

Electronic Supplementary Information

Highly electron-deficient 1-propyl-3,5-dinitropyridinium: evaluation of electron-accepting ability and application as an oxidative quencher for metal complexes

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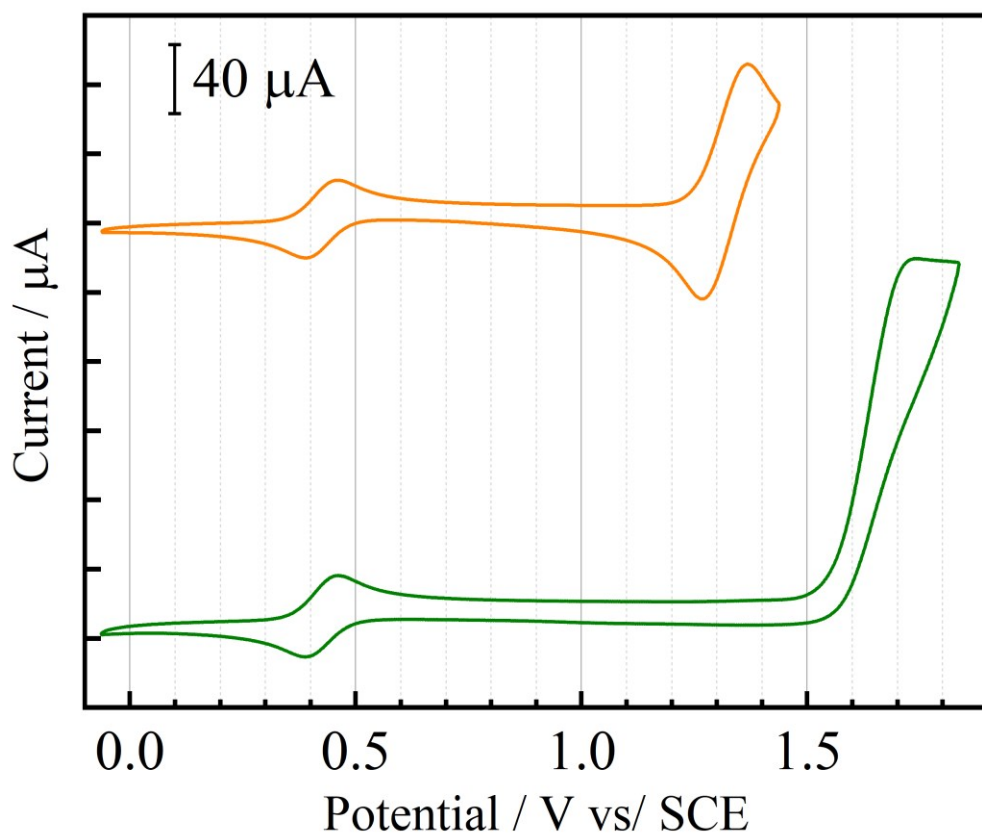


Fig. S1 Cyclic voltammograms of $[\text{Ru}(\text{bpy})_3](\text{PF}_6)_2$ (orange) and $[\text{Ir}(\text{ppy})_2(\text{bpy})]\text{PF}_6$ (green) in deaerated CH_3CN containing 0.1-M TBAPF_6 . Reversible waves at ~ 0.43 V represent redox couples of ferrocene as an internal standard.

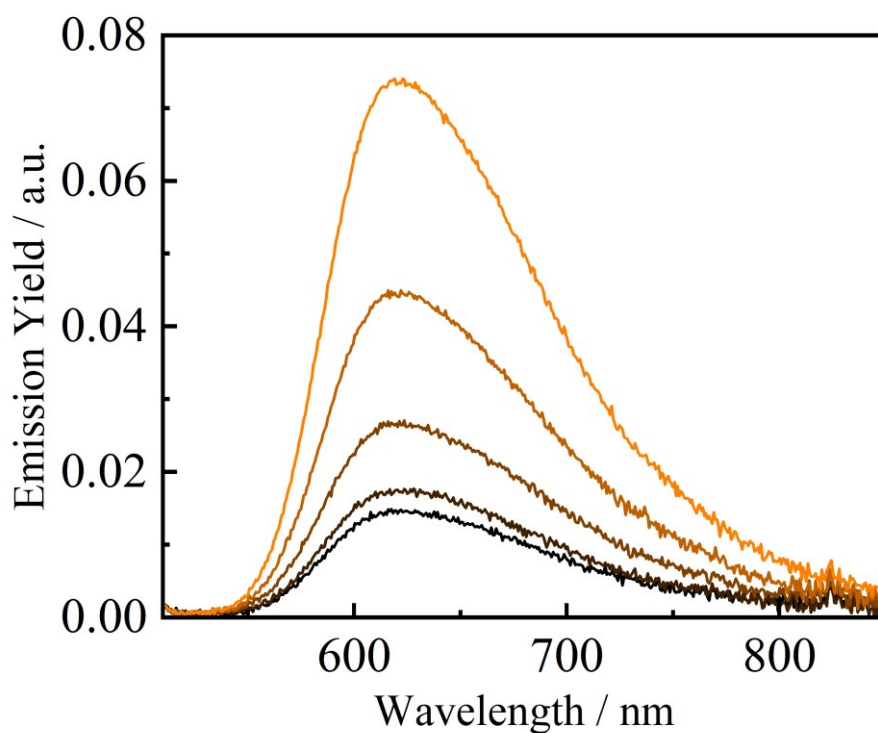


Fig. S2 Emission spectra of $[\text{Ru}(\text{bpy})_3](\text{PF}_6)_2$ (3.8×10^{-5} M, $\lambda_{\text{ex}} = 500$ nm) in the absence and presence of **5**·OTf ($(0.0\text{--}4.0) \times 10^{-3}$ M: orange \rightarrow black) in deaerated CH_3CN .

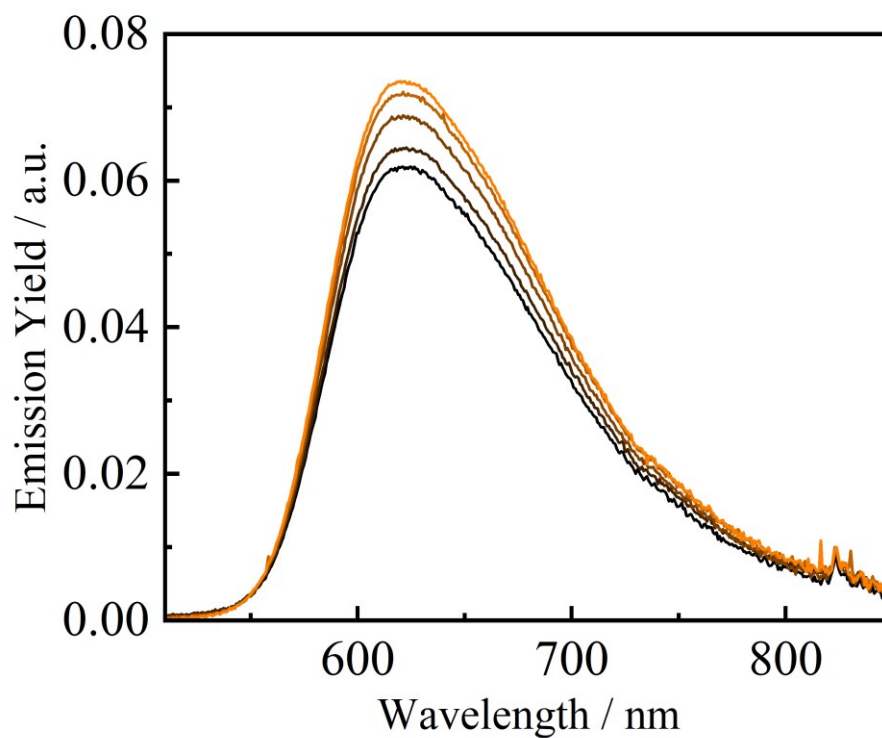


Fig. S3 Emission spectra of $[\text{Ru}(\text{bpy})_3](\text{PF}_6)_2$ (3.8×10^{-5} M, $\lambda_{\text{ex}} = 500$ nm) in the absence and presence of **6**·OTf ($(0.0\text{--}4.0) \times 10^{-3}$ M: orange \rightarrow black) in deaerated CH_3CN .

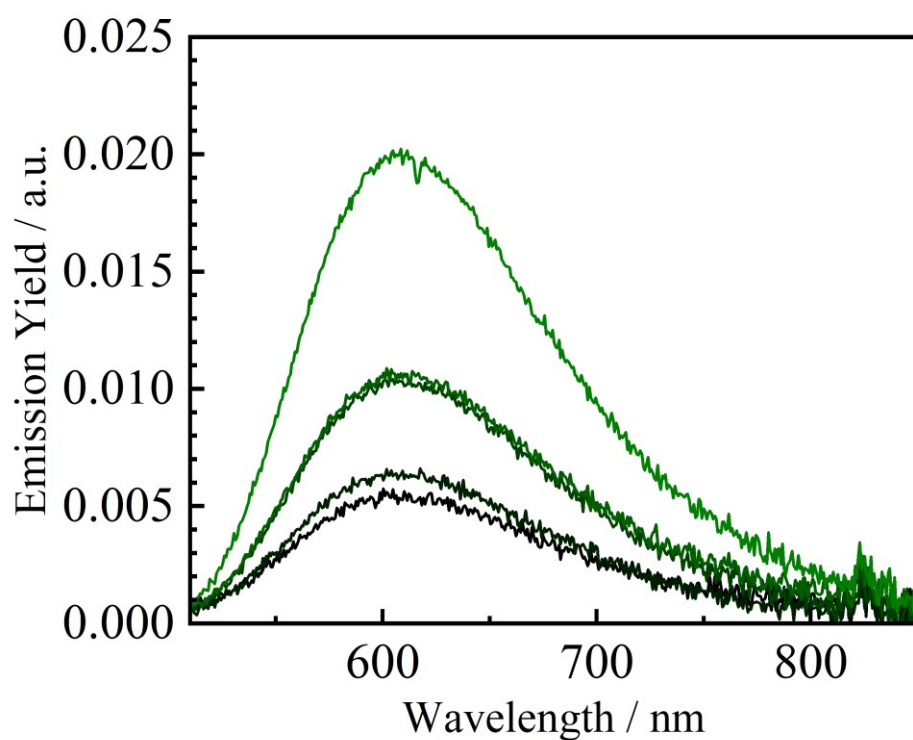


Fig. S4 Emission spectra of [Ir(ppy)₂(bpy)]PF₆ (3.8×10^{-6} M, $\lambda_{\text{ex}} = 470$ nm) in the absence and presence of 5·OTf ($(0.0\text{--}8.0) \times 10^{-3}$ M: orange \rightarrow black) in deaerated CH₃CN.

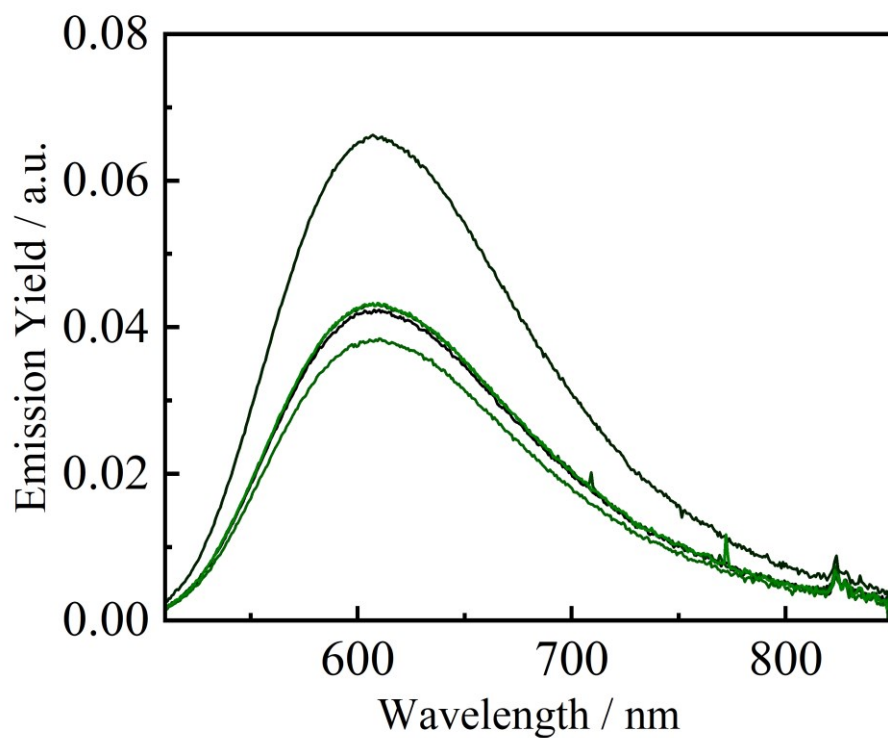


Fig. S5 Emission spectra of [Ir(ppy)₂(bpy)]PF₆ (3.8×10^{-6} M, $\lambda_{\text{ex}} = 470$ nm) in the absence and presence of 6·OTf ($(0.0\text{--}4.0) \times 10^{-3}$ M: orange \rightarrow black) in deaerated CH₃CN.