

Electronic supporting information

Novel enantiopure δ-thiolactones: synthesis, structural characterization, and reactivity studies

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I. Crystal structure of δ -thiolactones **1a** and **1b**

The crystals were obtained using vapor diffusion crystallization method. A mixture of miscible solvents was used. The compound was dissolved in the smallest amount of the most soluble solvent. Then a quantity of precipitating solvent was added, but in such quantities as to allow its dissolution. For 10 mg of compound a final volume of 4-5mL was used. The final solution was filtered into a vial and left for 7-10 days, under slow evaporation conditions.

The X-ray intensity data were measured at room temperature, 298 (2) K, using CuK α ($\lambda = 1.54184 \text{ \AA}$) for **1a**, and MoK α radiation ($\lambda = 0.71073 \text{ \AA}$) for **1c**, using a Bruker D8 Venture diffractometer. The frames were integrated with the Bruker SAINT software package using a narrow-frame algorithm and scaling and absorption correction using Multi-scan SADABS (BrukerSAINT and SADABS, Bruker AXS Inc., 2018). The crystal structures were solved by Direct Methods [1], and then completed by a difference Fourier map, refined using the program SHELX2018/3 [2] and the molecular and supramolecular graphics were carried out using the software Mercury [3]. Crystallographic data for the structures have been deposited in the Cambridge Crystallographic Data Center (CCDC) with deposition numbers - **1a** CCDC: 2387412, **1c**: 2387413. Crystal data, data collection, and structure refinement details are summarized in Table S1.

Table S1. Crystallographic data of δ -thiolactones **1a** and **1c**.

Crystal Data	Compound 1a	Compound 1c
Chemical Formula M_r	$C_7H_9NOS_3$ 219.33	$C_7H_9NOS_3$ 219.33
Crystalline system, space group	Monoclinic, $P2_1$	Orthorhombic, $P2_12_12_1$
$a, b, c \text{ (\AA)}$	11.4682 (5), 5.3017 (3), 15.1947 (7)	5.2493 (7), 12.0624 (18), 14.1479 (16)
$\alpha, \beta, \gamma (^\circ)$	90, 91.900 (2), 90	90, 90, 90
Volume, (\AA^3) $\rho, \text{ kg m}^{-3}$	923.34 (8) 1.578	895.8 (2) 1.626
Z	4	4
Temperature, (K)	298(2)	298(2)
Radiation type $\mu \text{ (mm}^{-1}\text{)}$	Cu K α 6.94	Mo K α 0.77
Theta range for data collection	$2.910^\circ < 2\theta < 72.310^\circ$ $-14 \leq h \leq 14,$ $-6 \leq k \leq 5,$ $-18 \leq l \leq 17$	$2.880^\circ < 2\theta < 26.601^\circ$ $-6 \leq h \leq 6,$ $-15 \leq k \leq 15,$ $-17 \leq l \leq 17$
Data collection		
Diffractometer	Bruker D8 Venture	Bruker D8 Venture
Absorption correction	Multi-Scan SADABS2016/2 (Bruker,2016/2)	Multi-Scan SADABS2016/2 (Bruker,2016/2)
Tmin, Tmax	0.421, 0.754	0.814, 0.941

No. of measured, independent and observed reflections [I>2σ(I)]	15886, 3542, 3270	29567, 1863, 1775
R _{int}	0.049	0.040
(sin θ/λ) _{max} (Å ⁻¹)	0.618	0.630
Refinement		
R[F ² > 2σ(F ²)], wR(F ²), S	0.039, 0.105, 1.06	0.038, 0.088, 1.25
No. of reflections	3542	1863
Refined parameters	217	109
No. of restraints	1	0
H-atoms treatment	H-atom parameters constrained	H-atom parameters constrained
Δρ _{max} , Δρ _{min} (e Å ⁻³)	0.42, -0.30	0.36, -0.27
Absolute structure	0.070 (12)	0.01 (4)
Absolute structure	Flack x determined using 1279 quotients [(I+)-(I-)]/[(I+)+(I-)] (Parsons, Flack and Wagner, Acta Cryst. B69 (2013) 249-259).	

II. NMR prediction methods

The conformational search for all the stereoisomers **1a-c** was performed through the LowMD method in MOE with chloroform as the implicit solvent (Born model, e=4.81). The initial optimization was made to the conformer with the lowest energy in MOE using the AMBER12:EHT force field, followed by the AM1 semi-empirical method.

Quantum

calculations were performed using Gaussian09. For optimization, the B3LYP method was used with a theory level of 6-311+g(d'; p') and chloroform as the solvent, modeled with SDM. Vibrational frequencies were calculated to check for the absence of imaginary values and confirm the reaching of a minimum. 1 H and 13 C NMR predictions were subsequently performed using the GIAO method with the same solvent model. The shifts for all the isomers 1a-d were calculated using TMS 1 H and 13 C shielding as the reference and compared with the experimental data.

Table S2: Optimized coordinates of 1a-d for NMR calculations

• 1a	• 1b
0 1	
H -1.84600500 0.37651900 1.59721500	H 1.82168800 0.52515000 1.45283700
C -1.63896700 0.40683500 0.52917300	C 1.64210100 0.51550000 0.36814700
C -2.57380900 -0.55305000 -0.20404700	S 1.60889200 2.28155400 -0.20356900
H -2.68189800 -0.29955500 -1.25707800	C -0.25041500 2.30622800 -0.21824300
H -3.55501100 -0.61996200 0.25977600	H -0.60813500 3.17918100 0.32317200
S -1.72382000 -2.18241000 -0.04702300	H -0.59514700 2.34628300 -1.25063900
S -1.74950400 2.18009000 -0.01665800	C -0.67731900 0.98925100 0.45516900
C 0.08199300 2.35319600 0.10483700	C -2.11274300 0.59930600 0.11910100
H 0.40405800 3.16303800 -0.54298300	O -3.00899300 1.41064100 0.15384000
H 0.37581800 2.56446500 1.13107400	H -0.67565200 1.12593900 1.55299500

<i>C</i>	0.58263300	0.99358200	-0.36470300	<i>N</i>	0.32124500	0.02414300	0.03277900
<i>H</i>	0.48984200	0.94295200	-1.46296600	<i>C</i>	0.16854200	-1.36156700	0.43070200
<i>N</i>	-0.24641700	-0.01530700	0.28525700	<i>C</i>	-0.92621400	-2.00672700	-0.39506600
<i>C</i>	-0.12016900	-1.32259800	-0.35571800	<i>H</i>	-1.12443000	-3.03041300	-0.07397800
<i>H</i>	-0.00078400	-1.21792900	-1.44132100	<i>H</i>	-0.66397300	-1.99945400	-1.45203800
<i>C</i>	1.08338600	-2.03686600	0.22708300	<i>S</i>	-2.54003600	-1.11632500	-0.19477600
<i>H</i>	0.93381000	-2.24903400	1.28279100	<i>H</i>	-0.03088900	-1.47912300	1.50472300
<i>H</i>	1.31460800	-2.95592100	-0.30968700	<i>S</i>	1.85482800	-2.10185700	0.08838100
<i>S</i>	2.61302600	-0.98477700	0.08380800	<i>C</i>	2.60278500	-0.44644400	-0.30702400
<i>C</i>	2.06175600	0.72817400	-0.09032200	<i>H</i>	3.60820100	-0.41096200	0.11074300
<i>O</i>	2.89928800	1.59756200	-0.08168400	<i>H</i>	2.63218300	-0.29796900	-1.38528400
<i>I 2 1.0</i>				<i>I 2 1.0</i>			
<i>2 3 1.0 7 1.0 13 1.0</i>				<i>2 3 1.0 11 1.0 19 1.0</i>			
<i>3 4 1.0 5 1.0 6 1.0</i>				<i>3 4 1.0</i>			
<i>4</i>				<i>4 5 1.0 6 1.0 7 1.0</i>			
<i>5</i>				<i>5</i>			
<i>6 14 1.0</i>				<i>6</i>			
<i>7 8 1.0</i>				<i>7 8 1.0 10 1.0 11 1.0</i>			
<i>8 9 1.0 10 1.0 11 1.0</i>				<i>8 9 2.0 16 1.0</i>			
<i>9</i>				<i>9</i>			
<i>10</i>				<i>10</i>			
<i>11 12 1.0 13 1.0 20 1.0</i>				<i>11 12 1.0</i>			
<i>12</i>				<i>12 13 1.0 17 1.0 18 1.0</i>			
<i>13 14 1.0</i>				<i>13 14 1.0 15 1.0 16 1.0</i>			
<i>14 15 1.0 16 1.0</i>				<i>14</i>			
<i>15</i>				<i>15</i>			
<i>16 17 1.0 18 1.0 19 1.0</i>				<i>16</i>			
<i>17</i>				<i>17</i>			
<i>18</i>				<i>18 19 1.0</i>			
<i>19 20 1.0</i>				<i>19 20 1.0 21 1.0</i>			
<i>20 21 2.0</i>				<i>20</i>			
<i>21</i>				<i>21</i>			
• 1c				• 1d			
<i>0 1</i>				<i>0 1</i>			
<i>H</i>	-2.10723600	0.14150600	1.54180000	<i>H</i>	1.63530800	1.12043100	1.77470700
<i>C</i>	-1.69160600	0.31041400	0.54710100	<i>C</i>	1.55168000	0.60731200	0.81750100
<i>S</i>	-1.85751400	2.10048900	0.12851800	<i>S</i>	1.59613800	1.98003000	-0.51513900
<i>C</i>	-0.05728700	2.27524400	-0.22571900	<i>C</i>	-0.24146500	2.00530900	-0.61426600
<i>H</i>	0.07804000	2.87559900	-1.12338400	<i>H</i>	-0.55893000	1.66780300	-1.60150700
<i>H</i>	0.42743200	2.76855100	0.61488800	<i>H</i>	-0.61843700	3.01203800	-0.44590000
<i>C</i>	0.47334300	0.85479200	-0.40966600	<i>C</i>	-0.73112100	1.04026800	0.51550700
<i>C</i>	1.98548000	0.80100500	-0.14931800	<i>C</i>	-2.15551200	0.56467800	0.24839600
<i>O</i>	2.69627800	1.77719800	-0.15399800	<i>O</i>	-3.07647800	1.35010100	0.27652100
<i>H</i>	0.31985400	0.53329700	-1.45255400	<i>H</i>	-0.78964100	1.61019500	1.44653900
<i>N</i>	-0.25640800	0.01699900	0.56116400	<i>N</i>	0.26013500	-0.01643400	0.70659600
<i>C</i>	-0.03360500	-1.41941400	0.54672300	<i>C</i>	0.24240700	-1.02402300	-0.36517000
<i>C</i>	1.19891800	-1.83644500	-0.24333900	<i>C</i>	-0.88103400	-2.00475700	-0.10526000
<i>H</i>	1.02748700	-1.77010700	-1.31783200	<i>H</i>	-0.76397700	-2.48182000	0.86715000
<i>H</i>	1.46843100	-2.86586400	-0.00742200	<i>H</i>	-0.93598500	-2.76604500	-0.88463500
<i>S</i>	2.67448500	-0.81933100	0.17193400	<i>S</i>	-2.51781700	-1.14114400	-0.15557600
<i>H</i>	0.02379900	-1.78910100	1.57021300	<i>H</i>	0.12218900	-0.57017100	-1.35500200
<i>S</i>	-1.58490000	-2.24728900	-0.19805200	<i>S</i>	1.90389900	-1.82704200	-0.31633700
<i>C</i>	-2.38263500	-0.61735500	-0.46579100	<i>C</i>	2.63777000	-0.45003500	0.67194900
<i>H</i>	-2.23003700	-0.28014600	-1.49098900	<i>H</i>	2.92340100	-0.83270200	1.65207500
<i>H</i>	-3.45190200	-0.70166100	-0.26923400	<i>H</i>	3.52508400	-0.06752000	0.16913200
<i>I 2 1.0</i>				<i>I 2 1.0</i>			
<i>2 3 1.0 11 1.0 19 1.0</i>				<i>2 11 1.0 19 1.0</i>			
<i>3 4 1.0</i>				<i>3 4 1.0</i>			
<i>4 5 1.0 6 1.0 7 1.0</i>				<i>4 5 1.0 6 1.0 7 1.0</i>			
<i>5</i>				<i>5</i>			
<i>6</i>				<i>6</i>			
<i>7 8 1.0 10 1.0 11 1.0</i>				<i>7 8 1.0 10 1.0 11 1.0</i>			
<i>8 9 2.0 16 1.0</i>				<i>8 9 2.0 16 1.0</i>			
<i>9</i>				<i>9</i>			
				<i>10</i>			

10	11 12 1.0	11 12 1.0
11 12 1.0	12 13 1.0 17 1.0	12 13 1.0 17 1.0 18 1.0
12 13 1.0 17 1.0	13 14 1.0 15 1.0 16 1.0	13 14 1.0 15 1.0 16 1.0
13 14 1.0 15 1.0 16 1.0	14	14
14	15	15
15	16	16
16	17	17
17	18 19 1.0	18 19 1.0
18 19 1.0	19 20 1.0 21 1.0	19 20 1.0 21 1.0
19 20 1.0 21 1.0	20	20
20	21	21
21		

III. HTE analytical-scale screening for δ -thiolactone preparation 1a-c

Three stock solutions of **2** (100 mg, 0.36 mmol) were prepared in the corresponding solvent DCM, DEC and MeCN (5 mL).

Reactions were performed in 1.5 mL Eppendorf tubes. For each case, in a tube containing 0.05 mmol of the coupling reagent, a stock solution of **2** in the corresponding solvent (0.5 mL, 0.042 mmol) was added. In the case of HATU and COMU, DIPEA (0.084 mmol, 15 μ L) was additionally added, and in the case of EDCI and DIC, catalytic amount of 4-DMAP was additionally added. The fifteen different reactions were conducted simultaneously in a digital Thermo Shaker with the appropriate module, at 25° C and 300 rpm, for 24 hours.

For the HPLC analysis, aliquots of 10 μ L were dissolved in 1 mL HPLC grade MeCN and filtered through 0.22 μ m PVDF filters. The yield was calculated as conversion percentage of L-2, calculated from the corresponding calibration curve, and verified from the quantification of 1a and 1c from the corresponding calibration curves.

Table S3: Analytical-scale screening for δ -thiolactone **1a-c** preparation.

Entry	Solvent	Coupling agent (eq.)	Aditives (eq)	1a-c Yield (%)
1		CDI (1.0)	-	0
2		COMU (1.2)	DIPEA (2)	0
3	DEC	HATU (1.2)	DIPEA (2)	0
4		EDCI (1.2)	4-DMAP (0.1)	10
5		DIC (1.2)	4-DMAP (0.1)	28
6		CDI (1.0)	-	0
7		COMU (1.2)	DIPEA (2)	16
8	ACN	HATU (1.2)	DIPEA (2)	14
9		EDCI (1.2)	4-DMAP (0.1)	15
10		DIC (1.2)	4-DMAP (0.1)	25
11		CDI (1.0)	-	0
12		COMU (1.2)	DIPEA (2)	40
13	DCM	HATU (1.2)	DIPEA (2)	5
14		EDCI (1.2)	4-DMAP (0.1)	80
15		DIC (1.2)	4-DMAP (0.1)	30

The High Performance Liquid Chromatography was conducted in an Agilent series 1200, equipped with automatic injector (inj. vol 5 μ L), quaternary pumps and diode array detector. A Kinetex NUCLEOSIL ® octadecylsilane column was used, 5 μ m p.d., 4.6x150 mm, with column oven thermostated at 25 °C. Monitored wavelengths were 210 and 250 nm. Elution was performed in isocratic mode with a mixture 65:35 of 0.1% formic acid and MeCN. The data was collected and analyzed using OpenLab software. Calibration curves for **L-2**, **1a** and **1c** were obtained with 25, 50, 75, 125 and 250 μ g/mL standards (Fig S1-3), prepared from a 500 μ g/mL stock solution in MeCN. In each case, the 250 μ g/mL standard was injected five times to evaluate system precision, obtaining RSD values of 1.31 for L-2, 1.07 for 1a and 0.09 for 1c. System aptitude was evaluated injecting a solution of L-2, 1a and 1c at 200 μ g/mL. For an elution order consisting of L-2 (rt: 3.38 min), 1c (rt: 5.87 min), 1a (rt: 7.85 min), selectivity indexes of 1.29 and 1.45 were obtained.

Figure S1: Calibration curve obtained for L-2, in a range of 25-250 μ g/mL, at a monitoring wavelength of 210 nm.

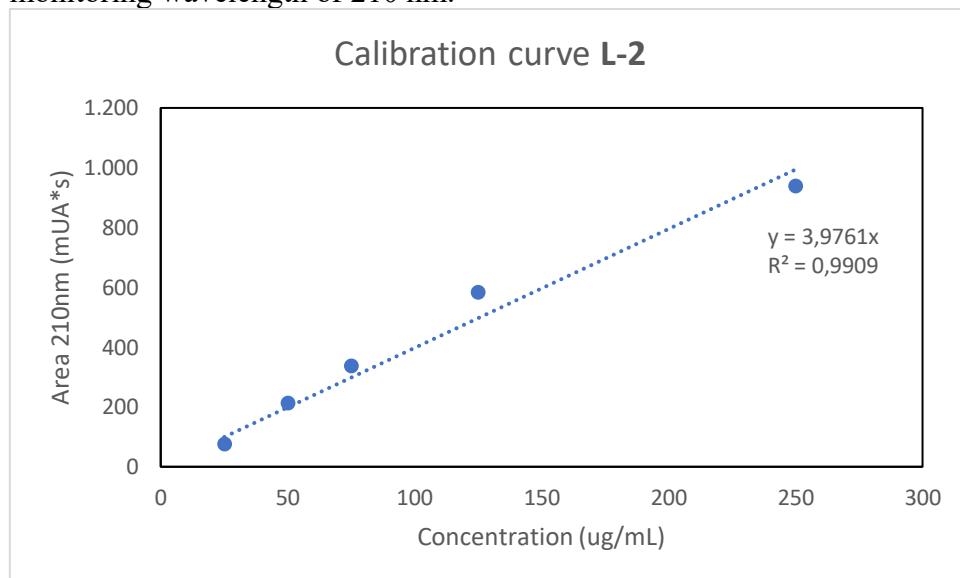


Figure S2: Calibration curve obtained for 1a, in a range of 25-250 µg/mL, at a monitoring wavelength of 250 nm.

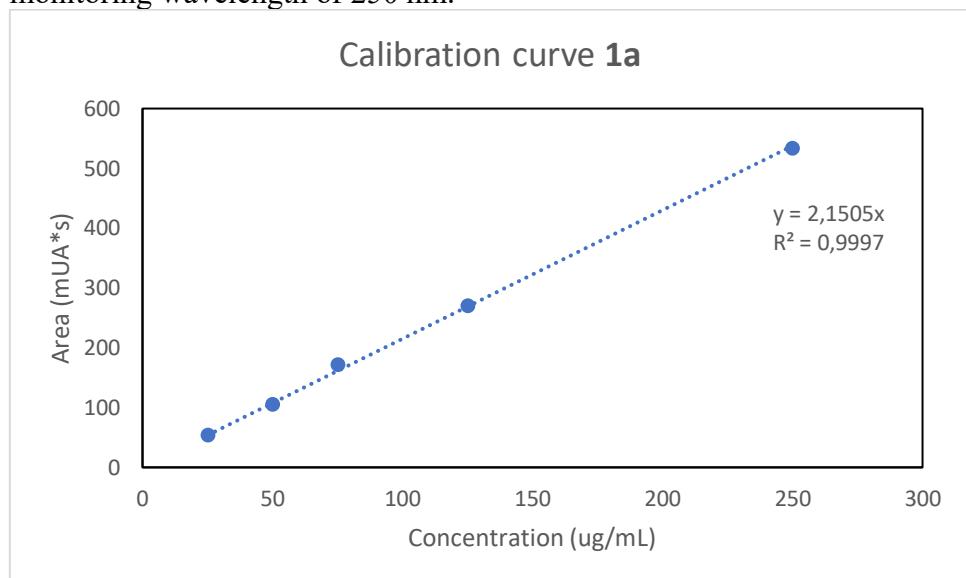
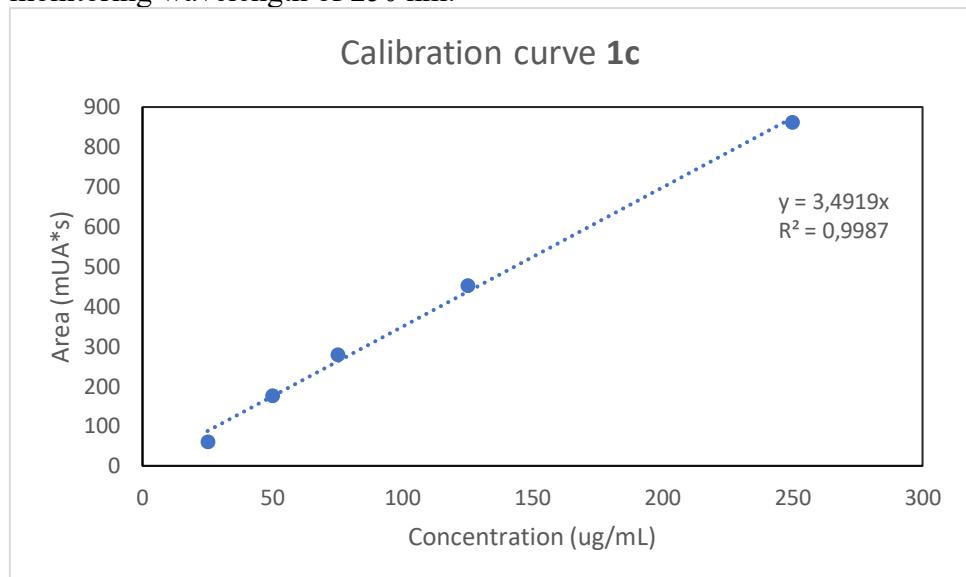


Figure S3: Calibration curve obtained for 1c, in a range of 25-250 µg/mL, at a monitoring wavelength of 250 nm.



IV. NMR Spectra of δ -thiolactones 1a-c and 4a, 4c

Figure S4: ^1H -NMR of **1a** in CDCl_3 (400 MHz)

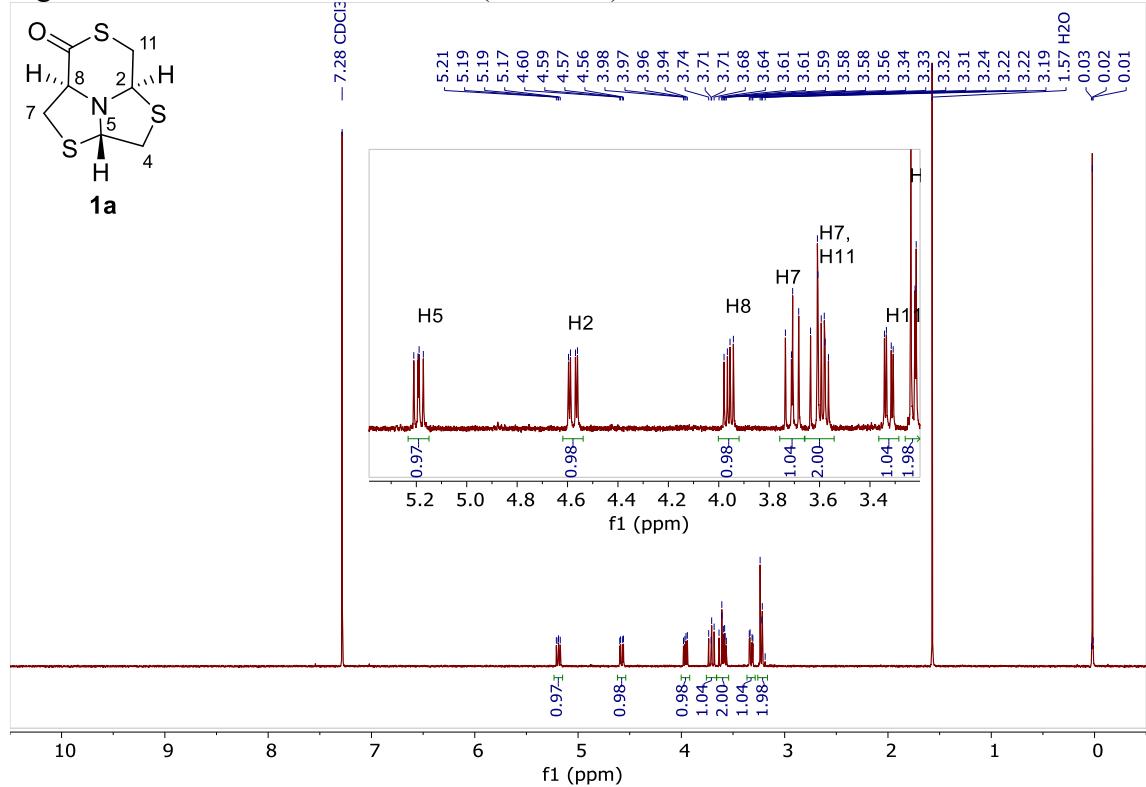


Figure S5: ^{13}C -NMR of **1a** in CDCl_3 (101 MHz)

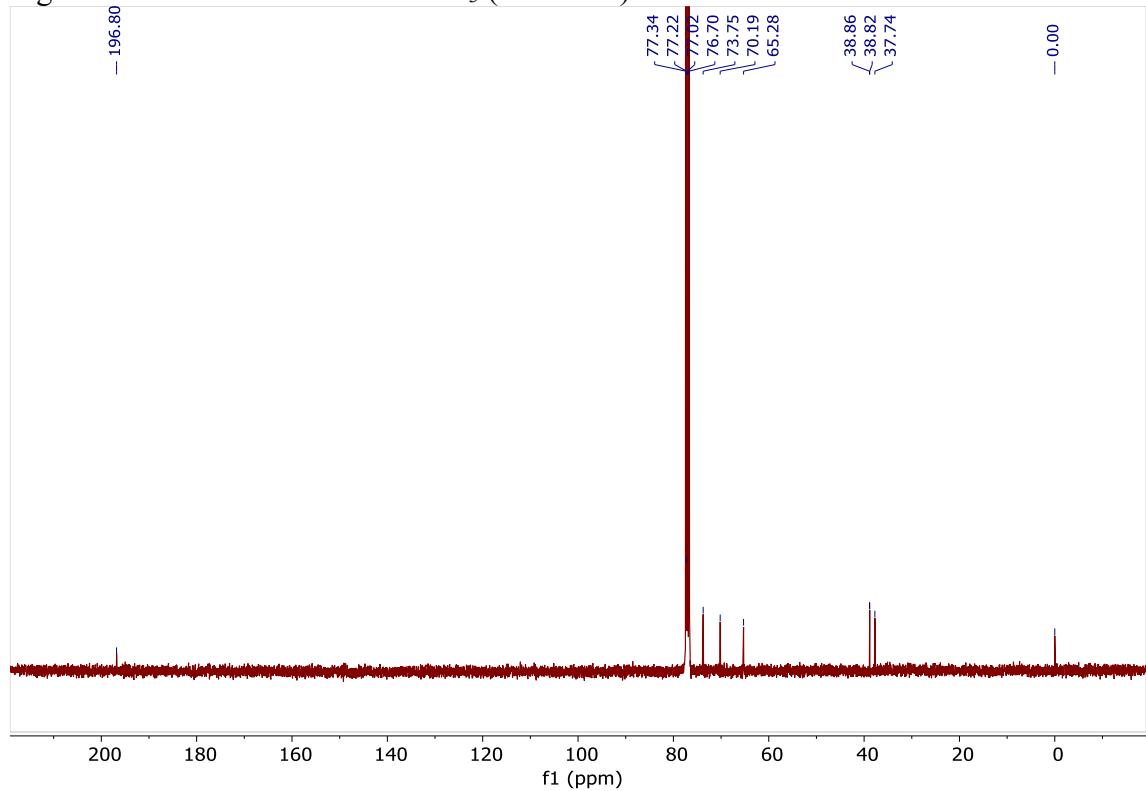


Figure S6: ^1H -NMR of **1b** in CDCl_3 (400 MHz)

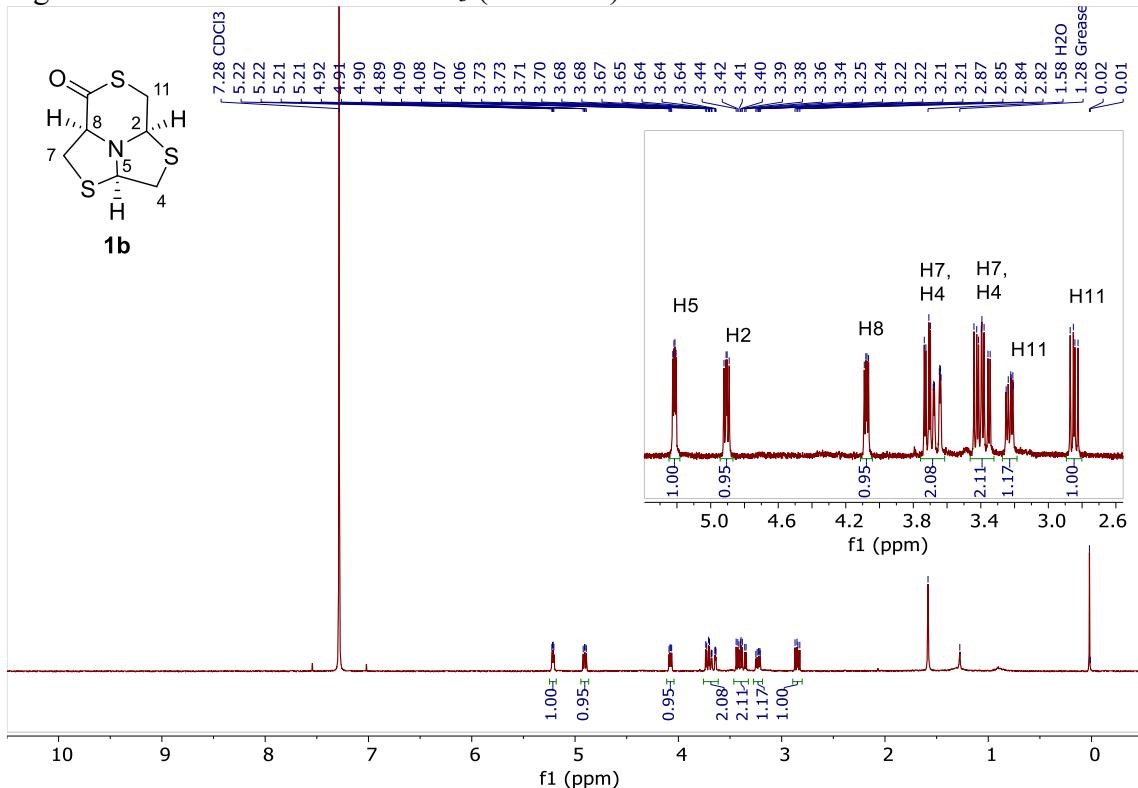


Figure S7: ^{13}C -NMR of **1b** in CDCl_3 (101 MHz)

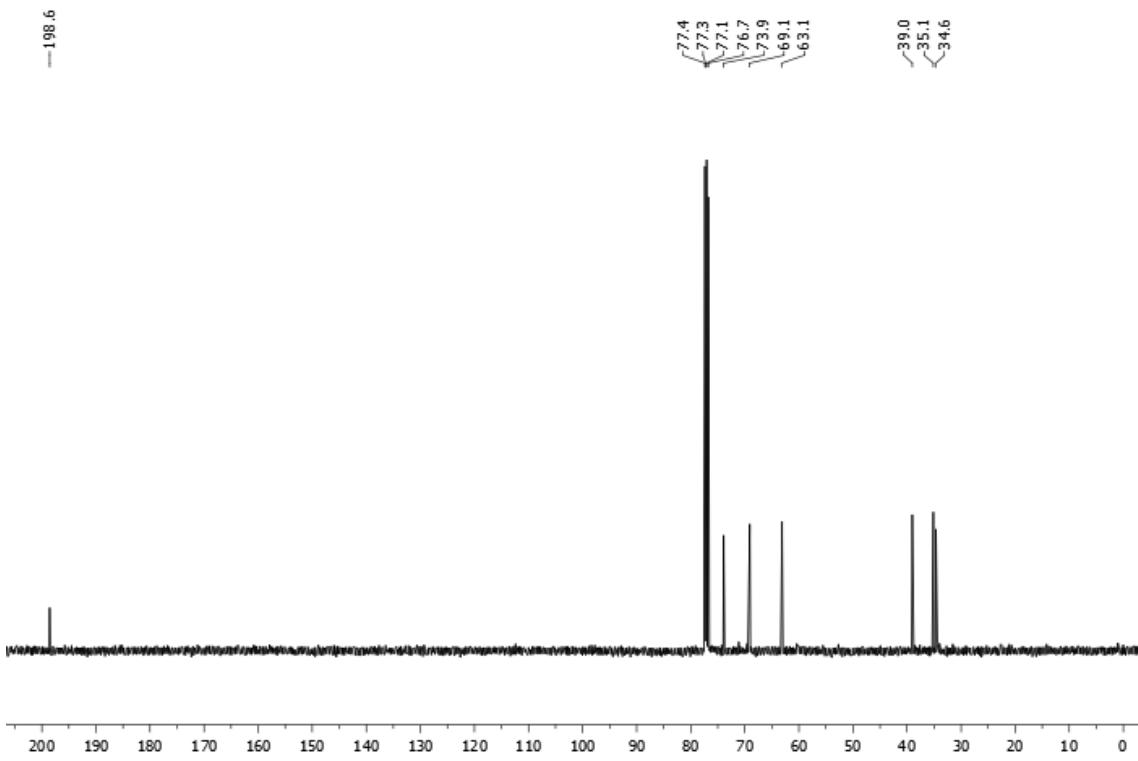


Figure S8: ^1H -NMR of **1c** in CDCl_3 (400 MHz)

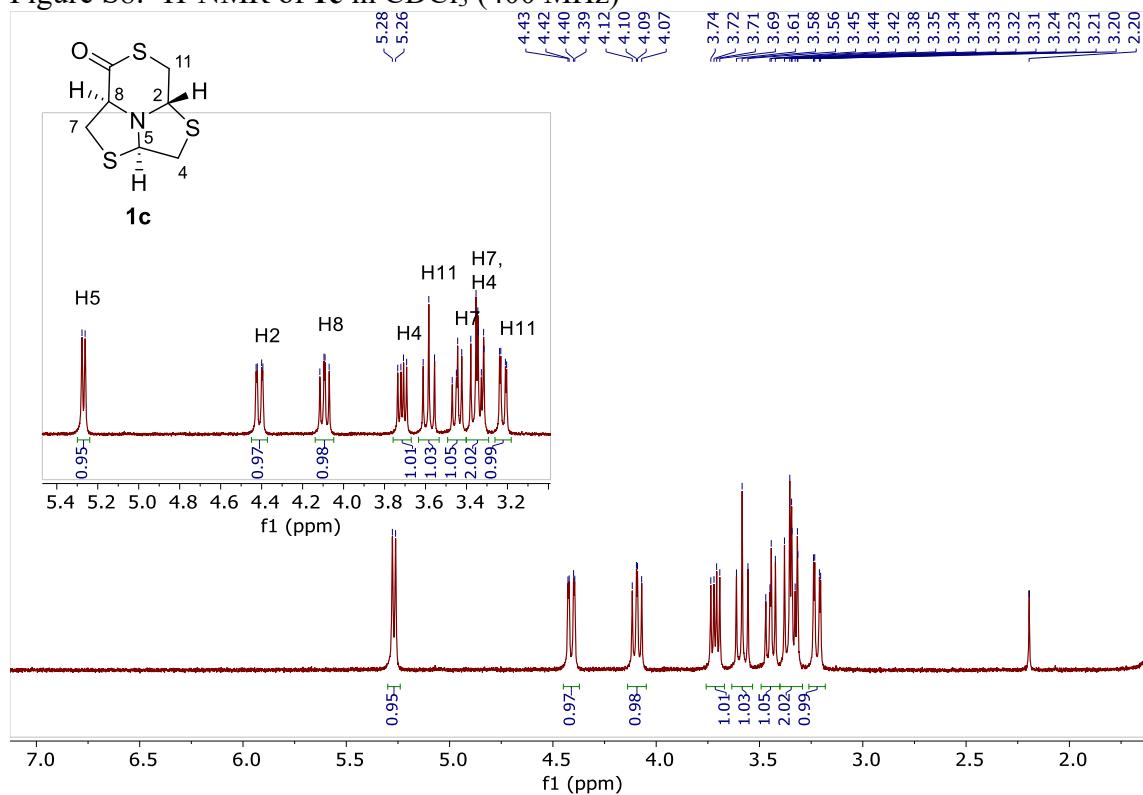


Figure S9: ^{13}C -NMR of **1c** in CDCl_3 (101 MHz)

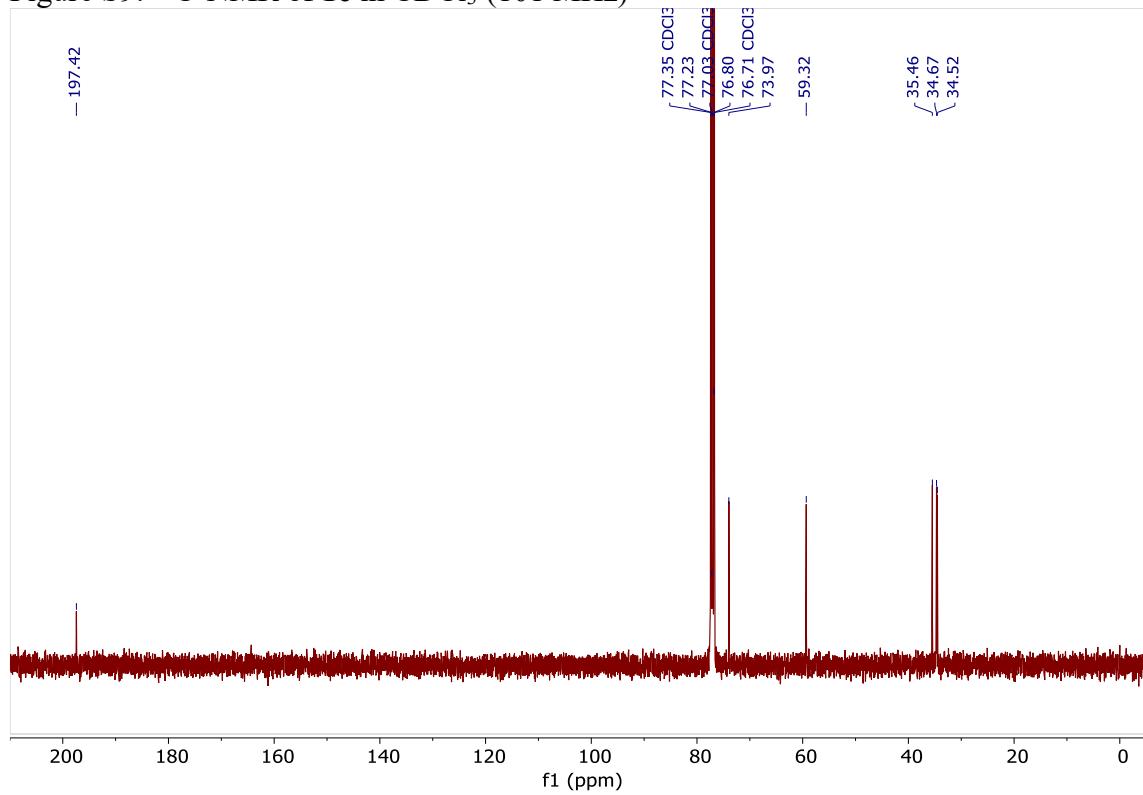


Figure S10: ^1H -NMR of **4a** in CDCl_3 (400 MHz)

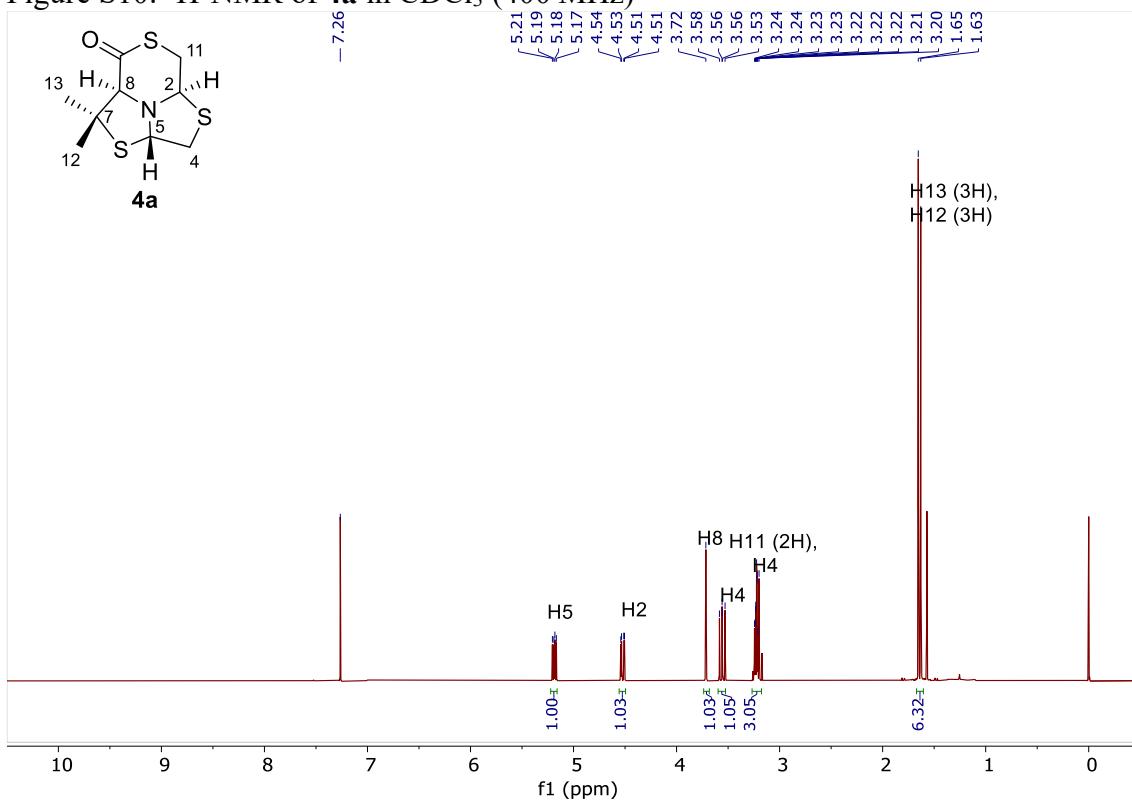


Figure S11: ^{13}C -NMR of **4a** in CDCl_3 (101 MHz)

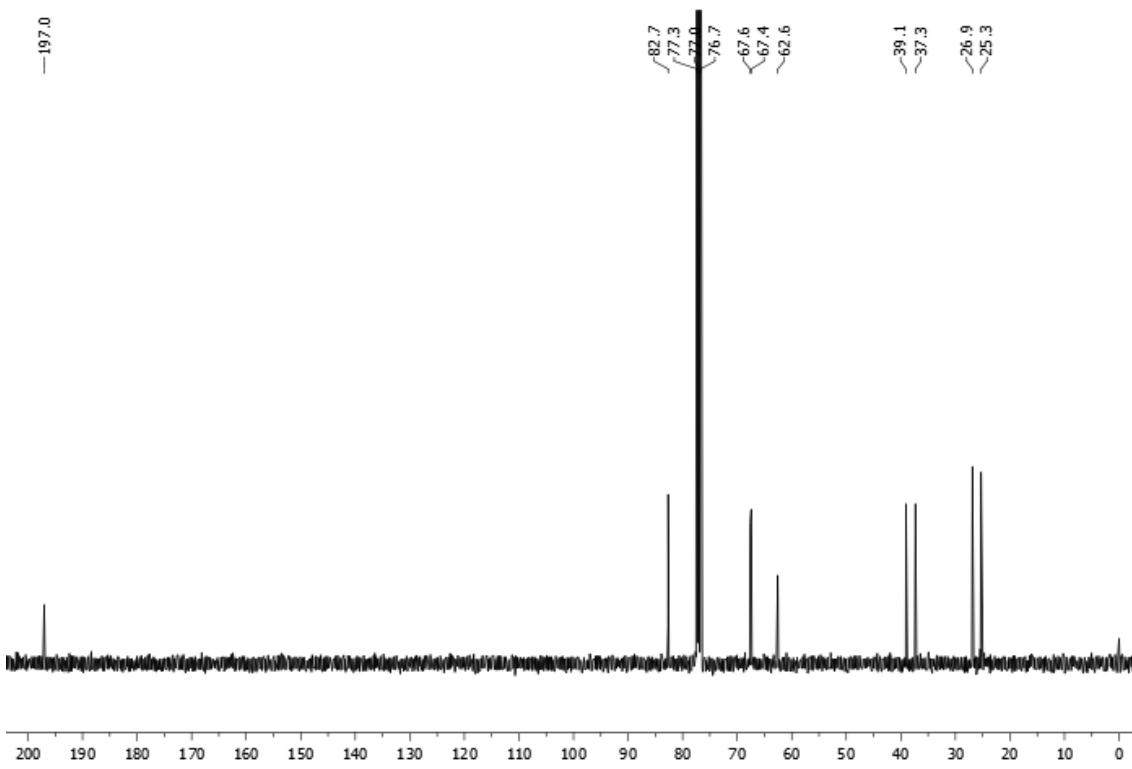


Figure S12: ^1H -NMR of **4c** in CDCl_3 (400 MHz)

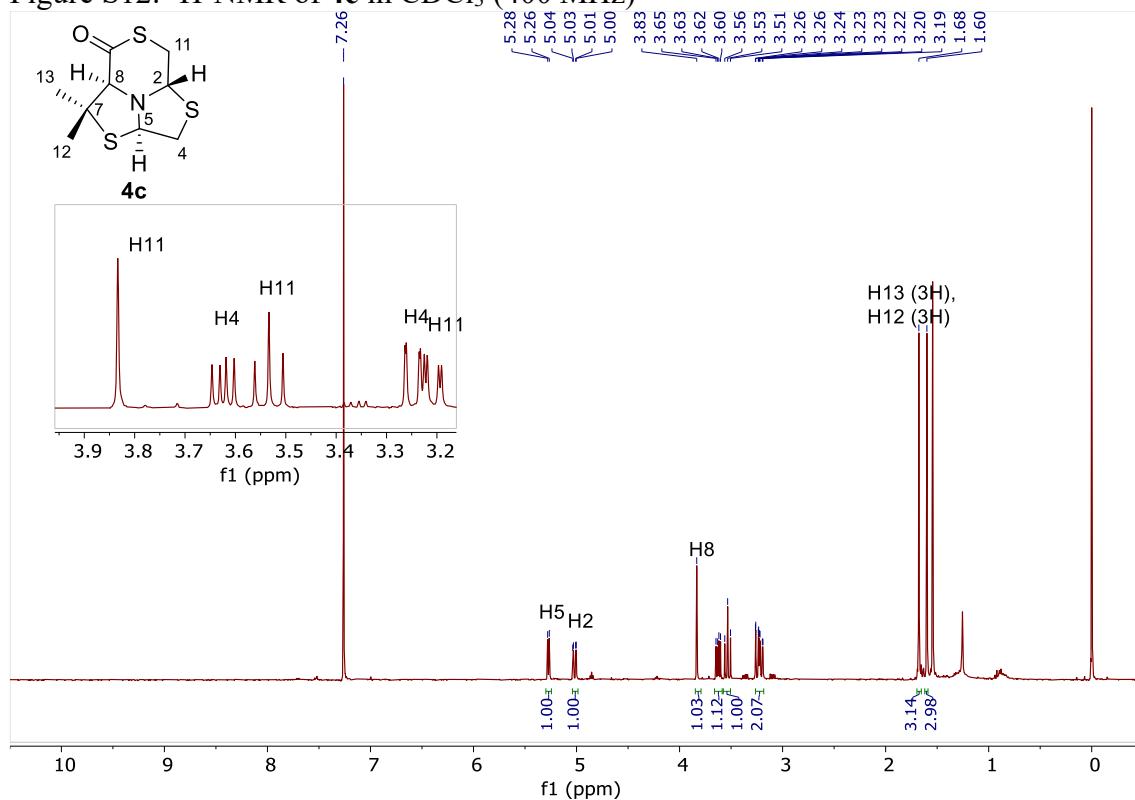
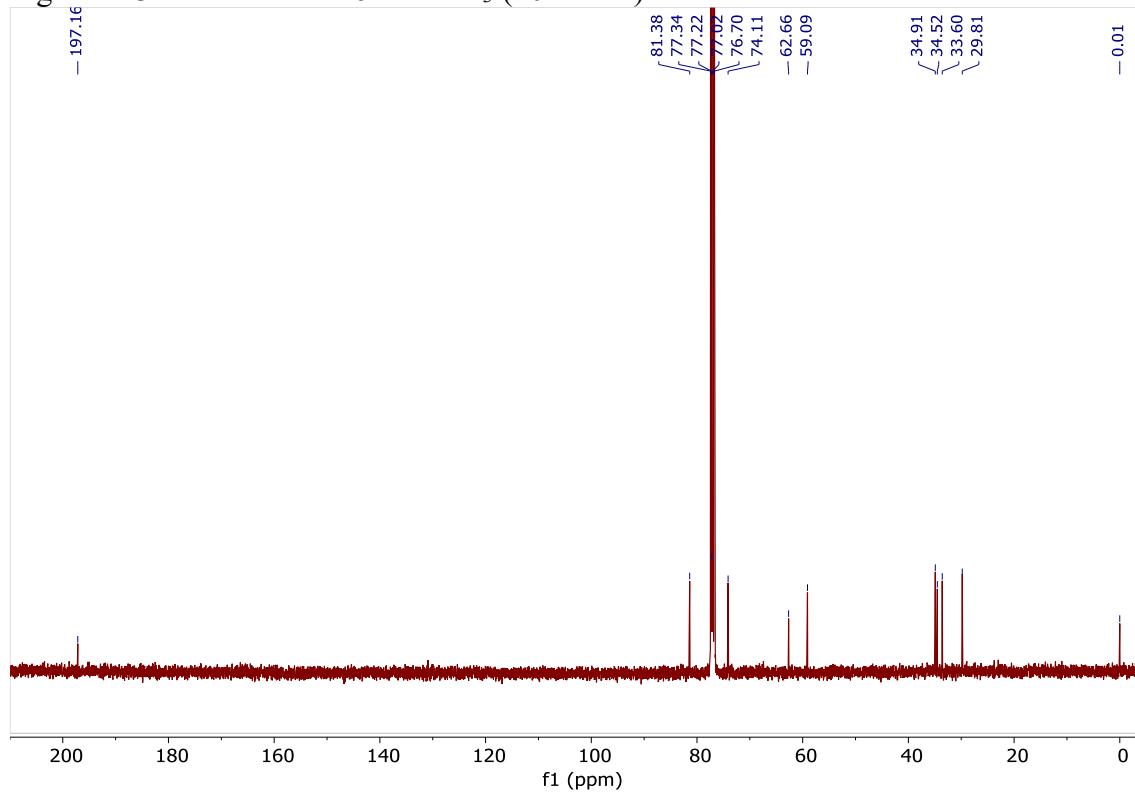


Figure S13: ^{13}C -NMR of **4c** in CDCl_3 (101 MHz)



V. 2D NOESY experiments of δ -thiolactones **1a-c** and **4a** and **4c**

Figure S14: 2D NOESY spectra of **1a**

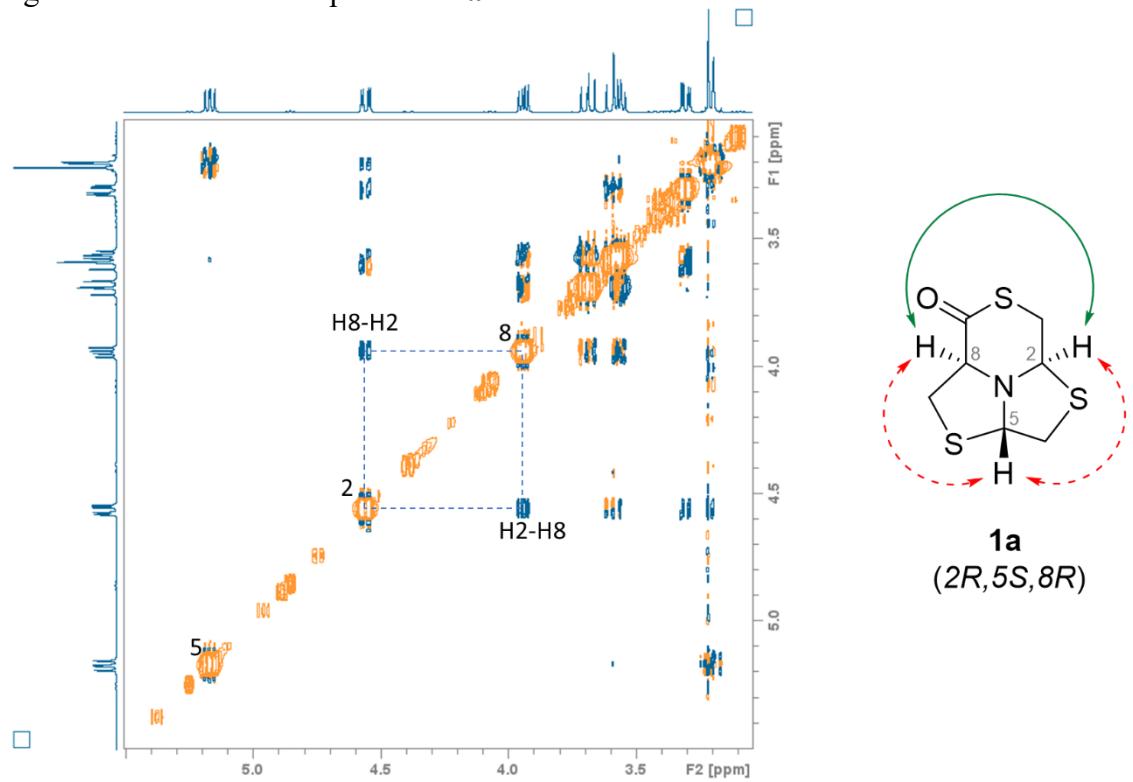


Figure S15: 2D NOESY spectra of **1b**

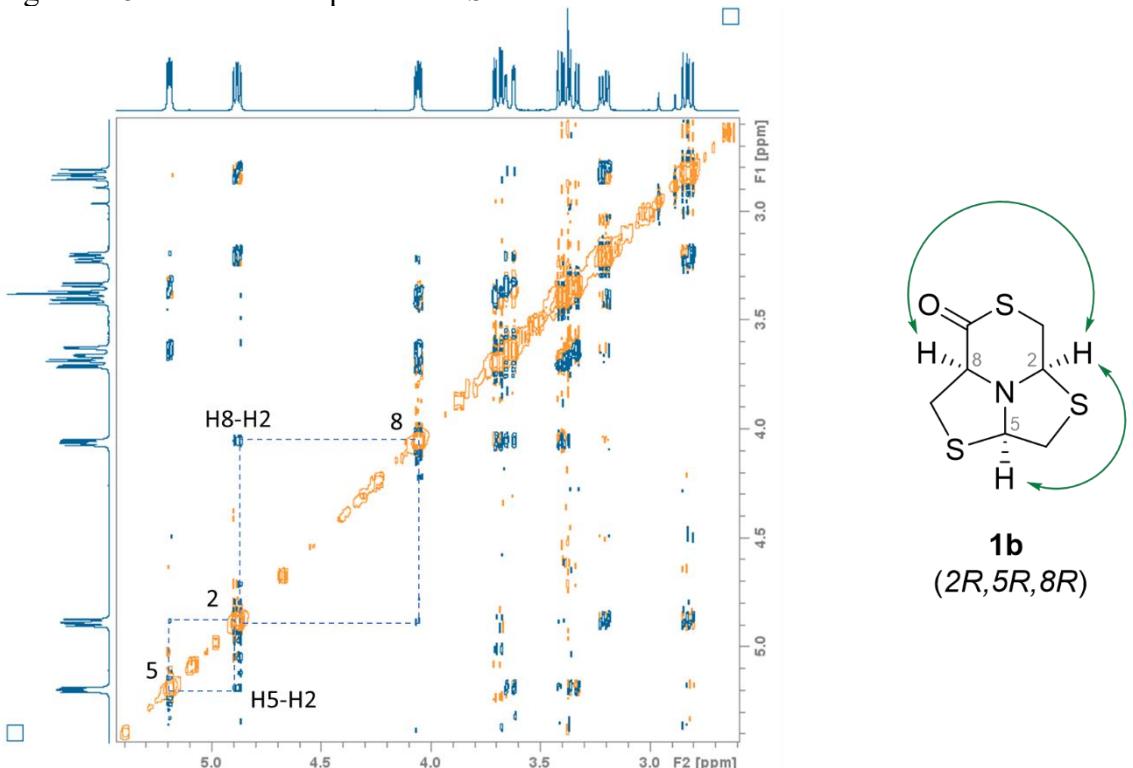


Figure S16: 2D NOESY spectra of **1c**

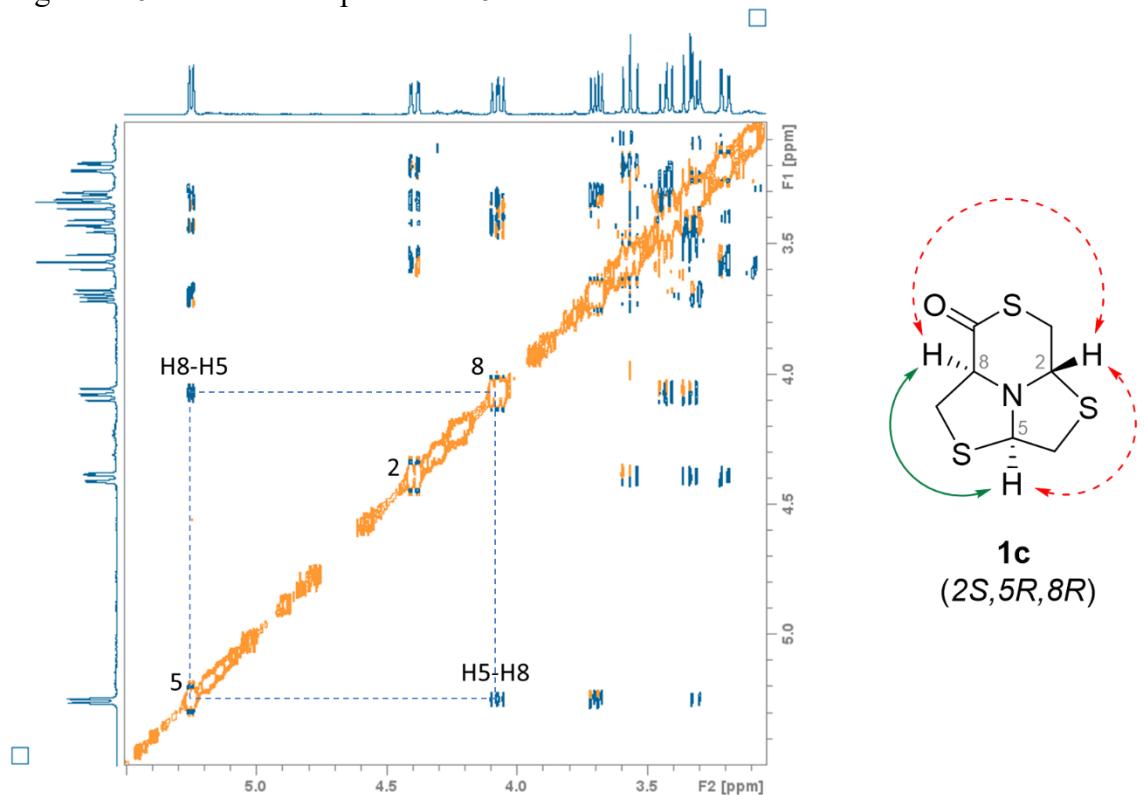


Figure S17: 2D NOESY spectra of **4a**

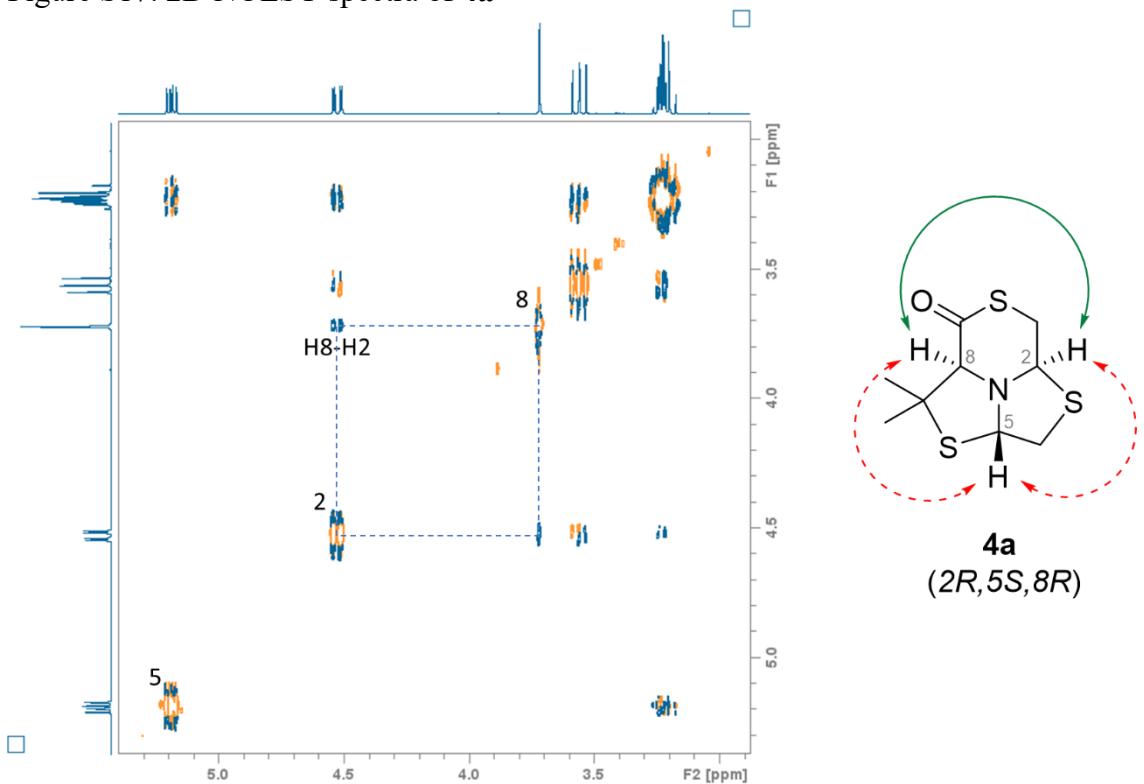
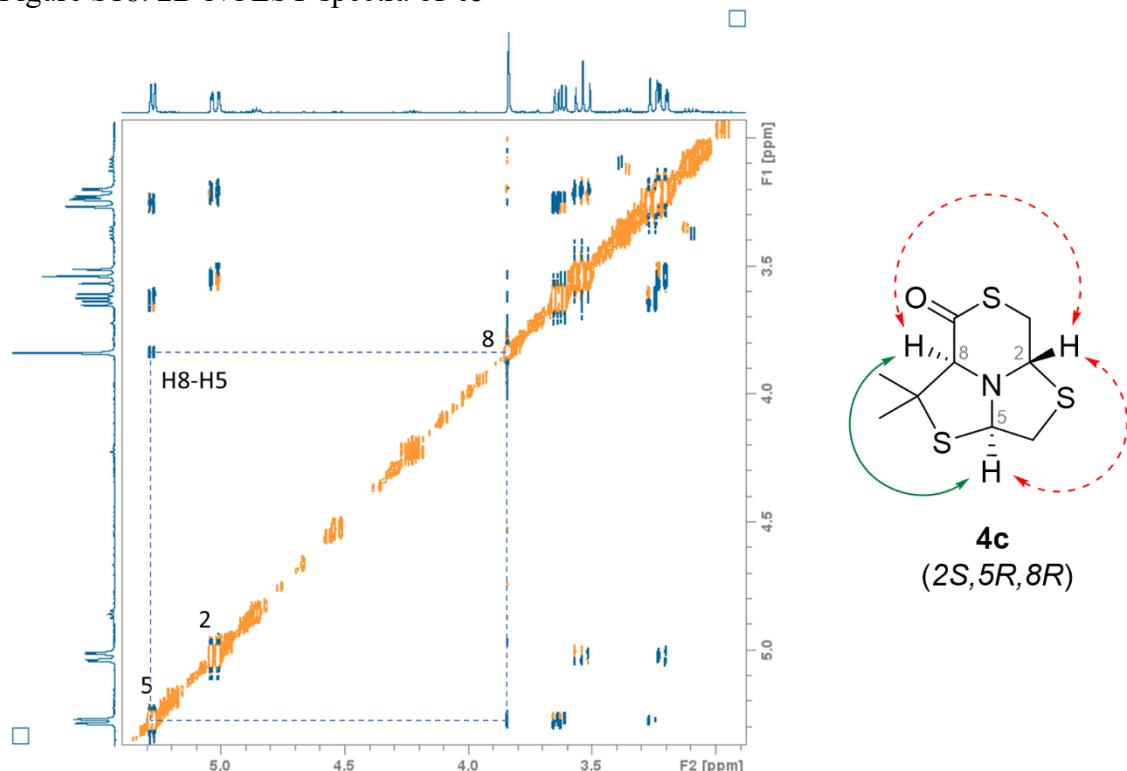


Figure S18: 2D NOESY spectra of **4c**



VI. Isomerization studies of δ -thiolactones 1

Figure S19: ^1H -NMR of time samples of isomerization reaction from **1a** to **1c** under DIPEA and microwave conditions.

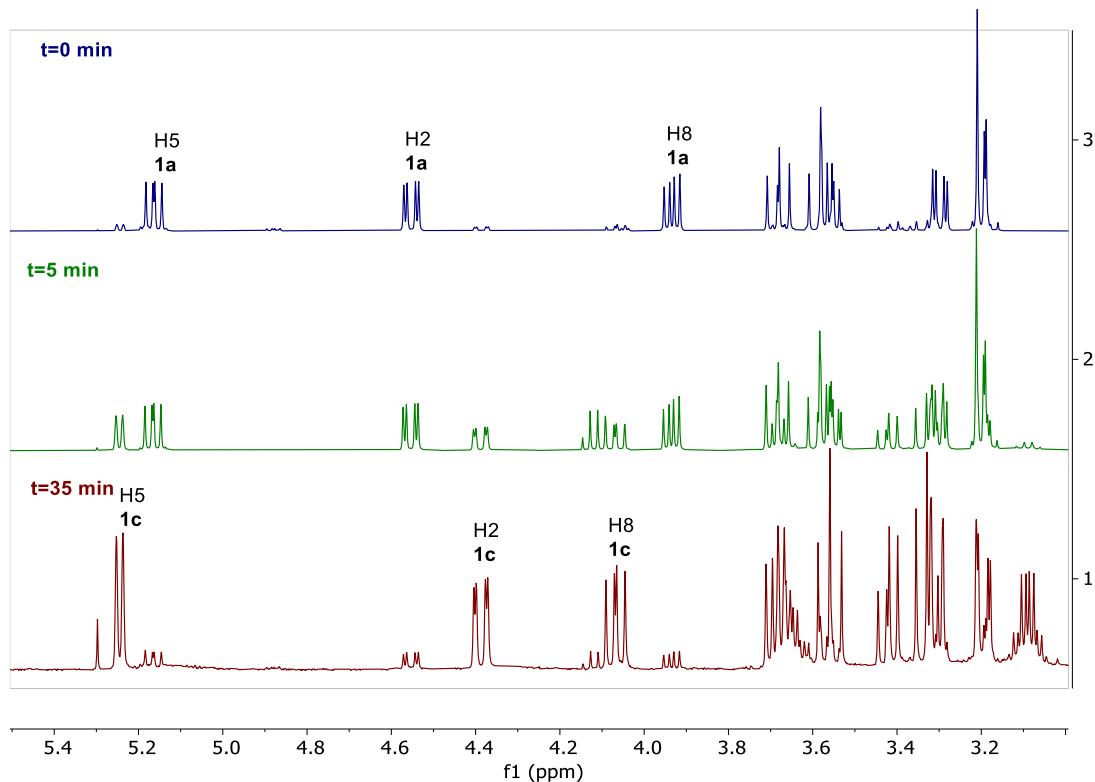
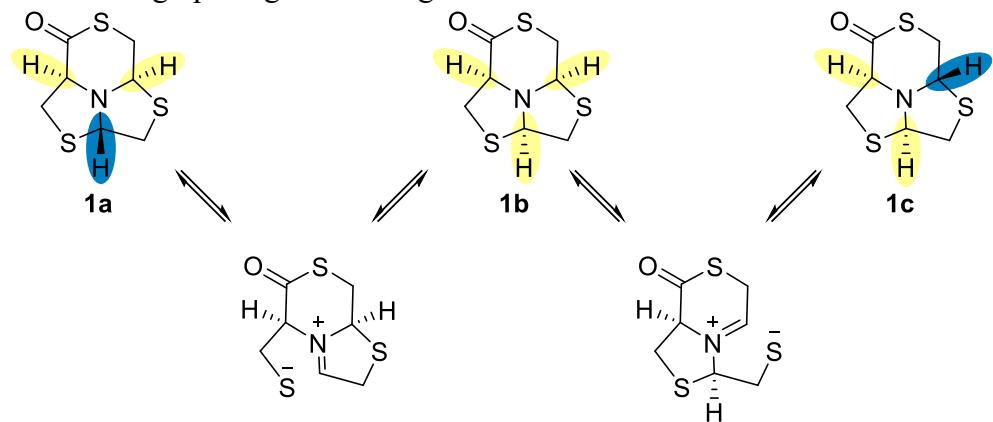


Figure S20: Proposed mechanism for the interconversion of thiolactone **1a** to **1c** through thiazolidine's ring opening and closing.



VII. NMR Spectra of δ -thiolactones opening products 5a-d, 6a-d, 8a, 9a-b and 10a-b

Figure S21: ^1H -NMR of **5a** in CDCl_3 (400 MHz)

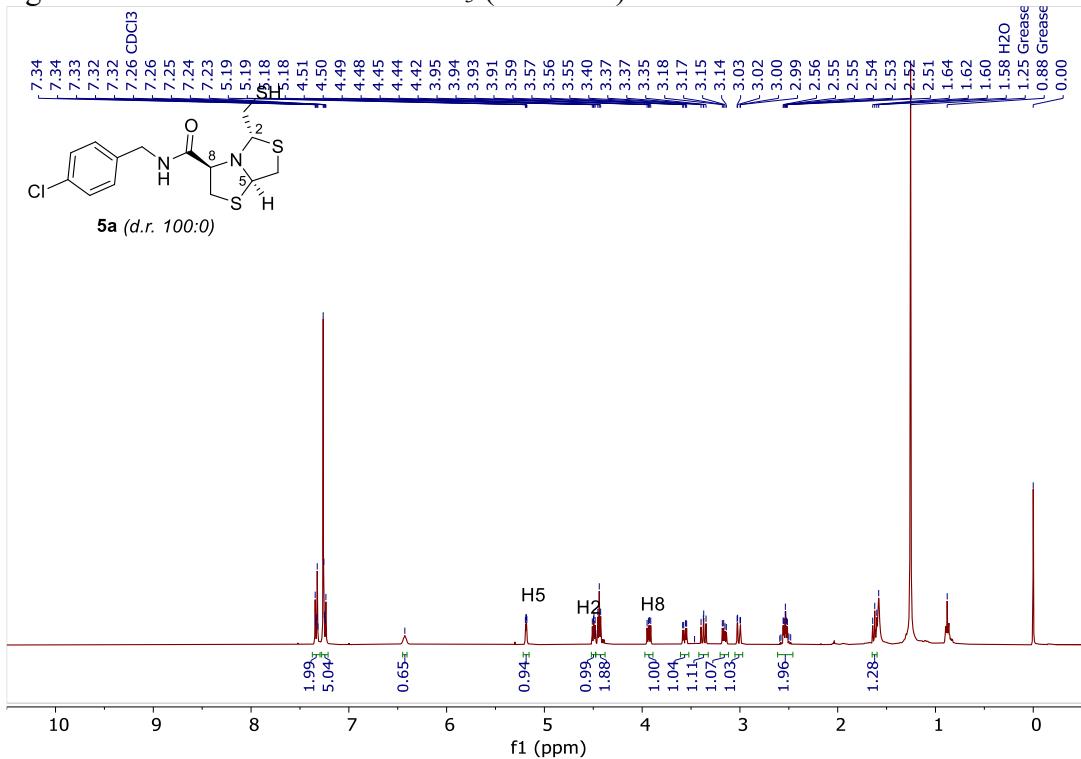


Figure S22: ^{13}C -NMR of **5a** in CDCl_3 (101 MHz)

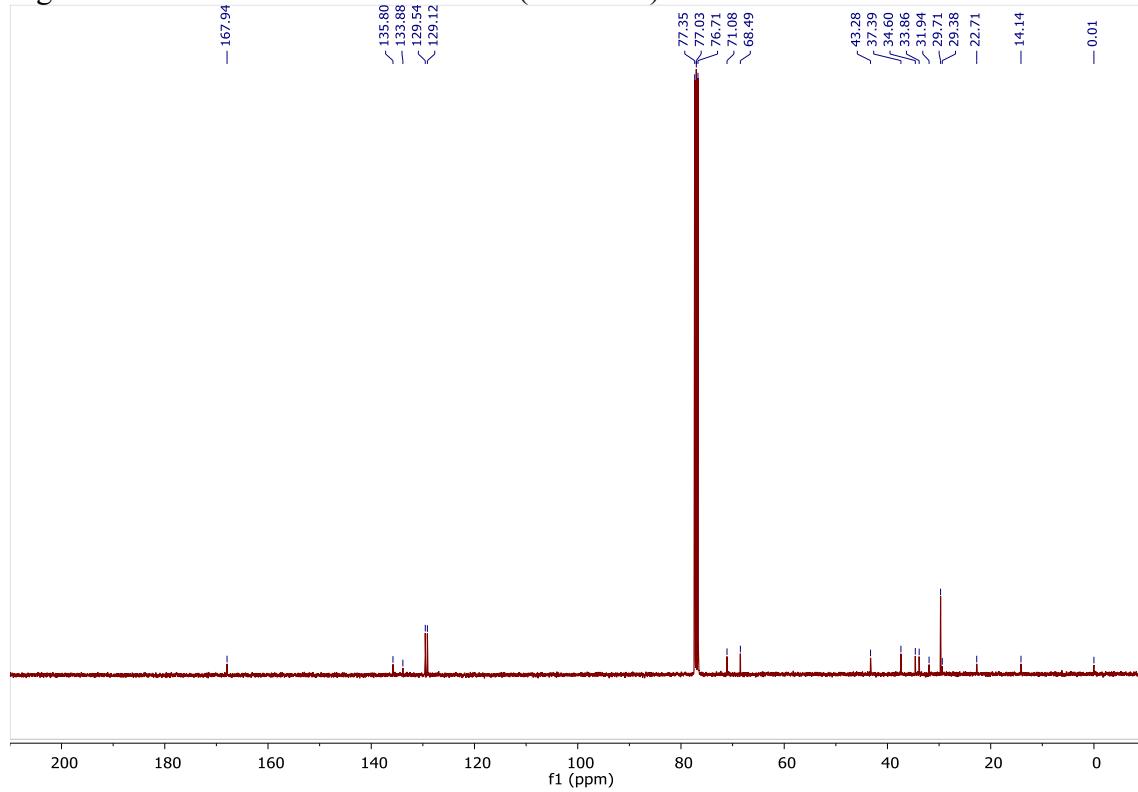


Figure S23: 2D NOESY NMR of **5a** in CDCl_3

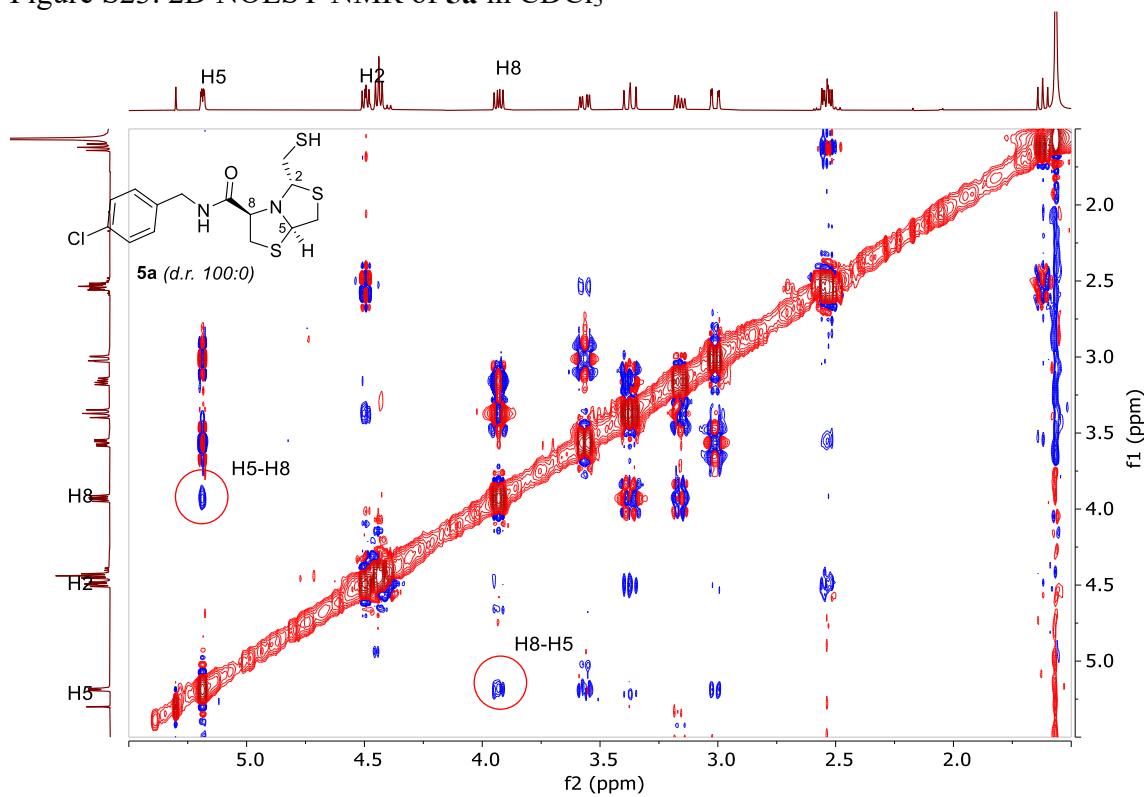


Figure S24: ^1H -NMR of **5b** in CDCl_3 (400 MHz)

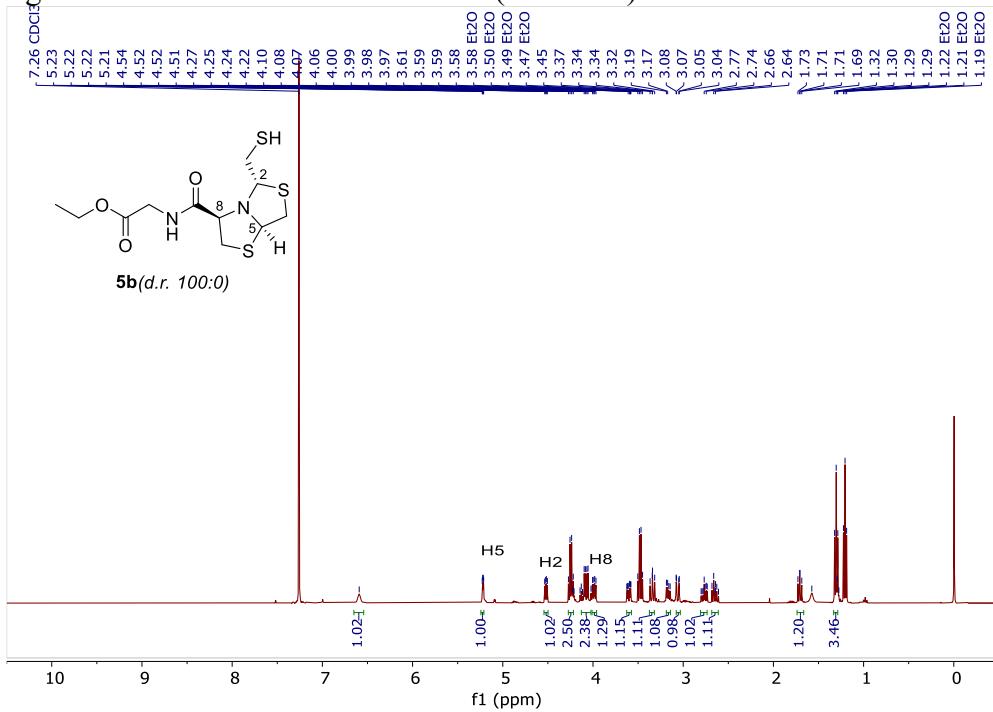


Figure S25: ^{13}C -NMR of **5b** in CDCl_3 (400 MHz) (the appearance of a second diastereomer could be observed)

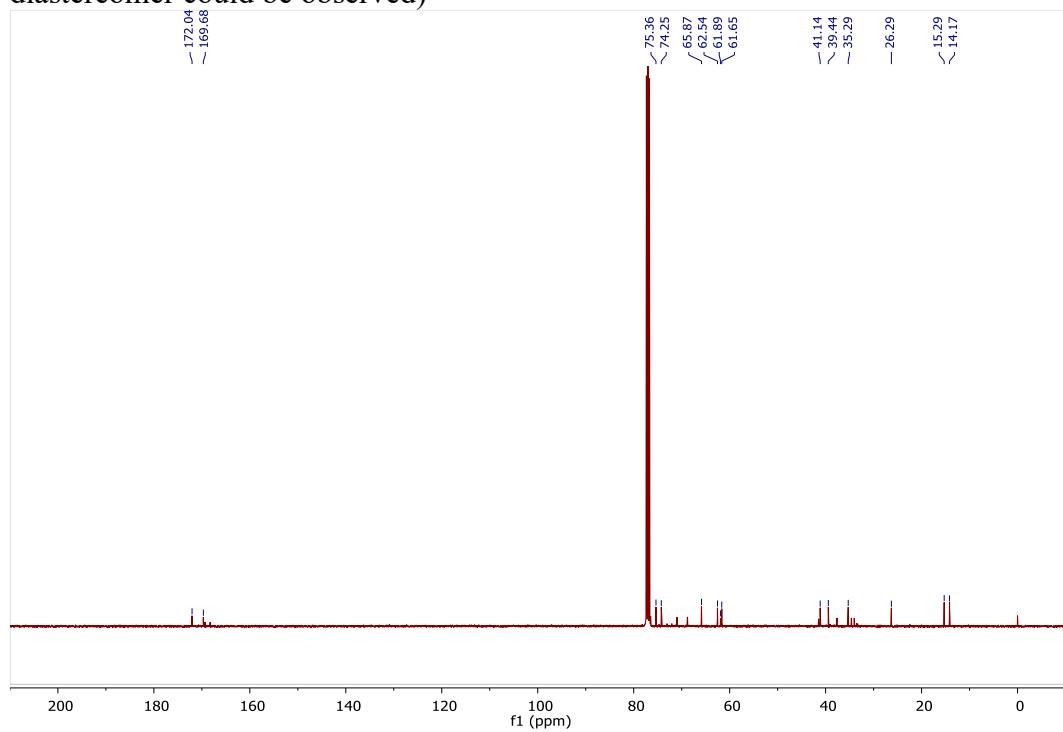


Figure S26: ^1H -NMR of **5b** in CDCl_3 after 24 h and removal of trace solvent (400 MHz)

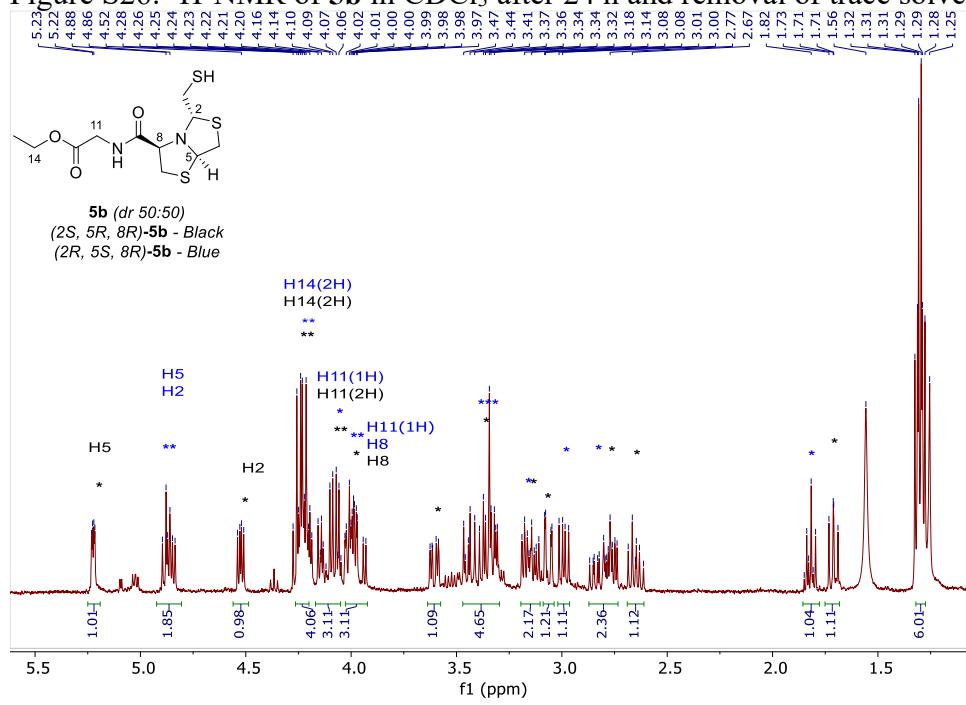


Figure S27: HSQC of diasteromers of 5b in CDCl₃ (after 24 h)

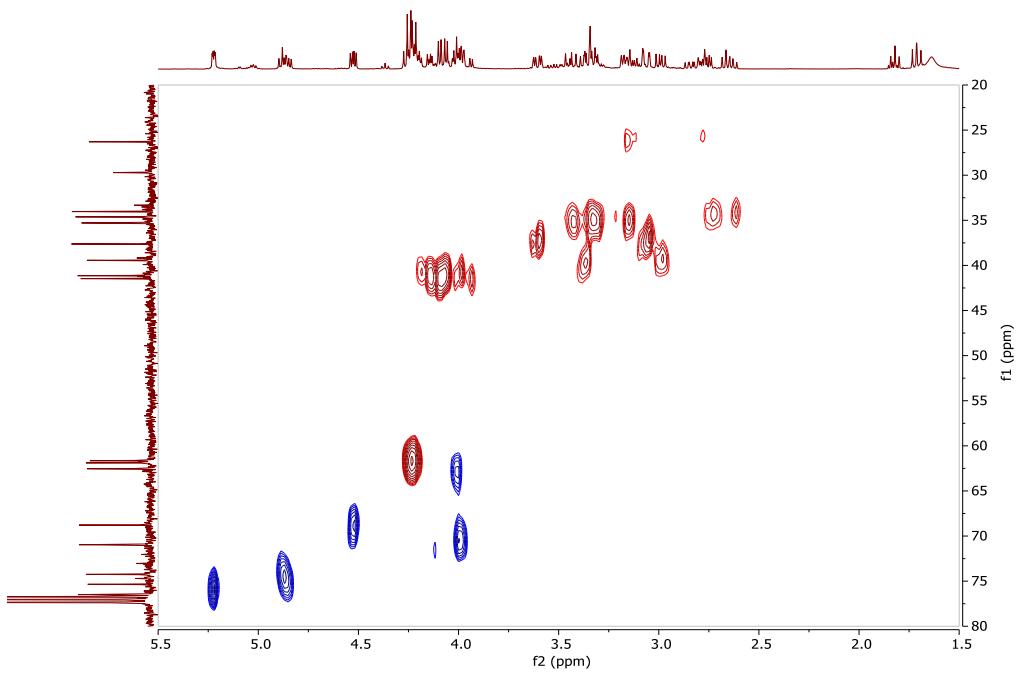


Figure S28: ^{13}C -NMR of diasteromers of **5b** in CDCl_3 after 24 h (400 MHz)

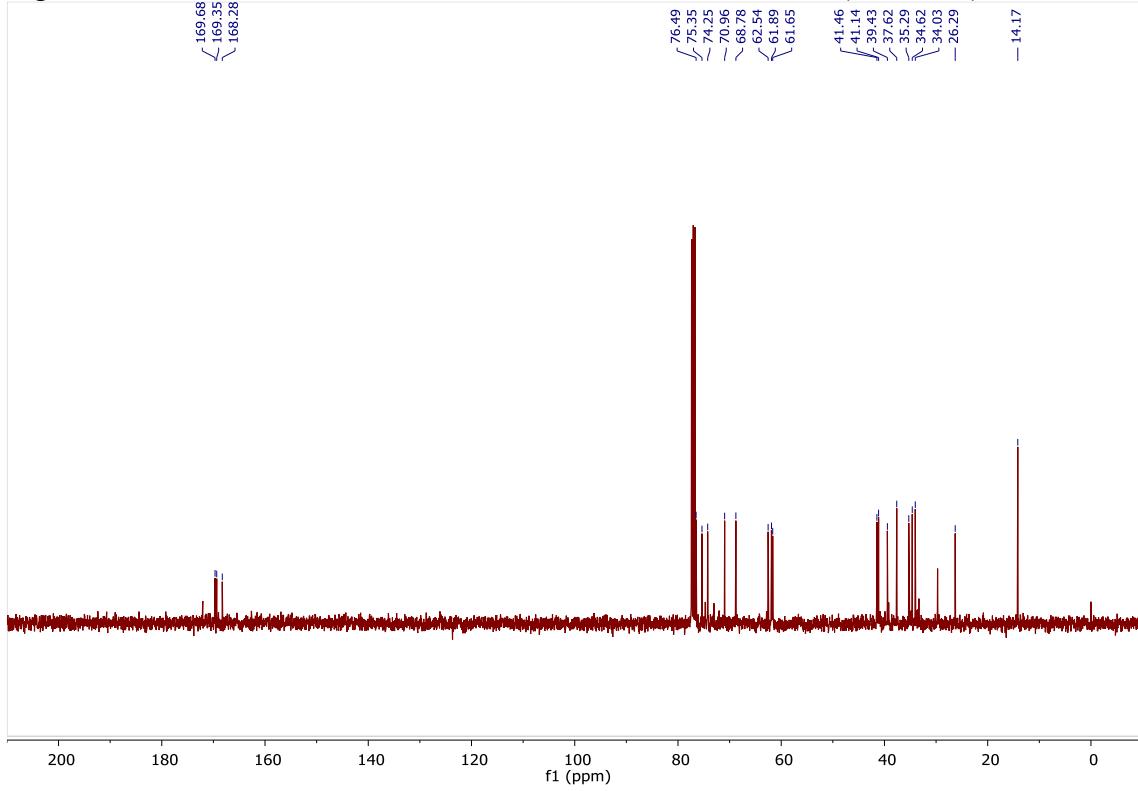


Figure S29: ^1H -NMR of **5c** in CDCl_3 (400 MHz)

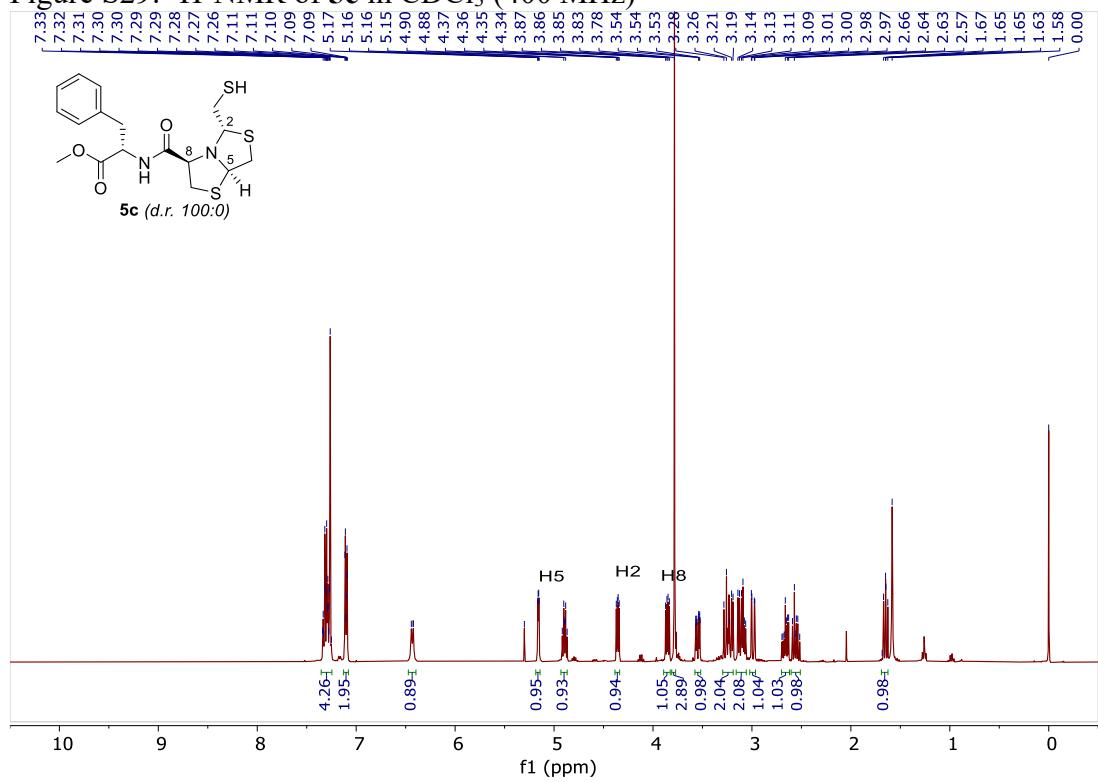


Figure S30: ^{13}C -NMR of **5c** in CDCl_3 (400 MHz)

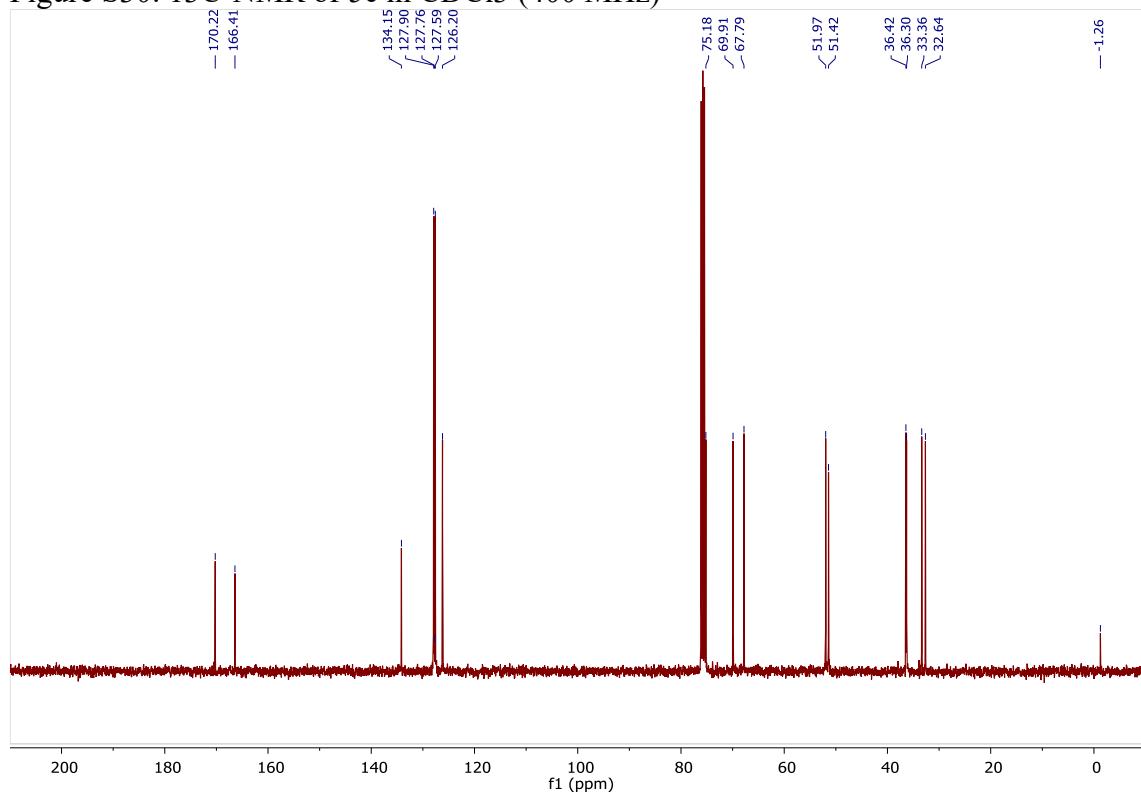


Figure S31: 2D NOESY NMR of **5c** in CDCl_3

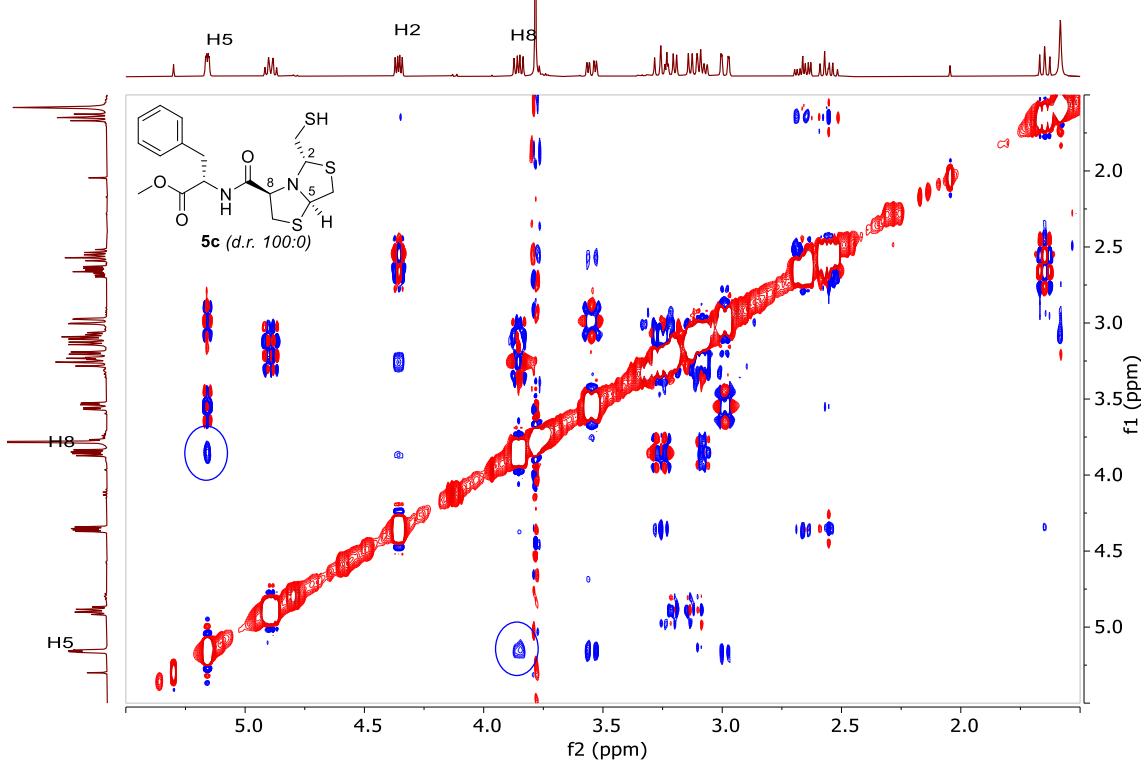


Figure S32: ^1H -NMR of **5d** in CDCl_3 (400 MHz)

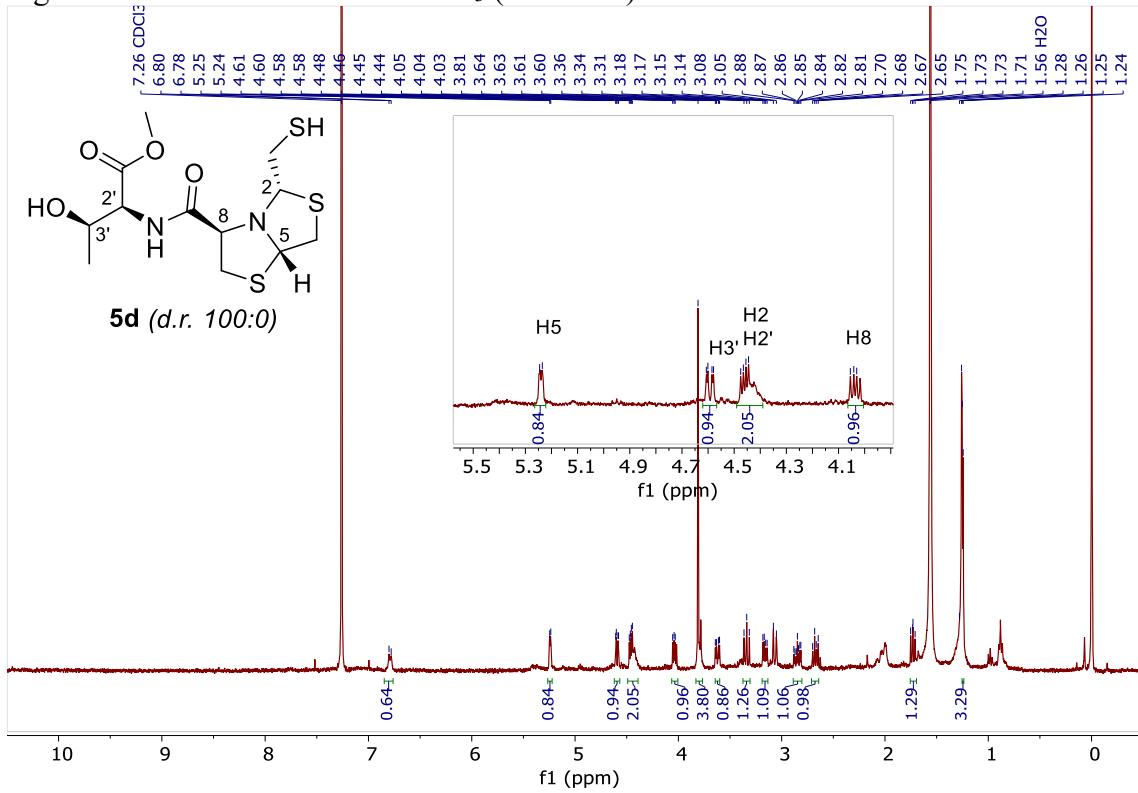


Figure S33: ^{13}C -NMR of **5d** in CDCl_3 (400 MHz)

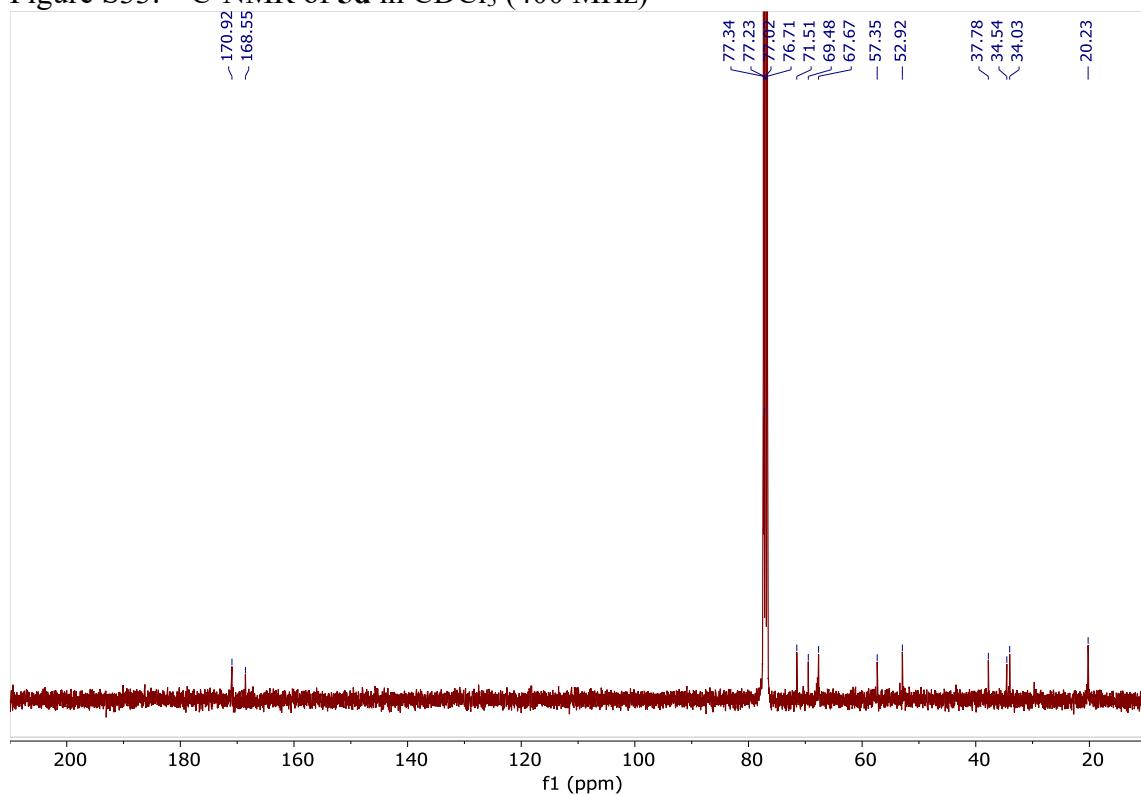


Figure S34: ^1H -NMR of **6a** in CDCl_3 (400 MHz)

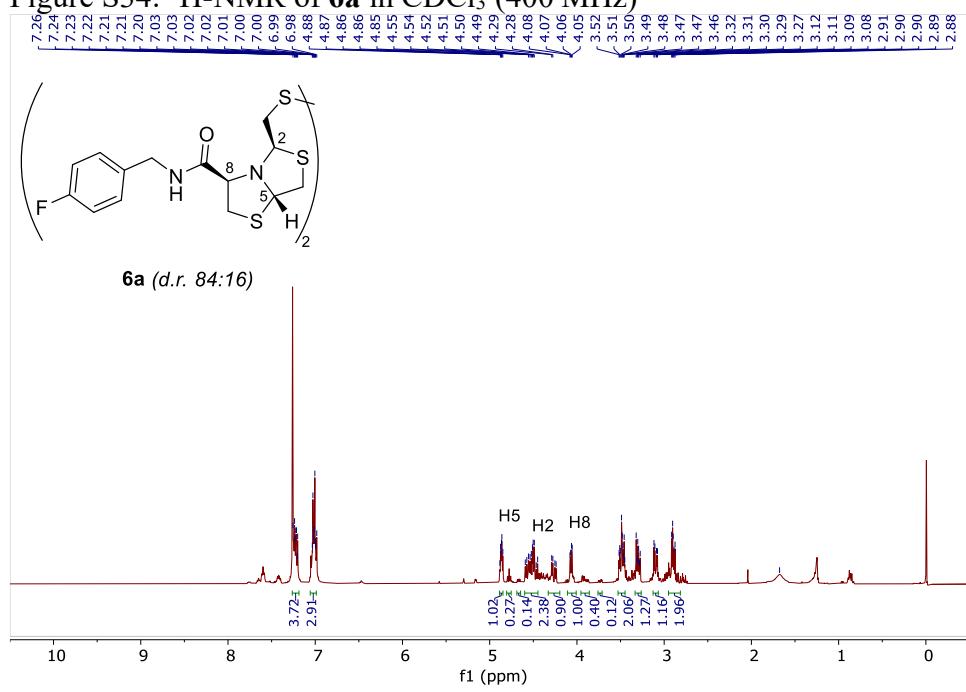


Figure S35: ^{13}C -NMR of **6a** in CDCl_3 (101 MHz)

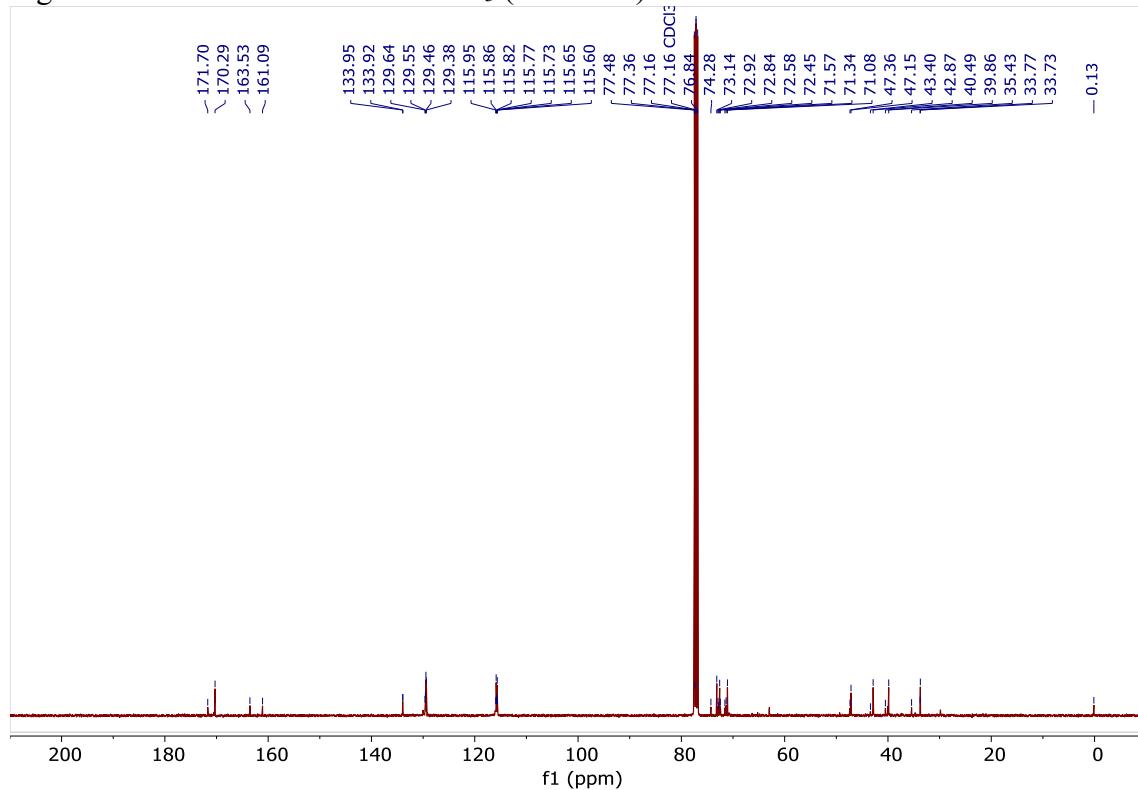


Figure S36: 1D NOESY NMR of **6a** in CDCl_3

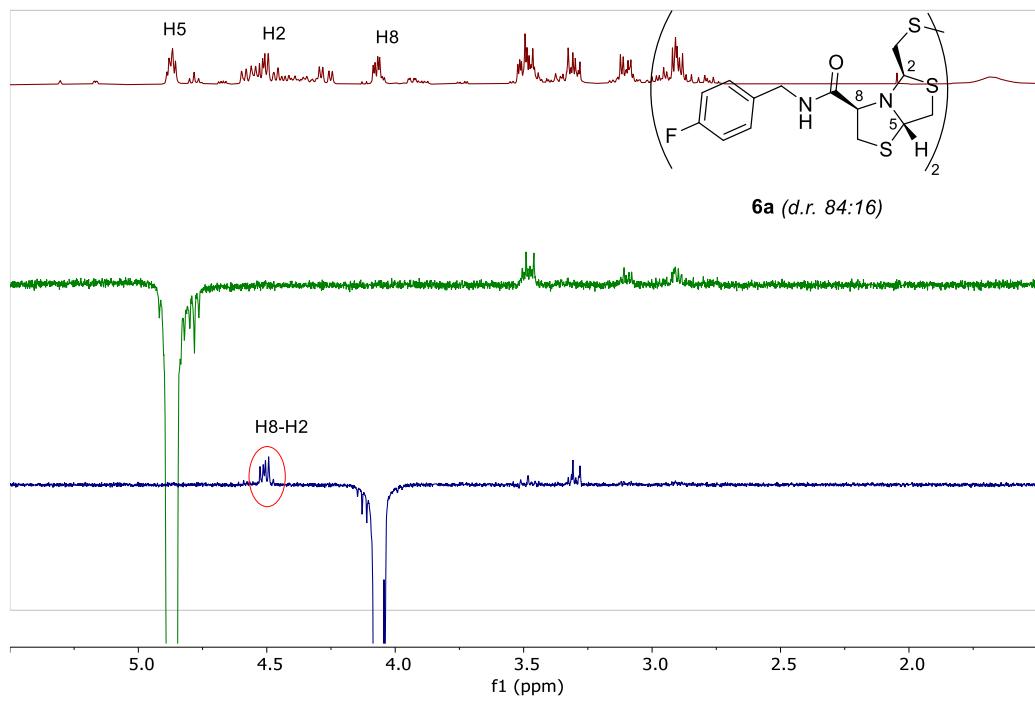


Figure S37: ^1H -NMR of **6b** in CDCl_3 (400 MHz)

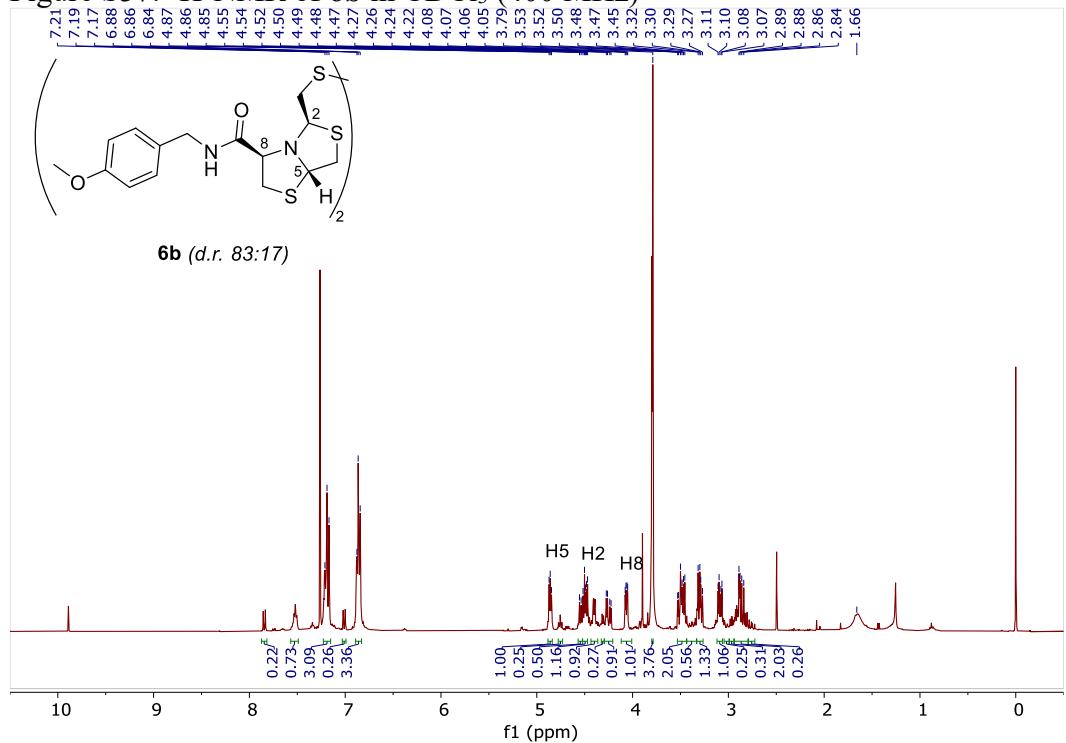


Figure S38: ^{13}C -NMR of **6b** in CDCl_3 (101 MHz)

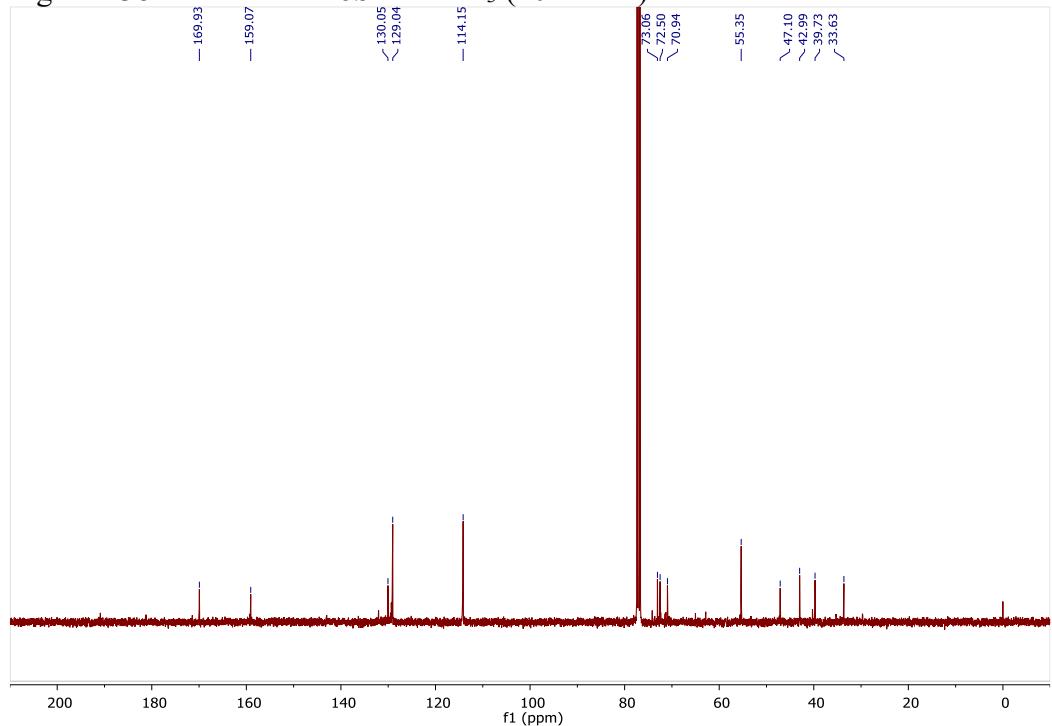


Figure S39: 1D NOESY NMR of **6b** in CDCl_3

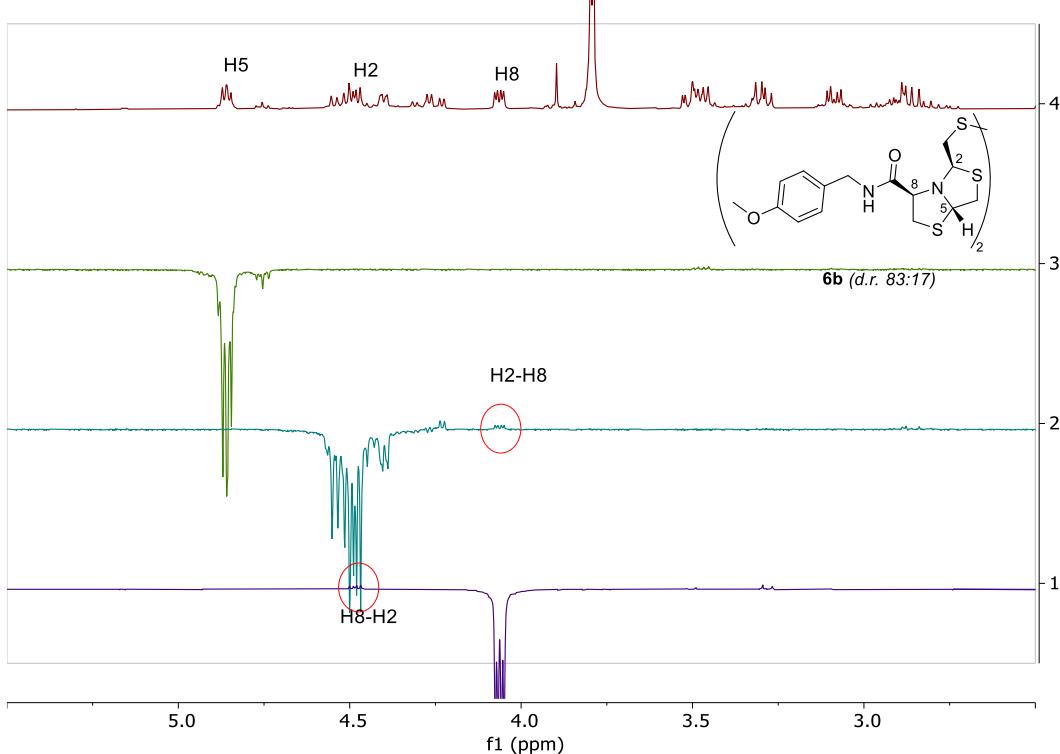


Figure S40: ^1H -NMR of **6c** in CDCl_3 (400 MHz)

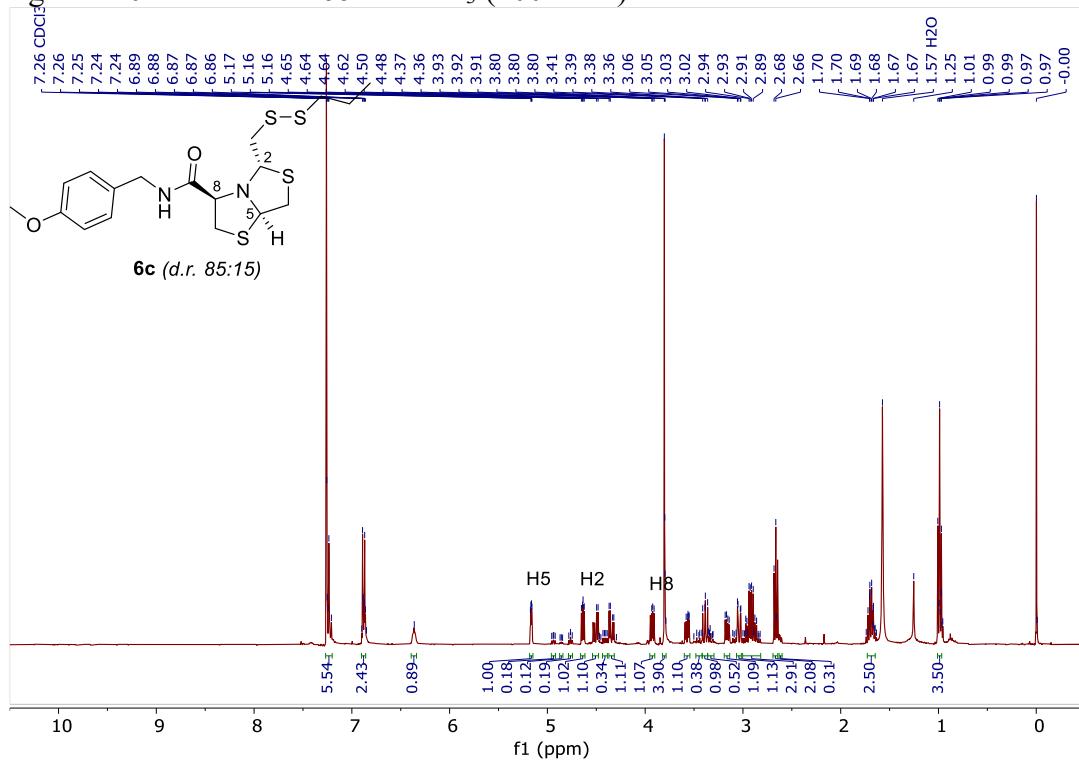


Figure S41: ^{13}C -NMR of **6c** in CDCl_3 (101 MHz)

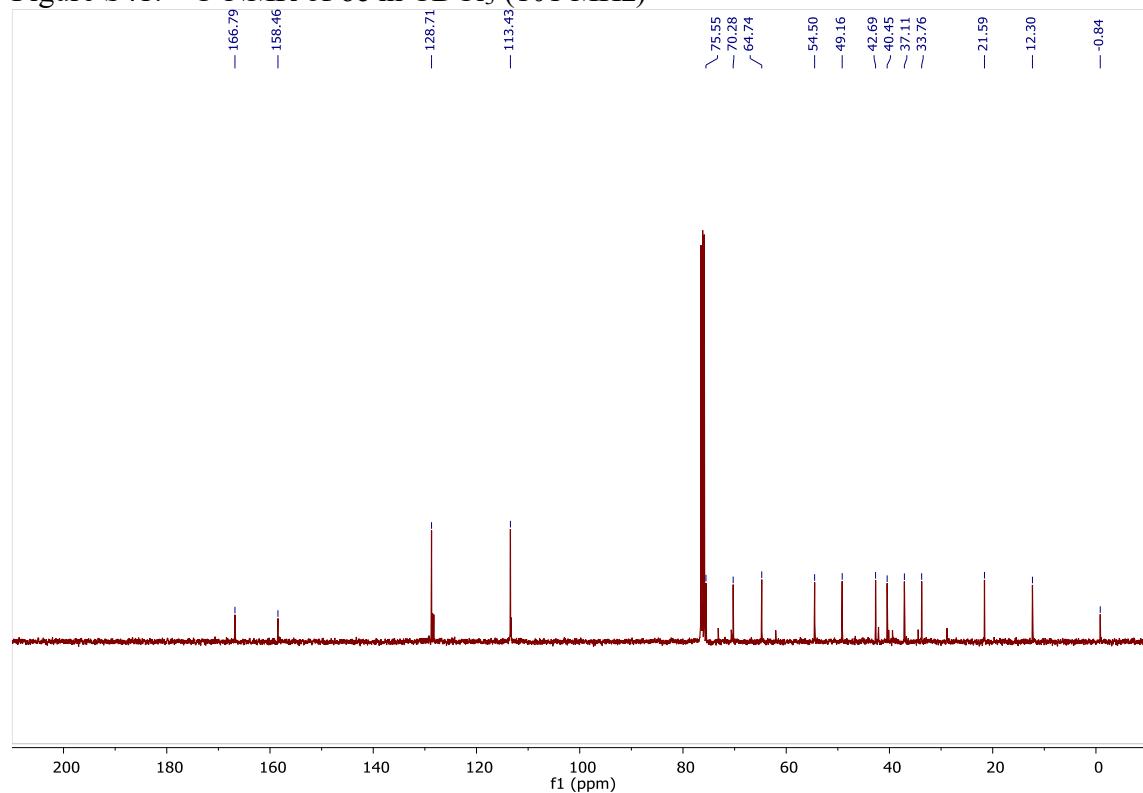


Figure S42: 1D NOESY NMR of **6c** in CDCl_3

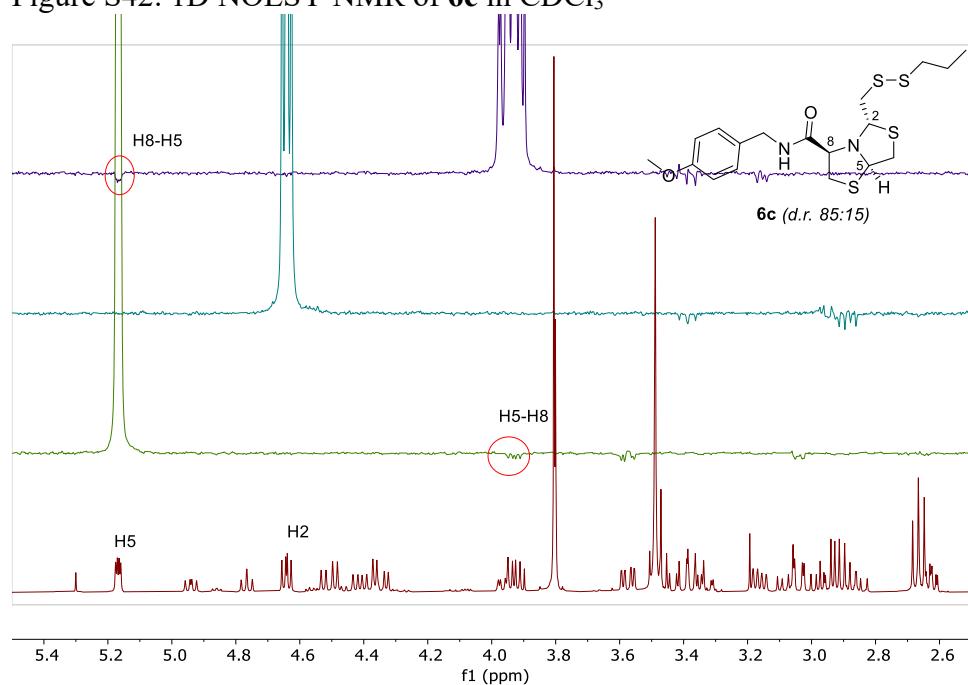


Figure S43: ^1H -NMR of **6d** idn CDCl_3 (400 MHz)

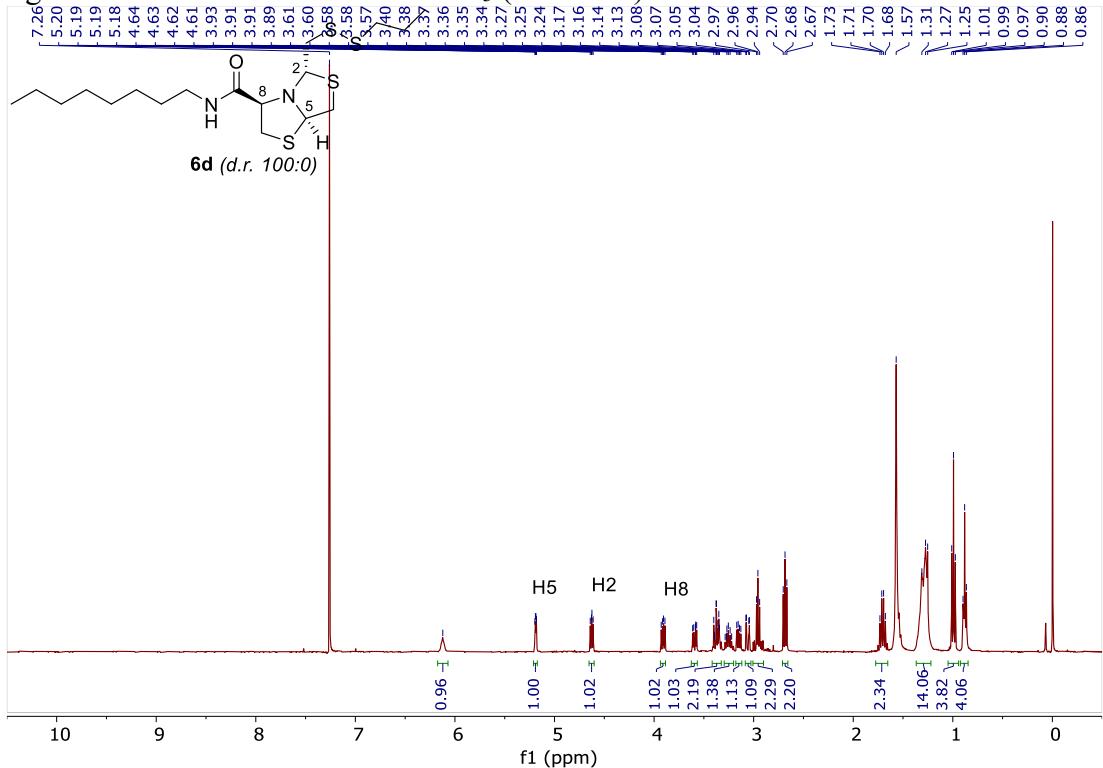


Figure S44: ^{13}C -NMR of **6d** in CDCl_3 (101 MHz)

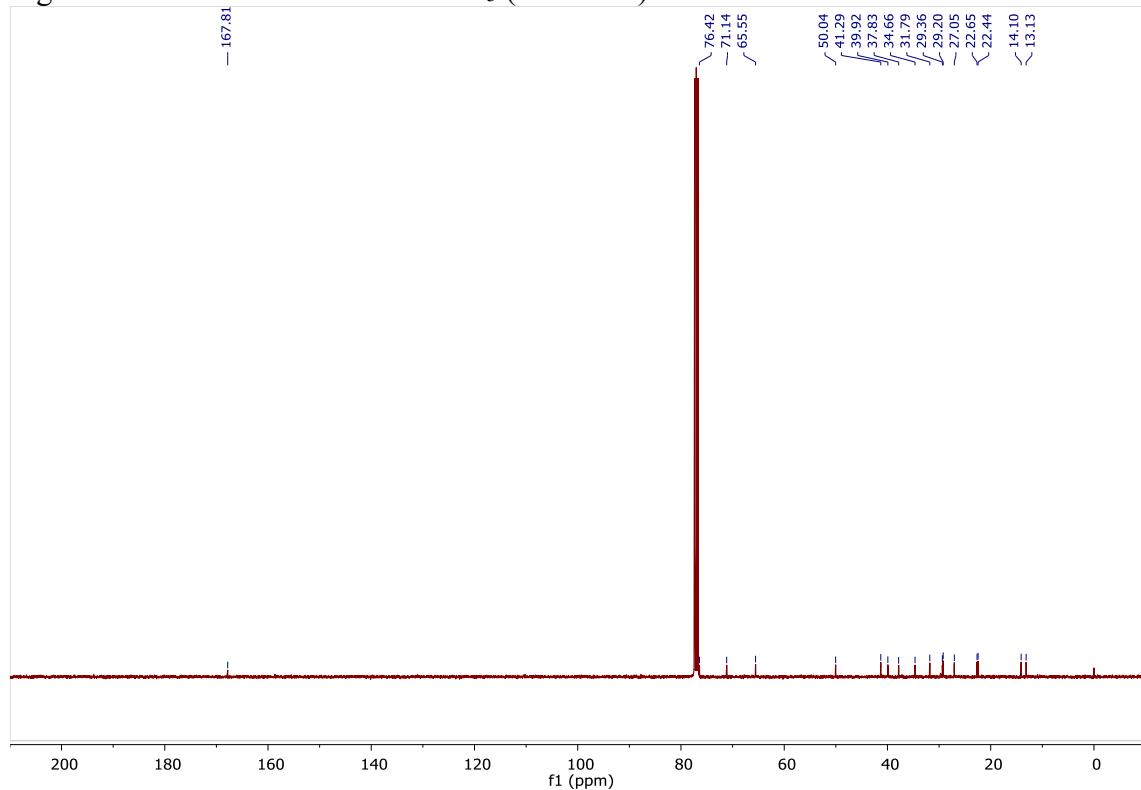


Figure S45: 2D NOESY NMR of **6d** in CDCl_3

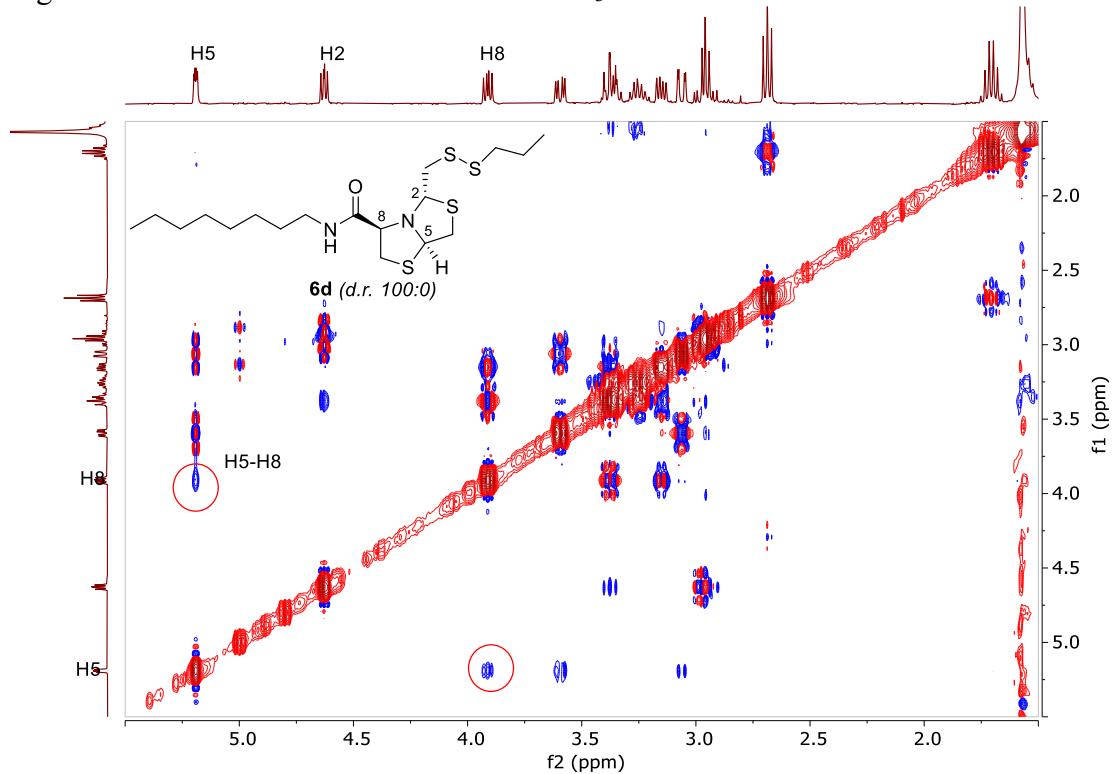


Figure S46: ^1H -NMR of **6e** in CDCl_3 (400 MHz)

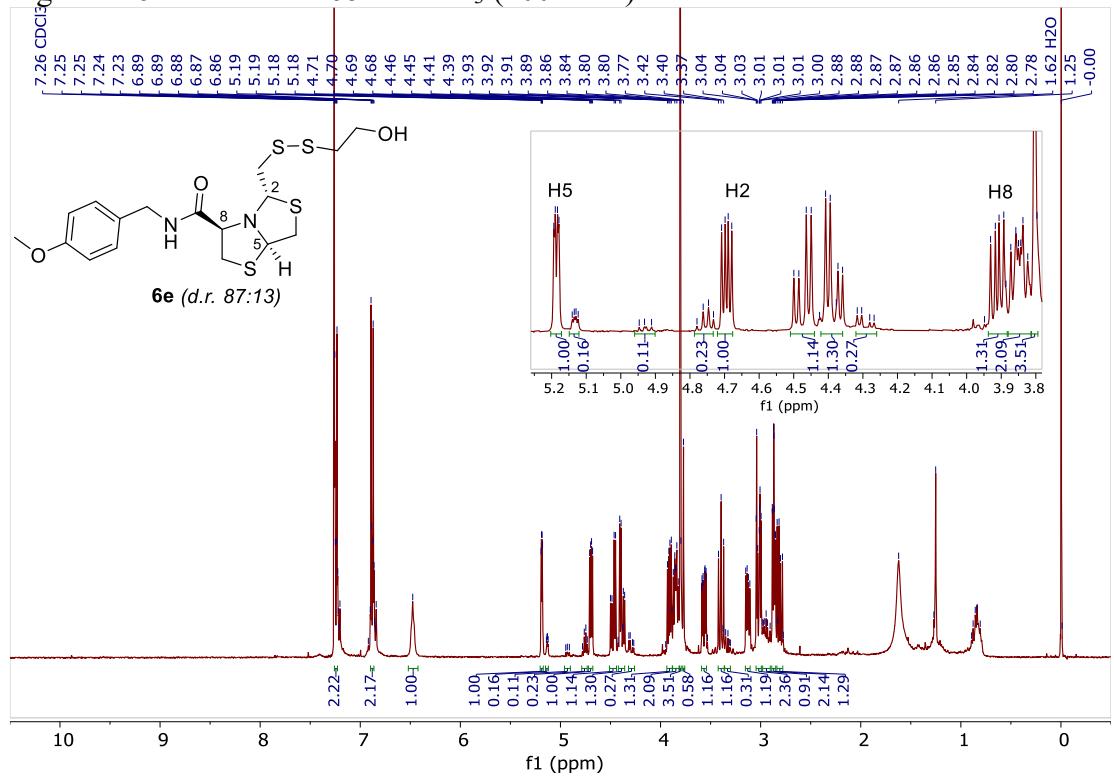


Figure S47: ^{13}C -NMR of **6e** in CDCl_3 (101 MHz)

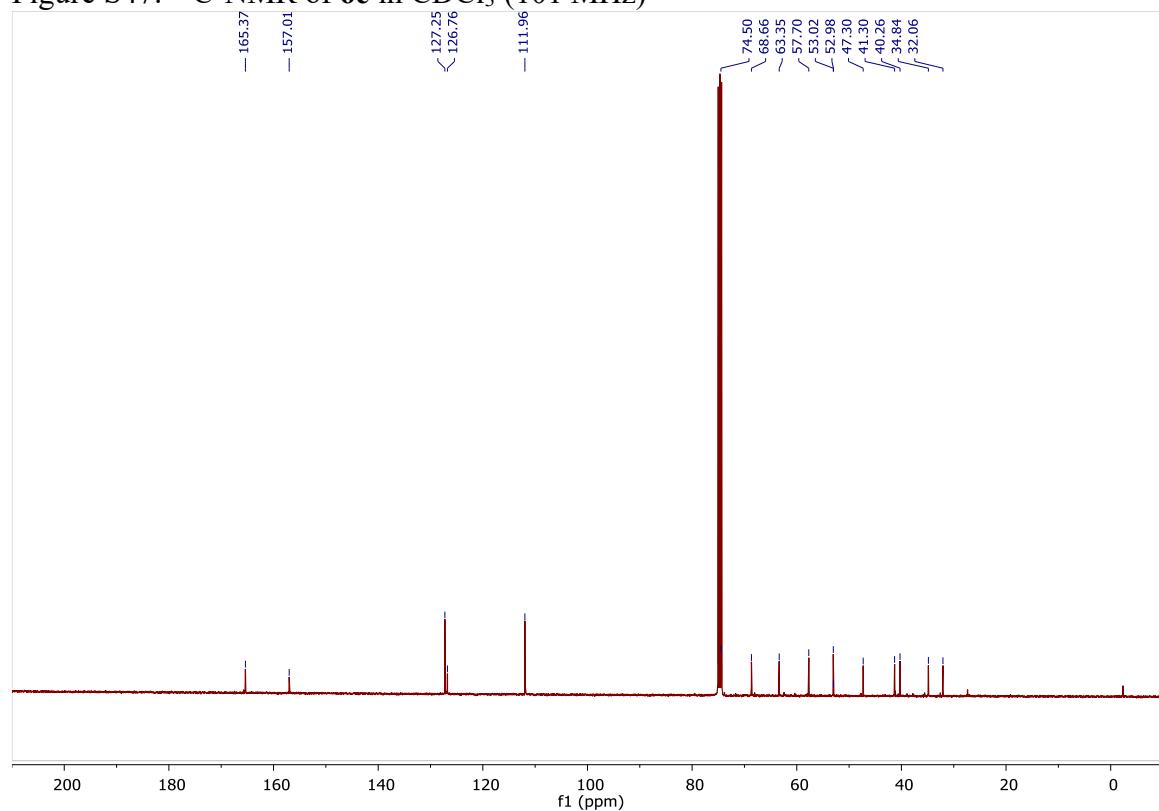


Figure S48: 2D NOESY NMR of **6e** in CDCl_3

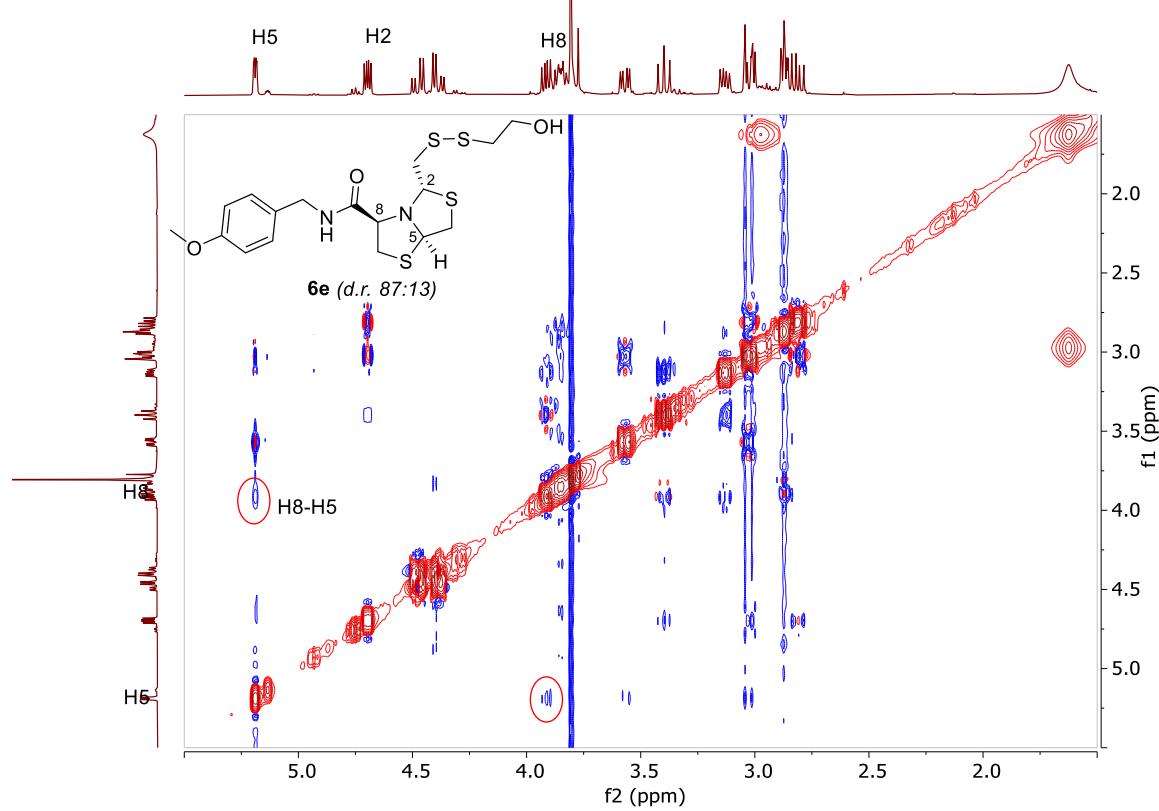


Figure S49: ^1H -NMR of **8a** in CDCl_3 (400 MHz)

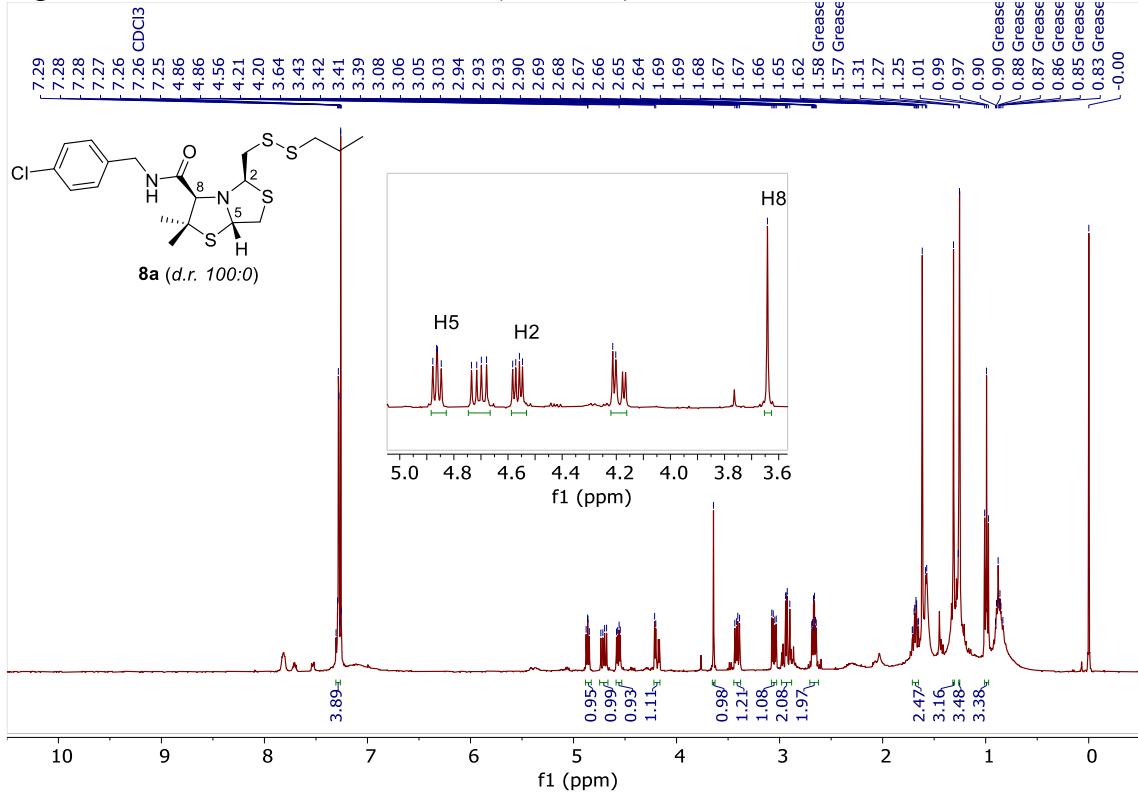


Figure S50: ^{13}C -NMR of **8a** in CDCl_3 (101 MHz)

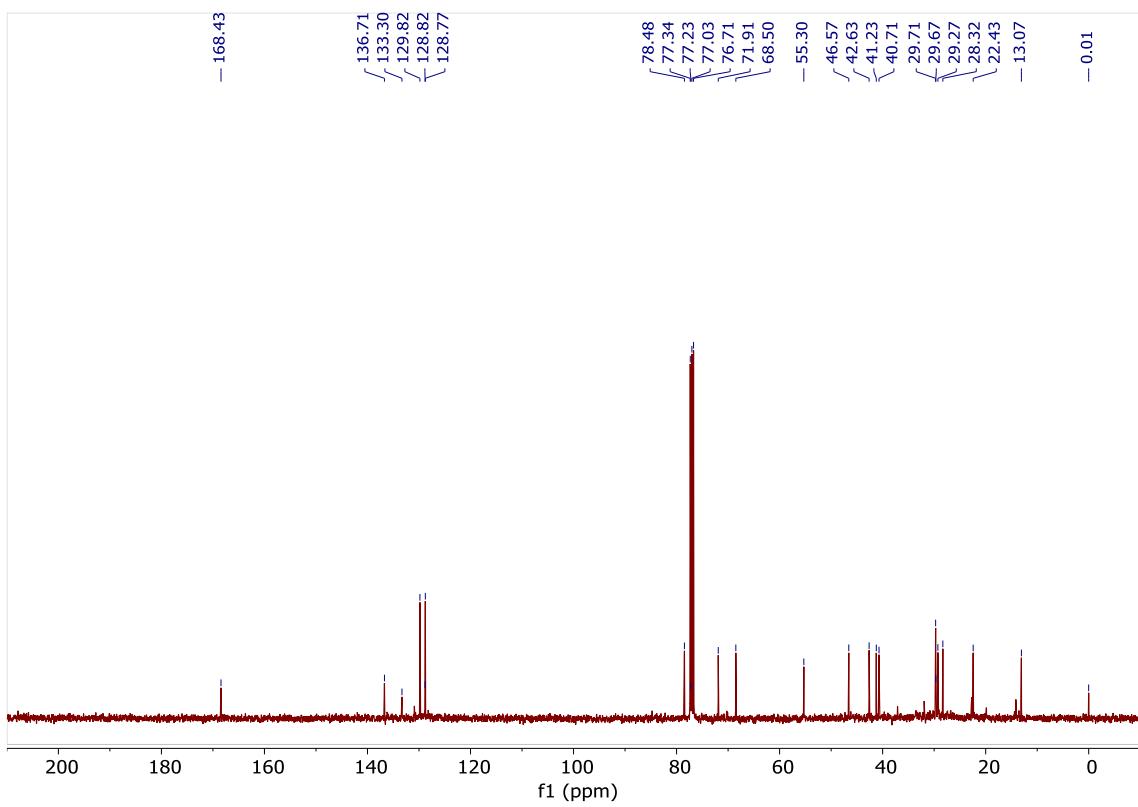


Figure S51: 2D NOESY NMR of **8a** in CDCl_3

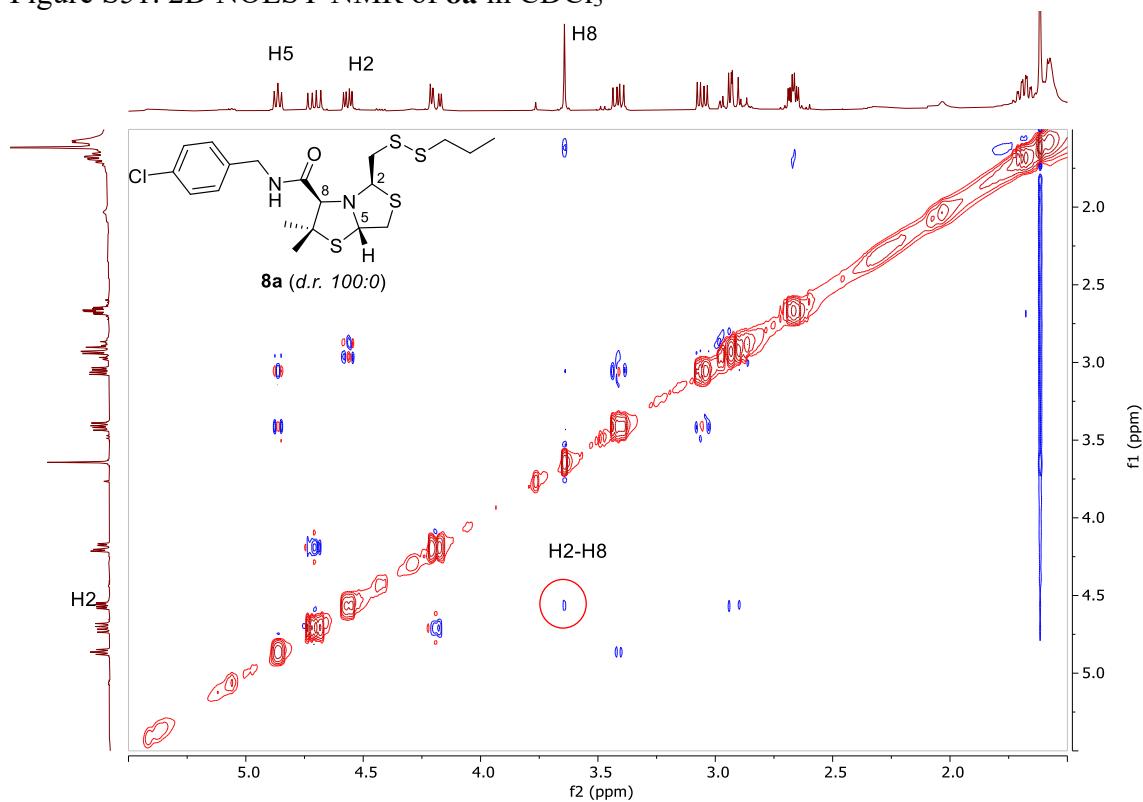


Figure S52: ^1H -NMR of **9a** in CDCl_3 (400 MHz)

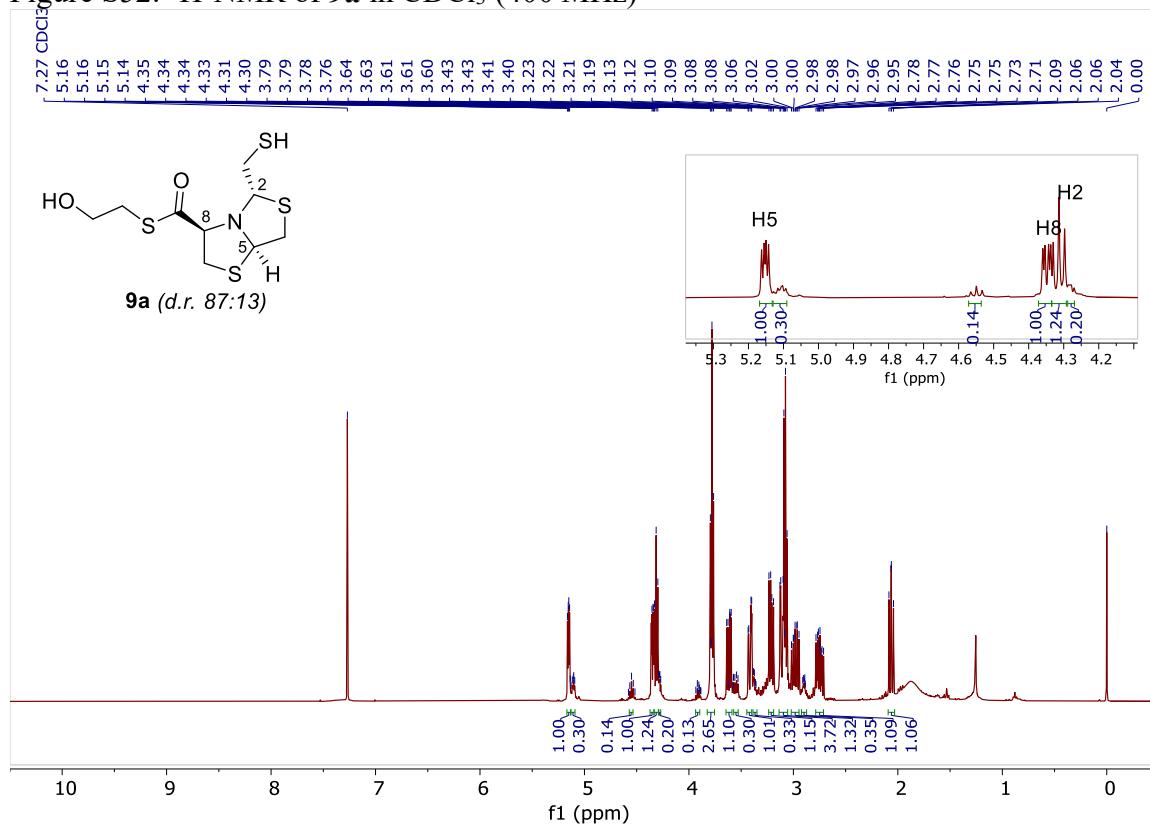


Figure S53: ^{13}C -NMR of **9a** in CDCl_3 (101 MHz)

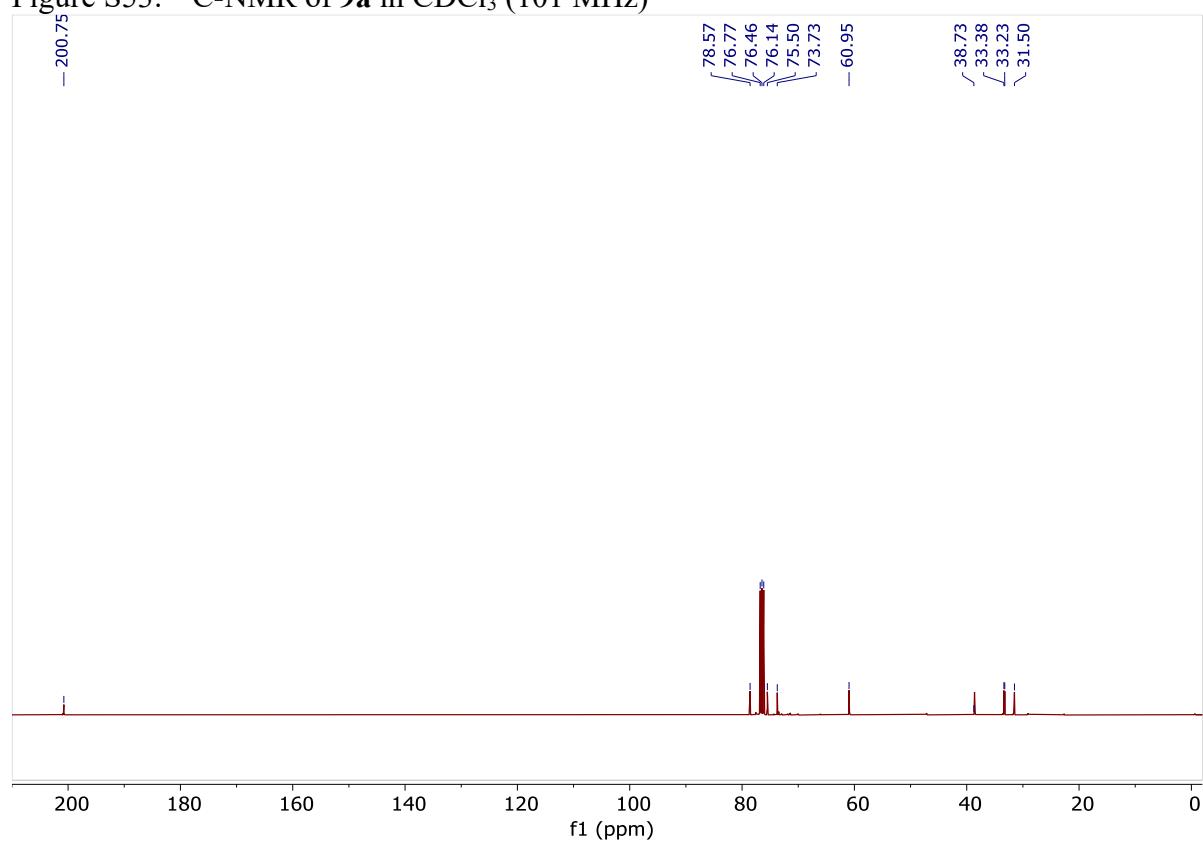


Figure S54: 2D NOESY NMR of **9a** in CDCl_3

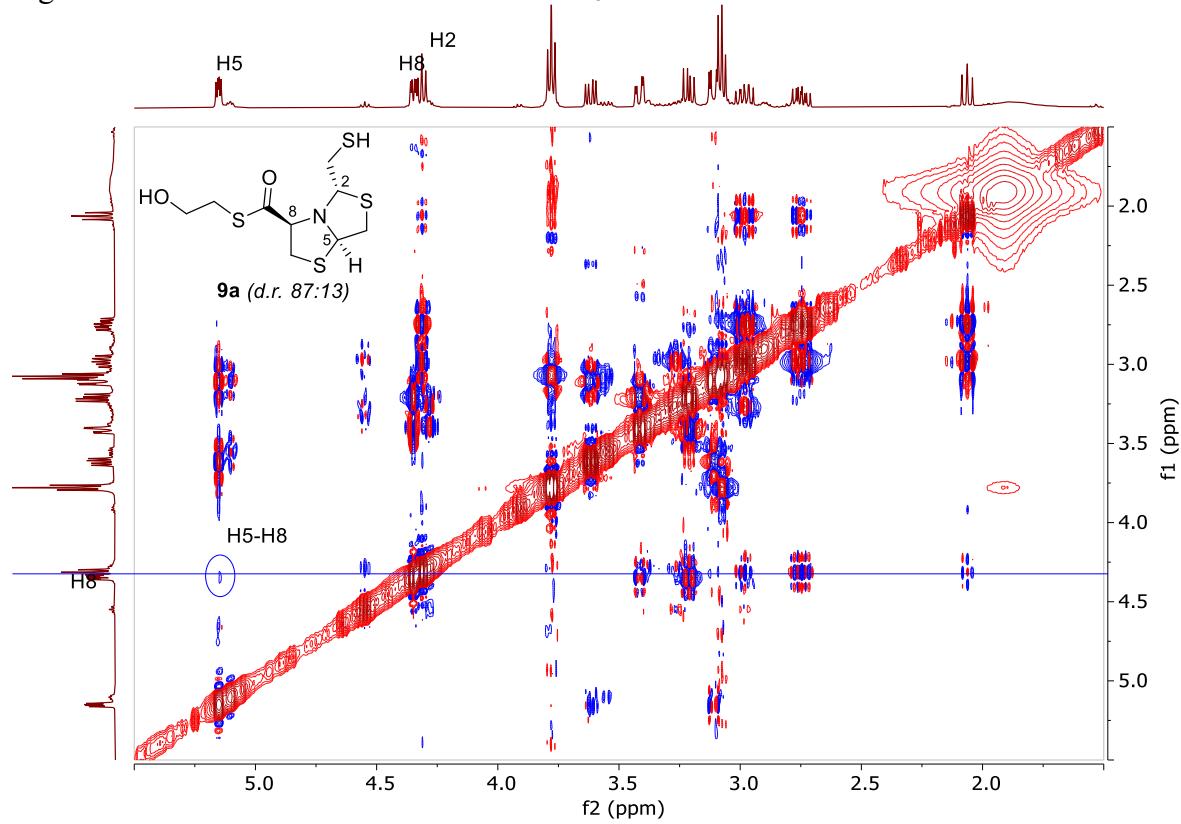


Figure S55: ^1H -NMR of **9b** in CDCl_3 (400 MHz)

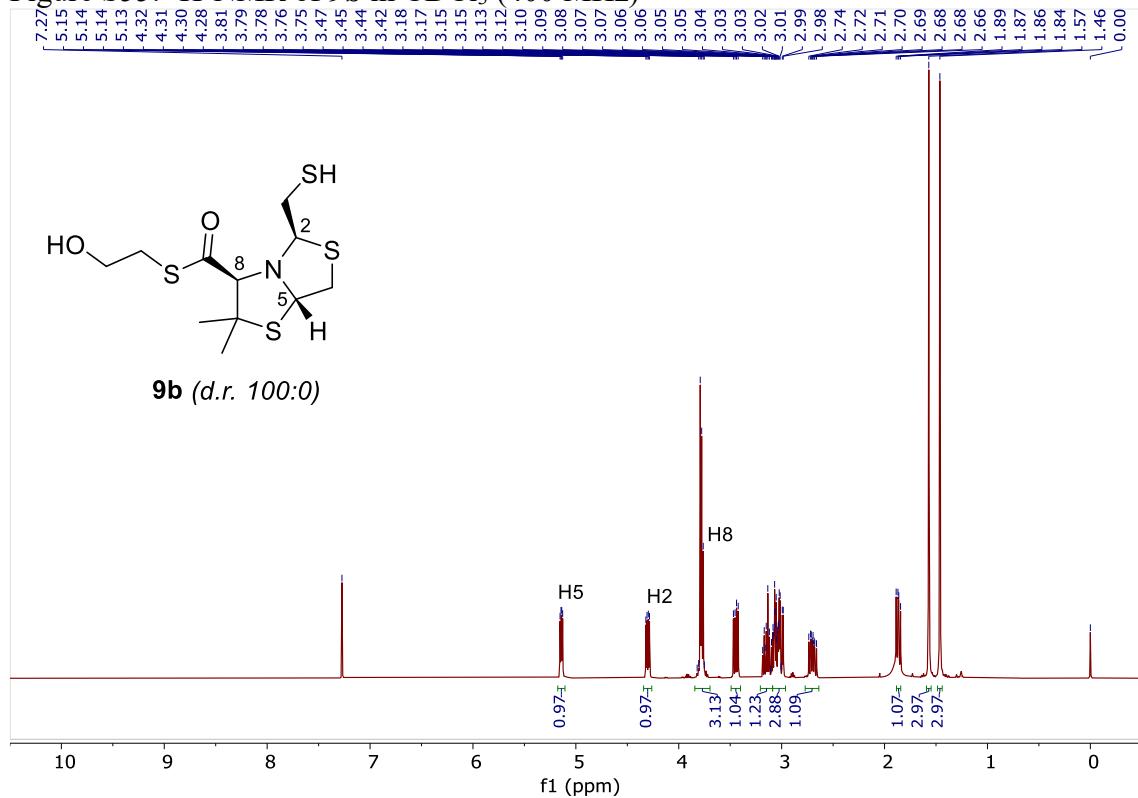


Figure S56: ^{13}C -NMR of **9b** in CDCl_3 (101 MHz)

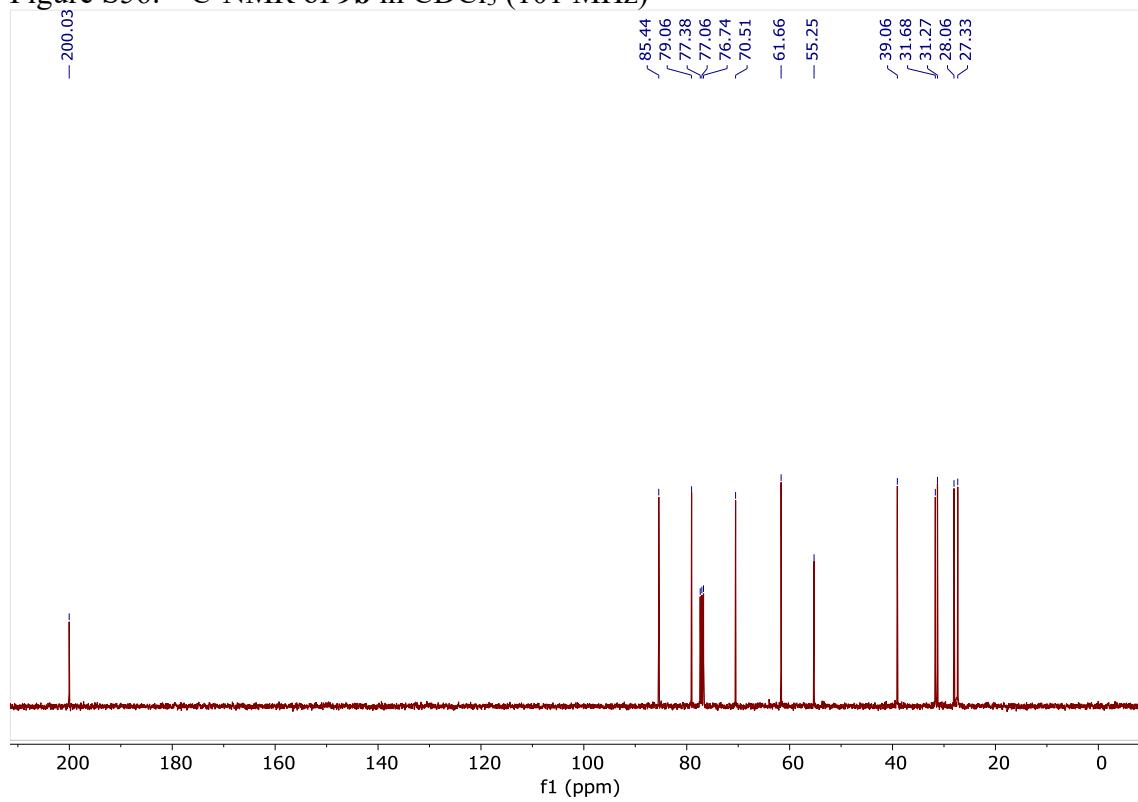


Figure S57: 2D NOESY NMR of **9b** in CDCl_3

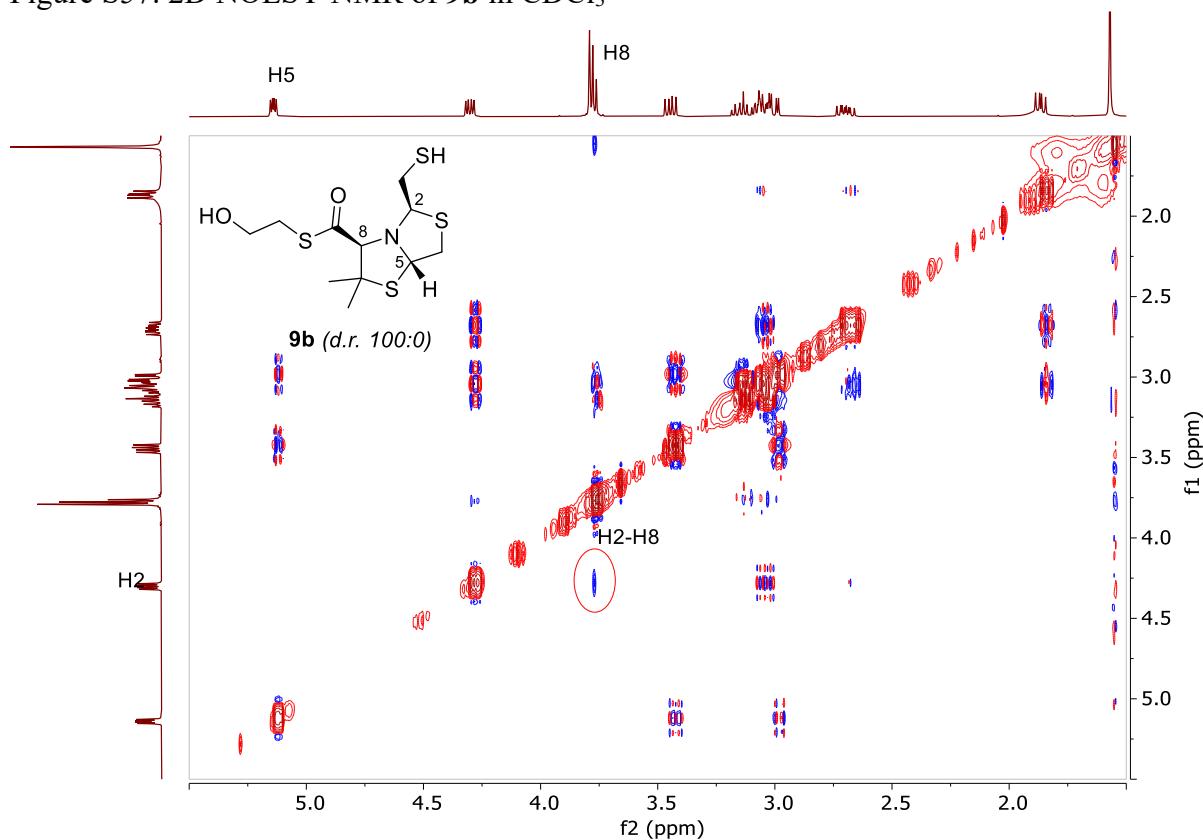


Figure S58: ^1H -NMR of **3'S-10a** in CDCl_3 (400 MHz)

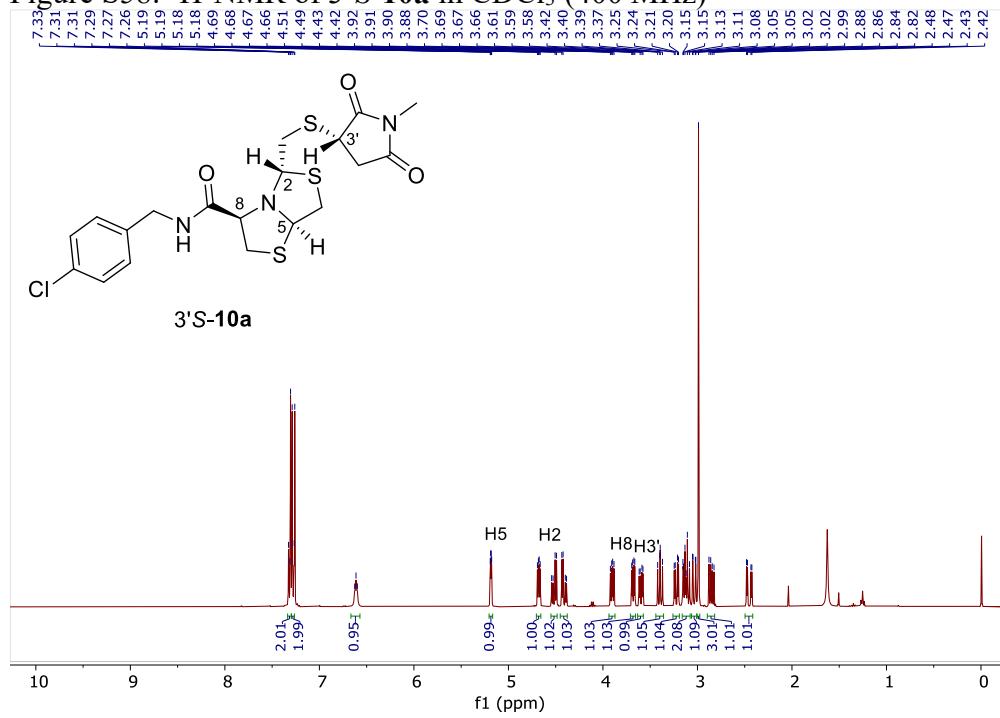


Figure S59: ^{13}C -NMR of 3'S-**10a** in CDCl_3 (101 MHz)

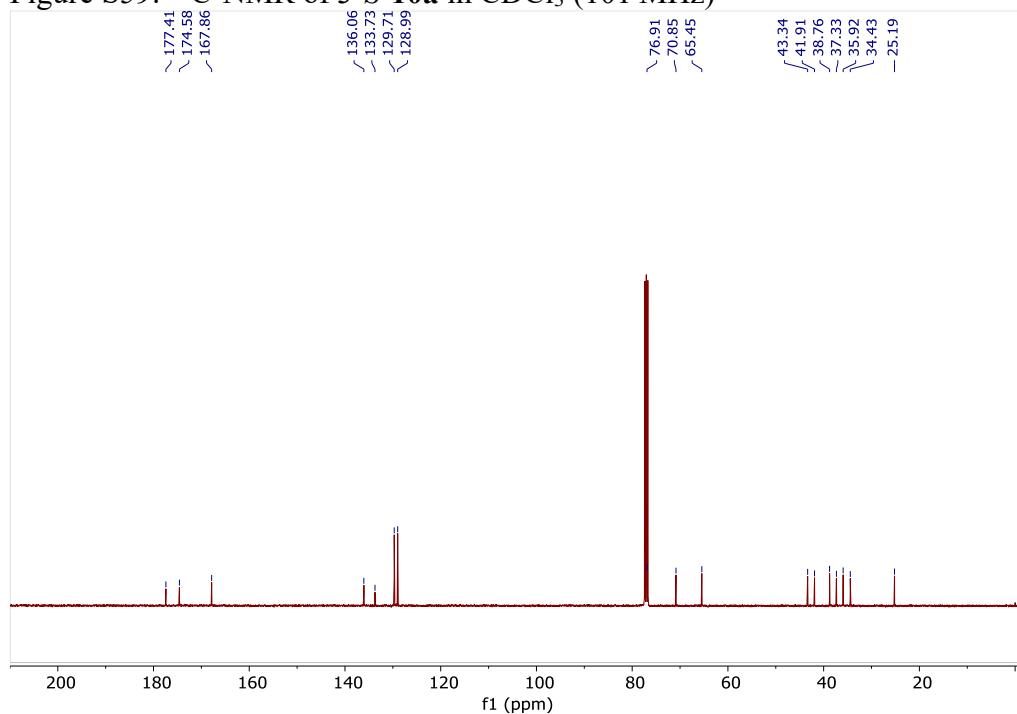


Figure S60: 2D NOESY NMR of 3'S-**10a** in CDCl_3

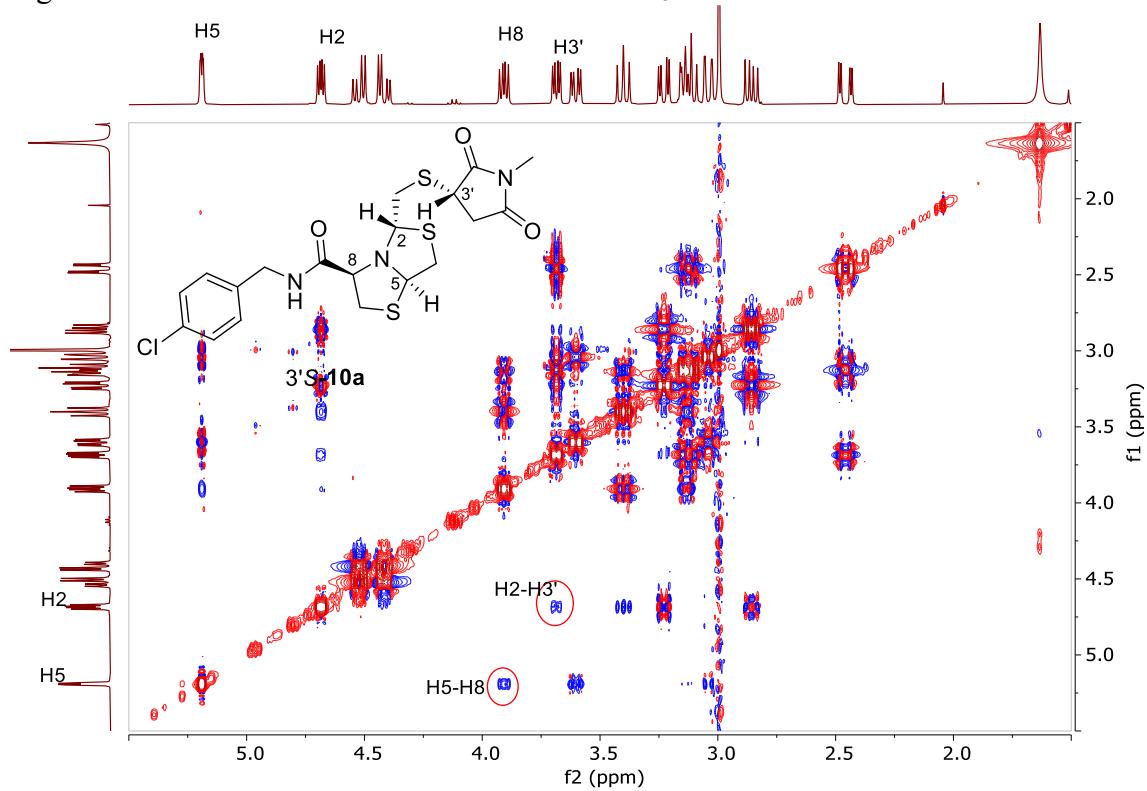


Figure S61: ^1H -NMR of 3'S-**10b** in CDCl_3 (400 MHz)

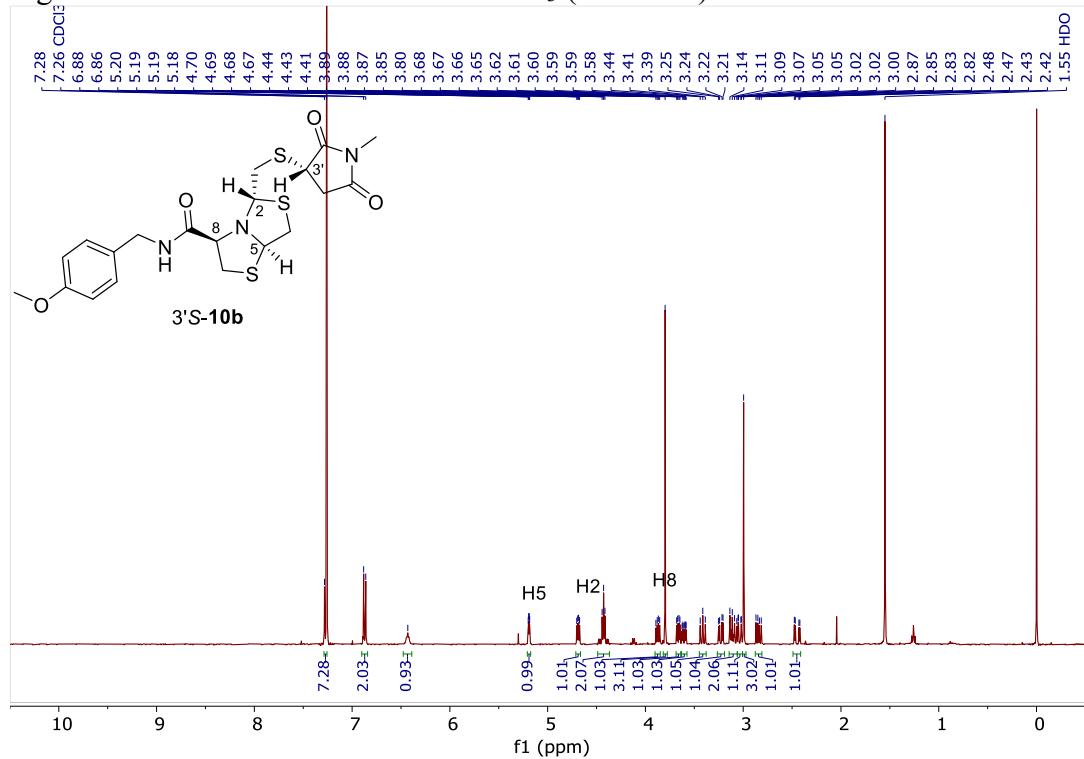


Figure S62: ^{13}C -NMR of 3'S-**10b** in CDCl_3 (101 MHz)

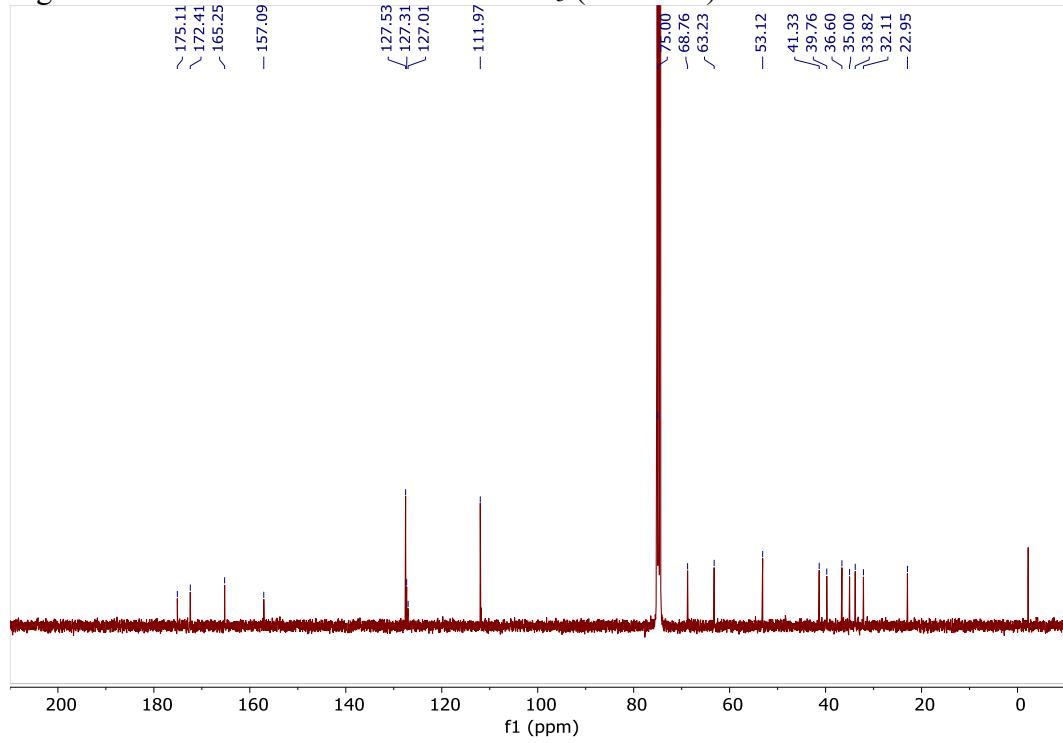
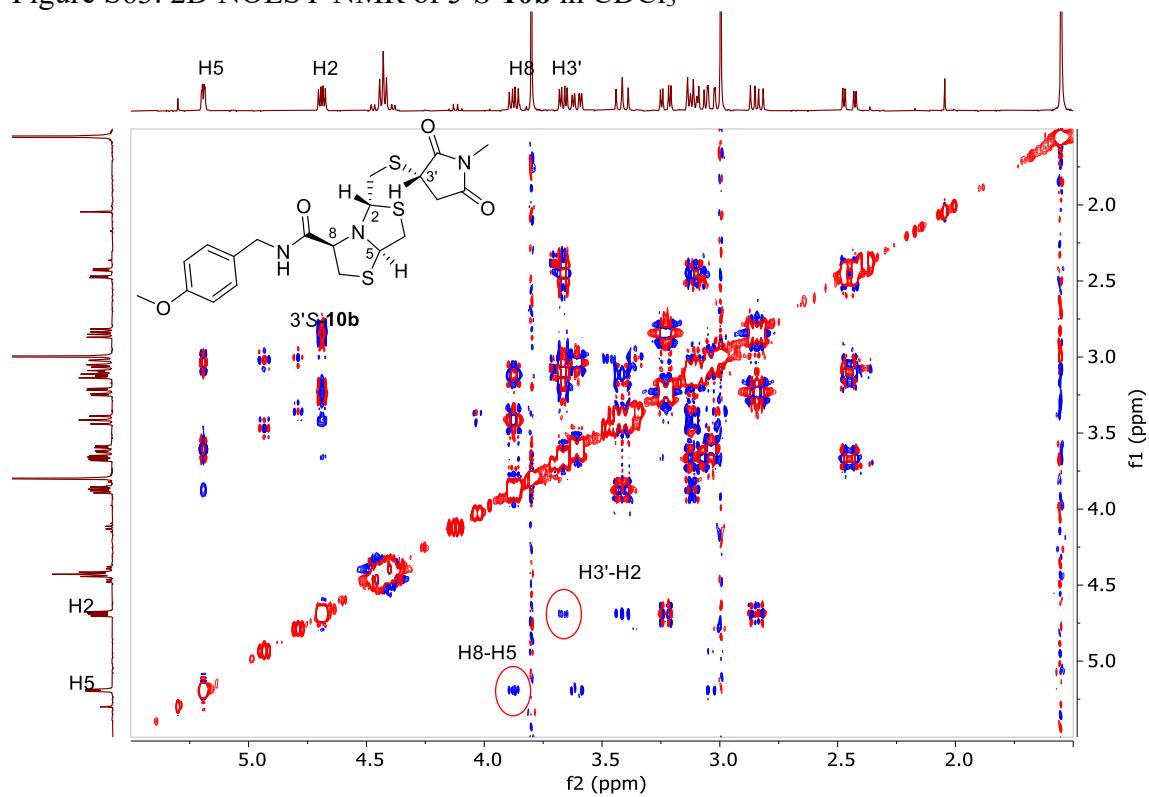


Figure S63: 2D NOESY NMR of 3'S-10b in CDCl₃



VIII. Opening of δ -thiolactones with dithiothreitol (DTT)

Condition TA-E: To a stirred solution of **1c** (40 mg, 0.18 mmol) dissolved in MeCN (1 mL) was added dithiothreitol (55 mg, 0.36 mmol) and dropwise a solution of L-Thr-OMe (39 mg, 0.22 mmol) in MeCN (0.5 mL). The reaction mixture was stirred for 24 h at rt, and the solvent evaporated under vacuum. The crude was extracted with HCl (5% aq. 10 mL) and AcOEt (3x15 mL), the organic layer was dried over Na₂SO₄ and concentrated under vacuum. The crude was purified by column chromatography (n-Hex:AcOEt 40:60) to give a mixture of **5d** distereomers (16 mg, 0.04 mmol, 22%, dr(58:24:18)) as an oil.

Figure S64: ¹H-NMR of **5d** distereomers in CDCl₃ (8.5 - 3.5 ppm, 400 MHz)

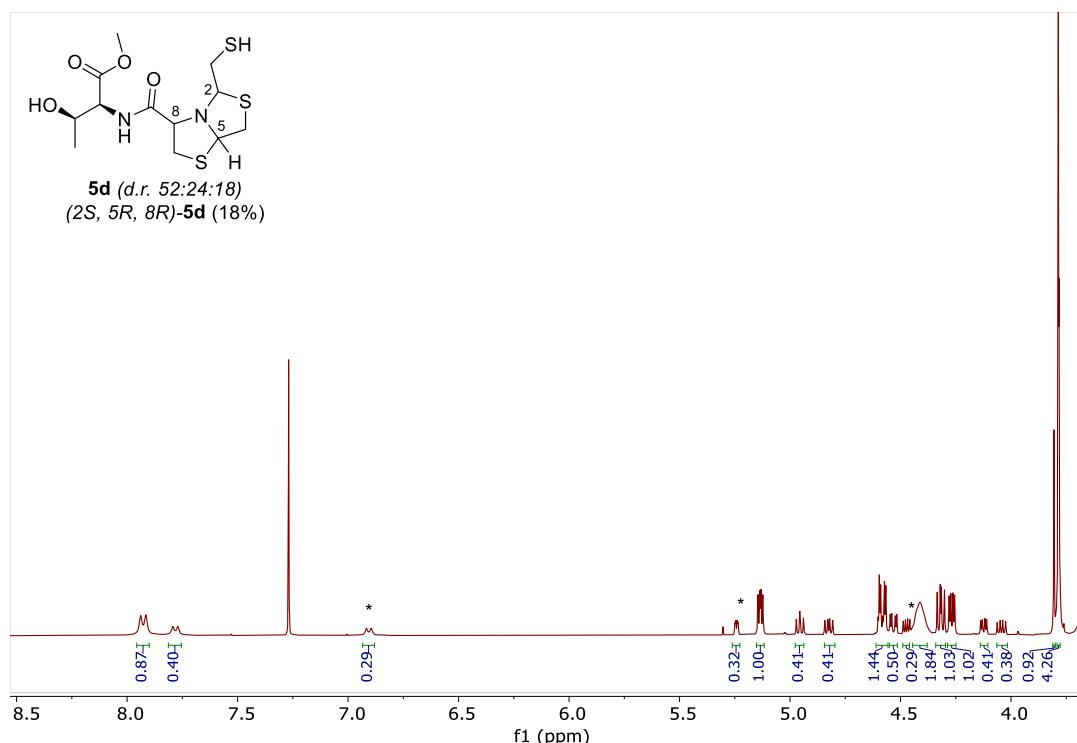
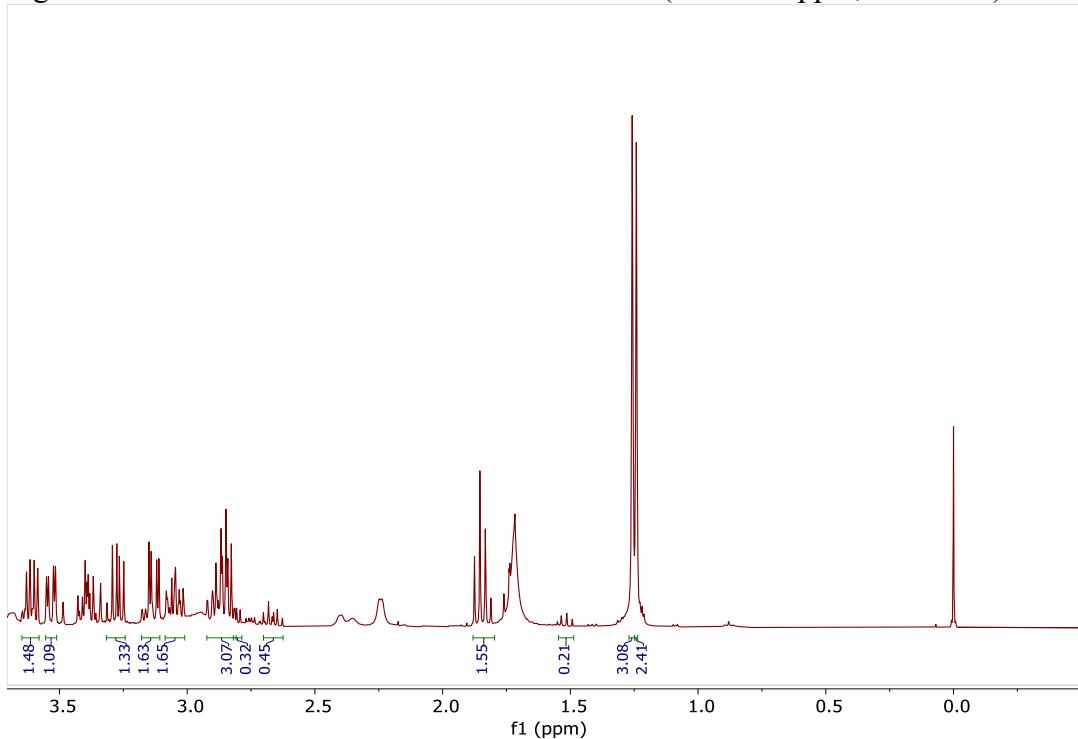
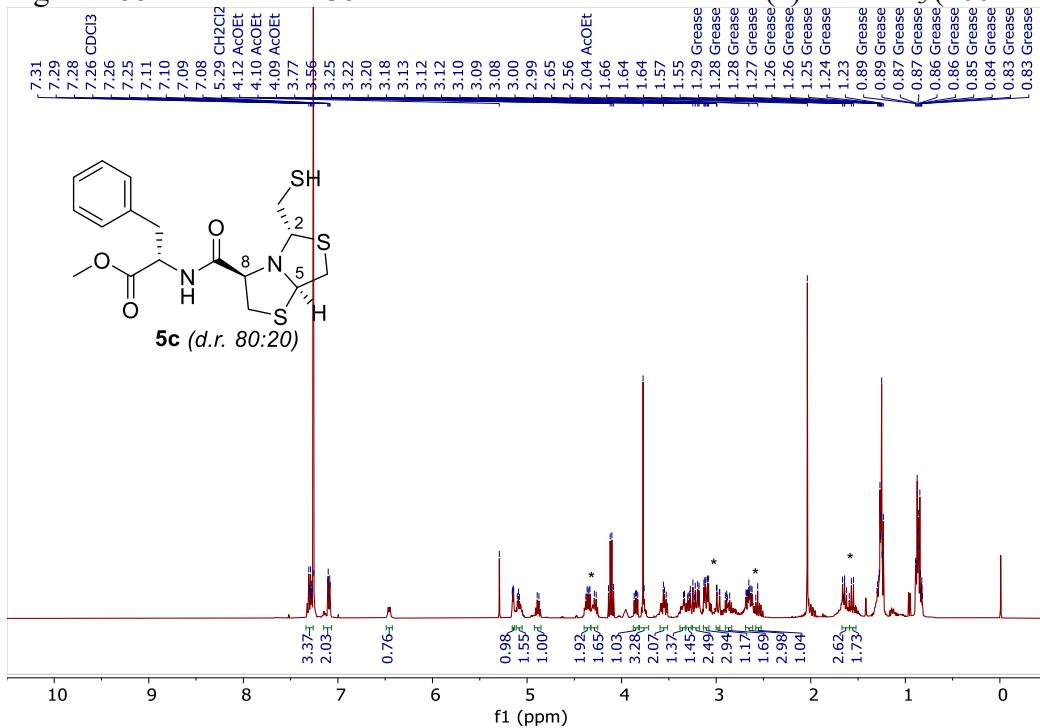


Figure S65: ^1H -NMR of **5d** distereomers in CDCl_3 (4.0 - -0.5 ppm, 400 MHz)



Condition TA-E: To a stirred solution of **1a** (40 mg, 0.18 mmol) dissolved in MeCN (1 mL) was added dithiothreitol (55 mg, 0.36 mmol) and dropwise a solution of L-Phe-OMe (39 mg, 0.22 mmol) in MeCN (0.5 mL). The reaction mixture was stirred for 5 h at rt, and the solvent evaporated under vacuum. The crude was extracted with HCl (5% aq. 10 mL) and AcOEt (3x15 mL), the organic layer was dried over Na_2SO_4 and concentrated under vacuum. The crude was purified by column chromatography (n-Hex:AcOEt 60:40) to give a irresolvable mixture of **5c** distereomers (40 mg, 0.10 mmol, 55%, dr(80:20)) and dithiothreitol.

Figure S66: ^1H -NMR of **5c** distereomers and dithiothreitol (*) in CDCl_3 (400 MHz)



IX. References

- [1] G.M. Sheldrick, A short history of SHELX, *Acta Crystallogr. Sect. A Found. Crystallogr.* 64 (2008) 112–122, doi: 10.1107/S0108767307043930
- [2] G.M. Sheldrick, Crystal structure refinement with SHELXL, *Acta Crystallogr. Sect. C: Struct. Chem.* 71 (2015) 3–8, doi: 10.1107/S2053229614024218.
- [3] C.F. Macrae, I.J. Bruno, J.A. Chisholm, P.R. Edgington, P. McCabe, E. Pidcock, L. Rodriguez-Monge, R. Taylor, J. van de Streek, P.A. Wood, Mercury CSD 2.0 - new features for the visualization and investigation of crystal structures, *J. Appl. Crystallogr.* 41 (2008) 466–470, doi: 10.1107/ S0021889807067908.