

*Supporting Information*

**Light-Driven Methane Conversion: Unveiling Methanol using  
TiO<sub>2</sub>/TiOF<sub>2</sub> Photocatalyst**

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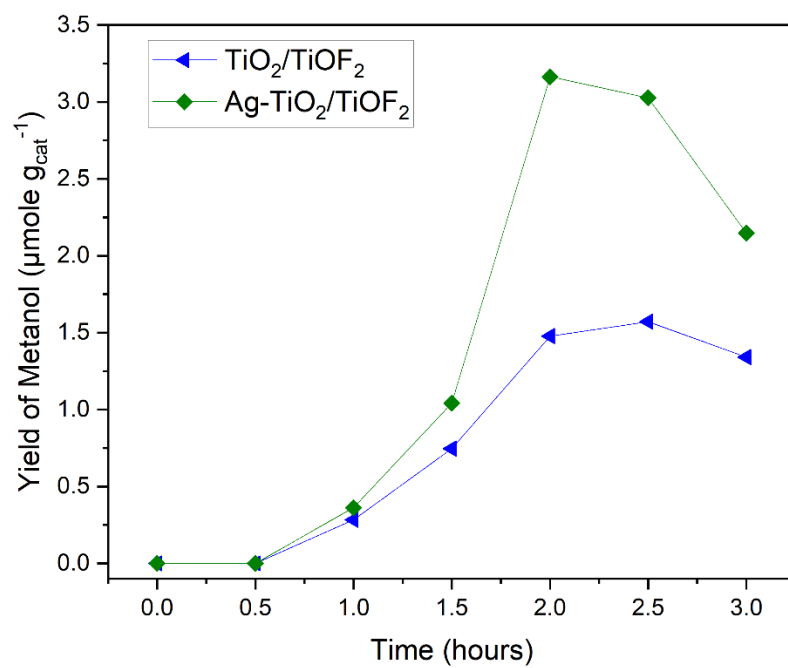
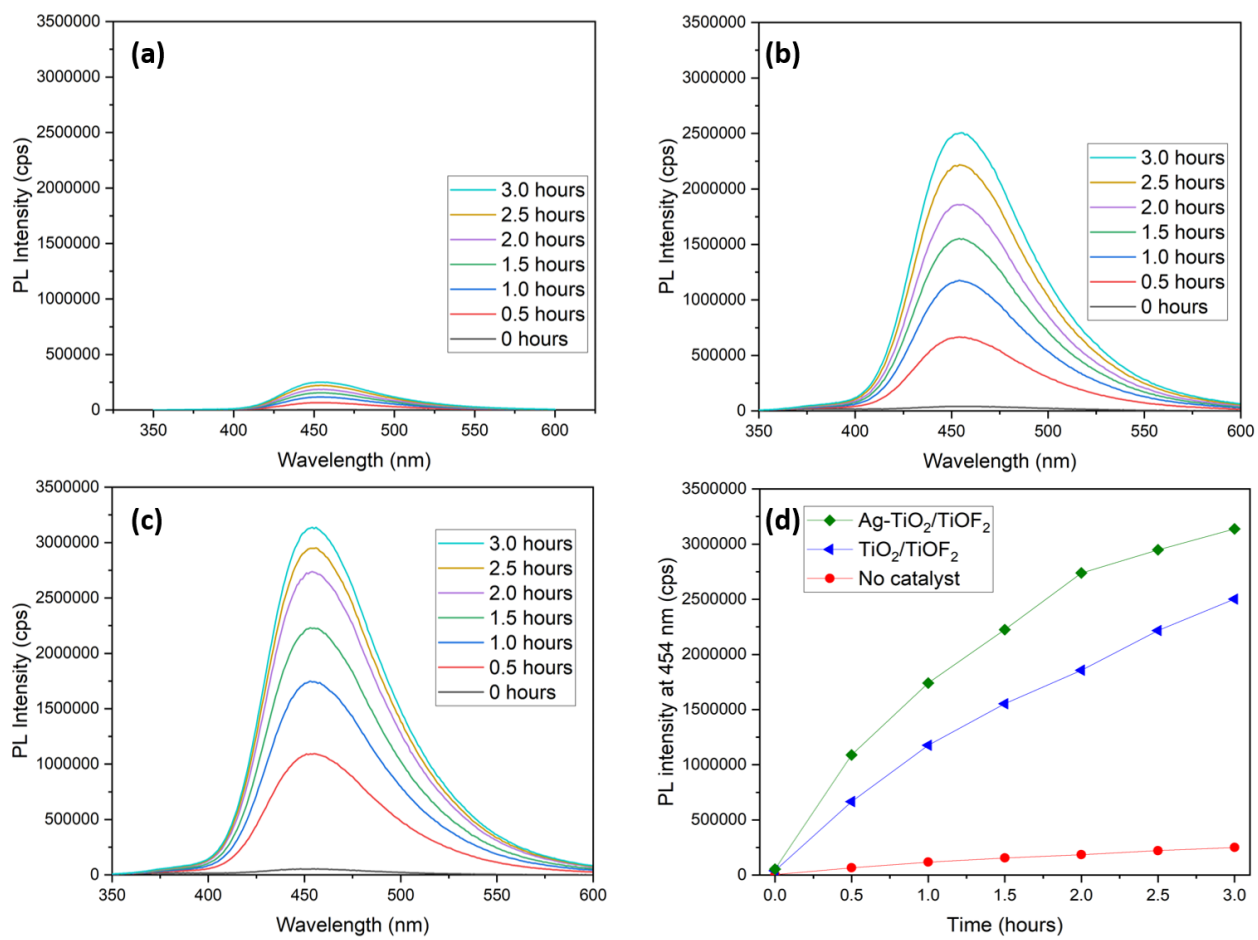


Figure S1. The yield of methanol comparison between  $\text{TiO}_2/\text{TiOF}_2$  and  $\text{Ag-TiO}_2/\text{TiOF}_2$  catalysts



**Figure S2.** Photoluminescence spectra for detection of hydroxyl radicals using coumarin as a probe molecule (a) without catalyst (b)  $\text{TiO}_2/\text{TiOF}_2$  (c)  $\text{Ag-TiO}_2/\text{TiOF}_2$  (d) Photoluminescence signal intensity over time at  $\lambda_{\text{max}} = 454$  nm.

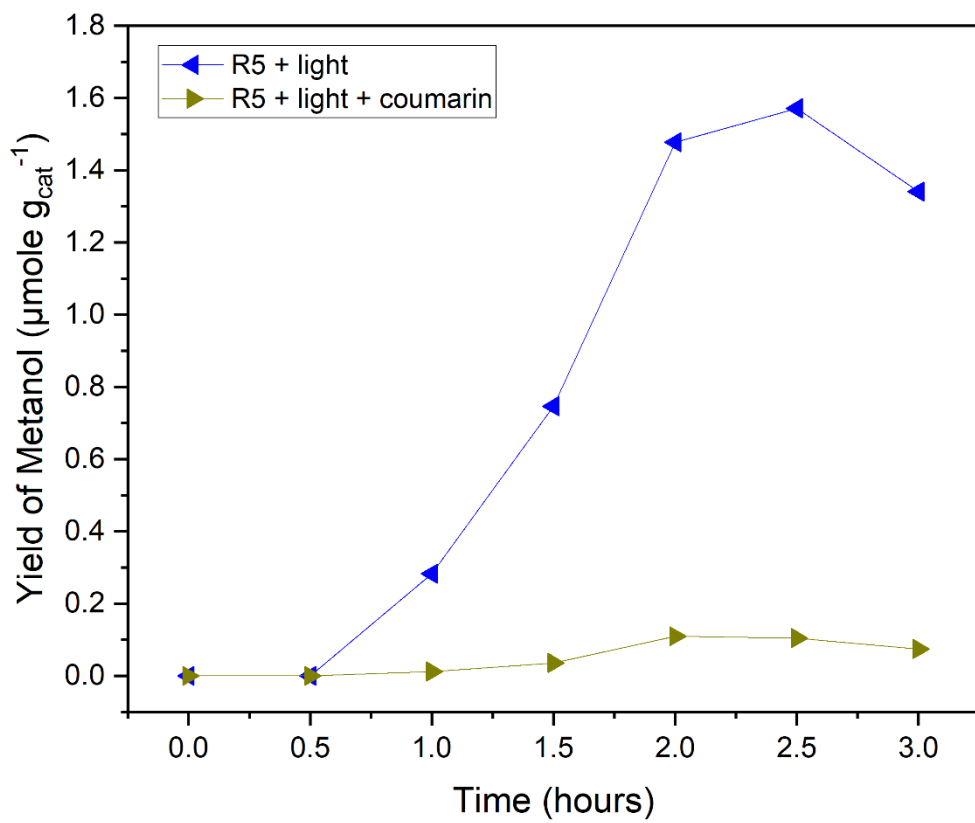


Figure S3. The yield of methanol comparison using  $\text{TiO}_2/\text{TiOF}_2$  (R5) catalysts in the presence and the absence of coumarin as a hydroxyl radical scavenger