·Supplementary Materials for

Three-Dimensional NiCoS Nanotubes@NiCo-LDH Nanosheets Core-Shell Heterostructure for High-Rate Capability Alkaline Zinc-Based Batteries

Linxi Dai, Shangshu Peng, Xinhai Wang, Bo Chen, Yang Wu, Quan Xie, Yunjun Ruan* Institute of Advanced Optoelectronic Materials and Technology, College of Big Data and Information Engineering, Guizhou University, Guiyang 550025, China. *Corresponding author E-mail: yjruan@gzu.edu.cn (Yunjun Ruan)



Figure S1. SEM image of (a)NF and (b)NiCo-LDH nanosheets on Ni foam.



Figure S2. GCD curves of the NiCoS@NiCo-LDH, NiCoS, and NiCo-LDH at 1 A g⁻¹.



Figure S3. (a) CV curves and (b) GCD curves of NiCo-LDH. (c) CV curves and (d) GCD curves of NiCoS.



Figure S4.(a) GCD curves for various current densities of NiCoS//Zn battery.(b) Rate performance and coulombic efficiency of NiCoS//Zn battery

Battery	Area capacity	Reference	
NiCoS@NiCo-LDH//Zn	$0.54 \text{ mAh cm}^{-2}(5 \text{ mA cm}^{-2})$	This work	
FCO//Zn	0.24 mAh cm ⁻² (4 mA cm ⁻²)	1	
COHF//Zn	0.265 mAh cm ⁻² (4 mA cm ⁻²)	2	
CNF@NiCo ₂ S ₄ //Zn	$0.32 \text{ mAh cm}^{-2} (2 \text{ mA cm}^{-2})$	3	
NiCo LDH@Ag NW//Zn	0.12 mAh cm ⁻² (0.1 mA cm ⁻²)	4	
P-NiCo ₂ O _{4-x} //Zn	0.24 mAh cm ⁻² (2 mA cm ⁻²)	5	
Ni@NiO//Zn	0.112 mAh cm ⁻² (4 mA cm ⁻²)	6	
CC-CF@NiO//CC-CF@ZnO	0.39 mAh cm ⁻² (0.5 mA cm ⁻²)	7	

 Table S1. Comparison of area capacity of alkaline Zn-based batteries.

Battery	Electrolyte	Cycle performance	Reference (
NiCoS@NiCo-LDH//Zn	6M KOH + sat. ZnO	95.9% after 3000cycles	This work
Ni-NiO/CC//Zn	6M KOH+0.5M ZnAc ₂	87.5% after 2000cycles	8
NiCo ₂ O ₄ //Zn plate	6M KOH+0.1M ZnAc ₂	63.2% after 1000cycles	9
Ni ₂ P//Zn@CF	1M KOH + 20mM ZnAc ₂	80.0% after 1500cycles	10
NiCo-90//Zn foil	2.5M KOH + sat. ZnO	73.0% after 850cycles	11
Co-Ni ₃ Se ₂ //Zn foil	1M KOH	77.9% after 100cycles	12
FNCP//Zn	1M KOH	90.6% after 2000cycles	13
Al-CoNiDH-5%//Zn	2.5M KOH + sat. ZnO	64.4% after 2000cycles	14
Ni ₃ S ₂ /Ov-Ni(OH) ₂ //Zn	1M KOH + 20mM ZnAc ₂	93.2% after 3000cycles	15
CNF@NiCo ₂ S ₄ //Zn	3M KOH+0.1M ZnAc ₂	83.0% after 2000 cycles	3
Ni ₃ S ₂ @PEDOT//Zn	1M KOH+20mM ZnAc ₂	97.3% after 2000 cycles	16
Co ₃ O ₄ @NiO//Zn@Cu	6М КОН	89% after 500cycles	17
foil			
CC-CF@NiO//CC-	2M KOH + sat. ZnO	72.9% after 2400cycles	7
CF@ZnO			

Table S2. Cycling performance of different alkaline Zn-based batteries.

Battery	Electrolyte	Energy density /Wh kg ⁻¹	Power density /kW kg ⁻¹	Reference
NiCoS@NiCo-LDH//Zn	6M KOH + sat. ZnO	435.3	4.1	This work
Ni ₃ S ₂ /OV-Ni(OH) ₂ //Zn	1M KOH+20mM ZnAc ₂	384.6	1.73	15
Ni ₁₂ P ₅ //Zn	1M K ₂ CO ₃ +2M kF+4M KOH + sat. ZnO	287.9	5.1	18
NCS@NCH//Zn	2M KOH+0.02M Zn(CH ₃ COO) ₂ ·2H ₂ O	194.2	0.72	19
Ni/NiO-BCF//Zn	6M KOH + 0.5mM Zn(Ac) ₂	313.4	0.66	20
Ni(OH) ₂ /CNFs//Zn	6M KOH+1M LiOH, and PAAS saturated	325	1.23	21
	with ZnO gel.			
R-Co ₃ O ₄ //Zn	6M KOH saturated with $Zn(Ac)_2$	295.5	0.84	22
NiCo ₂ O ₄ //Zn	1M KOH and 20mM Zn(Ac) ₂	248.3	2.2	23
Ni ₃ S ₂ @PANI//Zn	$6M \text{ KOH} + 0.2M \text{ Zn}(\text{CH}_3\text{COO})_2$	308	6.9	24
CC-CF@NiO//CC-	2M KOH + sat. ZnO	355.7	0.46	7
CF@ZnO				
Co ₃ O ₄ @NiO//Zn	6M KOH sat. ZnO	215.5	3.45	17

Table	S 3.	Electroc	chemical	performance	e of differ	ent alkaline	Zn-based	batteries.
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