

Supporting Information

Tungsten doped FeCoP₂ nanoparticles Embedded into Carbon for Highly Efficient Oxygen Evolution Reaction

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Number of pages: 12 (page S1-S12)

Number of tables: 5 (Table S1-S5)

Number of figures: 8 (Figure S1-S8)

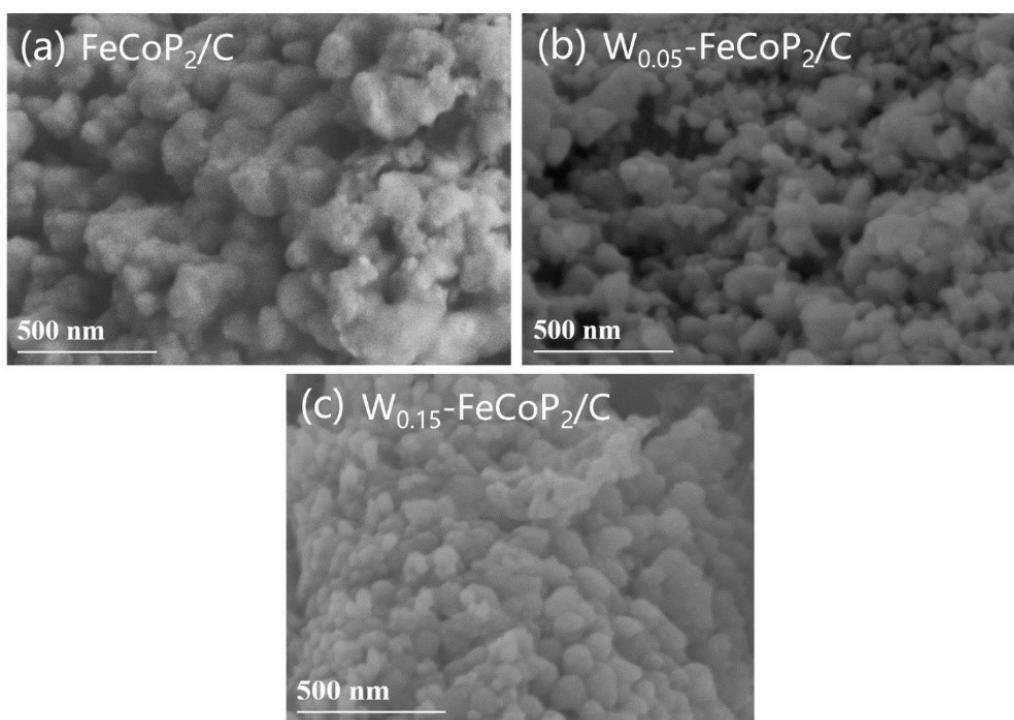


Figure S1. FESEM images of (a) FeCoP_2/C , (b) $\text{W}_{0.05}\text{-FeCoP}_2/\text{C}$ and (c) $\text{W}_{0.15}\text{-FeCoP}_2/\text{C}$

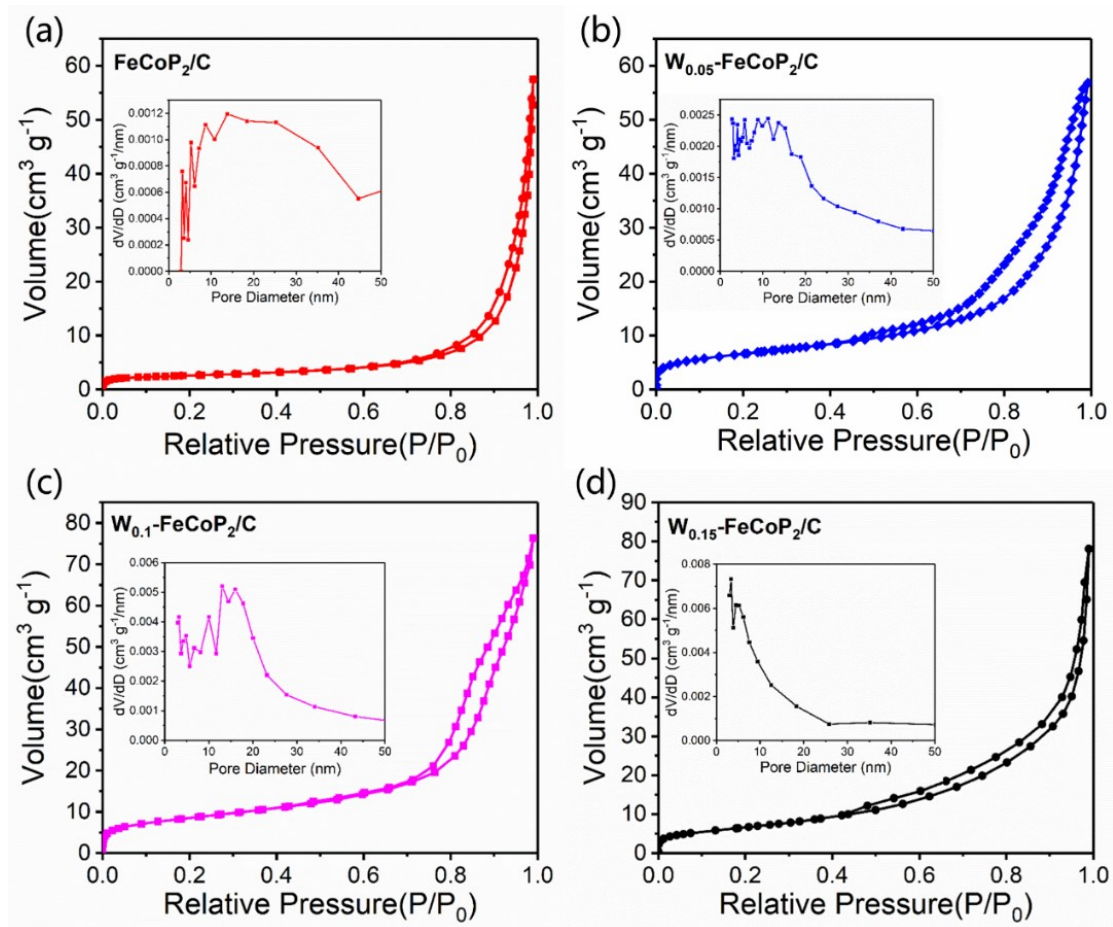


Figure S2. N₂ adsorption-desorption isotherms of (a) FeCoP₂/C, (b) W_{0.05}-FeCoP₂/C, (c) W_{0.1}-FeCoP₂/C, and (d) W_{0.15}-FeCoP₂/C

Table S1. The surface areas and pore volume of FeCoP₂/C, W_{0.05}-FeCoP₂/C, W_{0.1}-FeCoP₂/C and W_{0.15}-FeCoP₂/C

Samples	Surface area (m ² g ⁻¹)	Micropore volume (cm ³ g ⁻¹)	Mesopore volume (cm ³ g ⁻¹)
FeCoP ₂ /C	22.81	0.0095	0.074
W _{0.05} -FeCoP ₂ /C	28.22	0.0104	0.087
W _{0.1} -FeCoP ₂ /C	36.94	0.0124	0.121
W _{0.15} -FeCoP ₂ /C	30.67	0.0108	0.115

Table S2. The normalized atomic ratios of Fe, Co and W elements in $W_x\text{-FeCoP}_2/\text{C}$ measured by ICP-OES

Samples	Fe (Atomic)	Co (Atomic)	W (Atomic)	Fe: Co: W (Atomic)
$W_{0.05}\text{-FeCoP}_2/\text{C}$	4.22	4.32	0.19	1.11:1.14:0.05
$W_{0.1}\text{-FeCoP}_2/\text{C}$	3.87	3.90	0.36	1.08:1.09:0.1
$W_{0.15}\text{-FeCoP}_2/\text{C}$	4.11	3.89	0.51	1.21:1.14:0.15

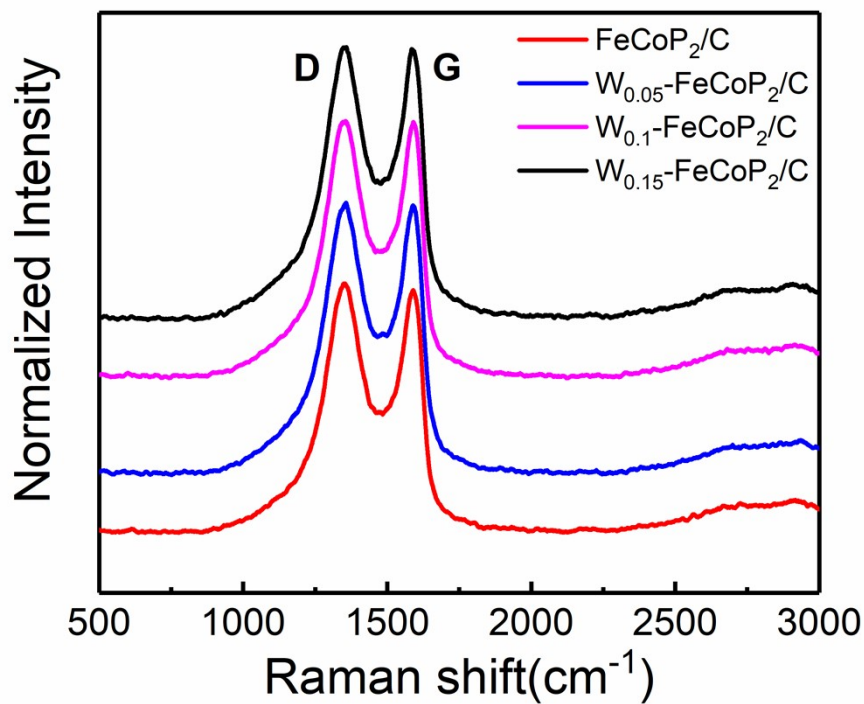


Figure S3. Raman spectra of FeCoP₂/C , W_{0.05}-FeCoP₂/C, W_{0.1}-FeCoP₂/C and W_{0.15}-FeCoP₂/C

Table S3. The ratios of I_D/I_G in different samples measured by Raman spectra

Samples	I _D /I _G
FeCoP ₂ /C	1.018
W _{0.05} -FeCoP ₂ /C	1.010
W _{0.1} -FeCoP ₂ /C	1.002
W _{0.15} -FeCoP ₂ /C	1.008

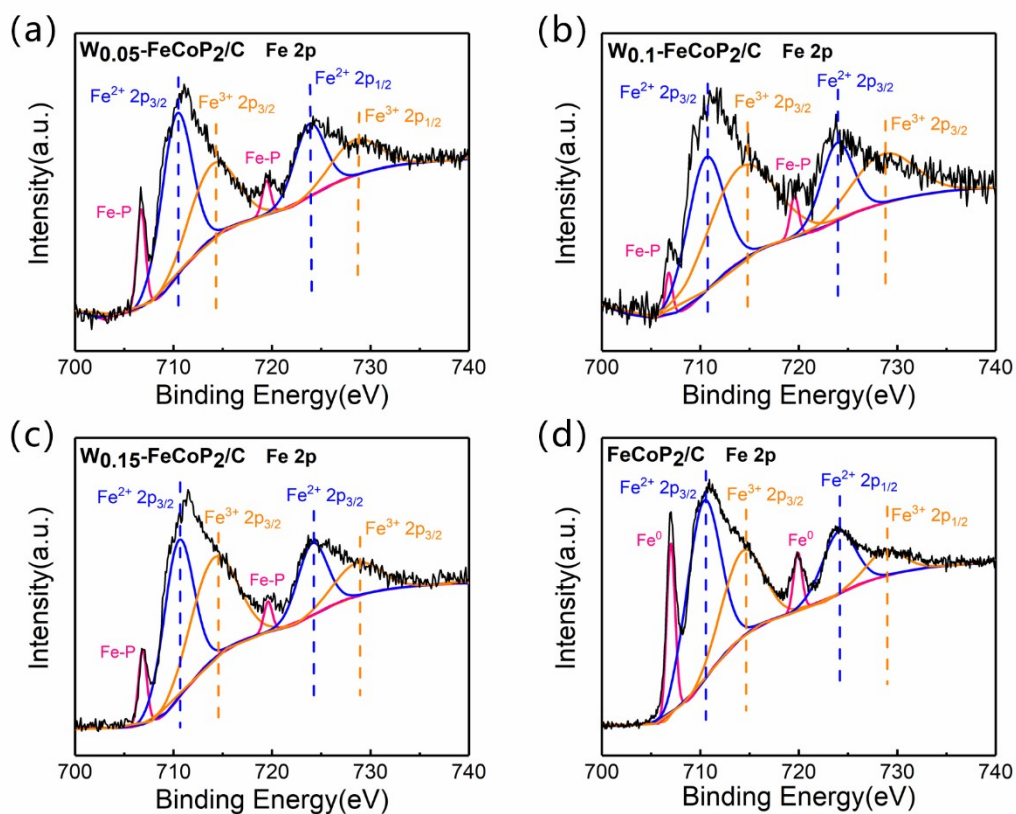


Figure S4. Fe 2p XPS spectra of FeCoP₂/C , W_{0.05}-FeCoP₂/C, W_{0.1}-FeCoP₂/C and W_{0.15}-FeCoP₂/C

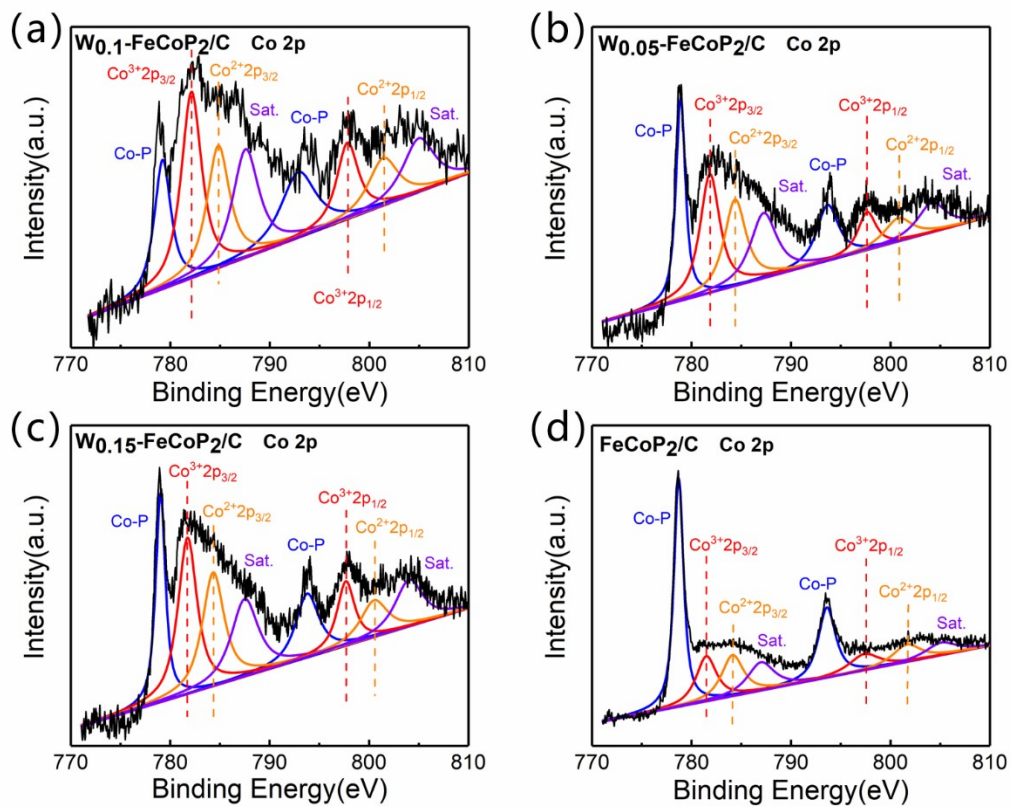


Figure S5. Co 2p XPS spectra of FeCoP₂/C, W_{0.05}-FeCoP₂/C, W_{0.1}-FeCoP₂/C and W_{0.15}-FeCoP₂/C

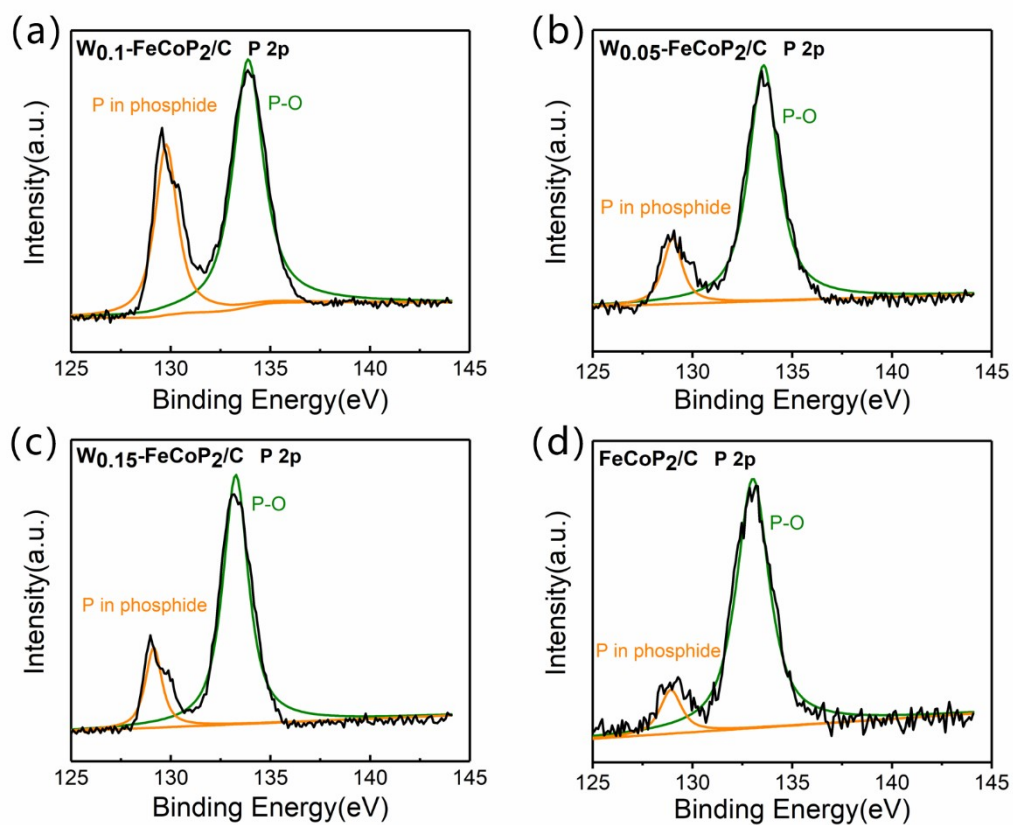


Figure S6. P 2p XPS spectra of FeCoP_2/C , $W_{0.05}\text{-FeCoP}_2/\text{C}$, $W_{0.1}\text{-FeCoP}_2/\text{C}$ and $W_{0.15}\text{-FeCoP}_2/\text{C}$

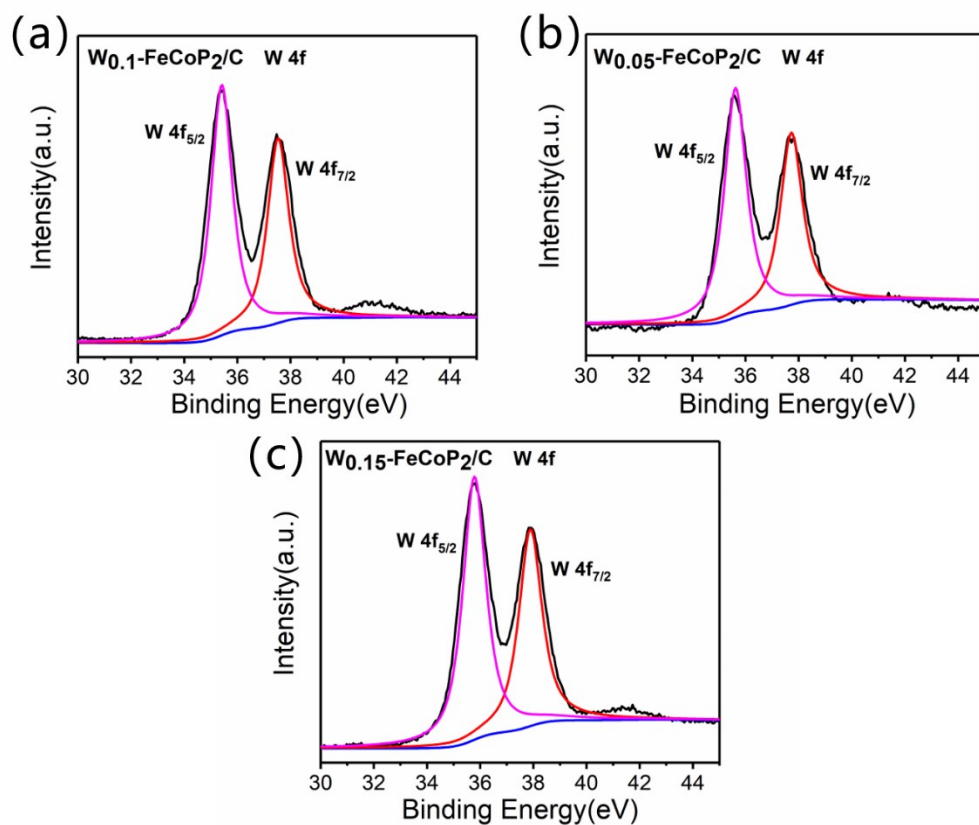


Figure S7. W 4f XPS spectra of FeCoP_2/C , $\text{W}_{0.05}\text{-FeCoP}_2/\text{C}$, $\text{W}_{0.1}\text{-FeCoP}_2/\text{C}$ and $\text{W}_{0.15}\text{-FeCoP}_2/\text{C}$

Table S4. OER performances of recently reported representative electrocatalysts in 1 M KOH electrolyte

Catalysts	Overpotential	Electrolyte	Reference
Fe-Co-P Alloy Hollow Spheres on GCE	252 mV@10 mA cm ⁻²	1.0 M KOH	[1]
NiCoP/C Nanoboxes on GCE	330 mV@10 mA cm ⁻²	1.0 M KOH	[2]
NiCoP/SCW	220 mV@10 mA cm ⁻²	1.0 M KOH	[3]
Mo-doped CoP Nanoarrays on carbon cloth	305 mV@10 mA cm ⁻²	1.0 M KOH	[4]
O-doped CoP@RGO on GCE	280 mV@10 mA cm ⁻²	1.0 M KOH	[5]
Co/CoP Nanoparticles on GCE	340 mV@10 mA cm ⁻²	1.0 M KOH	[6]
Amorphous NiFeP on GCE	219 mV@10 mA cm ⁻²	1.0 M NaOH	[7]
Sulfur-doped Co ₂ P on Ni Foam	360 mV@100 mA cm ⁻²	1.0 M KOH	[8]
FeCoNiP on carbon cloth	270 mV@100 mA cm ⁻²	1.0 M KOH	[9]
W_x-FeCoP₂/C	264 mV@10 mA cm⁻² 310 mV@100 mA cm⁻²	1.0 M KOH	This work

Table S5. The C_{dl} and the fitted R_{ct} of $FeCoP_2/C$, $W_{0.05}-FeCoP_2/C$, $W_{0.1}-FeCoP_2/C$ and $W_{0.15}-FeCoP_2/C$

Samples	C_{dl} (mF cm ⁻²)	R_{ct} (Ω)
$FeCoP_2/C$	5.78	16.14
$W_{0.05}-FeCoP_2/C$	6.25	12.57
$W_{0.1}-FeCoP_2/C$	7.15	7.86
$W_{0.15}-FeCoP_2/C$	6.78	9.85

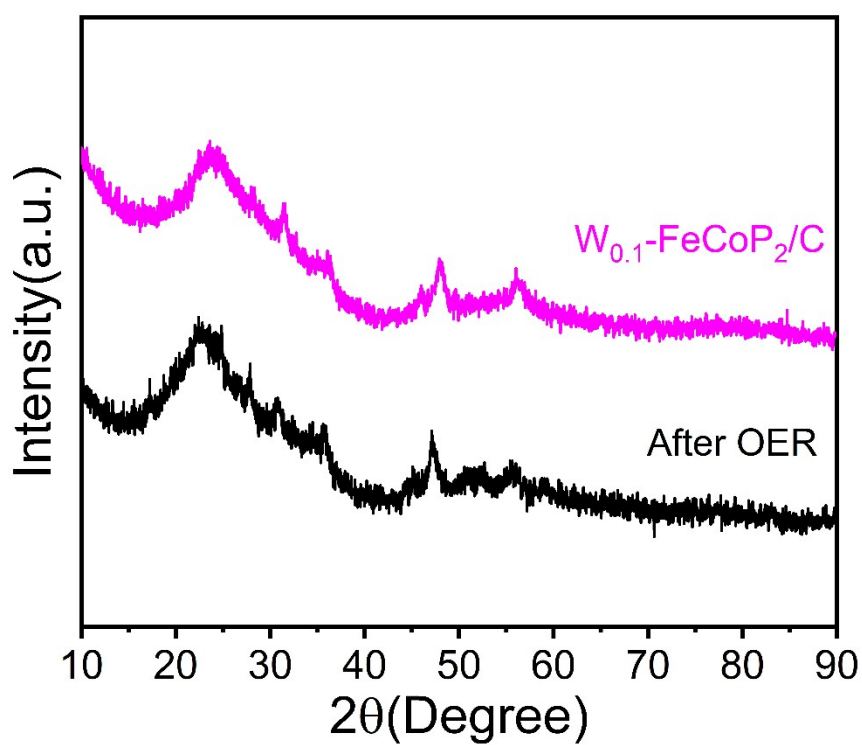


Figure S8. XRD patterns of the $W_{0.1}-FeCoP_2/C$ and the $W_{0.1}-FeCoP_2/C$ after OER tests

References:

- [1] K. Liu, C. Zhang, Y. Sun, G. Zhang, X. Shen, F. Zou, H. Zhang, Z. Wu, E. C. Wegener, C. J. Taubert, J. T. Miller, Z. Peng and Y. Zhu, *ACS Nano* **2018**, *12*, 158-167.
- [2] P. He, X.-Y. Yu and X. W. Lou, *Angewandte Chemie International Edition* **2017**, *56*, 3897-3900.
- [3] V. R. Jothi, R. Bose, H. Rajan, C. Jung and S. C. Yi, *Advanced Energy Materials* **2018**, *8*, 1802615.
- [4] C. Guan, W. Xiao, H. Wu, X. Liu, W. Zang, H. Zhang, J. Ding, Y. P. Feng, S. J. Pennycook and J. Wang, *Nano Energy* **2018**, *48*, 73-80.
- [5] G. Anandhababu, Y. Huang, D. D. Babu, M. Wu and Y. Wang, *Advanced Functional Materials* **2018**, *28*, 1706120.
- [6] Z.-H. Xue, H. Su, Q.-Y. Yu, B. Zhang, H.-H. Wang, X.-H. Li and J.-S. Chen, *Advanced Energy Materials* **2017**, *7*, 1602355.
- [7] F. Hu, S. Zhu, S. Chen, Y. Li, L. Ma, T. Wu, Y. Zhang, C. Wang, C. Liu, X. Yang, L. Song, X. Yang and Y. Xiong, *Advanced Materials* **2017**, *29*, 1606570.
- [8] M. A. R. Anjum, M. D. Bhatt, M. H. Lee and J. S. Lee, *Chemistry of Materials* **2018**, *30*, 8861-8870.
- [9] J. Xu, J. Li, D. Xiong, B. Zhang, Y. Liu, K.-H. Wu, I. Amorim, W. Li and L. Liu, *Chemical Science* **2018**, *9*, 3470-3476.