## Supplementary Information

# Nanoscale and Chiral Metal-Organic Framework for Asymmetric Reaction in Water: Bridging Lewis Acid Catalysis to Biological Systems 

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## 1. General experimental

Nuclear magnetic resonance (NMR) spectra were recorded on JEOL ECX-600 spectrometers, operating at 600 MHz for ${ }^{1} \mathrm{H}$ and 151 MHz for ${ }^{13} \mathrm{C} \mathrm{NMR} \mathrm{in} \mathrm{CDCl}_{3}$ and are reported relative to the solvent residual ${ }^{1} \mathrm{H}$-signal $\left(\mathrm{CHCl}_{3}, \delta(\mathrm{H})=7.26\right)$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and $\mathrm{CDCl}_{3}(\delta=77.0)$ for ${ }^{13} \mathrm{C}$ NMR. Infrared (IR) spectra were obtained using a SHIMADZU IRSpirit spectrometer, data are represented as frequency of absorption ( $\mathrm{cm}^{-1}$ ). High-performance liquid chromatography was carried out using the following apparatuses: SHIMADZU LC-10ATvp (liquid chromatograph), SHIMADZU SPD-10A (UV detector), and SHIMADZU C-R8A (Chromatopac) using Daicel Chiralpak ${ }^{\circledR}$ or Chiralcel ${ }^{\circledR}$ columns. Preparative thin-layer chromatography (PTLC) was carried out using Wakogel B-5F from Wako Pure Chemical Industries, Ltd. High-resolution mass spectra (HRMS) were recorded using a JEOL JMS T100TD (DART) or T100LC (ESI) spectrometer. Optical rotations were measured on a JASCO P2100 polarimeter using a 2 mL cell with 1 dm path length. Data are reported as follows: $[\alpha]_{\mathrm{D}}{ }^{\mathrm{T}}$ ( $c$ in $\mathrm{g} / 100 \mathrm{~mL}$, solvent). All melting points were determined on a YAZAWA micro melting point BY-1 apparatus and are uncorrected. Inductively Coupled Plasma (ICP) analysis was performed on Shimadzu ICPS-7510 equipment. Centrifugation was performed by using KOKUSAN H-36 centrifuge with 3500 rpm for 10 minutes, unless otherwise noted. TG analysis was performed on Rigaku Thermoplus TG 8120. XPS analysis was performed on JPS-9010MC with a Mg Ka X-ray source. The $\mathrm{Au} 4 \mathrm{f}^{7 / 2}$ line at 84.0 eV was used as a reference to correct the binding energy of C 1s. The corrected C 1s (283.4 eV ) was used as reference to correct the binding energy of other samples. PXRD analysis was performed on MiniFlex 600 from Rigaku. STEM/EDS images were obtained using a JEOL JEM2100 F instrument operated at 200 kV . All STEM specimens were prepared by placing a drop of the solution on carbon-coated grids and allowed to dry in air (without staining). SEM analysis was measured by a JEOL JSM-6700F. Nitrogen absorption/desorption isotherms were recorded on a BELSORP-mini Microtrac Bell. Calculation study to estimate the length of linkers was performed using the functional B3LYP together with $6-31 \mathrm{G}+(\mathrm{d}, \mathrm{p})$ basis set for $\mathrm{C}, \mathrm{H}, \mathrm{N}, \mathrm{O}$, and LanL2DZ for Sc. Deionized water from a MILLIPORE MilliQ machine (Integral 3) was used a as solvent without further treatment. All organic solvents used were commercially available dry solvents. Commercially available reagents (in liquid form) for the ring-opening reactions of epoxides were distilled before use.

## 2. Initial discovery through the design of chiral $\mathbf{H}_{2}$ BPVB linker 5

In our initial exploration, the chiral $\mathrm{H}_{2} \mathrm{BPV}$ linker $\mathbf{3}$ was synthesized from 2,6-dibromopyridine in 9 steps (Figure S1).


Figure S1. Synthesis of chiral $\mathrm{H}_{2} \mathrm{BPV} 3$.

Following the successful synthesis of chiral linker 3, the subsequent phase of our investigation entailed the preparation of the corresponding BPV-MOF through a solvothermal process. ${ }^{1}$ The reaction conditions for MOF construction were systematically explored and refined, as outlined in Table S1. After the completion of the reactions, the resulting precipitates underwent thorough ${ }^{1} \mathrm{H}-\mathrm{NMR}$ analysis to assess the degree of linker occupancy. Unexpectedly, our analysis uncovered the presence of undesired linkers $\mathbf{B}$ and $\mathbf{C}$, originating from intramolecular 1,4addition, during the self-assembly process (Figure S2).

Table S1. Optimization in the preparation of chiral BPV-MOF via solvothermal process.



| Entry | Heater (temp, ${ }^{\circ} \mathrm{C}$ ) | Modulator (x equiv) | Time (h) | Yield (mg) | $\mathrm{A}: \mathrm{B}: \mathrm{C}^{a}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | oven (120) | AcOH (30) | 48 | 9.9 | 24:76:0 |
| 2 | oven (120) | AcOH (30) | 24 | 7.8 | 23:71:6 |
| 3 | oven (110) | AcOH (30) | 24 | 7.1 | 41:54:5 |
| 4 | oven (100) | AcOH (30) | 40 | 3.9 | 67:30:3 |
| 5 | oven (90) | AcOH (30) | 70 | 6.5 | 85:2.5:12.5 |
| 6 | oven (80) | AcOH (30) | 135 | 4.2 | 53:27:20 |
| 7 | oven (90) | $\mathrm{AcOH}(10)$ | 70 | 9.7 | 42: $24: 34$ |
| 8 | oven (90) | - | 70 | 8.1 | 47:2:51 |
| $9^{\text {b }}$ | oven (90) | $\mathrm{AcOH}(30)$ | 70 | 12.8 | 36:9:55 |
| $10^{\text {b }}$ | oil bath (90) | $\mathrm{AcOH}(30)$ | 70 | 21.0 | 47:38.5: 14.5 |
| $11^{c}$ | oil bath (90) | $\mathrm{AcOH}(30)$ | 70 | 39.5 | 59:34:7 |
| 12 | microwave (90) | $\mathrm{AcOH}(30)$ | 8 | n.d. | - |
| 13 | oil bath (90) | $\mathrm{PhCO}_{2} \mathrm{H}$ (30) | 70 | n.d. | - |
| 14 | oil bath (90) | TFA (30) | 70 | n.d. | - |
| 15 | oil bath (90) | HCOOH (30) | 70 | 10.0 | 29: $65: 6$ |
| 16 | oil bath (90) | HCl (30) | 32 | 6.8 | 42: 49 : 9 |
| 17 | oil bath (90) | HCl (10) | 35 | 4.9 | 42: 17 : 41 |

[^0]
${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{K}_{3} \mathrm{PO}_{4} / \mathrm{D}_{2} \mathrm{O} / \mathrm{DMSO}-d_{6}\right)$

Table S1,entry 1


Table S1,entry 8


Figure S2. Assignment of linkers in BPV-MOF after solvothermal process.

To investigate whether the decomposition of the chiral linker impacted the asymmetric reaction in water, we explored the post-synthetic metalation of the prepared Zr -MOFs using $\mathrm{Sc}(\mathrm{OTf})_{3}$ and $\mathrm{Sc}(\mathrm{DS})_{3}$, as detailed in Table S 2 . The successful incorporation of scandium was confirmed through ICP analysis. These catalysts were subsequently evaluated for their performance in the asymmetric ring-opening reaction of epoxide in water, as summarized in Table S3. When BPV-MOF-Sc-OTf was employed (entries 1 and 2), low yields and enantioselectivities were obtained. Notably, the substitution of the counter anion to dodecyl sulfate (DS-) resulted in an improvement in reactivity and selectivity, with the desired product isolated in $70 \%$ yield with $35 \%$ ee. However, it is crucial to highlight that the observed low enantioselectivity was still evident, attributed to the deactivation of the chiral coordination site by Michael addition. Consequently, finding an alternative linker to suppress this issue is necessary.

Table S2. Preparation of chiral BPV-MOF-Sc catalysts.

${ }^{a}$ Based on ICP analysis. ${ }^{b}$ Few drops of $\mathrm{H}_{2} \mathrm{O}$ was added.

Table S3. Evaluation of chiral BPV-MOF-Sc catalysts in asymmetric ring-opening reaction of epoxide.


[^1]Table S4. Unsuccessful synthesis of chiral $\mathrm{H}_{2}$ BPDC linker 2.



| entry | conditions | remarks |
| :---: | :---: | :---: |
| 1 | 2.1 equiv $\mathrm{NaOMe}, \mathrm{MeOH}, \mathrm{rt}, 48 \mathrm{~h}$ | no conversion |


| 2 | 2.5 equiv $\mathrm{NaOH}, \mathrm{THF}, \mathrm{rt}, 18 \mathrm{~h}$ | messy |
| :---: | :---: | :---: |
| $3^{a}$ | 20 equiv $\mathrm{NaOMe}, \mathrm{MeOH}, 40^{\circ} \mathrm{C}$ | full conversion, messy |
| $4^{a}$ | 2.5 equiv $\mathrm{NaOMe}, \mathrm{MeOH}, 40^{\circ} \mathrm{C}$ | no conversion |
| $5^{a}$ | 5 equiv $\mathrm{NaOMe}, \mathrm{MeOH}, 40^{\circ} \mathrm{C}$ | $58 \%$ conv., complex mixture |
| $6^{a}$ | 10 equiv $\mathrm{NaOMe}, \mathrm{MeOH}, 40^{\circ} \mathrm{C}$ | full conversion, under purification |

${ }^{a}$ stirred for 18 h .

Table S5. Unsuccessful synthesis of chiral $\mathrm{H}_{2}$ BPY linker 4.






In response to the challenges encountered with the chiral $\mathrm{H}_{2} \mathrm{BPV}, \mathrm{H}_{2} \mathrm{BPDC}$, and $\mathrm{H}_{2} \mathrm{BPY}$ ligands, we have undertaken a modification of synthetic methods to design a new chiral linker by
introducing vinyl benzoate groups. We hypothesized that this modification not only addresses the linker decomposition or deprotection issues but also enhances hydrophobicity, rendering it a promising candidate for water-stable ligands. The proposed retrosynthesis of the chiral $\mathrm{H}_{2} \mathrm{BPVB}$ linker 5 is illustrated in Figure S3, outlining three key synthetic steps: asymmetric transfer hydrogenation, Wittig olefination, and Ullmann-homocoupling reaction.


chiral $\mathrm{H}_{2} \mathrm{BPVB}(5)$
Ullmann-homocoupling
Wittig olefination


Figure S3. Retrosynthesis of chiral $\mathrm{H}_{2}$ BPVB linker 5.

## 3. Synthesis of the chiral $\mathrm{H}_{2}$ BPVB linker 5



Figure S4. Overview synthetic pathways toward chiral $\mathrm{H}_{2} \mathrm{BPVB}$ linker 5.

## Synthesis of 2,6-Dibromonicotinaldehyde (S4) ${ }^{\mathbf{2}}$



S1



THF $,-78^{\circ} \mathrm{C}, 2 \mathrm{~h}$, then rt, 12 h
66\%

To an oven-dried 300 mL 3-necked round-bottom flask equipped with a stirring bar was charged diisopropylamine ( $16.9 \mathrm{~mL}, 120 \mathrm{mmol}$ ) under argon atmosphere, followed by the addition of 120 mL dry THF before setting at $-78^{\circ} \mathrm{C}$. A solution of ${ }^{n} \operatorname{BuLi}\left(1.56 \mathrm{M}\right.$ in ${ }^{n}$ hexane, $76.9 \mathrm{~mL}, 120$ mmol ) was added dropwise into the reaction flask via a dropping funnel at $-78{ }^{\circ} \mathrm{C}$. The mixture was stirred at $-78^{\circ} \mathrm{C}$ for 1 hour to obtain lithium diisopropylamide (LDA). To an oven-dried 1 L 3-necked round-bottom flask containing a solution of 2,6-dibromopyridine ( $23.68 \mathrm{~g}, 100$ mmol ) in 200 mL dry THF at $-78^{\circ} \mathrm{C}$, was slowly transferred the prepared LDA solution via a Teflon cannula, the mixture solution was stirred at $-78^{\circ} \mathrm{C}$ for 2 hours. Later, a solution of methyl formate ( $60.05 \mathrm{~g}, 1000 \mathrm{mmol}$ ) in 60 mL dry THF was added dropwise via a drooping funnel. The reaction mixture was then stirred at $-78^{\circ} \mathrm{C}$ for 2 hours before gradually warming up to room
temperature and stirred at room temperature for 12 hours. Then, the reaction was quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}(100 \mathrm{~mL})$ and extracted with EtOAc ( $3 \times 100 \mathrm{~mL}$ ). Organic layers were combined and washed with brine, dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated. The crude product was purified by column chromatography (silica gel, ${ }^{n}$ hexane $/ \mathrm{EtOAc}=20 / 1$ ) to give the title compound $\mathbf{S 4}$ as a white solid. $\mathbf{M p}$ : $125-126^{\circ} \mathrm{C}$; IR (neat): $v=2884,1665,1552,1424$, 1385, 1004, $830 \mathrm{~cm}^{-1} ;{ }^{1} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 10.28(\mathrm{dd}, J=1.8,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.01-$ $7.99(\mathrm{~m}, 1 \mathrm{H}), 7.63-7.60(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 190.0,146.4,144.1,139.3$, 129.6, 128.2; HRMS (DART) for $\mathrm{C}_{6} \mathrm{H}_{4} \mathrm{Br}_{2} \mathrm{NO}$ : calculated for $[\mathrm{M}+\mathrm{H}]^{+}$265.86392, found 265.86286 .

## Synthesis of 2,6-Dibromo-3-(1,3-dioxolan-2-yl)pyridine (S5) ${ }^{\mathbf{3}}$





To a solution of 2,6-dibromonicotinaldehyde ( $\mathbf{S 4}, 12.4 \mathrm{~g}, 47 \mathrm{mmol}$ ) and $p-\mathrm{TsOH} \cdot \mathrm{H}_{2} \mathrm{O}$ in 180 mL benzene, was added ethylene glycol ( $14.5 \mathrm{~g}, 234 \mathrm{mmol}$ ) via syringe. The reaction mixture was refluxed at $95^{\circ} \mathrm{C}$ for 16 hours. After completion, the reaction was allowed to cool down to room temperature and quenched with $\mathrm{H}_{2} \mathrm{O}(100 \mathrm{~mL})$. It was extracted with EtOAc $(3 \times 50 \mathrm{~mL})$ and the combined organic layer was washed with brine and dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. After concentration under reduced pressure, the crude product was purified by column chromatography (silica gel, ${ }^{n}$ hexane $/ \mathrm{EtOAc}=4 / 1$ ) to give the title compound $\mathbf{S 5}$ as a white solid. $\mathbf{M p}:=75-77{ }^{\circ} \mathrm{C}$; IR (neat): $v=2898,1564,1538,1414,1374,1108,821 \mathrm{~cm}^{-1} ;{ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}(600$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 7.72(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.49(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.97(\mathrm{~s}, 1 \mathrm{H}), 4.15-4.11(\mathrm{~m}$, 2H), 4.11 - 4.07 (m, 2H); ${ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 141.0,140.9,138.3,133.6,127.2$, 101.1, 65.6; HRMS (DART) for $\mathrm{C}_{8} \mathrm{H}_{8} \mathrm{Br}_{2} \mathrm{NO}_{2}$ : calculated for $[\mathrm{M}+\mathrm{H}]^{+}$309.89013, found 309.89146.

Synthesis of 1-(6-Bromo-3-(1,3-dioxolan-2-yl)pyridin-2-yl)-2,2-dimethylpropan-1-one (6) ${ }^{4}$


To an oven-dried 100 mL 3 -necked round-bottom flask equipped with a stirring bar was charged 2,6-dibromo-3-(1,3-dioxolan-2-yl)pyridine ( $\mathbf{S 5}, 6.2 \mathrm{~g}, 20 \mathrm{mmol}$ ), followed by addition of 60 mL dry THF under argon atmosphere. A solution of ${ }^{n} \operatorname{BuLi}\left(1.56 \mathrm{M}\right.$ in ${ }^{n}$ hexane, $14.2 \mathrm{~mL}, 22 \mathrm{mmol}$ ) was transferred to the reaction flask at $-78^{\circ} \mathrm{C}$ via a dropping funnel. After the reaction mixture was allowed to stir at $-78^{\circ} \mathrm{C}$ for 2 hours, a solution of methyl pivalate ( $4.7 \mathrm{~g}, 40 \mathrm{mmol}$ ) in 30 mL dry THF was transferred dropwise via a dropping funnel at $-78^{\circ} \mathrm{C}$ and the mixture was continuously stirred at the same temperature for 1 hour. The reaction was gradually warmed up to room temperature and allowed to stir at room temperature for 3 hours. After completion, the solution was quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}$ and extracted with $\mathrm{EtOAc}(3 \times 30 \mathrm{~mL})$. The organic phase was combined and washed with brine, dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure. The crude product was purified by column chromatography (silica gel, ${ }^{n}$ hexane/EtOAc $=4 / 1$ ) to give the title compound $\mathbf{6}$ as a colorless oil. IR (neat): $v=2969,2875$, 1695, 1481, 1392, 1280, 1110, $978 \mathrm{~cm}^{-1}{ }^{1}{ }^{1} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.78(\mathrm{~d}, J=8.2 \mathrm{~Hz}$, $1 \mathrm{H}), 7.50(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.98(\mathrm{~s}, 1 \mathrm{H}), 3.99(\mathrm{~s}, 4 \mathrm{H}), 1.32(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( 151 MHz , $\mathrm{CDCl}_{3}$ ): $\delta 208.9,156.6,140.1,137.5,131.4,128.3,99.7,65.2,44.1,27.3$; HRMS (DART) for $\mathrm{C}_{13} \mathrm{H}_{17} \mathrm{BrNO}_{3}$ : calculated for $[\mathrm{M}+\mathrm{H}]^{+} 314.03918$, found 314.03978 .

Synthesis of (S)-1-(6-Bromo-3-(1,3-dioxolan-2-yl)pyridin-2-yl)-2,2-dimethylpropan-1-ol (7) ${ }^{5}$

$\operatorname{RuCl}[(S, S)$-Tsdpen $]$ (mesitylene) ( $395.1 \mathrm{mg}, 0.6 \mathrm{mmol}$ ) was charged into 100 mL 2-necked round-bottom flask equipped with a stirring bar, then 30 mL degassed $\mathrm{H}_{2} \mathrm{O}$ was added under argon atmosphere. After stirring the suspension at $40^{\circ} \mathrm{C}$ for 1 hour, a solution of sodium formate
in 30 mL degassed $\mathrm{H}_{2} \mathrm{O}$ was added into the reaction flask followed by adding 1-(6-bromo-3-(1,3-dioxolan-2-yl)pyridin-2-yl)-2,2-dimethylpropan-1-one ( $6,4.0 \mathrm{~g}, 13 \mathrm{mmol}$ ) via syringe. The reaction mixture was allowed to stir at $40{ }^{\circ} \mathrm{C}$ for 72 hours. After completion, the reaction was extracted with EtOAc ( $3 \times 40 \mathrm{~mL}$ ). The organic phase was combined and washed with brine, dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure. The crude product was purified by column chromatography (silica gel, ${ }^{n}$ hexane/EtOAc $=4 / 1$ ). The resulting compound was recrystallized by using "hexane to give the title compound 7 as a colorless crystal with single (S)-enantiomer (>99.9\% ee). Mp: $73-74{ }^{\circ} \mathrm{C}$; $\mathbf{I R}$ (neat): $v=3523,2884,1574,1445,1362,1173$, $1044,844 \mathrm{~cm}^{-1} ;{ }^{\mathbf{1}} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.74(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.41(\mathrm{~d}, J=8.2 \mathrm{~Hz}$, $1 \mathrm{H}), 6.00(\mathrm{~s}, 1 \mathrm{H}), 4.71(\mathrm{~d}, J=9.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.18-4.10(\mathrm{~m}, 2 \mathrm{H}), 4.07-4.01(\mathrm{~m}, 2 \mathrm{H}), 3.49(\mathrm{~d}, J$ $=9.8 \mathrm{~Hz}, 1 \mathrm{H}), 0.95(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 160.3,141.4,137.6,130.7,126.9$, 99.5, 75.7, 65.7, 65.5, 36.9, 26.1; HRMS (DART) for $\mathrm{C}_{13} \mathrm{H}_{19} \mathrm{BrNO}_{3}$ : calculated for $[\mathrm{M}+\mathrm{H}]^{+}$ 316.05483, found 316.05424; HPLC (Daicel Chiralpak ${ }^{\circledR}$ AD-3, ${ }^{n}$ hexane $/ \mathrm{PrOH}=9 / 1$, flow rate $0.6 \mathrm{~mL} / \mathrm{min}$ at a wavelength of 254 nm$)$; $\mathrm{t}_{\mathrm{R}}=21.6 \mathrm{~min}($ major, $S), \mathrm{t}_{\mathrm{R}}=36.5($ minor, $R) ;[\alpha]_{\mathrm{D}}{ }^{26}=$ 13.6 ( $\mathrm{c}=1.00, \mathrm{CHCl}_{3},>99.9 \%$ ee [after recrystallization]).

## Synthesis of (S)-1-(6-Bromo-3-(1,3-dioxolan-2-yl)pyridin-2-yl)-2,2-dimethylpropyl acetate

 (8) ${ }^{6}$

To a solution of (S)-1-(6-bromo-3-(1,3-dioxolan-2-yl)pyridin-2-yl)-2,2-dimethylpropan-1-ol (7, $1.02 \mathrm{~g}, 3.2 \mathrm{mmol}$ ) and $N, N$-dimethylpyridin-4-amine ( $197 \mathrm{mg}, 1.6 \mathrm{mmol}$ ) in 10 mL dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, was added acetic anhydride ( $0.6 \mathrm{~mL}, 6.5 \mathrm{mmol}$ ) under argon atmosphere. The reaction mixture was stirred at room temperature for 12 hours. After completion, the solvent was removed under reduced pressure then the crude product was purified by column chromatography (silica gel, ${ }^{n}$ hexane $/ \mathrm{EtOAc}=4 / 1$ ) to give the title product $\mathbf{8}$ as a white solid. $\mathbf{M p}:=101-103{ }^{\circ} \mathrm{C}$; $\mathbf{I R}$ (neat): $v=2966,2878,1731,1571,1558,1437,1364,1098,828 \mathrm{~cm}^{-1},{ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta$ 7.77 (d, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.40(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.26(\mathrm{~s}, 1 \mathrm{H}), 5.70(\mathrm{~s}, 1 \mathrm{H}), 4.16-4.02(\mathrm{~m}, 4 \mathrm{H})$, 2.08 (s, 3H), 1.02 (s, 9H); ${ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 170.8,156.8,141.3,137.3,131.9$, 127.2, 99.6, 78.2, 65.5, 65.4, 35.5, 26.6, 21.0; HRMS (DART) for $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{BrNO}_{4}$ : calculated for $[\mathrm{M}+\mathrm{H}]^{+}$358.06540, found 358.06499; HPLC (Daicel Chiralpak ${ }^{\circledR}$ AD-H + AD-3,
${ }^{n}$ hexane $/ 2 \operatorname{PrOH}=9 / 1$, flow rate $0.6 \mathrm{~mL} / \mathrm{min}$ at a wavelength of 254 nm ); $\mathrm{t}_{\mathrm{R}}=15.8 \mathrm{~min}$ (minor, $R) \mathrm{t}_{\mathrm{R}}=19.7 \mathrm{~min}($ major, $S) ;[\alpha]_{\mathrm{D}}{ }^{26}=-7.3\left(\mathrm{c}=1.01, \mathrm{CHCl}_{3},>99.9 \%\right.$ ee $)$.

Synthesis of (S)-1-(6-Bromo-3-formylpyridin-2-yl)-2,2-dimethylpropyl acetate (S6) ${ }^{7}$


To a solution of ( $S$ )-1-(6-bromo-3-(1,3-dioxolan-2-yl)pyridin-2-yl)-2,2-dimethylpropyl acetate $(8,1.41 \mathrm{~g}, 3.9 \mathrm{mmol})$ in 20 mL dry THF was added 10 mL of $10 \% \mathrm{HCl}$ aqueous solution. The reaction mixture was stirred at room temperature for 3 hours. Another portion of $10 \% \mathrm{HCl}$ (5 mL ) was added, and the reaction mixture was continuously stirred at the same temperature for 1 hour. After completion, the solution was extracted with EtOAc $(3 \times 20 \mathrm{~mL})$. The organic layers were combined and washed with brine, dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure. The crude product was purified by column chromatography (silica gel, ${ }^{n}$ hexane/EtOAc $=4 / 1$ ) to give the title compound S6 as a colorless oil. IR (neat) $v=2961,2872$, $1738,1694,1568,1374,1247,821 \mathrm{~cm}^{-1} ;{ }^{\mathbf{1}} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 10.50(\mathrm{~d}, J=1.1 \mathrm{~Hz}$, $1 \mathrm{H}), 7.98(\mathrm{dd}, J=8.2,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.54(\mathrm{dd}, J=8.2,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.05(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.12$ $(\mathrm{d}, J=1.6 \mathrm{~Hz}, 3 \mathrm{H}), 1.01(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 9 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 189.7,170.7,160.1$, 145.8, 138.9, 129.8, 127.7, 78.7, 35.5, 26.3, 20.9; HRMS (DART) for $\mathrm{C}_{13} \mathrm{H}_{17} \mathrm{BrNO}_{3}$ : calculated for $[\mathrm{M}+\mathrm{H}]^{+} 314.03918$, found 314.04009 ; HPLC (Daicel Chiralpak ${ }^{\circledR}$ AD-H + Daicel Chiralcel ${ }^{\circledR}$ OD-H, ${ }^{n}$ hexane $/ \mathrm{PrOH}=9 / 1$, flow rate $0.6 \mathrm{~mL} / \mathrm{min}$ at a wavelength of 254 nm$) ; \mathrm{t}_{\mathrm{R}}=18.7 \mathrm{~min}$ $(\operatorname{minor}, R), \mathrm{t}_{\mathrm{R}}=20.8 \mathrm{~min}($ major, $S) ;[\alpha]_{\mathrm{D}}{ }^{23}=-23.8\left(\mathrm{c}=0.51, \mathrm{CHCl}_{3},>99.9 \% \mathrm{ee}\right)$.

Synthesis of Methyl (S,E)-4-(2-(2-(1-acetoxy-2,2-dimethylpropyl)-6-bromopyridin-3yl)vinyl)benzoate (9) ${ }^{8,9}$
1)



S6

18-crown-6 (12 mol\%) $\mathrm{K}_{2} \mathrm{CO}_{3}$ (6.0 equiv) DCM:THF (2:3), rt, 16 h
2) NBS (1.1 equiv) benzoyl peroxide ( $10 \mathrm{~mol} \%$ ) $\mathrm{CCl}_{4}$, reflux, 1 h

81\%, $96 \%$ ee
recrystallization (>99.9\% ee)


9

To an oven-dried 100 mL 2-necked round-bottom flask equipped with a stirring bar were charged (4-(methoxycarbonyl)benzyl)triphenylphosphonium bromide ( $1.60 \mathrm{~g}, 3.25 \mathrm{mmol}$ ), $\mathrm{K}_{2} \mathrm{CO}_{3}$ ( 2.25 $\mathrm{g}, 16.3 \mathrm{mmol}$ ), 18 -crown- 6 ether ( $87 \mathrm{mg}, 0.3 \mathrm{mmol}$ ) under argon atmosphere. Subsequently, 10 mL dry THF and 14 mL dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ were added, then the suspension was stirred at room temperature for 5 minutes. To the resulting suspension was slowly transferred a solution of ( $S$ )-1-(6-bromo-3-formylpyridin-2-yl)-2,2-dimethylpropyl acetate (S6, $852 \mathrm{mg}, 2.7 \mathrm{mmol}$ ) in 11 mL dry THF. The reaction mixture was stirred at room temperature for 16 hours. After completion, the suspension was filtered through a celite pad, the filtrate was concentrated and then extracted with EtOAc ( $3 \times 30 \mathrm{~mL}$ ). The combined organic phase and washed with brine, dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure. The crude product was filtered through a short column (silica gel, ${ }^{n}$ hexane/EtOAc $=4 / 1$ ) to obtain a mixture of $(E)$ - and $(Z)$ isomers. To an oven-dried 50 mL round-bottom flask were charged the obtained crude product, $15 \mathrm{~mL} \mathrm{CCl}_{4}, N$-Bromosuccinimide ( $530 \mathrm{mg}, 3.0 \mathrm{mmol}$ ), and benzoyl peroxide ( $65 \mathrm{mg}, 0.3 \mathrm{mmol}$ ). The reaction mixture was refluxed at $80{ }^{\circ} \mathrm{C}$ for 1 hour. After completion, the reaction was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 15 \mathrm{~mL})$. The combined organic phase and washed with brine, dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure. The crude product was purified by column chromatography (silica gel, "hexane/EtOAc/MeOH $=9 / 1 / 0.5$ ). The resulting compound was recrystallized by using ${ }^{n}$ hexane to give the title compound $\mathbf{9}$ as a white crystal with single (S)-enantiomer (>99.9\% ee). Mp: $=97-99^{\circ} \mathrm{C}$; $\mathbf{I R}$ (neat) $v=2955,1714,1604,1568$, $1435,1260,1135,767 \mathrm{~cm}^{-1}{ }^{\mathbf{1}}{ }^{1} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 8.04(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.67(\mathrm{~d}, J$ $=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.64(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.57(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.39(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.93$ (d, $J=16.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.69(\mathrm{~s}, 1 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H}), 2.10(\mathrm{~s}, 3 \mathrm{H}), 0.99(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( 151 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 171.0,166.7,155.9,141.0,140.0,136.5,132.0,131.9,130.1,129.6,127.2,126.6$, 126.5, 78.9, 52.1, 36.0, 26.4, 20.9; HRMS (DART) for $\mathrm{C}_{22} \mathrm{H}_{25} \mathrm{BrNO}_{4}$ : calculated for $[\mathrm{M}+\mathrm{H}]^{+}$ 446.09670, found 446.09564; HPLC (Daicel Chiralpak ${ }^{\circledR}$ AD-H + Daicel Chiralcel ${ }^{\circledR}$ AD-3,
${ }^{n}$ hexane $/ 2 \mathrm{PrOH}=9 / 1$, flow rate $0.6 \mathrm{~mL} / \mathrm{min}$ at a wavelength of 254 nm ); $\mathrm{t}_{\mathrm{R}}=25.3 \mathrm{~min}$ (minor, $R), \mathrm{t}_{\mathrm{R}}=29.0 \mathrm{~min}($ major, $S) ;[\alpha]_{\mathrm{D}}^{25}=139.2\left(\mathrm{c}=0.52, \mathrm{CHCl}_{3},>99.9 \%\right.$ ee [after recrystallization] $)$.

Synthesis of Dimethyl 4,4'-((1E,1'E)-(6,6'-bis((S)-1-acetoxy-2,2-dimethylpropyl)-[2,2'-bipyridine]-5,5'-diyl)bis(ethene-2,1-diyl))dibenzoate (S7) ${ }^{10}$


To an oven-dried 15 mL sealed tube containing a stirring bar were charged Methyl ( $S, E$ )-4-(2-(2-(1-acetoxy-2,2-dimethylpropyl)-6-bromopyridin-3-yl)vinyl)benzoate ( $9,500 \mathrm{mg}, 1.12 \mathrm{mmol}$ ), $\mathrm{PPh}_{3}(706 \mathrm{mg}, 2.7 \mathrm{mmol})$, and $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(319 \mathrm{mg}, 1.3 \mathrm{mmol})$. The tube was transferred into an argon-filled glovebox, $\mathrm{Zn}(176 \mathrm{mg}, 2.7 \mathrm{mmol})$, and 10 mL dry DMF were added before closing with a screw cap and removing from the glovebox. The suspension was stirred at $70^{\circ} \mathrm{C}$ for 18 hours. After completion, it was quenched with $\mathrm{NH}_{3}$ solution in ethanol ( 3 mL ) followed by filtration through a celite pad. The filtrate was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 15 \mathrm{~mL})$ and the combined organic phase and washed with brine, dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated under reduced pressure. The crude product was purified by column chromatography (silica gel, ${ }^{n}$ hexane/EtOAc $=4 / 1$ ) to give the title compound $\mathbf{S 7}$ as a white solid with single $(S, S)$ enantiomer ( $>99.9 \%$ ee ). Mp: $=249-250^{\circ} \mathrm{C} ; \mathbf{I R}$ (neat) $v=2955,2872,1717,1604,1432,1365$, $1275,764 \mathrm{~cm}^{-1} ;{ }^{1} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 8.41(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 8.06(\mathrm{~d}, J=8.3 \mathrm{~Hz}$, 4H), 7.98 (d, $J=8.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.80(\mathrm{~d}, J=16.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.61$ (d, $J=8.3 \mathrm{~Hz}, 4 \mathrm{H}), 7.02$ (d, $J=$ $16.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), $5.85(\mathrm{~s}, 2 \mathrm{H}), 3.94(\mathrm{~s}, 6 \mathrm{H}), 2.09(\mathrm{~s}, 6 \mathrm{H}), 1.08(\mathrm{~s}, 18 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR (151 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 171.0,166.8,154.3,153.9,141.5,134.8(\mathrm{~d}, J=16.6 \mathrm{~Hz}), 132.5(\mathrm{~d}, J=3.6 \mathrm{~Hz}), 131.4$, $130.8,130.0(\mathrm{~d}, J=1.3 \mathrm{~Hz}), 129.4,126.6(\mathrm{dd}, J=30.9,2.4 \mathrm{~Hz}), 119.9,79.7,52.2,35.9,26.7(\mathrm{q}$, $J=6.0 \mathrm{~Hz}$ ), $21.0\left(\mathrm{q}, J=12.2 \mathrm{~Hz}\right.$ ); HRMS (ESI) for $\mathrm{C}_{44} \mathrm{H}_{48} \mathrm{~N}_{2} \mathrm{NaO}_{8}$ : calculated for $[\mathrm{M}+\mathrm{Na}]^{+}$ 755.3303, found 755.3285; HPLC (Daicel Chiralpak ${ }^{\circledR}$ AD-3, ${ }^{n}$ hexane $/ \mathrm{PrOH}=9 / 1$, flow rate $0.6 \mathrm{~mL} / \mathrm{min}$ at a wavelength of 254 nm$) ; \mathrm{t}_{\mathrm{R}}=20.0 \mathrm{~min}($ major, $S, S), \mathrm{t}_{\mathrm{R}}=23.8 \mathrm{~min}($ minor $, R, R)$; $[\alpha]_{\mathrm{D}}{ }^{24}=99.7\left(\mathrm{c}=0.25, \mathrm{CHCl}_{3},>99.9 \%\right.$ ee $)$.

Synthesis of 4,4'-((1E,1'E)-(6,6'-bis((S)-1-hydroxy-2,2-dimethylpropyl)-[2,2'-bipyridine]-5,5'-diyl)bis(ethene-2,1-diyl))dibenzoic acid (5) ${ }^{11}$



To a solution of dimethyl 4,4'-((1E,1'E)-(6,6'-bis((S)-1-acetoxy-2,2-dimethylpropyl)-[2,2'-bipyridine]-5,5'-diyl)bis(ethene-2,1-diyl))dibenzoate (S7, $420 \mathrm{mg}, 0.6 \mathrm{mmol}$ ) in 7.5 mL THF and 2.5 mL MeOH , was slowly added a solution of $\mathrm{LiOH} \cdot \mathrm{H}_{2} \mathrm{O}\left(359 \mathrm{mg}, 8.6 \mathrm{mmol}\right.$ in $2.5 \mathrm{~mL} \mathrm{H} \mathrm{H}_{2} \mathrm{O}$. The reaction mixture was stirred at room temperature for 12 hours. After completion, the solution was acidified by $10 \% \mathrm{HCl}$ before sonication (at $20^{\circ} \mathrm{C}, 1$ hour) then centrifugation ( 3500 rpm , 10 minutes). The sonication and centrifugation were done repeatedly for 3 cycles (washing with $\mathrm{H}_{2} \mathrm{O}$ ), followed by filtration washed several times with $\mathrm{H}_{2} \mathrm{O}$. The filtered solid was dissolved in THF and EtOAc then washed with $\mathrm{H}_{2} \mathrm{O}$ (5 times), dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under reduced pressure to give the title compound $\mathbf{5}$ as a pale yellow solid. $\mathbf{M p}$ : $=$ $>355{ }^{\circ} \mathrm{C}$; IR (neat) $v=2956,2925,2865,1682,1605,1585,1422,1290,1050,843 \mathrm{~cm}^{-1} ;{ }^{1} \mathbf{H}$ NMR ( 600 MHz, DMSO- $d_{6}$ ): $\delta 12.90(\mathrm{~s}, 2 \mathrm{H}), 8.46(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 8.29(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H})$, $7.94(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 4 \mathrm{H}), 7.89(\mathrm{~d}, J=16.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.76(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 4 \mathrm{H}), 7.31(\mathrm{~d}, J=16.0$ $\mathrm{Hz}, 2 \mathrm{H}), 5.08(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.92(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 0.92(\mathrm{~s}, 18 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( 151 MHz , DMSO- $d_{6}$ ): $\delta 167.6,158.8,153.0,141.9,135.1,131.9,130.8,130.4,127.7,127.4,119.8,100.0$, $77.4,37.5,27.0$; HRMS (ESI) for $\mathrm{C}_{38} \mathrm{H}_{40} \mathrm{~N}_{2} \mathrm{NaO}_{6}$ : calculated for $[\mathrm{M}+\mathrm{Na}]^{+} 643.2779$, found $643.2764 ;[\alpha]_{\mathrm{D}}{ }^{25}=27.5(\mathrm{c}=0.26, \mathrm{DMSO})$.

## 4. Synthesis of BPVB-NMOF-Sc-DS

### 4.1 General procedure for the synthesis of BPVB-NMOF-Sc-DS



In a 20 mL vial, a solution of $\mathrm{ZrCl}_{4}(7 \mathrm{mg}, 0.03 \mathrm{mmol})$ in 5 mL dry DMF and $\mathrm{H}_{2} \mathrm{O}(5.4 \mu \mathrm{~L}, 0.3$ $\mathrm{mmol})$ was sonicated at $20^{\circ} \mathrm{C}$ for 10 minutes. $\mathrm{AcOH}(54.9 \mu \mathrm{~L}, 0.9 \mathrm{mmol})$ was added, and then sonicated at $20^{\circ} \mathrm{C}$ for 5 minutes, followed by the addition of the chiral $\mathrm{H}_{2} \mathrm{BPVB}(5,18.6 \mathrm{mg}$, 0.03 mmol ) and sonication at $20^{\circ} \mathrm{C}$ for 5 minutes. The resulting solution was filtered by a PTFE membrane filter $(\operatorname{mesh}=0.2 \mu \mathrm{~m})$ into a 50 mL vial. Covered by a non-tight cap, the solution was put in an oil bath at $40^{\circ} \mathrm{C}$. After standing at $40^{\circ} \mathrm{C}$ for 5 days, it was allowed to cool down to room temperature and sonicated (at $20^{\circ} \mathrm{C}, 10$ minutes) then centrifugation ( $3500 \mathrm{rpm}, 10$ minutes). The sonication and centrifugation were done repeatedly for 3 cycles (washing with DMF), followed by filtration washed several times with DMF and THF. The title BPVB-NMOF ( $10.3 \mathrm{mg}, 47 \%$ ) was obtained as a pale yellow solid after drying in vacuo at room temperature for 2 hours. To analyze linker occupancy in BPVB-NMOF, 10 mg of the resulting BPVB-NMOF
was digested in a $1: 1$ mixture of a saturated $\mathrm{K}_{3} \mathrm{PO}_{4} / \mathrm{D}_{2} \mathrm{O}$ solution and DMSO- $d_{6}(0.6: 0.6 \mathrm{~mL})$. The suspension was sonicated at room temperature for 10 minutes, after which the organic layer was subjected to ${ }^{1} \mathrm{H}-\mathrm{NMR}$ analysis (Figure S5).


Figure S5. ${ }^{1} \mathrm{H}$-NMR Analysis of BPVB-NMOF.

BPVB-NMOF ( $24.7 \mathrm{mg}, 0.0056 \mathrm{mmol}$, the weight of solvent-containing in MOF was omitted) was charged into a 2 mL vial, followed by the addition of acetone $(0.6 \mathrm{~mL}, 10 \mathrm{mM})$. The suspension was sonicated at $20^{\circ} \mathrm{C}$ for 1 hour, $\mathrm{Sc}(\mathrm{DS})_{3}(76.4 \mathrm{mg}, 0.067 \mathrm{mmol}, 2.0$ equiv to bipyridine scaffold, assumed that no defect of the bipyridine linker during MOF construction) was added, followed by adding $\mathrm{H}_{2} \mathrm{O}(60 \mu \mathrm{~L})$. The suspension was vigorously stirred at room temperature for 18 hours. After completion, it was sonicated (at $20^{\circ} \mathrm{C}, 10$ minutes) and then centrifugation ( $3500 \mathrm{rpm}, 10$ minutes). The sonication and centrifugation were done repeatedly for 3 cycles (washing with $\mathrm{H}_{2} \mathrm{O}$ ), followed by filtration washed several times with $\mathrm{H}_{2} \mathrm{O}$ and acetone. The title BPVB-NMOF-Sc-DS ( 41.1 mg ) was obtained as a yellow solid after drying in vacuo at room temperature for 12 hours. BPVB-NMOF-Sc-DS has Sc and Zr loadings of 0.451 and $0.818 \mathrm{mmol} / \mathrm{g}$, respectively. Under the identical procedure, BPVB-NMOF-Sc-OTf has Sc and Zr loadings of 1.370 and $1.329 \mathrm{mmol} / \mathrm{g}$, respectively.

Table S6. Optimization in the preparation of BPVB-NMOF.




| Entry | MOF Conditions | Results | $\mathrm{ZrCl}_{4}$ Source |
| :---: | :---: | :---: | :---: |
| 1 | $90^{\circ} \mathrm{C}, 2 \mathrm{~d}$ | $62 \%$ yield, $57 \%$ ee | Wako:LEG4496 |
| 2 | $90^{\circ} \mathrm{C}, 2 \mathrm{~d}$ | $82 \%$ yield, $70 \%$ ee | Wako:WTK0876 |
| 3 | $70^{\circ} \mathrm{C}, 3 \mathrm{~d}$ | $95 \%$ yield, $76 \%$ ee | Wako:WTK0876 |
| $4^{a}$ | $70^{\circ} \mathrm{C}, 3 \mathrm{~d}$ | $88 \%$ yield, $79 \%$ ee | Wako:WTK0876 |
| $5^{a}$ | $60^{\circ} \mathrm{C}, 4 \mathrm{~d}$ | $92 \%$ yield, $82 \%$ ee | Wako:WTK0876 |
| $6^{a}$ | $60^{\circ} \mathrm{C}, 3 \mathrm{~d}$ | $94 \%$ yield, $82 \%$ ee | Aldrich $(\geq 99.9 \%)$ |
| $7^{a}$ | $50^{\circ} \mathrm{C}, 5 \mathrm{~d}$ | $91 \%$ yield, $84 \%$ ee | Aldrich $(\geq 99.9 \%)$ |
| $8^{a}$ | $40^{\circ} \mathrm{C}, 7 \mathrm{~d}$ | $93 \%$ yield, $84 \%$ ee | Aldrich $(\geq 99.9 \%)$ |

${ }^{a} 1.0$ equiv of $\mathrm{ZrCl}_{4}$ was applied.

Table S7. Reproducibility of post-synthetic metalation step.


| Trial | Sc Loading $(\mathrm{mmol} / \mathrm{g})$ | Zr Loading $(\mathrm{mmol} / \mathrm{g})$ | $\mathrm{Sc} / \mathrm{Zr}$ ratio (\%) |
| :---: | :---: | :---: | :---: |
| $1^{\text {st }}$ | 0.451 | 0.818 | 55 |
| $2^{\text {nd }}$ | 0.499 | 0.972 | 51 |
| $3^{\text {rd }}$ | 0.496 | 0.901 | 55 |

Table S8. Elemental analysis for nitrogen loading of the $3^{\text {rd }}$ trial (in Table S7)

| BPVB-MOF-Sc-DS (mg) | $\mathrm{C}(\%)$ | $\mathrm{H}(\%)$ | $\mathrm{N}(\%)$ | $\mathrm{N}(\mathrm{mmol} / \mathrm{g})$ | $\mathrm{Sc} / \mathrm{N}$ ratio (\%) |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1.0839 | 45.52 | 6.97 | 1.36 | 0.971 | 102 |




Figure S6. Preparation of chiral BPVB-NMOFs. a, A solution before solvothermal process. b, After precipitation at $40^{\circ} \mathrm{C}$. c, BPVB-NMOF after filtration and drying under vacuum. d, BPVB-NMOF-ScDS after filtration and drying under vacuum.

## 5. Control experiments for epoxide ring-opening reactions

Table S9. Control experiments in BPVB-NMOF-Sc-DS-catalyzed asymmetric ring-opening reaction of epoxide in water.

|  |  |  |  |
| :---: | :---: | :---: | :---: |
| Entry | Deviation from standard conditions | Yield (\%) | Ee (\%) |
| 1 | none | >99 | 85 |
| 2 | w/o catalyst | n.d. | - |
| 3 | $\mathrm{Sc}(\mathrm{DS})_{3}(1 \mathrm{~mol} \%), 1$ (1.2 mol\%) | 87 | 93 |
| 4 | $\mathrm{Sc}(\mathrm{DS})_{3}(1 \mathrm{~mol} \%), 5$ (1.2 mol\%) | 61 | 49 |
| $5^{a}$ | BPVB-MOF (w/o Sc) | 20 | 3 |
| 6 | BPVB-MOF-OTf | 22 | 61 |
| 7 | $0.5 \mathrm{~mol} \%$ | 82 | 76 |
| 8 | $0.25 \mathrm{~mol} \%$ | 77 | 70 |
| 9 | 12 h | 93 | 84 |
| 10 | 6 h | 80 | 83 |
| 11 | THF | 12 | 23 |
| 12 | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 32 | 49 |
| 13 | MeCN | 58 | 35 |
| 14 | w/o solvent | 89 | 31 |
| 15 | MOPS buffer | 74 | 70 |
| 16 | entry 3 in MOPS buffer | 27 | 86 |
| Reaction conditions: $11(0.2 \mathrm{mmol})$, BPVB-NMOF-Sc-DS ( $1 \mathrm{~mol} \%, 0.002 \mathrm{mmol}$ ), 12a ( 0.3 mmol$), \mathrm{H}_{2} \mathrm{O}(1 \mathrm{~mL}), \mathrm{rt}, 24 \mathrm{~h}$, yield after isolation. ${ }^{a} \mathrm{Zr}$ loading was controlled as same as the standard conditions ( $2.8 \mathrm{~mol} \% \mathrm{Zr}$ ). MOPS buffer ( $0.2 \mathrm{M}, \mathrm{pH}=7.5$ ). DS $=$ dodecyl sulfate. n.d. $=$ not detected . |  |  |  |

Table S10. Leaching issue of the standard reaction conditions.

| Metal ion leaching | Aqueous phase (\%) | Crude product (\%) |
| :---: | :---: | :---: |
| Sc | UDL | UDL |
| Zr | UDL | UDL |

Respective limit of detection for Sc and Zr is $0.65 \%$ and $0.44 \%$, respectively.
(a)

(b)

|  |  |  |
| :---: | :---: | :---: |
| Sc loading of BPVB-NMOF-Sc-DS | Yield (\%) | Ee (\%) |
| 0.430 | 96 | 78 |
| 0.584 | 98 | 76 |

Table S11. Lot difference. (a) reaction in water; (b) reaction in MOPS buffer.

## 6. Typical procedure for epoxide ring-opening reactions in water



To a 2 mL vial equipped with a stirring bar were charged $11(39.3 \mathrm{mg}, 0.2 \mathrm{mmol})$ and BPVB-NMOF-Sc-DS ( $0.002 \mathrm{mmol}, 1 \mathrm{~mol} \%$ based on $\mathrm{Sc}^{3+}$ ), followed by adding degassed water ( 1 mL ). It was flushed by argon before closing a cap to stir at room temperature for 1 hour. Aniline $(27.4 \mu \mathrm{~L}, 0.3 \mathrm{mmol})$ was loaded and again argon-flushed before stirring at room temperature. After 24 hours, the reaction mixture was filtered then the filtrate was extracted with EtOAc (3 x 15 mL ). The combined organic phase was washed with brine, and dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The crude product was obtained after filtration and concentration. It was then purified by preparative thin-layer chromatography ( ${ }^{n}$ hexane $/ \mathrm{EtOAc}=3 / 1$ ) to give the desired product 13 a .


Figure S7. Appearance of the reaction in water after completion.
<Conditions reported by Ueno et al.>


Figure S8. Comparison with Ueno's result.

To a 2 mL vial equipped with a stirring bar, BPVB-NMOF-Sc-DS ( $1.2 \times 10^{-4} \mathrm{mmol}, 5 \mathrm{~mol} \%$ based on $\mathrm{Sc}^{3+}$ ), $30 \mu \mathrm{~L}$ of acetonitrile ( MeCN ), and $540 \mu \mathrm{~L}$ of MOPS buffer ( $10 \mathrm{mM}, \mathrm{pH} 7.5$ ) were added. This was followed by the addition of $\mathbf{1 1}\left(15 \mu \mathrm{~L}, 0.16 \mathrm{M}\right.$ in $\left.\mathrm{MeCN}, 2.4 \times 10^{-3} \mathrm{mmol}\right)$ and $\mathbf{1 2}\left(15 \mu \mathrm{~L}, 0.16 \mathrm{M}\right.$ in $\left.\mathrm{MeCN}, 2.4 \times 10^{-3} \mathrm{mmol}\right)$. The vial was flushed with argon before closing the cap and stirring at room temperature or $40^{\circ} \mathrm{C}$. After 48 hours, the reaction mixture was quenched by adding saturated $\mathrm{NaHCO}_{3}(1 \mathrm{~mL})$ and EtOAc ( 3 mL ), then centrifuged at 3500 rpm for 10 minutes. This centrifugation step was repeated three times. The combined organic phase was washed with brine and dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The crude product was obtained after filtration and concentration. Conversion and enantioselectivity were determined by ${ }^{1} \mathrm{H}$ NMR analysis and HPLC, respectively.

## 7. Spectroscopic data of the products

The obtained analytical data for literature-known compounds is in full agreement with reported data.
(1S,2S)-1,2-diphenyl-2-(phenylamino)ethan-1-ol (13a) ${ }^{12}$

|  | $\text { MR }\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.32-7.22(\mathrm{~m}, 10 \mathrm{H}), 7.11-7.07$ |
| :---: | :---: |
|  |  |
|  | $1 \mathrm{H}), 4.55$ (d, $J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.63$ (br s, 1H); ${ }^{13} \mathrm{C}$ | 140.5, 140.2, 129.0, 128.5, 128.2, 127.8, 127.5, 127.2, 126.5, 117.9, 114.1, 78.0, 64.7; HPLC (Daicel Chiralcel ${ }^{\circledR}$ OD-H, ${ }^{n}$ hexane/ ${ }^{i} \mathrm{PrOH}=90 / 10$, flow rate $1.0 \mathrm{~mL} / \mathrm{min}$ at a wavelength of 254 $\mathrm{nm}, 35^{\circ} \mathrm{C}$ ); $\mathrm{t}_{\mathrm{R}}=13.0 \mathrm{~min}$ (minor), $\mathrm{t}_{\mathrm{R}}=17.2 \mathrm{~min}$ (major).

(1S,2S)-1,2-diphenyl-2-(p-tolylamino)ethan-1-ol (13b) $)^{13}$


The crude mixture was purified by preparative TLC ( ${ }^{h}$ hexane/toluene/EtOAc $=3 / 1 / 1$ ). ${ }^{1} \mathbf{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.32-7.20(\mathrm{~m}, 10 \mathrm{H}), 6.92(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.51$ (d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 4.85(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.51(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.21(\mathrm{~s}, 3 \mathrm{H}) ;$ ${ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 144.9,140.6,140.2,129.5,128.4,128.1,127.8,127.4$, 127.3, 127.2, 126.6, 114.3, 78.0, 65.2, 20.3; HPLC (Daicel Chiralpak ${ }^{\circledR}$ AD-H, ${ }^{n}$ hexane $/$ ' $\mathrm{PrOH}=95 / 5$, flow rate $1.0 \mathrm{~mL} / \mathrm{min}$ at a wavelength of 254 nm ); $\mathrm{t}_{\mathrm{R}}=25.9$ $\min ($ major $), \mathrm{t}_{\mathrm{R}}=28.0 \mathrm{~min}($ minor $)$.

## (1S,2S)-2-((4-Bromophenyl)amino)-1,2-diphenylethan-1-ol (13c) ${ }^{13,14}$



The crude mixture was purified by preparative TLC ( ${ }^{n}$ hexane/ $\mathrm{Et}_{2} \mathrm{O}=3 / 2$ ).
${ }^{1}$ H NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.35-7.19(\mathrm{~m}, 10 \mathrm{H}), 7.14(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.40$ $(\mathrm{d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 4.86(\mathrm{~d}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.78(\mathrm{~s}, 1 \mathrm{H}), 4.49(\mathrm{~d}, J=5.3 \mathrm{~Hz}, 1 \mathrm{H})$, $2.49(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 146.2,140.4,139.7,131.7,128.6$, 128.3, 128.0, 127.6, 127.2, 126.4, 115.6, 109.4, 77.9, 64.5; HPLC (Daicel Chiralcel ${ }^{\circledR}$ OD-H + Daicel Chiralpak ${ }^{\circledR}$ AD-H, ${ }^{n}$ hexane $/ \mathrm{PrOH}=95 / 5$, flow rate 1.0 $\mathrm{mL} / \mathrm{min}$ at a wavelength of 254 nm$) ; \mathrm{t}_{\mathrm{R}}=70.9 \mathrm{~min}($ major $), \mathrm{t}_{\mathrm{R}}=82.4 \mathrm{~min}($ minor $)$.
(1S,2S)-2-((4-Methoxyphenyl)amino)-1,2-diphenylethan-1-ol (13d) ${ }^{13}$


13d

The crude mixture was purified by twice preparative TLC ( ${ }^{n}$ hexane/EtOAc $=4 / 1$ then ${ }^{n}$ hexane/dichloromethane $=1 / 2$ ).
${ }^{1} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.35-7.11(\mathrm{~m}, 10 \mathrm{H}), 6.70-6.63(\mathrm{~m}, 2 \mathrm{H}), 6.54-$ $6.51(\mathrm{~m}, 2 \mathrm{H}), 4.82(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.50-4.23(\mathrm{~m}, 2 \mathrm{H}), 3.68(\mathrm{~s}, 3 \mathrm{H}), 2.85(\mathrm{br}$ $\mathrm{s}, 1 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 152.5,141.3,140.6,140.2,128.4,128.1$, $127.8,127.4,127.3,126.7,115.7,114.6,78.1$ (d, $J=3.1 \mathrm{~Hz}$ ), $66.2,55.6$ (q, $J=3.1$ Hz); HPLC (Daicel Chiralcel ${ }^{\circledR}$ OD-3, ${ }^{n}$ hexane $/ \mathrm{PrOH}=90 / 10$, flow rate $1.0 \mathrm{~mL} / \mathrm{min}$ at a wavelength of 254 nm ); $\mathrm{t}_{\mathrm{R}}=24.7 \mathrm{~min}$ (major), $\mathrm{t}_{\mathrm{R}}=30.1 \mathrm{~min}$ (minor).
(1S,2S)-2-((2-Methoxyphenyl)amino)-1,2-diphenylethan-1-ol (13e) ${ }^{12}$

$13 e$

The crude mixture was purified by preparative TLC ( ${ }^{n}$ hexane/EtOAc $=9 / 1$ ). ${ }^{1} \mathbf{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.37-7.11(\mathrm{~m}, 10 \mathrm{H}), 6.78(\mathrm{dd}, J=7.8,1.4 \mathrm{~Hz}$, $1 \mathrm{H}), 6.72-6.68(\mathrm{~m}, 1 \mathrm{H}), 6.67-6.63(\mathrm{~m}, 1 \mathrm{H}), 6.44-6.40(\mathrm{~m}, 1 \mathrm{H}), 5.30(\mathrm{br} \mathrm{s}$, 1 H ), 4.89 (d, $J=6.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.54$ (d, $J=6.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.89 (s, 3H), 2.80 (br s, $1 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 147.4,140.6,140.1,137.1,128.4,128.1$, 127.7, 127.4, 127.3, 126.7, 121.0, 117.1, 111.7, 109.6, 78.3 (d, $J=3.3 \mathrm{~Hz}$ ), 64.9, 55.6 (q, $J=3.2$ Hz ); HPLC (Daicel Chiralpak ${ }^{\circledR}$ AS-H, ${ }^{n}$ hexane $/ \mathrm{PrOH}=95 / 5$, flow rate $0.8 \mathrm{~mL} / \mathrm{min}$ at a wavelength of 254 nm ); $\mathrm{t}_{\mathrm{R}}=15.1 \mathrm{~min}($ major $), \mathrm{t}_{\mathrm{R}}=19.2 \mathrm{~min}($ minor $)$.

## (1S,2S)-1,2-Diphenyl-2-((3-(trifluoromethyl)phenyl)amino)ethan-1-ol (13f) ${ }^{14}$


${ }^{13 f}$

The crude mixture was purified by preparative TLC ( ${ }^{n}$ hexane $/ \mathrm{Et}_{2} \mathrm{O}=3 / 2$ ).
${ }^{1} \mathbf{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.35-7.23(\mathrm{~m}, 10 \mathrm{H}), 7.12(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.88$ (d, $J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.77(\mathrm{~s}, 1 \mathrm{H}), 6.61(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.99(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.90(\mathrm{~d}$, $J=5.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.56(\mathrm{~d}, J=5.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.49(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( 151 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 147.4,140.4,139.6,131.2(\mathrm{q}, J=31.7 \mathrm{~Hz}), 129.4,128.7,128.3,128.0$, 127.7, 127.2, 126.4, $124.2(\mathrm{q}, J=272.6 \mathrm{~Hz}), 116.5,114.1(\mathrm{q}, J=3.9 \mathrm{~Hz}), 110.5(\mathrm{q}, J=4.0 \mathrm{~Hz})$, 77.9, 64.3; HPLC (Daicel Chiralpak ${ }^{\circledR}$ AD-H, ${ }^{n}$ hexane $/$ $\operatorname{PrOH}=95 / 5$, flow rate $1.0 \mathrm{~mL} / \mathrm{min}$ at a wavelength of 254 nm ); $\mathrm{t}_{\mathrm{R}}=15.2 \mathrm{~min}$ (major), $\mathrm{t}_{\mathrm{R}}=17.6 \mathrm{~min}$ (minor).
(1S,2S)-2-(Naphthalen-1-ylamino)-1,2-diphenylethan-1-ol (13g) ${ }^{12}$


13g

The crude mixture was purified by preparative TLC ( ${ }^{n}$ hexane/EtOAc $=4 / 1$ ).
${ }^{1} \mathbf{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.07-8.02(\mathrm{~m}, 1 \mathrm{H}), 7.84-7.79(\mathrm{~m}, 1 \mathrm{H}), 7.58-$ $7.47(\mathrm{~m}, 2 \mathrm{H}), 7.40-7.27(\mathrm{~m}, 10 \mathrm{H}), 7.24-7.13(\mathrm{~m}, 2 \mathrm{H}), 6.40-6.32(\mathrm{~m}, 1 \mathrm{H})$, $5.58(\mathrm{~s}, 1 \mathrm{H}), 5.04(\mathrm{~d}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.74(\mathrm{~d}, J=5.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.56(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.1,140.6,139.9,134.2,128.6,128.6,128.3$, $128.0,127.5,127.2,126.5,126.4,125.6,124.8,123.9,120.0,117.7,106.6,78.2(\mathrm{~d}, J=3.4 \mathrm{~Hz})$, 64.4 (d, $J=1.7 \mathrm{~Hz}$ ); HPLC (Daicel Chiralcel ${ }^{\circledR}$ OD-3, ${ }^{n}$ hexane $/ \mathrm{iPrOH}=95 / 5$, flow rate 1.0 $\mathrm{mL} / \mathrm{min}$ at a wavelength of 254 nm ); $\mathrm{t}_{\mathrm{R}}=37.7 \mathrm{~min}($ minor $), \mathrm{t}_{\mathrm{R}}=80.0 \mathrm{~min}$ (major).
(1S,2S)-2-(Methyl(phenyl)amino)-1,2-diphenylethan-1-ol (13h) ${ }^{12}$


13h

The crude mixture was purified by preparative TLC ( ${ }^{n}$ hexane/EtOAc $=4 / 1$ ). ${ }^{1} \mathbf{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.43(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.31(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.26$ (t, $J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.22-7.15(\mathrm{~m}, 4 \mathrm{H}), 7.04(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.03-6.99(\mathrm{~m}$, $2 \mathrm{H}), 6.96-6.93(\mathrm{~m}, 1 \mathrm{H}), 5.32(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.91(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.01$ (br s, 1H), 2.73 (s, 3H); ${ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 151.3,140.6,134.6,129.1$, 128.7, 128.2, 127.9, 127.7, 127.6, 127.6, 120.3, 117.7, 73.7 (d, $J=2.8 \mathrm{~Hz}), 71.4(\mathrm{~d}, J=1.8 \mathrm{~Hz})$, 32.7 (d, $J=2.0 \mathrm{~Hz}$ ); HPLC (Daicel Chiralpak ${ }^{\circledR}$ AD-H, ${ }^{n}$ hexane $/ \mathrm{iPrOH}=95 / 5$, flow rate 0.8 $\mathrm{mL} / \mathrm{min}$ at a wavelength of 254 nm ); $\mathrm{t}_{\mathrm{R}}=15.3 \mathrm{~min}($ minor $), \mathrm{t}_{\mathrm{R}}=30.7 \mathrm{~min}($ major $)$.
( $\mathbf{1 R}, \mathbf{2 R}$ )-2-(1-Methyl-1H-indol-3-yl)-1,2-diphenylethan-1-ol (13i) ${ }^{12}$

$13 i$

The crude mixture was purified by preparative TLC ( ${ }^{n}$ hexane/EtOAc $=3 / 1$ ).
${ }^{1} \mathbf{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.46(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.30(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H})$, $7.25-7.12(\mathrm{~m}, 11 \mathrm{H}), 7.10-7.07(\mathrm{~m}, 1 \mathrm{H}), 7.06-7.02(\mathrm{~m}, 1 \mathrm{H}), 5.33(\mathrm{dd}, J=$ $8.1,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.58(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 2.52(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H})$; ${ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.4,141.9,137.1,128.6,128.1,128.0,127.9,127.3,127.2$, $126.8,126.2,121.9,119.5,119.1,113.7,109.2,77.7(\mathrm{~d}, J=1.5 \mathrm{~Hz}), 52.2,32.9(\mathrm{q}, J=2.1 \mathrm{~Hz})$; HPLC (Daicel Chiralcel ${ }^{\circledR}$ OD-H, ${ }^{n}$ hexane $/$ $\operatorname{PrOH}=80 / 20$, flow rate $0.8 \mathrm{~mL} / \mathrm{min}$ at a wavelength of 254 nm ); $\mathrm{t}_{\mathrm{R}}=16.3 \mathrm{~min}($ minor $), \mathrm{t}_{\mathrm{R}}=24.9 \mathrm{~min}($ major $)$.
(1S,2S)-1,2-Diphenyl-2-(phenylthio)ethan-1-ol (13j) ${ }^{12}$

$\mathrm{Hz}, 1 \mathrm{H}), 3.30(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 140.4,139.2,134.1,132.4,128.9,128.5$, 128.1, 128.0, 127.8, 127.4, 127.3, 126.9, 76.8, 64.0; HPLC (Daicel Chiralcel ${ }^{\circledR}$ OD-H, ${ }^{n}$ hexane $/ \mathrm{PrOH}=95 / 5$, flow rate $1.0 \mathrm{~mL} / \mathrm{min}$ at a wavelength of 254 nm ); $\mathrm{t}_{\mathrm{R}}=15.4 \mathrm{~min}$ (minor), $\mathrm{t}_{\mathrm{R}}=18.1 \mathrm{~min}$ (major).

## 8. Typical procedure for biocompatible epoxide ring-opening reactions



To a 3.5 mL vial equipped with a stirring bar were charged 11 ( $39.3 \mathrm{mg}, 0.2 \mathrm{mmol}$ ), BPVB-NMOF-Sc-DS ( $0.004 \mathrm{mmol}, 2 \mathrm{~mol} \%$ based on $\mathrm{Sc}^{3+}$ ), and an additive ( 50 mg ) followed by adding MOPS buffer ( $0.2 \mathrm{M}, \mathrm{pH} 7.5,2 \mathrm{~mL}$ ) and subsequent aniline ( $0.3 \mathrm{mmol}, 27.4 \mu \mathrm{~L}$ ). It was flushed by argon before closing a cap to stir at room temperature. After 48 hours, the reaction mixture was filtered then the filtrate was extracted with EtOAc ( $3 \times 10 \mathrm{~mL}$ ). The combined organic phase was washed with brine, and dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The crude product was obtained after filtration and concentration. It was then purified by preparative thin-layer chromatography ( ${ }^{n}$ hexane/EtOAc $=3 / 1$ ) to give the desired product 13a.


Figure S9. Appearance of the reaction in buffer in the presence of pepsin (after completion).

Table S12. Biocompatible conditions in BPVB-NMOF-Sc-DS-catalyzed asymmetric ring-opening reaction of epoxide.

|  | $\mathrm{PhNH}_{2}$ | + additive | $\xrightarrow[\substack{\text { MOPS buffer }(0.2 \mathrm{M}, \mathrm{M}, \mathrm{pH} 7.5,2 \mathrm{~mL}), \mathrm{H}, 48 \mathrm{~h}}]{\text { BPVB-NMO-S. }}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Entry | Additive | Yield (\%) | Ee (\%) | Sc Leaching (\%) | Zr Leaching (\%) |
| 1 | glutathione | >99 | 85 | UDL | UDL |
| 2 | lipase | 85 | 87 | 3.1 | 0.8 |
| 3 | pepsin | 83 | 84 | UDL | UDL |
| 4 | diastase | 60 | 84 | UDL | UDL |

Respective limit of detection for Sc and Zr is $0.65 \%$ and $0.44 \%$, respectively.
Table S13. Comparison of non-immobilized $\mathrm{Sc}(\mathrm{DS})_{3}$ for biocompatible conditions.

|  | $\mathrm{PhNH}_{2}+$ |  |  |
| :---: | :---: | :---: | :---: |
|  | 12a |  |  |
| Entry | Additive | Yield (\%) | Ee (\%) |
| 1 | glutathione | >99 (68) ${ }^{\text {a,b }}$ | $90(88)^{a, b}$ |
| 2 | lipase | 59 (n.d.) ${ }^{\text {a,c }}$ | $87(-)^{a, c}$ |
| 3 | pepsin | 39 (n.d.) ${ }^{\text {a,c }}$ | $84(-)^{a, c}$ |
| 4 | diastase | trace (n.d.) ${ }^{a, c}$ | $-(-)^{a, c}$ |

${ }^{a} 1 \mathrm{~mol} \% \mathrm{Sc}(\mathrm{DS})_{3}, 1.2 \mathrm{~mol} \%$ Bolm's ligand. ${ }^{b}$ Three halves equivalents of glutathione was used at $40^{\circ} \mathrm{C}, 24 \mathrm{~h} .{ }^{c}$ Run for 24 h .

## 9. General procedure for recovery and reuse experiments

To a 3.5 mL vial equipped with a stirring bar were charged $11(0.4 \mathrm{mmol}, 39.3 \mathrm{mg})$ and BPVB-NMOF-Sc-DS ( $0.008 \mathrm{mmol}, 2 \mathrm{~mol} \%$ based on $\mathrm{Sc}^{3+}$ ) followed by adding degassed water ( 2 mL ). It was flushed by argon before closing a cap to stir at room temperature for 1 hour. Aniline ( 0.6 $\mathrm{mmol}, 54.8 \mu \mathrm{~L}$ ) was loaded and again argon-flushed before stirring at room temperature. After 24 h , the reaction mixture was carefully filtered and washed with $\mathrm{H}_{2} \mathrm{O}$ and acetone, the obtained yellow solid was dried in vacuo at room temperature for 12 hours and directly applied for the next run (the reaction scale of the next run was calculated based on the remaining catalyst after recovery). On the other hand, the filtrate was extracted with EtOAc ( $3 \times 15 \mathrm{~mL}$ ). The combined organic phase was washed with brine, and dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The crude product was obtained after filtration and concentration. It was then purified by preparative thin-layer chromatography ( ${ }^{n}$ hexane/EtOAc $=3 / 1$ ) to give the desired product 13a. The leaching of metal ions was determined by ICP analysis in both aqueous solution and part of the crude product.


Table S14. Reuse experiment.


|  | $1^{\text {st }}$ Run | $2^{\text {nd }}$ Run | $3^{\text {rd }}$ Run |
| :---: | :---: | :---: | :---: |
| Yield, Ee | $96 \%$ yield, $87 \%$ ee | $>99 \%$ yield, $81 \%$ ee | $>99 \%$ yield, $78 \%$ ee |
| Sc Leaching | UDL | UDL | UDL |
| Zr Leaching | 1.1 | 1.5 | UDL |

[^2]
## 10. TG analysis

Thermogravimetric (TG) analysis of the prepared MOFs was conducted to assess their thermal characteristics. For BPVB-NMOF, a 20\% solvent loss was detected, leading to a calculated yield of $47 \%$ (Figure S10a). Furthermore, the thermal stability of the MOFs before and after Sccomplexation were compared. Remarkably, after subjecting BPVB-NMOF-Sc-DS to temperatures exceeding $600^{\circ} \mathrm{C}$, a higher residue was observed. This observation indicates that the presence of Sc species has a substantial impact on the thermal stability of the MOF (Figure S10b).
a)

b)


Figure S10. TG analysis. a, BPVB-NMOF. b, BPVB-NMOF-Sc-DS.

## 11. SEM analysis

The SEM image of BPVB-NMOF confirmed the successful preparation of nano-sized UiO-67 MOF. The analysis of particle size distribution provided a normal distribution curve, showcasing a well-defined pattern with an average particle diameter of 125 nm (Figure S11).
a)


Figure S11. SEM Analysis of BPVB-NMOF. a, SEM Image of BPVB-NMOF. b, Size distribution curve. c, Size distribution histogram.


Figure S12. SEM Image of BPVB-NMOF-Sc-DS.
12. PXRD analysis


Figure S13. PXRD spectrum of chiral $\mathrm{H}_{2} \mathrm{BPVB} 5$.


Figure S14. PXRD patterns of BPVB-NMOFs.

## 13. Water stability test of BPVB-NMOF and BPVB-NMOF-Sc-DS

To a 2 mL vial equipped with a stirring bar, BPVB-NMOF ( 10 mg ), and degassed water ( 1 mL ) were added. The vial was flushed with argon before closing the cap and stirring at room temperature for 24 hours. The resulting MOF was recovered by centrifugation, filtration, and drying in vacuo before PXRD and digestive ${ }^{1} \mathrm{H}-\mathrm{NMR}$ analyses.


Figure S15. PXRD patterns of water-treated MOFs.


Figure S16. Digestive ${ }^{1}$ H NMR spectra of BPVB-NMOF (above) and BPVB-NMOF-Sc-DS (below) $\left(\mathrm{K}_{3} \mathrm{PO}_{4} / \mathrm{D}_{2} \mathrm{O} / \mathrm{DMSO}-d_{6}\right)$.

## 14. STEM/EDS analysis




Figure S17. STEM/EDS analysis of BPV-MOF.



Figure S18. STEM/EDS analysis of BPV-MOF-Sc-DS.



Figure S19. STEM/EDS analysis of BPV-MOF-Sc-OTf.



Figure S20. STEM/EDS analysis of BPVB-NMOF.



Figure S21. STEM/EDS analysis of BPVB-NMOF-Sc-DS.



Figure S22. STEM/EDS analysis of BPVB-NMOF-Sc-OTf.
<Area analysis of BPVB-NMOF-Sc-OTf>














Figure S23. STEM/EDS analysis of BPVB-NMOF-Sc-DS (after the reaction with lipase).

## 15. $\mathbf{N}_{\mathbf{2}}$ adsorption/desorption isotherms

Nitrogen sorption experiments were conducted at 77 K after drying the samples ( $50-55 \mathrm{mg}$ ) at $50^{\circ} \mathrm{C}$ for 24 hours under vacuum.


Figure S24. $\mathrm{N}_{2}$ Adsorption/ desorption isotherms of BPVB-NMOF and BPVB-NMOF-Sc-DS.


Figure S25. Pore size distributions of BPVB-NMOF (red) and BPVB-NMOF-Sc-DS (blue).
16. XPS study


Figure S26. N 1s binding energy of BPVB-NMOF.


Figure S27. N 1s, Sc $2 p^{3 / 2}$, and Sc $2 p^{1 / 2}$ binding energy of BPVB-NMOF-Sc-DS.

## 17. Computational details

All DFT geometry optimization and frequency calculations were performed with Gaussian 16 program. ${ }^{15}$ Optimization were conducted in the gas phase with the B3LYP functional and 6$31 \mathrm{G}+(\mathrm{d}, \mathrm{p})$ basis sets for C, H, N, O and LanL2DZ for Sc. The optimization of $\mathrm{Sc}^{3+}-\mathrm{H}_{2} \mathrm{BPVB}$ complex was performed with the solute electron density (SMD) ${ }^{16}$ implicit solvation model for solvents. All stationary points were characterized by vibrational frequency calculations to confirm the minima for ground state optimizations (no imaginary frequencies) or transition state calculations (one imaginary frequency) and to obtain zero-point energy (ZPE) corrections. All energy values are reported in Hartrees and are the sum of electronic and thermal free energies as computed by Gaussian 16.

## Cartesian coordinate

The thermal corrections were computed at 298.15 K and 1 atm pressure. The Cartesian coordinates are in $\AA$.

## B3LYP optimized structures

## $\mathrm{H}_{2}$ BPDC

C - $0.662844000 .33480300-0.52846000$ C $0.66278200-0.33455400-0.52840500$ C - $0.767611001 .73180800-0.55639500$ C - $2.039277002 .27858500-0.57141000$ C - 3.17227400 1.44655200 -0.56995800 C - $2.968222000 .04654300-0.50639300$ N -1.73512100 -0.46513400 -0.48839400 N $1.735059000 .46540300-0.48841300$ C $2.96814000-0.04629300-0.50648500$ C $3.17217300-1.44630500-0.57004600$ C $2.03921900-2.27835600-0.57129000$ C $0.76754300-1.73155700-0.55621400$ C - $4.07809700-0.98481300-0.37243900$ O -3.71716800-2.09482000-1.20374300 C $4.078113000 .98490600-0.37230900$ O 3.71723000 2.09522200-1.20317900 C -4.33202900-1.42251400 1.10968500 C - 3.10672500-2.12242300 1.71710600 C -4.70600100-0.19173500 1.95455900 C - $5.52368100-2.399613001 .10727800$ C 4.332155001 .422007001 .10997800 C 3.106734002 .121323001 .71785300 C 4.706527000 .190922001 .95421700 C 5.523582002 .399388001 .10788700 H $0.129045002 .33847400-0.57245000$ H - $2.187281003 .35236600-0.59370100$ H $2.18718800-3.35214500-0.59348600$ H - $0.12912500-2.33820700-0.57214600$

$$
\begin{aligned}
& \text { H -5.00654000 -0.54865200 -0.73837400 } \\
& \text { H - } 4.52780300-2.56086900-1.43940300 \\
& \text { H } 5.006492000 .54875700-0.73844800 \\
& \text { H } 4.52782000 \text { 2.56167300 - } 1.43820000 \\
& \text { H -2.77450500-2.94433600 } 1.07789800 \\
& \text { H -3.36299300-2.52591400 } 2.70328200 \\
& \text { H -4.95982700 -0.50145100 } 2.97371800 \\
& \text { H - } 5.574832000 .332427001 .54089300 \\
& \text { H -3.87862100 } 0.520767002 .02398900 \\
& \text { H -6.40882500 - } 1.948498000 .64164900 \\
& \text { H - } 5.28274800-3.326562000 .57650400 \\
& \text { H } 2.774151002 .943344001 .07897600 \\
& \text { H } 2.267202001 .432852001 .83946800 \\
& \text { H } 5.57549600-0.332788001 .54025300 \\
& \text { H } 3.87937600-0.521872002 .02335100 \\
& \text { H } 6.408738001 .948757000 .64180500 \\
& \text { H } 5.282327003 .326606000 .57773200 \\
& \text { H } 5.791152002 .673248002 .13379700 \\
& \text { H } 4.960332000 .500201002 .97351300 \\
& \text { H -2.26695200 - } 1.434268001 .83880400 \\
& \text { H } 3.363024002 .524551002 .70413100 \\
& \text { H -5.79108700-2.67403200 } 2.13308200 \\
& \text { C - } 4.473195002 .17336100-0.61104200 \\
& \text { О -4.60727800 3.32763100-0.25023100 } \\
& \text { О -5.51805600 1.47873900-1.12988300 } \\
& \text { H - } 6.27841300 \text { 2.08481500 -1.11366500 } \\
& \text { C 4.47316700-2.17299800 -0.61135200 } \\
& \text { O 4.60771600-3.32685000-0.24942600 } \\
& \text { O 5.51738600-1.47876500-1.13199900 }
\end{aligned}
$$

H 6.27792700-2.08461100-1.11582400

## $\mathrm{H}_{2} \mathrm{BPV}$

C -0.72616300 0.14816100-0.55750700
C $0.72611900-0.14810300-0.55754400$
C - $1.195710001 .47005500-0.60175800$
C -2.56443100 1.67255300-0.61302200
C - $3.451001000 .57908200-0.56892600$
C - $2.87530900-0.71591300-0.51702800$ N - $1.55493700-0.90306200-0.51421700$ N $1.554895000 .90311800-0.51425200$ C $2.875268000 .71597300-0.51706500$ C $3.45095400-0.57902900-0.56897000$ C $2.56437900-1.67249500-0.61308200$ C $1.19566100-1.46999800-0.60180800$ C - $3.71139900-1.97959600-0.40723800$ O -3.16938300-2.91443400-1.34012000 C $3.711355001 .97966500-0.40731300$ O $3.169177002 .91456800-1.34002400$ C -3.79480400-2.55280900 1.04738700 C - $2.42211100-3.016739001 .55817400$ C - $4.36784800-1.475364001 .98453100$ C - $4.76273000-3.752676001 .02954700$ C 3.794972002 .552761001 .04735300 C 2.422341003 .016637001 .55836300 C 4.368155001 .475272001 .98435700 C 4.762891003 .752634001 .02947600 H - $0.486591002 .28723200-0.64695800$ H -2.96213900 $2.67921800-0.68930300$ H $2.96209500-2.67915700-0.68937700$ H $0.48654100-2.28717300-0.64700900$ H -4.74277400 -1.74510000 -0.69735800 H -3.75552000 -3.68053400-1.37253100 H $4.742684001 .74521000-0.69763300$ H $3.755150003 .68080100-1.37224500$ H-1.96331200 -3.72102800 0.85923400 H - $2.53868800-3.512225002 .52857000$ H -4.50635900-1.88915200 2.98875700 H -5.34100900 -1.11119300 1.63678200 H -3.69669400 -0.61560100 2.06825100 H -5.74268400 -3.47156800 0.62487300 H -4.36926400 -4.59042800 0.44211700 H 1.963428003 .720976000 .85955100 H 1.734657002 .177375001 .68098900 H 5.341302001 .111173001 .63649500 H 3.697055000 .615471002 .06809000 H 5.742791003 .471553000 .62465100 H 4.369362004 .590443000 .44216900 H 4.918510004 .127255002 .04629900 H 4.506738001 .889006002 .98859600 H-1.73440000 -2.17749500 1.68077000 H 2.539043003 .512041002 .52878700 H -4.91821700 -4.12737500 2.04636100 C - 4.89776600 0.79524000 -0.58587700

H -5.51940300 0.00415700-0.99104000 C $4.89771400-0.79521300-0.58587500$ H 5.51938500-0.00415300-0.99102600 C -5.52332500 1.90139700-0.13742500 H - 4.992290002 .727174000 .32311700 C $5.52322700-1.90142200-0.13748300$ H $4.99215800-2.727214000 .32299200$ C - $6.978770002 .09252100-0.19884500$ O -7.554329003 .092110000 .19369400 O -7.65648400 1.03897900-0.73711400 H -8.59624200 1.28313300-0.72435800 С $6.97866600-2.09258700-0.19889600$ O 7.55419100-3.09222100 0.19358100 O $7.65642500-1.03901200-0.73704600$ H 8.59617600-1.28319400-0.72427900

## $\mathrm{H}_{2} \mathrm{BPY}$

С $0.71163000-0.20739500-0.46935500$ C - $0.711624000 .20732100-0.46934700$ C $1.06655800-1.56458000-0.51320000$ C $2.41320100-1.88710900-0.51184400$ С $3.37473500-0.85916700-0.46590800$ C 2.91938200 0.48454200-0.42020900 N $1.622336000 .77756500-0.42316100$ $\mathrm{N}-1.62232900-0.77763500-0.42312800$ C -2.91937900 -0.48460600-0.42017200 C - $3.374727000 .85909300-0.46588900$ C -2.41318600 1.88704200-0.51183600 C - $1.066548001 .56451100-0.51319900$ C $3.892322001 .64603700-0.36128100$ O $3.460365002 .58817900-1.34632100$ C - $3.89234000-1.64608600-0.36125900$ O -3.46041800-2.58819200-1.34633600 C 4.033896002 .282339001 .05989500 C 2.725631002 .943968001 .51941400 C 4.456187001 .195665002 .06451600 C 5.147363003 .345723000 .98861900 C - $4.03394300-2.282404001 .05990900$ C - $2.72574200-2.944167001 .51942000$ C - $4.45611600-1.195697002 .06454400$ C -5.14751600-3.34567600 0.98864500 H $0.28973500-2.31774300-0.55284600$ H $2.74138300-2.92037500-0.55035000$ H -2.74137400 $2.92030600-0.55033900$ H -0.28971700 $2.31766600-0.55285200$ H $4.882093001 .26026700-0.63367300$ H $4.178526003 .21576200-1.49359900$ H -4.88210100-1.26027600 -0.63363500 H -4.17840900 -3.21604900 -1.49328400 H 2.371253003 .663144000 .77619500 H 2.892110003 .471583002 .46524900 H 4.627939001 .644735003 .04829900 H 5.381890000 .699169001 .75577500 H 3.681615000 .431073002 .18104200

H 6.088096002 .915396000 .62527000 H 4.870173004 .183917000 .33922700 Н - $2.37147400-3.663431000 .77623300$ H - 1.93414900 -2.20712800 1.67254300 H -5.38172500 -0.69904100 1.75577000 H -3.68143000 -0.43123100 2.18112900 H -6.08822300 -2.91525300 0.62534500 H - 4.87044900 -4.18388600 0.33921500 H -5.33437000 -3.76316500 1.98340100 H -4.62799600 -1.64476300 3.04830600 H 1.934132002 .206841001 .67261600 H -2.89225500 -3.47171100 2.46528900 H 5.334224003 .763188001 .98338400 C $4.75759700-1.17723100-0.46433000$ C 5.94024400-1.45568400-0.45946200 C - $4.757580001 .17720500-0.46431800$ C -5.94020200 1.45575300-0.45947100 С $7.36232200-1.64943900-0.42511100$ O 8.17446500-0.74668000-0.34262700 О $7.70906500-2.95728200-0.49166400$ H 8.68076400-2.99187600-0.46037900 C - $7.362267001 .64959500-0.42518800$ O -8.17451100 $0.74691300-0.34310500$ O -7.70885800 $2.95754000-0.49128700$ H-8.68055400 2.99222300-0.46007400

## $\mathrm{H}_{2} \mathrm{BPVB}$

C $0.74201700-0.01101900-0.59841000$ C -0.73941000-0.00548200-0.60189700
C $1.46354200-1.21375100-0.64277200$
С $2.84602400-1.14432700-0.65186600$
C $3.504107000 .09988700-0.59977500$
C $2.682657001 .25464500-0.54676000$
N $1.351155001 .18153600-0.55101900$ N - $1.34866800-1.19781200-0.55271500$ C - $2.68017500-1.27127100-0.55386000$ C - $3.50160500-0.11670100-0.61361600$ C - $2.843216001 .12728700-0.66874400$ C - $1.460852001 .19698200-0.65457900$ C $3.258291002 .65518400-0.42527500$ O $2.580782003 .46841100-1.38411600$ C - $3.25526500-2.67214100-0.43256300$ O -2.56871300-3.48715800-1.38332900 С 3.189864003 .239927001 .02505300 C 1.741184003 .458992001 .48789400 C 3.909876002 .280990001 .98942000 C 3.934102004 .590256001 .03026200 C - 3.19752100-3.25269500 1.01991400 C - $1.75203500-3.465519001 .49546300$ C - $3.92886000-2.293715001 .97565100$ C - $3.93717900-4.605510001 .02237300$ H $0.92899700-2.15430400-0.69354400$ Н 3.43384600-2.05293000 -0.73442500 H -3.43056400 2.03559500-0.75754600

H -0.92629600 2.13741400-0.70740300 H $4.324504002 .61815000-0.68108600$ H $2.985890004 .34446200-1.37569100$ H -4.31932700 -2.63755100 -0.69737900 H -2.97615900 -4.36215400 -1.37950800 H 1.191013004 .074831000 .77138200 H 1.737998003 .965556002 .45978500 H 3.942339002 .715482002 .99409400 H 4.941119002 .087428001 .67411200 H 3.394089001 .318746002 .05700800 H 4.962467004 .482891000 .66409100 H 3.424391005 .345591000 .42028000 H -1.19361800 -4.08056800 0.78471300 H -1.22037300 -2.51764300 1.59913600 H -4.95858400 -2.10572800 1.65209300 H -3.41777400-1.32901300 2.04408900 H -4.96249200 -4.50287300 0.64642700 H -3.41915500 -5.36079200 0.41945600 H -3.99489500 -5.00314900 2.04080300 H -3.96705200 -2.72521500 2.98141500 H 1.205211002 .513268001 .58855300 H -1.75516900 -3.97033300 2.46825600 Н 3.983725004 .990836002 .04796800 C $4.966874000 .16775900-0.60909300$ H $5.411102001 .04555300-1.06794200$ C - $4.96419300-0.18496700-0.62643200$ Н -5.40659400-1.06805500 -1.07651600 C $5.78276800-0.76742400-0.07903700$ H 5.33833600-1.60742500 0.45143600 C -5.78229300 $0.75563900-0.10939300$ H -5.33996000 1.603243000 .41060600 С $7.24538100-0.74852900-0.09461800$ С $7.997464000 .15600100-0.87210900$ C $7.94519500-1.675711000 .70315400$ C $9.385323000 .14506400-0.83923400$ H $7.492193000 .86359300-1.52061100$ C $9.33380700-1.691955000 .73951400$ H $7.38373300-2.385463001 .30399200$ C 10.06711000-0.77873100-0.02989200 H $9.950245000 .84436200-1.44426600$ H 9.86934500-2.40423000 1.35740600 C - $7.244721000 .73470300-0.12705700$ C -7.94607200 1.683865000 .64159300 C -7.99567300 -0.19403800-0.87835900 C - 9.336265001 .702164000 .67720100 H - 7.385692002 .411414001 .22183300 C -9.38188100-0.18182900-0.84538800 H - $7.48973900-0.92158700-1.50377000$ C-10.06665100 $0.76471400-0.06581400$ H -9.85913000 2.437703001 .27694100 H -9.95920200 -0.89460200-1.42392800 C $11.54560200-0.830335000 .03661200$ O $12.18870600-1.608780000 .71689400$ O 12.15058500 0.09789300-0.75257700

H 13.10717700-0.02265600-0.63534200 C - $11.547112000 .73619500-0.06662600$ O -12.23110100 -0.05428800-0.69103200 O -12.10268300 1.699547000 .71652100 H-13.06522100 1.59145700 0.64471800

## $\mathbf{S c}^{3+}-\mathrm{H}_{2} \mathbf{B P V B}$

C $0.73267700-0.99795800-0.01013400$
C - $0.73267600-0.997947000 .01073300$
C 1.51863800-2.14508600-0.12221800
C $2.89443000-2.00731000-0.20302100$
C $3.51149800-0.74319900-0.12736100$
C 2.653487000 .366060000 .04302600
N 1.313551000 .223294000 .06294300
N - $1.313544000 .22328200-0.06264200$
C - $2.653462000 .36609500-0.04283100$
C -3.51148700-0.74313000 0.12776200
C -2.89443500-2.00722200 0.20381300
C - $1.51864500-2.145054000 .12308500$
С 3.131317001 .786305000 .22925600
O $2.030981002 .59211500-0.28381600$
C - $3.131329001 .78628100-0.22927800$
O - 2.031008002 .592216000 .28371600
Sc - $0.000016002 .04565100-0.00006200$ C 3.474076002 .195555001 .69983500 C 2.303842001 .920284002 .65359400 C 4.712297001 .420347002 .17951800 С 3.805131003 .699674001 .70920900 C - 3.47407500 2.19541100-1.69986500 C -2.30379200 1.92013900-2.65357600 C - 4.71223500 1.42010900-2.17951500 C -3.80520600 3.69952100-1.70934500 H $1.06442000-3.12528200-0.18947800$ H 3.50198700-2.89137300-0.35630600 H - $3.50204500-2.891212000 .35731500$ H - $1.06444000-3.125240000 .19056000$ H $3.996431001 .99306000-0.40329600$ H $2.280021003 .52912400-0.34432100$ H - 3.996433001 .993043000 .40328700 H -2.28018600 3.529200000 .34403700 H 1.409103002 .486615002 .36579400 H 2.578306002 .243379003 .66258300 H 4.979057001 .772578003 .18089300 H 5.572206001 .585370001 .52338800 H 4.524291000 .345835002 .24421800 H 4.586140003 .938168000 .97845300

H 2.925663004 .319135001 .50040600 H -1.40912600 $2.48665100-2.36594200$ H -2.05137300 $0.85668100-2.69699400$ H -5.57214800 1.58501300-1.52336100 H -4.52413800 $0.34561500-2.24426900$ H -4.58625700 3.93804100-0.97864200 H -2.92577600 4.31904500-1.50055400 H -4.17016800 3.98334300-2.70114700 H -4.97906300 1.77234800-3.18087200 H 2.051570000 .856814002 .69729300 H -2.57838500 2.24289500-3.66263800 H 4.170134003 .983562002 .70097400 C $4.95724700-0.58161900-0.25694100$ H 5.30944100 0.41381900-0.50042200 C - $4.95725300-0.581564000 .25717200$ H -5.30948200 0.413869000 .50065900 C $5.85745500-1.57218700-0.07680000$ H 5.51086700-2.56646300 0.19663200 C -5.85742700-1.57212100 0.07681300 H -5.51079600 -2.56637400 -0.19665000 С $7.30977300-1.44403600-0.18483700$ С $7.96988800-0.22026700-0.42351700$ C $8.09793000-2.60122200-0.02936300$ C $9.35676000-0.16692500-0.50759500$ H $7.399793000 .69593800-0.53684600$ С $9.48662200-2.54408700-0.11958400$ H 7.60750600-3.55185300 0.16158000 С 10.13807300-1.32613600-0.36005100 H $9.847104000 .78327800-0.68781100$ Н $10.07345200-3.447913000 .00024200$ C - $7.30976100-1.443982000 .18470900$ C -8.09788900 -2.60120100 0.02929300 C - $7.96990400-0.220203000 .42320100$ C -9.48658300-2.54407700 0.11940500 H -7.60742300 -3.55184100 -0.16149700 C -9.35678900-0.16687300 0.50720300 H -7.39984800 0.696036000 .53645900 C - 10.13806900-1.32610100 0.35972400 H -10.07341500 -3.44790400 -0.00037800 H -9.84714700 0.783347000 .68729500 C - $11.65113600-1.261980000 .45243000$ O -12.18703500 -0.12097400 0.62019300 O -12.29456200-2.35403500 0.35465700 C $11.65114200-1.26195600-0.45283600$ O $12.29456100-2.35407500-0.35566300$ O 12.18702300-0.12088000-0.62010300

## 18. References

1. K. Manna, T. Zhang, F. X. Greene, W. Lin, Bipyridine- and phenanthroline-based metal-organic frameworks for highly efficient and tandem catalytic organic transformations via directed C-H activation. J. Am. Chem. Soc. 2015, 137, 2665-2673.
2. C. Chen, P. Pan, Z. Deng, D. Wang, Q. Wu, L. Xu, T. Hou, S. Cui, Discovery of 3,6-diaryl1 H -pyrazolo[3,4-b]pyridines as potent anaplastic lymphoma kinase (ALK) inhibitors. Bioorg. Med. Chem. Lett. 2019, 29, 912-916.
3. P. Bałczewski, A. Bodzioch, E. Róz ycka-Sokołowska, B. Marciniak, P. Uznan ski, First approach to nitrogen-containing fused aromatic hydrocarbons as targets for organoelectronics utilizing a new transformation of $o$-protected diaryl methanols. Chem. Eur. J. 2010, 16, 23922400.
4. C. Bolm, M. Zehnder, D. Bur, Optically active bipyridines in asymmetric catalysis. Angew. Chem. Int. Ed. Engl. 1990, 29, 205-207.
5. J. Mao, B. Wan, F. Wu, S. Lu, First example of asymmetric transfer hydrogenation in water induced by a chiral amino alcohol hydrochloride. Tetrahedron Lett. 2005, 46, 7341-7344.
6. K. C. Nicolaou, C. F. Claiborne, P. G. Nantermet, E. A. Couladouros, E. J. Sorensen, Synthesis of novel toxoids. J. Am. Chem. Soc. 1994, 116, 1591-1592.
7. X. Wu, S. Sun, S. Xu, J. Cheng, Rh-Catalyzed annulation of ortho-C-H bonds of 2arylimidazoles with 1,4,2-dioxazol-5-ones toward 5-arylimidazo[1,2-c]quinazolines. Adv. Synth. Catal. 2018, 360, 1111-1115.
8. Z. Dogan, R. Paulini, J. A. R. Stütz, S. Narayanan, C. Richert, 5'-Tethered stilbene derivatives as fidelity- and affinity-enhancing modulators of DNA duplex stability. J. Am. Chem. Soc. 2004, 126, 4762-4763.
9. Md. M. Baag, A. Kar, N. P. Argade, N-Bromosuccinimide-dibenzoyl peroxide/ azabisisobutyronitrile: a reagent for $Z$ - to $E$-alkene isomerization. Tetrahedron 2003, 59, 6489-6492.
10. S. Lauzon, T. Ollevier, 2,2'-Bipyridine- $\alpha, \alpha$ '-trifluoromethyl-diol ligand: synthesis and application in the asymmetric $\mathrm{Et}_{2} \mathrm{Zn}$ alkylation of aldehydes. Chem. Commun., 2021, 57, 11025-11028.
11. J. M. Gawel, A. E. Shouksmith, Y. S. Raouf, N. Nawar, K. Toutah, S. Bukhari, P. Manaswiyoungkul, O. O. Olaoye, J. Israelian, T. B. Radu, A. D. Cabral, D. Sina, A. Sedighi, E. D. de Araujo, P. T. Gunning, PTG-0861: A novel HDAC6-selective inhibitor as a
therapeutic strategy in acute myeloid leukaemia. Eur. J. Med. Chem. 2020, 201, 112411112435.
12. T. Kitanosono, F. Lu, K. Masuda, Y. Yamashita, S. Kobayashi, Efficient recycling of catalystsolvent couples from Lewis acid catalyzed asymmetric reactions in water. Angew. Chem. Int. Ed. 2022, 61, e202202335.
13. B. Gao, Y. Wen, Z. Yang, X. Huang, X. Liu, X. Feng, Asymmetric ring opening of mesoepoxides with aromatic amines catalyzed by a new proline-based $N, N^{\prime}$ - dioxide-indium tris(triflate) complex. Adv. Synth. Catal. 2008, 350, 385-390.
14. C. Ogawa, S. Azoulay, S. Kobayashi, Bismuth triflate-chiral bipyridine complex catalyzed asymmetric ring opening reactions of meso-epoxide in water. Heterocycles, 2005, 66, 201206.
15. Gaussian 16, Revision B.01, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, G. A. Petersson, H. Nakatsuji, X. Li, M. Caricato, A. V. Marenich, J. Bloino, B. G. Janesko, R. Gomperts, B. Mennucci, H. P. Hratchian J. V. Ortiz, A. F. Izmaylov, J. L. Sonnenberg, D. Williams-Young, F. Ding, F. Lipparini, F. Egidi, J. Goings, B. Peng, A. Petrone, T. Henderson, D. Ranasinghe, V. G. Zakrzewski, J. Gao, N. Rega, G. Zheng, W. Liang, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, K. Throssell, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. J. Bearpark, J. J. Heyd, E. N. Brothers, K. N. Kudin, V. N. Staroverov, T. A. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. P. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, J. M. Millam, M. Klene, C. Adamo, R. Cammi, J. W. Ochterski, R. L. Martin, K. Morokuma, O. Farkas, J. B. Foresman, D. J. Fox, Gaussian, Inc., Wallingford CT, 2016.
16. A. V. Marenich, C. J. Cramer, D. G. Truhlar, Universal Solvation Model Based on the Generalized Born Approximation with Asymmetric Descreening. J. Chem. Theory Comput. 2009, 5, 2447-2467.
17. ${ }^{1} \mathrm{H} \&{ }^{13} \mathrm{C}$ NMR spectra and HPLC chromatograms


WS0003A-13C
N
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Nis



WS0015-1H



WS0015-13C




WS0730A

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$\stackrel{\text { \% }}{\substack{\circ \\ i}}$



WS0026A-13C




[^3]mV


Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area\% | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 21.595 | 274058 | 50.079 | 11409 | 50.079 |
| 2 | 36.515 | 273191 | 49.921 | 6907 | 49.921 |
| Total |  | 547249 | 100.000 | 18315 |  |

mV


Peak Table
Det.A 254 nm

| Peak | Ret. Time | Area | Area\% | Height | Conc. |
| ---: | :---: | :---: | ---: | ---: | ---: |
| 1 | 21.702 | 1296040 | 100.000 | 52724 | 100.000 |
| Total |  | 1296040 | 100.000 | 52724 |  |





WS0464-13C








Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area\% | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 15.823 | 2887304 | 50.048 | 196682 | 50.048 |
| 2 | 19.697 | 2881813 | 49.952 | 174069 | 49.952 |
| Total |  | 5769117 | 100.000 | 370752 |  |

mV


Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area $\%$ | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | :---: |
| 1 | 19.450 | 868318 | 100.000 | 52963 | 100.000 |
| Potal |  | 868318 | 100.000 | 52963 |  |






Peak Table
Det.A 254 nm

| $\|r\| r e a k$ | Peak | Ret. Time | Area | Area\% | Height |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 18.728 | 1948051 | 49.927 | 102636 | Conc. |
| 2 | 20.701 | 1953769 | 50.073 | 947773 | 50.027 |
| Total |  | 3901819 | 100.000 | 197409 |  |



Peak Table

| Peak Table |  |  |  |
| :--- | :---: | :---: | :---: |
| Det.A 254 nm      <br> Peak\# Ret. Time Area Area $\%$ Height Conc. <br> 1 20.773 7652002 100.000 353064 100.000 <br> Total  7652002 100.000 353064  |  |  |  |




WS0861-13C

$\left.\begin{array}{llllllllllllllllllllllllll}\hline 0 & 210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 \\ f 1(\mathrm{ppm})\end{array}\right)$



Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area $\%$ | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 30.550 | 494087 | 100.000 | 18549 | 100.000 |
| Total |  | 494087 | 100.000 | 18549 |  |




WS0922-13C






| 10 | 210 | 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 <br> $f 1(\mathrm{ppm})$ | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | -1 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |

mV


Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area $\%$ | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 20.035 | 197057 | 50.177 | 4465 | 50.177 |
| 2 | 23.802 | 195664 | 49.823 | 3602 | 49.823 |
| Total |  | 392721 | 100.000 | 8067 |  |

mV


Peak Table
Det.A 254 nm

| Deak\# | Ret. Time | Area | Area $\%$ | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 21.966 | 181904 | 100.000 | 3199 | 100.000 |
| Potal |  | 181904 | 100.000 | 3199 |  |





13a



Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area $\%$ | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 12.840 | 10910629 | 50.001 | 446897 | 50.001 |
| 2 | 17.290 | 10910142 | 49.999 | 338205 | 49.999 |
| Total |  | 21820771 | 100.000 | 785102 |  |



Peak Table




mV


Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area\% | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 25.970 | 15087690 | 49.819 | 431086 | 49.819 |
| 2 | 28.015 | 15197150 | 50.181 | 403049 | 50.181 |
| Total |  | 30284840 | 100.000 | 834134 |  |



Peak Table

| Det.A 254 nm Peak |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Peak\# | Ret. Time | Area | Area\% | Height | Conc. |
| 1 | 25.889 | 9848182 | 91.690 | 282695 | 91.690 |
| 2 | 27.966 | 892545 | 8.310 | 27809 | 8.310 |
| Total |  | 10740727 | 100.000 | 310504 |  |



mV


Peak Table
Det.A 254 nm
Det.A 254 nm

|  |  |  |  |  |  |
| ---: | ---: | ---: | ---: | ---: | ---: |
| Peak\# | Ret. Time | Area | Area $\%$ | Height | Conc. |
| 1 | 70.750 | 22225974 | 50.087 | 264085 | 50.087 |
| 2 | 82.049 | 22148604 | 49.913 | 229850 | 49.913 |
| Total |  | 44374578 | 100.000 | 493935 |  |

mV


Peak Table
Det.A 254 nm

| $\|r\| r e a$ | Peak\# | Ret. Time | Area | Area $\%$ | Height |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 70.887 | 18026472 | 92.472 | 214916 | 92.472 |
| 2 | 82.408 | 1467458 | 7.528 | 16048 | 7.528 |
| Total |  | 19493930 | 100.000 | 230964 |  |


|  |  | $\stackrel{\square}{\circ}$ |
| :---: | :---: | :---: |
|  | ササ | 9 |




mV


Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area $\%$ | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 25.736 | 178979 | 49.273 | 5986 | 49.273 |
| 2 | 30.324 | 184262 | 50.727 | 5199 | 50.727 |
| Total |  | 363241 | 100.000 | 11184 |  |

mV


Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area $\%$ | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 24.737 | 2107024 | 86.768 | 58687 | 86.768 |
| 2 | 30.116 | 321314 | 13.232 | 7770 | 13.232 |
| Total |  | 2428338 | 100.000 | 66457 |  |






Peak Table
Det.A 254 nm

| Peak Table |  |  |  |  |  |
| ---: | ---: | ---: | ---: | ---: | ---: |
| Peak\# | Ret. Time | Area | Area $\%$ | Height | Conc. |
| 1 | 15.098 | 14821632 | 91.387 | 511862 | 9.387 |
| 2 | 19.228 | 1396955 | 8.613 | 28287 | 8.613 |
| Total |  | 16218587 | 100.000 | 540148 |  |



mV


Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area\% | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 15.917 | 16150709 | 49.935 | 716324 | 49.935 |
| 2 | 18.427 | 16192635 | 50.065 | 639652 | 50.065 |
| Total |  | 32343344 | 100.000 | 1355977 |  |


Det.A 254 nm

| Peak\# | Ret. Time | Area | Area\% | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 15.234 | 3288419 | 92.832 | 152805 | 92.832 |
| 2 | 17.603 | 253902 | 7.168 | 11537 | 7.168 |
| Total |  | 3542321 | 100.000 | 164342 |  |





13g


mV


Peak Table
Det.A 254nm

| Peak\# | Ret. Time | Area | Area\% | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 40.515 | 12256747 | 50.006 | 198916 | 50.006 |
| 2 | 81.866 | 12253623 | 49.994 | 100531 | 49.994 |
| Total |  | 24510370 | 100.000 | 299447 |  |

mV


Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area $\%$ | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 37.703 | 2993650 | 10.478 | 65562 | 10.478 |
| 2 | 80.020 | 25577915 | 89.522 | 216746 | 89.522 |
| Total |  | 28571566 | 100.000 | 282308 |  |

##  <br>  <br> $\stackrel{N}{\mathrm{O}}$ <br> $\bar{\square}$ <br> $\stackrel{N}{\stackrel{N}{\sim}}$



13h


mV


Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area $\%$ | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 15.513 | 2996178 | 49.867 | 152236 | 49.867 |
| 2 | 31.299 | 301258 | 50.133 | 85557 | 50.133 |
| Total |  | 6008337 | 100.000 | 237792 |  |

mV


Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area $\%$ | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 15.314 | 988606 | 6.020 | 48740 | 6.020 |
| 2 | 30.771 | 15432469 | 93.980 | 415638 | 93.980 |
| Total |  | 16421076 | 100.000 | 464378 |  |



mV


Peak Table
???A 254 nm

| Peak\# | Ret. Time | Area | Area\% | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 15.958 | 2311099 | 50.110 | 68842 | 50.110 |
| 2 | 28.656 | 2300926 | 49.890 | 37832 | 49.890 |
| Total |  | 4612024 | 100.000 | 106674 |  |

mV


Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area\% | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 16.314 | 150008 | 10.006 | 4988 | 10.006 |
| 2 | 24.944 | 1349143 | 89.994 | 27029 | 89.994 |
| Total |  | 1499151 | 100.000 | 32018 |  |




13j


mV


Peak Table

| Peak\# | Ret. Time | Area | Area\% | Height | Conc. |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | 13.843 | 12450290 | 50.006 | 394715 | 50.006 |
|  | 20.668 | 12447336 | 49.994 | 296333 | 49.994 |
| Tota |  | 24897626 | 100.000 | 691048 |  |



Peak Table
Det.A 254 nm

| Peak\# | Ret. Time | Area | Area\% | Height | Conc. |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 15.382 | 558858 | 16.182 | 10668 | 16.182 |
| 2 | 18.086 | 2894724 | 83.818 | 88160 | 83.818 |
| Total |  | 3453582 | 100.000 | 98828 |  |


[^0]:    ${ }^{a}$ Ratio of occupancy was determined by ${ }^{1} \mathrm{H}$ NMR analysis. ${ }^{b} 0.09 \mathrm{mmol}$ scale. ${ }^{c}$ Combination of 4 portions.

[^1]:    ${ }^{a}$ Isolated yield.

[^2]:    Respective limit of detection for Sc and Zr is $0.65 \%$ and $0.44 \%$, respectively.

[^3]:    $\left.\begin{array}{lllllllllllllllllllllllllll}10 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 \\ \mathrm{f} 1(\mathrm{ppm})\end{array}\right)$

