

**Supplementary Materials: Synthesis and Characterization of Soluble Pyridinium Containing Copolyimides**

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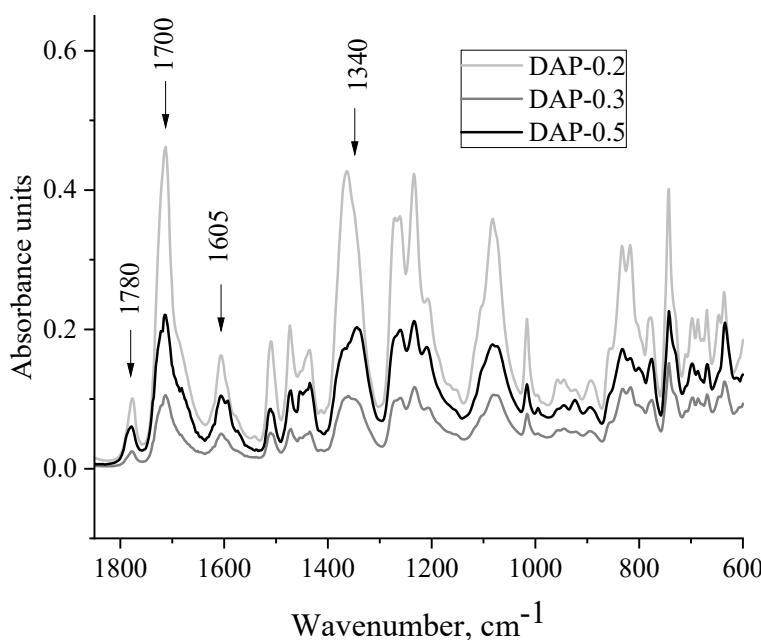
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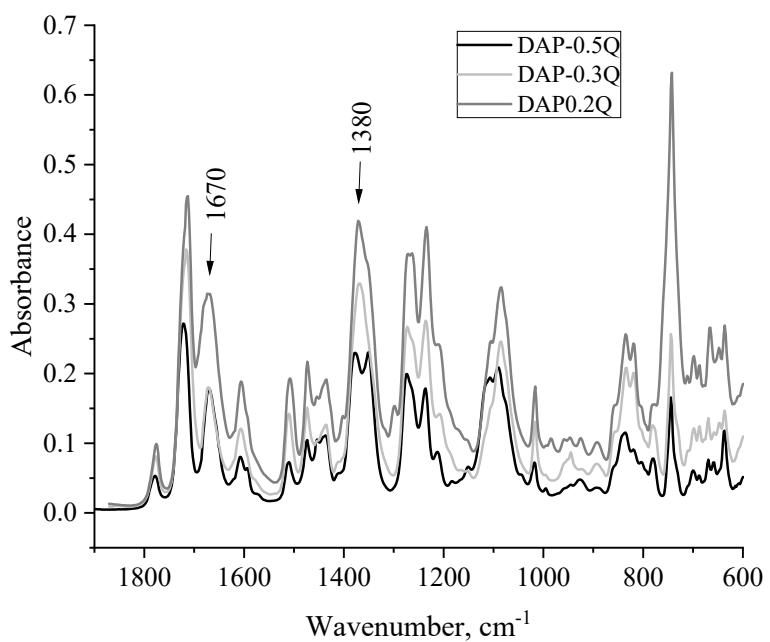
Charles University, Prague 180 00, Czech Republic

**Table S1.** Monomers ratio in synthesized coPIs calculated from  $^1\text{H}$  NMR spectra.

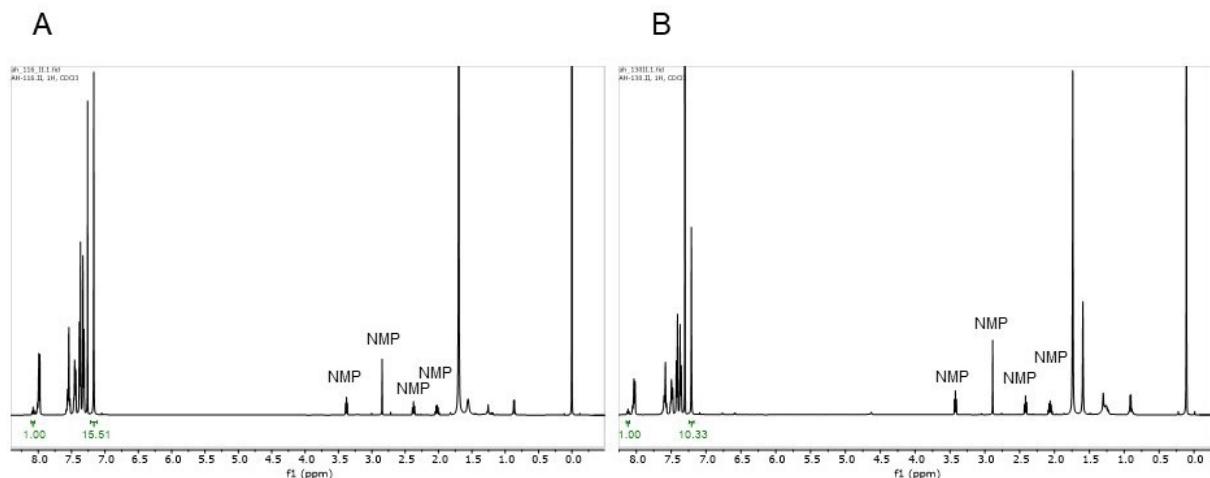
Sample	Theoretical	NMR calculated
	ODPA:BIS P:DAP	
DAP-0.5	1:0.5:0.5	1:0.53:0.47
DAP-0.3	1:0.7:0.3	1:0.7:0.3
DAP-0.2	1:0.8:0.2	1:0.78:0.22



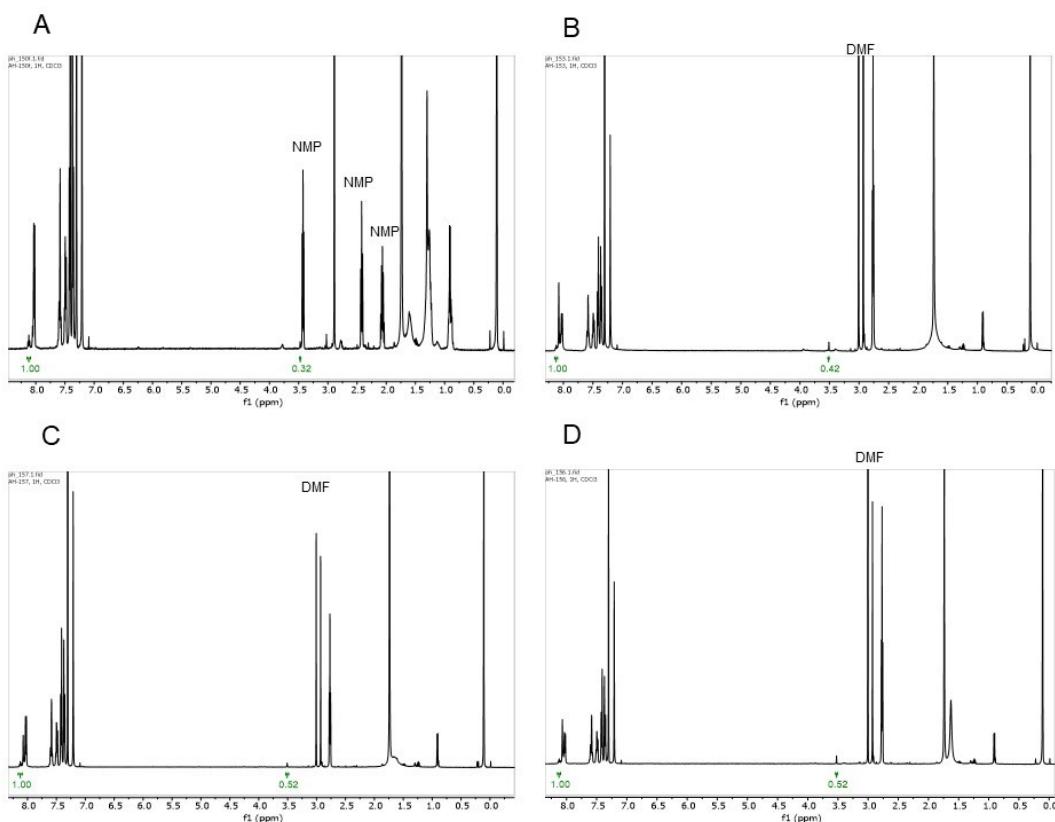
**Figure S1.** FTIR spectra of coPIs



**Figure S2.** FTIR spectra of coPI-Qs



**Figure S3.** NMR spectra of coPIs: DAP-0.2 (A), DAP-0.3 (B)



**Figure S4.** NMR spectra of coPI-Qs: DAP-0.2Q (A), DAP-0.3Q (B), DAP-0.2Q2-(C), DAP-0.3Q2 (D)

### SEC Measurements

Molar mass and its distribution was determined using Waters SEC-RI chromatograph (Waters 515 pump, Waters 2414 refractive index detector, ECOM 4 channel degasser). Separation of polymers (~2 mg/ml solution in chloroform) was done at 25°C on Phenogel 5 µm linear column (7.8x300 mm) using chloroform (HPLC grade, stabilized by amylyene) as mobile phase (1 ml/min). Evaluation of data was done in Clarity 6.0 GPC software using conventional calibration based on narrow polystyrene standards (580 – 1233000 g/mol) providing apparent molar mass values (polystyrene equivalents).

**Table S2.** Molar mass characteristics of PI and *co*PIs

Sample	M <sub>n</sub> , g/mol	M <sub>w</sub> , g/mol	D
DAP-0.5	22 100	522 000	23.6
DAP-0.3	10 900	32 700	3.0
DAP-0.2	45 100	93 400	2.1
DAP-0	38 900	89 600	2.3