Electronic Supplementary Material (ESI) for Reaction Chemistry & Engineering. This journal is © The Royal Society of Chemistry 2024

Selective hydrogenation of 5-hydroxymethylfurfural to 2,5bis(hydroxymethyl)furan over Ni–Ga intermetallic catalysts and its kinetic studies

5 Weixiao Sun[†], Fuzeyu Zhong[†], Xiaohu Ge^{*}, Wenyao Chen, Gang Qian, Yueqiang Cao, Xuezhi Duan, Xinggui Zhou, Jing Zhang^{*}

State Key Laboratory of Chemical Engineering, School of Chemical Engineering, East China University of Science and Technology; Shanghai, 200237, China.

[†]These authors contributed equally to this work.

10 *Corresponding author email: xhge@mail.ecust.edu.cn, jingzhang8507@ecust.edu.cn

	Supporting Figures
	Fig. S1
	Fig. S24
5	Fig. S35
	Fig. S46
	Fig. S57
	Fig. S6
	Fig. S79
10	Fig. S810
	Fig. S911
	Fig. S1012
	Fig. S1113
	Fig. S1214
15	Fig. S1315
	Supporting Tables
	Table S116
	Table S217
20	Table S318
	Table S419
	Table S5
	Scheme S1

Contents

Supporting Figures



Fig. S1. EDX line scan of the Ni_1Ga_1 intermetallic catalyst.

	- 3—	AI		lt k			Map Sum	Spectrum
	- 2-		Element	Line Type	k factor	Absorption Correction	Wt%	Wt% Sigma
/e/	-		Al	K series	1.03564	1.00	90.74	0.35
çb			Ni	K series	1.26753	1.00	4.00	0.23
	1_		Ga	K series	1.60318	1.00	5.26	0.27
	- 2 (Ga	Total:				100.00	
		M M	Ni	a				
	0		5	10 15	5 20	25	30	35 keV

Fig. S2. The EDX spectra of the Ni_1Ga_1 intermetallic catalyst and total distribution map spectrum.



Fig. S3. EDS elemental mapping of the Ni₁Ga₁ intermetallic catalyst.



Fig. S4. NH₃-TPD profiles of the referred Ni, Ni₃Ga₁, and Ni₁Ga₁ catalysts.



Fig. S5. (a) N₂ adsorption-desorption isotherms and (b) pore size distributions of the referred Ni, Ni₃Ga₁, Ni₁Ga₁ intermetallic catalysts and Al₂O₃ support.



Fig. S6. HAADF-STEM images with corresponding particle size distribution of the post-reaction (a) referred Ni, (b) Ni₃Ga₁, and (c) Ni₁Ga₁ intermetallic catalysts.



Fig. S7. XRD spectra of the Ni₁Ga₁/MgAl intermetallic catalyst.



Fig. S8. HRTEM images with lattice fringes and FFT patterns, HAADF-STEM images with corresponding particle size distribution of the Ni₁Ga₁/MgAl intermetallic catalyst.



Fig. S9. EDX line scan of the $Ni_1Ga_1/MgA1$ intermetallic catalyst.



Fig. S10. EDS elemental mapping of the $Ni_1Ga_1/MgA1$ intermetallic catalyst.



Fig. S11. NH₃-TPD (a) and CO₂-TPD (b) profiles of the referred Ni/MgAl and Ni₁Ga₁/MgAl catalysts.



Fig. S12. Conversion of HMF (a) and selectivity of BHF (b) over the referred Ni/MgA1 and Ni₁Ga₁/MgA1 catalysts (reaction conditions: 20 mL H₂O, 25 mg catalyst, 40 mg HMF, 3.0 MPa H₂, 120 °C).



Fig. S13. The Ni atoms coordination environment of Ni_3Ga_1 and Ni_1Ga_1 intermetallics.

Supporting Tables

Table S1. Binding energies & surface compositions (wt.%) of the referred Ni, Ni₃Ga₁, and Ni₁Ga₁ catalysts measured by *in situ* XPS/ICP-MS.

Catalyat	Binding energy (eV)		Fraction (%)		Ni (wt.%)		Ga (wt.%)	
Cataryst	${ m Ni}^{0} 2 p_{3/2}$	$\mathrm{Ga}^{0} 2p_{3/2}$	Ni ⁰ /(Ni ⁰ +Ni ²⁺)	Ga ⁰ /(Ga ⁰ +Ga ³⁺)	XPS	ICP	XPS	ICP
Ni	852.9	/	74.4	/	5.2	5.1	/	/
Ni ₃ Ga ₁	852.7	1116.2	78.0	61.3	6.5	5.0	3.1	1.8
Ni ₁ Ga ₁	852.5	1116.4	81.9	54.4	12.2	5.3	10.6	5.1

5

Catalyst	Cycle	c _{Ni} (mg/L)
	1	1.85
NI:	2	0.97
181	3	0.90
	4	0.88
	1	0.18
N: Ca	2	0.10
N ₁₃ Ga ₁	3	0.09
	4	0.05
	1	0.15
N: Ca	2	0.05
m ₁ Ga ₁	3	0.08
	4	0.04

Table S2. ICP analysis of the post-reaction solution from the referred Ni, Ni_3Ga_1 , and Ni_1Ga_1 intermetallic catalysts after the recyclability test.

Table S3. Catalytic performance of Al2O3 catalysts.

Catalyst	Conv. (%)	Select. BHF (%)	Select. BHTF (%)
Al ₂ O ₃	<0.1	-	-

Catalyst	Acidity (mmol/g)
Al ₂ O ₃	1.548
Ni	1.480
Ni ₃ Ga ₁	1.503
Ni ₁ Ga ₁	1.466

Table S4. Acidity of intermetallic Ni-Ga over Al₂O₃.

Catalyst	Acidity (mmol/g)	Basicity (mmol/g)
Ni/MgAl	0.114	0.180
Ni ₁ Ga ₁ /MgAl	0.142	0.150

Table S5. Acidity and basicity of $Ni_1Ga_1/MgA1$ intermetallic catalysts.



Scheme S1. The schematic diagram of Ni₁Ga₁ catalyst preparation.