

Supplementary Section: Synergistic Effects of β -NaFeO₂ Ferrite Nanoparticles for Photocatalytic Degradation, Antibacterial, and Antioxidant Applications

Tahira Jabeen ^{1,*}, Muhammad Shahid Khan ^{1,*}, Sana Javaid ¹, Waqar Azeem ², Rabia Ayoub
¹, Martin Motola ^{3,*}

1. Institute of Physics, The Islamia University of Bahawalpur, Bahawalpur 63100, Pakistan
2. Faculty of Resilience, Rabdan Academy, Abu Dhabi, United Arab Emirates
3. Department of Inorganic Chemistry, Faculty of Natural Sciences, Comenius University
Bratislava, Ilkovicova 6, 842 15 Bratislava, Slovakia

*Corresponding Author:

T. Jabeen: (✉ tahira.jabeen1999@gmail.com; ☎ +923016979528)

M. S. Khan (✉ mshahid.khan@iub.edu.pk; ☎ +923006421403)

M. Motola (✉ martin.motola@uniba.sk; +421 2 9014 9374)

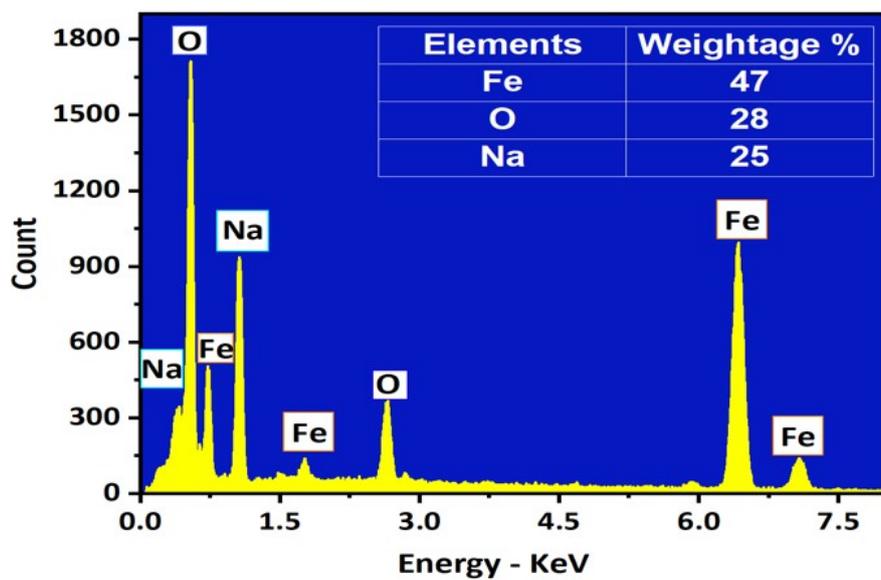


Fig. S1. Elemental composition of the material β -NaFeO₂ by EDS technique, with iron (Fe) constituting 47% of the overall composition, oxygen (O) comprising 28%, and sodium (Na) accounting for 25%

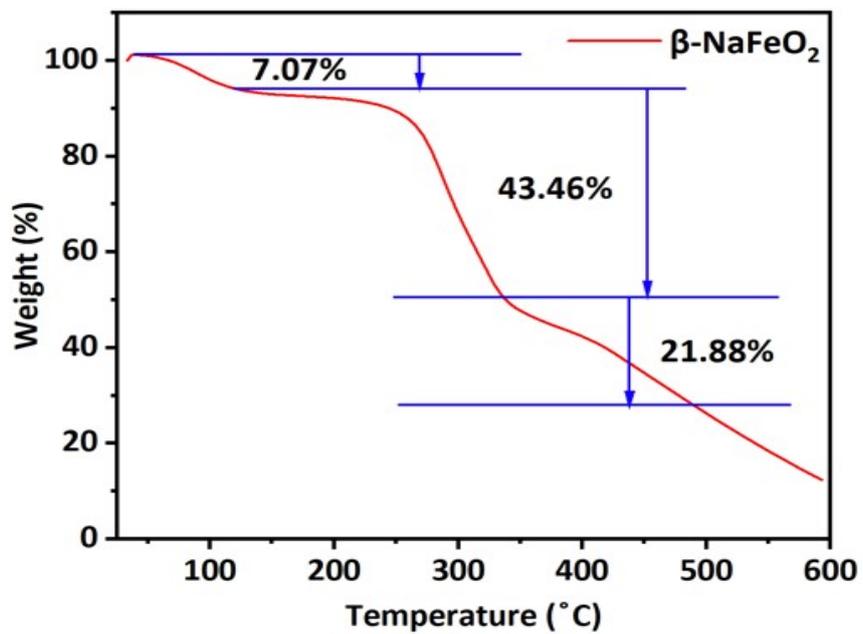


Fig. S2. Thermogravimetric Analysis (TGA) with weight loss patterns across temperature ranges, reflecting volatile component removal and organic decomposition during calcination.

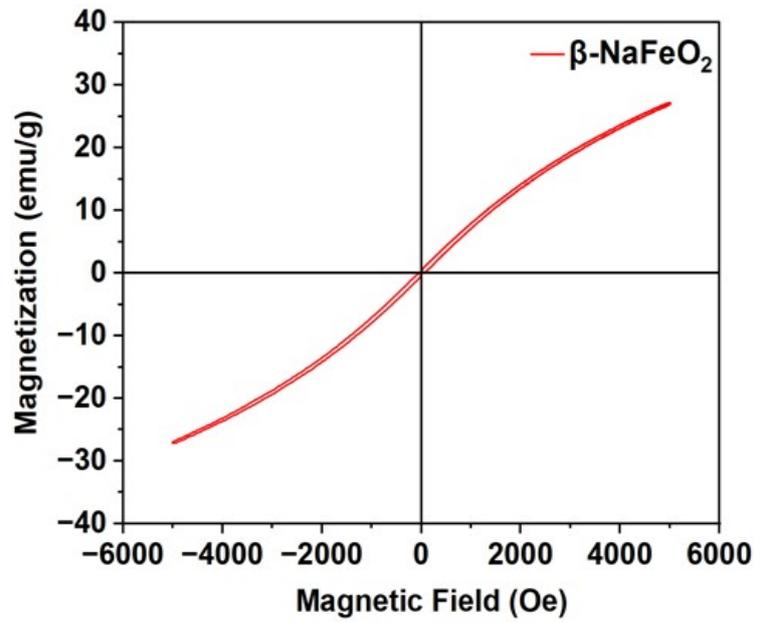


Fig. S3. Magnetic behavior analysis (M-H curve) shows superparamagnetic properties of β -NaFeO₂ nanoparticles with no coercivity (H_c) or remanence magnetization (M_r) and a saturation magnetization (M_s) of 27.11 emu/g at room temperature.

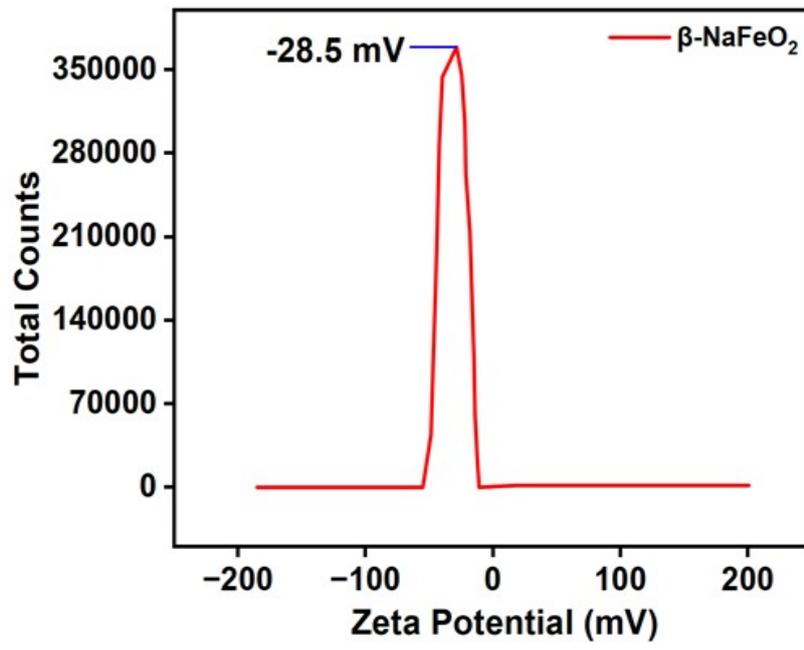


Fig. S4. Surface charge and colloidal stability of β -NaFeO₂ by zeta potential technique

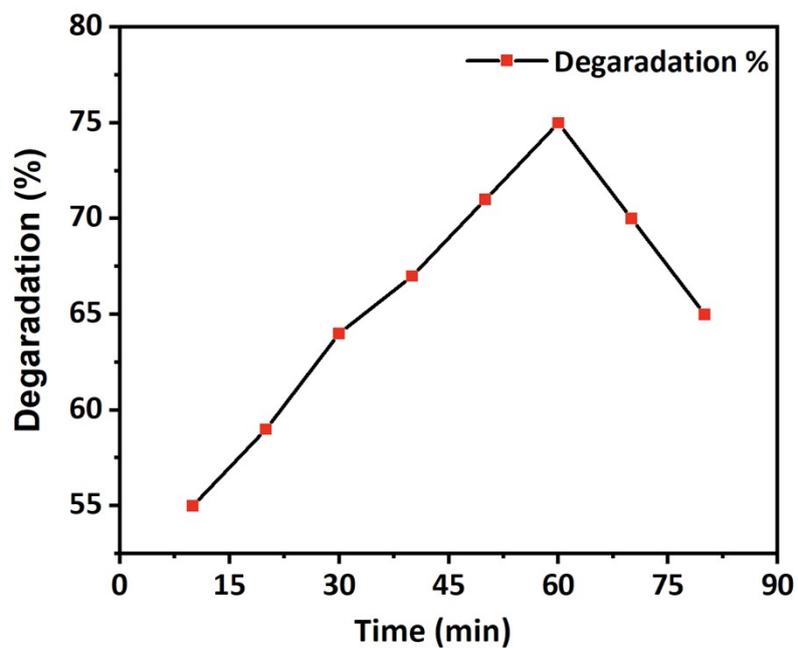


Fig. S5. Effect of TEMPO scavenger on photocatalytic degradation of Methyl Red Dye

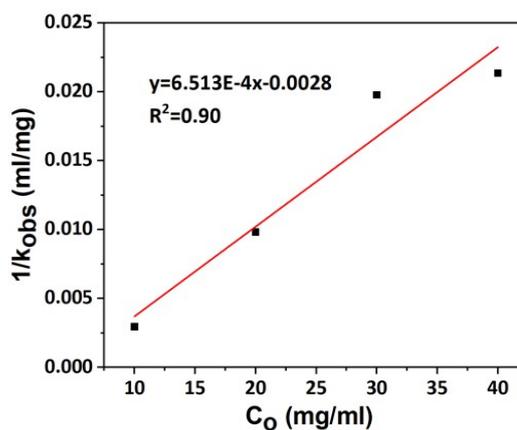
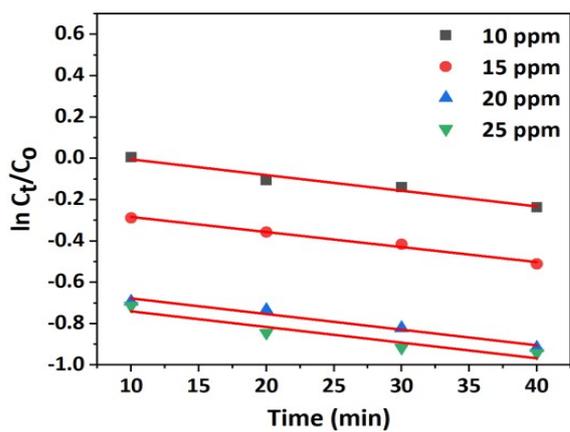


Fig. S6. Application of Langmuir-Hinshelwood model to analyze MR dye degradation kinetics, demonstrating strong correlation ($R^2 = 0.90$) and efficiency of β -NaFeO₂ nanoparticles ($k = 0.000613 \text{ min}^{-1}$)