Support Information for:
Exploring the thermal decomposition and detonation mechanisms of
2,4-dinitroanisole by TG-FTIR-MS and molecular simulations
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23 1. Validation of the ReaxFF/lg description

Parameters	Exp	Reaxff-lg	Deviation/%
a/Å	8.77 (14.91)	8.66 (14.82)	-1.26 (-0.60)
b/Å	12.65 (6.03)	12.48 (6.00)	-1.27 (-0.50)
c/Å	15.43 (19.68)	15.23 (19.57)	-1.27 (-0.56)
β/°	81.89 (-)	81.89 (-)	0.00 (-)
$\rho/g \cdot cm^{-3}$	1.56 (1.70)	1.61 (1.73)	3.21 (1.76)

24 **Table S1** Experiment and computed lattice parameters of DNAN and TNT. The data

outside and in the parenthesis are for DNAN and TNT, respectively.

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Table S2 Experiment and computed bond lengths of DNAN and TNT. The data

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outside and in the parenthesis are for DNAN and TNT, respectively.

Bond/Å	Exp	Reaxff-lg	DFT
C-N	1.47 (1.41)	1.52 (1.52)	1.48 (1.49)
C-C	1.38 (1.39)	1.36 (1.43)	1.39 (1.39)
C-C _{CH3}	- (1.50)	- (1.53)	- (1.51)
C-O	1.39 (-)	1.44 (-)	1.38 (-)
N-O	1.20 (1.23)	1.28 (1.28)	1.22 (1.22)
C _{ph} -H	1.09 (0.97)	1.13 (1.13)	1.08 (1.08)
С _{СН3} -Н	1.09 (0.99)	1.12 (1.12)	1.09 (1.09)

30 2. Detonation of DNAN and TNT during fast heating



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32 Fig. S1 The potential energy (PE) evolution of DNAN (blue) and TNT (red) crystals

upon fast heating from 300 to 2500 K in 2 ps.





Fig. S2 Snapshots of bimolecular H (a) and O (b) transfer of TNT in condensed

36 phase. The C, H, O, and N atoms are colored in gray, white, red and blue,

respectively.

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39 **3. Detonation of DNAN during slow heating**



41 Fig. S3 The evolution of the numbers of direct bonding between C, H, O and N atoms

42 during the decomposition of DNAN in condensed phase at 2500 K (a) and 3500 K (b).





Fig. S4 The snapshots and initial time for the four types of bimolecular reactions
during the initial decomposition of DNAN in condensed phase at 2500K. P6 and P8
are the A+A type oxygen transfers, P9 is the A+B type oxygen transfer, while P10 is
the A+B type hydrogen transfer (see Table 2 in the text). The C, H, O, and N atoms
are colored in gray, white, red and blue, respectively.



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Fig. S5 The formation of NO from a nucleophilic attack at the phenyl ring by NO₂
 during the initial decomposition of the DNAN crystal at 2500 K. The C, H, O, and N
 atoms are colored in gray, white, red and blue, respectively.





62 Fig. S7 Three carbon clusters formed during the intermediate decomposition stage of

- 63 the DNAN crystal at 2500 K. The C, H, O, and N atoms are colored in gray, white,
 - red and blue, respectively.

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Fig. S8 The final gas products (a) and carbon clusters (b) formed from the thermal
decomposition of the DNAN crystal at 2500 K. The C, H, O, and N atoms are colored
in gray, white, red and blue, respectively.

4. Key reactions of DNAN and TNT during thermal decomposition



82 and detonation

Fig. S9 The structures of the reactant dimer, transition state and product of the reactions in Fig. 11 (b) of the text: (a) NO₂-O transfer, (b) DNAN-O transfer, (c) TNT- α -H transfer, (d) DNAN-H transfer. The C, H, O, and N atoms are colored in gray, white, red and blue, respectively.