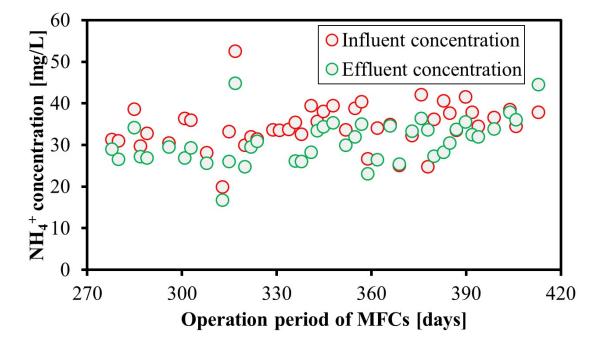
SUPPLEMENTARY INFORMATION 1 2 Application of natural zeolite adsorption in cooperation with photosynthesis 3 for the post-treatment of microbial fuel cells 4 5 6 Que Nguyen Ho<sup>1</sup>, Taira Hidaka<sup>2</sup>, Mukhlis A Rahman<sup>3</sup>, Naoko Yoshida<sup>1\*</sup> 7 8 <sup>1</sup> Department of Civil Engineering, Nagoya Institute of Technology, Nagoya, Japan 9 <sup>2</sup> Department of Environmental Engineering, Graduate School of Engineering, Kyoto University, 10 Kyoto University Katsura, Nishikyo, Kyoto 615-8540, Japan 11 <sup>3</sup>Advanced Membrane Technology Research Centre (AMTEC), School of Chemical and Energy 12 Engineering (FCEE), Universiti Teknologi Malaysia, 81310 UTM Skudai, Johor, Malaysia 13 14

15 I. The degradation of NH<sub>4</sub><sup>+</sup> following treatment with MFCs

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**Fig S1.** The degradation of  $NH_4^+$  concentration following treatment with MFCs. The influent concentration represented the  $NH_4^+$  concentration in sewage, while the effluent concentration indicated the  $NH_4^+$  concentration after treatment with MFCs.

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Fig S1 shows the degradation of  $NH_4^+$  after MFCs treatment, which coincided with the period using natural zeolite adsorption and photosynthesis. The average influent concentration of  $NH_4^+$ was approximately 35.16 mg/L, decreasing to 31.9 mg/L after passing through the MFCs. These results demonstrated that the MFCs system, which used an anion exchange membrane as a separator, was not efficient in removing  $NH_4^+$  from sewage. Therefore, a second step was necessary for  $NH_4^+$  removal. The integration of natural zeolite adsorption and photosynthesis clearly proved to be effective in removing  $NH_4^+$ , as shown in Fig 4(c1) and 4(c2) of the manuscript.

## 26 II. Comparison of NH<sub>4</sub><sup>+</sup> removal efficiency in this study with other reported systems

Table S1 presents the most popular combinations of MFCs with other technologies for the removal of nitrogenous compounds from wastewater. These studies achieved ammonium removal efficiencies exceeding 75%. Similar to these previous studies, our results demonstrated the effectiveness of this system in removing ammonium following MFCs treatment. Based on these review systems, the combination of MFC and membrane produced good output values. However, this approach raises several operational issues, including high costs, membrane fouling, and significant energy requirements for processes like air scouring to control bacterial growth on the 34 membranes (Al-Asheh, Bagheri et al. 2021, Bhattacharya and Banerjee 2023). In contrast, our 35 system showed promise due to its low cost and ease of operation. It utilized natural zeolite, which 36 is abundant and inexpensive (Wang and Peng 2010), and did not require additional energy for 37 operation (e.g., LED lights).

Table S1. Comparison the NH4<sup>+</sup> removal of this study and other reported systems

	Ammonium concentration [mg/L]		Highest removal-EC	References
Type of systems				
	Inf	Eff	[%]	
MFC-Membrane	32±4.28	$0.8 \pm 0.41$	97.7±1.9	(Malaeb, Katuri et al. 2013)
bioreactor				
Regular MFC-	21.4±10.2	4.9±3.8	77	(Zhang, Ge et al. 2013)
Denitrifying MFC				
MFC- FO	780	114.7	85.3±3.5	(Qin, Hynes et al. 2017)
MFC-CDI	21.4	0.6±0.1	97.6	(Feng, Tsai et al. 2017)
MFC-AA/O	100-130	16-25	80.7-84	(Liu, Tursun et al. 2017)
UFCW-MFC	39	3.51	91	(Oon, Ong et al. 2015)
MFC-ZP	29.6-34.7	10.4-3.1	60-84.5	This study

40 Inf: influent concentration; Eff: effluent concentration; EC: efficiency; FO: forward osmotic, CDI:
41 capacitive deionization; AA/O: anaerobic–anoxic–oxic; UFCW-MFC: up-flow constructed
42 wetland-microbial fuel cell; ZP: natural zeolite adsorption integrated with photosynthesis.

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