Found: $[\text{M}+\text{H}]^+$ 436.1911.



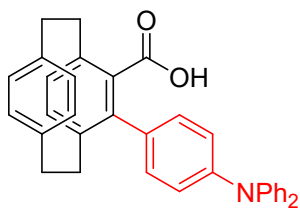
3f:

White solid (34.4 mg, 67%). The product **3f** was obtained by flash column chromatography on silica gel using petroleum ether/acetone = 8:1 as the eluent. ^1H NMR (500 MHz, CDCl_3) δ 12.40 (s, 1H), 8.98 (s, 1H), 8.30 (m, 1H), 7.30 (dd, $J = 7.9$, 1.9 Hz, 1H), 6.76 (d, $J = 7.8$ Hz, 1H), 6.73 – 6.63 (m, 3H), 6.61 (d, $J = 7.8$ Hz, 1H), 3.74 (s, 4H), 3.53 – 3.37 (m, 6H), 3.14 (t, $J = 11.2$ Hz, 1H), 3.08 – 2.97 (m, 2H), 2.91 (m, 3H), 1.46 (s, 9H). ^{13}C $\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 171.0, 159.5, 154.7, 140.5, 140.2, 139.2, 138.6, 137.0, 135.0, 133.4, 133.0, 132.3, 131.4, 130.6, 129.6, 122.5, 80.2, 43.9, 35.3, 34.8, 34.7, 33.1, 31.6, 28.4, 22.7, 14.1. HRMS (ESI) m/z : Calcd for $\text{C}_{30}\text{H}_{34}\text{N}_4\text{O}_4$: $[\text{M}+\text{H}]^+$ 515.2653; Found: $[\text{M}+\text{H}]^+$ 515.2654.



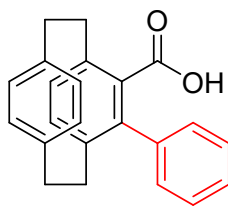
3g:

White solid (11.6 mg, 35%). The product **3g** was obtained by flash column chromatography on silica gel using petroleum ether/ethyl acetate = 5:1 as the eluent. ^1H NMR (500 MHz, CDCl_3) δ 7.41 (dd, $J = 5.0, 2.9$ Hz, 1H), 7.35 – 7.30 (m, 1H), 7.11 (d, $J = 4.8$ Hz, 1H), 6.94 (dd, $J = 7.9, 1.9$ Hz, 1H), 6.70 (dd, $J = 7.8, 1.9$ Hz, 1H), 6.68 – 6.62 (m, 2H), 6.57 (dd, $J = 8.0, 1.9$ Hz, 1H), 6.51 (d, $J = 7.8$ Hz, 1H), 3.31 (d, $J = 9.1$ Hz, 2H), 3.13 (t, $J = 10.2$ Hz, 1H), 3.03 (ddd, $J = 14.9, 9.5, 5.9$ Hz, 1H), 2.97 – 2.90 (m, 2H), 2.84 (m, 2H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 172.6, 147.6, 139.7, 139.7, 139.1, 138.7, 137.0, 134.1, 133.0, 132.9, 131.9, 131.5, 129.9, 129.3, 124.8, 123.2, 122.6, 35.2, 35.0, 34.4, 33.1. HRMS (ESI) m/z : Calcd for $\text{C}_{21}\text{H}_{18}\text{O}_2\text{S}$: $[\text{M}+\text{H}]^+$ 335.1100; Found: $[\text{M}+\text{H}]^+$ 335.1106.



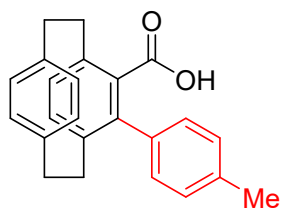
(R_p)/(S_p)-3h:

White solid (32.1 mg, 65%). The product **3h** was obtained by flash column chromatography on silica gel using dichloromethane/petroleum ether = 10:1 as the eluent. R_p $[\alpha]_D^{25} = 107$ ($c = 0.1$, CH_2Cl_2), S_p $[\alpha]_D^{25} = -105$ ($c = 0.1$, CH_2Cl_2). ^1H NMR (500 MHz, CDCl_3) δ 7.53 – 7.35 (m, 1H), 7.27 (m, 3H), 7.25 (s, 1H), 7.15 (d, $J = 7.9$ Hz, 4H), 7.09 (d, $J = 8.1$ Hz, 2H), 7.03 (t, $J = 7.3$ Hz, 3H), 6.88 (dd, $J = 7.9, 1.8$ Hz, 1H), 6.78 – 6.62 (m, 3H), 6.55 (dd, $J = 7.9, 1.8$ Hz, 1H), 6.48 (d, $J = 7.8$ Hz, 1H), 3.42 – 3.27 (m, 2H), 3.20 – 3.10 (m, 1H), 3.07 – 2.89 (m, 4H), 2.82 (m, 1H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 172.6, 147.6, 147.1, 147.1, 139.7, 139.7, 139.6, 139.1, 138.7, 138.7, 137.0, 134.1, 133.0, 132.9, 132.9, 131.9, 131.6, 131.5, 129.9, 129.5, 129.4, 129.3, 124.8, 123.2, 122.6, 35.2, 35.0, 34.4, 33.1. HRMS (ESI) m/z : Calcd for $\text{C}_{35}\text{H}_{29}\text{NO}_2$: $[\text{M}+\text{H}]^+$ 496.2271; Found: $[\text{M}+\text{H}]^+$ 496.2278.



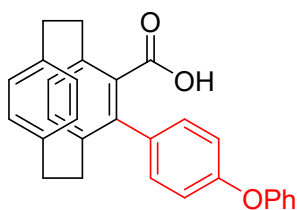
3i:

White solid (12.7 mg, 39%). The product **3i** was obtained by preparative thin-layer chromatography using dichloromethane/petroleum ether/triethylamine = 20:1:1 as the eluent. ^1H NMR (500 MHz, CDCl_3) δ 7.56 (s, 1H), 7.49 – 7.25, (m, 3H), 7.07 (s, 1H), 6.86 (dd, $J = 8.0, 1.8$ Hz, 1H), 6.66 – 6.56 (m, 3H), 6.50 – 6.39 (m, 2H), 3.28 (m, 2H), 3.12 – 3.04 (m, 1H), 2.95 – 2.80 (m, 3H), 2.80 – 2.67 (m, 2H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 171.3, 138.8, 138.7, 138.6, 138.4, 138.1, 137.6, 137.2, 136.0, 133.3, 131.8, 130.9, 130.5, 128.9, 128.4, 127.4, 126.4, 34.2, 33.9, 33.3, 32.0. HRMS (ESI) m/z : Calcd for $\text{C}_{23}\text{H}_{20}\text{O}_2$: $[\text{M}+\text{H}]^+$ 329.1536; Found: $[\text{M}+\text{H}]^+$ 329.1544.



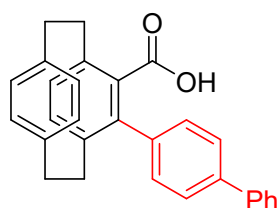
3j:

White solid (22.6 mg, 66%). The product **3j** was obtained by preparative thin-layer chromatography using dichloromethane/petroleum ether/triethylamine = 20:1:1 as the eluent. ^1H NMR (500 MHz, CDCl_3) δ 7.45 (s, 1H), 7.19 (m, 2H), 6.98 (s, 1H), 6.84 (dd, $J = 8.0, 1.0$ Hz, 1H), 6.65 – 6.56 (m, 3H), 6.47 (d, $J = 7.9$ Hz, 1H), 6.43 (d, $J = 7.8$ Hz, 1H), 3.31 – 3.22 (m, 2H), 3.07 (t, $J = 10.6$ Hz, 1H), 2.92 – 2.76 (m, 4H), 2.71 (ddd, $J = 13.2, 9.8, 6.0$ Hz, 1H), 2.37 (s, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 172.3, 138.8, 138.7, 138.7, 138.0, 137.7, 136.1, 136.0, 135.4, 133.2, 131.8, 131.8, 130.8, 130.5, 128.9, 128.5, 128.1, 34.2, 33.9, 33.3, 32.0, 28.7. HRMS (ESI) m/z : Calcd for $\text{C}_{24}\text{H}_{22}\text{O}_2$: $[\text{M}+\text{H}]^+$ 343.1693; Found: $[\text{M}+\text{H}]^+$ 343.1692.



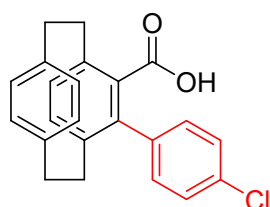
3k:

White solid (24.8 mg, 59%). The product **3k** was obtained by preparative thin-layer chromatography using dichloromethane/petroleum ether/triethylamine = 20:1:1 as the eluent. ^1H NMR (500 MHz, CDCl_3) δ 7.48 (s, 1H), 7.29 (m, 1H), 7.26 (m, 1H), 7.18 (s, 1H), 7.03 (p, $J = 11.4, 9.4$ Hz, 5H), 6.83 (dd, $J = 8.0, 1.8$ Hz, 1H), 6.69 – 6.57 (m, 3H), 6.51 – 6.39 (m, 2H), 3.33 – 3.21 (m, 2H), 3.08 (q, $J = 11.8, 11.1$ Hz, 1H), 2.94 – 2.79 (m, 4H), 2.74 (ddd, $J = 12.8, 9.4, 5.2$ Hz, 1H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 173.0, 155.9, 155.7, 138.7, 138.6, 138.3, 138.0, 137.7, 136.3, 136.1, 133.3, 133.2, 131.9, 130.9, 130.4, 128.8, 128.8, 128.5, 122.6, 118.4, 117.3, 34.2, 33.9, 33.3, 31.9. HRMS (ESI) m/z : Calcd for $\text{C}_{29}\text{H}_{24}\text{O}_3$: $[\text{M}+\text{K}]^+$ 459.1357; Found: $[\text{M}+\text{K}]^+$ 459.1352.



3l:

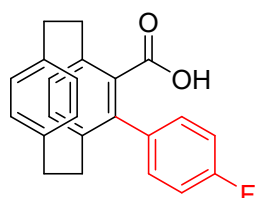
White solid (24.6 mg, 61%). The product **3l** was obtained by preparative thin-layer chromatography using dichloromethane/petroleum ether/triethylamine = 20:1:1 as the eluent. ^1H NMR (500 MHz, CDCl_3) δ 7.61 (d, $J = 7.8$ Hz, 5H), 7.41 (s, 1H), 7.38 (s, 1H), 7.30 (t, $J = 7.4$ Hz, 1H), 7.18 (s, 1H), 6.85 (d, $J = 7.5$ Hz, 1H), 6.66 – 6.55 (m, 3H), 6.48 (d, $J = 8.0$ Hz, 1H), 6.42 (d, $J = 7.8$ Hz, 1H), 3.23 (m, 2H), 3.02 (t, $J = 11.0$ Hz, 1H), 2.95 – 2.87 (m, 2H), 2.81 (m, 2H), 2.76 – 2.68 (m, 1H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 172.6, 139.6, 139.0, 138.7, 138.6, 138.5, 138.2, 137.7, 137.6, 137.4, 136.1, 133.4, 131.8, 130.9, 130.5, 128.9, 128.4, 127.8, 126.4, 126.0, 126.0, 34.2, 33.9, 33.3, 32.0. HRMS (ESI) m/z : Calcd for $\text{C}_{29}\text{H}_{24}\text{O}_2$: $[\text{M}+\text{K}]^+$ 443.1408; Found: $[\text{M}+\text{K}]^+$ 443.1407.



3m:

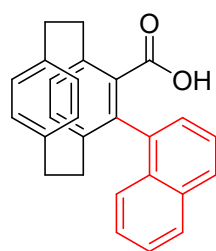
White solid (17.7 mg, 49%). The product **3m** was obtained by preparative thin-layer chromatography using dichloromethane/petroleum ether/triethylamine = 20:1:1 as the eluent. ^1H NMR (500 MHz, CDCl_3) δ 7.50 – 7.25 (m, 3H), 7.02 (s, 1H), 6.81 (dd, $J =$

8.0, 1.9 Hz, 1H), 6.68 – 6.56 (m, 3H), 6.48 (dd, $J = 8.0, 2$ Hz, 1H), 6.43 (d, $J = 7.5$ Hz, 1H), 3.32 – 3.18 (m, 2H), 3.14 – 3.04 (m, 1H), 2.94 – 2.83 (m, 2H), 2.82 – 2.69 (m, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 172.9, 138.7, 138.5, 138.2, 137.6, 137.5, 136.9, 136.1, 133.6, 132.6, 131.9, 131.0, 130.4, 129.8, 128.7, 128.5, 127.6, 34.2, 33.9, 33.3, 31.9. HRMS (ESI) m/z : Calcd for $\text{C}_{23}\text{H}_{19}\text{ClO}_2$: $[\text{M}+\text{H}]^+$ 363.1146; Found: $[\text{M}+\text{H}]^+$ 363.1141.



3n:

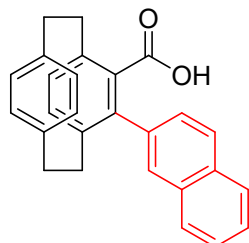
White solid (18.3 mg, 53%). The product **3n** was obtained by preparative thin-layer chromatography using dichloromethane/petroleum ether/triethylamine = 20:1:1 as the eluent. ^1H NMR (500 MHz, CDCl_3) δ 7.50 (s, 1H), 7.05 (m, 3H), 6.83 (dd, $J = 7.9, 1.8$ Hz, 1H), 6.68 – 6.56 (m, 3H), 6.46 – 6.40 (m, 2H), 3.31 – 3.20 (m, 2H), 3.14 – 3.05 (m, 1H), 2.95 – 2.85 (m, 2H), 2.83 – 2.69 (m, 3H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 172.3, 162.3, 160.3, 138.7, 138.6, 138.1, 137.7, 137.6, 136.1, 134.4, 134.4, 133.5, 131.9, 131.0, 130.4, 128.7, 128.6, 34.2, 33.9, 33.2, 31.9. HRMS (ESI) m/z : Calcd for $\text{C}_{23}\text{H}_{19}\text{FO}_2$: $[\text{M}+\text{K}]^+$ 385.1001; Found: $[\text{M}+\text{K}]^+$ 385.1005.



(R_p)/(S_p)-3o:

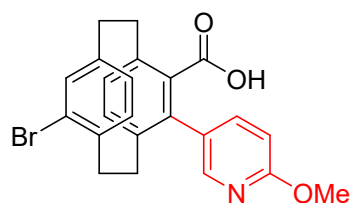
White solid (17.4 mg, 46%). The product **3o** was obtained by preparative thin-layer chromatography using dichloromethane/petroleum ether/triethylamine = 20:1:1 as the eluent. R_p $[\alpha]_D^{25} = 144$ ($c = 0.1$, CH_2Cl_2), S_p $[\alpha]_D^{25} = -124$ ($c = 0.1$, CH_2Cl_2). ^1H NMR (500 MHz, CDCl_3) δ 7.82 (t, $J = 7.2$ Hz, 2H), 7.71 (d, $J = 7.2$ Hz, 1H), 7.58 (d, $J = 8.4$ Hz, 1H), 7.55 – 7.48 (m, 1H), 7.44 (t, $J = 7.5$ Hz, 1H), 7.32 (t, $J = 7.6$ Hz, 1H), 6.96 (d, $J = 8.2$ Hz, 1H), 6.76 (d, $J = 7.8$ Hz, 1H), 6.69 (t, $J = 7.2$ Hz, 3H), 6.57 (d, $J = 7.6$ Hz, 1H), 3.40 – 3.25 (m, 2H), 3.13 (t, $J = 11.0$ Hz, 1H), 2.96 – 2.89 (m, 1H), 2.83 (m, 2H),

2.73 (m, 1H), 2.31 (ddd, $J = 13.6, 9.7, 5.6$ Hz, 1H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 140.5, 140.0, 139.9, 135.8, 134.9, 134.8, 133.5, 132.9, 131.8, 131.5, 129.6, 128.1, 128.0, 126.9, 126.8, 126.2, 125.9, 125.8, 125.7, 125.3, 116.7, 113.9, 34.3, 33.0, 30.9, 29.7. HRMS (ESI) m/z : Calcd for $\text{C}_{27}\text{H}_{22}\text{O}_2$: $[\text{M}+\text{H}]^+$ 379.1693; Found: $[\text{M}+\text{H}]^+$ 379.1690.



(R_p)/(S_p)-3p**:**

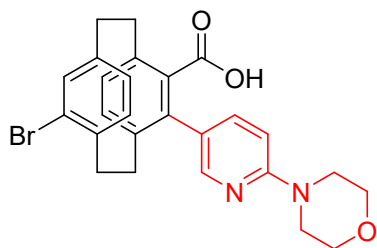
White solid (18.5 mg, 49%). The product **3p** was obtained by preparative thin-layer chromatography using dichloromethane/petroleum ether/triethylamine = 20:1:1 as the eluent. R_p $[\alpha]_D^{25} = 140$ ($c = 0.1$, CH_2Cl_2), S_p $[\alpha]_D^{25} = -125$ ($c = 0.1$, CH_2Cl_2). ^1H NMR (500 MHz, CDCl_3) δ 8.03– 7.72 (m, 3H), 7.70– 7.18 (m, 4H), 6.89 (m, 1H), 6.64 (dd, $J = 7.6, 2.0$ Hz, 2H), 6.61– 6.38 (m, 3H), 3.20 (m, 2H), 3.04 (t, $J = 10.3$ Hz, 1H), 2.96 – 2.88 (m, 1H), 2.75 (ddd, $J = 45.5, 12.9, 7.4$ Hz, 4H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 172.9, 138.8, 138.6, 138.4, 138.0, 137.7, 136.2, 136.1, 136.0, 133.4, 132.4, 131.9, 131.5, 130.9, 130.5, 129.3, 128.7, 128.7, 127.3, 126.9, 126.7, 125.7, 125.2, 34.2, 33.9, 33.3, 32.1. HRMS (ESI) m/z : Calcd for $\text{C}_{27}\text{H}_{22}\text{O}_2$: $[\text{M}+\text{H}]^+$ 379.1693; Found: $[\text{M}+\text{H}]^+$ 379.1699.



3q:

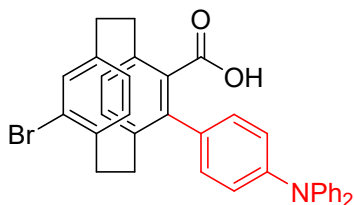
White solid (26.2 mg, 60%). The product **3q** was obtained by flash column chromatography on silica gel using petroleum ether/tetrahydrofuran = 3:1 as the eluent. ^1H NMR (500 MHz, $\text{THF-}d_8$) δ 8.24 – 7.20 (m, 2H), 7.21 (d, $J = 7.8$ Hz, 1H), 6.95 (dd, $J = 7.8, 1.6$ Hz, 1H), 6.68 (m, 2H), 6.42 (m, 2H), 3.82 (s, 3H), 3.25 – 3.19 (m, 1H), 3.16 (td, $J = 11.2, 10.6, 3.9$ Hz, 1H), 3.07 (ddd, $J = 13.2, 9.8, 3.2$ Hz, 1H), 2.98 – 2.92 (m, 2H), 2.90 (m, 1H), 2.82 (m, 2H). $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, $\text{THF-}d_8$) δ 170.2, 163.2,

142.2, 138.5, 138.2, 138.2, 138.1, 138.0, 136.4, 136.0, 133.2, 132.7, 131.7, 131.3, 130.2, 128.4, 126.4, 109.9, 52.5, 34.0, 33.8, 33.2, 31.6. HRMS (ESI) m/z : Calcd for $C_{23}H_{20}BrNO_3$: $[M+H]^+$ 438.0699; Found: $[M+H]^+$ 438.0703.



3r:

White solid (28.5 mg, 58%). The product **3r** was obtained by flash column chromatography on silica gel using petroleum ether/tetrahydrofuran = 3:1 as the eluent. 1H NMR (500 MHz, THF- d_8) δ 10.86 (s, 1H), 7.01 (dd, J = 7.9, 1.8 Hz, 1H), 6.81 (d, J = 8.8 Hz, 1H), 6.70 (dd, J = 7.8, 2.9 Hz, 2H), 6.65 (dd, J = 7.8, 1.8 Hz, 1H), 6.49 (m, 2H), 5.79 (s, 1H), 3.78 (t, J = 4.8 Hz, 4H), 3.57 (m, 4H), 3.37–3.28 (m, 2H), 3.11 (m, 1H), 3.02 – 2.95 (m, 2H), 2.92 – 2.87 (m, 1H), 2.82 (dd, J = 10.4, 5.7 Hz, 2H). $^{13}C\{^1H\}$ NMR (126 MHz, THF- d_8) δ 170.2, 158.2, 139.5, 139.3, 138.3, 138.2, 136.0, 135.9, 133.8, 132.6, 131.8, 131.7, 131.7, 131.6, 131.3, 129.5, 125.3, 105.4, 66.5, 45.4, 34.9, 34.6, 33.9, 32.8. HRMS (ESI) m/z : Calcd for $C_{26}H_{25}BrN_2O_3$: $[M+H]^+$ 493.1121; Found: $[M+H]^+$ 493.1118.

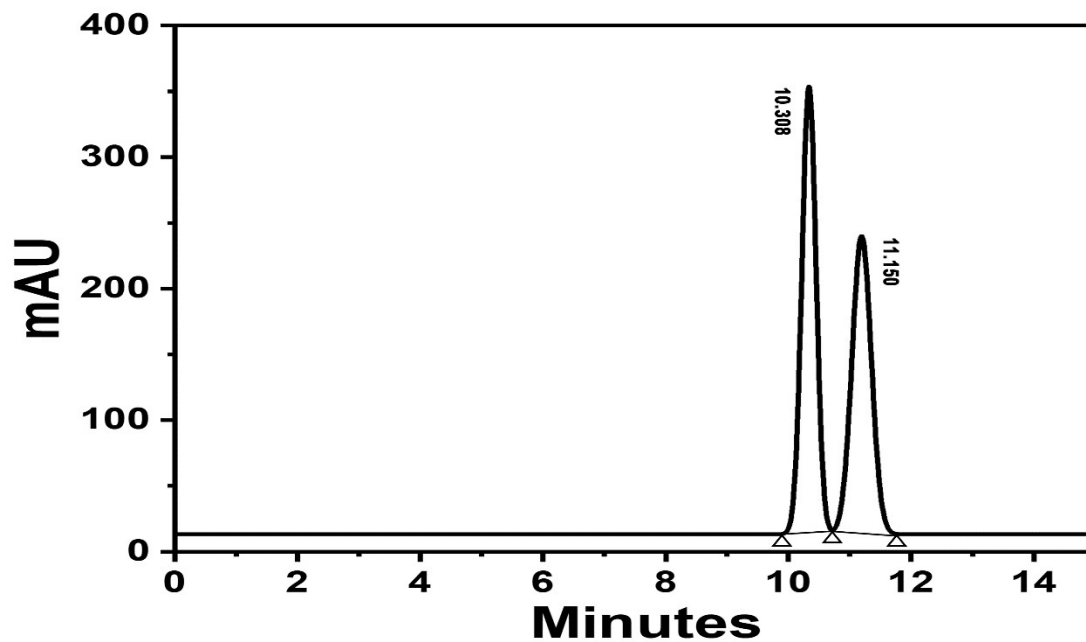


3s:

White solid (35.5 mg, 62%). The product **3s** was obtained by flash column chromatography on silica gel using petroleum ether/tetrahydrofuran = 3:1 as the eluent. 1H NMR (500 MHz, THF- d_8) δ 10.74 (s, 1H), 7.38 (s, 1H), 7.19 – 7.09 (m, 6H), 7.04 – 7.01 (m, 4H), 6.97 (d, J = 7.9 Hz, 2H), 6.94 – 6.88 (m, 3H), 6.67 (d, J = 1.7 Hz, 1H), 6.51 – 6.44 (m, 1H), 6.40 – 6.32 (m, 1H), 3.20 (td, J = 11.7, 11.0, 1.9 Hz, 1H), 3.17 – 3.07 (m, 2H), 3.06 – 2.96 (m, 2H), 2.95 – 2.91 (m, 1H), 2.90 – 2.86 (m, 1H), 2.80 (ddd, J = 11.9, 9.3, 4.0 Hz, 1H). $^{13}C\{^1H\}$ NMR (126 MHz, THF- d_8) δ 170.4, 147.8, 146.8, 142.1, 139.2, 138.6, 137.9, 137.8, 136.8, 136.2, 136.0, 134.8, 134.1, 133.6, 132.7, 132.2, 131.6, 131.4, 130.2, 130.1, 129.1, 126.4, 124.5, 122.9, 122.4, 34.0, 33.9, 33.2,

31.9. HRMS (ESI) m/z: Calcd for C₃₅H₂₈BrNO₂: [M+K]⁺ 612.0935; Found: [M+K]⁺ 612.0931.

5. HPLC Chromatograms

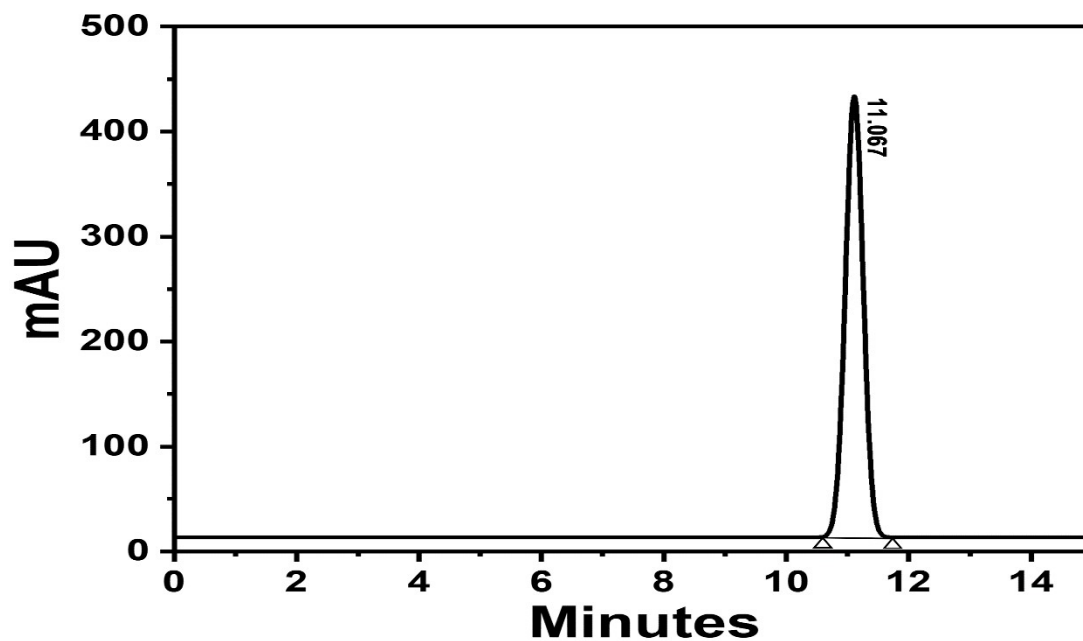


<Peak Table>

PDA Ch1 270 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	10.318	6218.754	358.413	51.017
2	11.150	5970.859	251.858	48.983
Total		12189.614	610.271	100.000

Figure S1. Chromatogram of the racemic of **1a**.

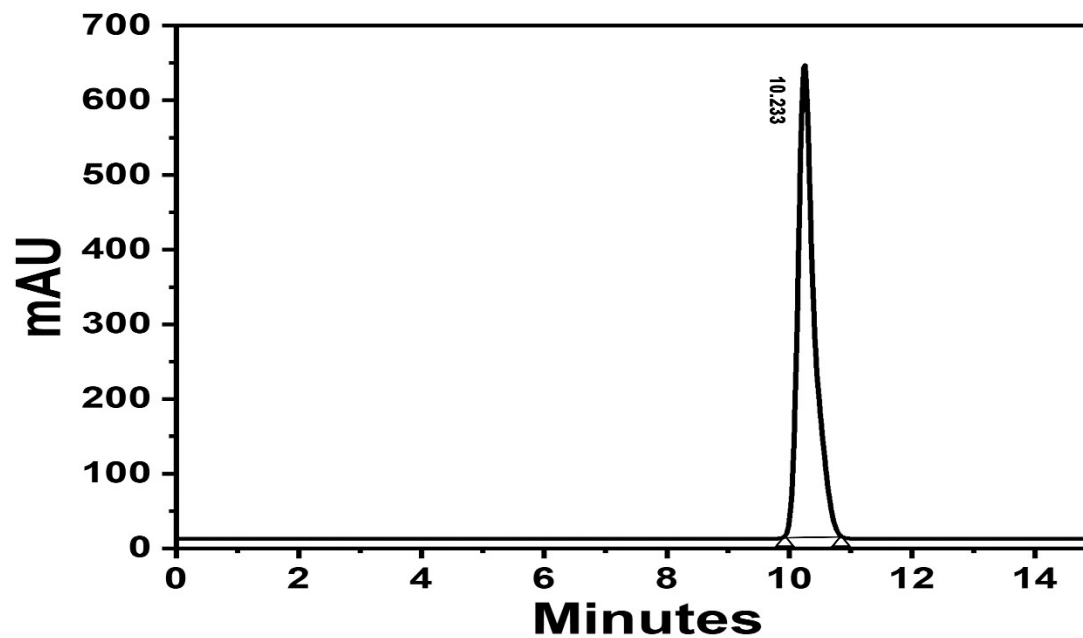


<Peak Table>

PDA Ch1 270 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	11.067	9714.872	445.415	100.000
Total		9714.872	445.415	100.000

Figure S2. Chromatogram of (*R_p*)-1a.

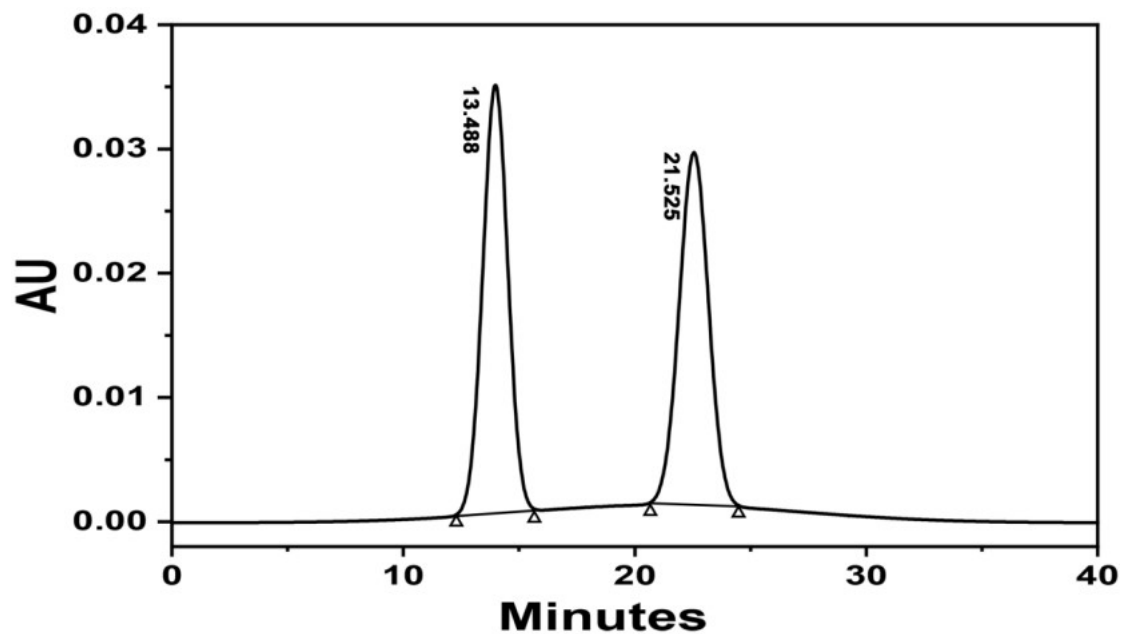


<Peak Table>

PDA Ch1 270 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	10.233	11455.167	626.811	100.000
Total		11455.167	626.811	100.000

Figure S3. Chromatogram of (*S_p*)-**1a**.

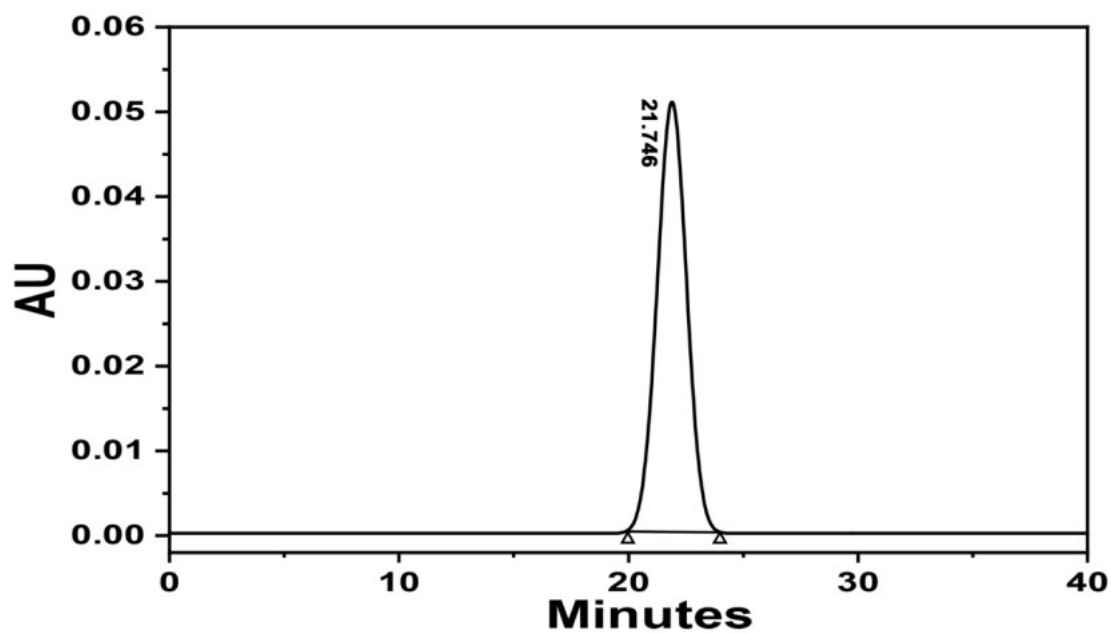


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	13.488	2973678	38090	49.90
2	21.525	2985988	32139	50.10
Total		5959666	70229	100.00

Figure S4. Chromatogram of the racemic of **3a**.

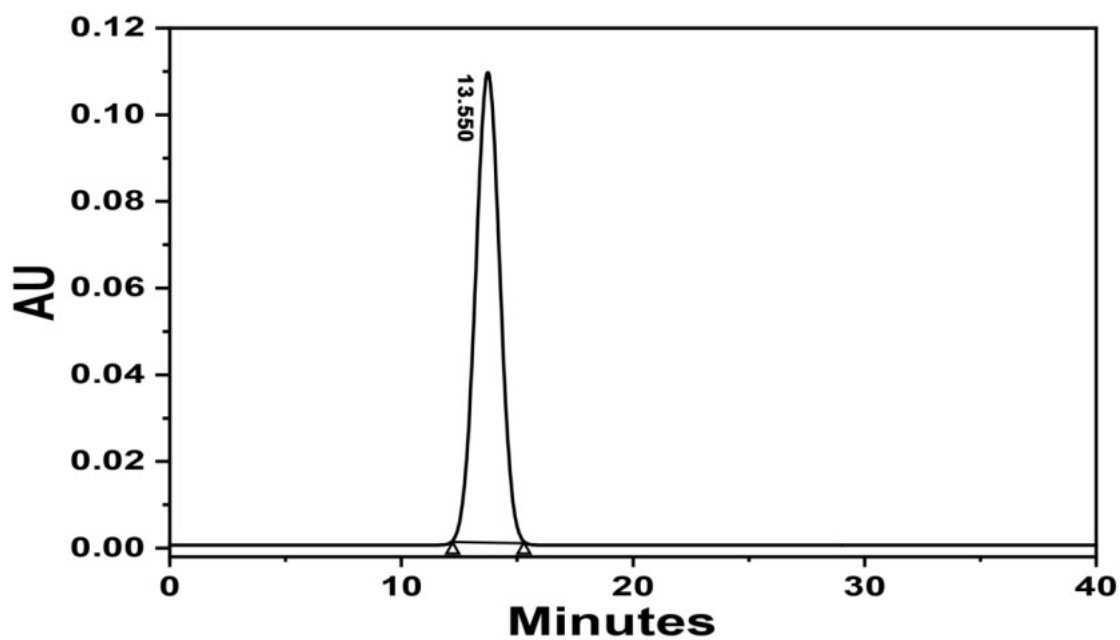


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	21.746	3890908	41194	100.00
Total		3890908	41194	100.00

Figure S5. Chromatogram of (*R_p*)-3a.



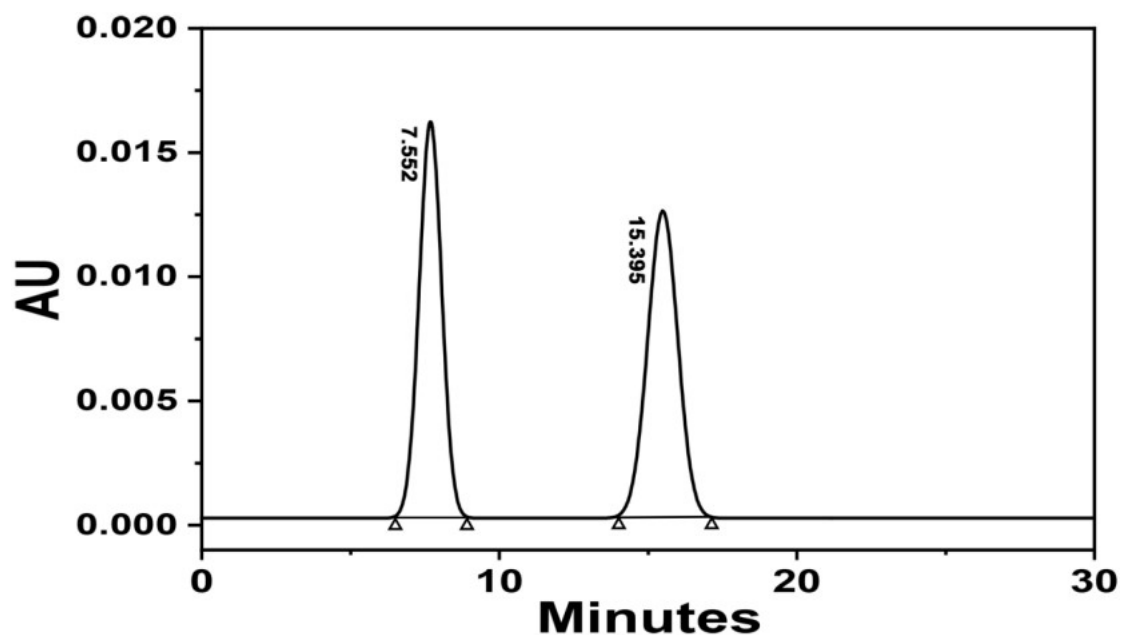
<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	13.550			

1	13.550	9389548	118039	100.00
Total		9389548	118039	100.00

Figure S6. Chromatogram of (*S_p*)-**3a**.

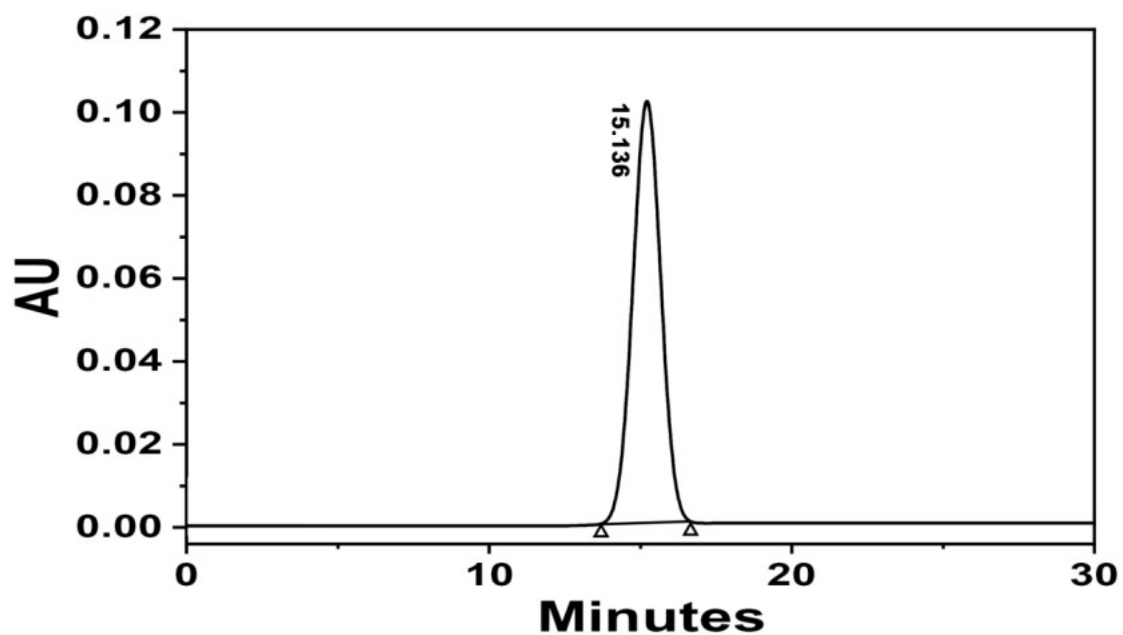


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	7.552	1052560	17116	50.57
2	15.395	1028980	13161	49.43
Total		2081540	30277	100.00

Figure S7. Chromatogram of the racemic of **3c**.

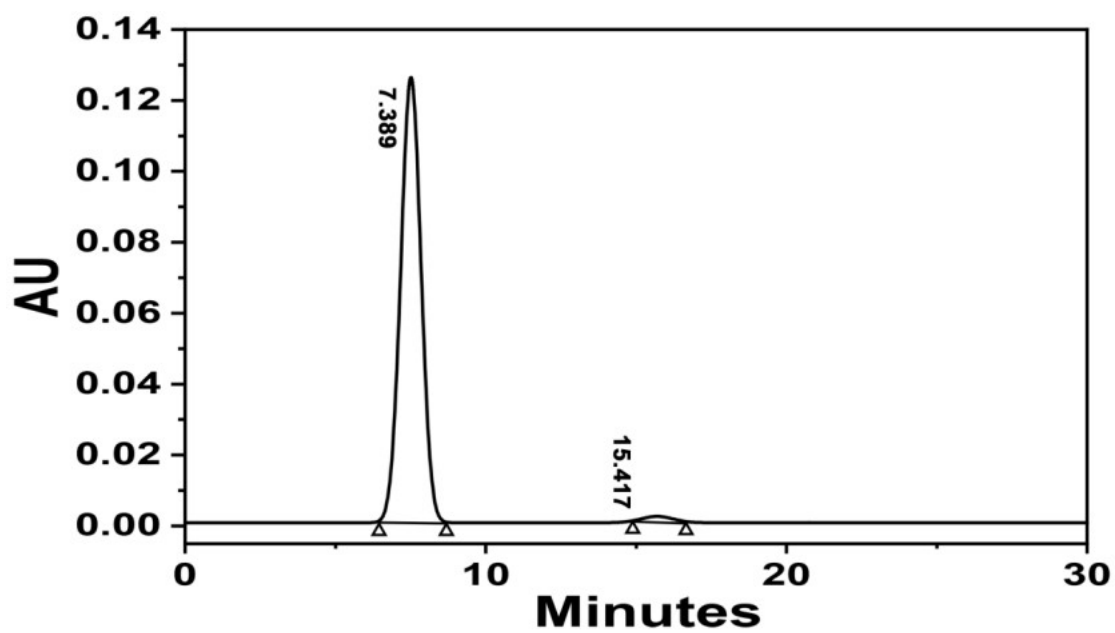


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	15.136	7569353	108559	100.00
Total		7569353	108559	100.00

Figure S8. Chromatogram of (*R_p*)-3c.

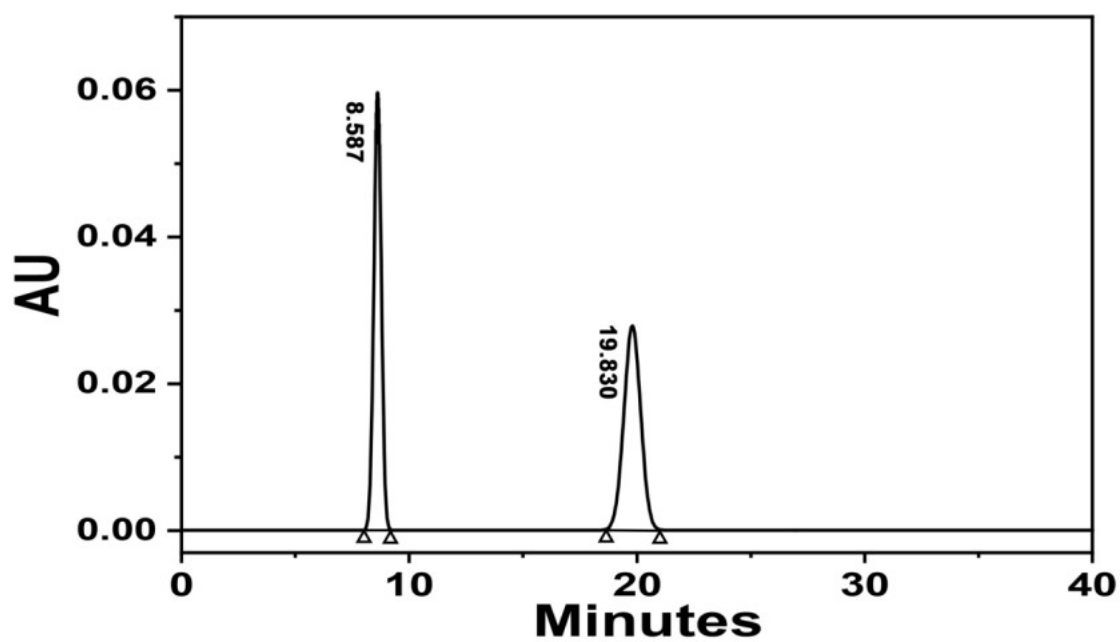


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	7.389	7283196	134442	99.71
2	15.417	51136	1349	0.29
Total		7334332	135791	100.00

Figure S9. Chromatogram of (*S_p*)-3c.

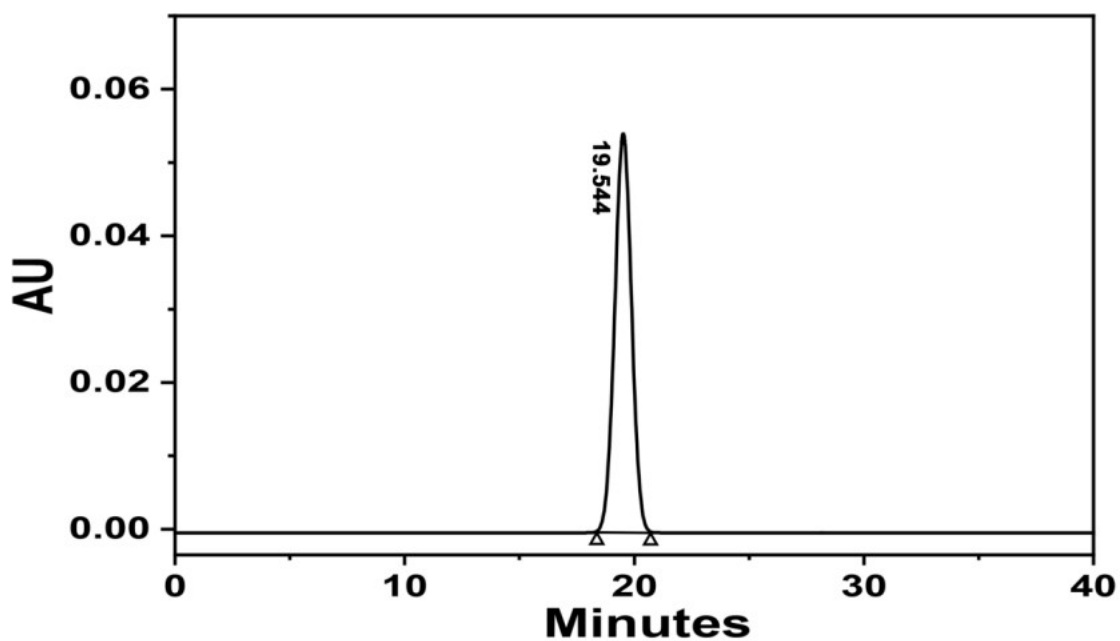


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	8.587	1535869	62766	50.07
2	19.830	1531705	29745	49.93
Total		3067574	92511	100.00

Figure S10. Chromatogram of the racemic of **3h**.

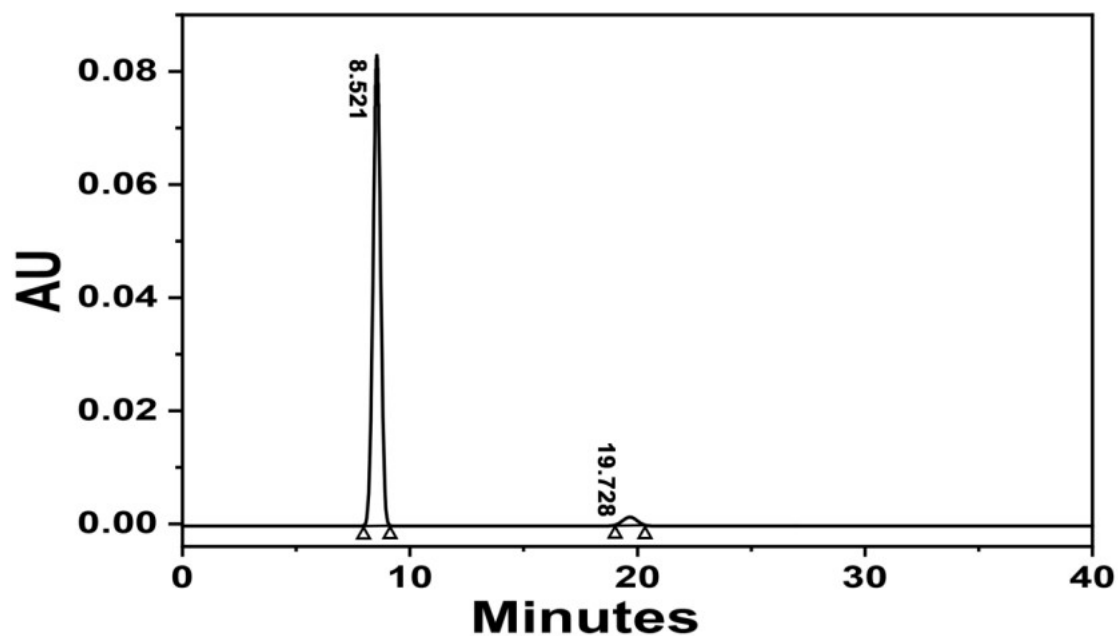


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	19.544	2922142	58309	100.00
Total		2922142	58309	100.00

Figure S11. Chromatogram of (R_p)-3h.

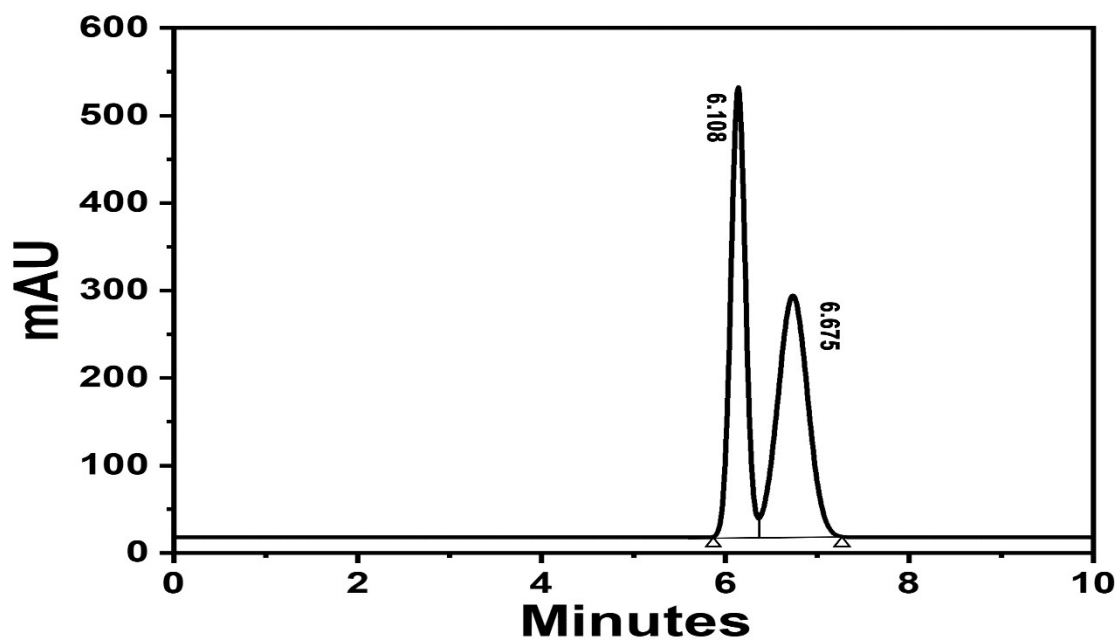


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	8.521	2122028	87848	99.68
2	19.728	16741	741	0.32
Total		2138769	88589	100.00

Figure S12. Chromatogram of (S_p)-3h.

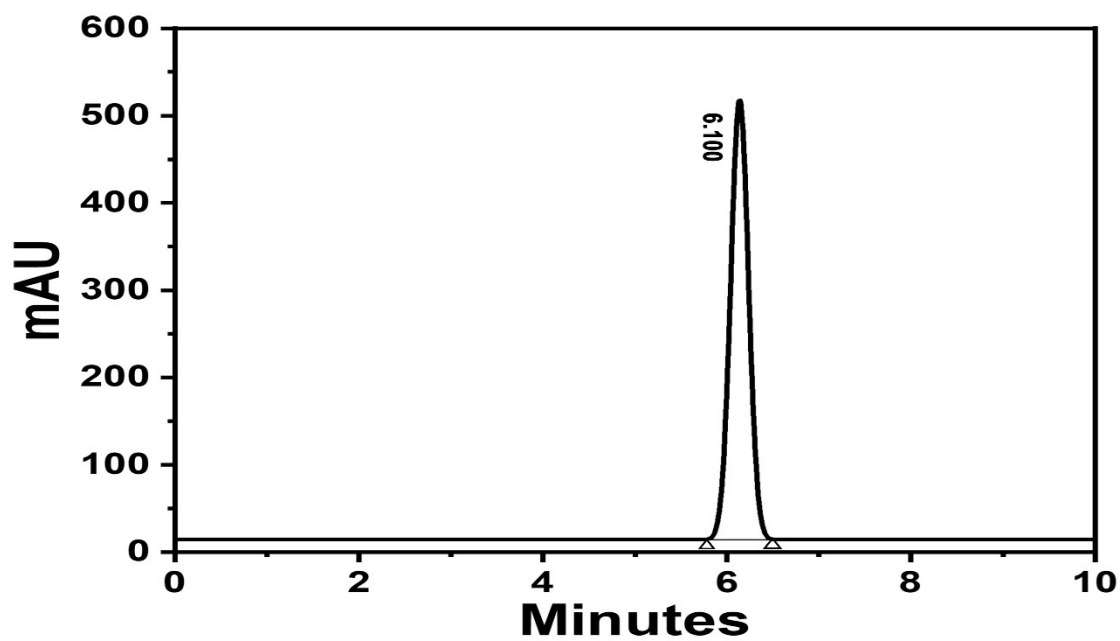


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	6.108	6651.650	549.790	49.697
2	6.675	6732.852	292.428	50.303
Total		13384.502	842.218	100.000

Figure S13. Chromatogram of the racemic of **30**.

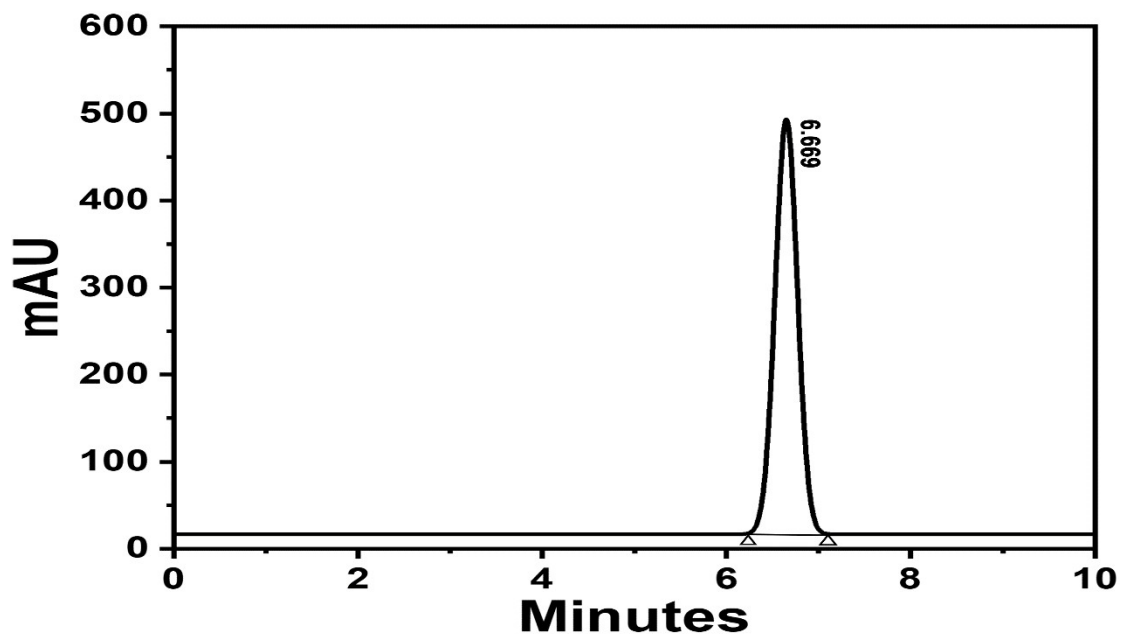


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	6.100	8919.920	525.104	100.000
Total		8919.920	555.104	100.000

Figure S14. Chromatogram of (*R_p*)-**3o**.

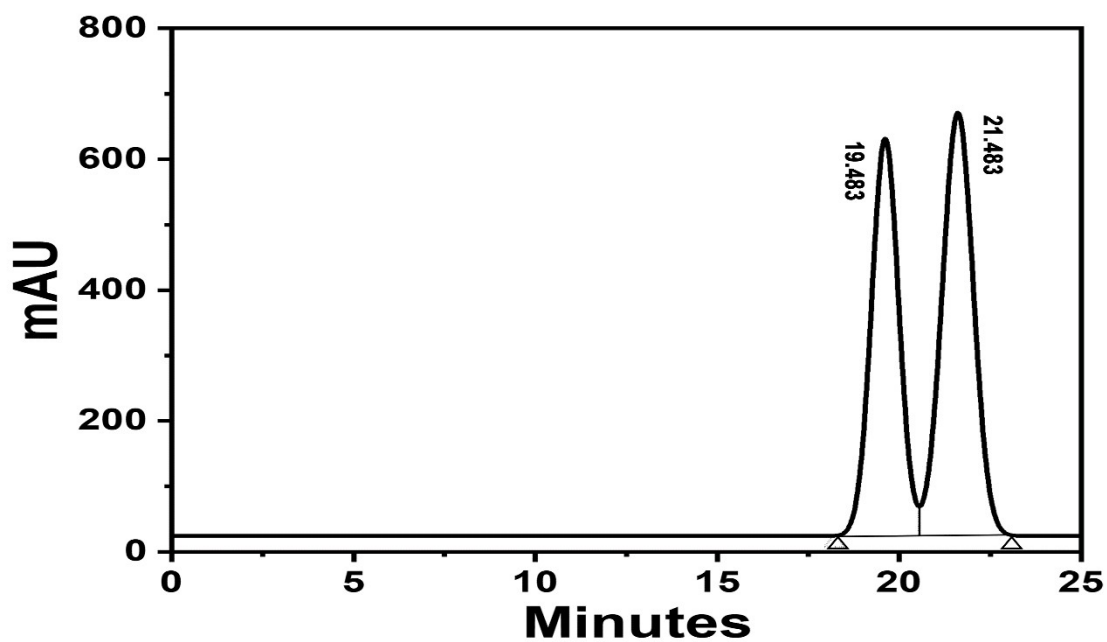


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	6.669	9557.340	495.047	100.000
Total		9557.340	495.047	100.000

Figure S15. Chromatogram of (*S_p*)-**3o**.

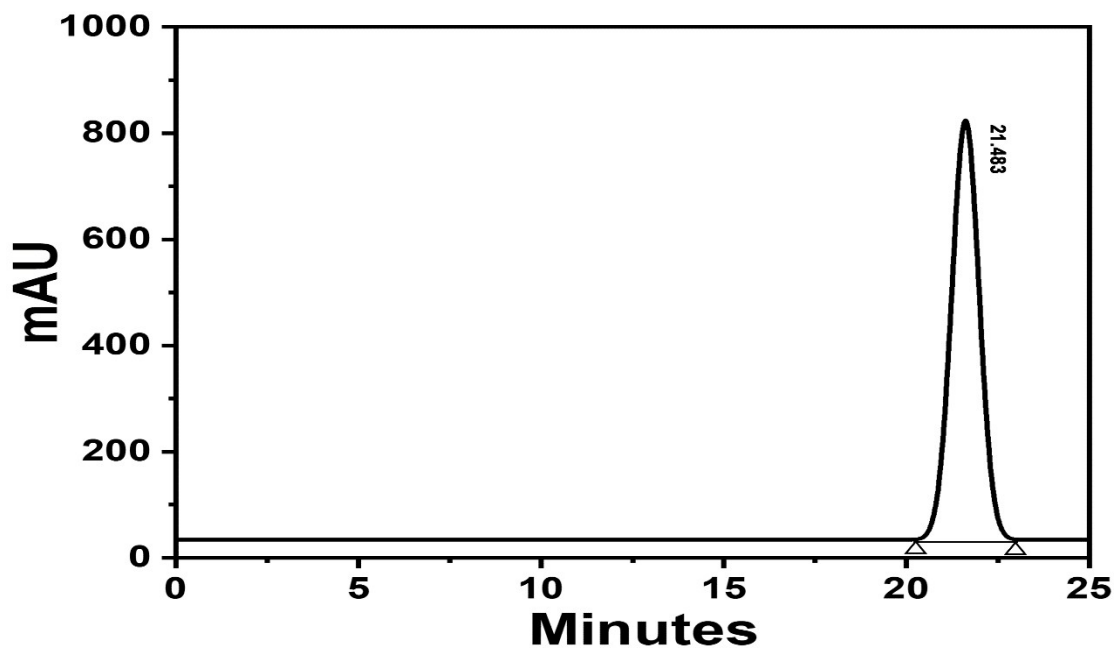


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	19.483	38157.742	652.737	48.942
2	21.483	39806.692	699.812	51.058
Total		77964.434	1352.549	100.000

Figure S16. Chromatogram of the racemic of 3p.

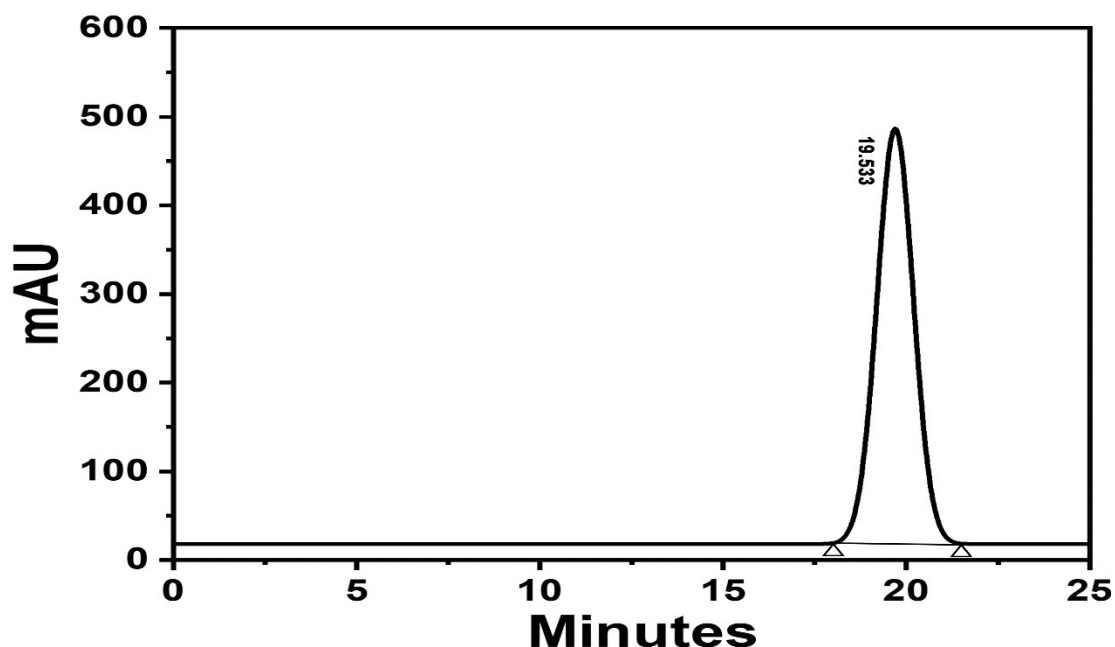


<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	21.483	52398.164	848.223	100.000
Total		52398.164	848.223	100.000

Figure S17. Chromatogram of (*R_p*)-**3p**.



<Peak Table>

PDA Ch1 257 nm

Peak Name	Retention Time	Area	Peak Height	Area %
1	19.533	27820.675	464.756	100.000
Total		27820.675	464.756	100.000

Figure S18. Chromatogram of (*S_p*)-**3p**.

6. General Procedure for Crystal Preparation and Measurement.

The single crystal of compound **3a** was grown by slow evaporation of solvent at room temperature. Intensity data of **3a** was collected on a Rigaku Oxford Diffraction Synergy Custom DW system X-ray diffractometer with a Hypix detector using Cu- $K\alpha$ radiation ($\lambda=1.54184$ Å) at 140 K; The structure was solved by direct methods and refined by full-matrix least-squares methods with SHELX-2018 program. Displacement parameters were refined anisotropically, and the positions of the H-atoms were generated geometrically, assigned isotropic thermal parameters, and allowed to ride on their parent carbon atoms before the final cycle

of refinement. Basic information pertaining to crystal parameters and structure refinement are summarized in Table S2, and hydrogen bonds are listed in Table S3. CCDC 2281238 contains the supplementary crystallographic data for this paper.

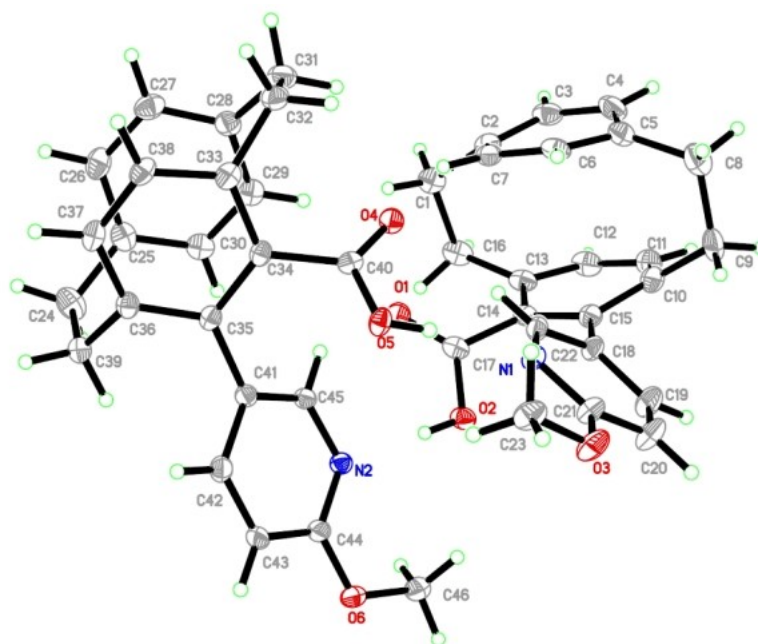


Figure S19. Single crystals of compound **3a** (Ellipsoids are drawn to 30% probability).

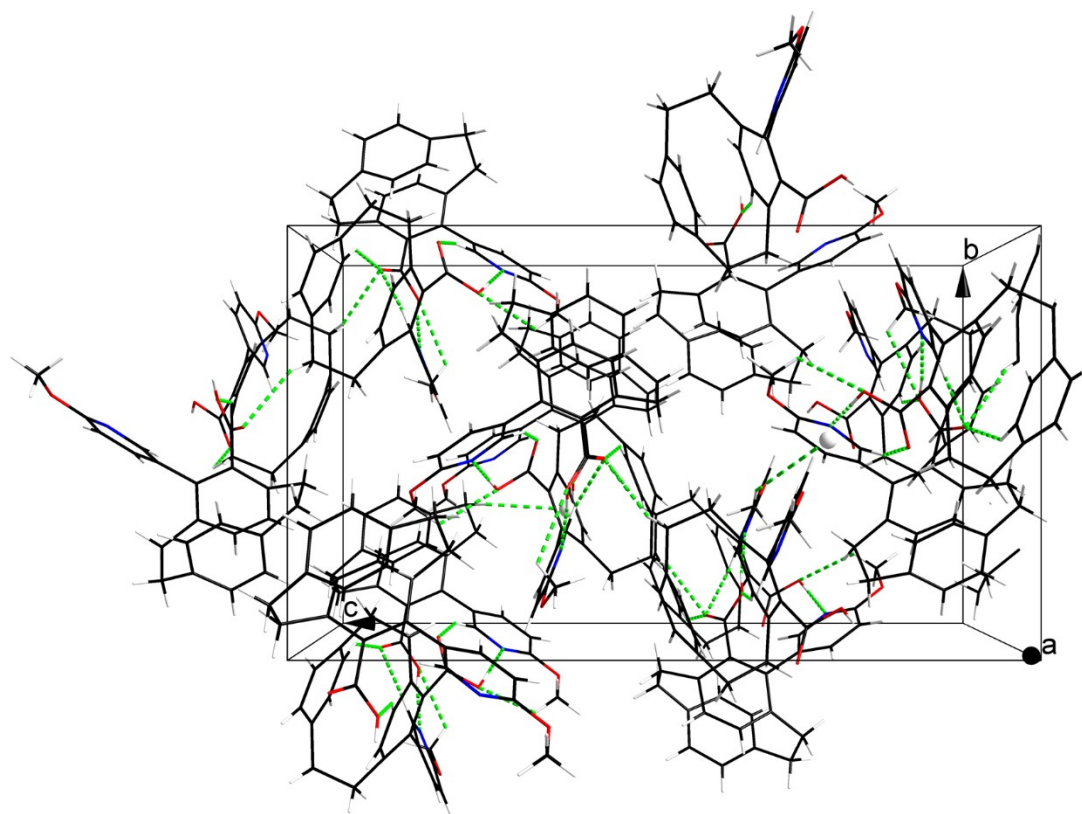


Figure S20. Molecular packing structure of **3a** along *a* axis. The green dotted lines show weak intermolecular interactions.

Table S2. Crystallographic data and structure refinement for **3a**.

Compound	3a
Empirical formula	C ₄₆ H ₄₂ N ₂ O ₆
CCDC number	2281238
Formula weight	718.81
Temperature	140(2) K
Crystal system	Orthorhombic
space group	P2(1)2(1)2(1)
Unit cell dimensions	a = 11.3401 (5) Å; $\alpha = 90^\circ$ b = 13.7292 (5) Å; $\beta = 90^\circ$ c = 23.8310 (10) Å; $\gamma = 90^\circ$
Volume	3710.3(3) Å ³
Z	4
Cal. Density	1.287 g/cm ³
Index ranges	-9 ≤ h ≤ 13 -13 ≤ k ≤ 16 -28 ≤ l ≤ 28
F (000)	1520
Crystal size	0.900 x 0.080 x 0.080 mm ³
GOF	1.070
R indices	R ₁ = 0.0575, wR ₂ = 0.1127

Table S3. Hydrogen bonds for **3a** [Å and °]

D–H⋯A	D(H⋯A)	∠(DHA)
-------	--------	--------

O5–H5⋯N1	1.8845 (30) Å	174.120 (186)°
O2–H2⋯N2	1.9123 (33) Å	172.715 (186)°
C45–H45⋯O1	2.3660 (23) Å	137.766 (212)°
C46–H46C⋯O2	2.7078 (24) Å	114.020 (259)°
C7–H7⋯O4	2.6670 (23) Å	139.913 (212)°
C8–H8B⋯O4	2.4867 (24) Å	143.615 (306)°
C22–H22⋯O4	2.3787 (24) Å	137.440 (243)°
C12–H12⋯O5	2.4922 (23) Å	157.653 (214)°
C23–H23C⋯O5	2.6865 (25) Å	116.534 (245)°
C23–H23B⋯C42 (π)	2.8842 (38) Å	-
C24–H24A⋯C13 (π)	2.8836 (39) Å	-

7. Photophysical Properties.

Theoretical calculations

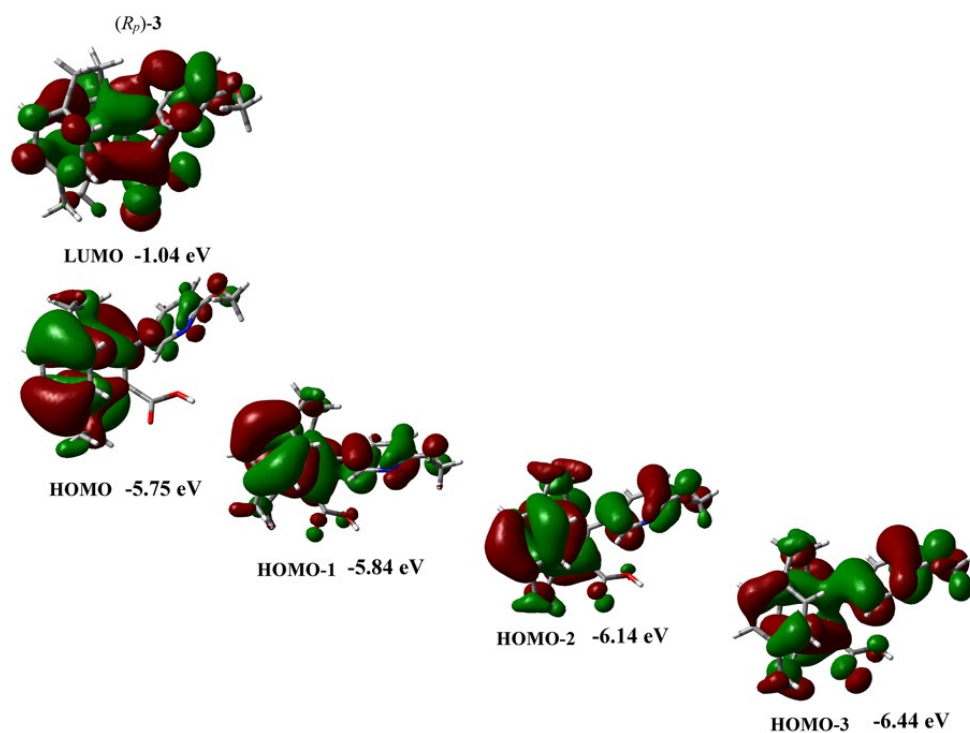


Figure S21. The angular nodal patterns of the LUMO, HOMO, HOMO-1, HOMO-2, and HOMO-3 of **3a**, calculated by using B3LYP/6-31G(d) basis set with the G09 program package.

Table S4. Related wave functions, oscillator strengths, and calculated electronic excitation energies of **3a**.

Compound	State ^{[a] [b]}	λ [nm]	$f^{[c]}$	E_g (eV)	Orbital (coefficient) ^[d]
(<i>R_p</i>)- 3a	S ₁	313.71	0.0214	3.95	H-L (67.8 %)

S_2	301.17	0.0425	4.12	H_1-L (67.3 %)
S_3	279.98	0.0170	4.43	H_2-L (62.9 %)
S_6	268.59	0.1482	4.62	H_3-L (38.7 %)

[^a]Only selected excited states were considered; [^b]DCM was employed as the solvent for the DFT calculations; [^c]Oscillator strength; [^d]MOs involved in the transitions. H = HOMO, L = LUMO.

Coefficient of the wavefunction for each excitation.

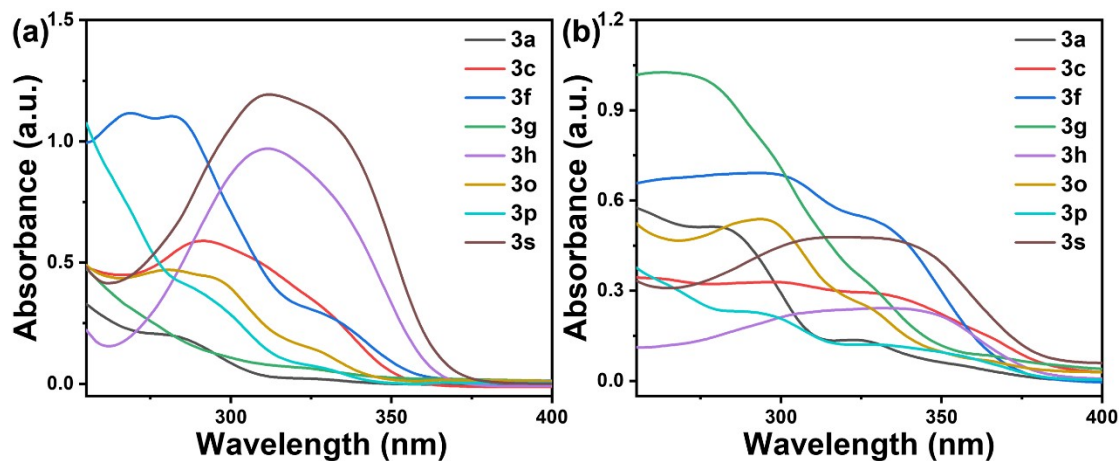


Figure S22. The absorption spectra of **3a**, **3c**, **3f**, **3g**, **3h**, **3o**, **3p**, and **3s** in (a) THF solutions ($50 \mu\text{M}$) and (b) PMMA films.

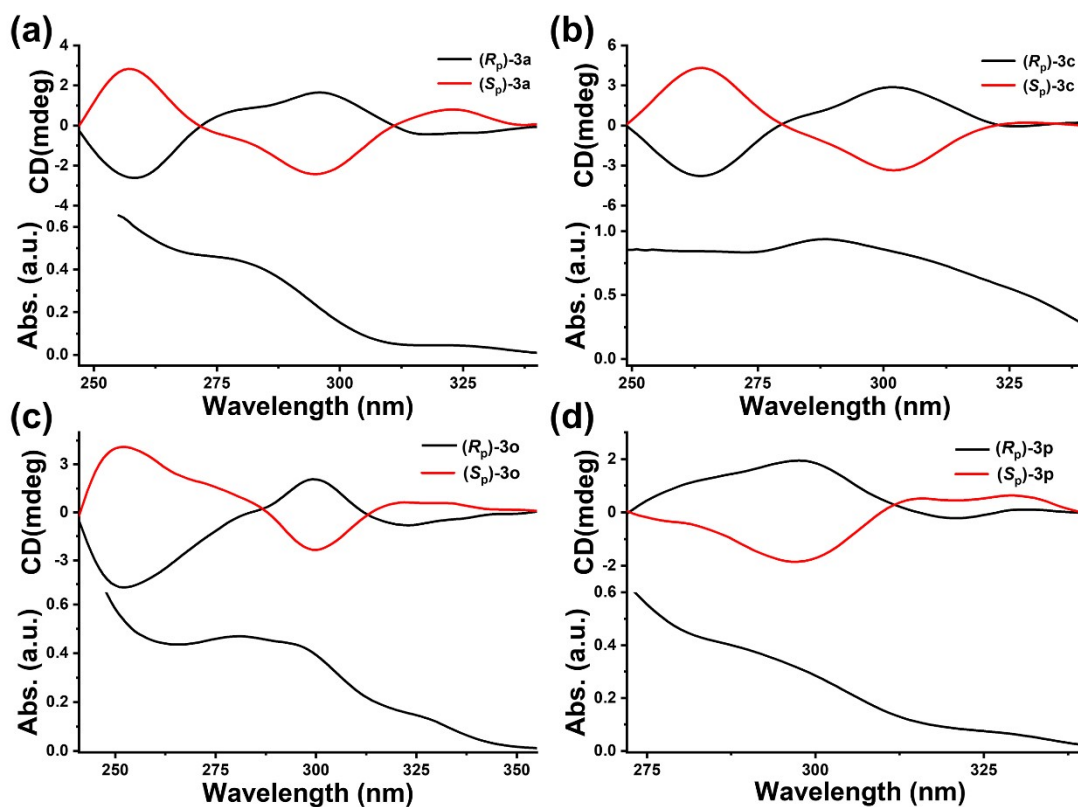


Figure S23. The CD and absorption spectra of (a) $(R_p)/(S_p)$ -**3a**, (b) $(R_p)/(S_p)$ -**3c**, (c)

(*R*_p)/(*S*_p)-**3o**, and (d) (*R*_p)/(*S*_p)-**3p** in THF solutions (50 μM).

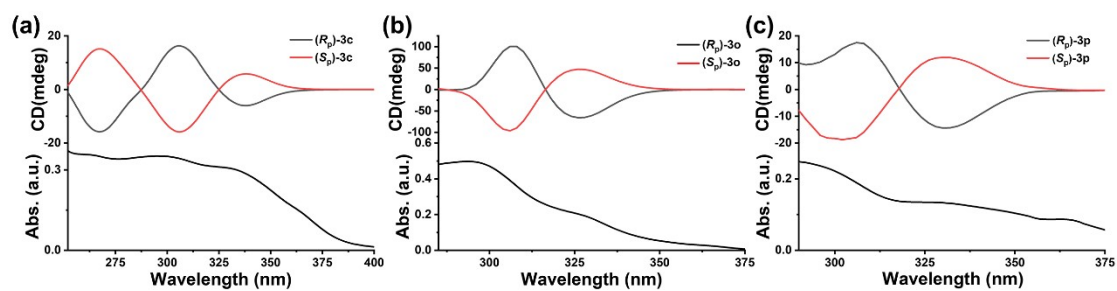
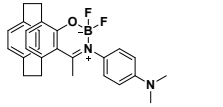
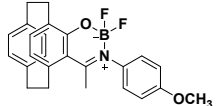
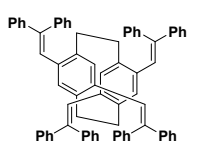
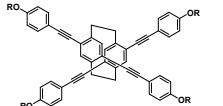


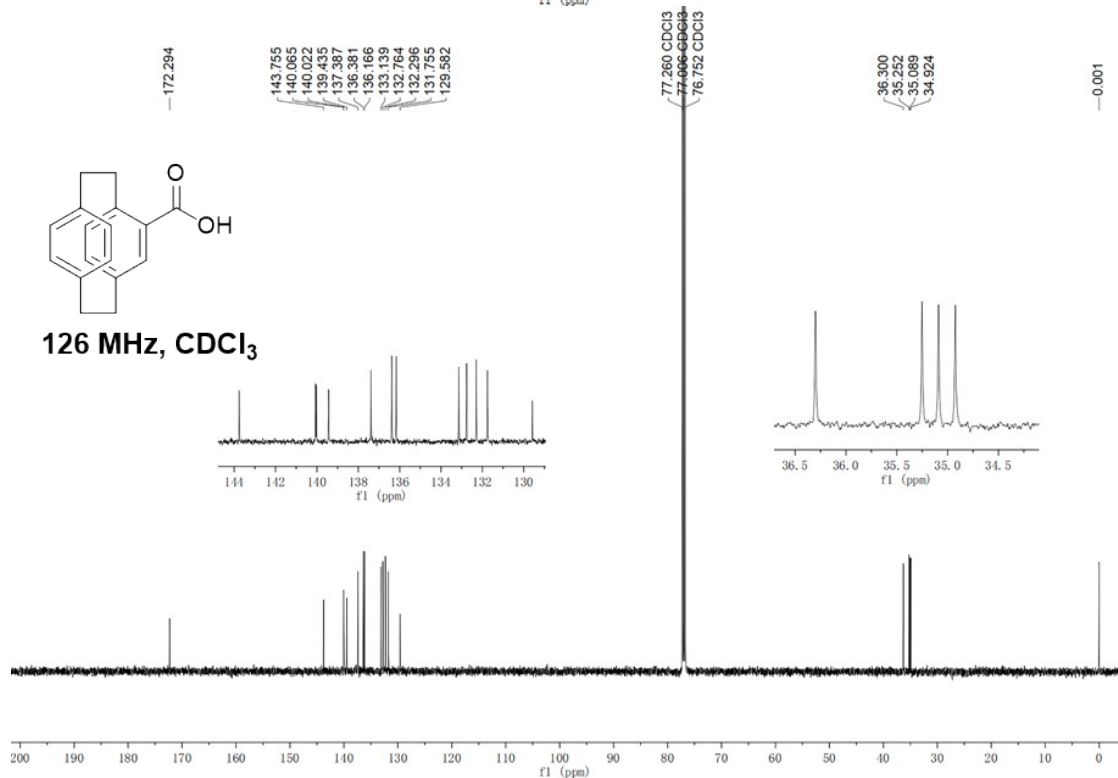
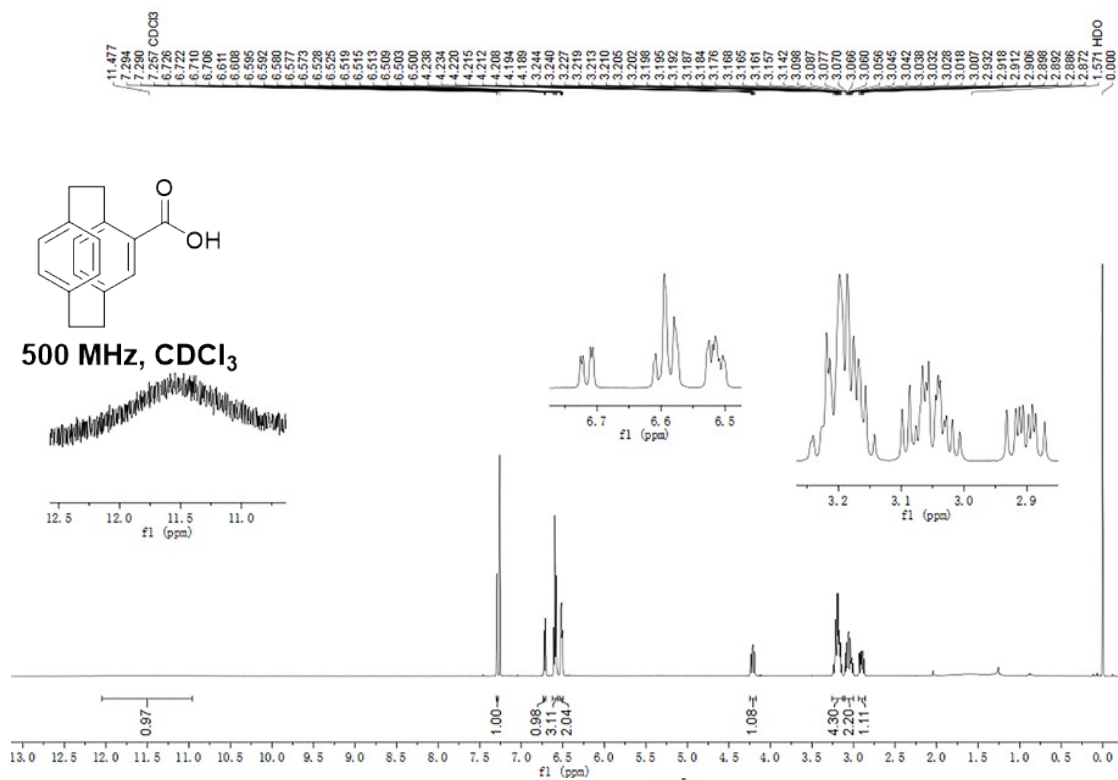
Figure S24. The CD and absorption spectra of (a) (*R*_p)/(*S*_p)-**3c**, (b) (*R*_p)/(*S*_p)-**3o**, and (c) (*R*_p)/(*S*_p)-**3p** in PMMA films.

Table S5. Comparison of CPL performances of [2.2]paracyclophane derivatives.

Structure	State	CPL emission wavelength (nm)	Φ_f (%)	$ g_{lum} $	Ref.
	thin film	490	8	1.7×10^{-3}	This work
	solid	555	9	3.9×10^{-4}	Ref.6
	solid	662	97	1.9×10^{-3}	Ref.7
	doped in PMMA	393	75	4.8×10^{-3}	Ref.8
	doped in PMMA	395	70	2.9×10^{-3}	Ref.8
	doped in PMMA	404	70	5×10^{-4}	Ref.8
	thin film	548	47	2.7×10^{-3}	Ref.9

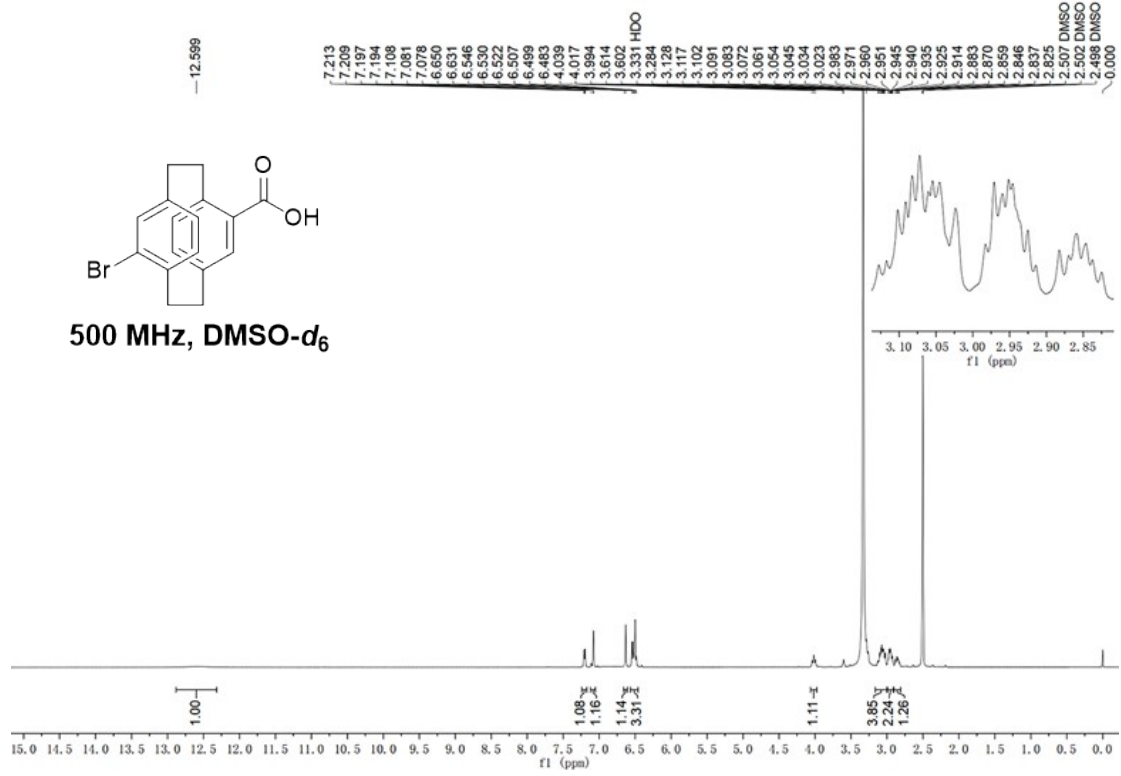
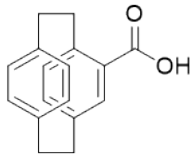
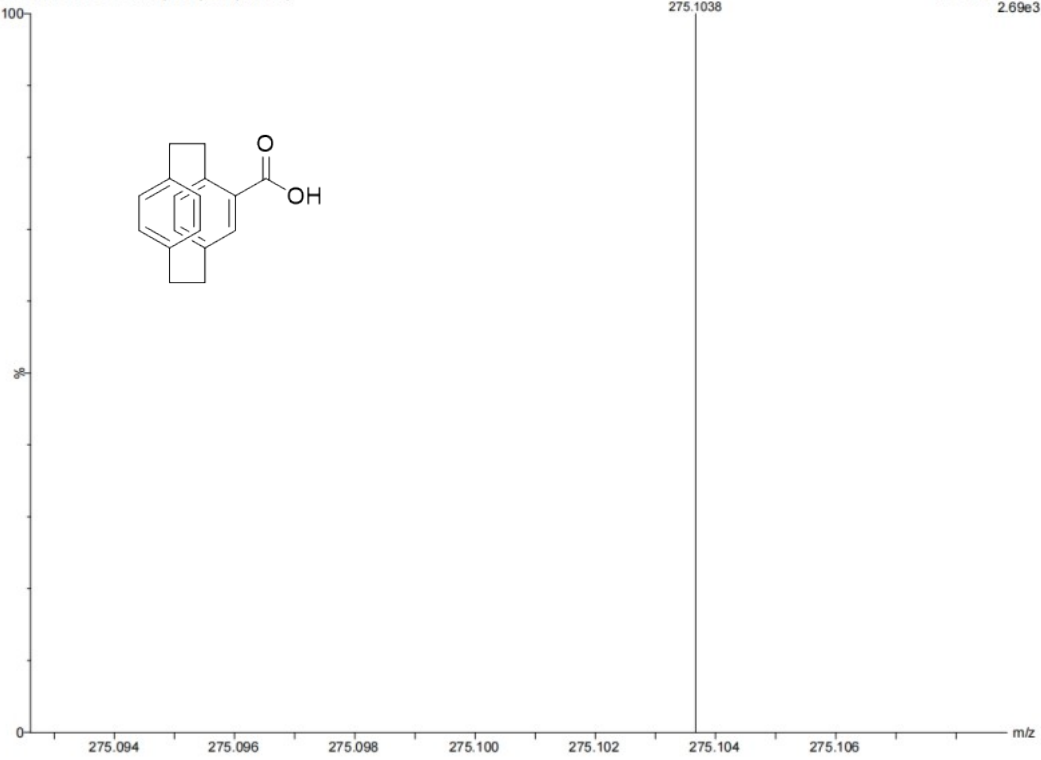
	solid	562	5	3.0×10^{-4}	Ref.10
	solid	488	8	1.8×10^{-3}	Ref.10
	in aggregation state	494	58	9.0×10^{-4}	Ref.11
	thin film	415	63	2.1×10^{-3}	Ref.12

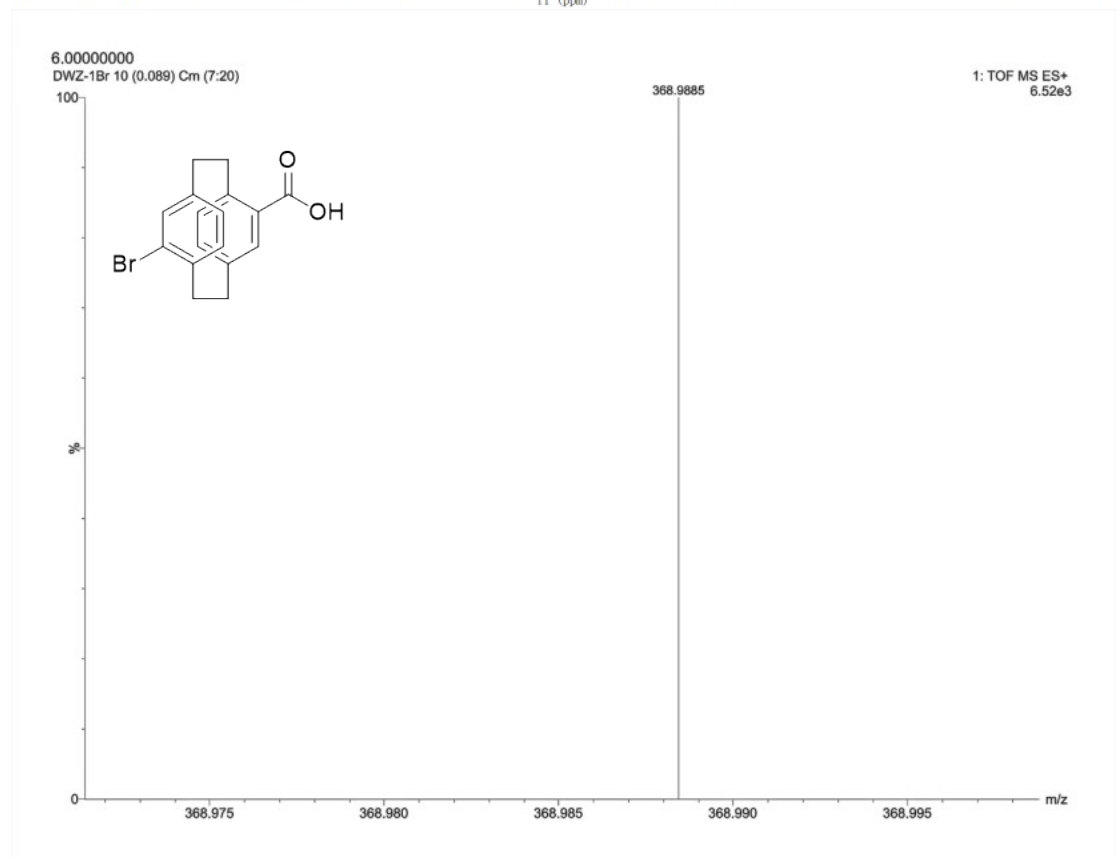
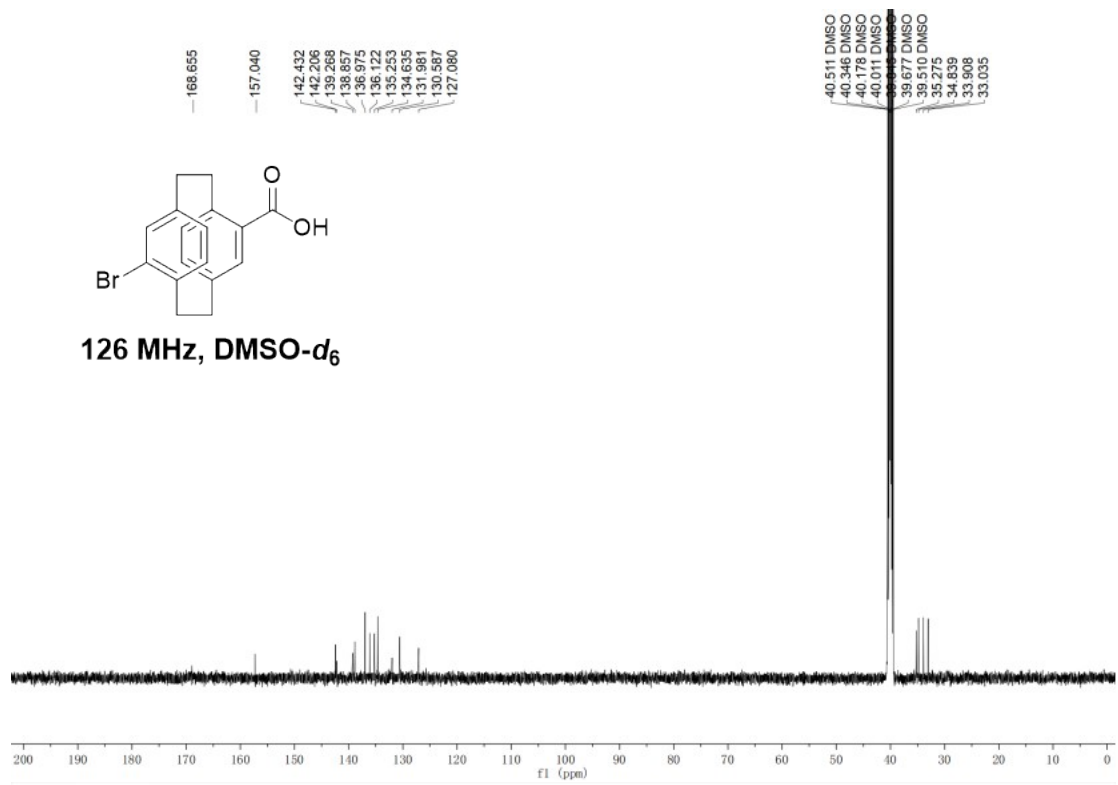
8. Copies of NMR and HRMS Spectra.



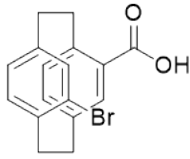
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DWZ-LWJ-0525-POS 127 (0.781) Cm (111:141)

1: TOF MS ES+
2.69e3

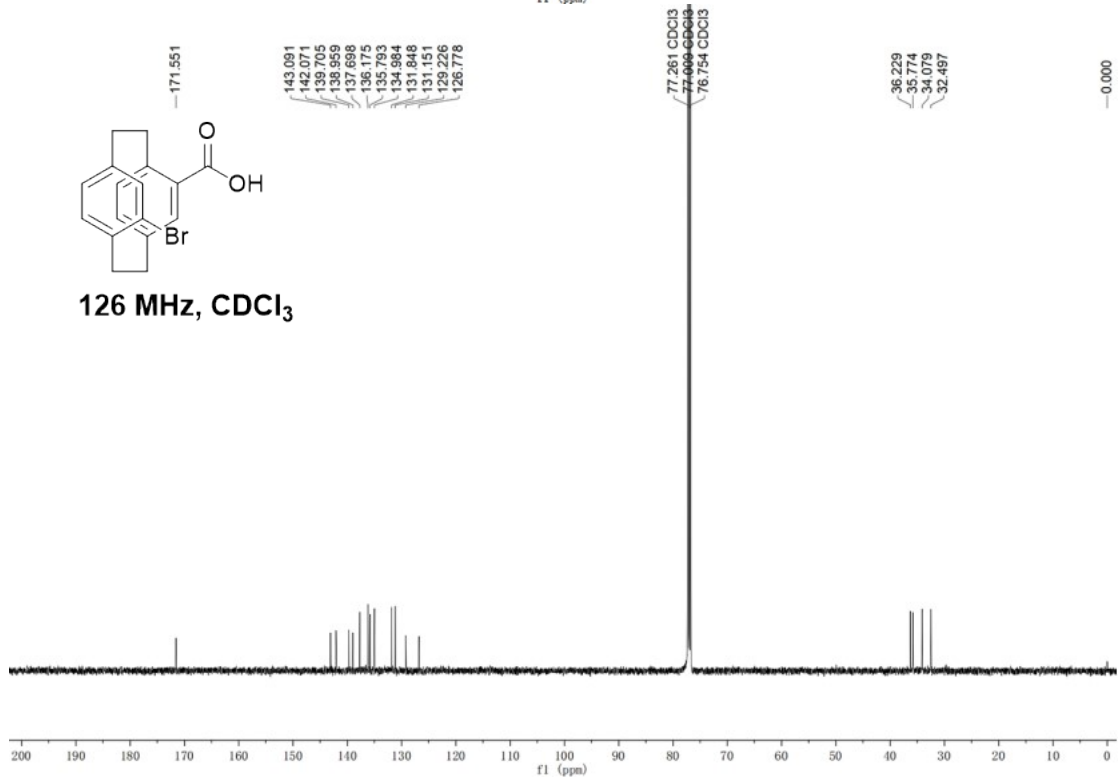
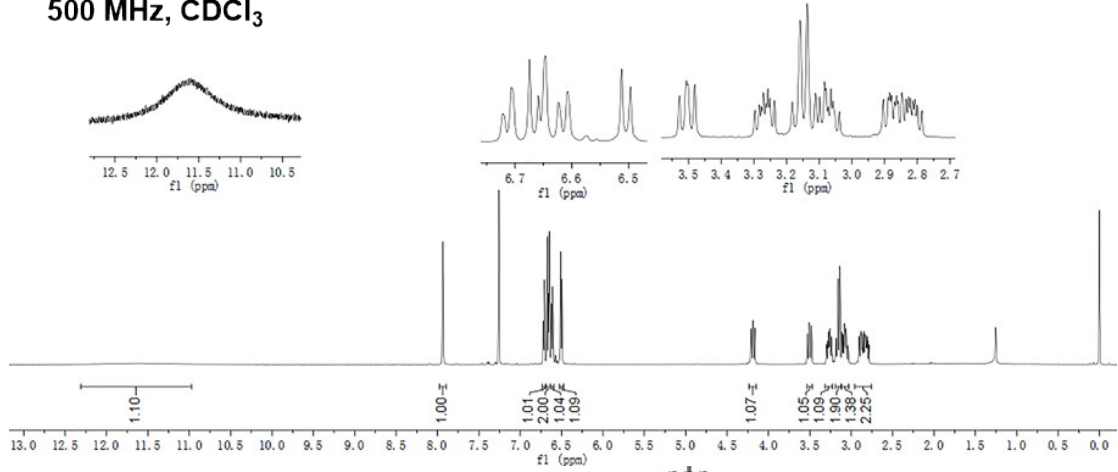




11.602
7.936
7.932
7.257 CDCl₃
6.723
6.719
6.707
6.674
6.659
6.649
6.646
6.624
6.620
6.609
6.605
6.577
6.572
6.566
6.512
6.497
4.211
4.192
4.184
4.168
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3.526
3.508
3.503
3.499
3.482
3.479
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3.271
3.263
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3.214
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3.137
3.112
3.099
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3.079
3.073
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3.058
3.053
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0.000

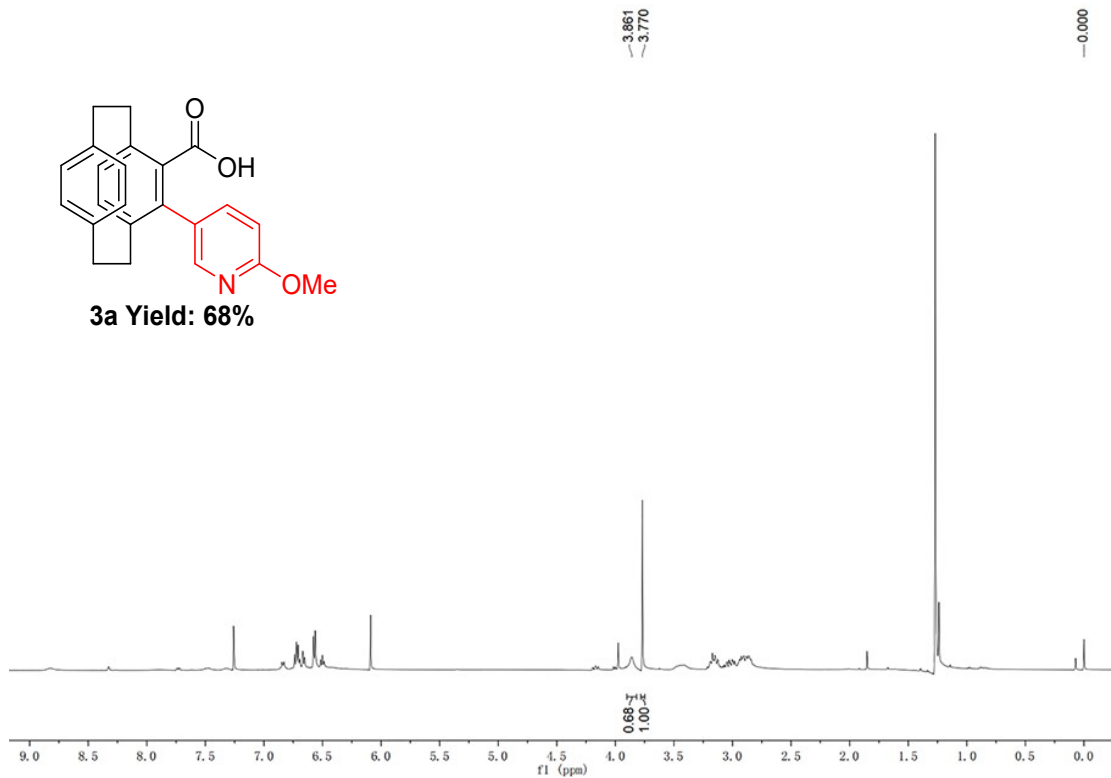
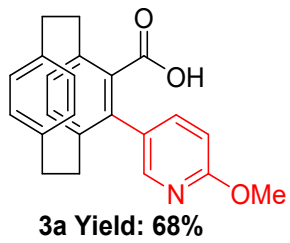
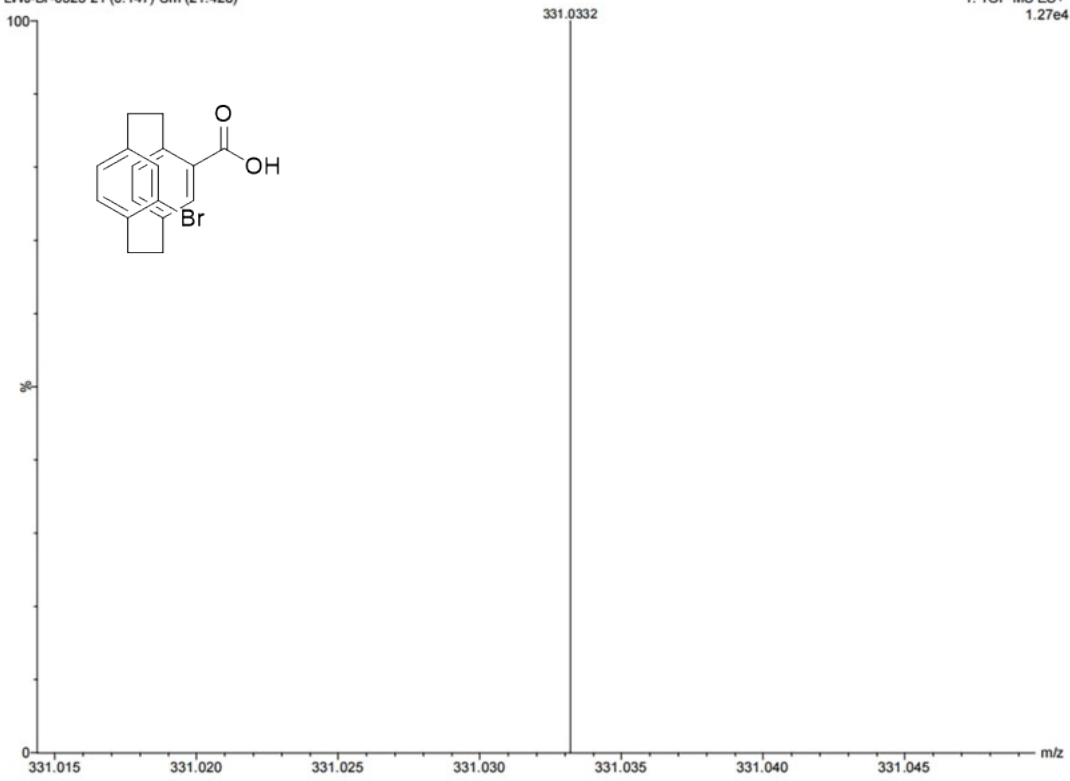


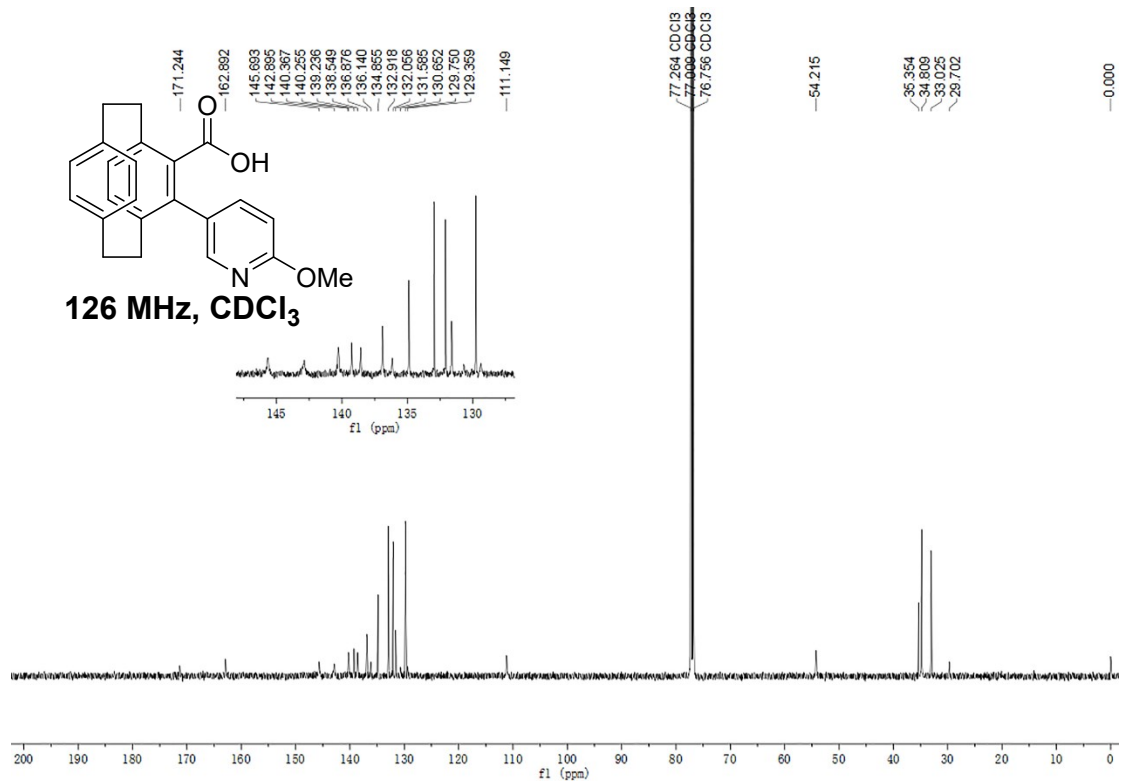
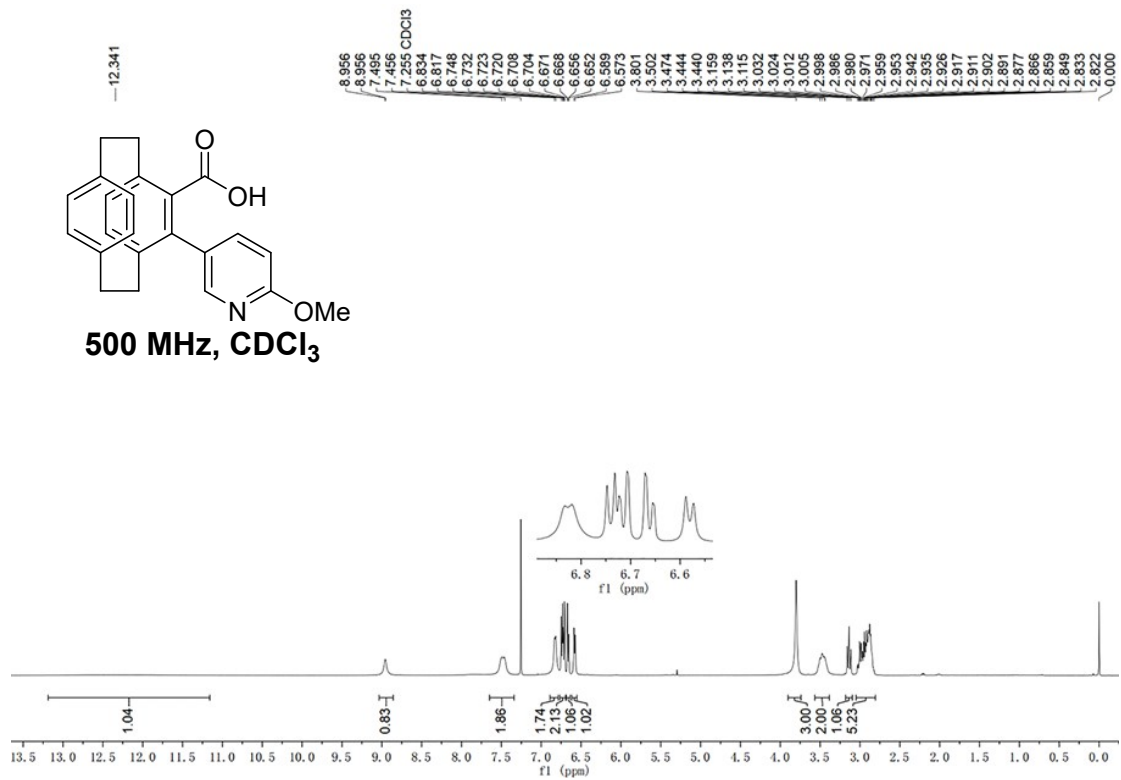
500 MHz, CDCl₃

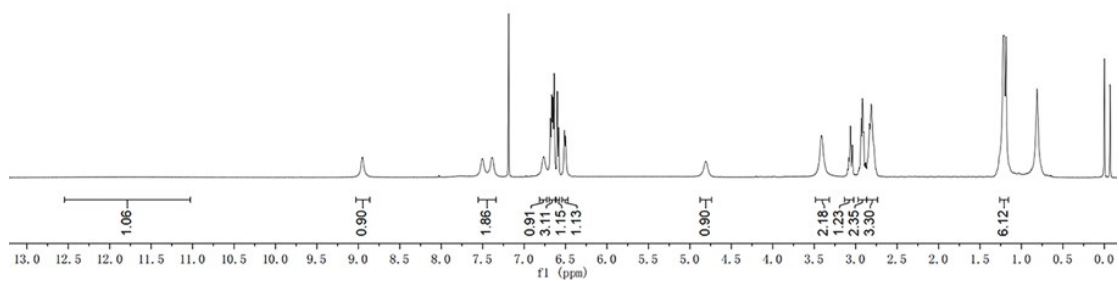
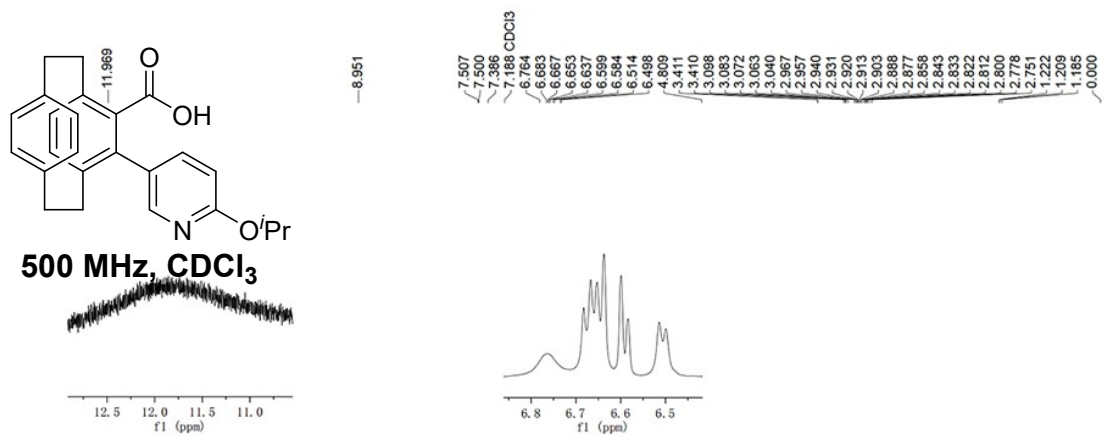
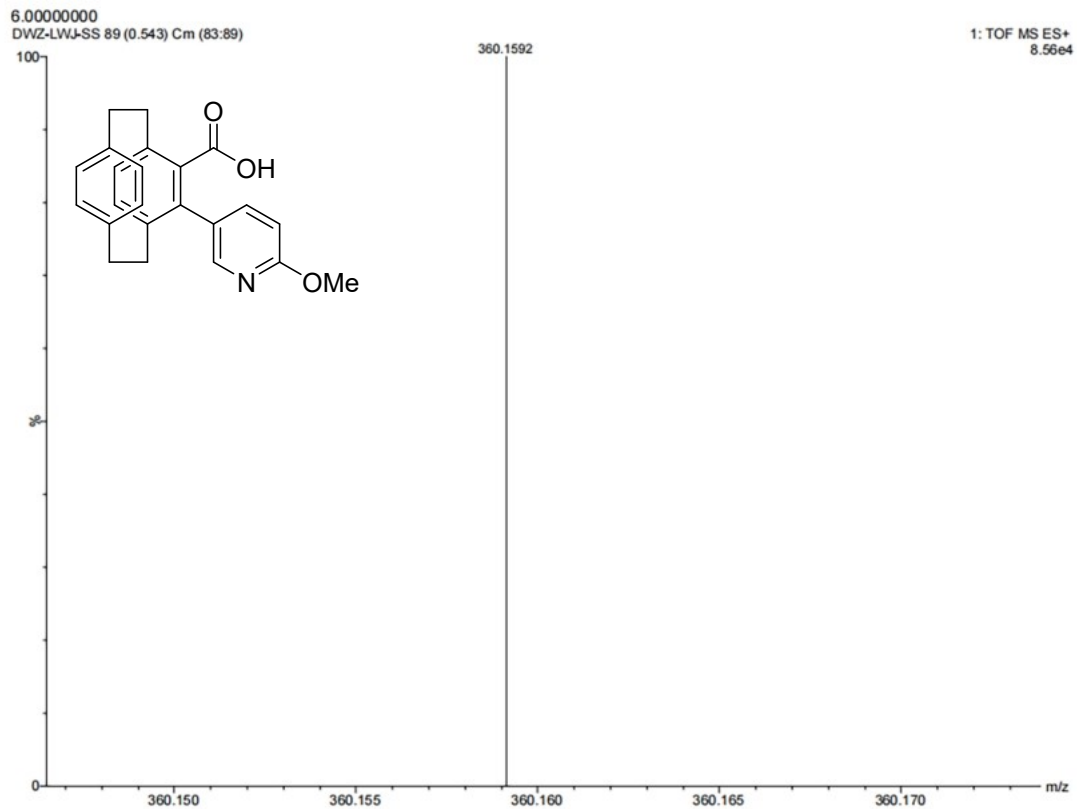


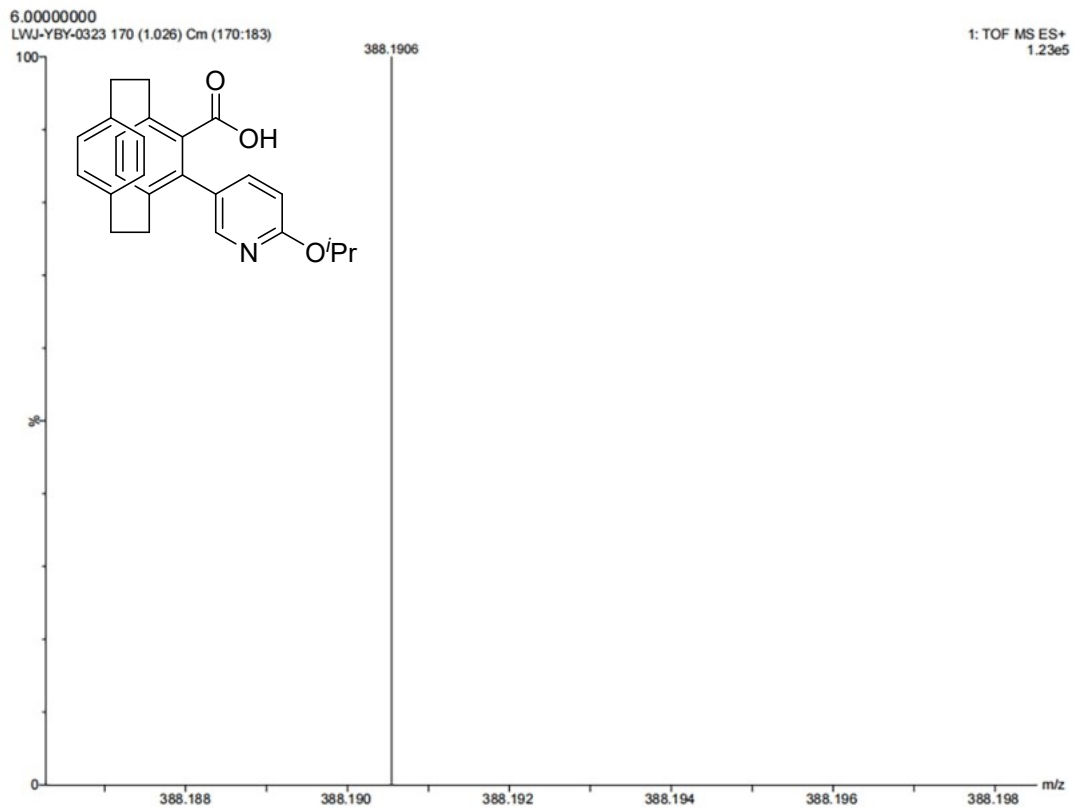
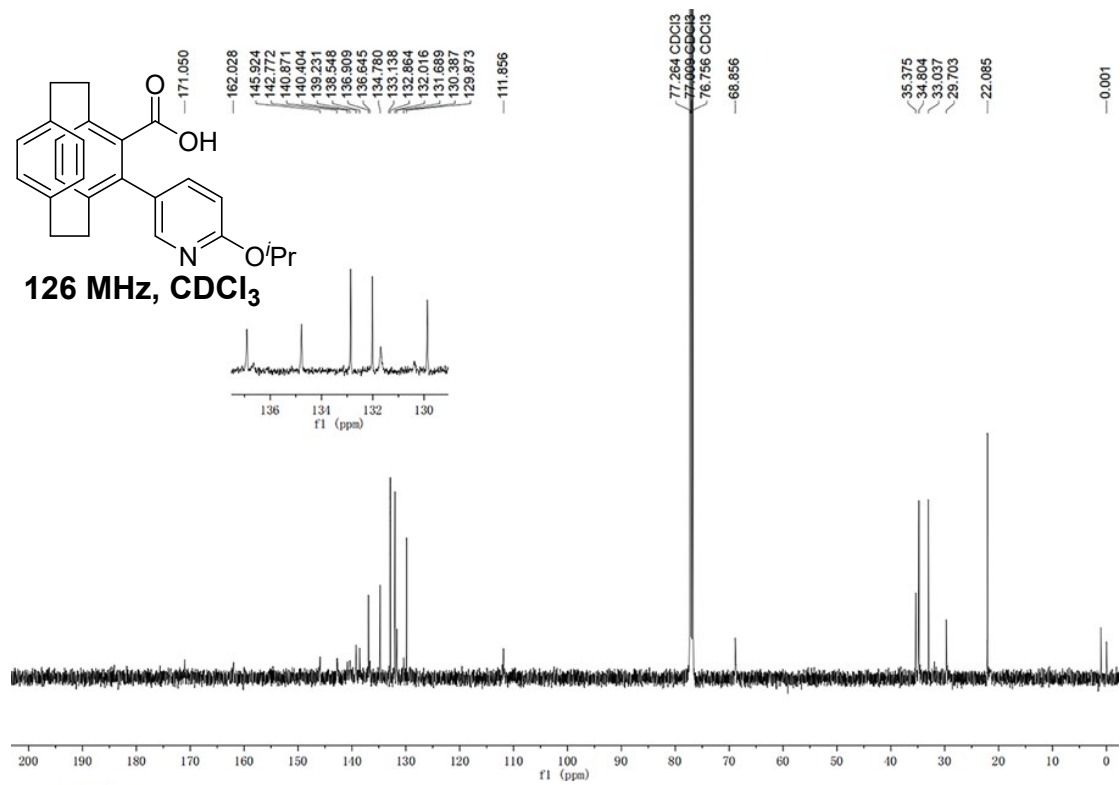
6.00000000
LWJ-Br-0323 21 (0.147) Cm (21:428)

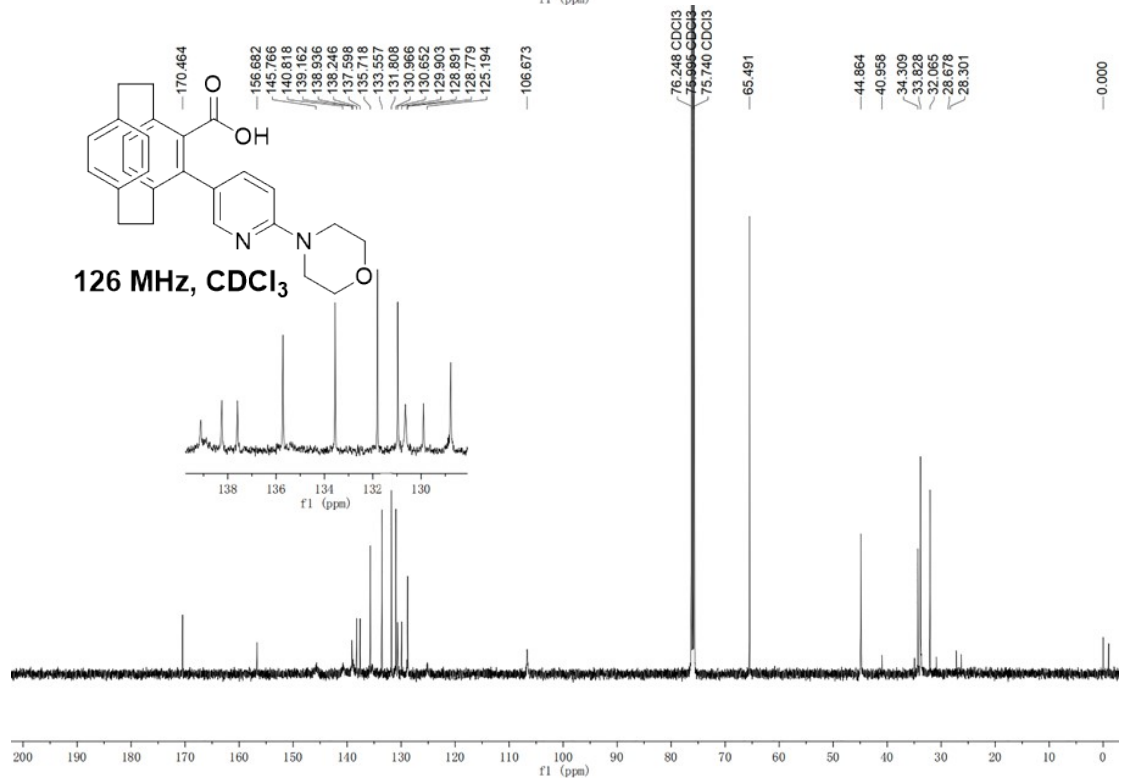
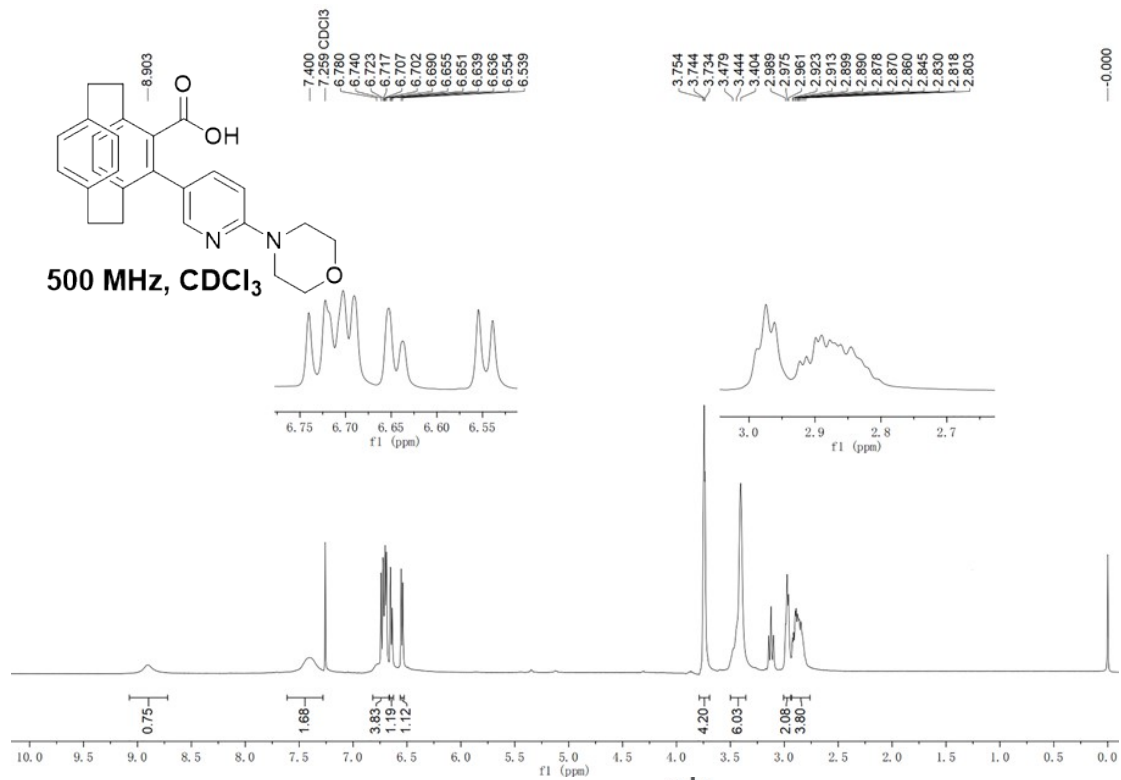
1: TOF MS ES+
1.27e4





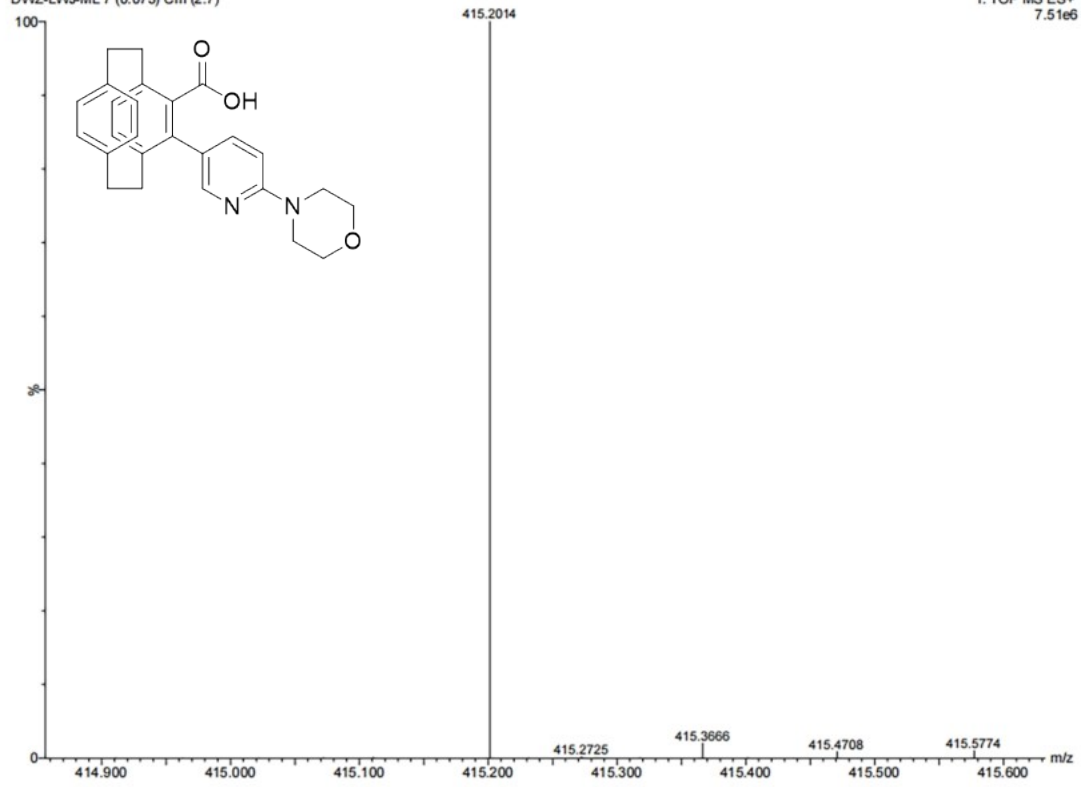




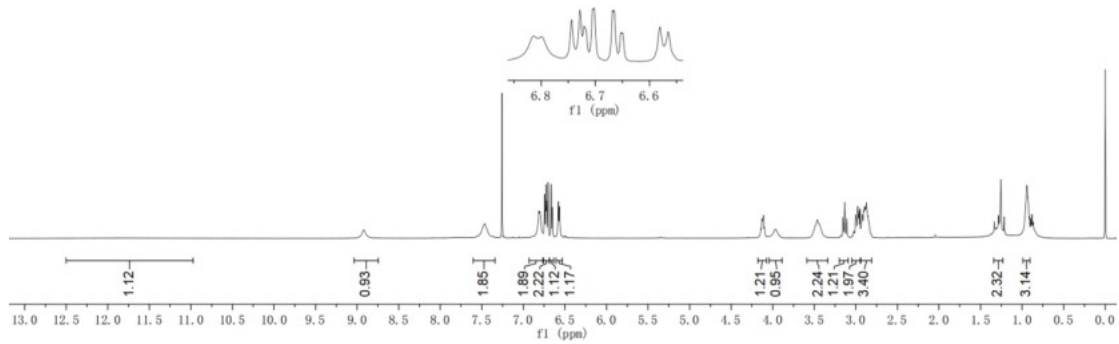
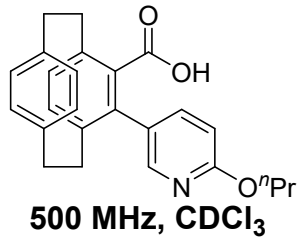


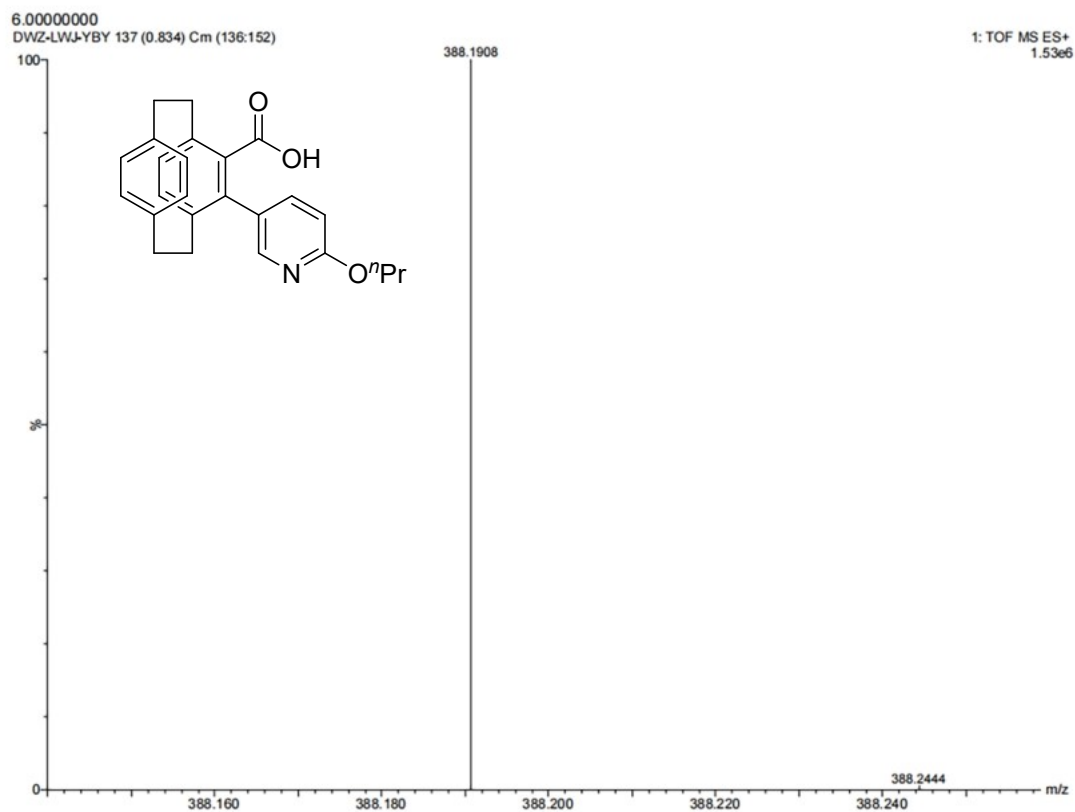
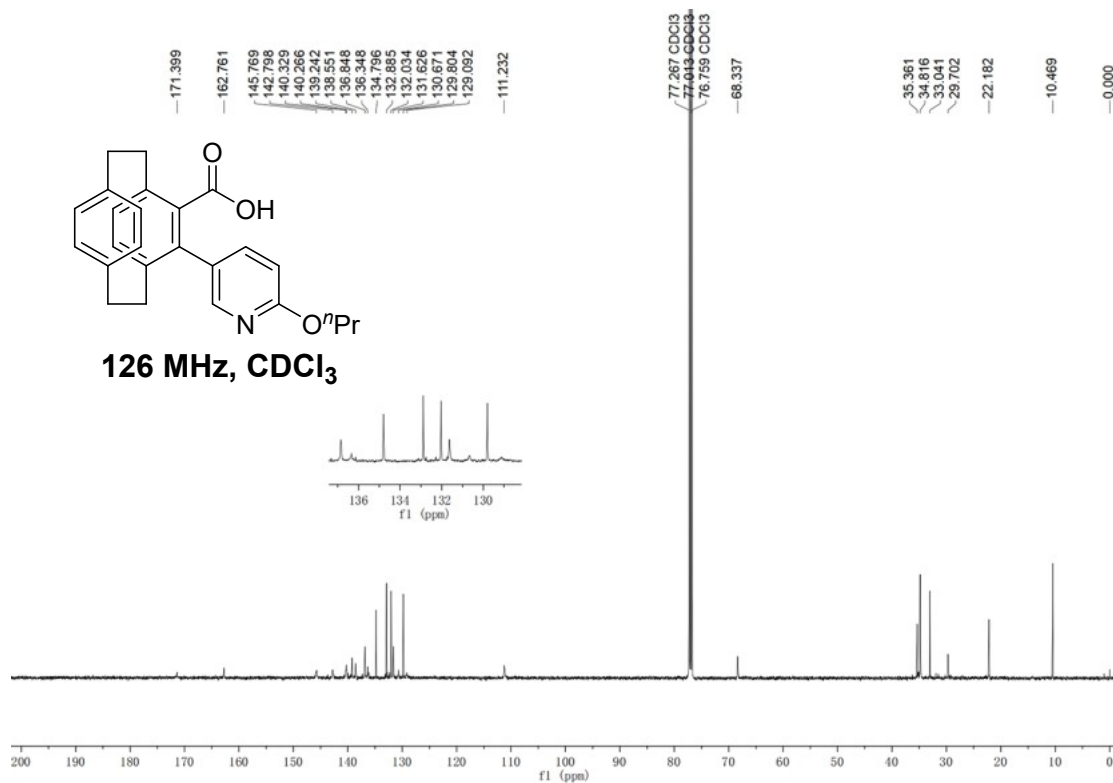
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DWZ-LWJ-ML 7 (0.073) Cm (2:7)

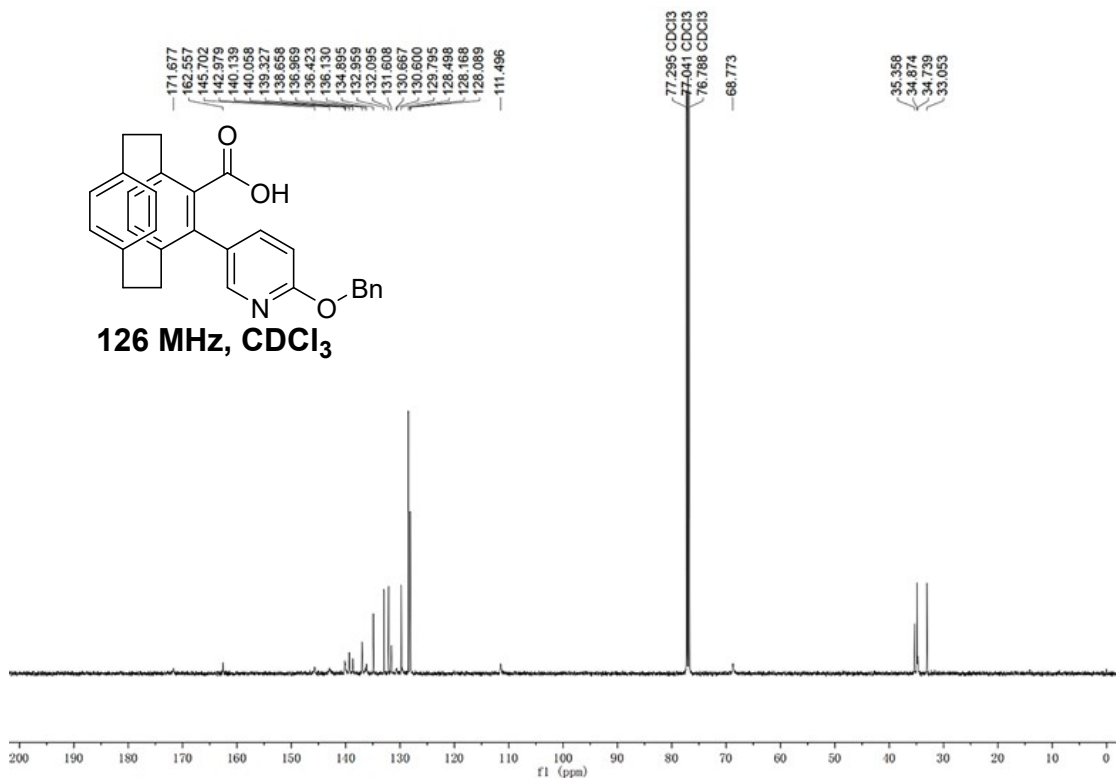
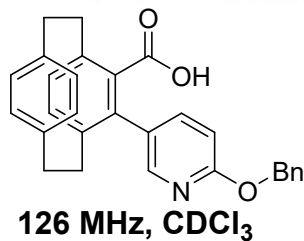
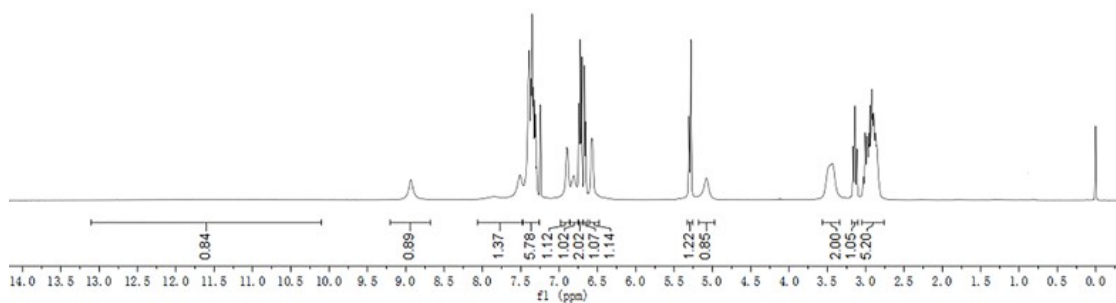
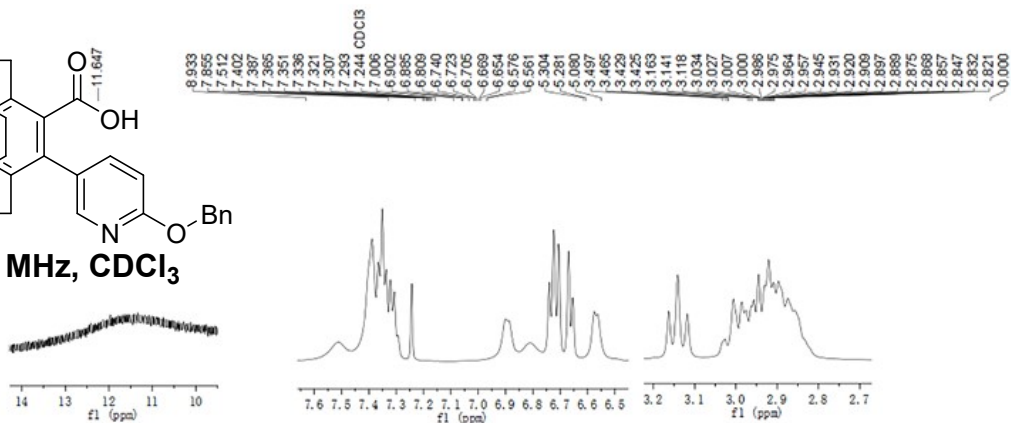
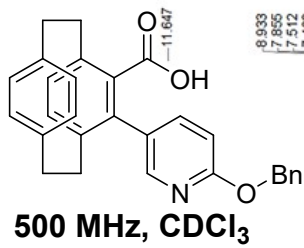
1: TOF MS ES+
7.51e6



11.773
8.919
7.466
7.257 CDCl3
6.815
6.798
6.744
6.728
6.721
6.717
6.705
6.702
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6.665
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6.581
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4.111
3.086
3.060
3.050
3.178
3.155
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3.009
3.002
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0.911
0.000

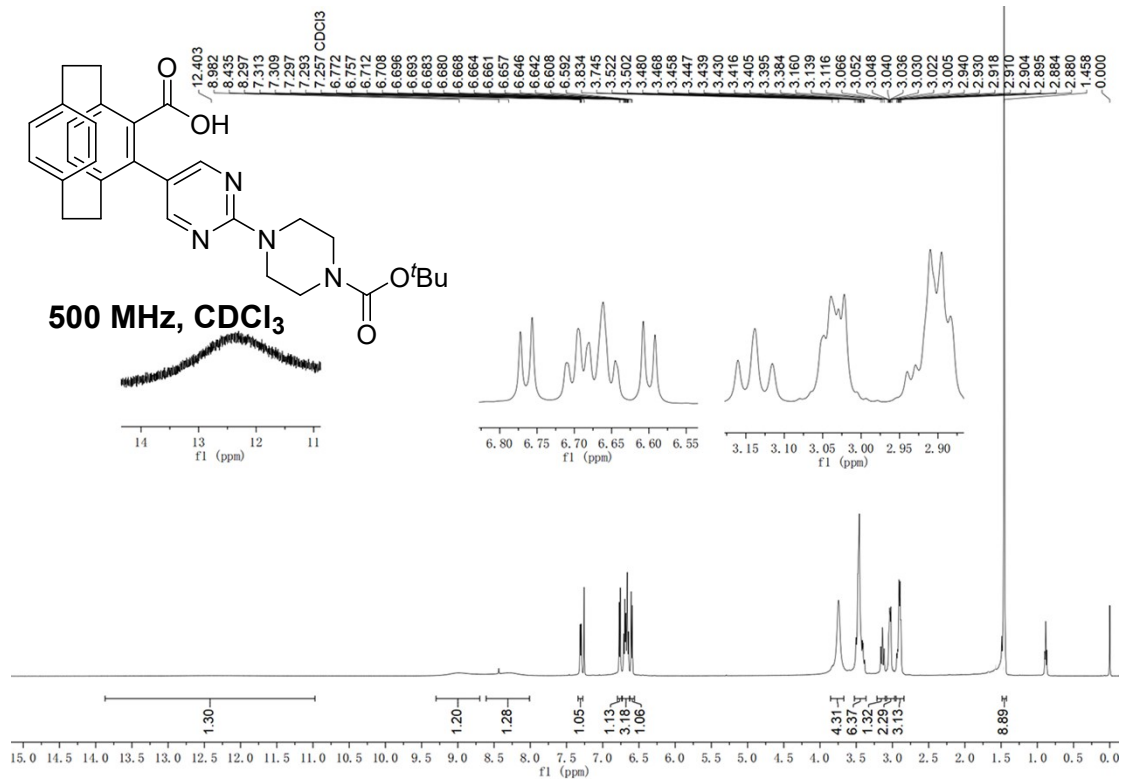
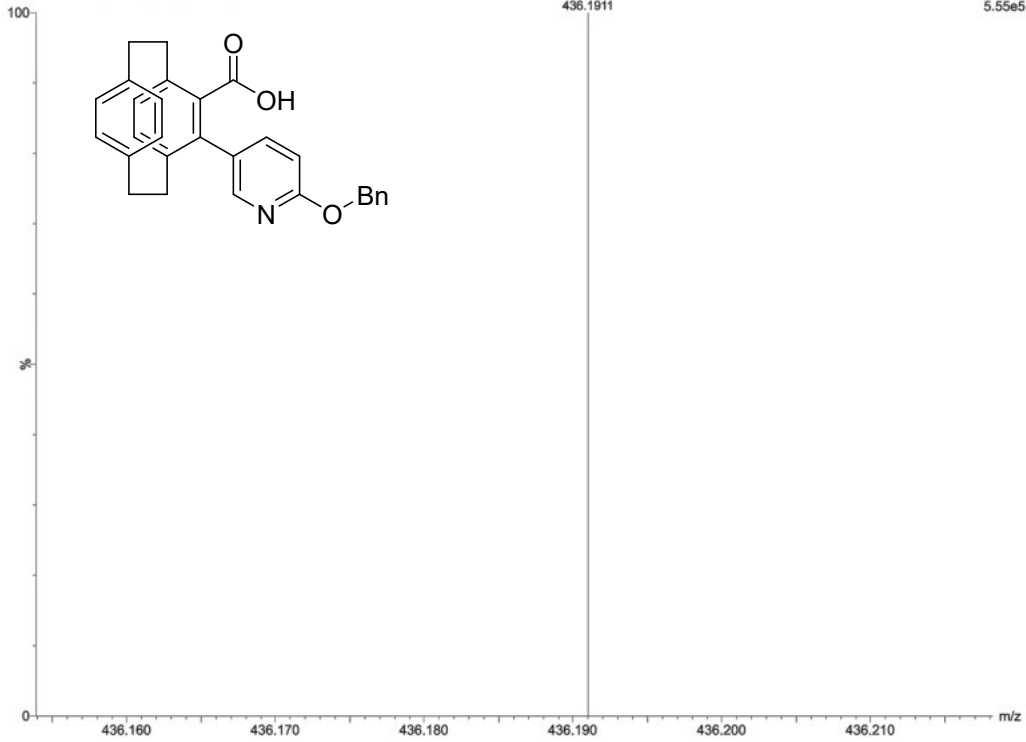


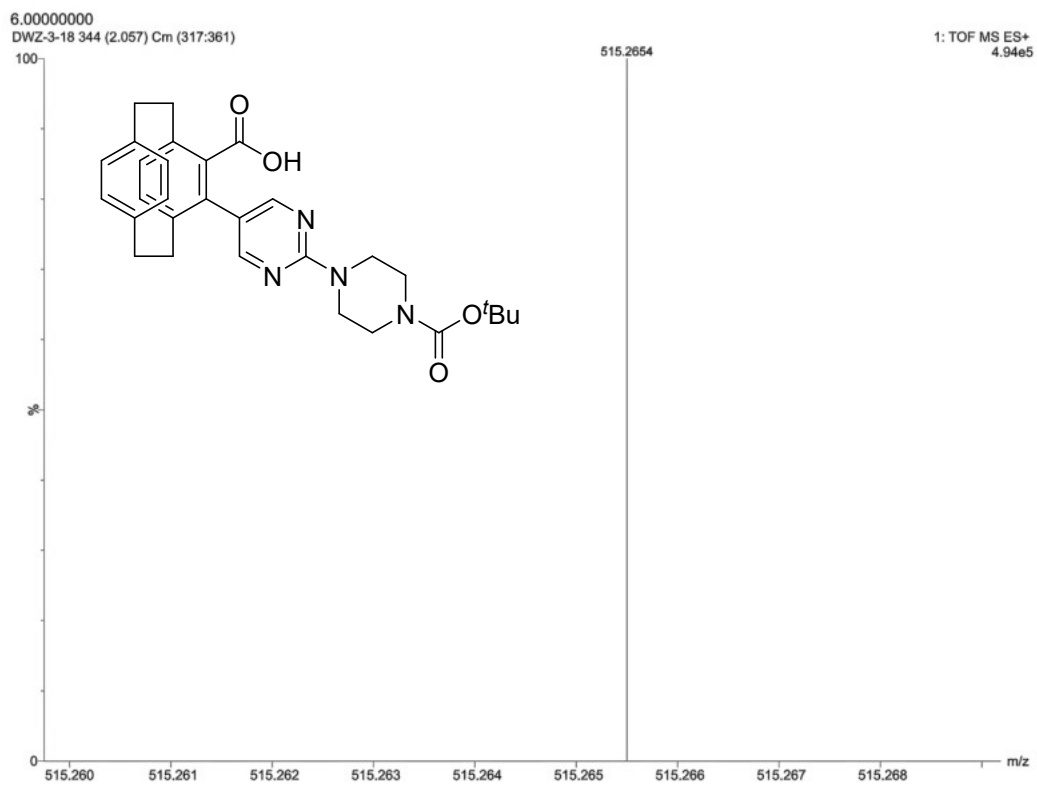
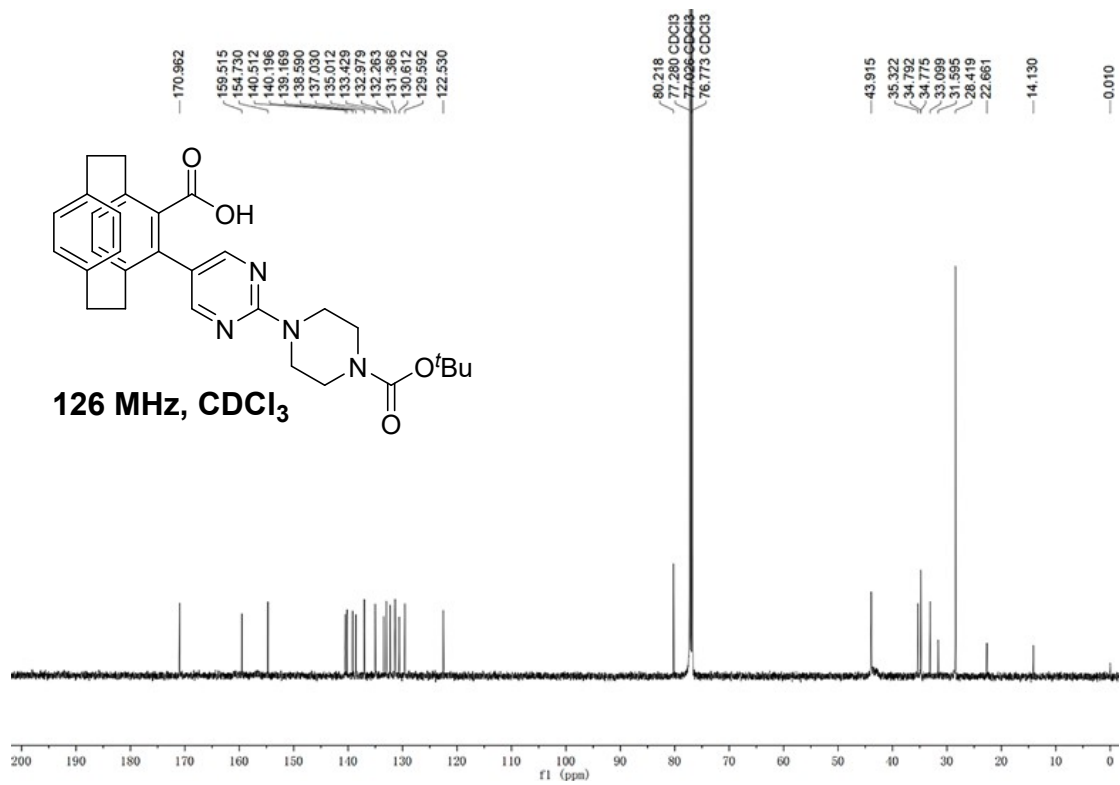


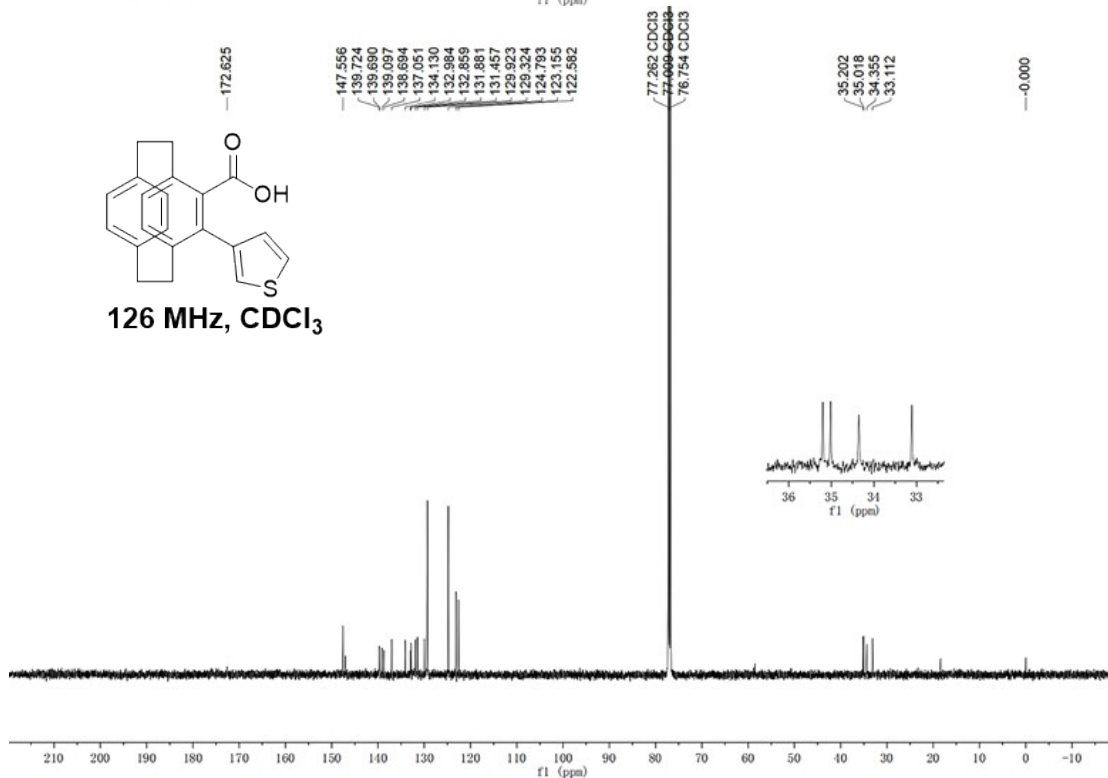
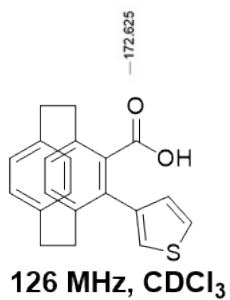
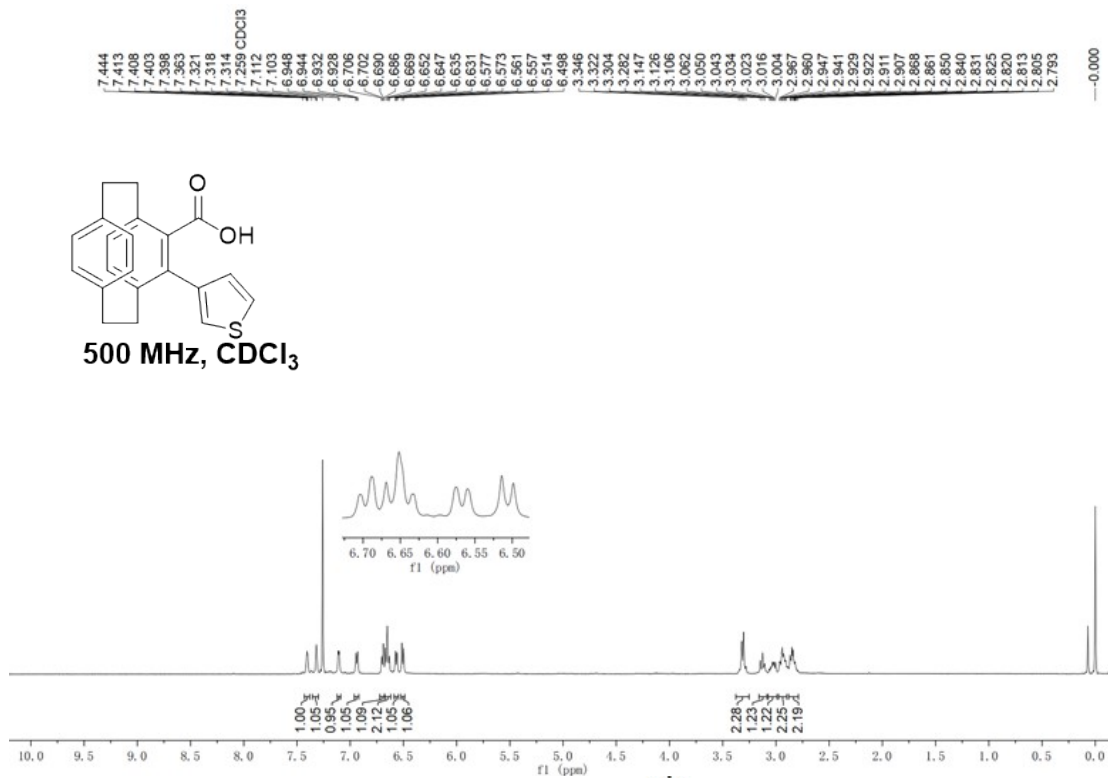


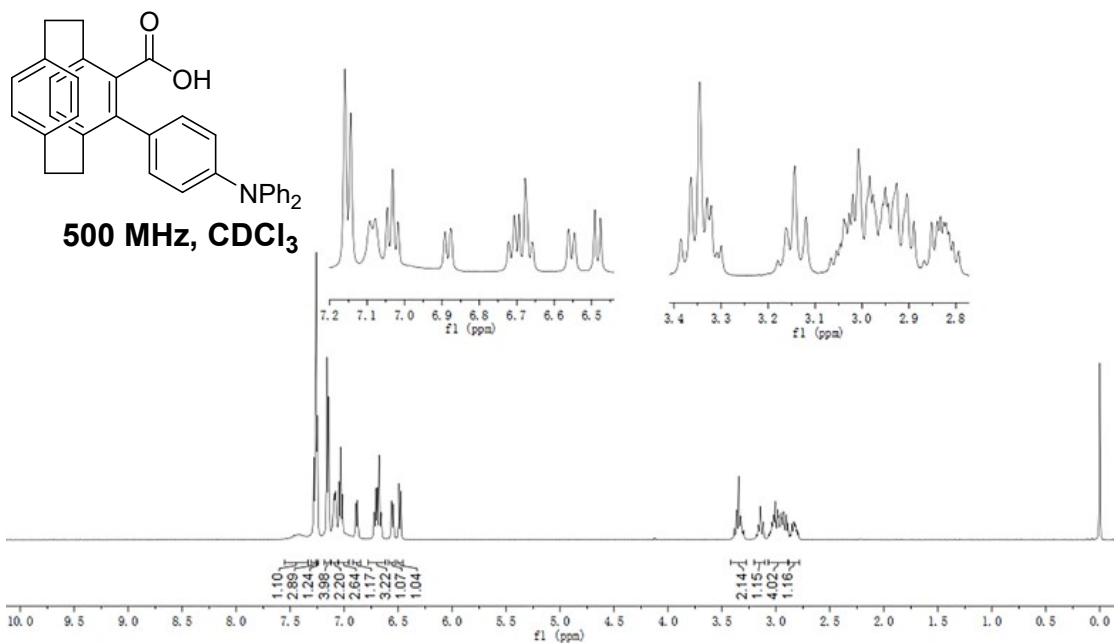
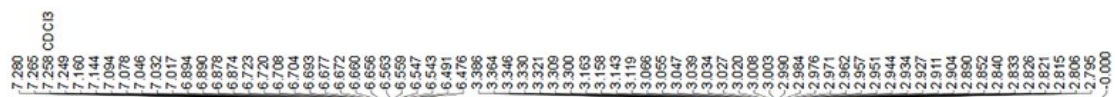
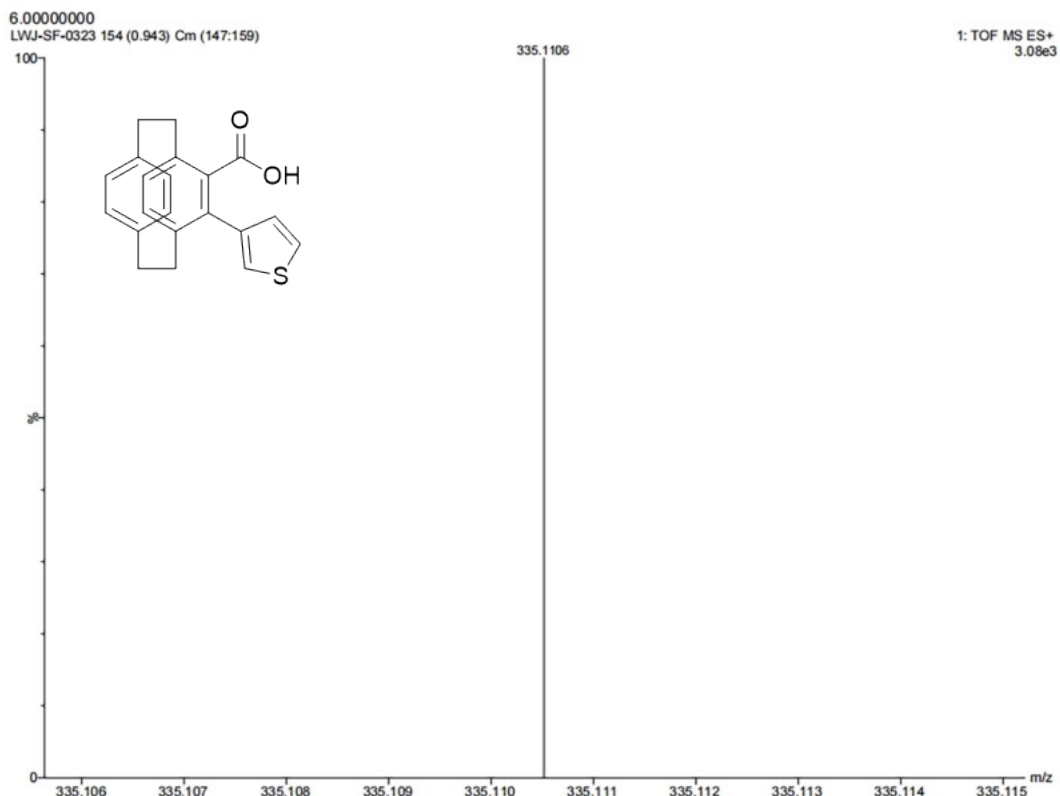
6.0000000
DWZ-2-170 287 (1.719) Cm (286:300)

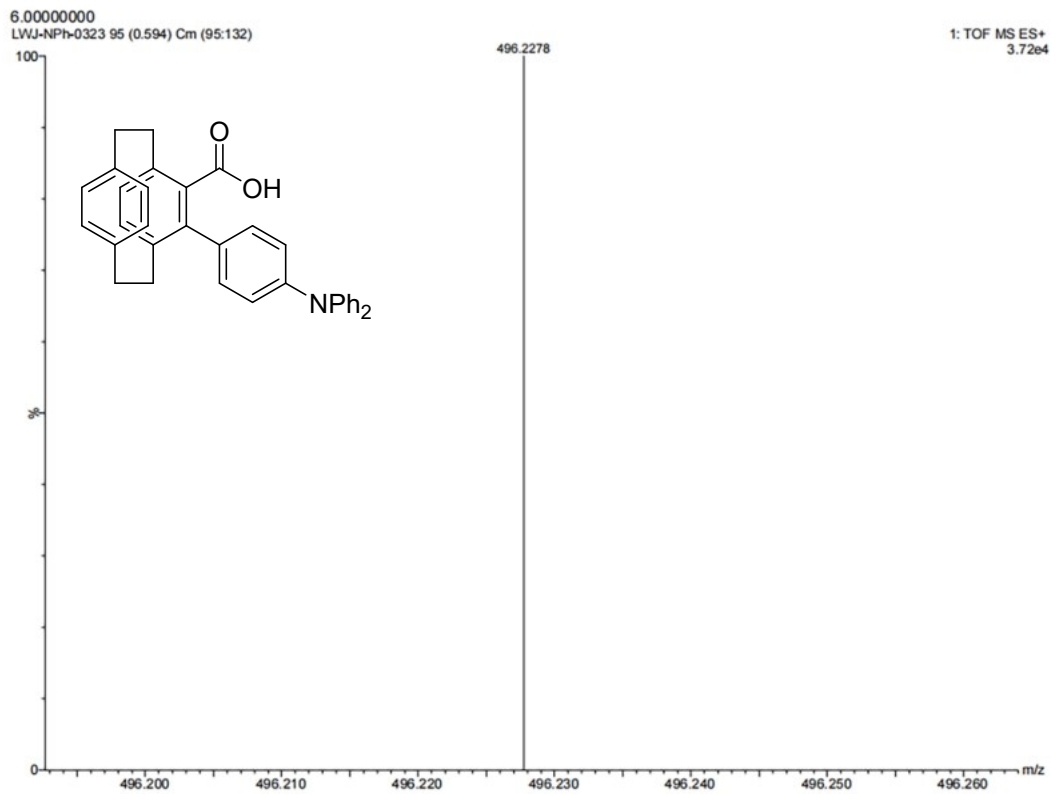
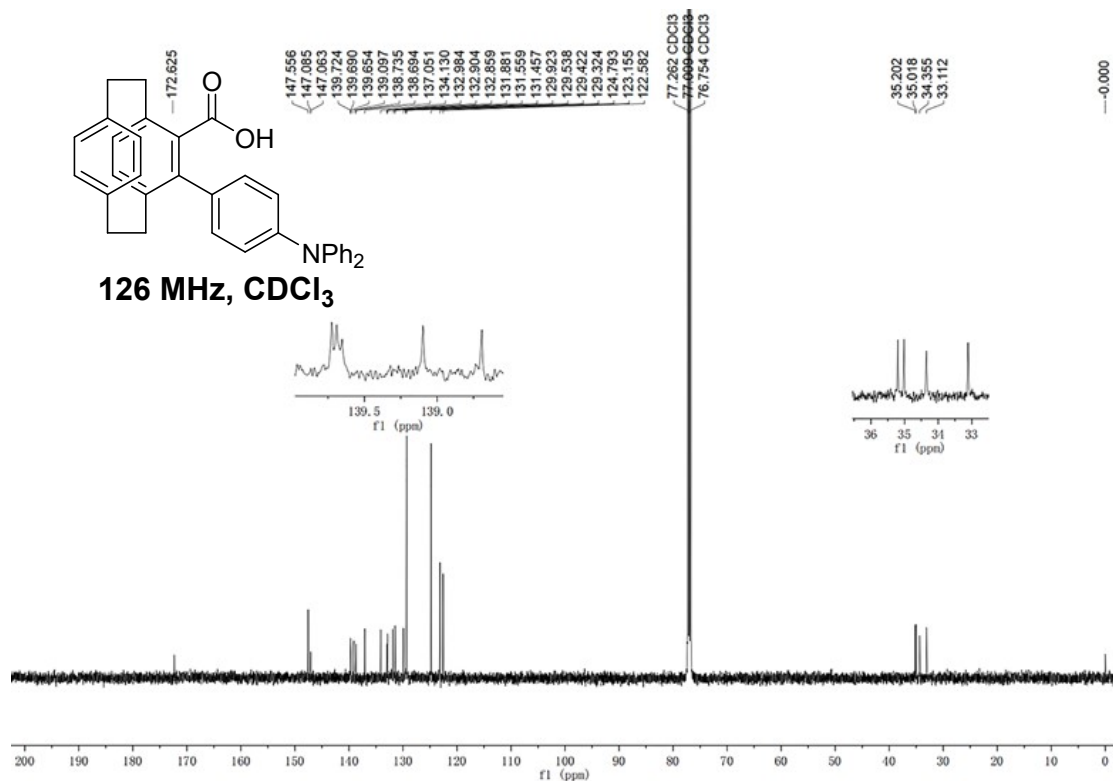
1: TOF MS ES+
5.55e5

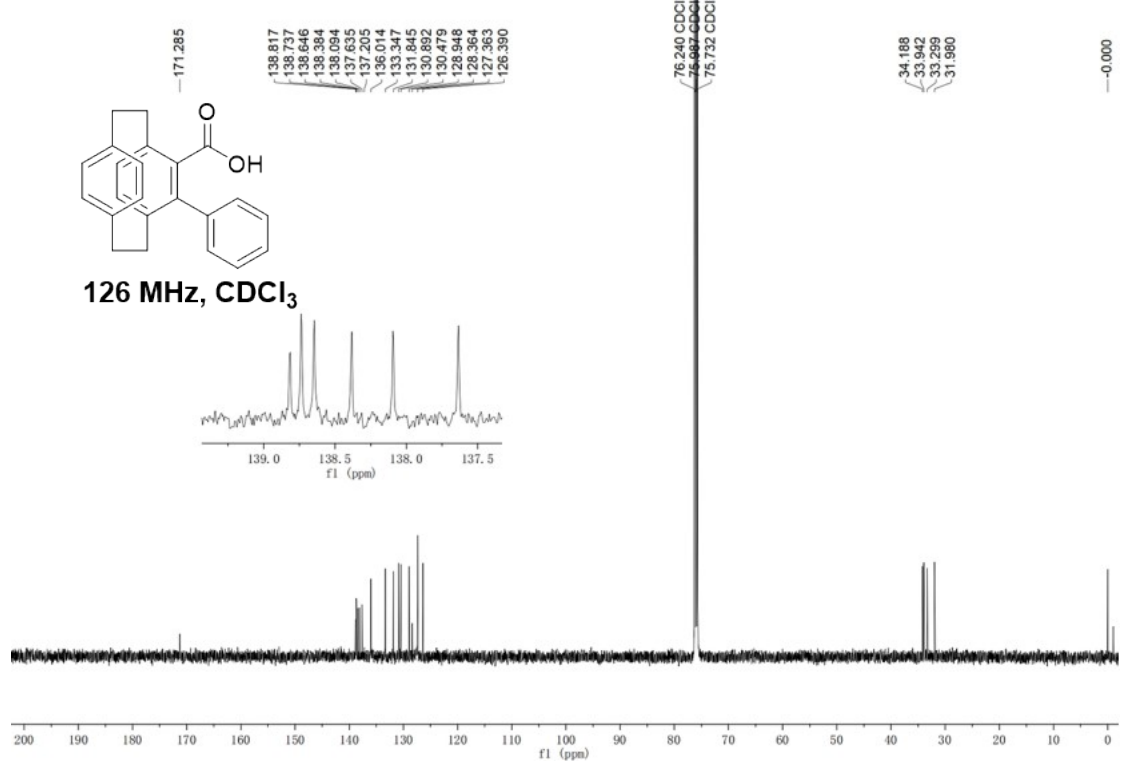
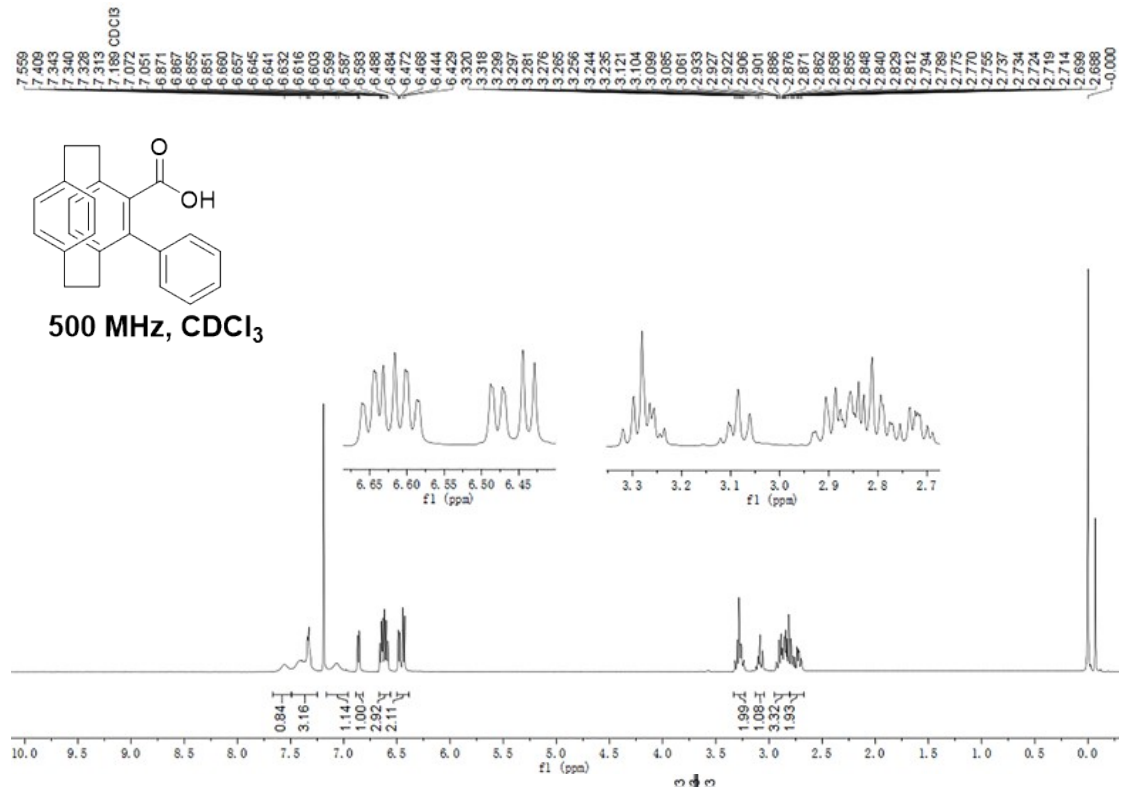


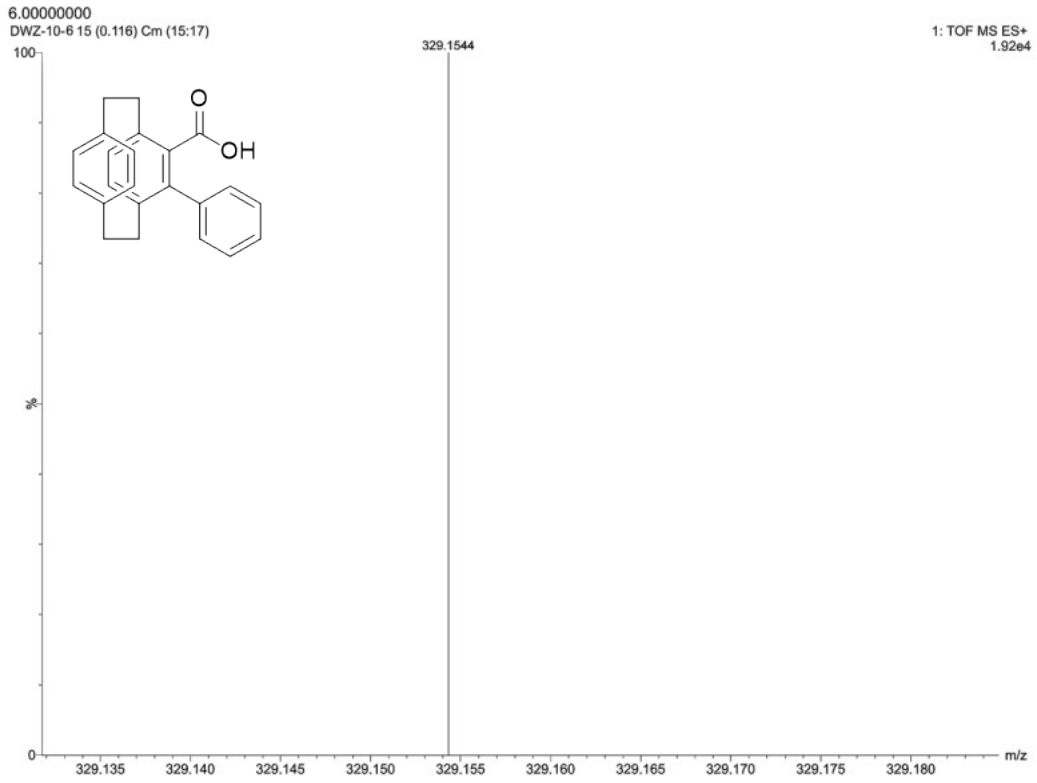




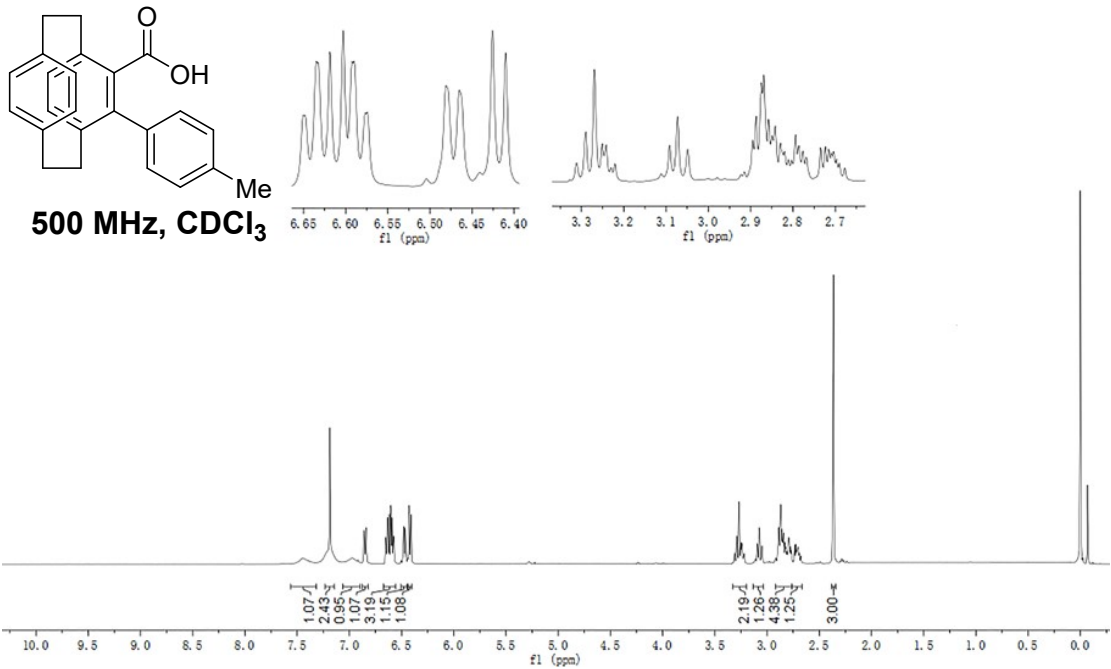


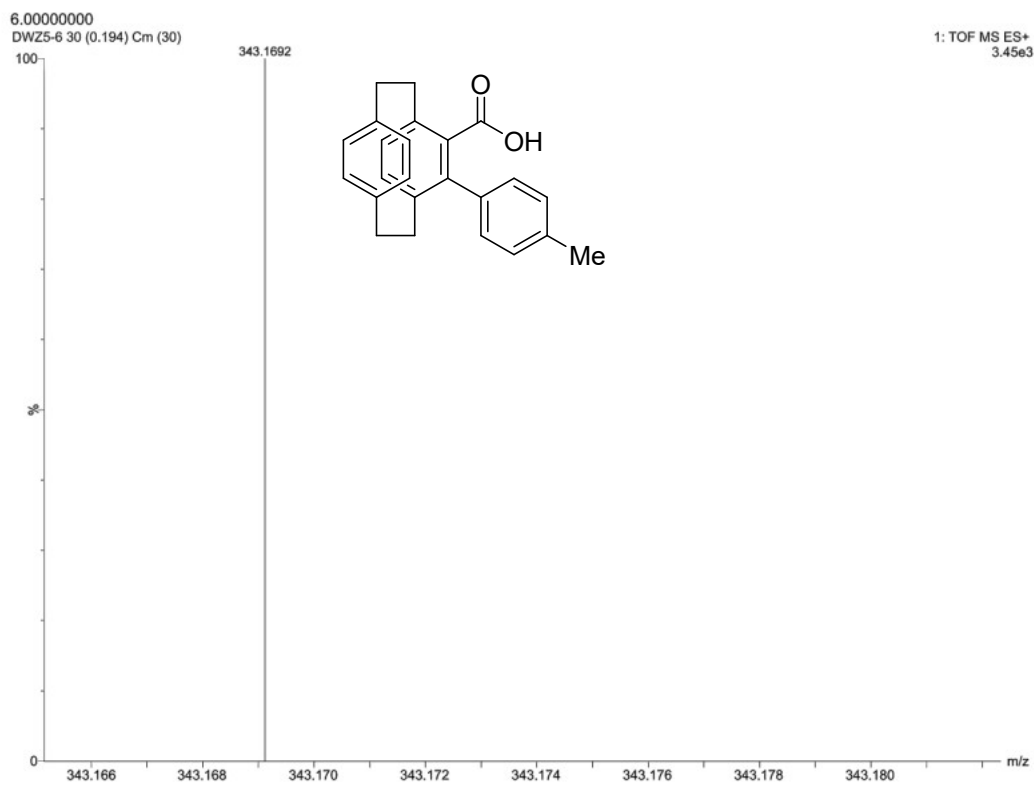
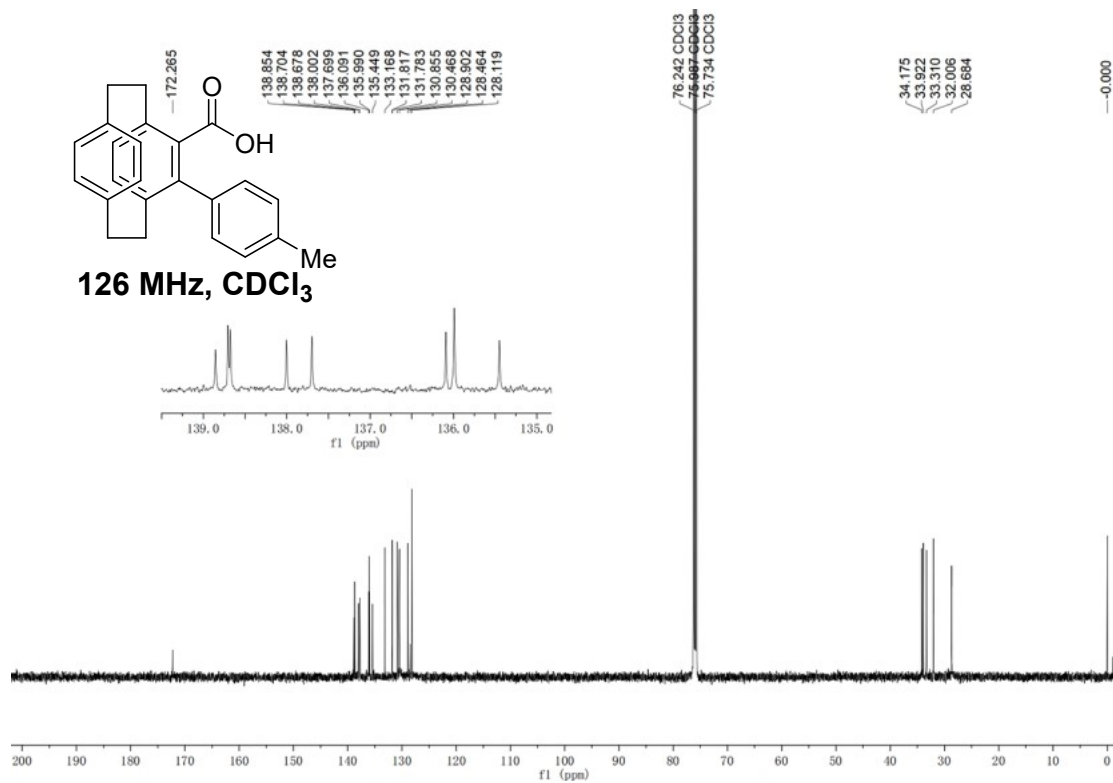


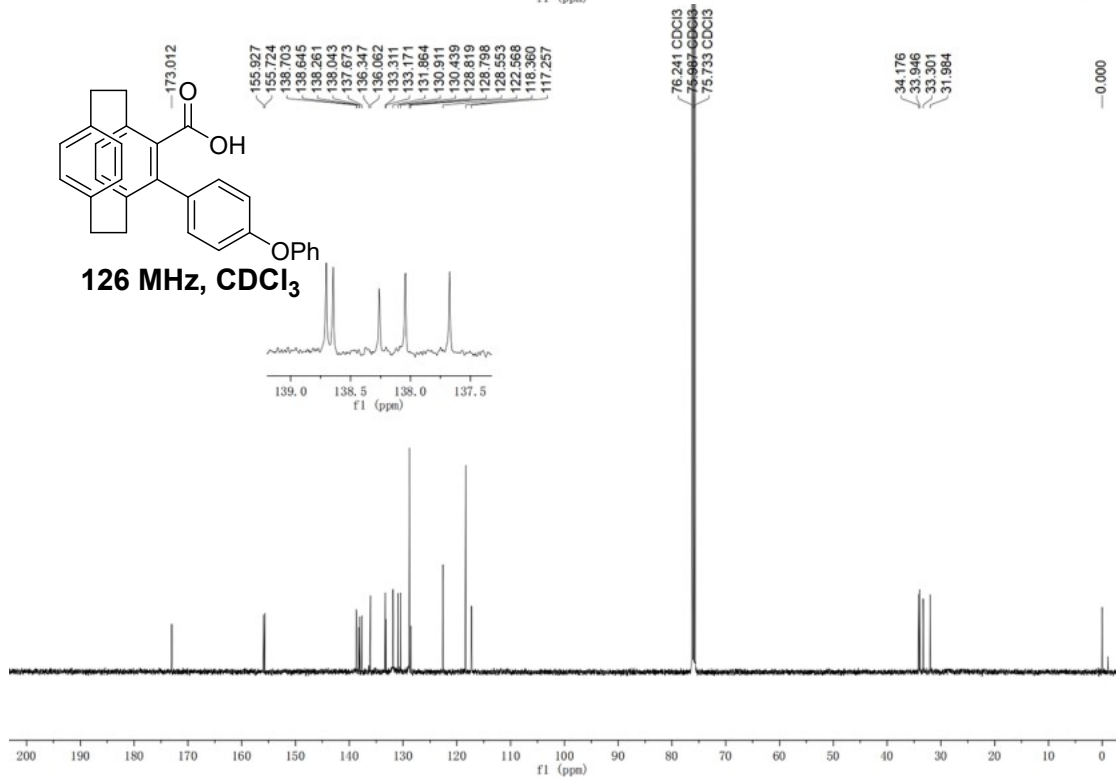
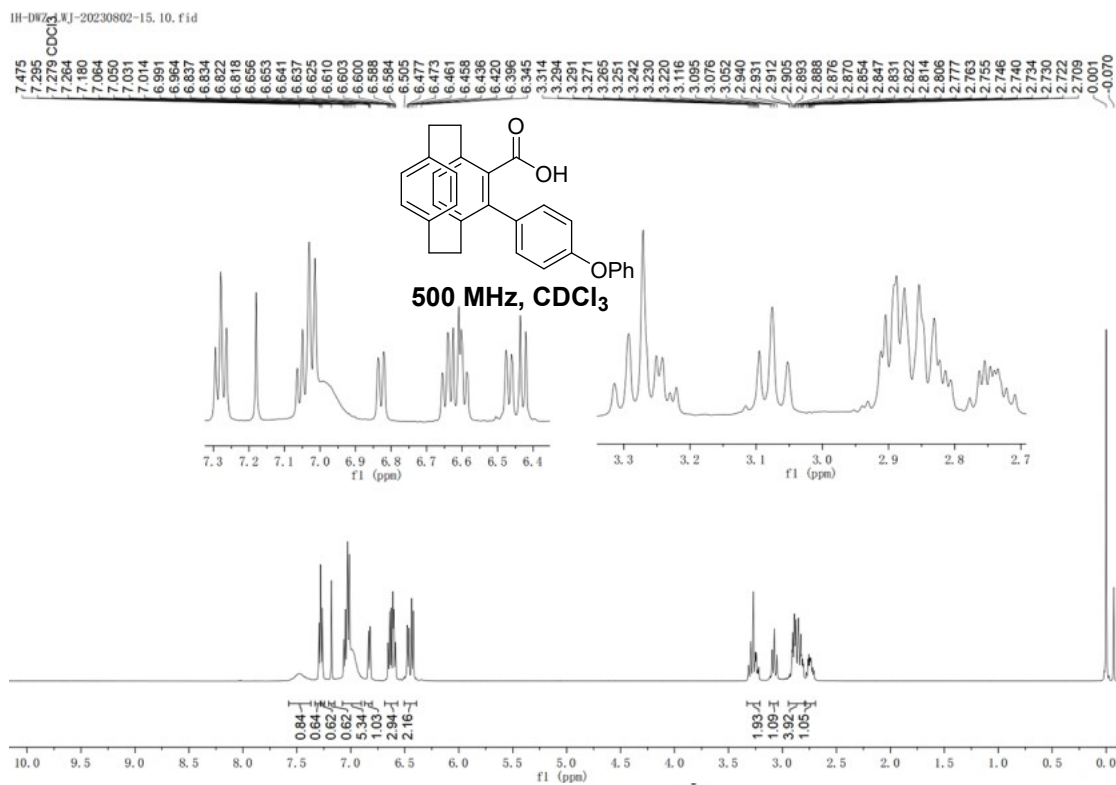




7.463
7.213
7.186
7.173
6.982
6.869
6.857
6.843
6.841
6.650
6.633
6.619
6.603
6.592
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6.577
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3.230
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2.857
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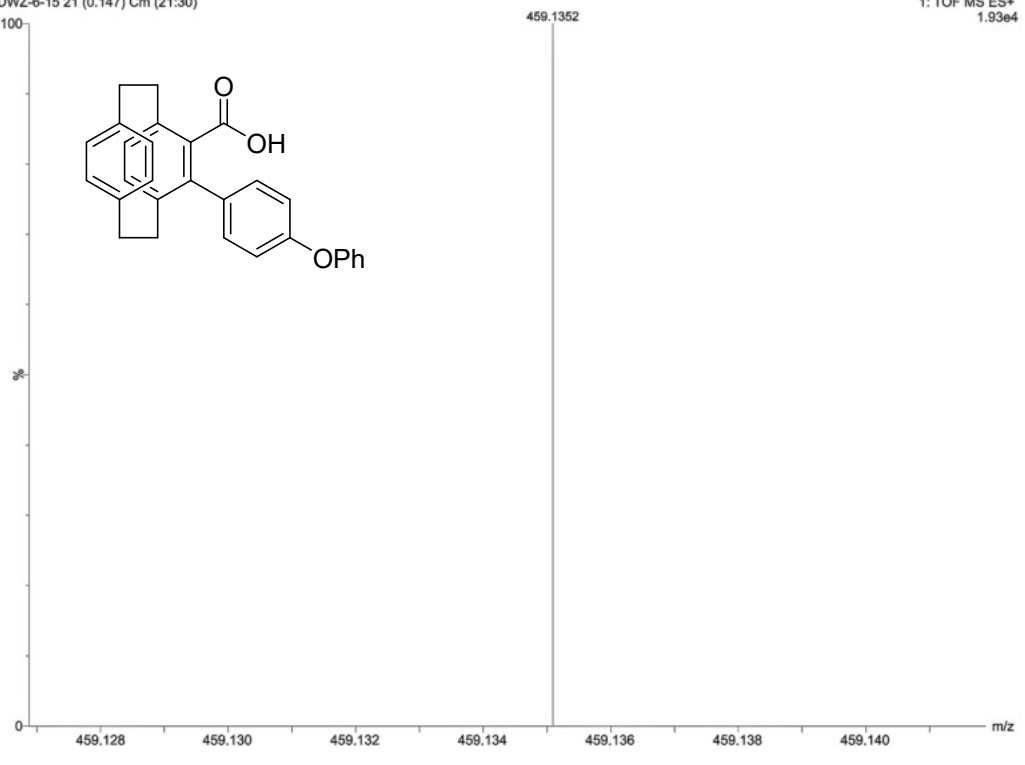




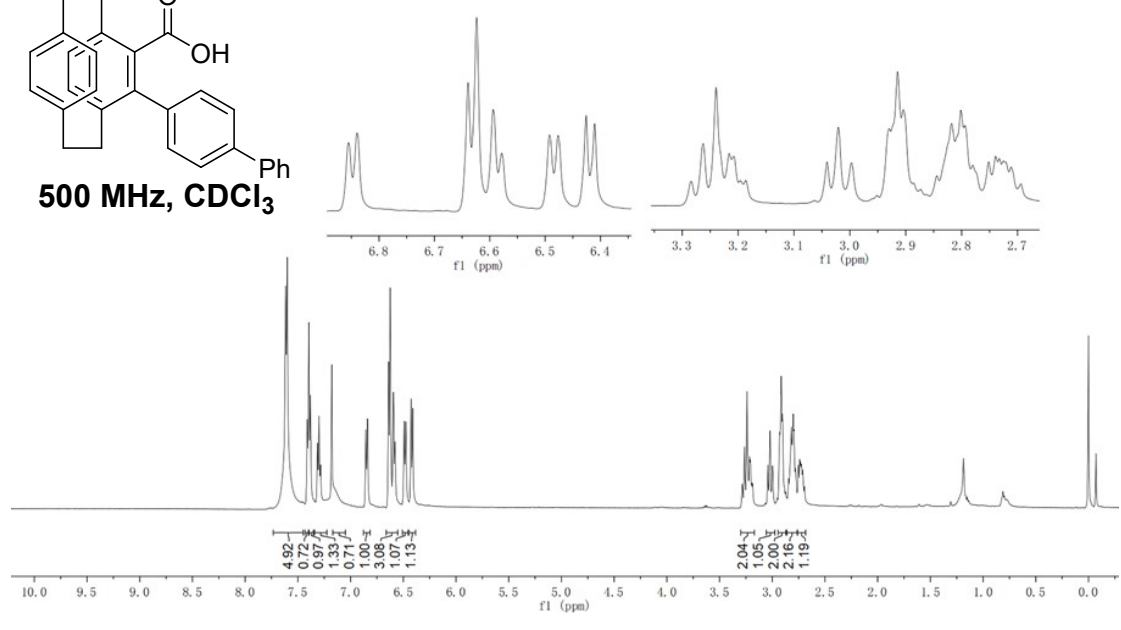
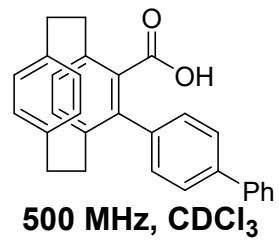


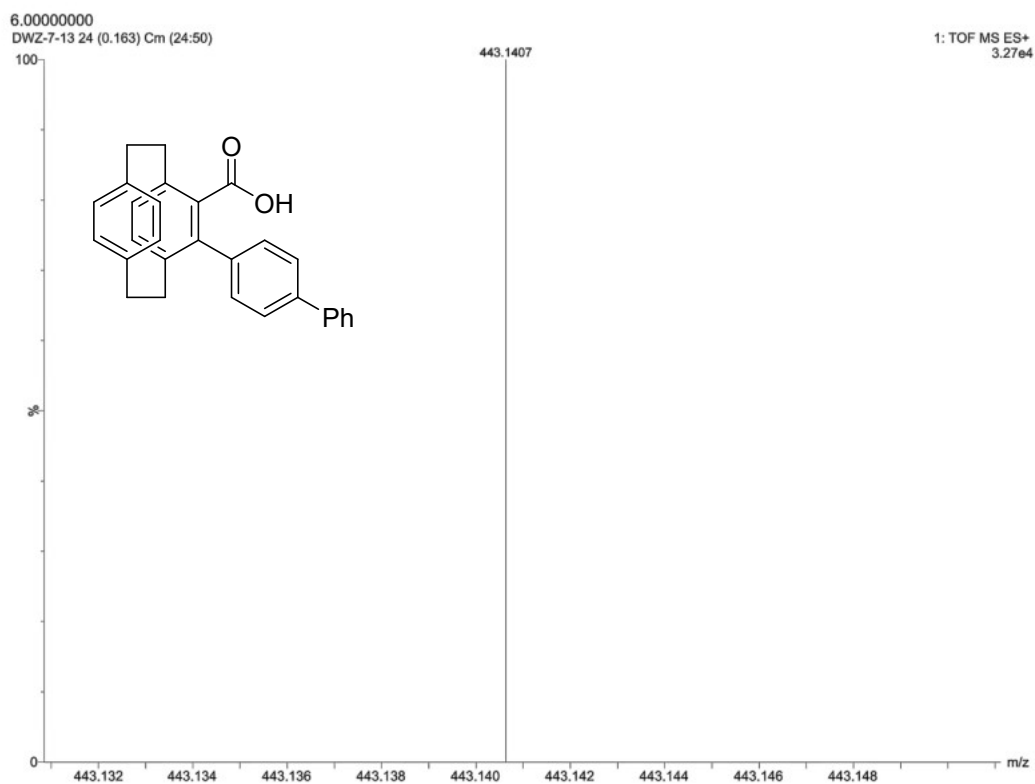
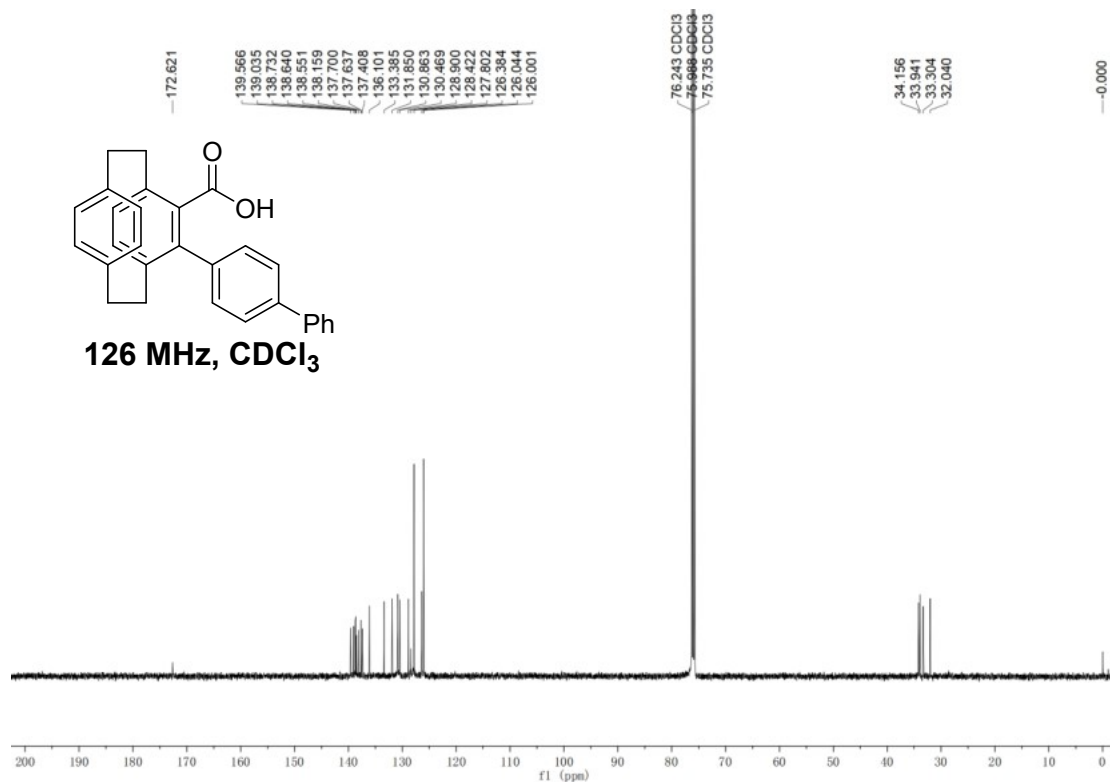
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DWZ-6-15 21 (0.147) Cm (21:30)

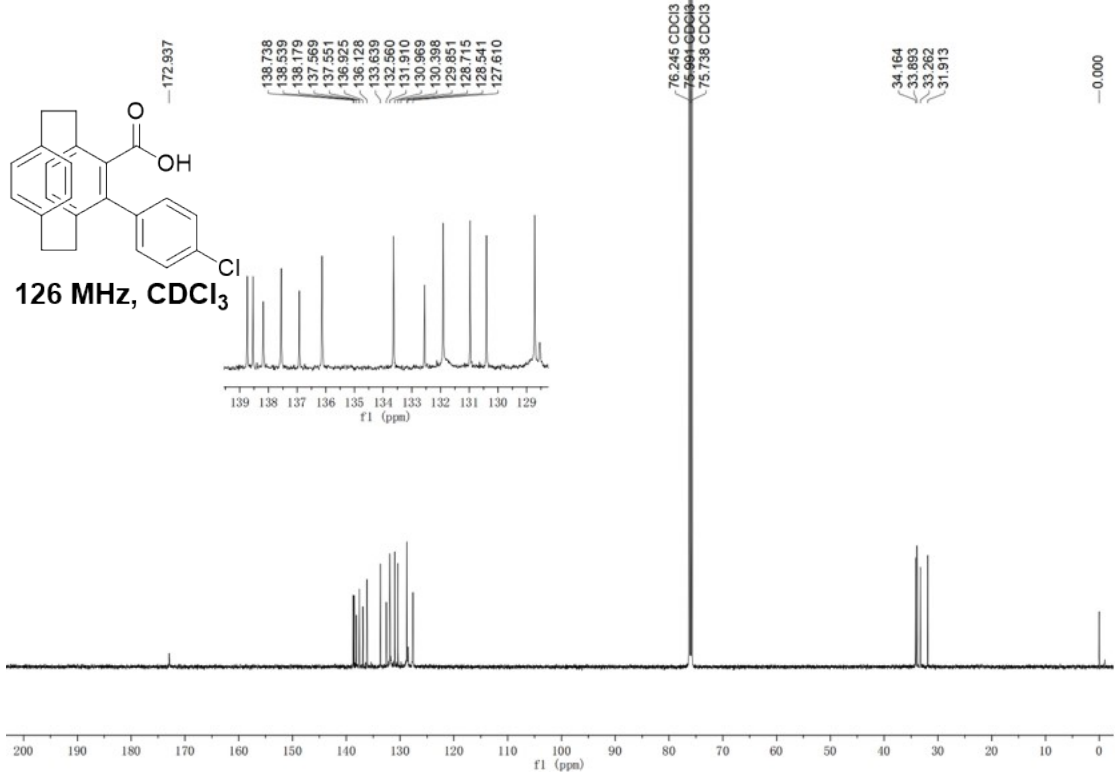
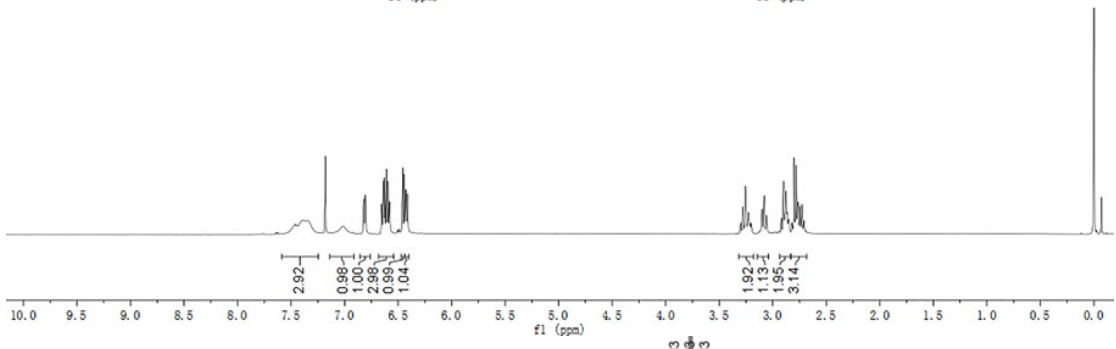
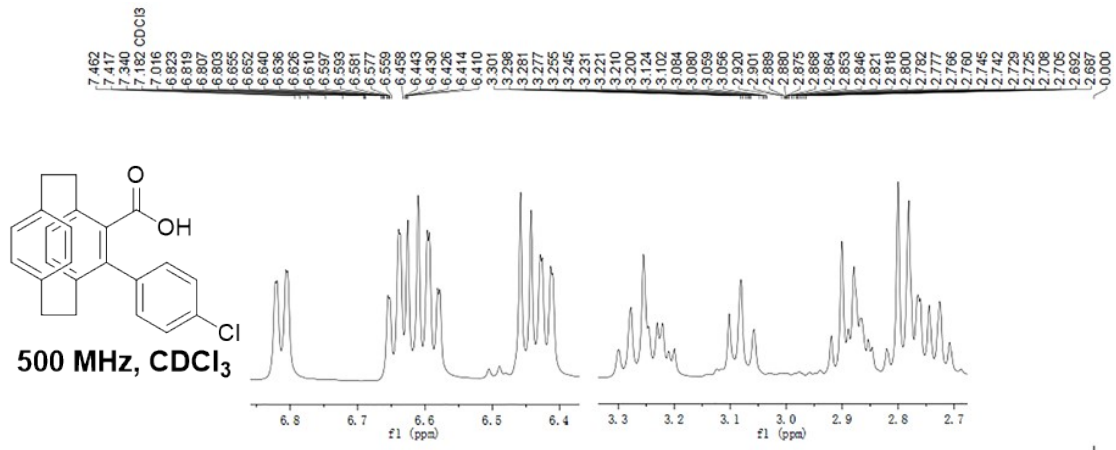
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1.93e4

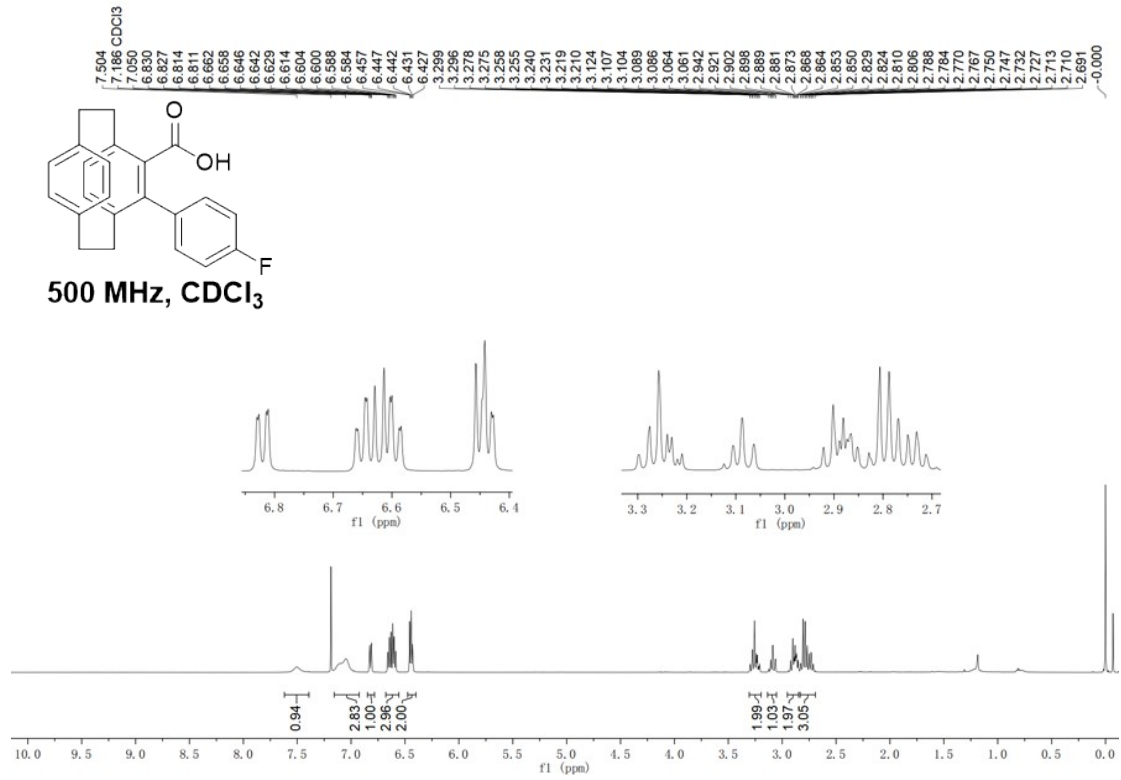
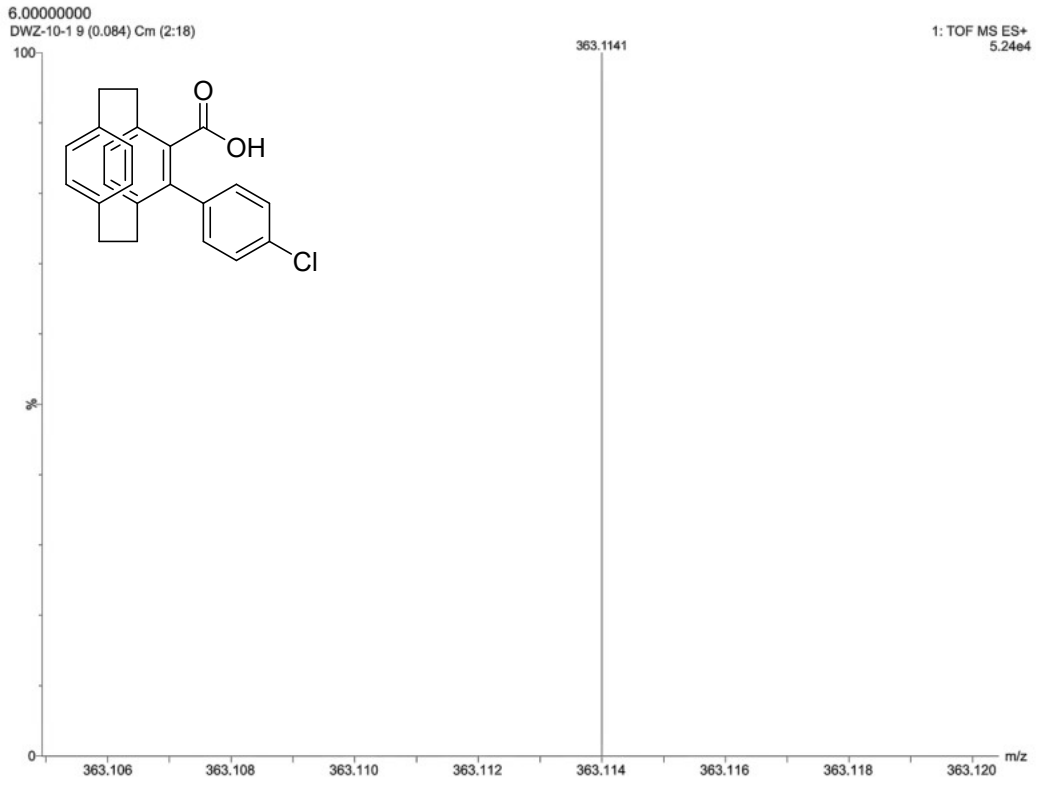


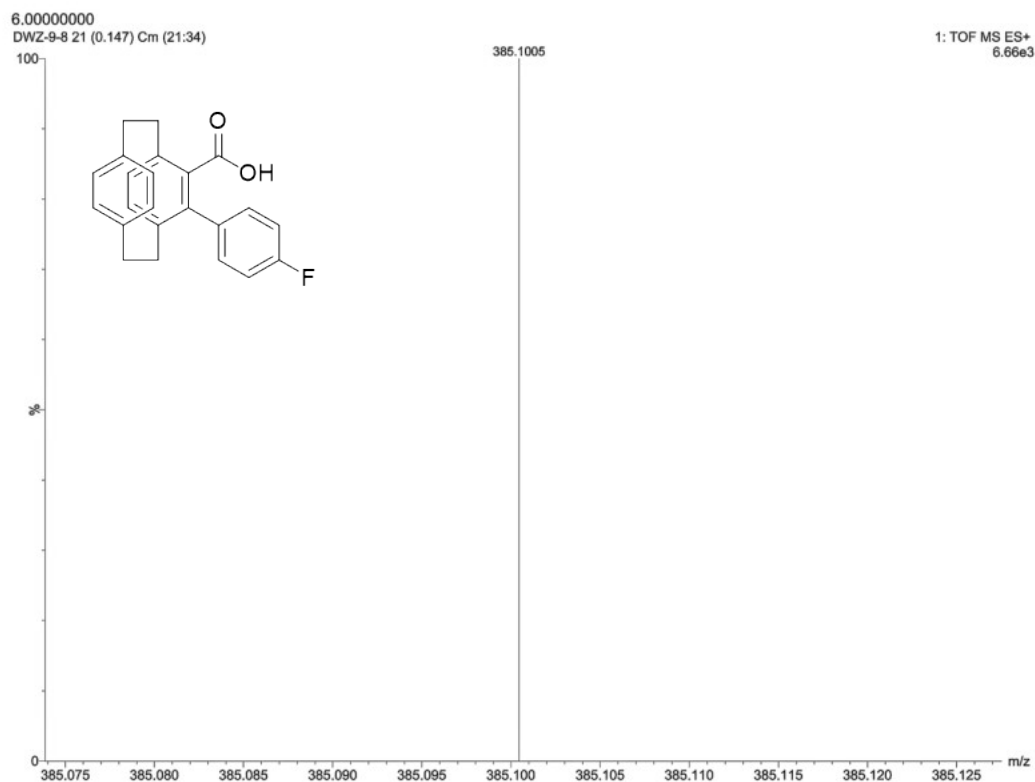
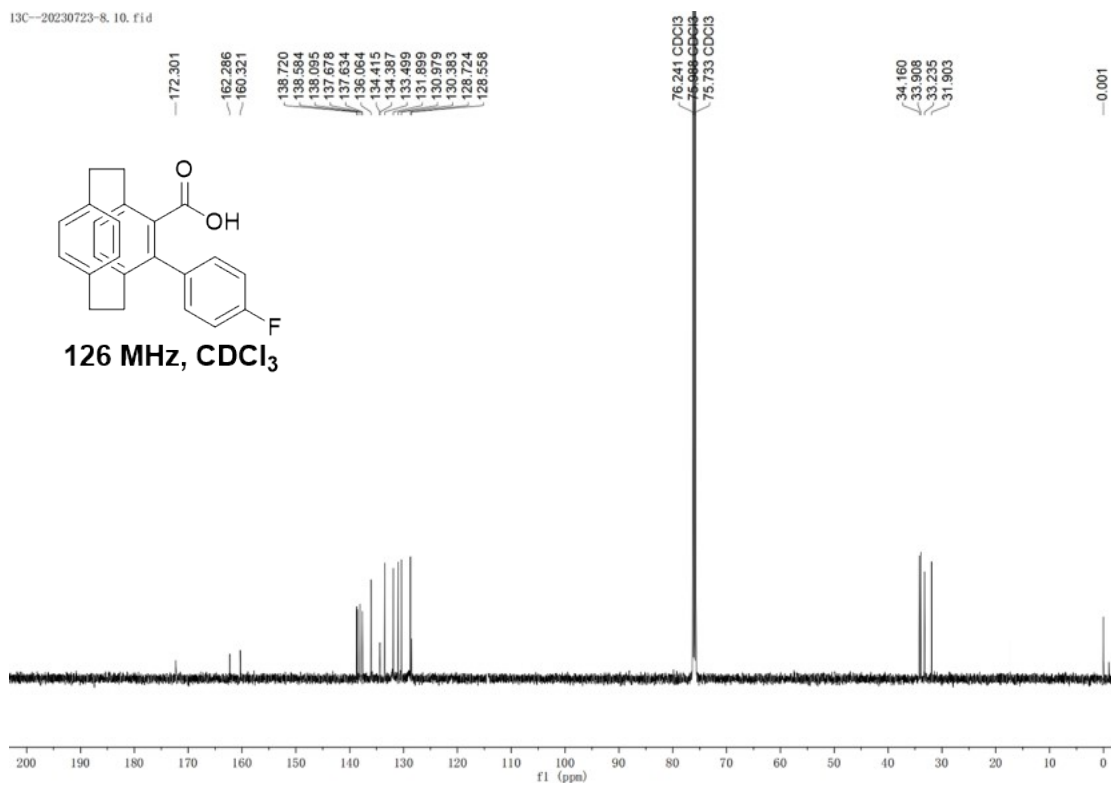
7.617
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7.586 CDCl3
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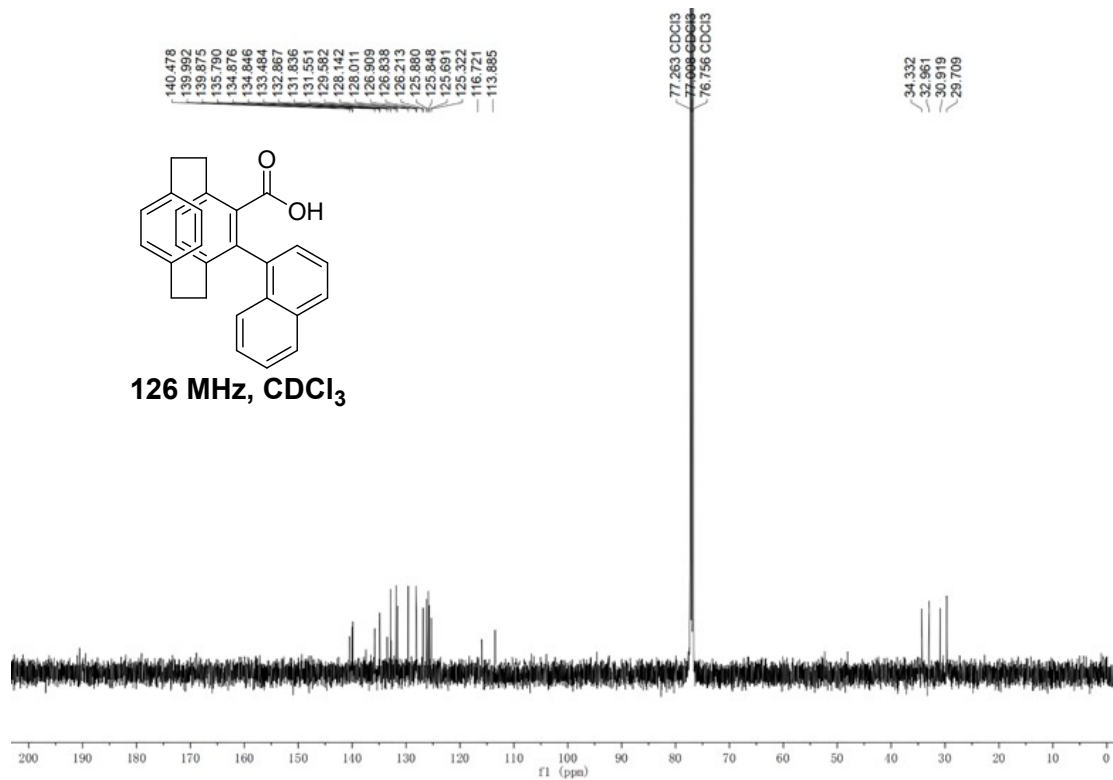
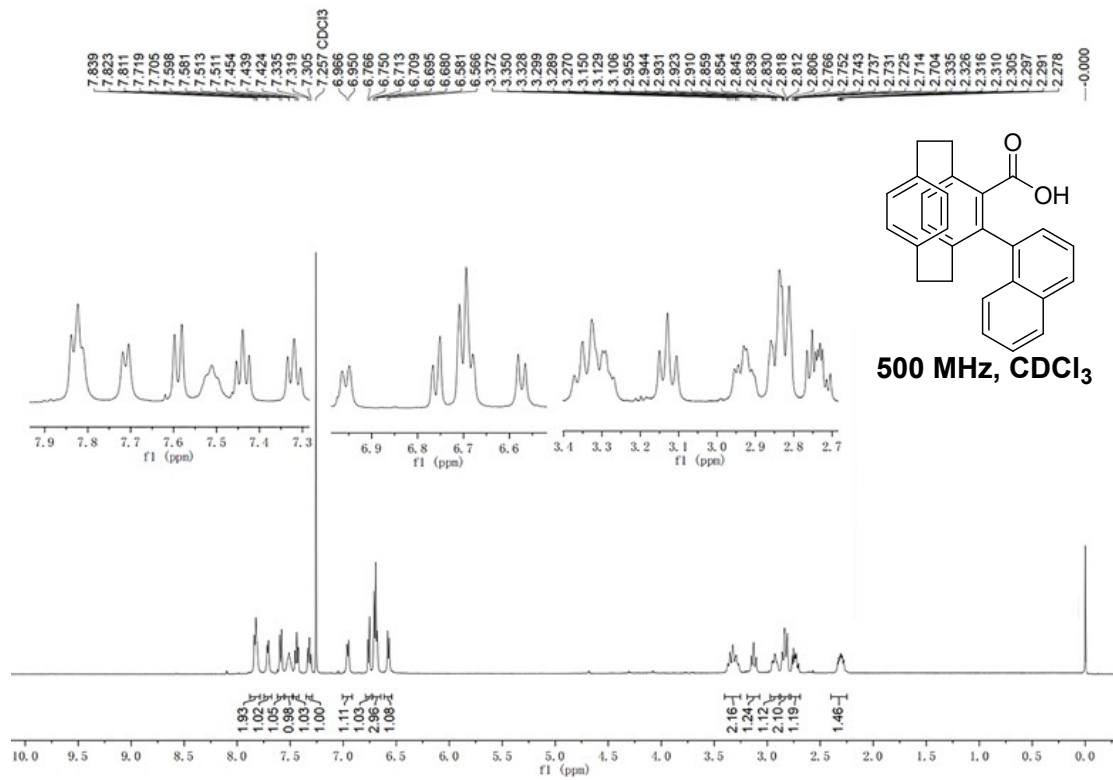






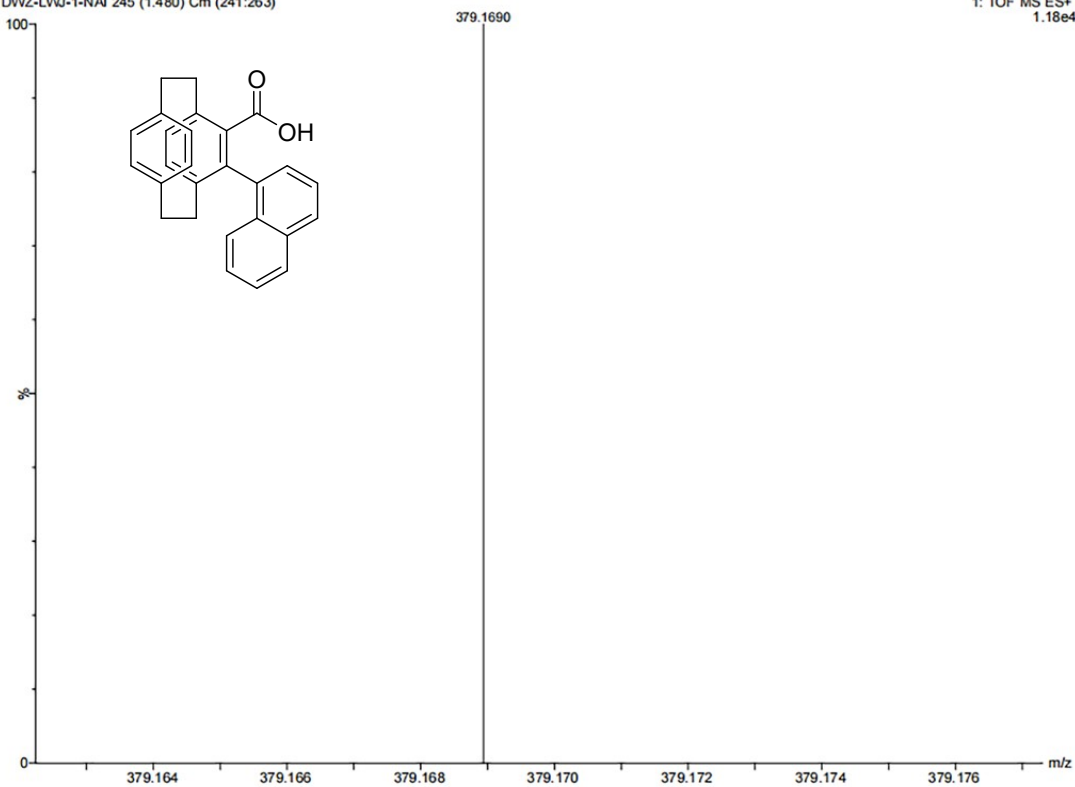




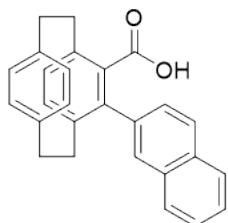


6.00000000
DWZ-LWJ-1-NAI 245 (1.480) Cm (241:263)

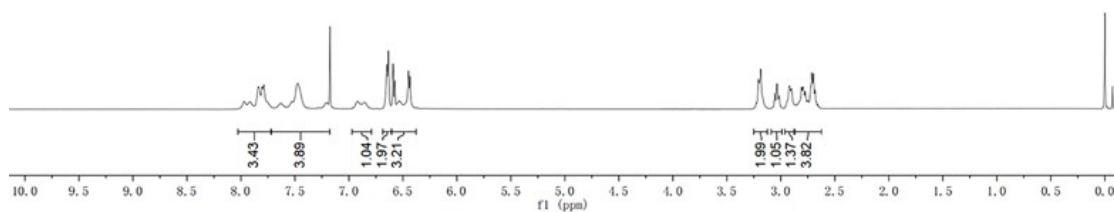
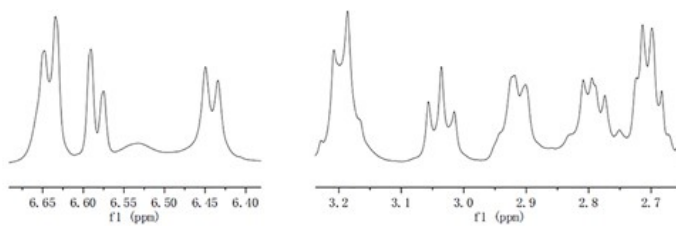
1: TOF MS ES+
1.18e4

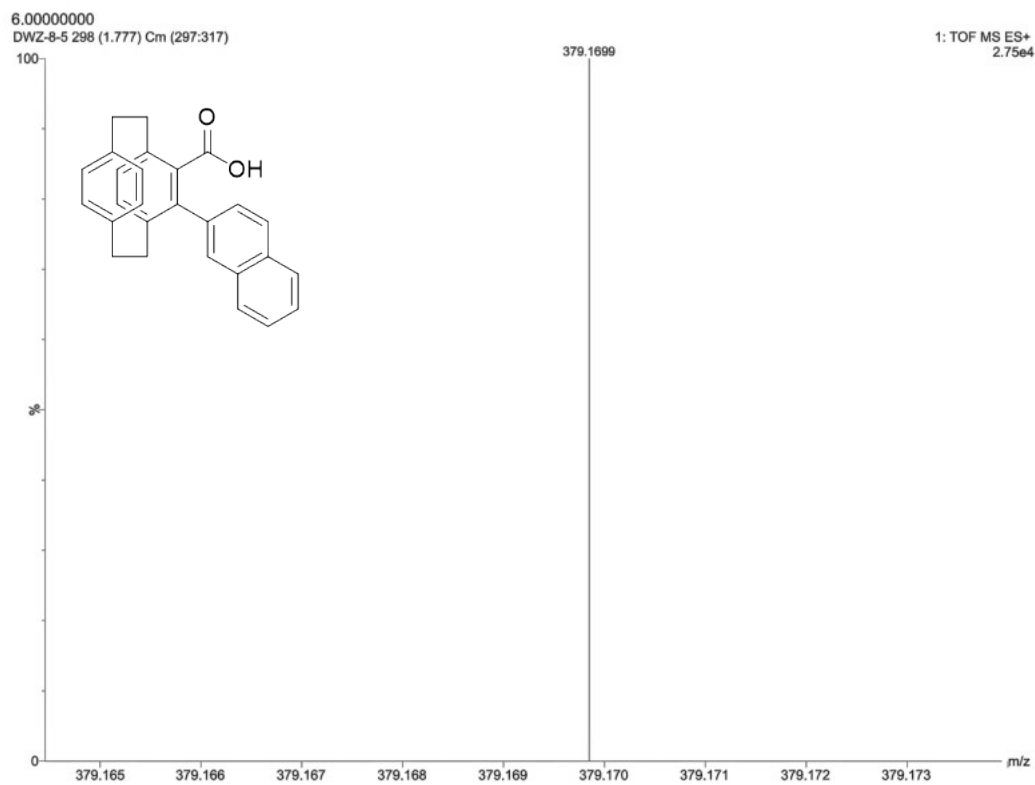
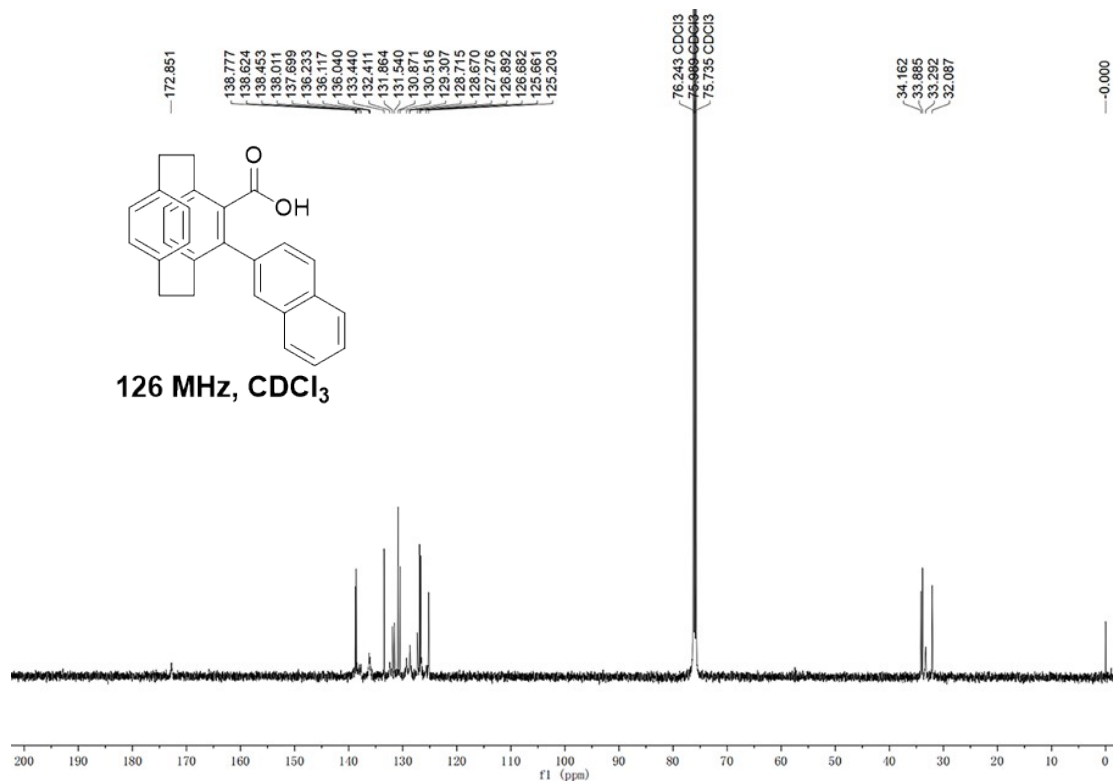


7.971 7.917 7.840 7.826 7.804 7.787 7.631 7.532 7.474 7.209 7.176 CDCl₃ 6.920 6.853 6.851 6.646 6.631 6.593 6.590 6.578 6.574 6.532 6.450 6.434 3.229 3.208 3.186 3.165 3.057 3.036 3.015 2.951 2.943 2.925 2.916 2.900 2.832 2.809 2.795 2.788 2.774 2.750 2.725 2.714 2.688 2.683 2.671 2.656 -0.000

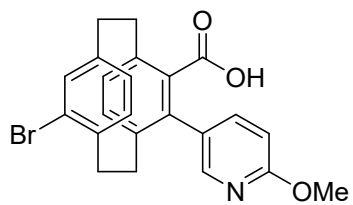
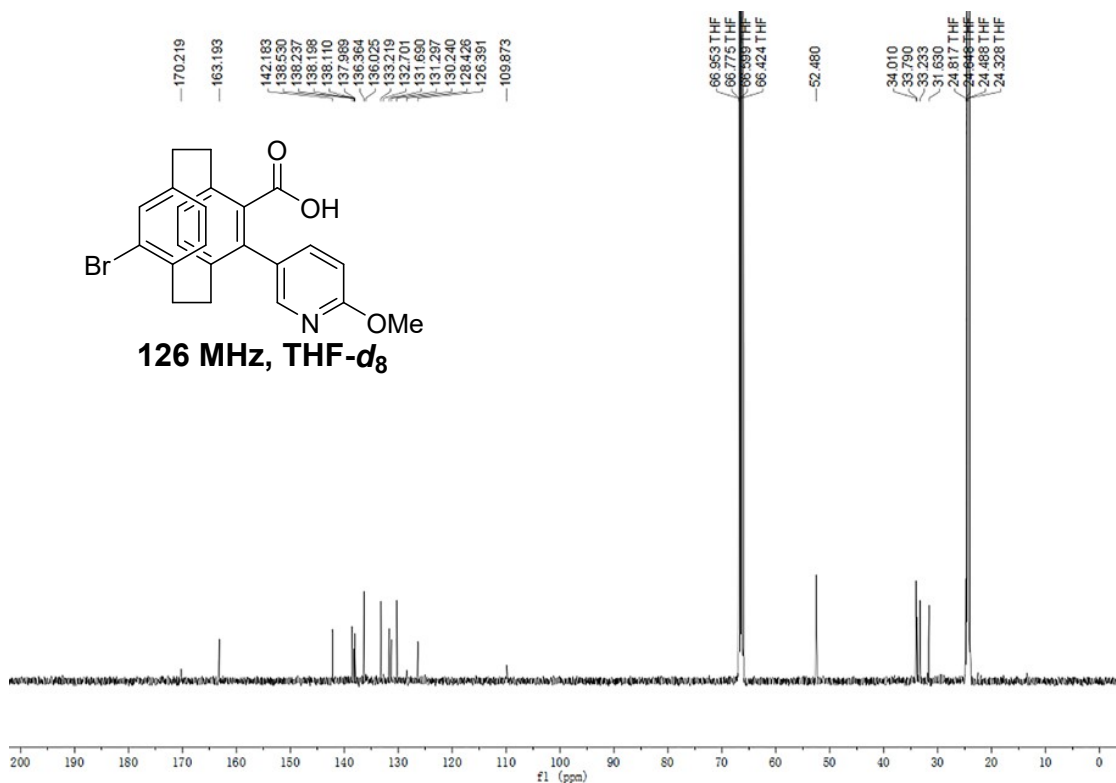
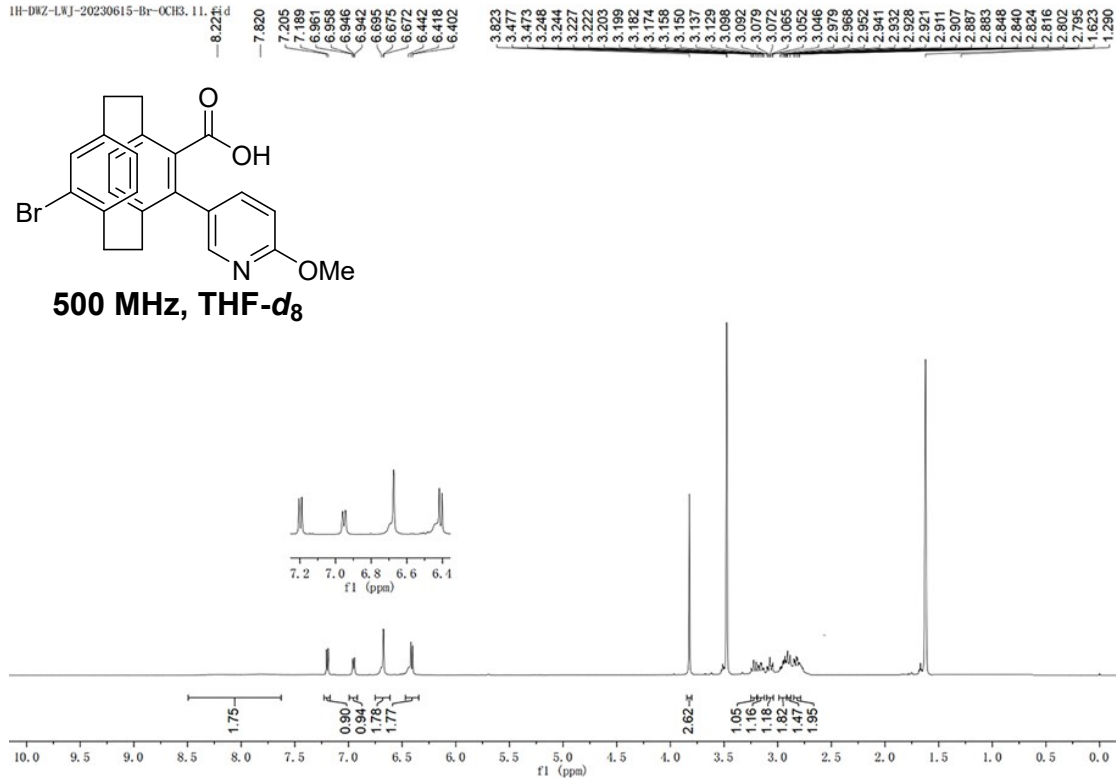


500 MHz, CDCl₃



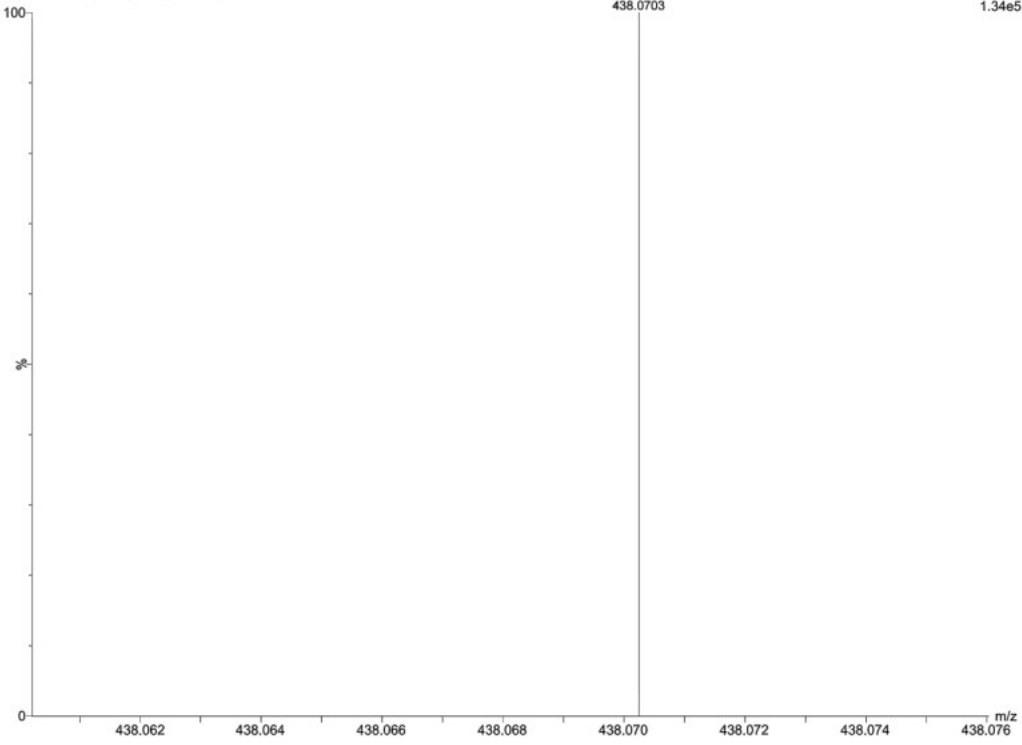


1H-DWZ-LWJ-20230615-Br-OCH3_11.f1.d

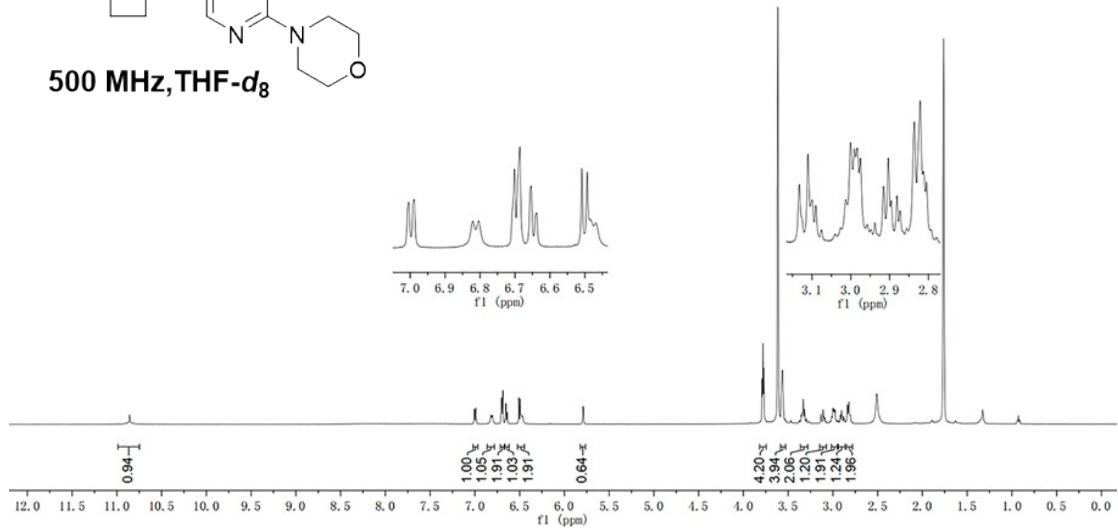
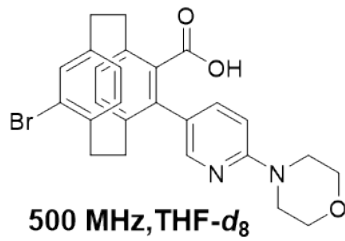


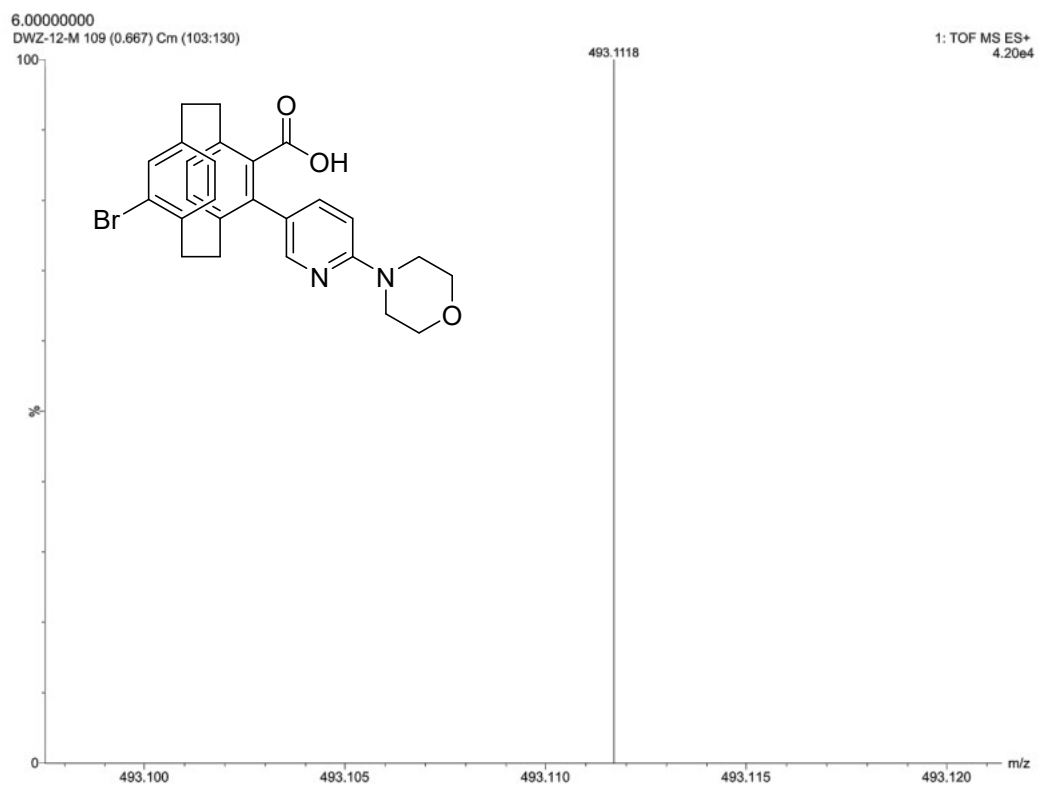
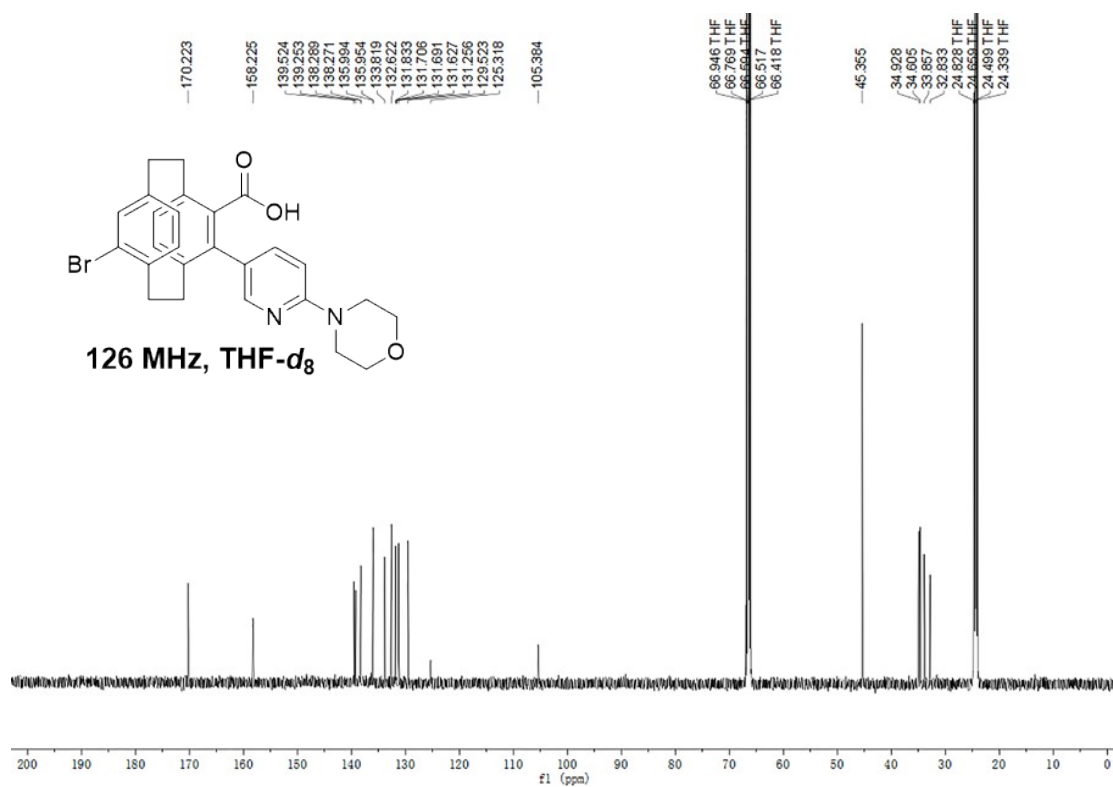
6.0000000
 DWZ-11-0 351 (2.094) Cm (341:362)

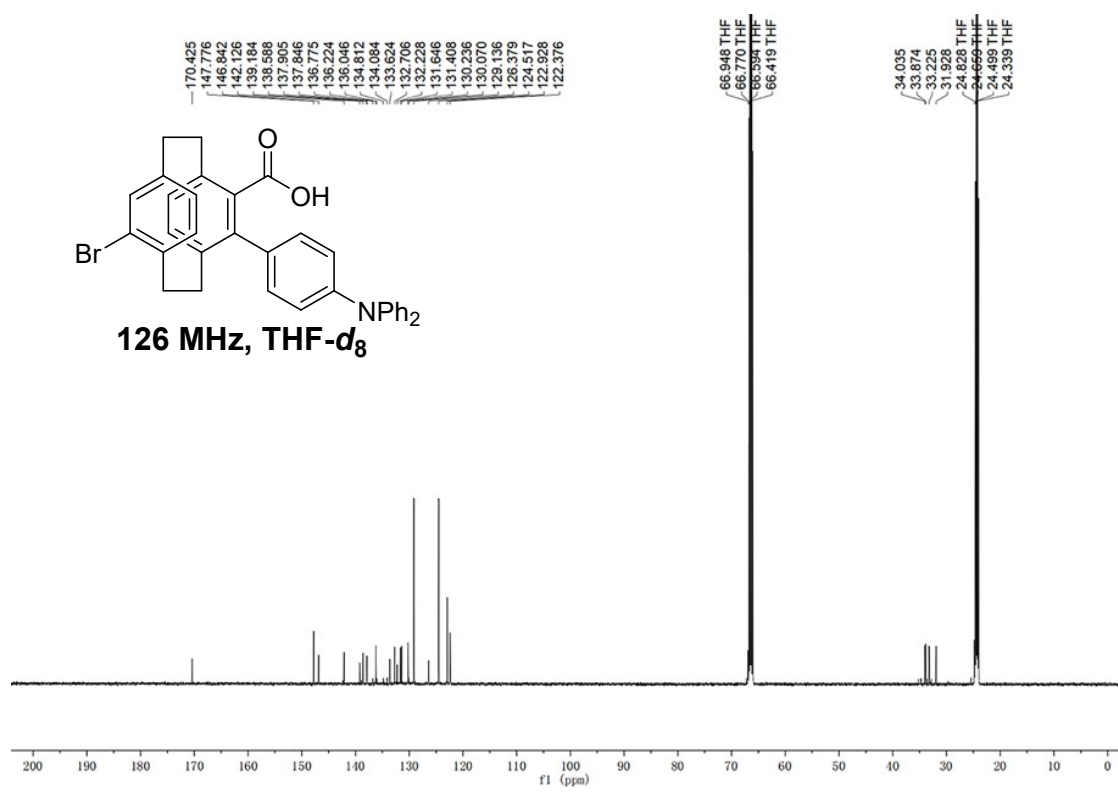
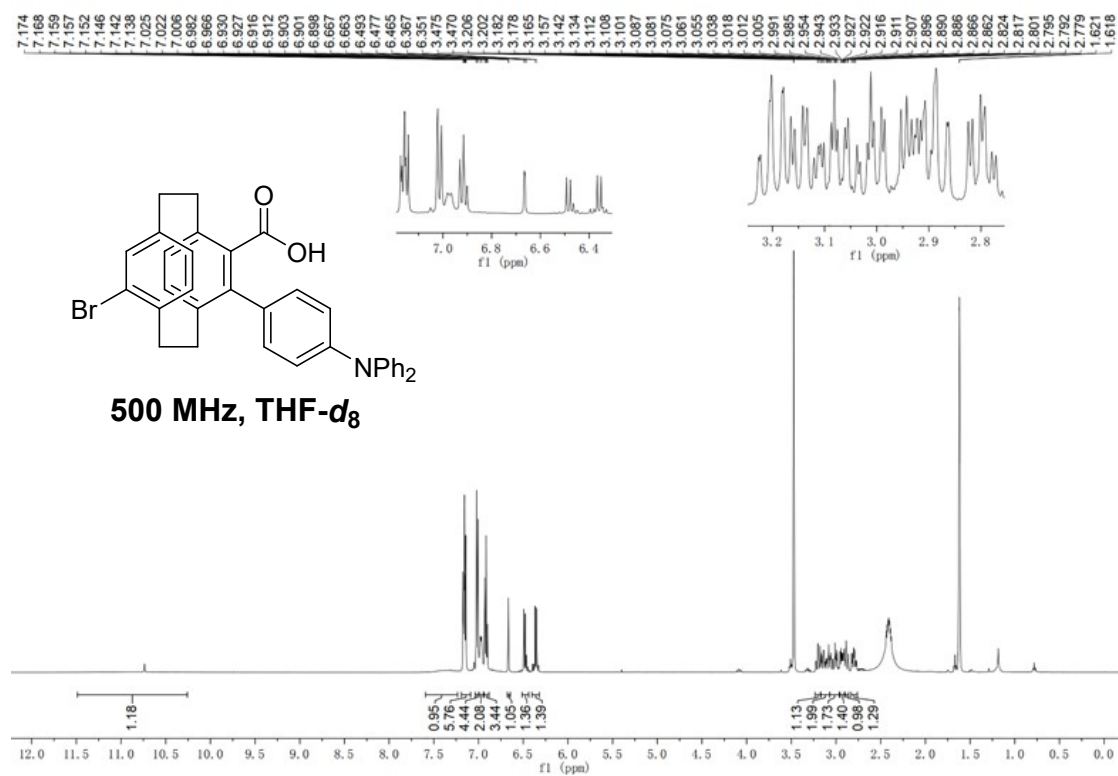
1: TOF MS ES+
 1.34e5

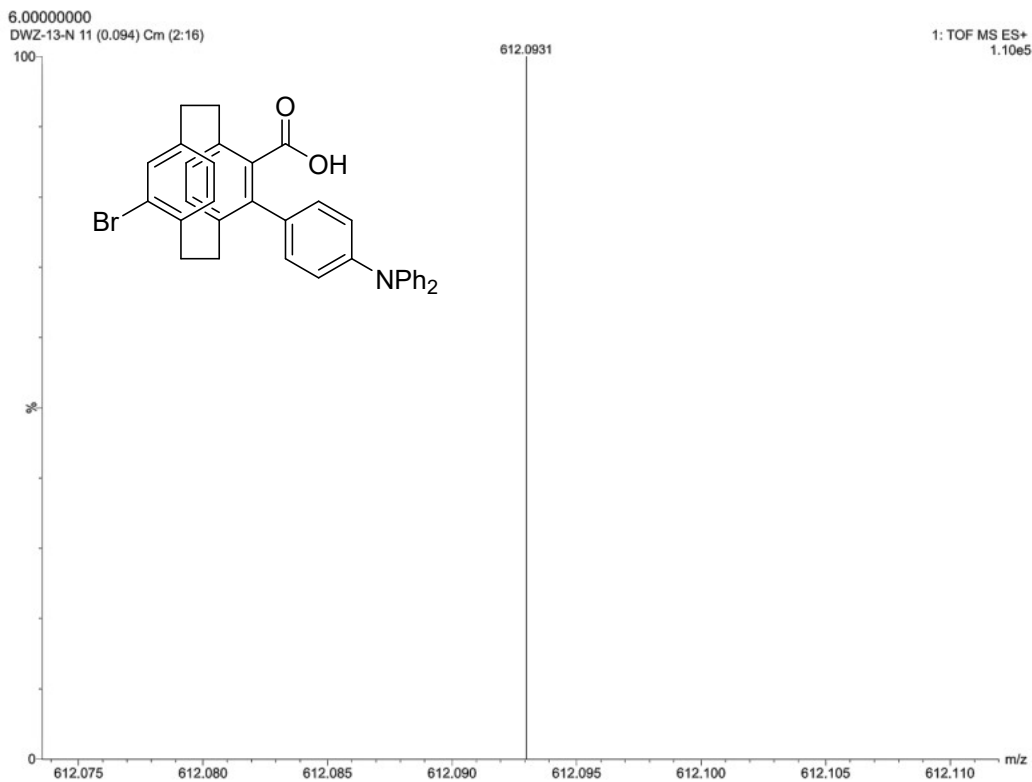


10.857
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 6.638
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 3.562
 3.558
 3.553
 3.374
 3.353
 3.345
 3.339
 3.331
 3.318
 3.296
 3.265
 3.133
 3.126
 3.110
 3.100
 3.090
 3.075
 3.041
 3.027
 3.013
 3.001
 2.981
 2.963
 2.975
 2.957
 2.948
 2.916
 2.903
 2.896
 2.881
 2.873
 2.856
 2.837
 2.827
 2.821
 2.812
 2.805
 2.793
 1.763









9. Coordinates of Optimized Structures

Coordinates of **3a** in the S₀ state

Charge = 0 Multiplicity = 1

N	7.7995	3.3048	15.2587
O	5.3193	6.9979	16.1672
O	5.4026	5.2476	17.561
H	6.1124	5.6136	17.8213
O	9.1012	1.4877	15.8714
C	3.0943	7.942	14.1274
H	2.464	8.6203	13.777
H	3.9514	8.3963	14.3249
C	3.3232	6.8861	13.0744
C	2.2782	6.2849	12.395

H	1.4665	6.7623	12.2695
C	2.3952	5.0083	11.8988
H	1.6644	4.6114	11.4395
C	3.5827	4.291	12.0659
C	4.7066	5.0085	12.4645
H	5.5683	4.6157	12.3886
C	4.5781	6.2891	12.9703
H	5.3513	6.7655	13.2491
C	3.5963	2.7799	12.0952
H	4.469	2.4622	11.7524
H	2.893	2.4428	11.4854
C	3.3609	2.1686	13.5291
H	2.607	1.5285	13.4856
H	4.1691	1.6652	13.8003
C	3.059	3.2091	14.5755
C	1.7929	3.7739	14.6143
H	1.0515	3.2681	14.3028
C	1.5815	5.0466	15.0914
H	0.6982	5.3949	15.1239
C	2.6446	5.8293	15.5267
C	3.8479	5.1475	15.7952
C	4.0653	3.8397	15.332
C	2.5064	7.3262	15.4718
H	2.9801	7.724	16.2448
H	1.5488	7.5661	15.5466

C	4.9304	5.8994	16.5002
C	5.3979	3.1836	15.4822
C	5.5414	1.8943	16.0117
H	4.773	1.408	16.2867
C	6.7879	1.33	16.135
H	6.8937	0.4511	16.4797
C	7.8885	2.0793	15.7424
C	6.5604	3.8354	15.1377
H	6.488	4.7172	14.7918
C	10.2304	2.2243	15.3785
H	10.0499	2.5255	14.4636
H	11.0232	1.6482	15.3823
H	10.3892	3.0021	15.9532

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