Supporting Information

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Figure 2 High-resolution XPS spectra for Ni 2p of Ni_3S_2



Figure 3 High-resolution XPS spectra for Mo 2p of MoS_2



Figure 4 Polarization curves of HER



Figure 5 Polarization curves of OER



Figure 6 Polarization curve of two-electrode system



Figure 7 Chemisorption models of $Ni_3S_2[1 \ 0 \ 1]$ (a), $FeS_2[200]$ (b), Fe (NiS_2)₂ [1 1 1] (c), CoS_2 [2 0 0], MoS_2 and CoMoS.

Table1 Cdl and ECSA of HER

Catalysts	Cdl(mF.cm ⁻²)	ECSA (cm ²)
CoMoS/FF	33.3	832.5
CoMoS/NF	29.9	747.5
NiFeCoMoS/NFF	32.2	805
NiFeMoS/NFF	12.4	310
NiFeCoS/NFF	15.3	382.5
NiFeS	21.2	530

Table2 Cdl and ECSA of OER

Catalysts	Cdl(mF.cm ⁻²)	ECSA (cm ²)
CoMoS/FF	7.5	187.5
CoMoS/NF	21.3	532.5
NiFeCoMoS/NFF	31.9	797.5
NiFeMoS/NFF	30.6	765
NiFeCoS/NFF	32.5	812.5
NiFeS	32.7	817.5

Table3 Comparison of activity of the NiFeCoMoS/NFF with recently reported non-precious metal based electrocatalysts. The * indicates data was corrected by the iR loss.

Electrocatalyst	Electrolyte	$\eta_{\rm 10}$ / mV	Substrate	Reference
Ni3S2-CoMoSx	1М КОН	1.52*	Ni foam	1
Co-Fe-NiSe2	1М КОН	1.52*	carbon cloth	2
MoS2/NiS2	1М КОН	1.59*	carbon cloth	3
MoS2/NiS	1М КОН	1.61	Ni foam	4
P- Co304	1М КОН	1.63	Ni foam	5
Ni/Mo2C-NCNF	1М КОН	1.64	carbon cloth	6
Fe-Ni@NC-CNTs	1М КОН	1.7	carbon cloth	7
NiCo/NiCoOx	1М КОН	1.72	Ni foam	8
δ-FeOOH	1М КОН	1.65	Ni foam	9
NiCo2S4	1М КОН	1.7	Ni foam	10

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