

## **ELECTRONIC SUPPLEMENTARY INFORMATION**

### **Reactivity of Metal Hydrides with CO<sub>2</sub>: Going Beyond Formate with a High-Valent Cationic Pentahydride Mo(VI) Complex**

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# Table of Content

|                                                                                                                                                             |           |
|-------------------------------------------------------------------------------------------------------------------------------------------------------------|-----------|
| <b>I. General Information .....</b>                                                                                                                         | <b>3</b>  |
| I.1. Experimental .....                                                                                                                                     | 3         |
| I.2. Nuclear Magnetic Resonance.....                                                                                                                        | 3         |
| I.3. Crystallographic data collection and refinement .....                                                                                                  | 3         |
| I.4. Miscellaneous.....                                                                                                                                     | 3         |
| <b>II. Syntheses, Characterization Data and Spectra for the New Compounds.....</b>                                                                          | <b>4</b>  |
| II.1. [Mo(depe) <sub>2</sub> H <sub>5</sub> ][B(C <sub>6</sub> H <sub>5</sub> ) <sub>4</sub> ] ( <b>1.BPh<sub>4</sub></b> ) .....                           | 4         |
| II.2. [Mo(depe) <sub>2</sub> (HCOO)H <sub>2</sub> ][B(C <sub>6</sub> H <sub>5</sub> ) <sub>4</sub> ] ( <b>3.BPh<sub>4</sub></b> ) .....                     | 9         |
| II.3. [Mo(depe) <sub>2</sub> (CO) <sub>2</sub> H][B(C <sub>6</sub> H <sub>5</sub> ) <sub>4</sub> ] ( <b>4.BPh<sub>4</sub></b> ) .....                       | 21        |
| II.4. [Mo(depe) <sub>2</sub> (CH <sub>2</sub> S <sub>2</sub> )H] ( <b>6.BPh<sub>4</sub></b> ).....                                                          | 28        |
| II.5. [Mo(depe) <sub>2</sub> ( <sup>i</sup> PrNCHN <sup>i</sup> Pr)H <sub>2</sub> ] ( <b>7.BPh<sub>4</sub></b> ).....                                       | 33        |
| <b>III. Computational Details .....</b>                                                                                                                     | <b>39</b> |
| III.1. General Information.....                                                                                                                             | 39        |
| III.2. Detailed Results.....                                                                                                                                | 39        |
| III.3. Cartesian Coordinates and E+ZPE and free energies (G) of optimized structures .....                                                                  | 44        |
| <b>IV. Crystallographic data .....</b>                                                                                                                      | <b>64</b> |
| IV.1. [Mo(depe) <sub>2</sub> H <sub>5</sub> ][B(C <sub>6</sub> H <sub>5</sub> ) <sub>4</sub> ] ( <b>1.BPh<sub>4</sub></b> ) – Molecular structure .....     | 64        |
| IV.2. [Mo(depe) <sub>2</sub> (CO) <sub>2</sub> H][B(C <sub>6</sub> H <sub>5</sub> ) <sub>4</sub> ] ( <b>4.BPh<sub>4</sub></b> ) – Molecular structure ..... | 64        |
| IV.3. [Mo(depe) <sub>2</sub> H <sub>5</sub> ][B(C <sub>6</sub> H <sub>5</sub> ) <sub>4</sub> ] ( <b>1.BPh<sub>4</sub></b> ) – Tables.....                   | 65        |
| IV.4. [Mo(depe) <sub>2</sub> (HCOO)H <sub>2</sub> ][B(C <sub>6</sub> H <sub>5</sub> ) <sub>4</sub> ] ( <b>3.BPh<sub>4</sub></b> ) – Tables.....             | 66        |
| IV.5. [Mo(depe) <sub>2</sub> (CO) <sub>2</sub> H][B(C <sub>6</sub> H <sub>5</sub> ) <sub>4</sub> ] ( <b>4.BPh<sub>4</sub></b> ) – Tables.....               | 67        |
| IV.6. [Mo(depe) <sub>2</sub> ( <sup>i</sup> PrNCHN <sup>i</sup> Pr)H <sub>2</sub> ] ( <b>7.BPh<sub>4</sub></b> ) – Tables .....                             | 68        |
| <b>V. Supplementary references .....</b>                                                                                                                    | <b>69</b> |

## I. General Information

### I.1. Experimental

All reactions were performed in oven-dried glassware with rigorous exclusion of air and moisture, using a nitrogen-filled *Jacomex* glove box ( $O_2 < 1$  ppm,  $H_2O < 1$  ppm) or an argon-filled MBraun glove box ( $O_2 < 1$  ppm,  $H_2O < 1$  ppm), both equipped with vacuum line. Solvents used were pre-dried (tetrahydrofuran and *n*-pentane by passing through a Puresolv MD 7 solvent purification machine; benzene and 1,2-dichlorobenzene by distillation over  $CaH_2$ ), degassed by freeze-pump-thaw cycles, dried over molecular sieves and stored in the glove box. THF- $d_8$  (purchased from *Eurisotop*) and *o*- $C_6D_4Cl_2$  (purchased from *Deutero*) were degassed by freeze-pump-thaw cycles, dried over molecular sieves and stored in the glove box.

$[MoH_4(depe)_2]$ ,<sup>[1]</sup>  $[MoH_5(depe)_2][HB(C_6F_5)_3]$ <sup>[2]</sup> and  $[NEt_3H][B(C_6H_5)_4]$ <sup>[3]</sup> were synthesized according to reported procedures and stored in the glove box.

### I.2. Nuclear Magnetic Resonance

$^1H$ ,  $^{11}B$ ,  $^{19}F$  and  $^{31}P$  NMR spectra were recorded in THF- $d_8$  by using Wilmad quick pressure valve NMR tubes or J. Young high-vacuum valve NMR tubes on a *Bruker* Avance III 400 spectrometer. Chemical shifts are in parts per million (ppm) downfield from tetramethylsilane and are referenced to the residual solvent resonances as the internal standard ( $C_6HD_3Cl_2$ :  $\delta$  reported = 7.17 ppm, 6.90 ppm;  $C_4HD_7O_2$ :  $\delta$  reported = 3.58, 1.72 for  $^1H$  NMR).  $^{11}B$ ,  $^{19}F$  and  $^{31}P$  NMR spectra were calibrated according to the IUPAC recommendation using a unified chemical shift scale based on the proton resonance of tetramethylsilane as primary reference.<sup>[4,5]</sup> Data are reported as follows: chemical shift, multiplicity (br = broad, s = singlet, d = doublet, t = triplet, q = quartet, quin = quintet, sext = sextet; m = multiplet), coupling constant (Hz), and integration.

### I.3. Crystallographic data collection and refinement

Data for compounds (**1.BPh<sub>4</sub>**), (**3.BPh<sub>4</sub>**), (**4.BPh<sub>4</sub>**) and (**7.BPh<sub>4</sub>**) were collected at low temperature (100 K) either on a *Bruker* Kappa Apex II diffractometer using a Mo-K $\alpha$  radiation ( $\lambda = 0.71073\text{\AA}$ ) micro-source or on a XtaLAB Synergy, Dualflex, HyPix diffractometer using micro-focus sealed X-ray tube, PhotonJet (Cu) X-ray source. The structures have been solved using the new dual-space algorithm program SHELXT<sup>[6]</sup> and refined by means of least-squares procedures using the SHELXL-2018,<sup>[7]</sup> program included in the software package WinGX version 1.63<sup>[8]</sup> or with the aid of the software package Crystal<sup>[9]</sup>. The Atomic Scattering Factors were taken from International tables for X-Ray Crystallography<sup>[10]</sup>. Hydrogen atoms were placed geometrically and refined using a riding model. All non-hydrogens atoms were anisotropically refined. Drawing of molecules in the following figures were performed with the program Mercury<sup>[11]</sup> with 30% probability displacement ellipsoids for non-hydrogen atoms. The crystal structures have been deposited at the Cambridge Crystallographic Data Centre and allocated the deposition numbers CCDC **2365476**, **2365477**, **2365478** and **2365557**.

### I.4. Miscellaneous

Infrared (IR) spectra were recorded in a nitrogen filled *Jacomex* glove box ( $O_2 < 1$  ppm,  $H_2O < 1$  ppm) on an *Agilent* Cary 630 FT-IR spectrophotometer equipped with a transmission module and are reported in wavenumbers ( $cm^{-1}$ ).

## II. Syntheses, Characterization Data and Spectra for the New Compounds

### II.1. [Mo(depe)<sub>2</sub>H<sub>5</sub>][B(C<sub>6</sub>H<sub>5</sub>)<sub>4</sub>] (1.BPh<sub>4</sub>)

#### II.1.1. Experimental procedure

[MoH<sub>4</sub>(depe)<sub>2</sub>] (60.0 mg, 117 μmol, 1.00 equiv.) was dissolved in THF (1 mL). A solution of [HNEt<sub>3</sub>][B(C<sub>6</sub>H<sub>5</sub>)<sub>4</sub>] (50.0 mg, 117 μmol, 1.00 equiv.) in THF (1 mL) was added. The reaction mixture was left at room temperature for 30 min, under continuous stirring. The resulting pale-yellow solution was then dropwise transferred in a large volume (ca. 10 mL) of precooled pentane (-40°C), triggering immediate precipitation of an off-white solid. The solution was decanted off and the solid washed twice with cold pentane (2 x 5 mL). The resulting off-white powder was then dried under an Ar atmosphere at room temperature yielding the desired compound (85.0 mg, 87%). Single crystals suitable for X-ray diffraction were obtained by storing at -40°C a concentrated THF solution of the complex layered by pentane over a few days.

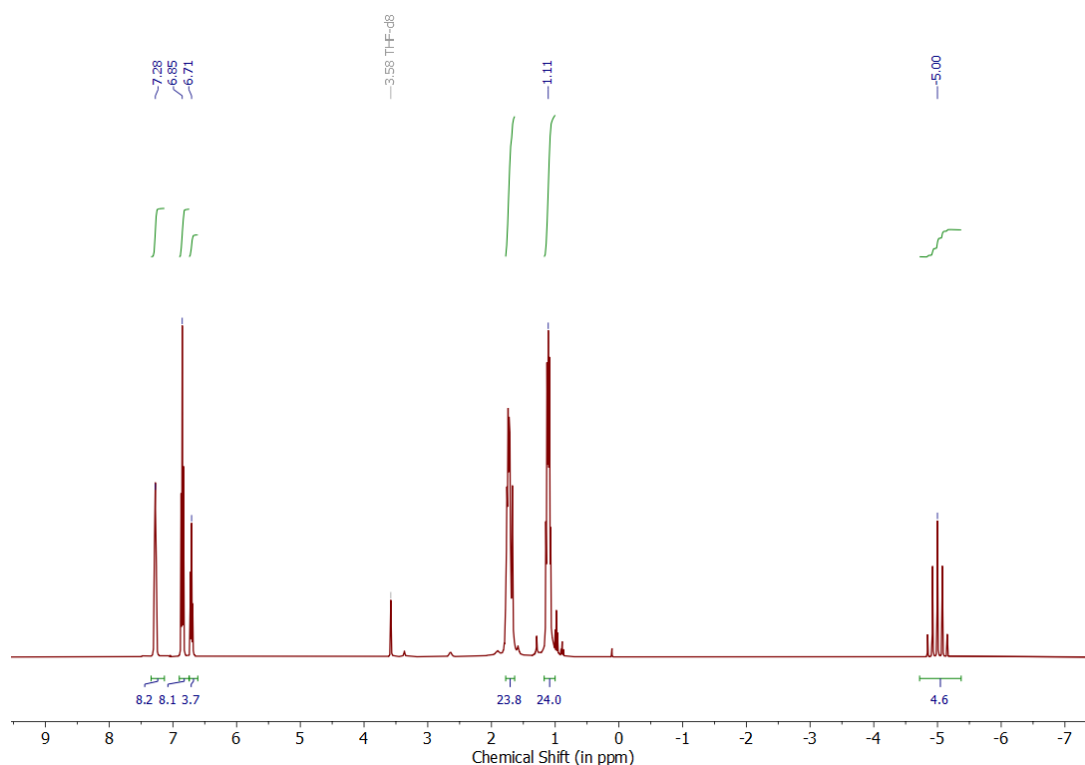
**<sup>1</sup>H-NMR** (400 MHz, THF-d<sub>8</sub>) δ 7.28 (br. m, 8H, *meta*-C<sub>6</sub>H<sub>5</sub>), 6.85 (t, <sup>3</sup>J<sub>(H-H)</sub> = 7.4 Hz, 8H, *ortho*-C<sub>6</sub>H<sub>5</sub>), 6.71 (t, <sup>3</sup>J<sub>(H-H)</sub> = 7.2 Hz, 4H, *para*-C<sub>6</sub>H<sub>5</sub>), 1.85 – 1.58 (m, 24H, -CH<sub>2</sub>-CH<sub>3</sub> & -CH<sub>2</sub>-CH<sub>2</sub>-), 1.11 (m, 24H, -CH<sub>2</sub>-CH<sub>3</sub>), -5.00 (quin, <sup>2</sup>J<sub>(P-H)</sub> = 30.8 Hz, 5H, Mo-H).

**<sup>11</sup>B-NMR** (128 MHz, THF-d<sub>8</sub>) δ -6.8.

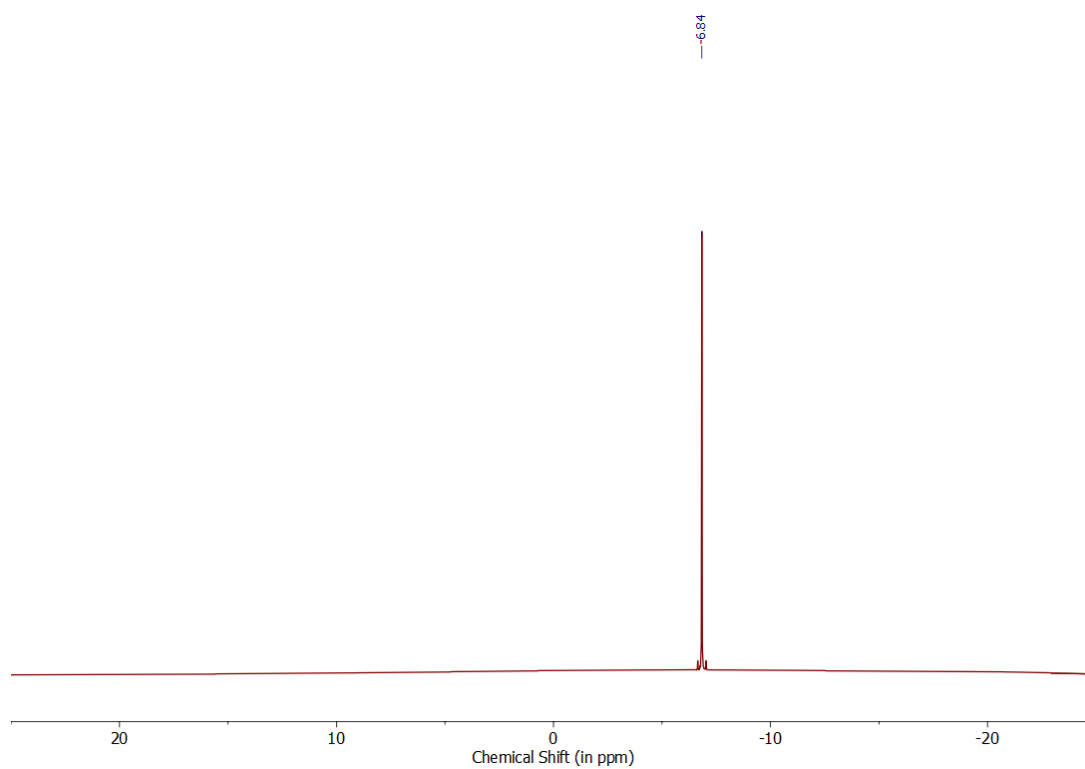
**<sup>13</sup>C{<sup>1</sup>H}-NMR** (101 MHz, THF-d<sub>8</sub>) δ 164.0 (q, <sup>2</sup>J<sub>(B-C)</sub> = 50.1 Hz, *ipso*-C<sub>6</sub>H<sub>5</sub>), 135.9 (br. s, *meta*-C<sub>6</sub>H<sub>5</sub>), 124.4 (m, *ortho*-C<sub>6</sub>H<sub>5</sub>), 120.5 (s, *para*-C<sub>6</sub>H<sub>5</sub>), 25.7 (m, -CH<sub>2</sub>-), 25.1 (m, -CH<sub>2</sub>-), 7.7 (s, -CH<sub>3</sub>).

**<sup>31</sup>P{<sup>1</sup>H}-NMR** (162 MHz, THF-d<sub>8</sub>) δ 73.7 (s). **<sup>31</sup>P{<sup>1</sup>H}<sub>sel</sub>-NMR** (162 MHz, THF-d<sub>8</sub>) δ 73.7 (sext, <sup>2</sup>J<sub>(P-H)</sub> = 30.8 Hz).

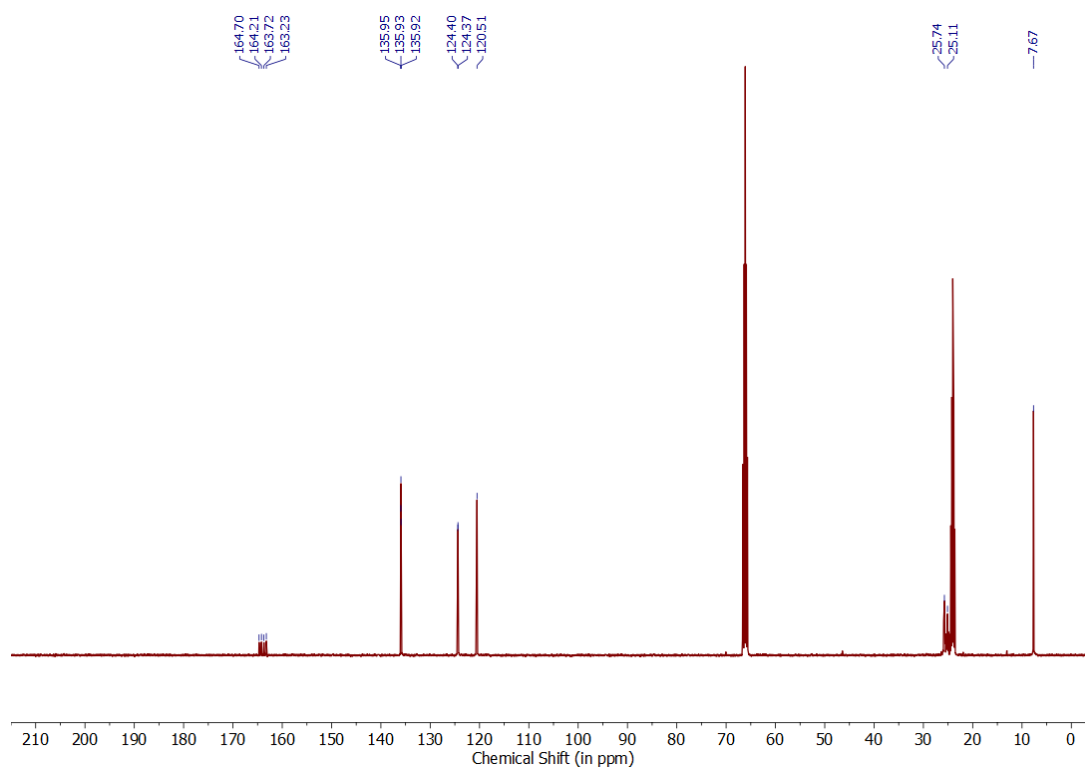
#### II.1.2. <sup>1</sup>H-NMR spectrum (400 MHz, THF-d<sub>8</sub>)



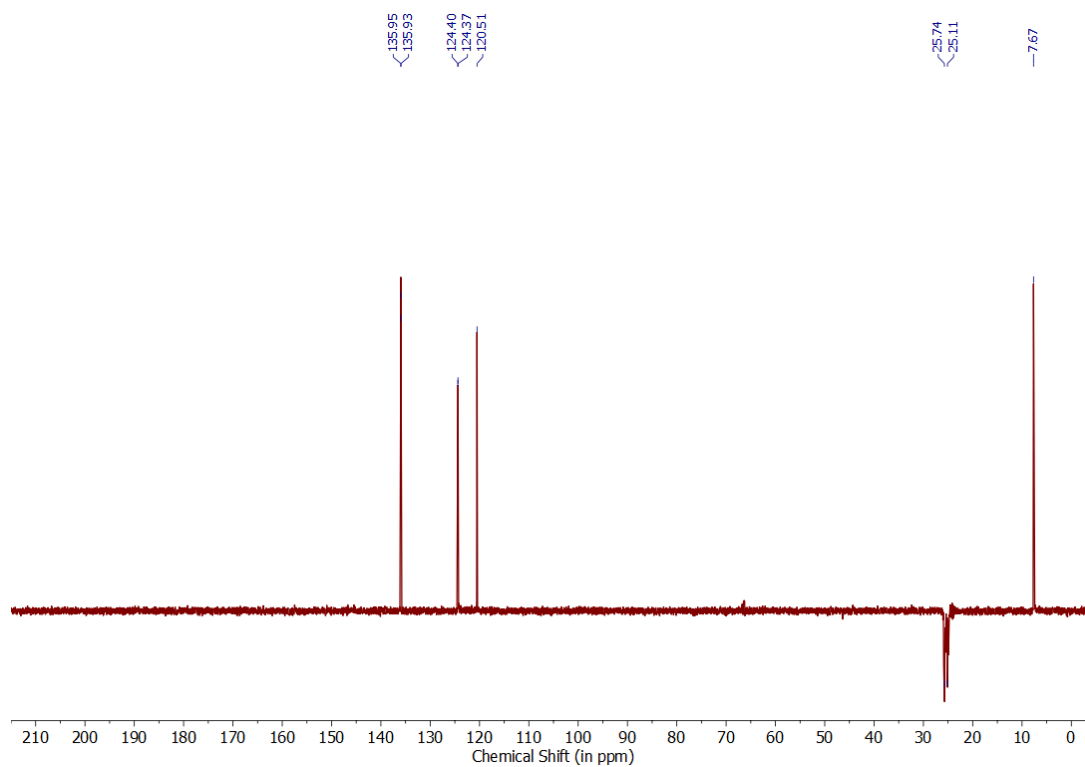
**II.1.3.  $^1\text{B}\{^1\text{H}\}$ -NMR spectrum (128 MHz, THF- $\text{d}_8$ )**



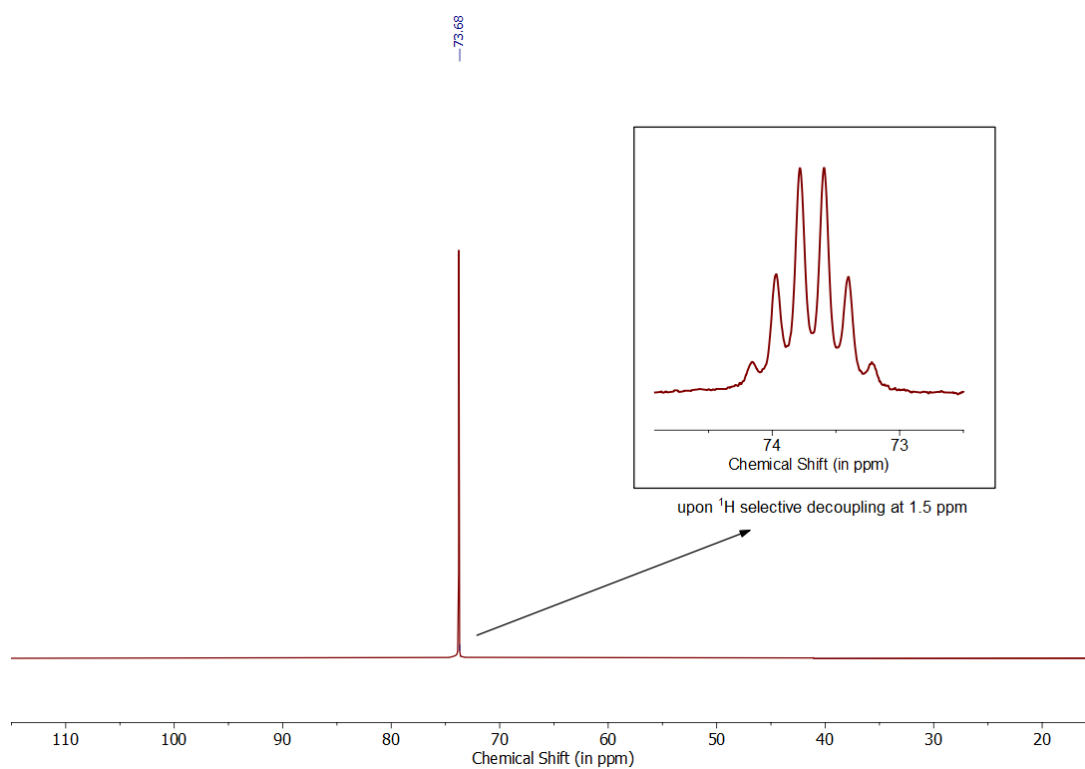
**II.1.4.  $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum (101 MHz, THF- $\text{d}_8$ )**



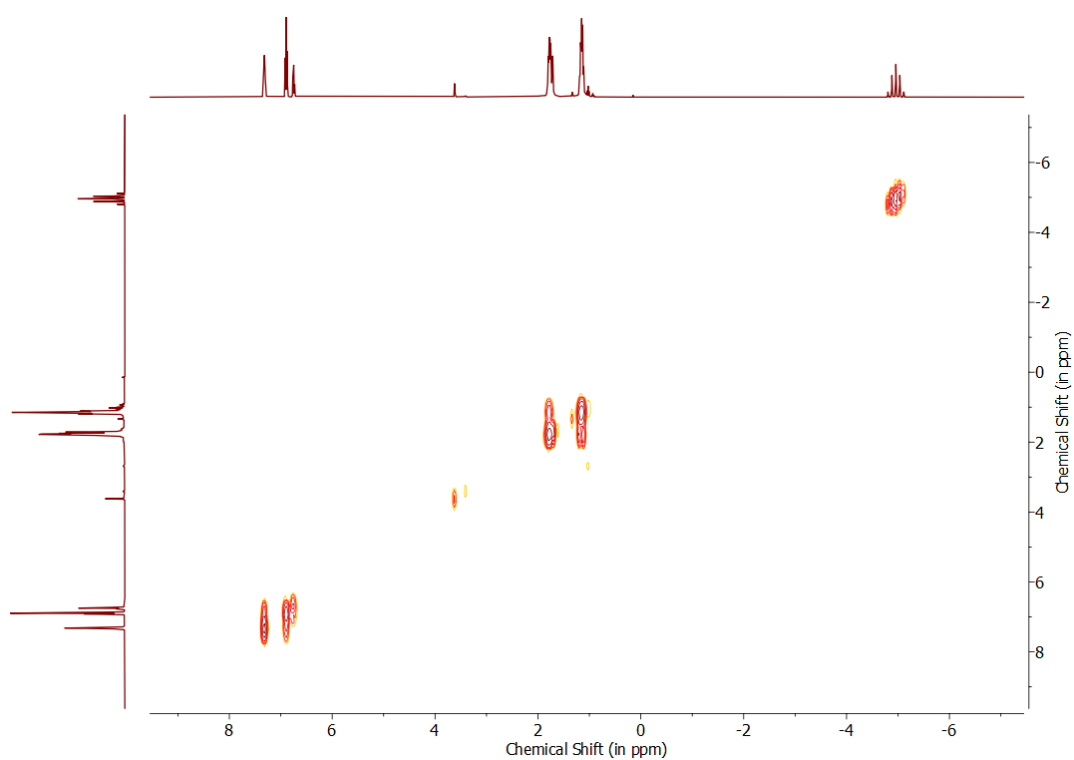
**II.1.5.  $^{13}\text{C}$ -DEPT NMR spectrum (101 MHz, THF- $d_8$ )**



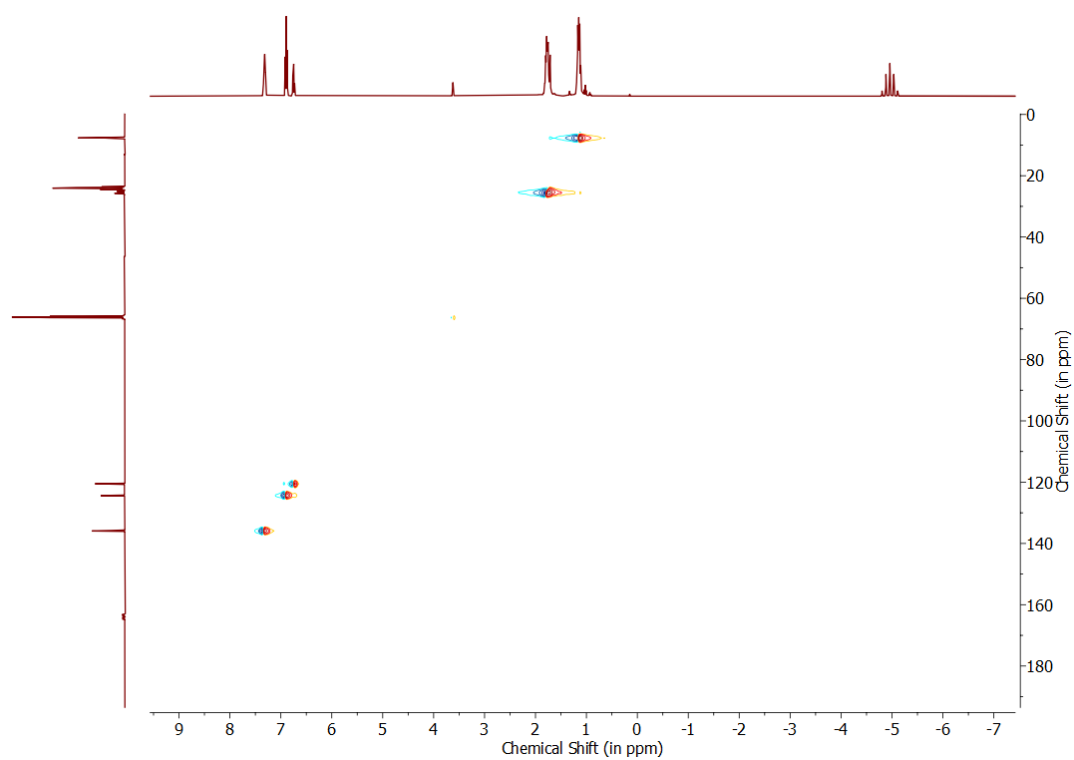
**II.1.6.  $^{31}\text{P}\{^1\text{H}\}$ -NMR spectrum (162 MHz, THF- $d_8$ )**



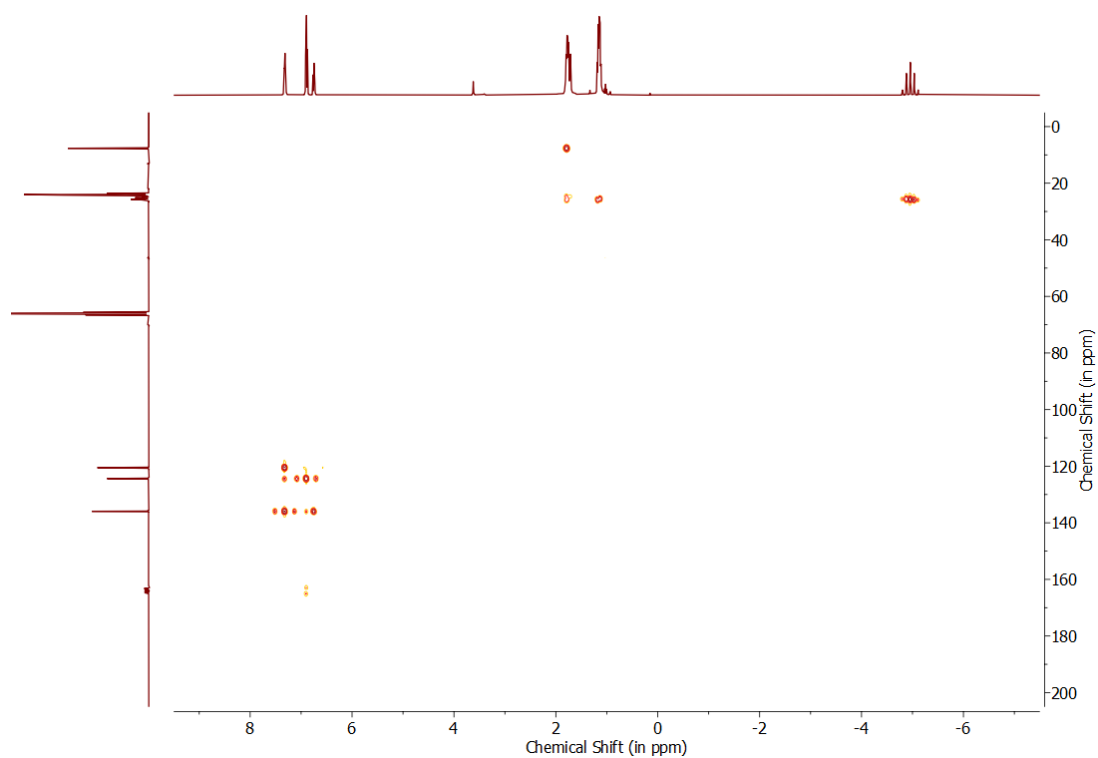
**II.1.7.  $^1\text{H}$ - $^1\text{H}$  COSY NMR spectrum**



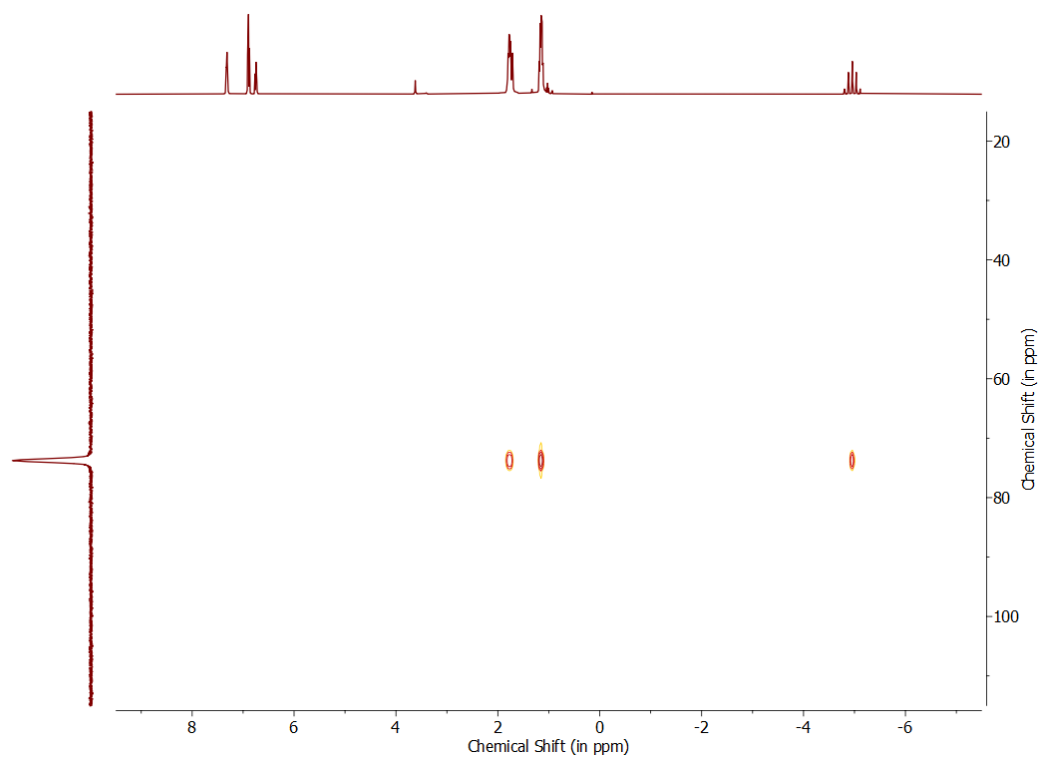
**II.1.8.  $^{13}\text{C}$ - $^1\text{H}$  HSQC NMR spectrum**



**II.1.9.  $^{13}\text{C}$ - $^1\text{H}$  HMBC NMR spectrum**

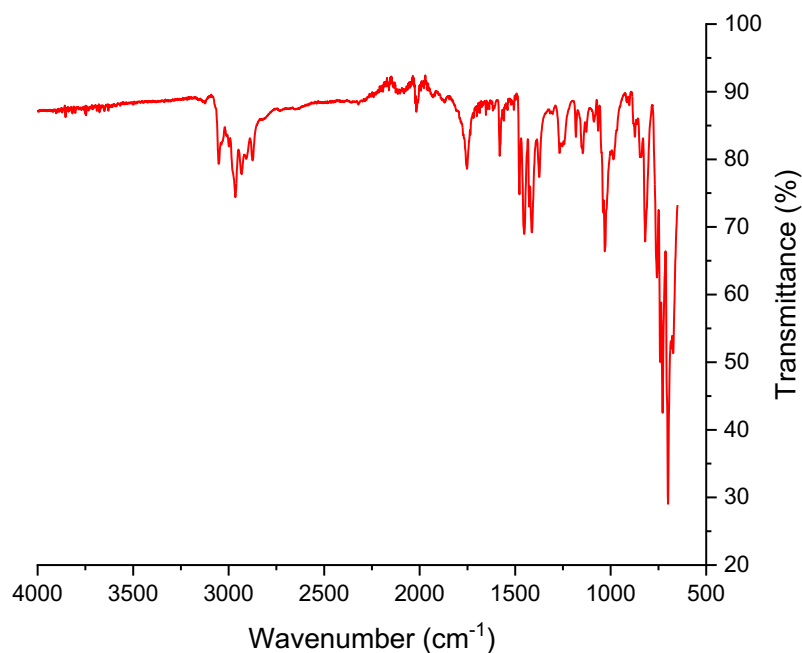


**II.1.10.  $^{31}\text{P}$ - $^1\text{H}$  HMQC NMR spectrum**





### II.1.11. FT-IR spectrum (ATR)



## II.2. [Mo(depe)<sub>2</sub>(HCOO)H<sub>2</sub>][B(C<sub>6</sub>H<sub>5</sub>)<sub>4</sub>] (3.BPh<sub>4</sub>)

### II.2.1. Experimental procedure

In an Wilmad pressure NMR tube, [Mo(depe)<sub>2</sub>H<sub>5</sub>][BPh<sub>4</sub>] (20.0 mg, 24.0 μmol, 1.00 equiv.) was dissolved in THF (1 mL). The tube was placed at 0°C, evacuated and finally pressurized with carbon dioxide (3 bar). The reaction mixture was left at room temperature for 24h. After pressure was carefully released, the solution was slowly poured in cold pentane (ca. 5 mL), triggering immediate precipitation of a pink solid. The solution was decanted off and the solid washed twice with cold pentane (2 x 5 mL). The solid was then dried under an argon flux at room temperature to yield the desired compound (15.5 mg, 74%). Single crystals suitable for X-ray diffraction were obtained by storing at -40°C a concentrated THF solution of the complex **5.BPh<sub>4</sub>**, layered by pentane over a few days.

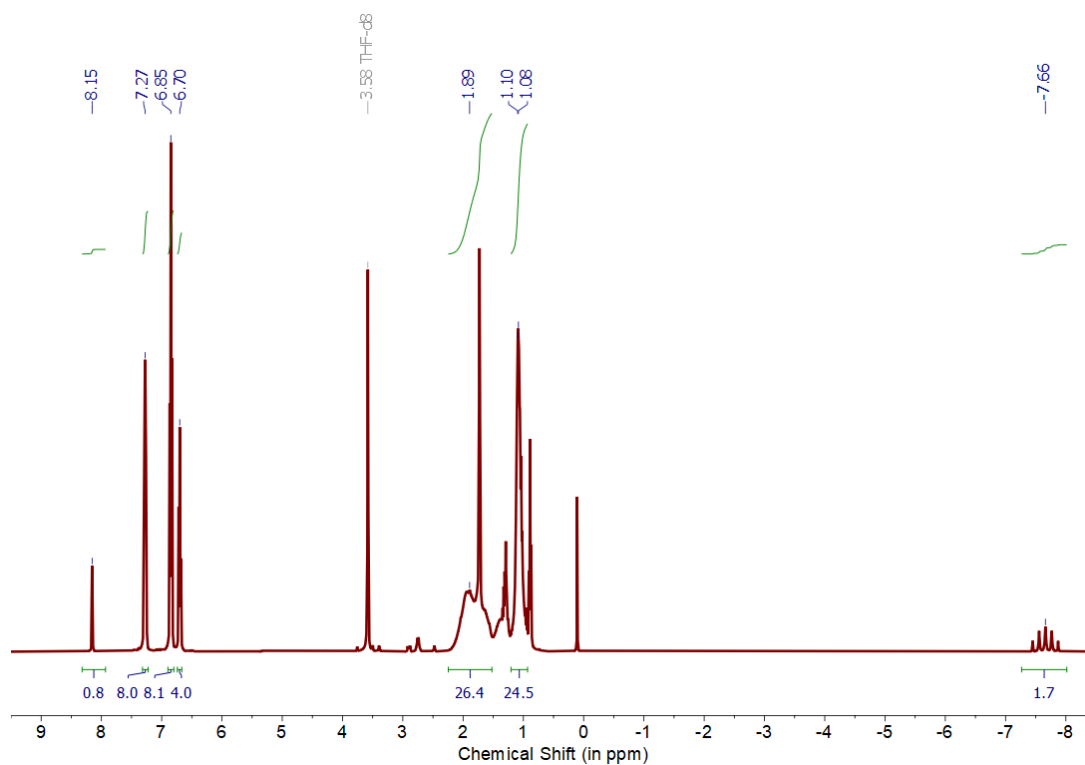
**<sup>1</sup>H-NMR** (400 MHz, THF-*d*<sub>8</sub>) δ 8.15 (s, 1H, HCOO), 7.27 (br. m, 8H, *meta*-C<sub>6</sub>H<sub>5</sub>), 6.85 (t, <sup>3</sup>J<sub>(H-H)</sub> = 7.4 Hz, 8H, *ortho*-C<sub>6</sub>H<sub>5</sub>), 6.69 (t, <sup>3</sup>J<sub>(H-H)</sub> = 7.2 Hz, 4H, *para*-C<sub>6</sub>H<sub>5</sub>), 2.21-1.52 (m, 24H, CH<sub>2</sub> and CH<sub>3</sub> from depe), 1.22-0.94 (m, 24H, CH<sub>2</sub> and CH<sub>3</sub> from depe), -7.66 (pseudo-quintet, <sup>2</sup>J<sub>(P-H),app</sub> = 42.6 Hz, 2H).

**<sup>11</sup>B-NMR** (128 MHz, THF-*d*<sub>8</sub>) δ -6.87

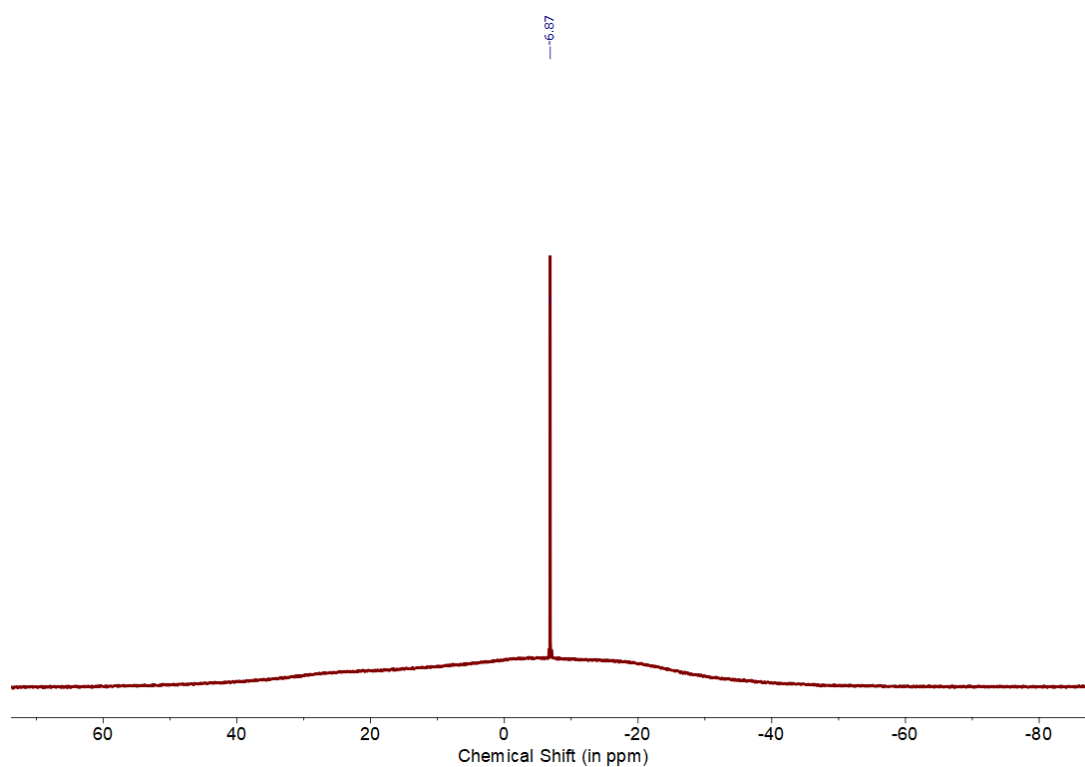
**<sup>13</sup>C{<sup>1</sup>H}-NMR** (101 MHz, THF-*d*<sub>8</sub>) δ 176.9 (m, HCOO), 163.9 (q, <sup>2</sup>J<sub>(B-C)</sub> = 50.3 Hz, *ipso*-C<sub>6</sub>H<sub>5</sub>), 135.9 (br. s, *meta*-C<sub>6</sub>H<sub>5</sub>), 124.3 (m, *ortho*-C<sub>6</sub>H<sub>5</sub>), 120.4 (s, *para*-C<sub>6</sub>H<sub>5</sub>), 22.85, 20.43, 17.11, 13.41, 7.48, 7.06.

**<sup>31</sup>P{<sup>1</sup>H}-NMR** (162 MHz, THF-*d*<sub>8</sub>) δ 78.3 (s, 2P), 44.0 (s, 2P). **<sup>31</sup>P{<sup>1</sup>H}<sub>sel</sub>-NMR** (162 MHz, THF-*d*<sub>8</sub>) δ 78.3 (t, <sup>2</sup>J<sub>(P-H)</sub> = 42.6 Hz), 48.3 (t, <sup>2</sup>J<sub>(P-H)</sub> = 36.7 Hz).

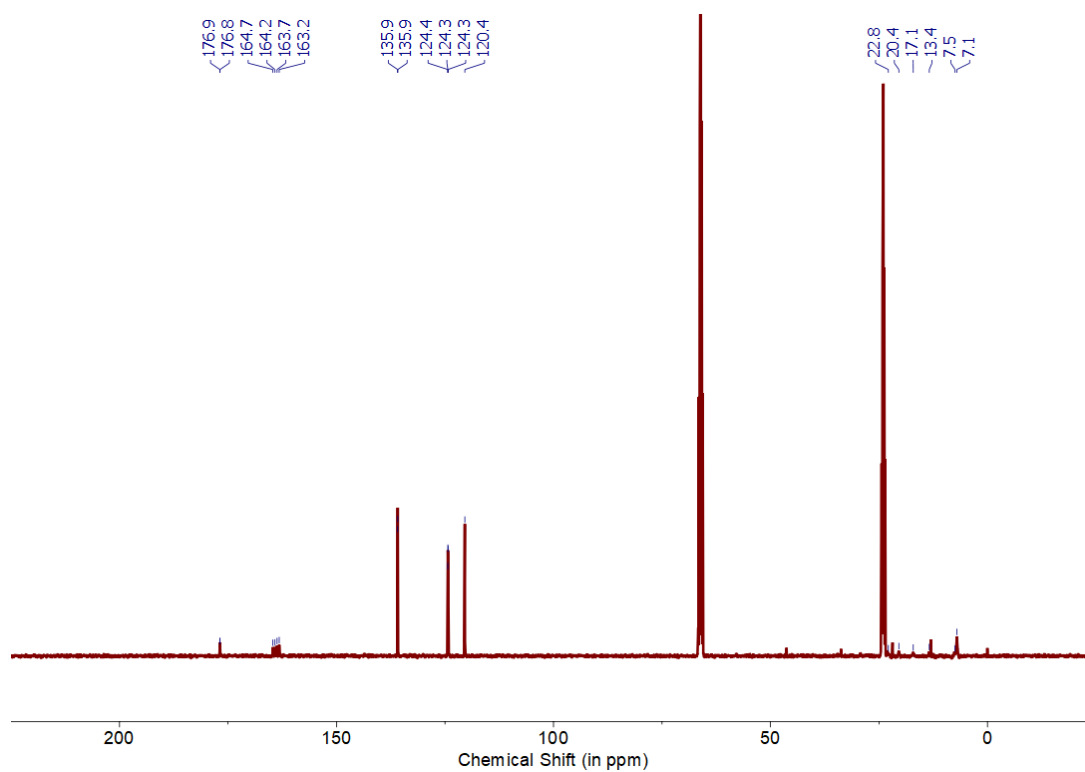
**II.2.2.  $^1\text{H}$ -NMR spectrum (400 MHz, THF-d<sub>8</sub>)**



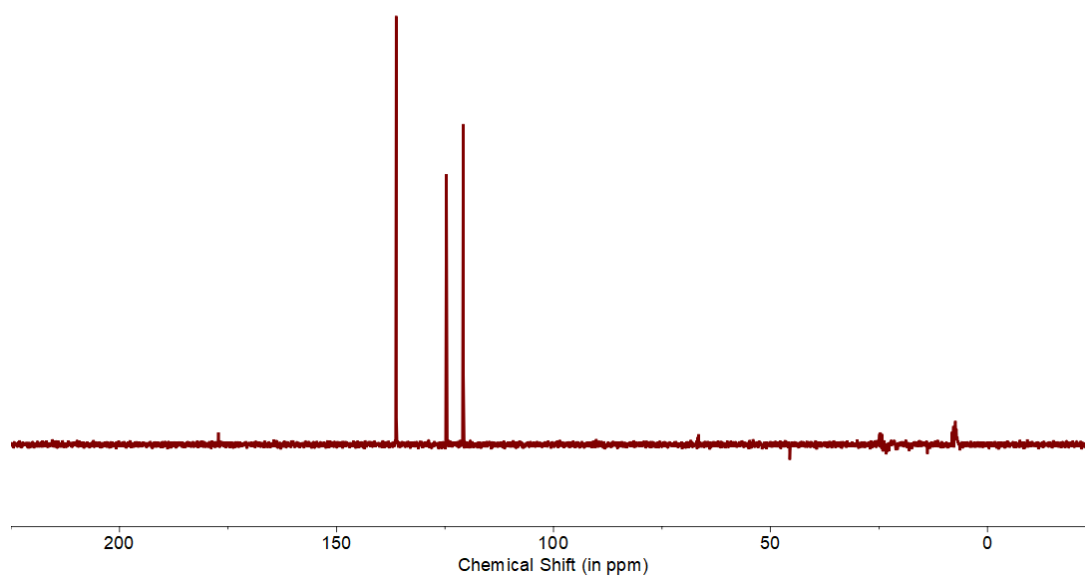
**II.2.3.  $^{11}\text{B}\{^1\text{H}\}$ -NMR spectrum (128 MHz, THF-d<sub>8</sub>)**



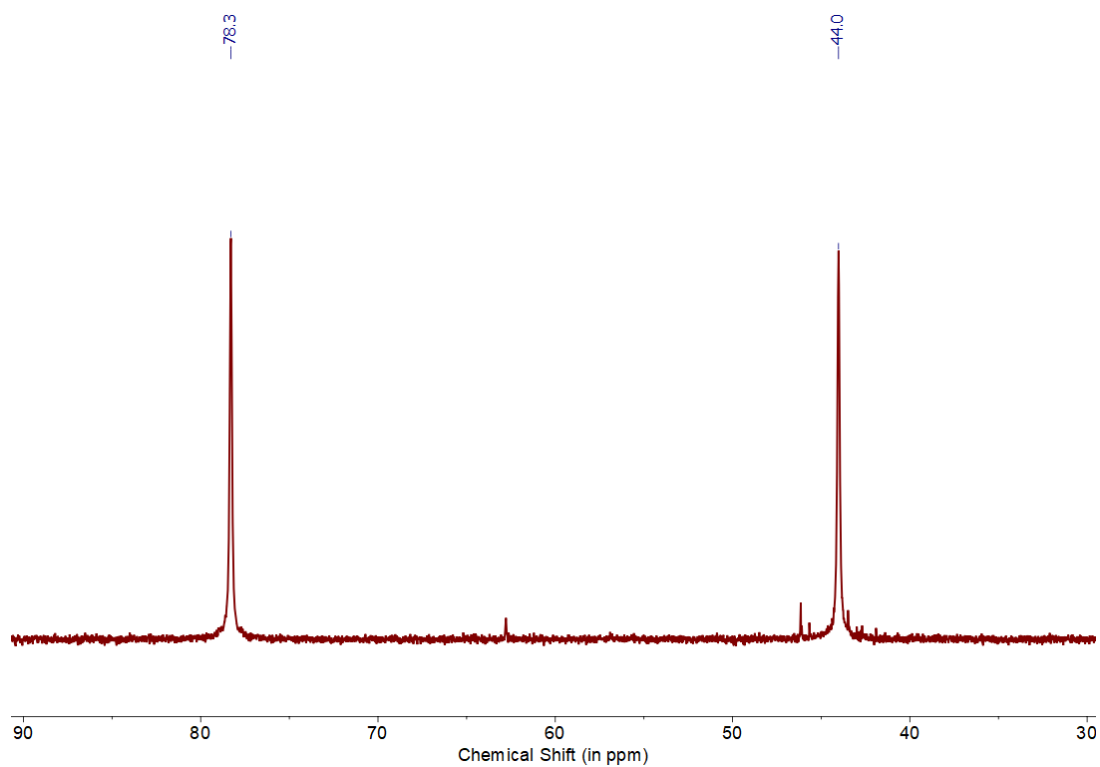
**II.2.4.  $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum (101 MHz, THF-d8)**



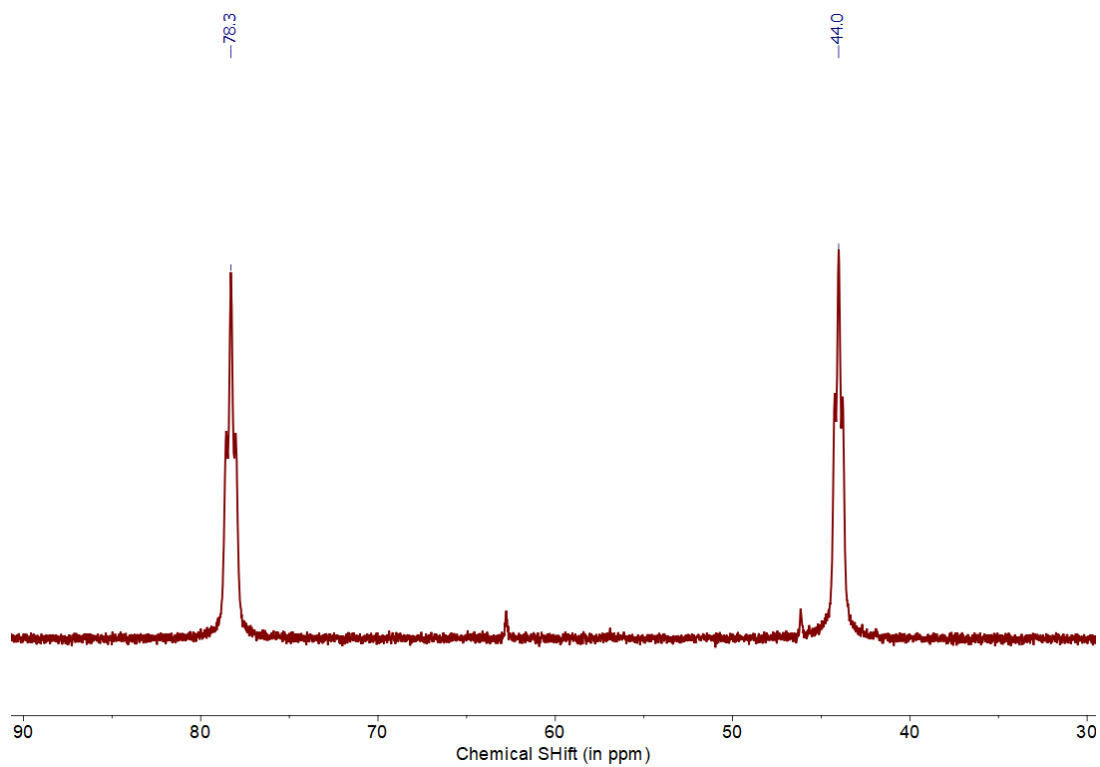
**II.2.5.  $^{13}\text{C}$ -DEPT NMR spectrum (101 MHz, THF-d8)**



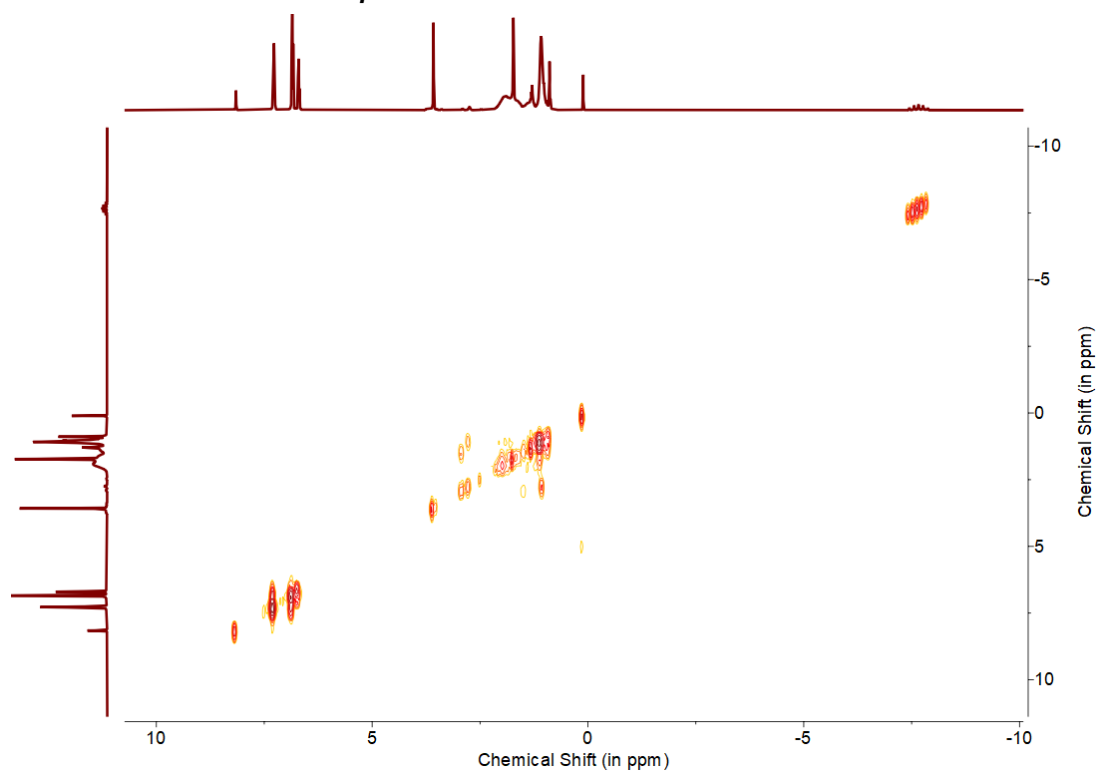
**II.2.6.  $^{31}\text{P}\{\text{^1H}\}$ -NMR spectrum (162 MHz, THF-d8)**



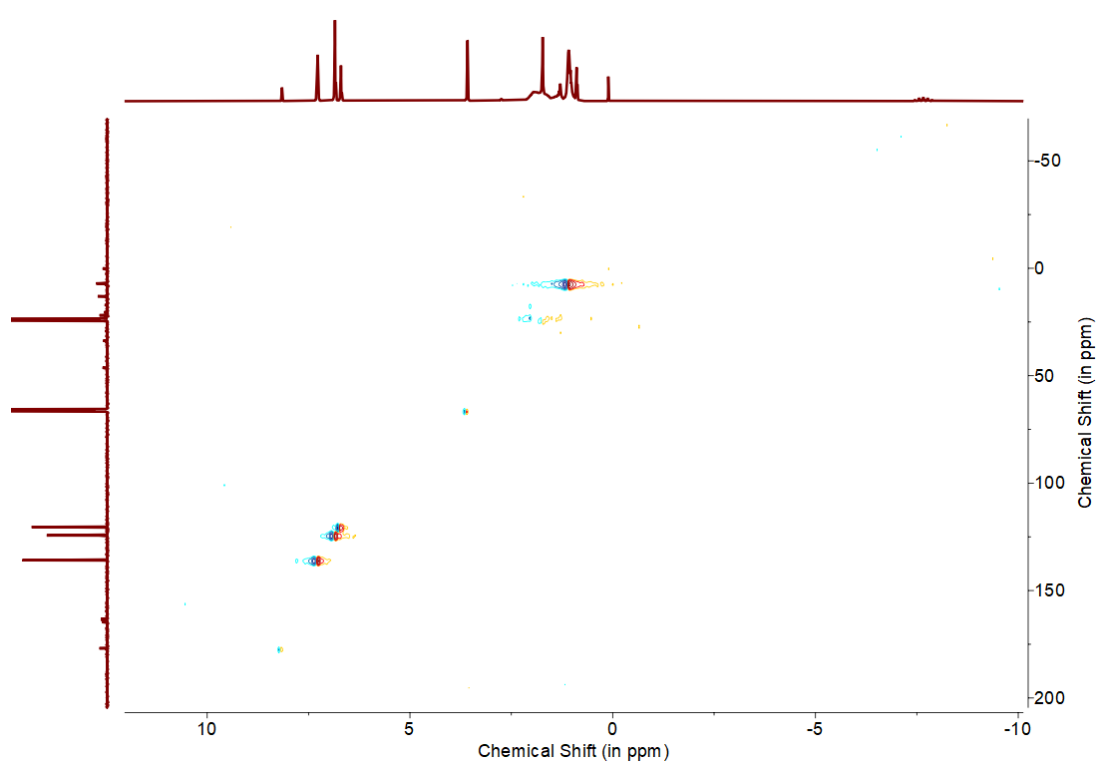
**II.2.7.  $^{31}\text{P}\{\text{^1H}\}$  sel ( $\delta = 1.5$  ppm)-NMR spectrum (162 MHz, THF-d8)**



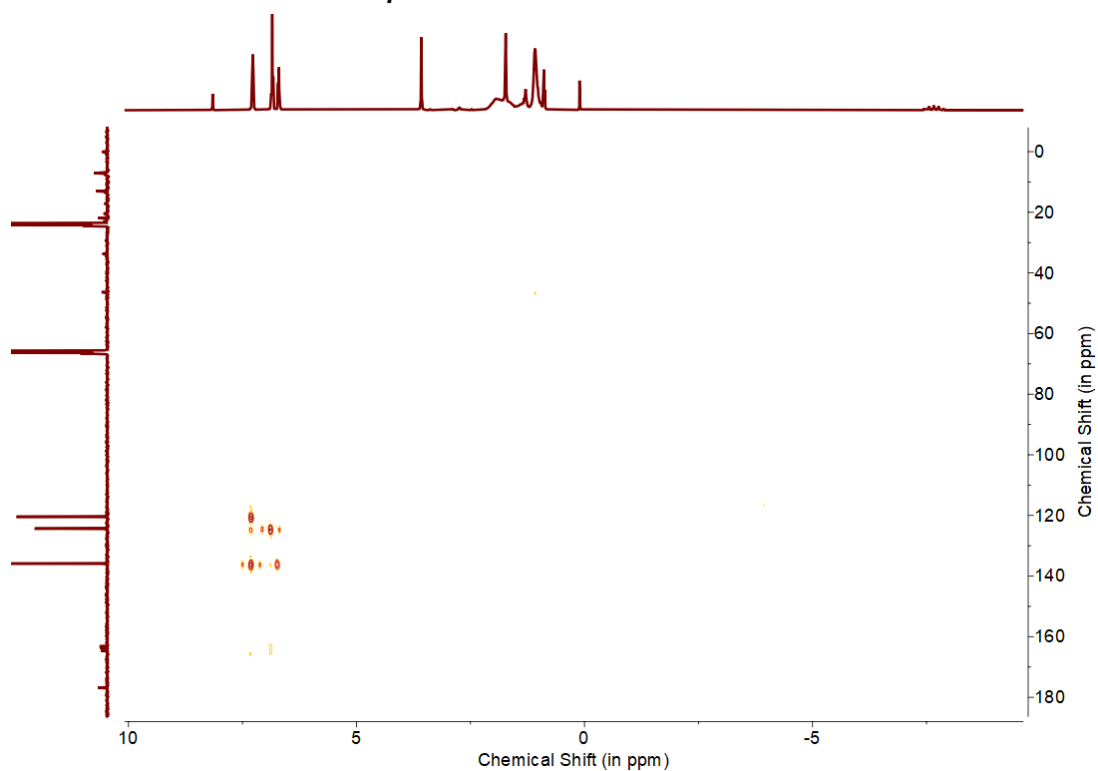
**II.2.8.  $^1\text{H}$ - $^1\text{H}$  COSY NMR spectrum**



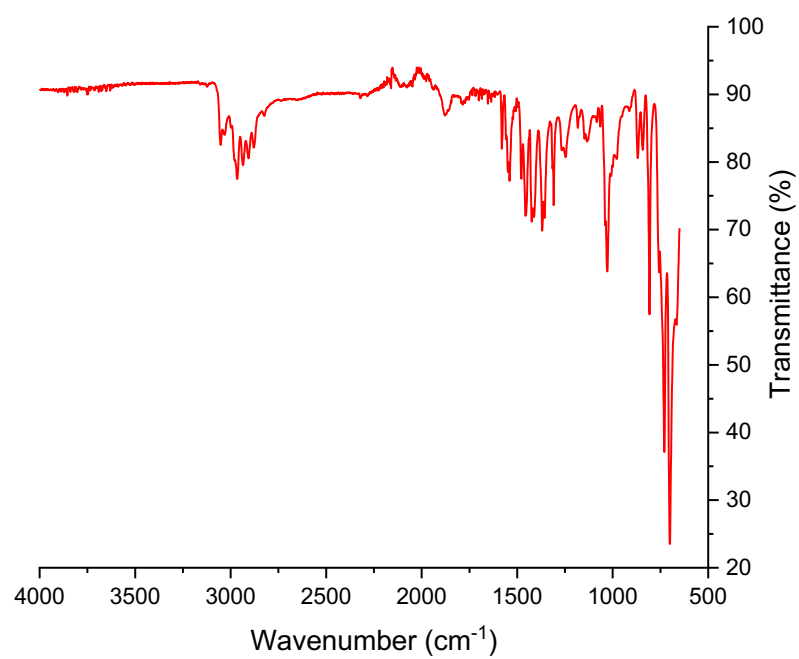
**II.2.9.  $^{13}\text{C}$ - $^1\text{H}$  HSQC NMR spectrum**



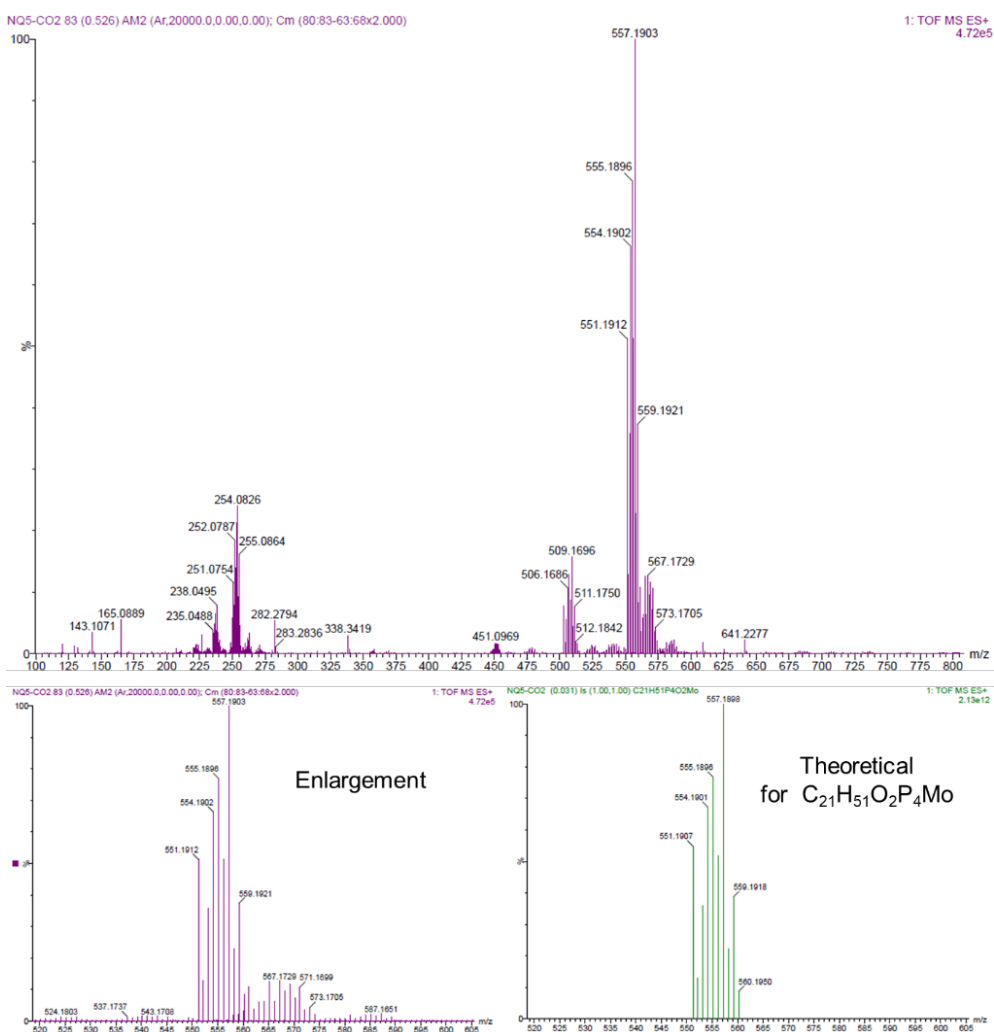
**II.2.10.  $^{13}\text{C}$ - $^1\text{H}$  HMBC NMR spectrum**



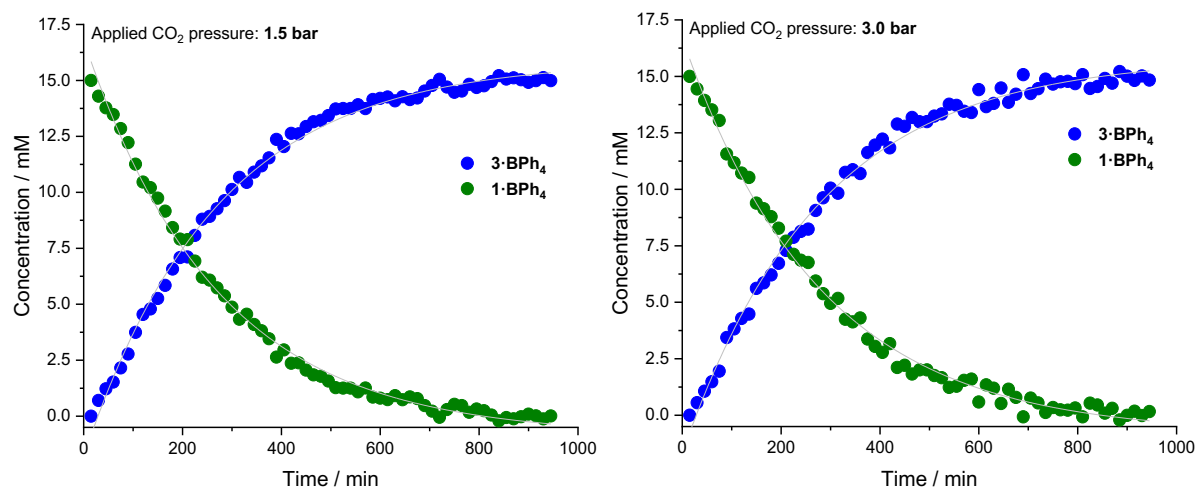
**II.2.11. FT-IR spectrum (ATR)**



## II.2.12. ESI-HRMS spectrum



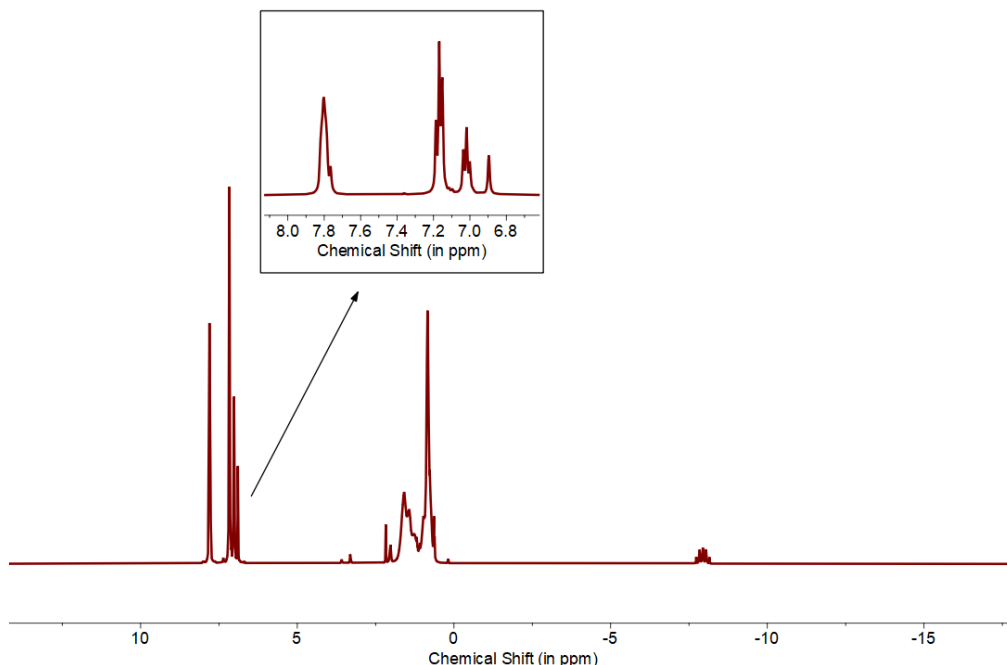
## II.2.13. CO<sub>2</sub> pressure effect on reaction kinetics



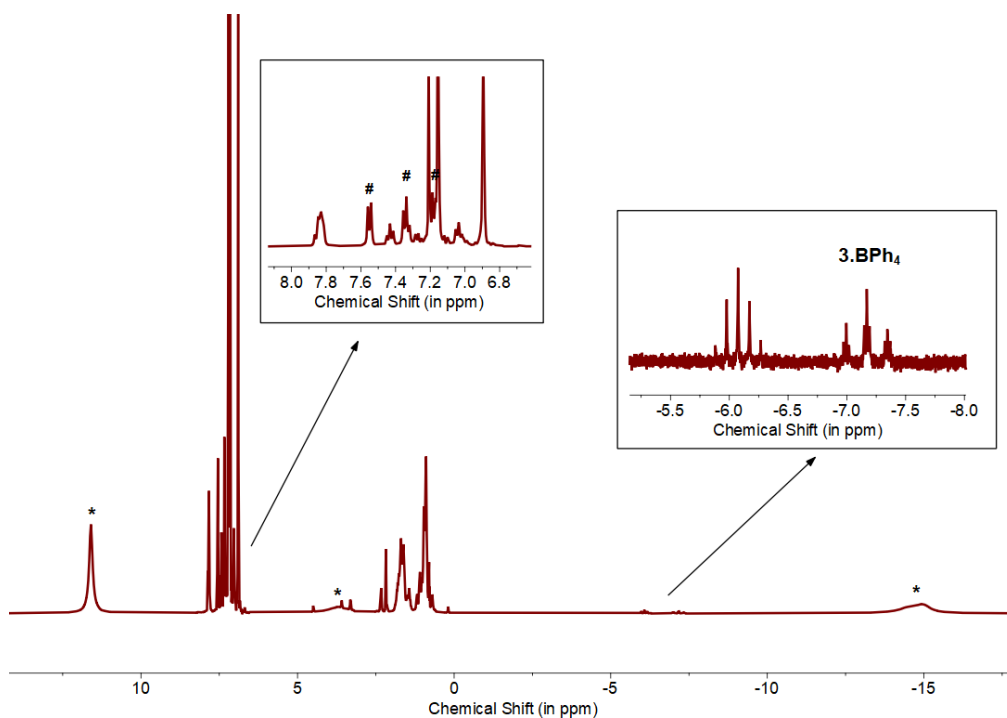
#### II.2.14. Thermal Stability in $C_6D_4Cl_2$ (100°C, $CO_2$ ) – case of the $BPh_4^-$ counter-anion

Further reactivity with  $CO_2$  was investigated at 100°C, in two different solvents (thf-d8 or  $C_6D_4Cl_2$ ). In a typical experiment 30  $\mu$ mol of **3.BPh<sub>4</sub>** was dissolved in the desired solvent (ca. 700  $\mu$ L). The tube was placed at 0°C, evacuated, pressurized with carbon dioxide (1 bar) and finally placed at 100°C. Evolution of the reaction medium was monitored by NMR spectroscopies.

(A)  $^1H$ -NMR spectrum of a solution of **3.BPh<sub>4</sub>** in  $C_6D_4Cl_2$ , before heating at 100°C for 24h.

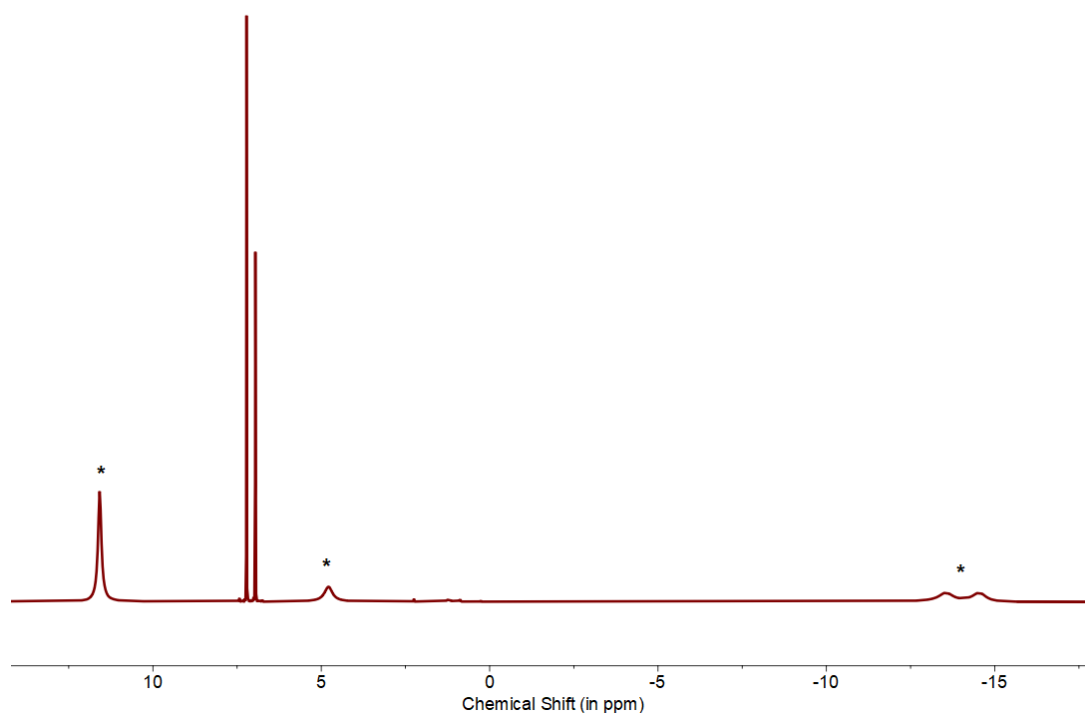


(B)  $^1H$ -NMR spectrum of a solution of **3.BPh<sub>4</sub>** in  $C_6D_4Cl_2$ , after heating at 100°C for 24h. \* indicates the location of signals associated with paramagnetic species (see authentic sample  $^1H$ -NMR spectrum on II.2.12.C). # indicates the location of paramagnetism-affected signals associated with biphenyl (see authentic sample  $^1H$ -NMR spectrum on II.2.12.D). Insert boxes display the formation of the characteristic hydride signal of **4.BPh<sub>4</sub>** at -7.2 ppm and the formation of new aromatic signals assigned to oxidative degradation of  $BPh_4^-$  anion.

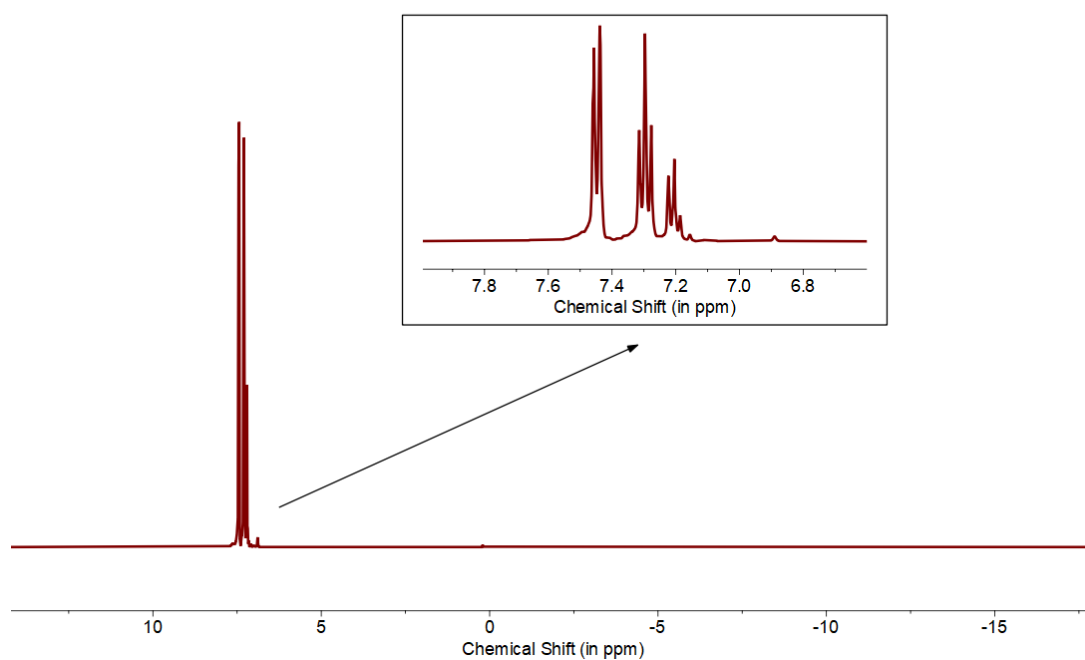




- (C)  $^1\text{H}$ -NMR spectrum of a solution of  $[\text{Mo}(\text{depe})_2\text{Cl}_2]$  in  $\text{C}_6\text{D}_4\text{Cl}_2$ . This complex is paramagnetic and the large, poorly-resolved signals displayed properly match those of the paramagnetic products formed in the thermal decomposition of **3.BPh<sub>4</sub>** under  $\text{CO}_2$ .

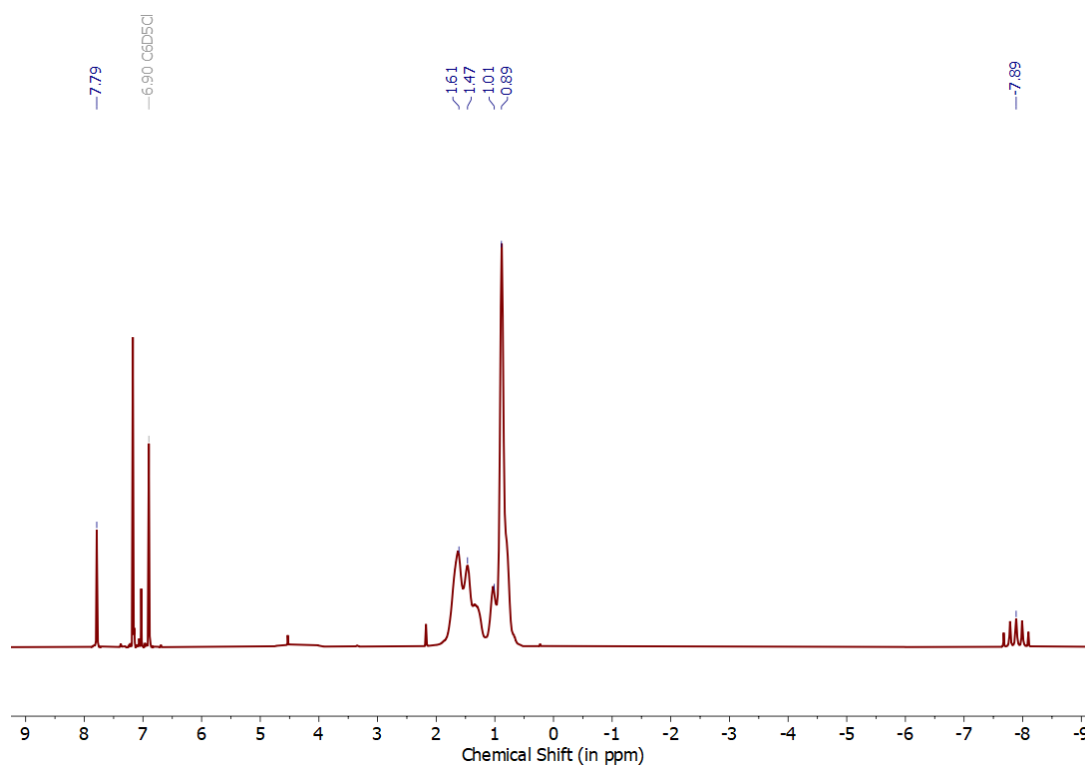


- (D)  $^1\text{H}$ -NMR spectrum of a solution of  $[\text{Mo}(\text{depe})_2\text{Cl}_2]$  in  $\text{C}_6\text{D}_4\text{Cl}_2$ . This complex is paramagnetic and the large, poorly-resolved signals displayed properly match those of the paramagnetic products formed in the thermal decomposition of **3.BPh<sub>4</sub>** under  $\text{CO}_2$ .

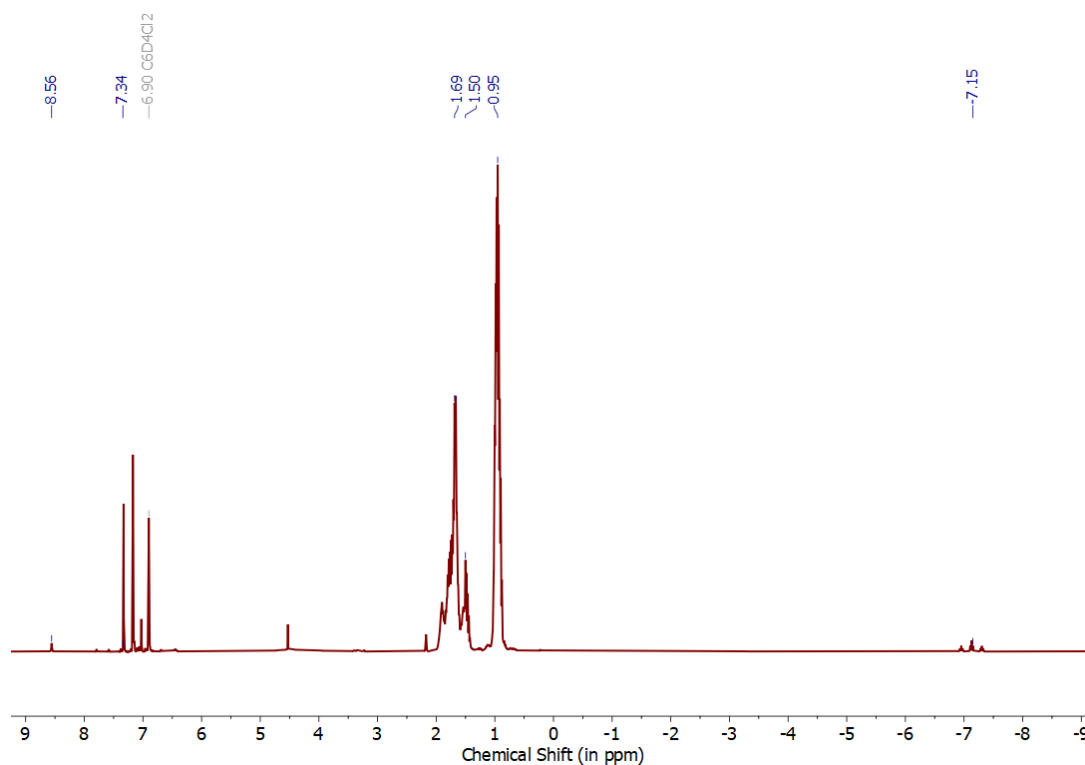


## II.2.15. Thermal Stability in $C_6D_4Cl_2$ (100°C, $CO_2$ ) – case of the $HB(C_6F_5)_3^-$ counter-anion

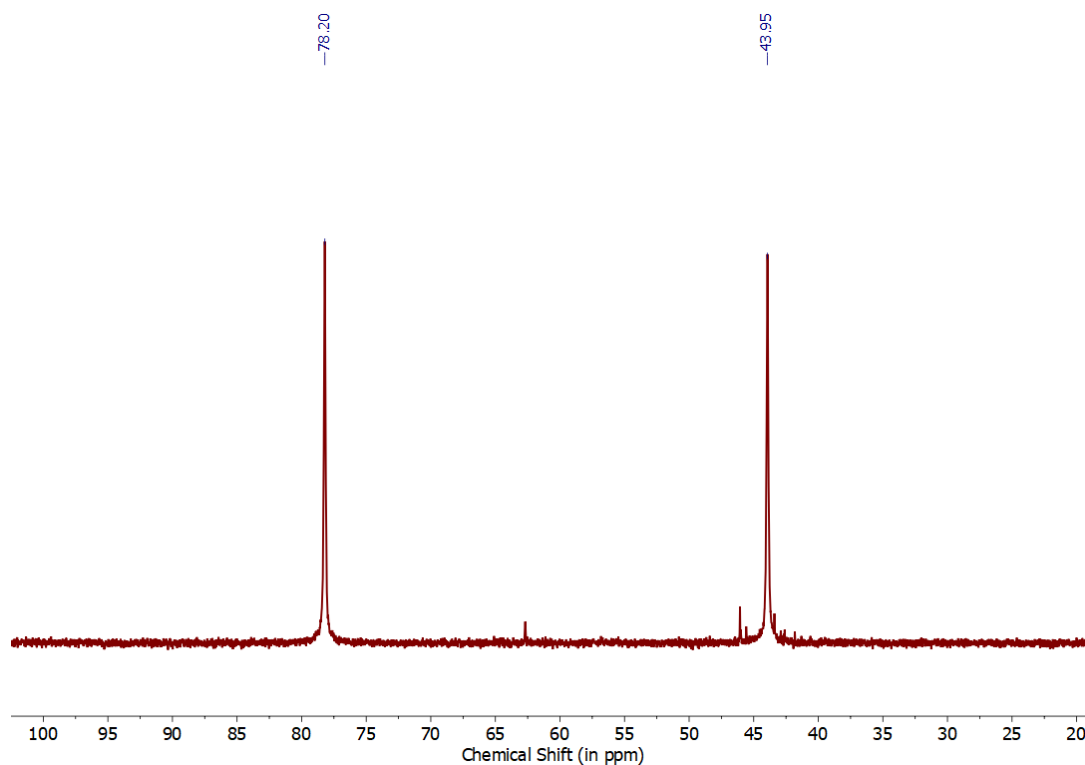
(A)  $^1H$ -NMR spectrum of a solution of  $3.HB(C_6F_5)_3$  in  $C_6D_4Cl_2$ , before heating at 100°C for 36h.



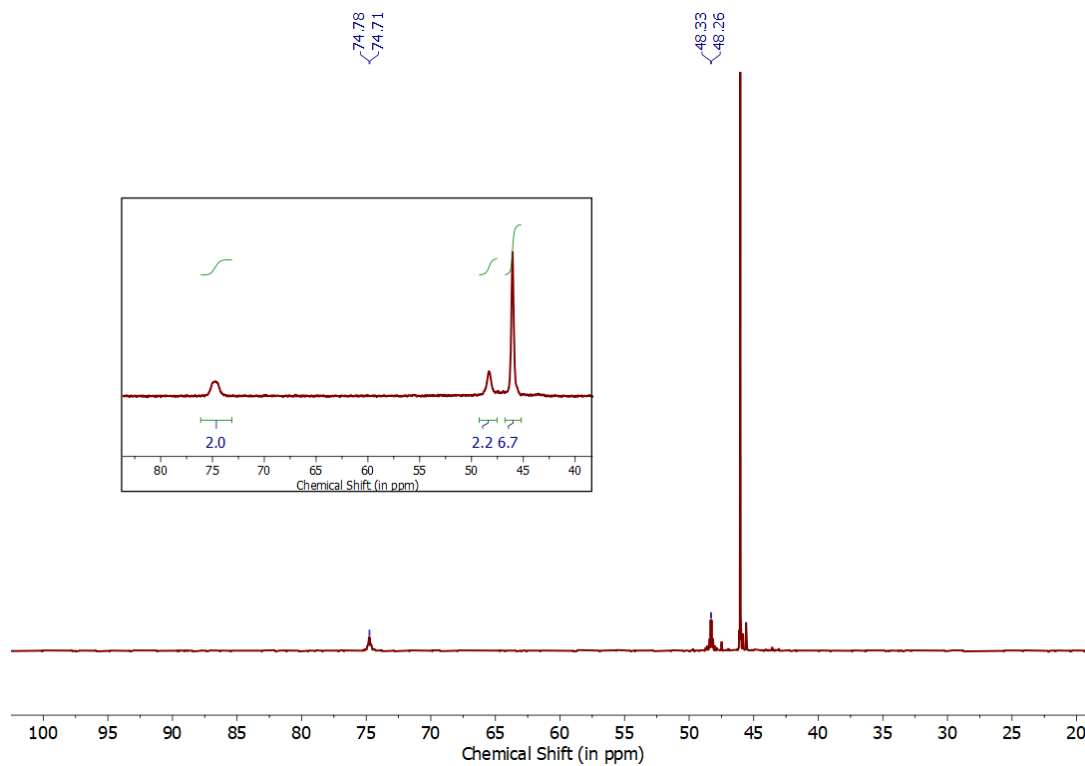
(B)  $^1H$ -NMR spectrum of a solution of  $3.HB(C_6F_5)_3$  in  $C_6D_4Cl_2$ , after heating at 100°C for 36h. Of particular interest are the free formate anion signal at 8.56 ppm, the new formate-containing species at 7.34 ppm and the characteristic hydride signal of  $4.HB(C_6F_5)_3$  at -7.2 ppm



(C)  $^{31}\text{P}$ -NMR spectrum of a solution of **3.HB(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>** in  $\text{C}_6\text{D}_4\text{Cl}_2$ , before heating at 100°C for 36h.

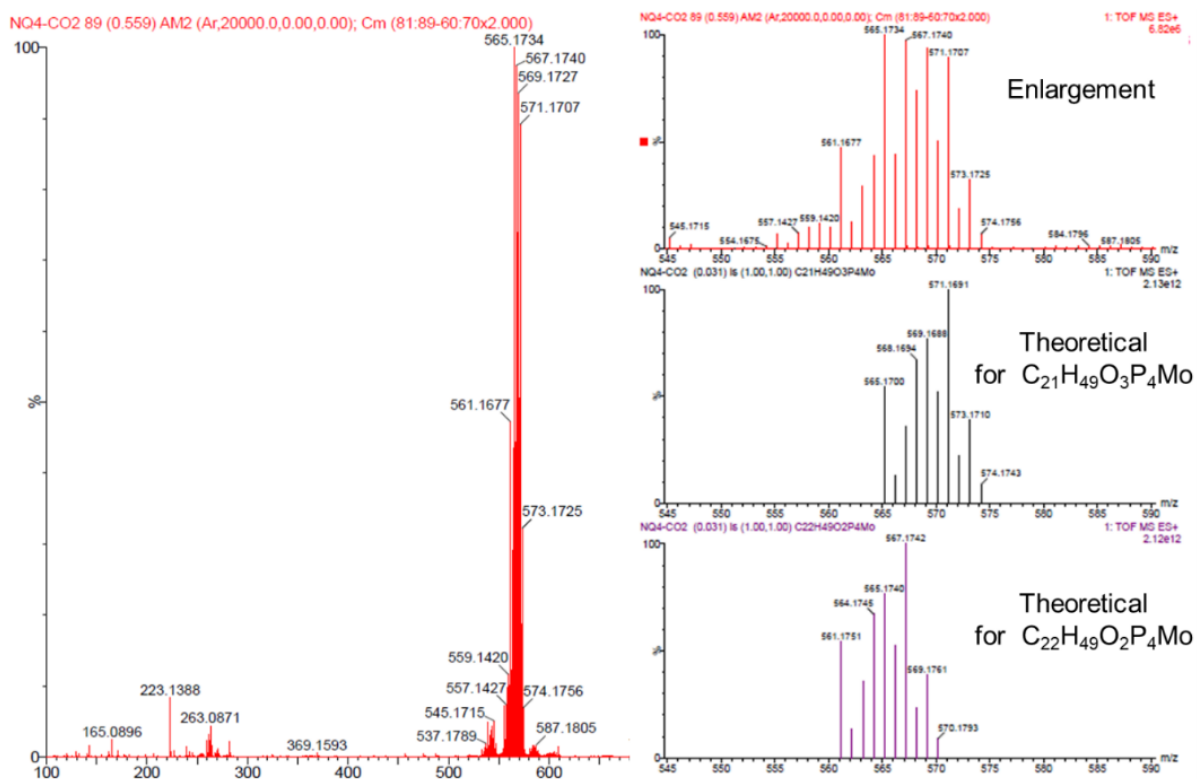


(D)  $^{31}\text{P}\{^1\text{H}\}$ -NMR spectrum of a solution of **3.HB(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>** in  $\text{C}_6\text{D}_4\text{Cl}_2$ , after heating at 100°C for 36h. Of particular interest are the diagnostic signal of **4.HB(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>** at 74.8 and 48.3 ppm. To better assess the relative proportion of **4.HB(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>** and **5.HB(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>** by the end of the reaction, undecoupled  $^{31}\text{P}$  integrations were used (insert).



## II.2.16. HRMS analysis of the thermal decomposition of 3.HB(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>

ESI-HRMS analysis of the thermal decomposition of 3.HB(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>: experimental spectrum and enlargement of the main feature (red traces), theoretical isotopic pattern calculated for 5<sup>+</sup>, [MoO(depe)<sub>2</sub>(HCOO)]<sup>+</sup> (black traces) and theoretical isotopic pattern calculated for 4<sup>+</sup>, [Mo(CO)<sub>2</sub>(depe)<sub>2</sub>H]<sup>+</sup> (purple traces)



## II.3. [Mo(depe)<sub>2</sub>(CO)<sub>2</sub>H][B(C<sub>6</sub>H<sub>5</sub>)<sub>4</sub>] (4.BPh<sub>4</sub>)

### II.3.1. Experimental procedure

In an Wilmad pressure NMR tube, [Mo(depe)<sub>2</sub>H<sub>5</sub>][B(C<sub>6</sub>H<sub>5</sub>)<sub>4</sub>] (20.0 mg, 24.0 μmol, 1.00 equiv.) was dissolved in THF (1 mL). The tube was placed at 0°C, evacuated and finally pressurized with carbon monoxide (3 bar). The reaction mixture was left at room temperature for 24h. After pressure was carefully released, the pale-yellow solution was poured in cold pentane (ca. 5 mL), triggering immediate precipitation of an off-white solid. The solution was decanted off and the solid washed twice with cold pentane (2 x 5 mL). The solid was then dried under high vacuum at room temperature to yield the desired compound (15.9 mg, 75%). Single crystals suitable for X-ray diffraction were obtained by storing at -40°C a concentrated THF solution of the complex layered by pentane over a few days.

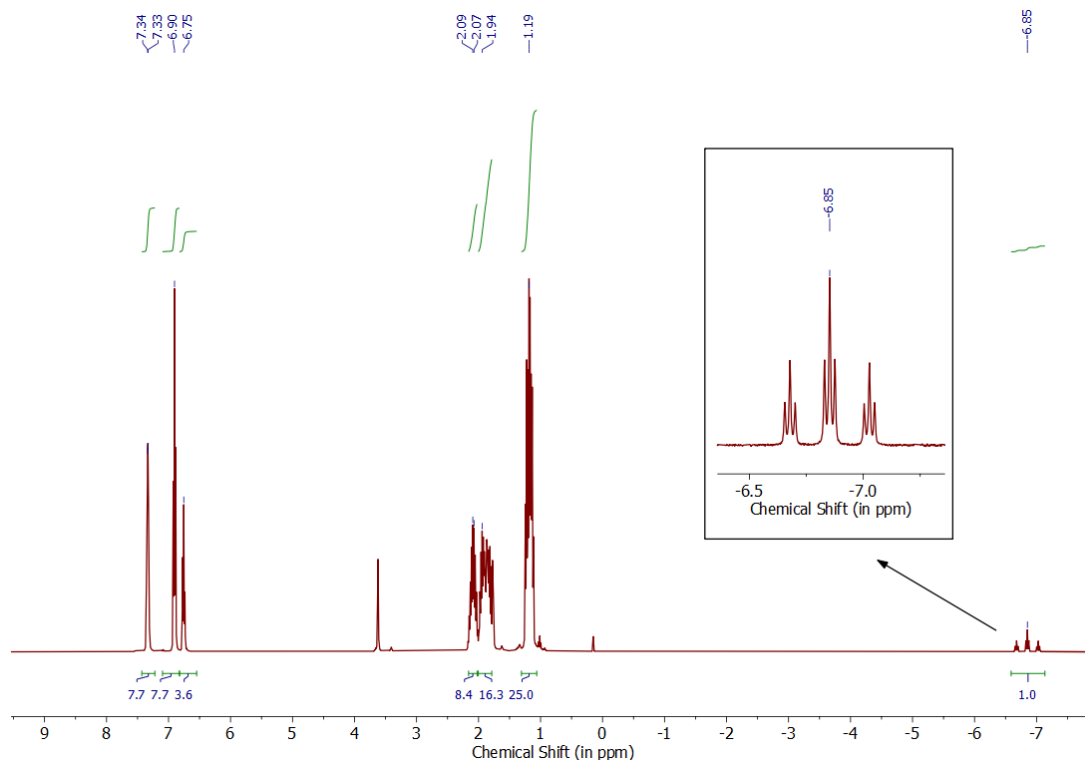
**<sup>1</sup>H-NMR** (400 MHz, THF-d<sub>8</sub>) δ 7.33 (br. m, 8H, *meta*-C<sub>6</sub>H<sub>5</sub>), 6.90 (t, <sup>3</sup>J<sub>(H-H)</sub> = 7.3 Hz, 8H, *ortho*-C<sub>6</sub>H<sub>5</sub>), 6.75 (t, <sup>3</sup>J<sub>(H-H)</sub> = 7.3 Hz, 4H, *para*-C<sub>6</sub>H<sub>5</sub>), 2.10 (m, 8H, -CH<sub>2</sub>-CH<sub>2</sub>-), 2.02 – 1.79 (m, 16H, -CH<sub>2</sub>-CH<sub>3</sub>), 1.18 (m, 24H, -CH<sub>2</sub>-CH<sub>3</sub>), -6.85 (tt, <sup>2</sup>J<sub>(P-H)</sub> = 69.8, 9.1 Hz, 1H, Mo-H).

**<sup>11</sup>B-NMR** (128 MHz, THF-d<sub>8</sub>) δ -6.8.

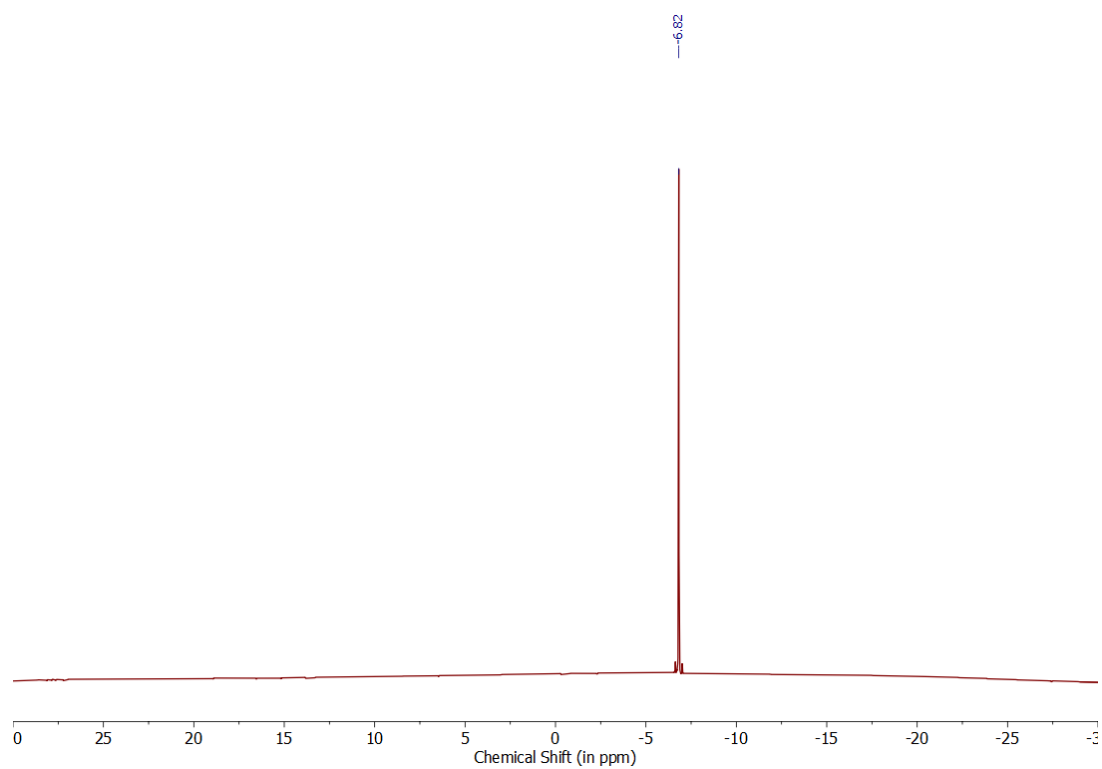
**<sup>13</sup>C{<sup>1</sup>H}-NMR** (101 MHz, THF-d<sub>8</sub>) δ 209.8 (s; Mo-CO), 163.9 (q, <sup>2</sup>J<sub>(B-C)</sub> = 50.3 Hz, *ipso*-C<sub>6</sub>H<sub>5</sub>), 135.9 (br. s, *meta*-C<sub>6</sub>H<sub>5</sub>), 124.4 (m, *ortho*-C<sub>6</sub>H<sub>5</sub>), 120.5 (s, *para*-C<sub>6</sub>H<sub>5</sub>), 23.3 (m, -CH<sub>2</sub>-), 21.6 (m, -CH<sub>2</sub>-), 7.5 (br. s, -CH<sub>3</sub>), 7.3 (br. s, -CH<sub>3</sub>).

**<sup>31</sup>P{<sup>1</sup>H}-NMR** (162 MHz, THF-d<sub>8</sub>) δ 74.9 (m), 48.3 (m). **<sup>31</sup>P{<sup>1</sup>H}<sub>sel</sub>-NMR** (162 MHz, THF-d<sub>8</sub>) δ 74.9 (d, <sup>2</sup>J<sub>(P-H)</sub> = 68.9 Hz), 48.3 (br. s).

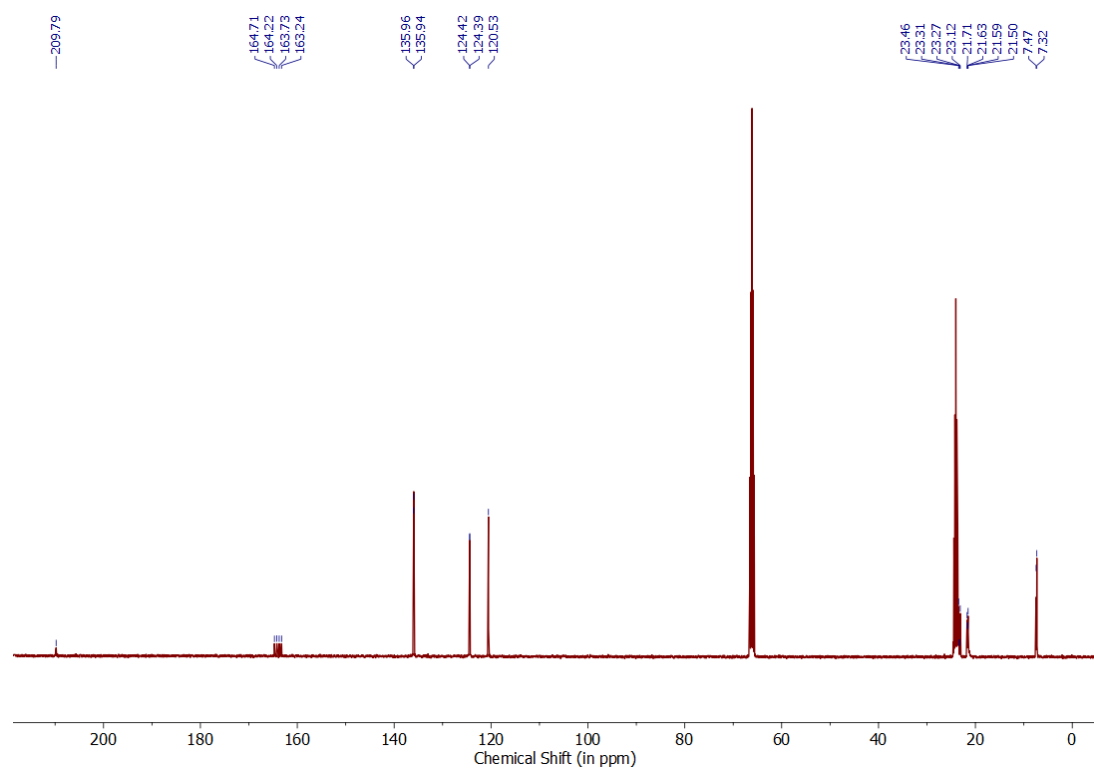
### II.3.2. <sup>1</sup>H-NMR spectrum (400 MHz, THF-d<sub>8</sub>)



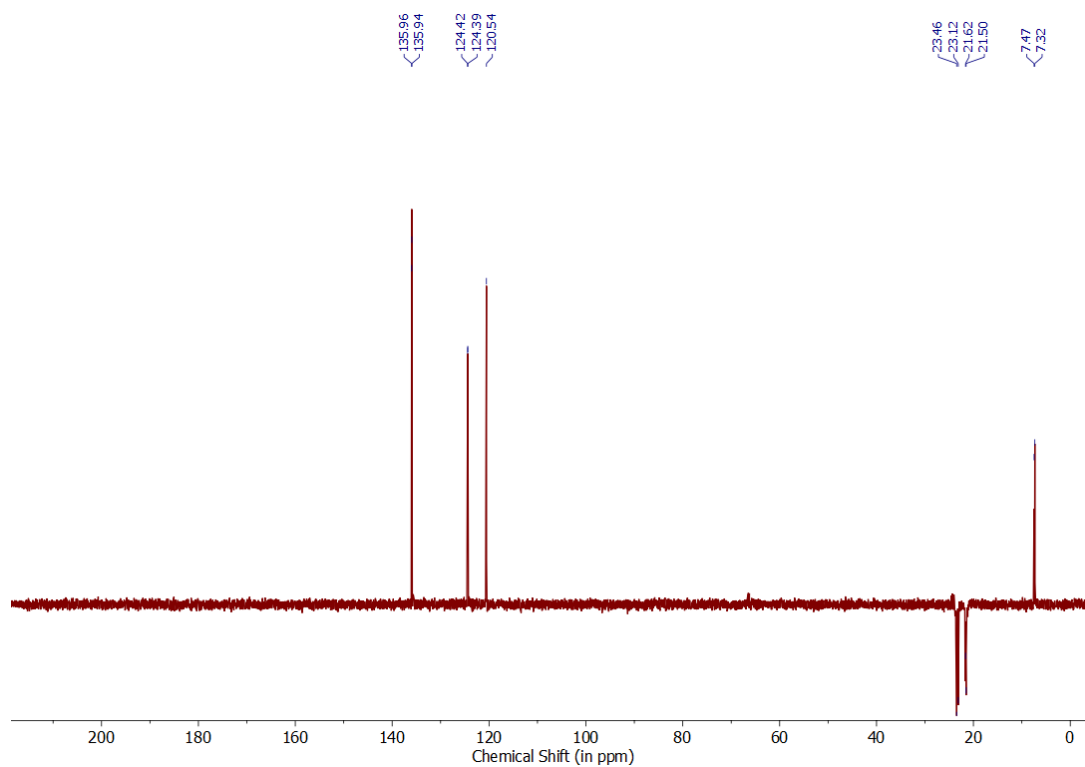
**II.3.3.  $^1\text{B}\{^1\text{H}\}$ -NMR spectrum (128 MHz,  $\text{THF-d}_8$ )**



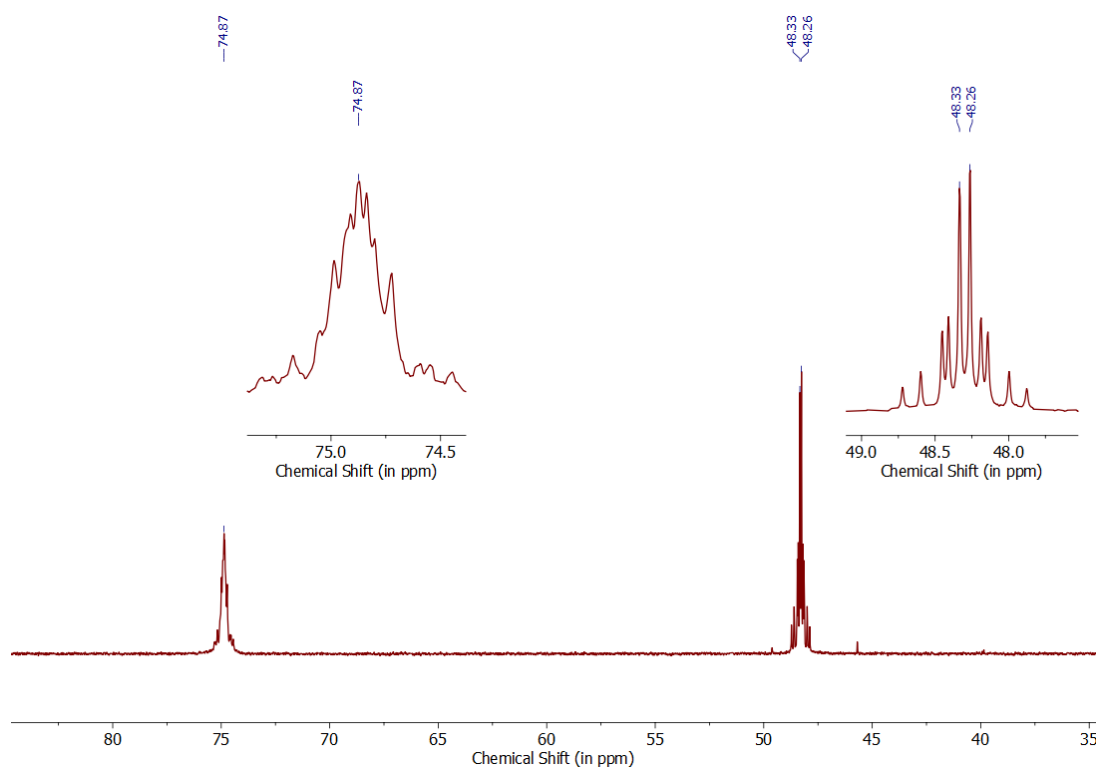
**II.3.4.  $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum (101 MHz,  $\text{THF-d}_8$ )**



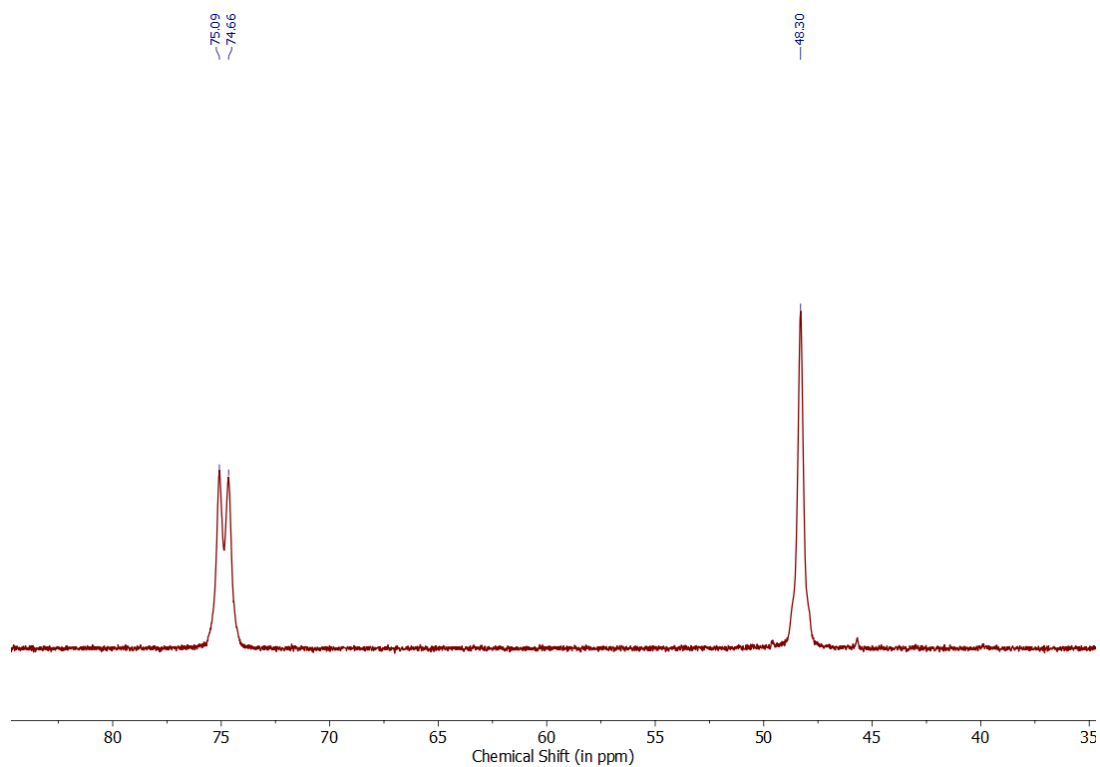
**II.3.5.  $^{13}\text{C}$ -DEPT NMR spectrum (101 MHz, THF- $d_8$ )**



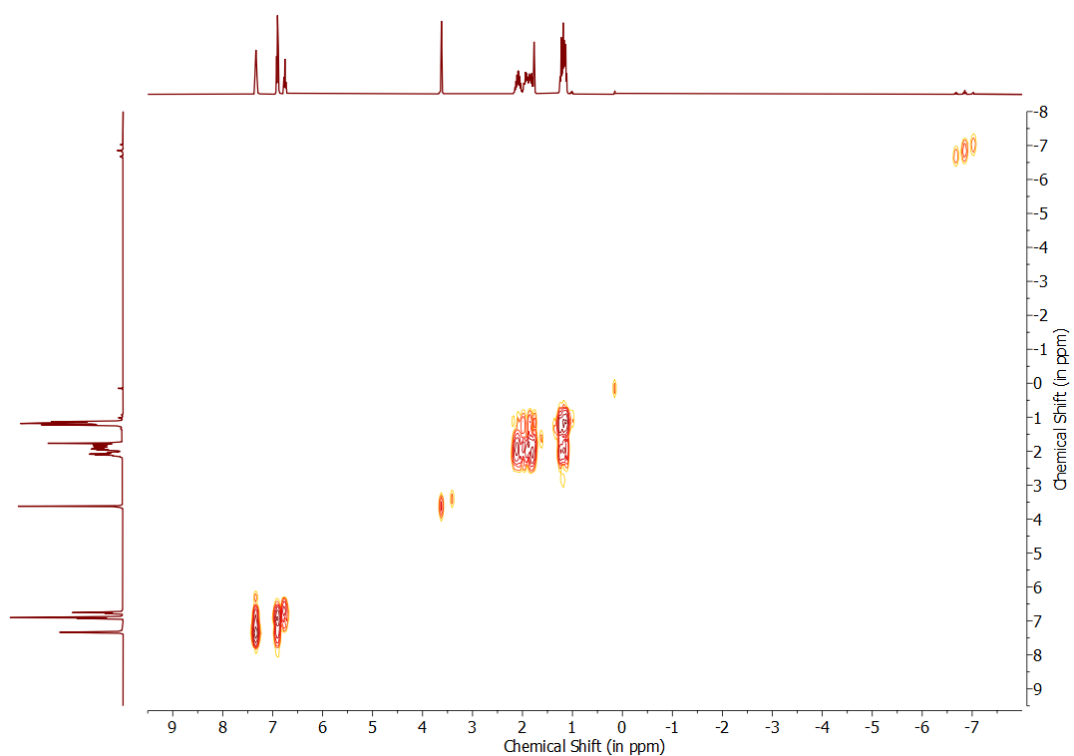
**II.3.6.  $^{31}\text{P}\{^1\text{H}\}$ -NMR spectrum (162 MHz, THF- $d_8$ )**



**II.3.7.  $^{31}\text{P}\{^1\text{H}\}$  sel ( $\delta = 1.5 \text{ ppm}$ )-NMR spectrum (162 MHz, THF- $d_8$ )**

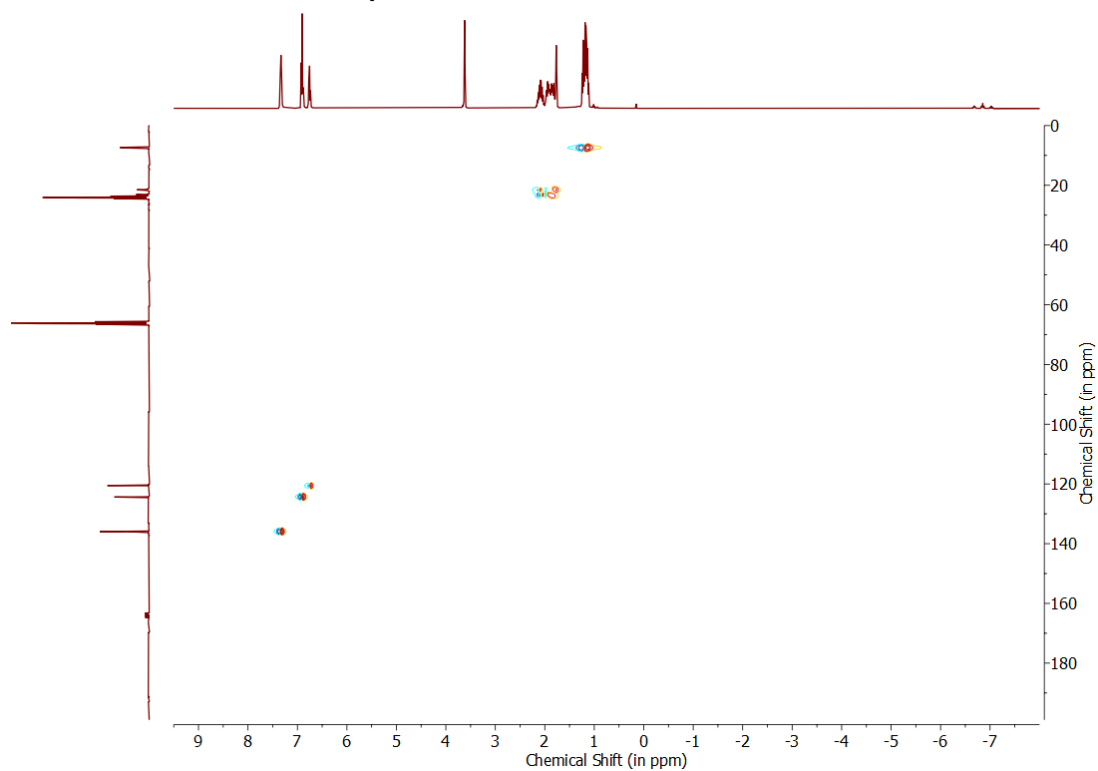


**II.3.8.  $^1\text{H}$ - $^1\text{H}$  COSY NMR spectrum**

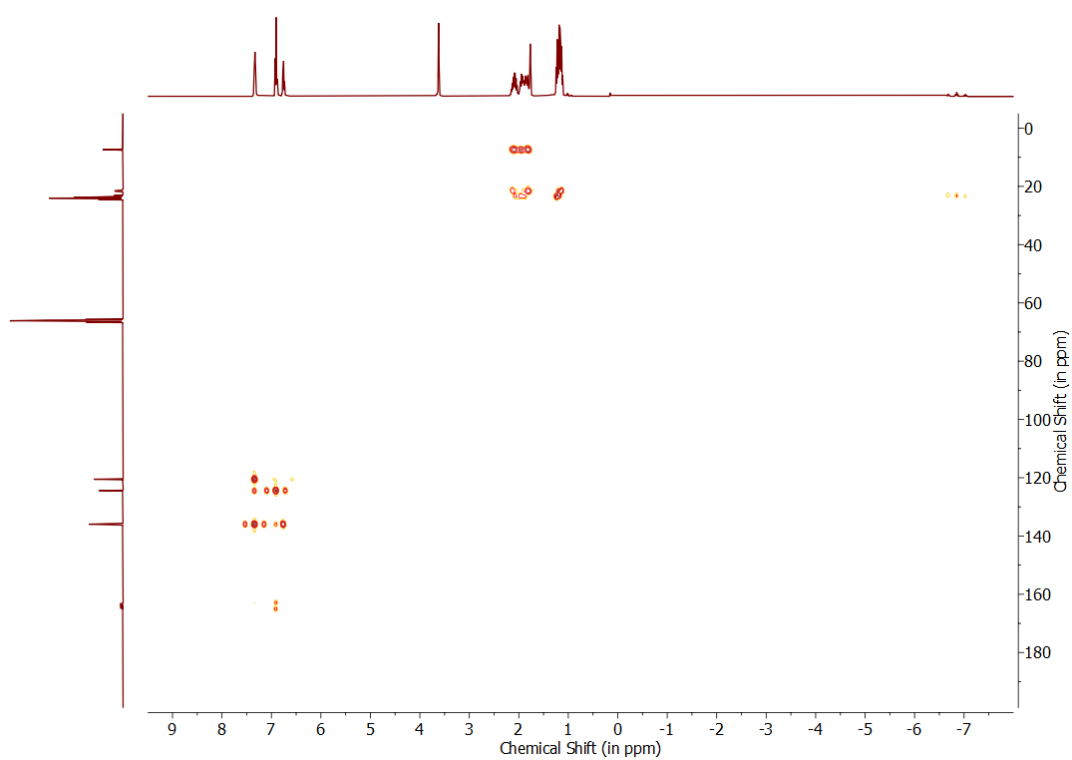




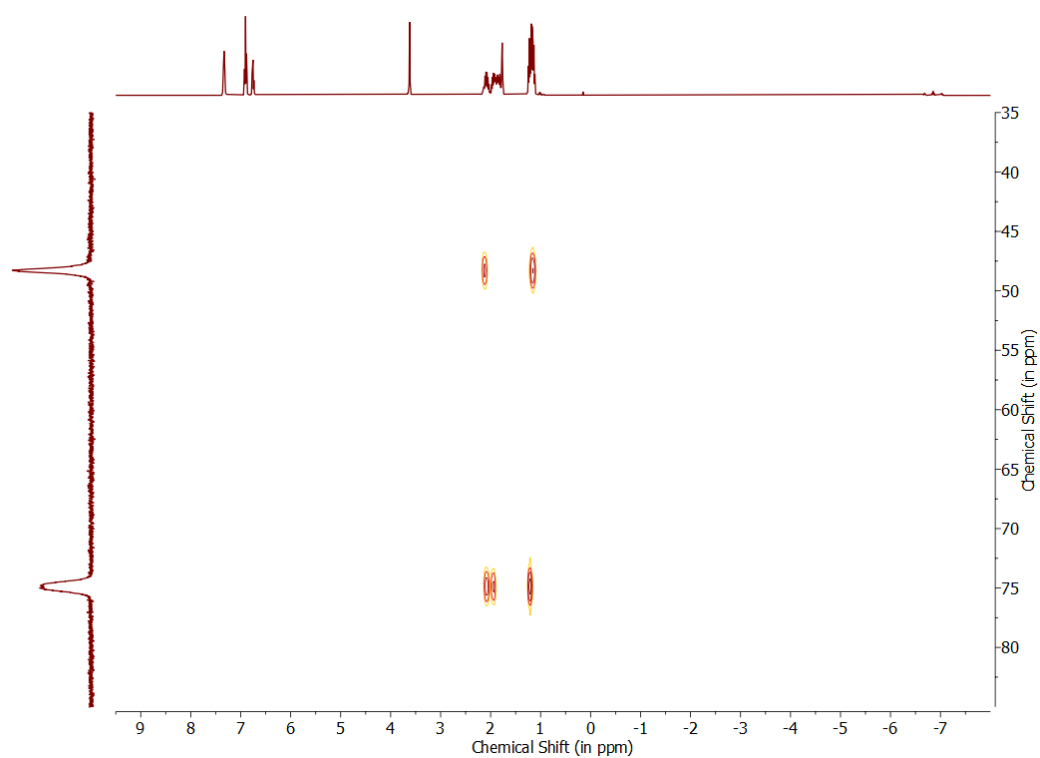
**II.3.9.  $^{13}\text{C}$ - $^1\text{H}$  HSQC NMR spectrum**



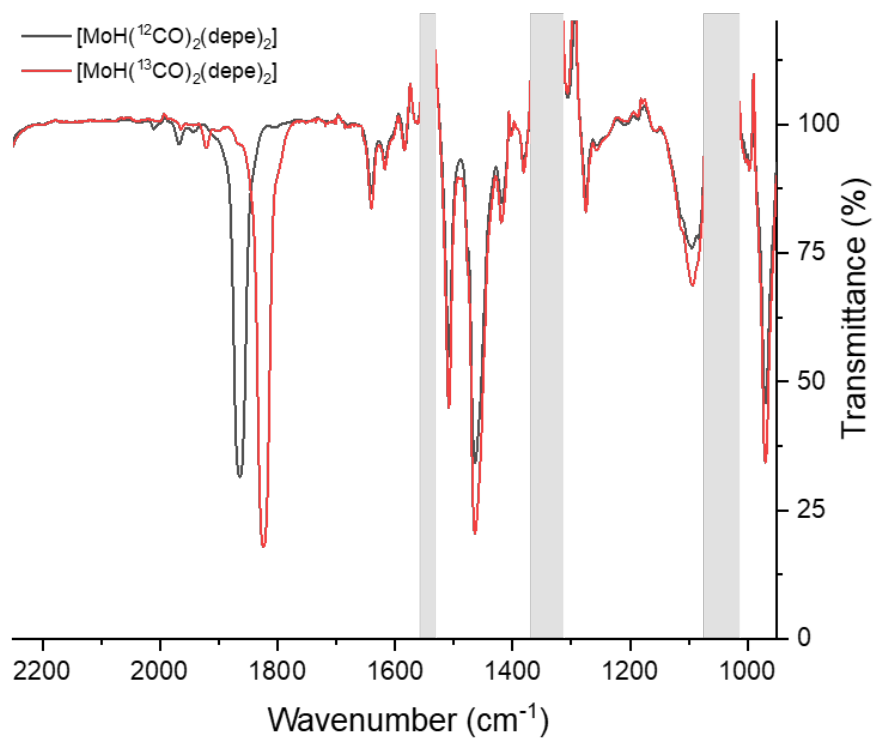
**II.3.10.  $^{13}\text{C}$ - $^1\text{H}$  HMBC NMR spectrum**



**II.3.11.  $^{31}\text{P}$ - $^1\text{H}$  HMQC NMR spectrum**

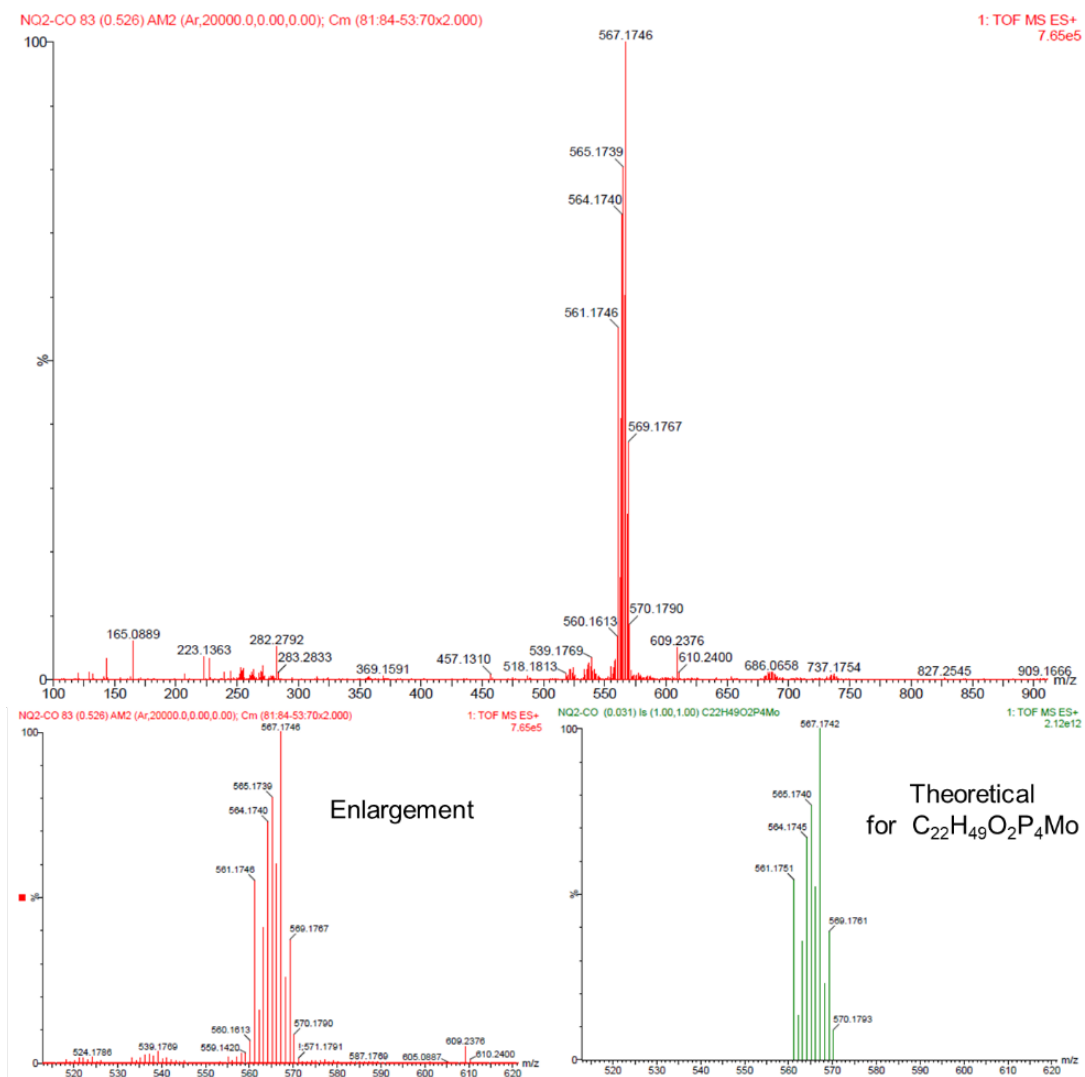


**II.3.12. FT-IR spectrum ( $^{13}\text{CO}$  vs  $^{12}\text{CO}$ ) – liquid phase**



Gray boxes correspond to solvent strong absorption area.

### II.3.13. ESI-HRMS spectrum



## II.4. [Mo(depe)<sub>2</sub>(CH<sub>2</sub>S<sub>2</sub>)H] (6.BPh<sub>4</sub>)

### II.4.1. Experimental procedure

In a Fisher-Porter flask, [Mo(depe)<sub>2</sub>H<sub>4</sub>] (20.0 mg, 39.0 μmol, 1.00 equiv.) was dissolved in THF (1 mL). A solution of [HNEt<sub>3</sub>][B(C<sub>6</sub>H<sub>5</sub>)<sub>4</sub>] (16.4 mg, 39.0 μmol, 1.00 equiv.) and CS<sub>2</sub> (3.0 mg, 39.0 μmol, 1.00 equiv.) in THF (1 mL) was added. The solution immediately turns brown. The reaction mixture was left at room temperature for 24h, under continuous stirring. Pentane (ca. 5 mL) was added to the solution, triggering immediate precipitation of a brown solid. The solution was decanted off and the resulting solid was washed twice with cold pentane (2 x 5 mL). The brown powder was then dried under high vacuum at room temperature to yield the title compound with a purity of ca. 92%, based on <sup>31</sup>P-NMR spectroscopy (27.5 mg, 78%).

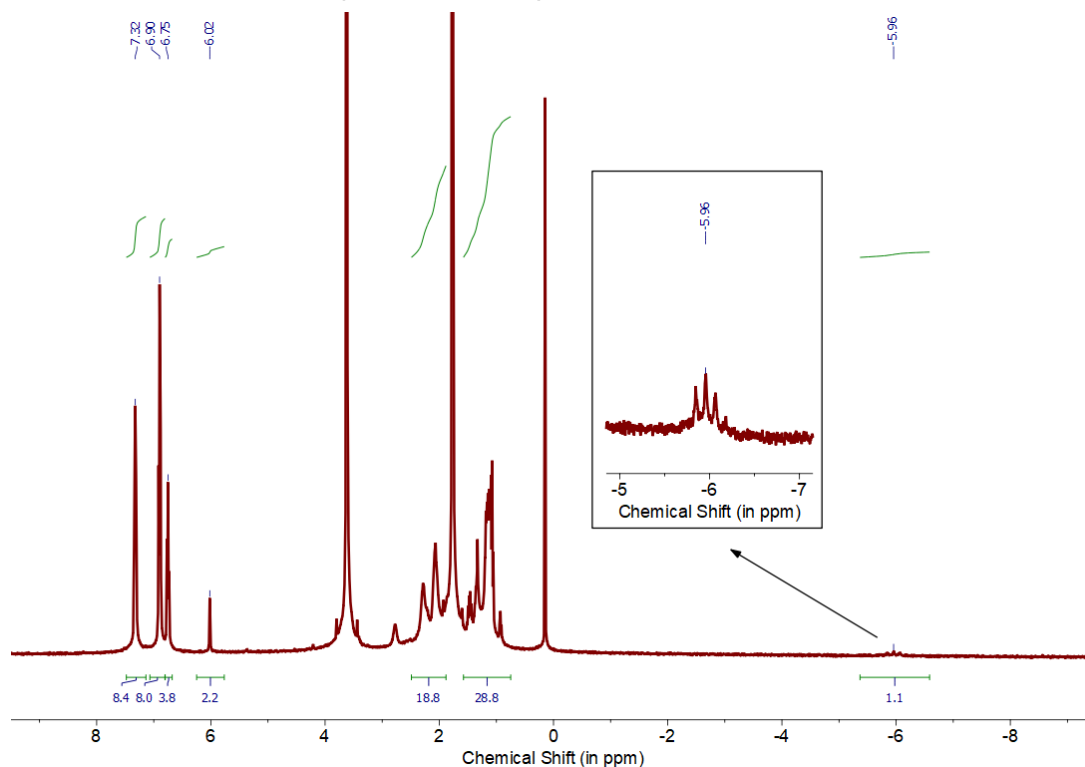
**<sup>1</sup>H-NMR** (400 MHz, THF-d<sub>8</sub>) δ 7.28 (br. m, 8H, *meta*-C<sub>6</sub>H<sub>5</sub>), 6.86 (t, <sup>3</sup>J<sub>(H-H)</sub> = 7.3 Hz, 8H, *ortho*-C<sub>6</sub>H<sub>5</sub>), 6.71 (t, <sup>3</sup>J<sub>(H-H)</sub> = 7.3 Hz, 4H, *para*-C<sub>6</sub>H<sub>5</sub>), 5.98 (s, 2H, -CH<sub>2</sub>S<sub>2</sub>-), 2.35-1.85 (m, 18H), 1.50 – 0.79 (m, 30H), -5.99 (quintet, <sup>2</sup>J<sub>(P-H)</sub> = 43.1 Hz, 1H, Mo-H).

**<sup>11</sup>B-NMR** (128 MHz, THF-d<sub>8</sub>) δ -6.8.

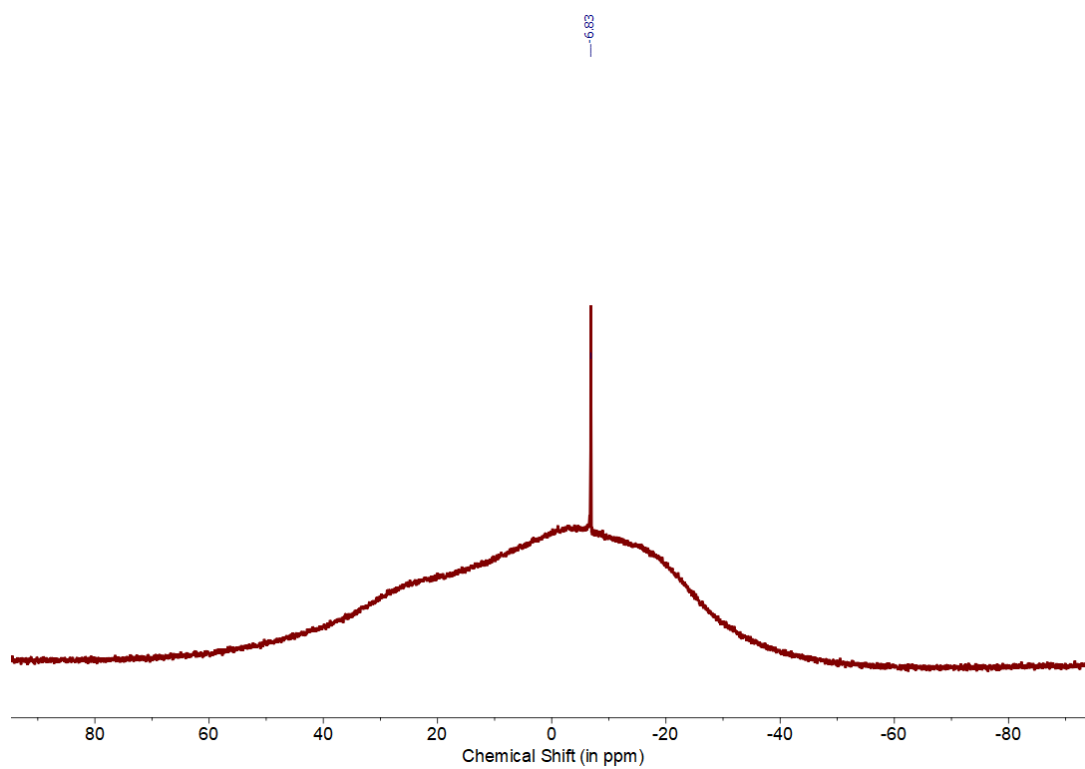
**<sup>13</sup>C{<sup>1</sup>H}-NMR** (101 MHz, THF-d<sub>8</sub>) δ 163.9 (q, <sup>2</sup>J<sub>(B-C)</sub> = 50.3 Hz, *ipso*-C<sub>6</sub>H<sub>5</sub>), 135.9 (br. s, *meta*-C<sub>6</sub>H<sub>5</sub>), 124.4 (m, *ortho*-C<sub>6</sub>H<sub>5</sub>), 120.5 (s, *para*-C<sub>6</sub>H<sub>5</sub>), 61.5 (s, -S-CH<sub>2</sub>-S-), 21.1 (m, -CH<sub>2</sub>-), 8.6 (s, -CH<sub>3</sub>), 7.9 (s, -CH<sub>3</sub>).

**<sup>31</sup>P{<sup>1</sup>H}-NMR** (162 MHz, THF-d<sub>8</sub>) δ 69.0 (s). **<sup>31</sup>P{<sup>1</sup>H}<sub>sel</sub>-NMR** (162 MHz, THF-d<sub>8</sub>) δ 69.0 (d, <sup>2</sup>J<sub>(P-H)</sub> = 43.1 Hz).

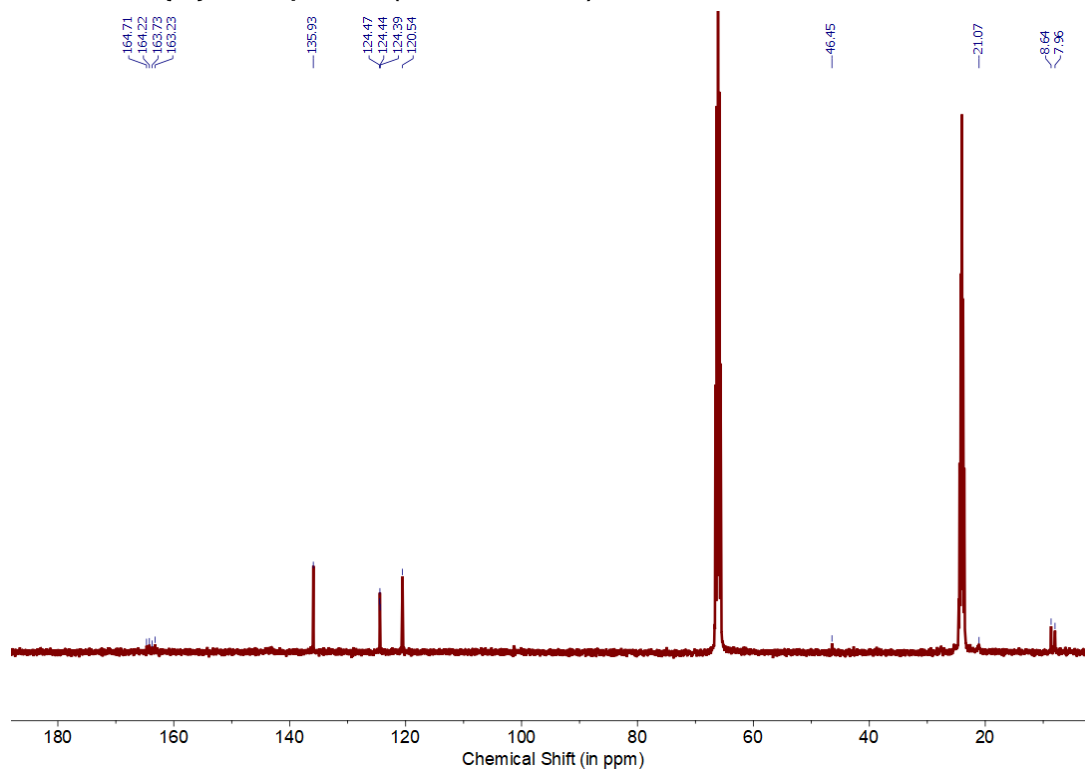
### II.4.2. <sup>1</sup>H-NMR spectrum (400 MHz, THF-d<sub>8</sub>)



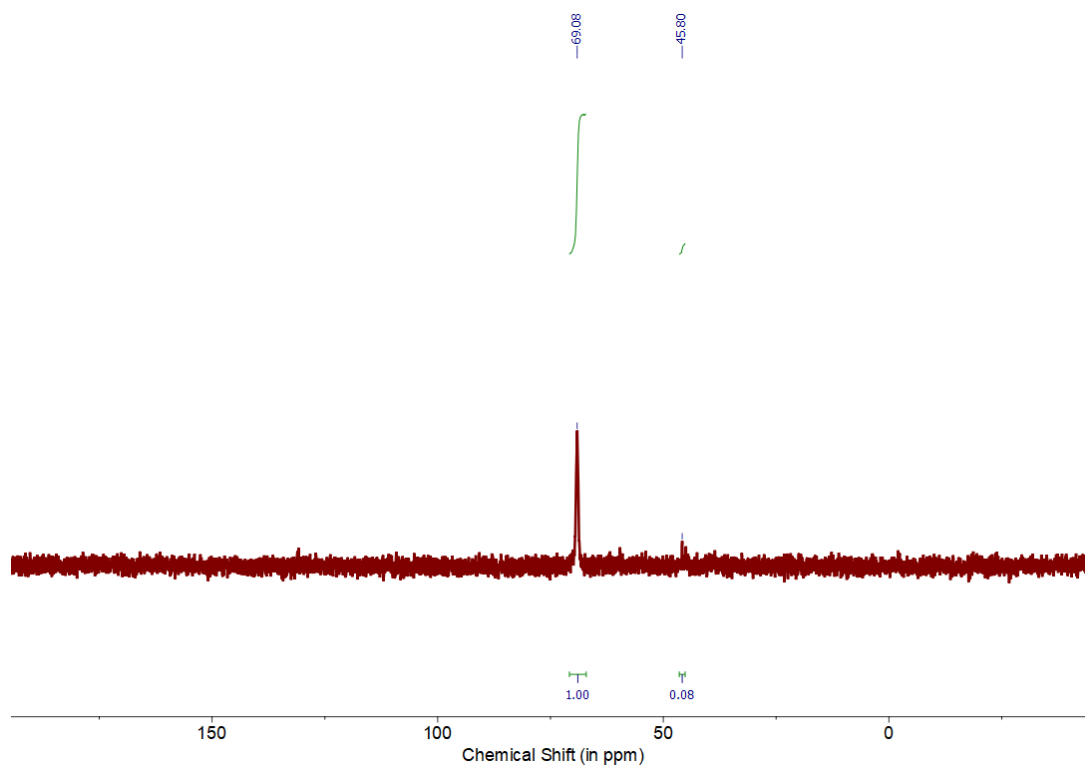
**II.4.3.  $^{11}\text{B}\{^1\text{H}\}$ -NMR spectrum (128 MHz,  $\text{THF-d}_8$ )**



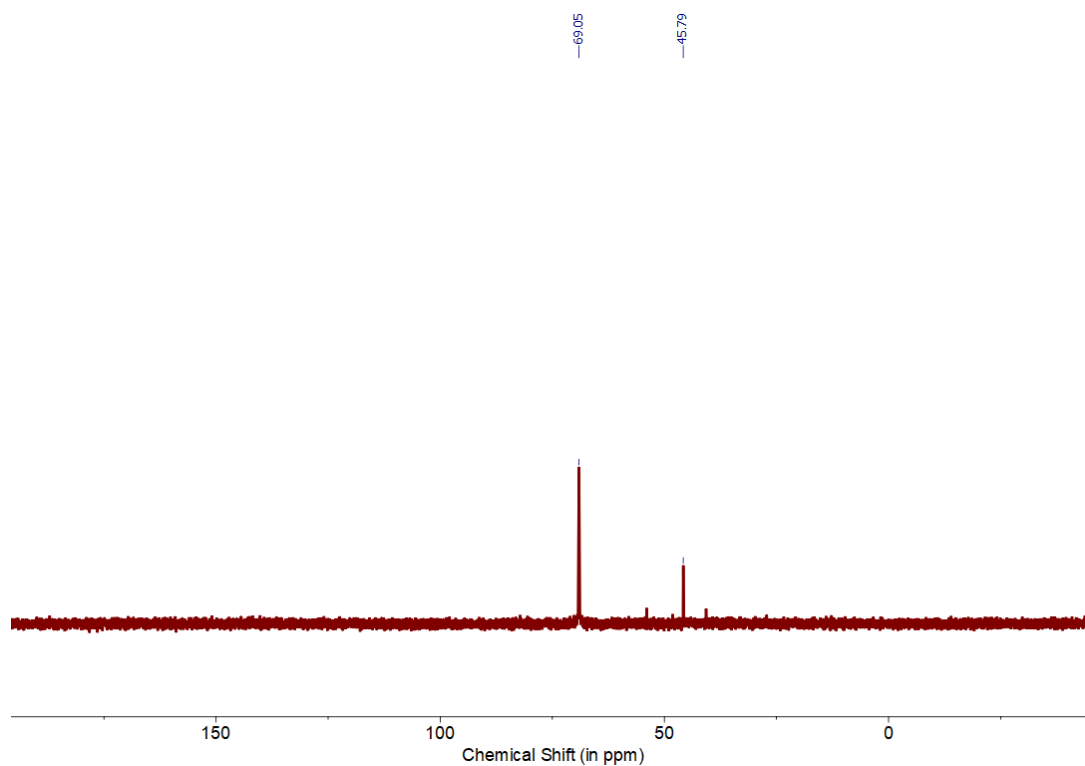
**II.4.4.  $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum (101 MHz,  $\text{THF-d}_8$ )**



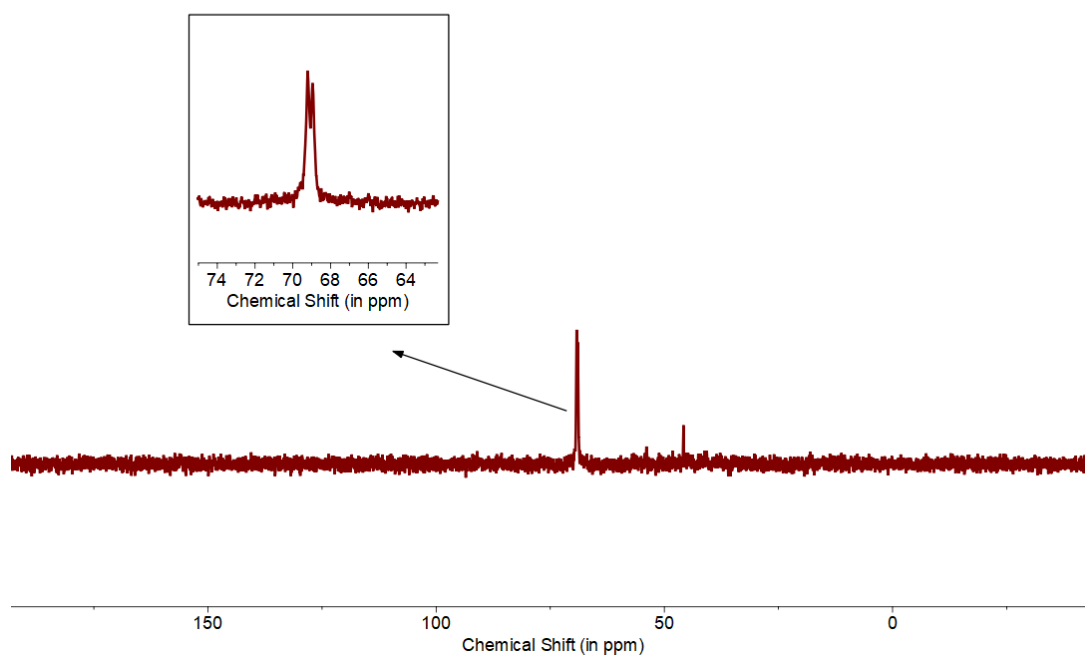
**II.4.5.  $^{31}\text{P}$ -NMR spectrum (162 MHz,  $\text{THF-d}_8$ )**



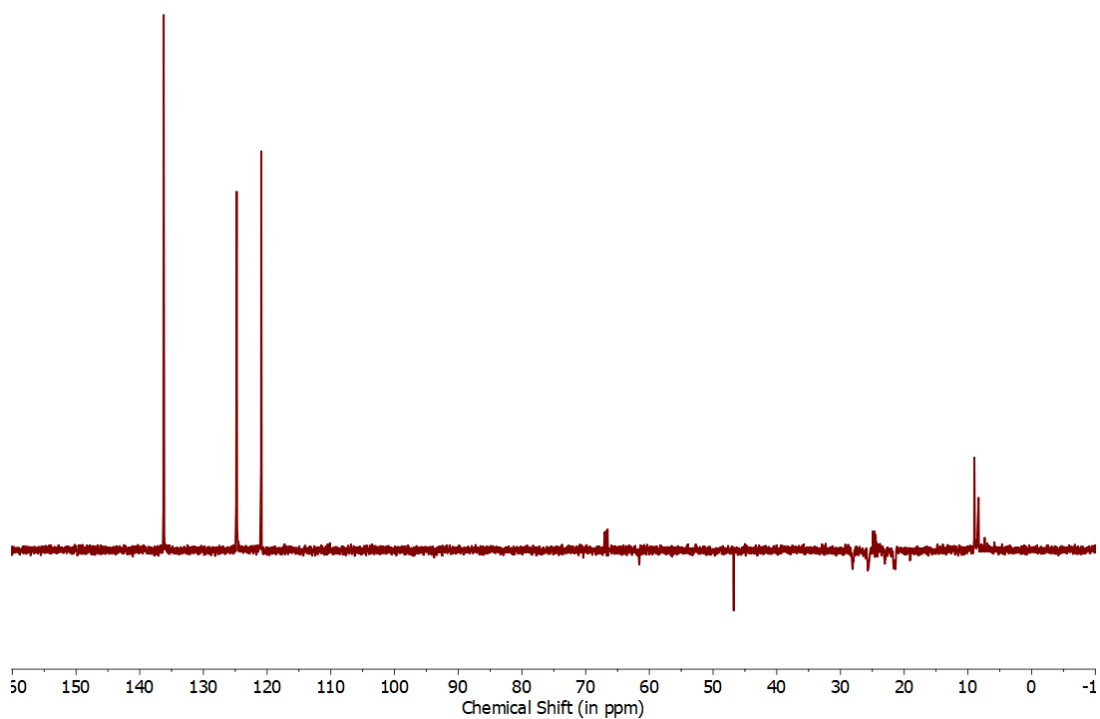
**II.4.6.  $^{31}\text{P}\{^1\text{H}\}$  NMR spectrum (162 MHz,  $\text{THF-d}_8$ )**



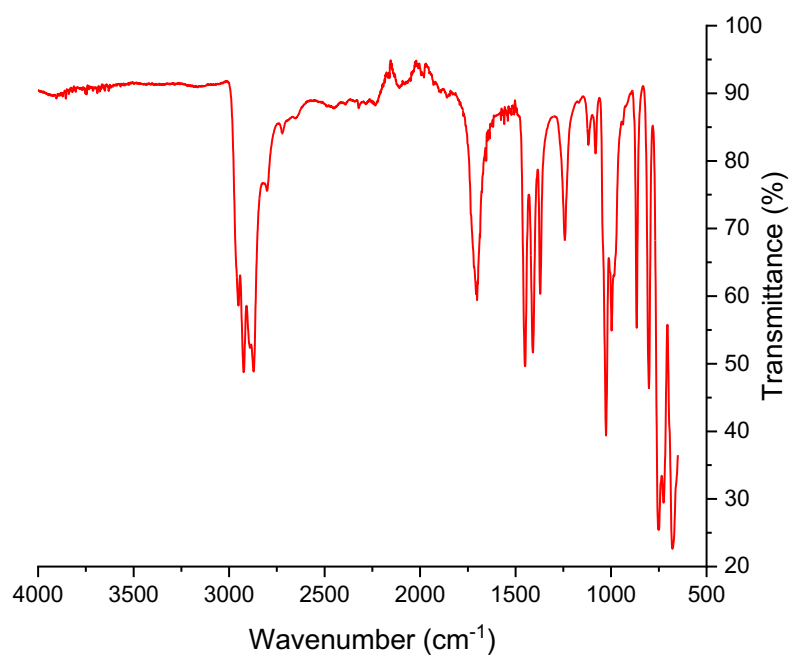
**II.4.7.  $^{31}\text{P}\{^1\text{H}\}_{\text{sel}}$  ( $\delta = 1.5 \text{ ppm}$ )-NMR spectrum (162 MHz, THF- $d_8$ )**



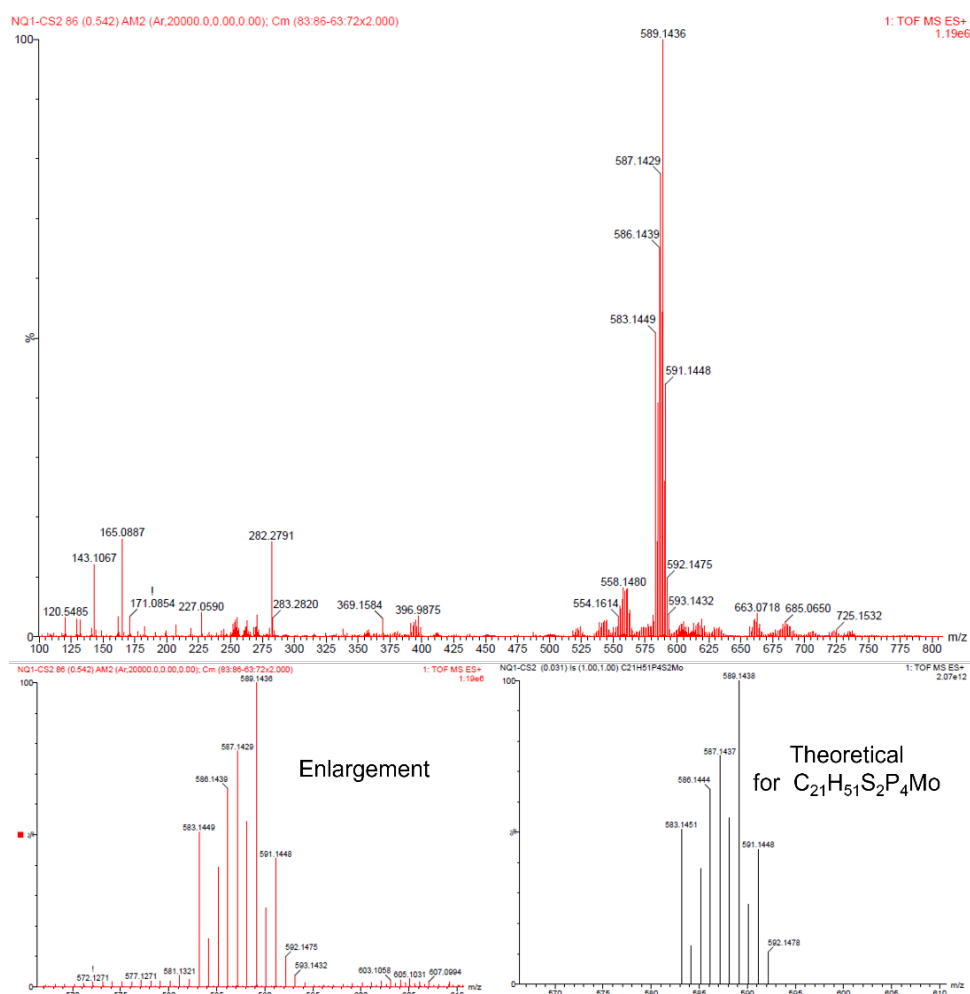
**II.4.8.  $^{13}\text{C}$ -DEPT NMR spectrum (101 MHz, THF- $d_8$ )**



#### II.4.9. FT-IR spectrum (ATR)



#### II.4.10. ESI-HRMS spectrum





## II.5. [Mo(depe)<sub>2</sub>(<sup>i</sup>PrNCHN<sup>i</sup>Pr)H<sub>2</sub>] (7.BPh<sub>4</sub>)

### II.5.1. Experimental procedure

In a Fisher-Porter flask, [Mo(depe)<sub>2</sub>H<sub>4</sub>] (20.0 mg, 39.0 μmol, 1.00 equiv.) was dissolved in THF (1 mL). A solution of [HNEt<sub>3</sub>][B(C<sub>6</sub>H<sub>5</sub>)<sub>4</sub>] (16.4 mg, 39.0 μmol, 1.00 equiv.) and <sup>i</sup>PrNCHN<sup>i</sup>Pr (4.9 mg, 39.0 μmol, 1.00 equiv.) in THF (1 mL) was added. The reaction mixture was left at room temperature for 24h, under continuous stirring. Pentane (ca. 10 mL) was added to the orange solution, triggering immediate phase separation of a dark orange oil. The solution was decanted off and the oil triturated twice with cold pentane (2 x 5 mL). The solid was then dried under high vacuum at room temperature to yield the desired compound (12.7 mg, 34%). Single crystals suitable for X-ray diffraction were obtained by storing at -40°C a concentrated THF solution of the complex layered by pentane over a few days.

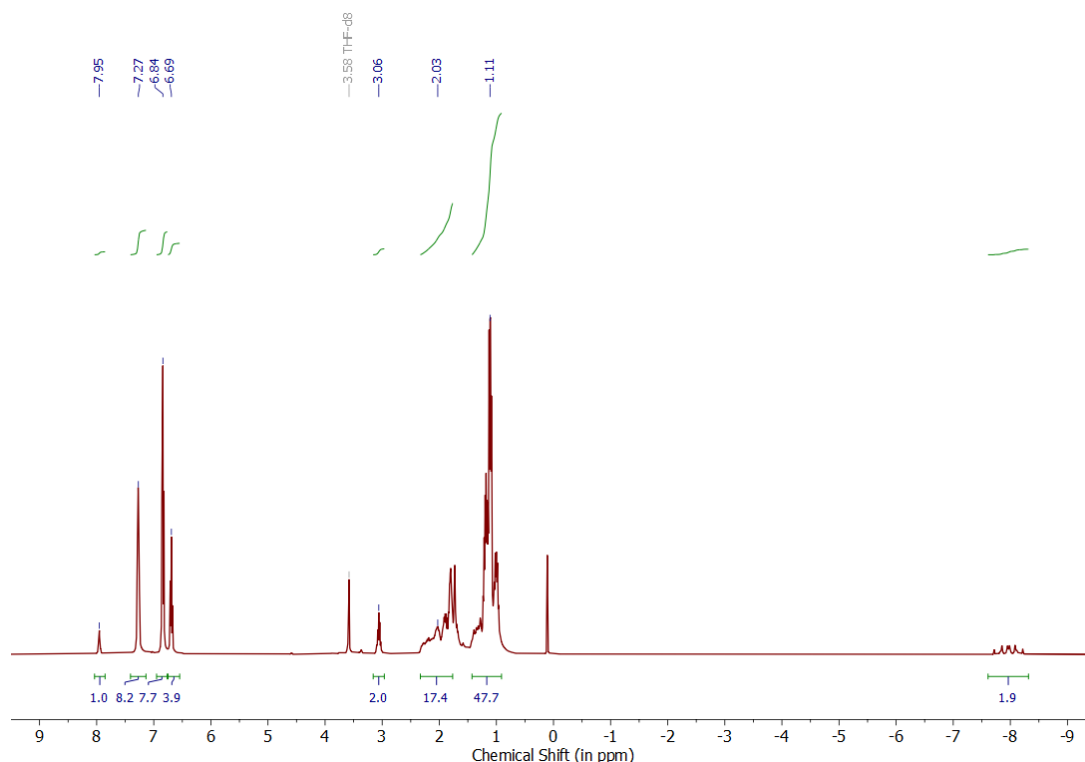
**<sup>1</sup>H-NMR** (400 MHz, THF-d<sub>8</sub>, 293K) δ 7.95 (m, 1H, <sup>i</sup>PrNCHN<sup>i</sup>Pr), 7.27 (br. m, 8H, *meta*-C<sub>6</sub>H<sub>5</sub>), 6.84 (t, <sup>3</sup>J<sub>(H-H)</sub> = 7.4 Hz, 8H, *ortho*-C<sub>6</sub>H<sub>5</sub>), 6.69 (t, <sup>3</sup>J<sub>(H-H)</sub> = 7.2 Hz, 4H, *para*-C<sub>6</sub>H<sub>5</sub>), 3.06 (hept., 2H, -CH(CH<sub>3</sub>)<sub>2</sub>), 2.33-1.76 (m, 16H, -CH<sub>2</sub>-CH<sub>3</sub>), 1.11 (m, 44H, -CH<sub>2</sub>-CH<sub>2</sub>- & -CH(CH<sub>3</sub>)<sub>2</sub> & -CH<sub>2</sub>-CH<sub>3</sub>), -7.97 (m, 2H).

**<sup>11</sup>B-NMR** (128 MHz, THF-d<sub>8</sub>) δ -6.5.

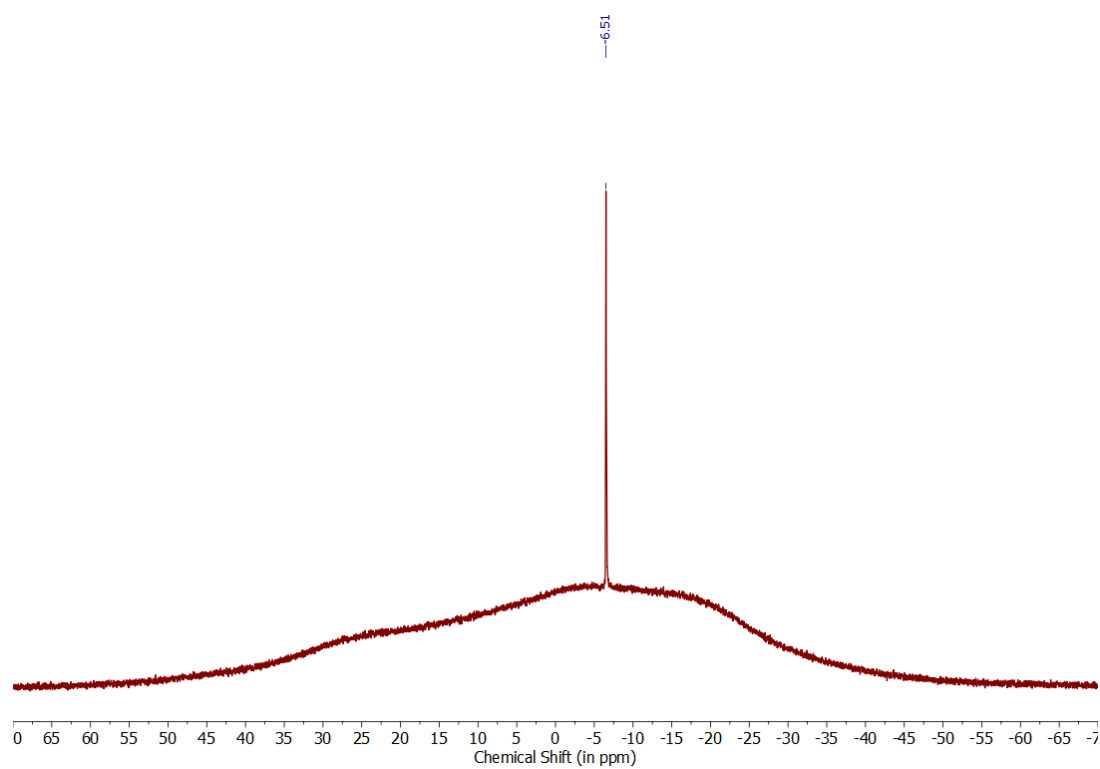
**<sup>13</sup>C{<sup>1</sup>H}-NMR** (101 MHz, THF-d<sub>8</sub>) δ 164.3 (q, <sup>2</sup>J<sub>(B-C)</sub> = 49.5 Hz, *ipso*-C<sub>6</sub>H<sub>5</sub>), 159.4 (m, <sup>i</sup>PrNCHN<sup>i</sup>Pr), 136.2 (br. s, *meta*-C<sub>6</sub>H<sub>5</sub>), 124.6 (m, *ortho*-C<sub>6</sub>H<sub>5</sub>), 120.7 (s, *para*-C<sub>6</sub>H<sub>5</sub>), 49.8 (-CH(CH<sub>3</sub>)<sub>2</sub>), 27.0 (-CH<sub>2</sub>-), 26.1 (-CH<sub>2</sub>-), 21.9 (-CH<sub>3</sub>), 20.8 (-CH<sub>2</sub>-), 13.6 (-CH<sub>2</sub>-), 8.5 (-CH<sub>3</sub>), 7.8 (-CH<sub>3</sub>), 7.5 (-CH<sub>3</sub>).

**<sup>31</sup>P{<sup>1</sup>H}-NMR** (162 MHz, THF-d<sub>8</sub>) δ 71.0 (m), 41.5 (m). **<sup>31</sup>P{<sup>1</sup>H}<sub>sel</sub>-NMR** (162 MHz, THF-d<sub>8</sub>) δ 71.0 (t, <sup>2</sup>J<sub>(P-H)</sub> = 50.8 Hz), 48.3 (t, <sup>2</sup>J<sub>(P-H)</sub> = 43.5 Hz).

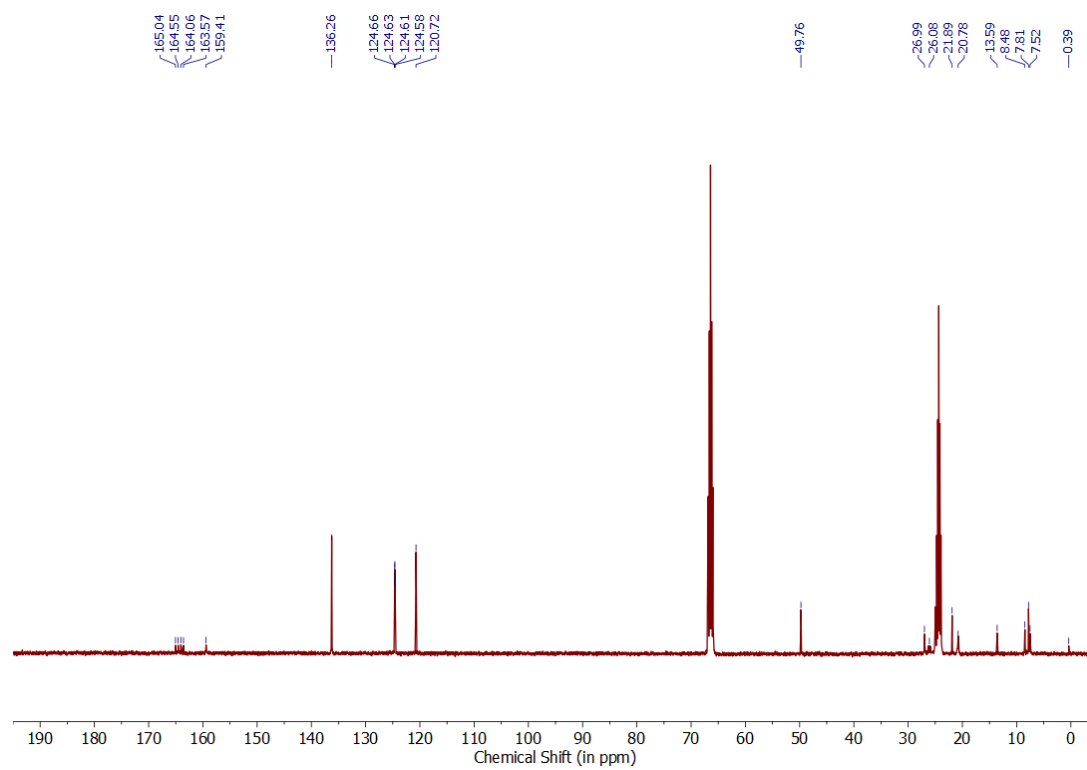
### II.5.2. <sup>1</sup>H-NMR spectrum (400 MHz, THF-d<sub>8</sub>)



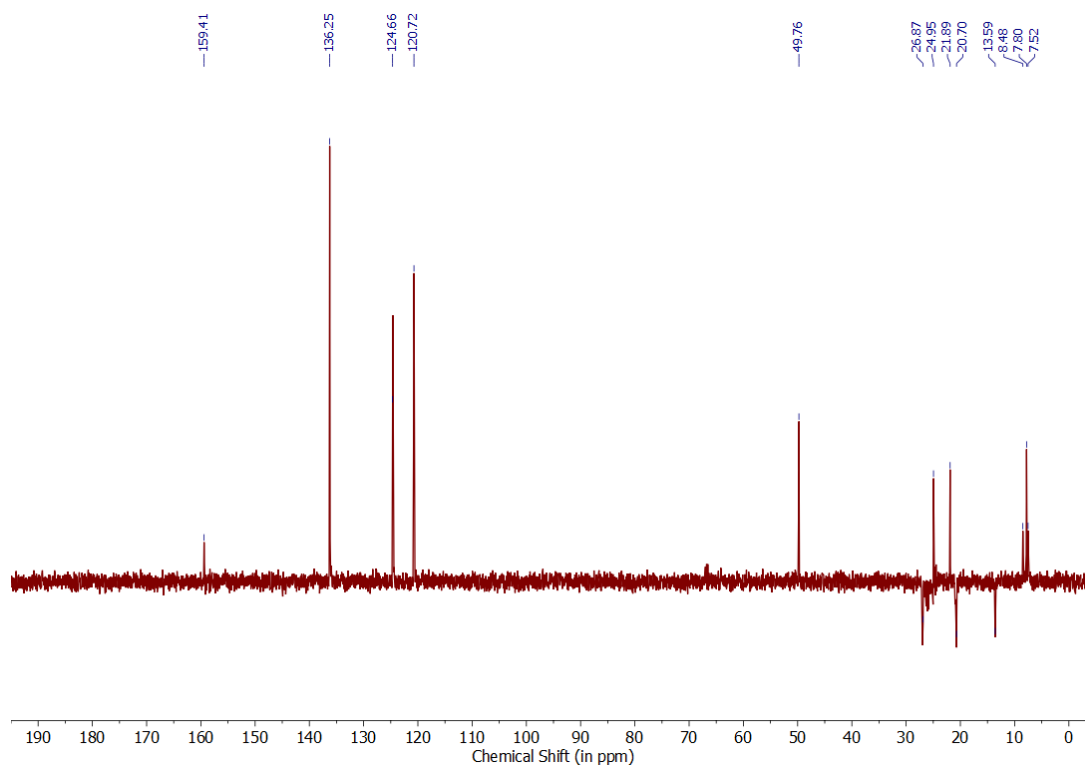
**II.5.3.  $^{11}\text{B}\{^1\text{H}\}$ -NMR spectrum (128 MHz,  $\text{THF-d}_8$ )**



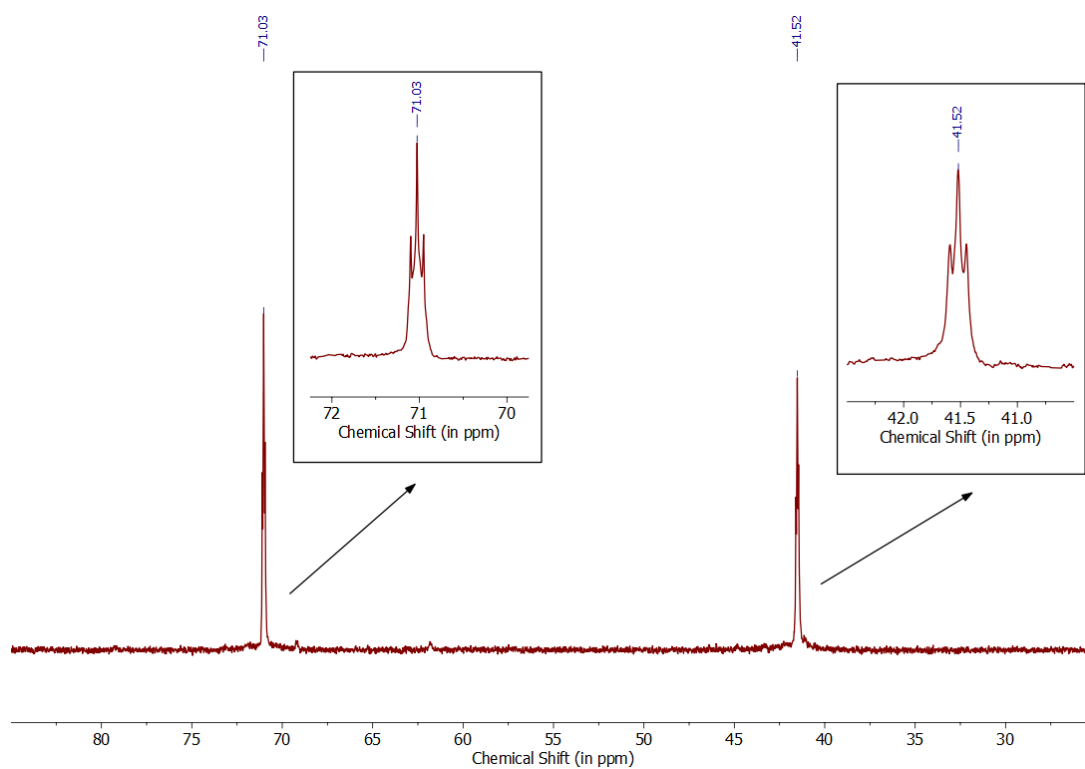
**II.5.4.  $^{13}\text{C}\{^1\text{H}\}$ -NMR spectrum (101 MHz,  $\text{THF-d}_8$ )**



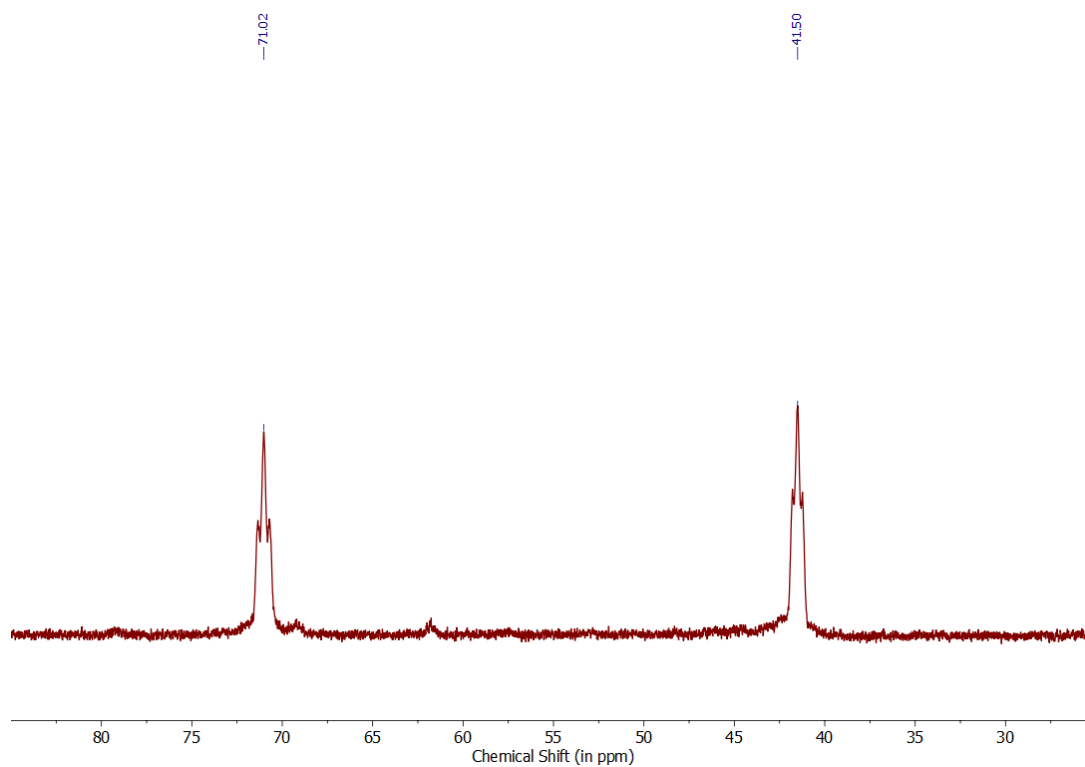
**II.5.5.  $^{13}\text{C}$ -DEPT NMR spectrum (101 MHz, THF- $d_8$ )**



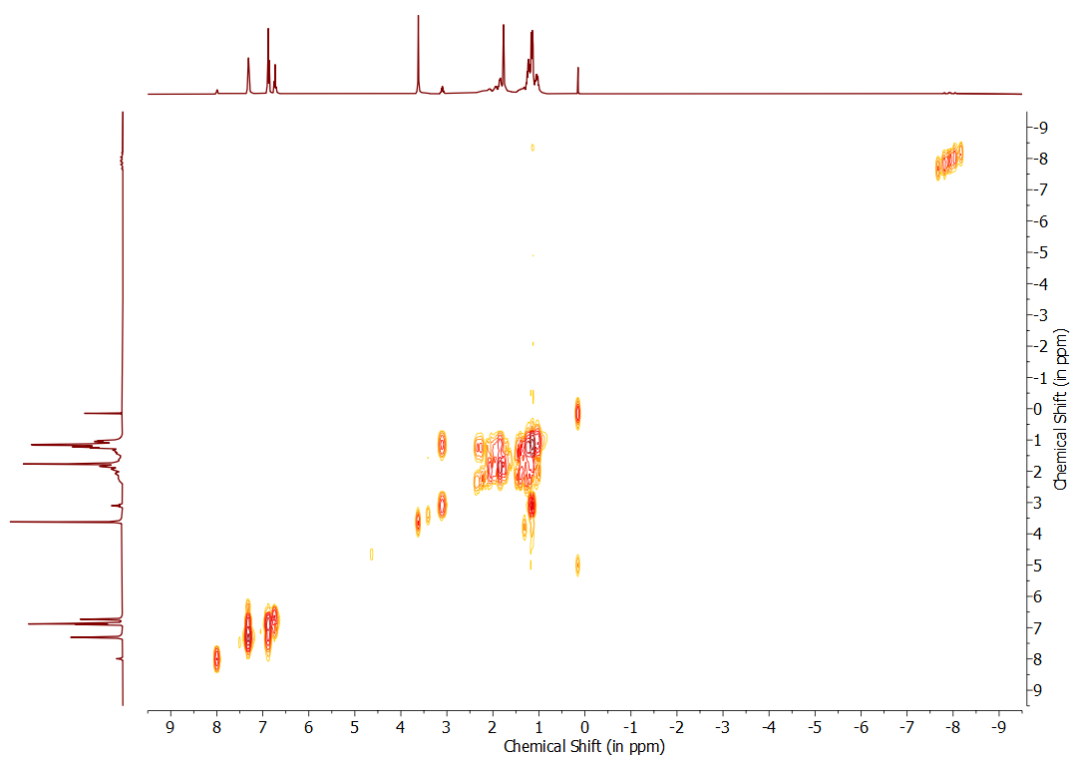
**II.5.6.  $^{31}\text{P}\{^1\text{H}\}$ -NMR spectrum (162 MHz, THF- $d_8$ )**



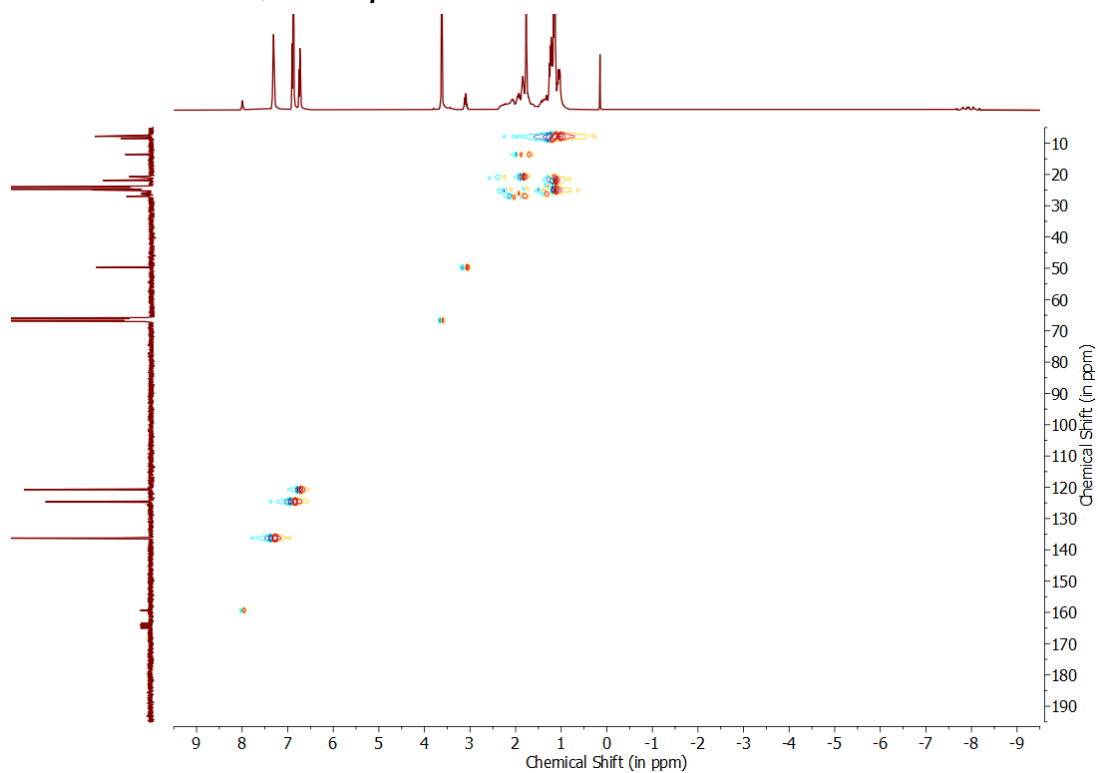
**II.5.7.  $^{31}\text{P}\{^1\text{H}\}$  sel ( $\delta = 1.5 \text{ ppm}$ )-NMR spectrum (162 MHz, THF- $d_8$ )**



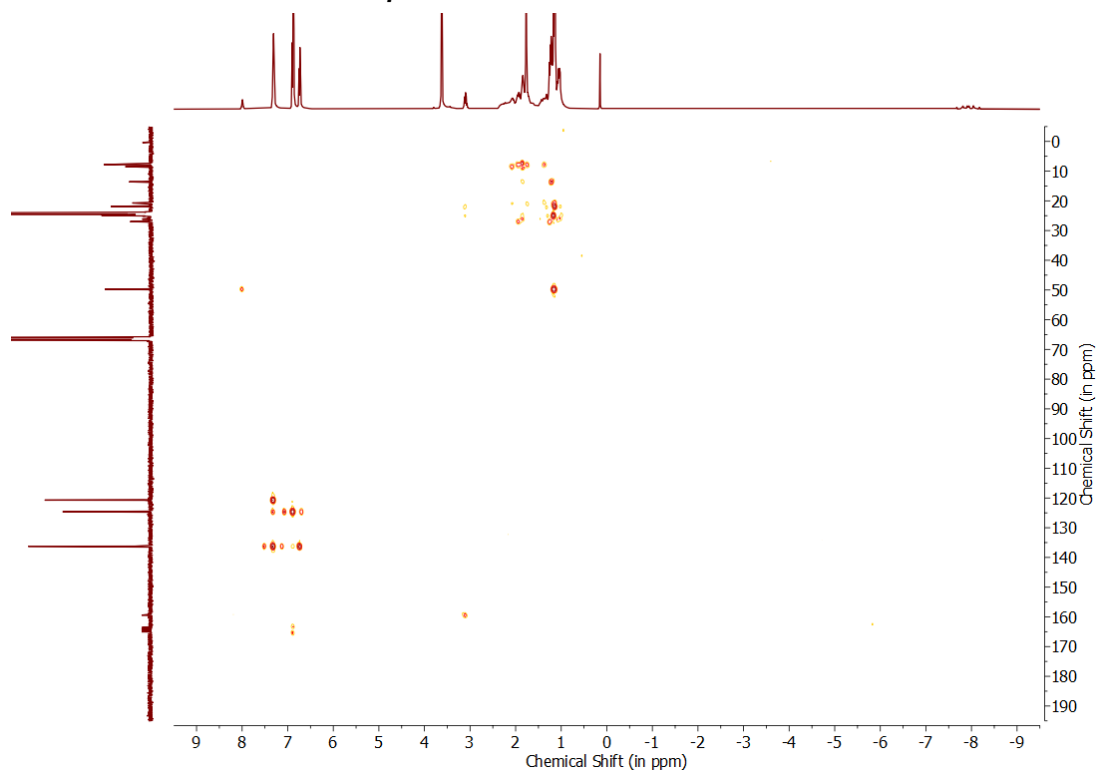
**II.5.8.  $^1\text{H}$ - $^1\text{H}$  COSY NMR spectrum**



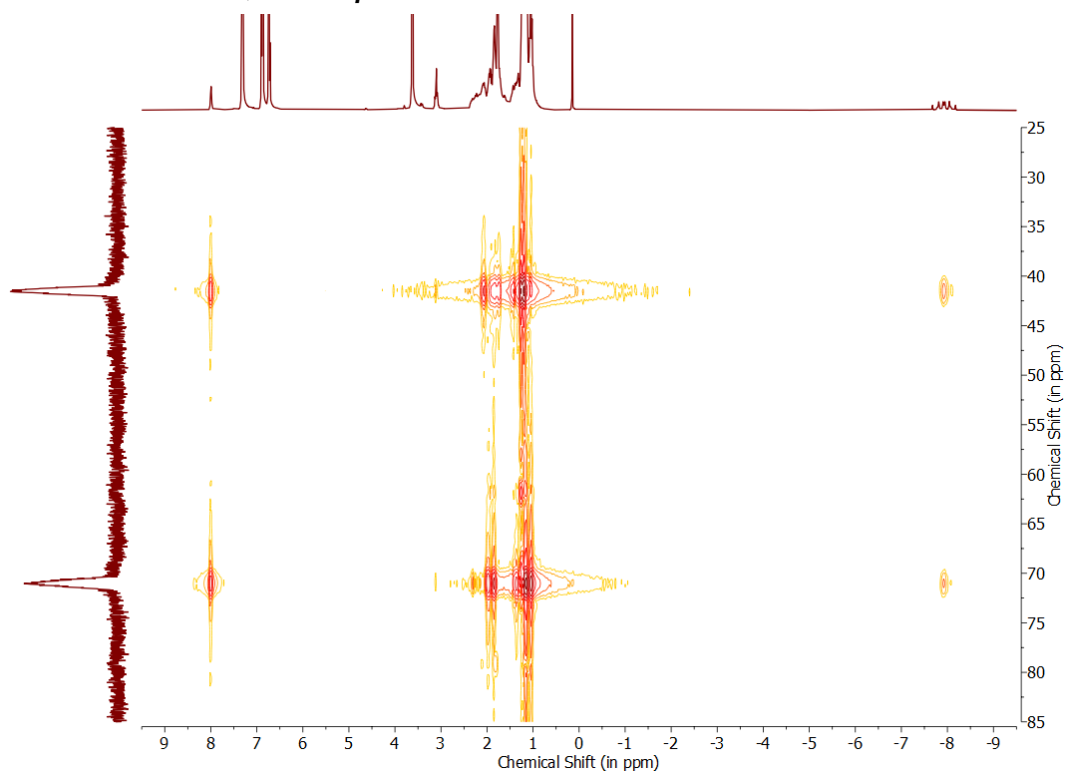
**II.5.9.  $^{13}\text{C}$ - $^1\text{H}$  HSQC NMR spectrum**



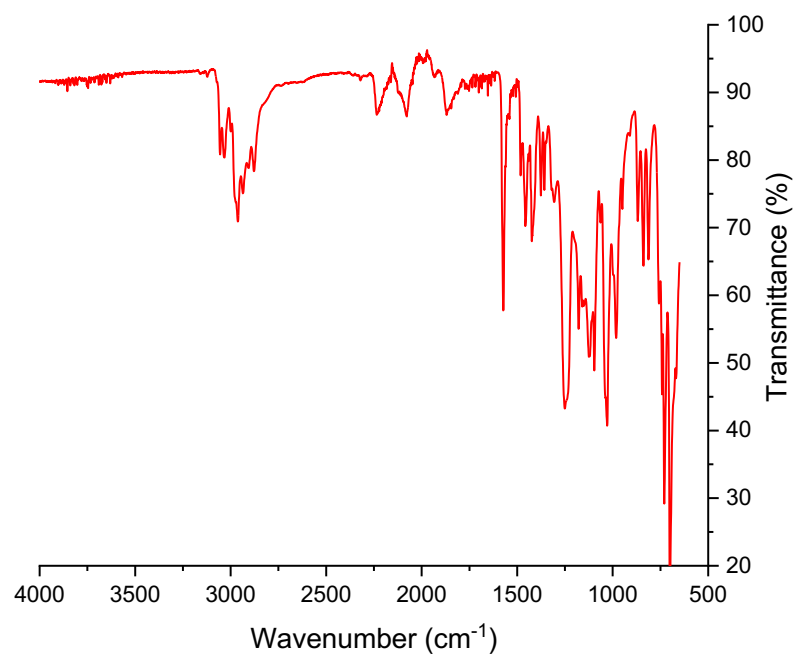
**II.5.10.  $^{13}\text{C}$ - $^1\text{H}$  HMBC NMR spectrum**



**II.5.11.  $^{31}\text{P}$ - $^1\text{H}$  HMQC NMR spectrum**



**II.5.12. FT-IR spectrum (ATR)**



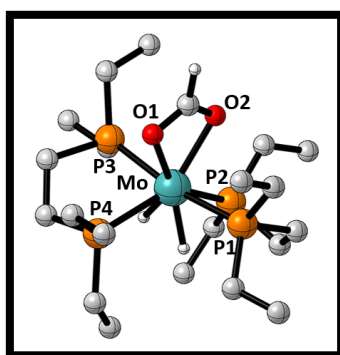
### III. Computational Details

#### III.1. General Information

All geometry optimizations and frequency calculations were carried out without symmetry constraints with the Gaussian16 Revision B.01 suite of programs<sup>[12]</sup> using the B3LYP<sup>[13–15]</sup> functional in conjunction with the Grimme D3(BJ) dispersion correction.<sup>[16,17]</sup> The Molybdenum atom was modelled using the relativistic effective core potential SDD<sup>[18–20]</sup> basis set, augmented with an f-type polarization function having an exponent of 1.043.<sup>[21]</sup> The 6-31G(d,p)<sup>[22,23]</sup> basis set was used for other atoms (H, C, O, P). Therefore, the level of theory of the optimizations can be described as B3LYP-D3(BJ)/SDD+f(Mo), 6-31G\*\* (other atoms). Further energy refinement single-point calculations were performed at the higher M06-L<sup>[24]</sup>-D3/def2-TZVPP<sup>[25,26]</sup> level of theory considering solvents (tetrahydrofuran: THF) effects by means of the Polarizable Continuum Model (PCM)<sup>[27–29]</sup> method. Frequency calculations were performed on minima (reactants, intermediates and products) resulting in positive definite Hessian matrices. Transition states (TS) were identified by a single negative eigenvalue in their diagonalized force constant matrices. To confirm the TSs nature, the eigenvectors associated with the negative eigenvalue were confirmed to correspond to the motion along the reaction coordinate employing the Intrinsic Reaction Coordinate (IRC) method.<sup>[30]</sup> Molecular structure images were generated utilizing Cylview software.<sup>[31]</sup> For the thermal evolution of compound **3**<sup>+</sup> into **4**<sup>+</sup> and **5**<sup>+</sup>, the GoodVibes package<sup>[32]</sup> was employed to correct free energy at 373.15 K.

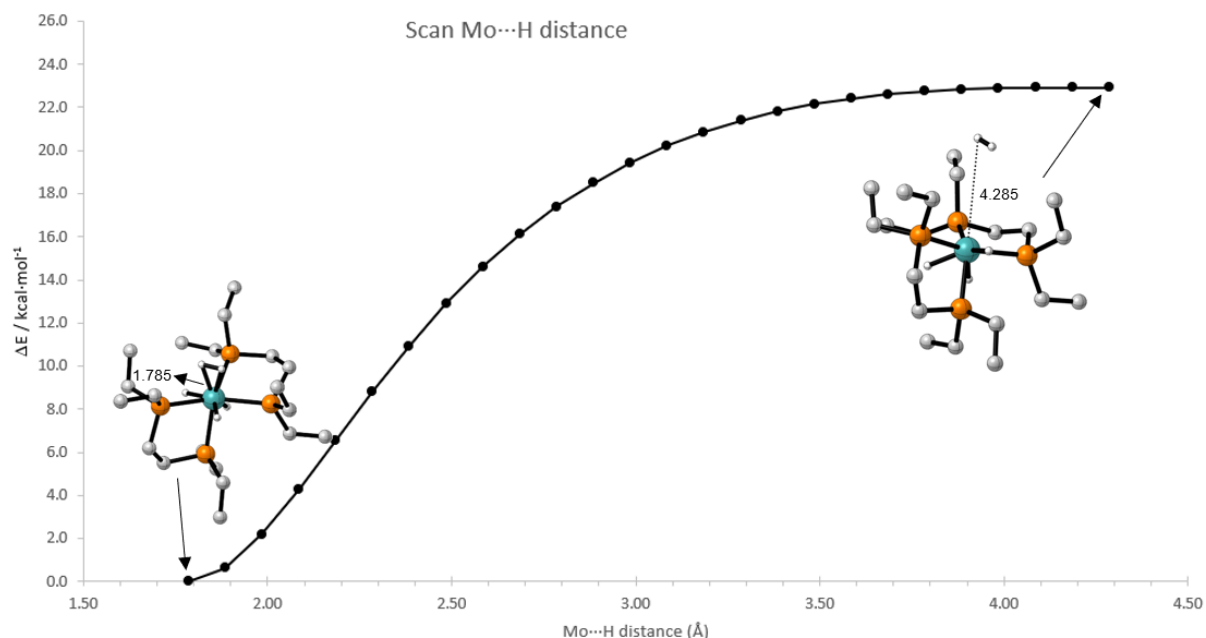
#### III.2. Detailed Results

**Table S1.** Comparison of the X-ray diffraction structure of naked cationic **3**<sup>+</sup> *versus* the DFT-computed structure. Distances and bond angles are given in angstroms and °, respectively. The image shows the DFT-optimized geometry at B3LYP-D3(BJ)/SDD+f(Mo), 6-31G\*\* (other atoms) level of theory. Hydrogen atoms in the diphosphine ligands have been omitted for clarity.



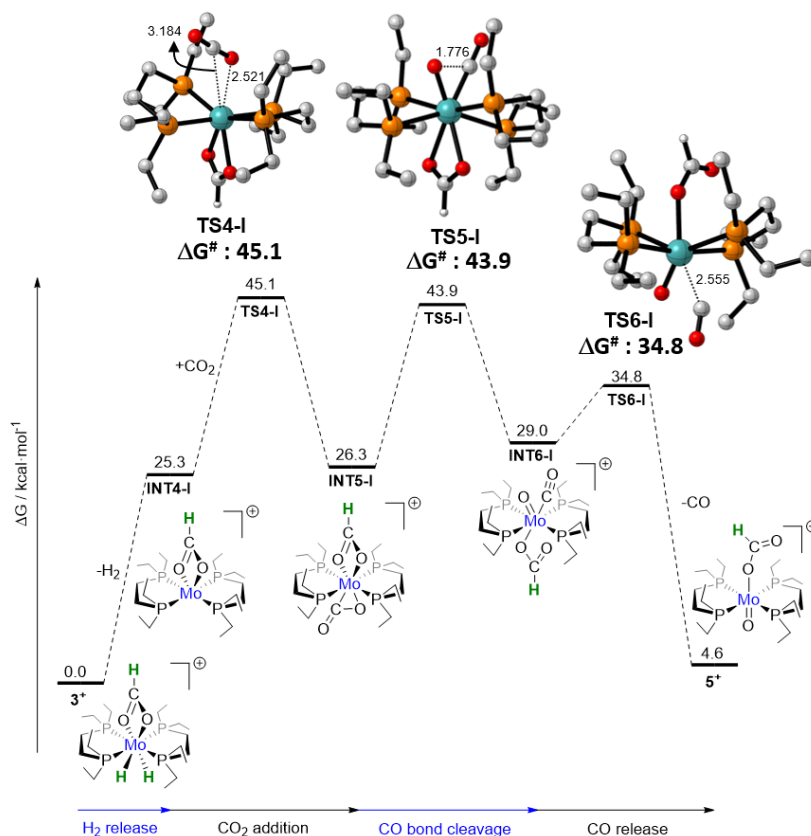
|                 | X-ray value | DFT value | Difference | Relative error (%) |
|-----------------|-------------|-----------|------------|--------------------|
| <i>Mo-O1</i>    | 2.254       | 2.282     | 0.028      | 1.2                |
| <i>Mo-O2</i>    | 2.274       | 2.284     | 0.010      | 0.4                |
| <i>Mo-P1</i>    | 2.463       | 2.499     | 0.036      | 1.5                |
| <i>Mo-P2</i>    | 2.414       | 2.441     | 0.027      | 1.1                |
| <i>Mo-P3</i>    | 2.485       | 2.487     | 0.002      | 0.1                |
| <i>Mo-P4</i>    | 2.425       | 2.441     | 0.016      | 0.7                |
| <i>P1-Mo-P2</i> | 81.5        | 81.9      | 0.4        | 0.5                |
| <i>P3-Mo-P4</i> | 82.2        | 82.1      | -0.1       | -0.1               |

**Figure S1.** Energy scan of the H<sub>2</sub> release from **INT1**, leading to **INT1'** + H<sub>2</sub>. The Mo...H distance has been increased from 1.785 to 4.285 Å by step of 0.100 Å. Energy ( $\Delta E$  in kcal·mol<sup>-1</sup>) and distances in angstroms. All data have been computed at the B3LYP-D3(BJ)/SDD+f(Mo), 6-31G\*\* (other atoms) level of theory. Hydrogen atoms in the diphosphine ligands have been omitted for clarity.



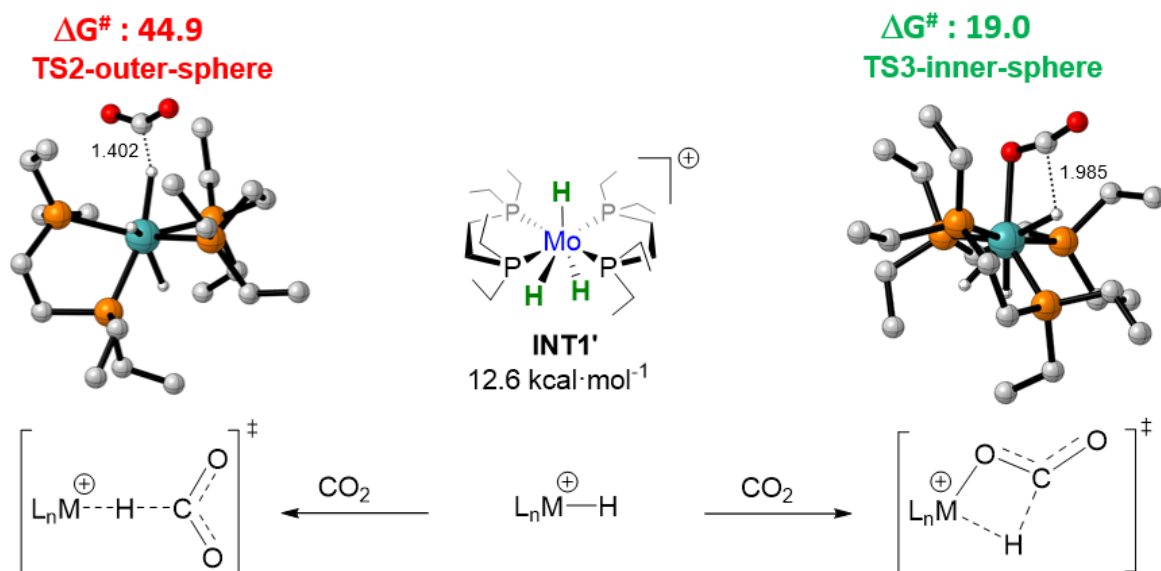
This scan shows that the release of H<sub>2</sub> from **INT1** affording **INT1'** + H<sub>2</sub> is barrierless.

**Figure S2. Pathway I:** Energy profile for the thermal evolution of complex **3<sup>+</sup>** into complex **5<sup>+</sup>**. All data have been computed at the PCM(o-dichlorobenzene)-M06-L-D3/def2-TZVPP//B3LYP-D3(BJ)/SDD+f(Mo), 6-31G\*\* (other atoms) level of theory. Free energies ( $\Delta G$ ) have been corrected at 373.15 K. All activation barriers are referred to **3<sup>+</sup>**. Distances and energies are given in Å and kcal·mol<sup>-1</sup>, respectively.

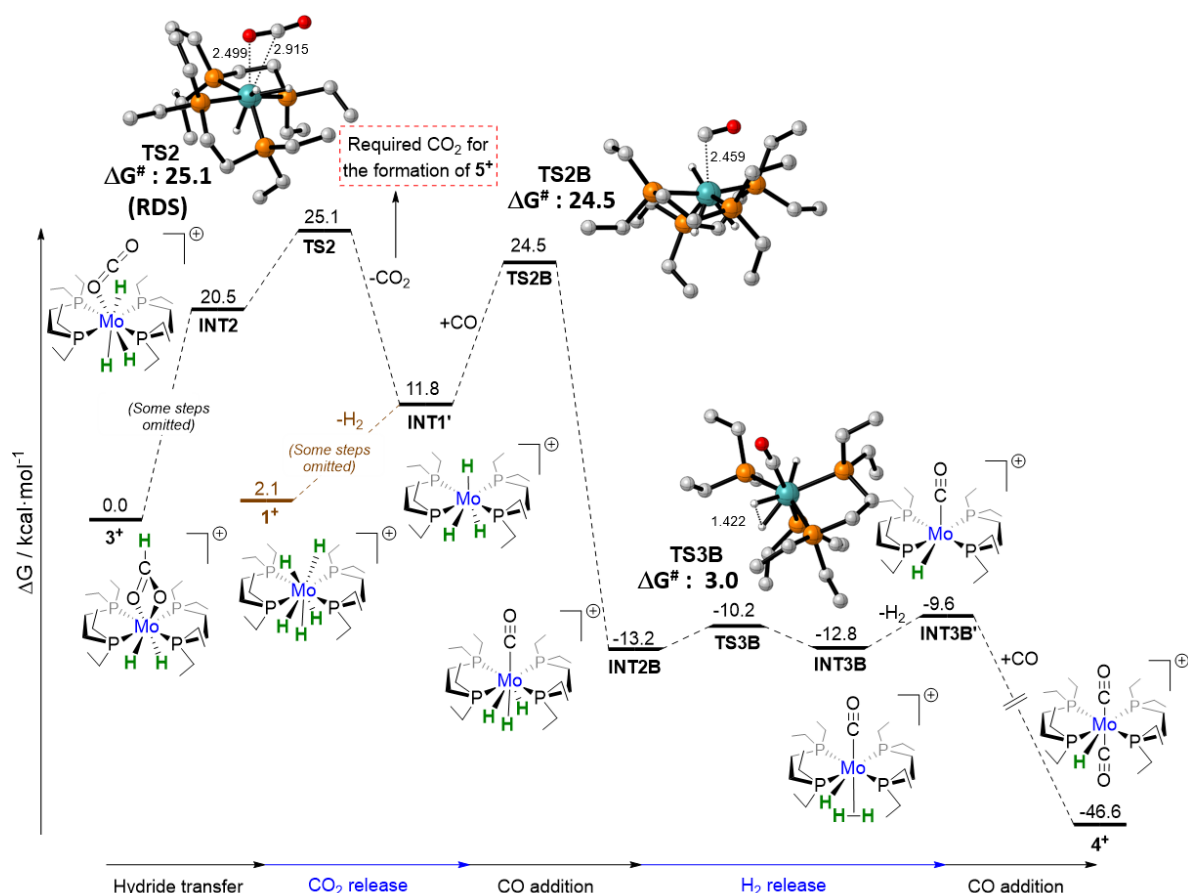




**Scheme S1.** Competitive pathways (outer or inner-sphere mechanisms) for the hydride transfer from the Molybdenum hydride intermediate **INT1'** to the CO<sub>2</sub> molecule. Distances are given in Å. Structures of the TS and Gibbs free energy ( $\Delta G$  in kcal·mol<sup>-1</sup> from initial reactants) have been computed at the PCM(o-dichlorobenzene)-M06-L-D3/def2-TZVPP//B3LYP-D3(BJ)/SDD+f(Mo), 6-31G\*\* (other atoms) level of theory. Hydrogen atoms in the diphosphine ligands have been omitted for clarity.

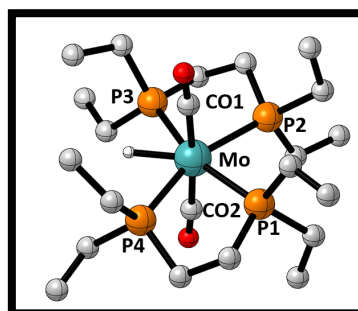


**Figure S3.** Computed profile for the thermal evolution of complex **3<sup>+</sup>** and its reactivity with carbon monoxide, affording complex **4<sup>+</sup>**. All data have been computed at the PCM(*o*-dichlorobenzene)-M06-L-D3/def2-TZVPP//B3LYP-D3(BJ)/SDD+f(Mo), 6-31G\*\* (other atoms) level of theory. Free energies have been corrected at 373.15 K. All activation barriers are referred to **3<sup>+</sup>**. Distances and energies are given in Å and kcal·mol<sup>-1</sup>, respectively.



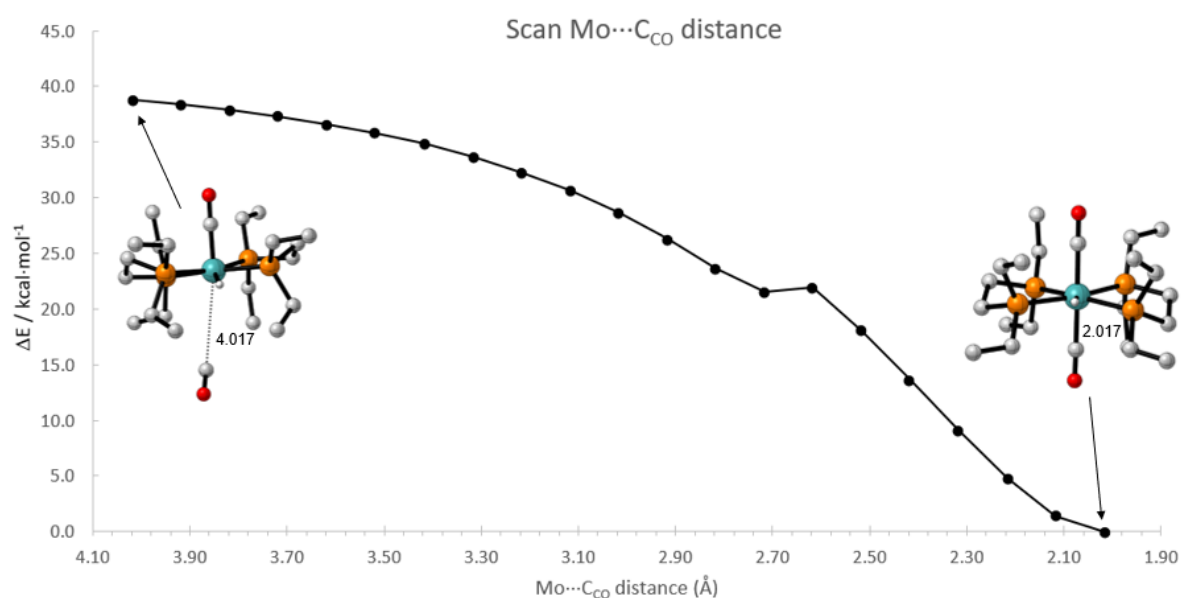
We investigated two processes forming complex **4<sup>+</sup>**: (i) the direct reaction of complex **1<sup>+</sup>** with CO at room temperature (25°C) and (ii) the thermal evolution of complex **3<sup>+</sup>**, when heated at 100 °C. Note that these two processes involve the same intermediate **INT1'** and share the subsequent steps. In the first process, no transition state associated to the direct addition of CO was found on the potential energy surface due to the high coordination number of **1<sup>+</sup>**. As a result, the initial step involves the release of H<sub>2</sub> forming **INT1'**. This step has been omitted in this profile for clarity and has been described in the Fig. 2 of the manuscript. The second process, associated to the thermal evolution, starts by the formation of **INT2** which has been previously described in the main text in Fig. 3 and has been omitted in this profile for clarity, as well. The transition state from **INT2** to **INT1'**, associated to release of CO<sub>2</sub> (**TS2**), was computed in Fig. 2 but at 25°C. In this new profile, we corrected the energies at 373.15 K using GoodVibes software. It lies at 25.1 kcal·mol<sup>-1</sup> from **3<sup>+</sup>**. It is important to mention that this release of CO<sub>2</sub> is crucial for the last step of the profile computed for the thermal evolution of **3<sup>+</sup>** into **5<sup>+</sup>** (see Fig.3 in the main text). Once **INT1'** is formed, the rest of the profile is shared between the two processes. The subsequent addition of CO occurs through **TS2B**, which lies at 12.7 kcal·mol<sup>-1</sup> from **INT1'** (ΔG<sup>#</sup>: 24.5 kcal·mol<sup>-1</sup> from the free complex **3<sup>+</sup>**). Remarkably, this addition is highly exergonic, ΔG: -25.0 kcal·mol<sup>-1</sup>, affording **INT2B**, where CO is coordinated to the metal centre by the carbon atom. Afterwards, in order to add the second molecule of CO in *trans* to the first one, the release of a second molecule of H<sub>2</sub> is required by a new reductive elimination process, giving the Mo(II) intermediate **INT3B'**. This H<sub>2</sub> release is even energetically easier than the first one (**TS3B** with ΔG<sup>#</sup>: 3.0 kcal·mol<sup>-1</sup>). Then, the coordination of the second molecule of CO was found to be barrierless according to an energy scan where the CO was located at 4.0 Å and gradually approached to the Mo centre until forming the final product **4<sup>+</sup>** (Figure S4). In line with experimental observations, the formation of **4<sup>+</sup>** (see Table S2 for geometrical features) from **3<sup>+</sup>** after successive releases of H<sub>2</sub> and additions of CO is strongly exergonic, with an overall ΔG value of up to -46.6 kcal·mol<sup>-1</sup> (-49.8 kcal·mol<sup>-1</sup> from complex **1<sup>+</sup>**). The rate-determining step corresponds to CO<sub>2</sub> release with an activation barrier of 25.1 kcal·mol<sup>-1</sup>, further confirming the feasibility of the reaction.

**Table S2.** Comparison of the X-ray diffraction structure of naked cationic **4<sup>+</sup>** *versus* the DFT-computed structure. Distances and bond angles are given in angstroms and °, respectively. The image shows the DFT-optimized geometry at B3LYP-D3(BJ)/SDD+f(Mo), 6-31G\*\* (other atoms) level of theory. Hydrogen atoms in the diphosphine ligands have been omitted for clarity.



|                 | X-ray value | DFT value | Difference | Relative error (%) |
|-----------------|-------------|-----------|------------|--------------------|
| <b>Mo-CO1</b>   | 2.015       | 2.017     | 0.002      | 0.1                |
| <b>Mo-CO2</b>   | 2.009       | 2.006     | -0.003     | -0.1               |
| <b>Mo-P1</b>    | 2.451       | 2.468     | 0.017      | 0.7                |
| <b>Mo-P2</b>    | 2.531       | 2.555     | 0.024      | 0.9                |
| <b>Mo-P3</b>    | 2.451       | 2.477     | 0.026      | 1.1                |
| <b>Mo-P4</b>    | 2.526       | 2.551     | 0.025      | 1.0                |
| <b>P1-Mo-P2</b> | 77.2        | 77.5      | 0.3        | 0.4                |
| <b>P3-Mo-P4</b> | 77.5        | 78.0      | 0.5        | 0.6                |

**Figure S4.** Energy scan of the CO addition to **INT3B<sup>+</sup>**. The Mo...C<sub>CO</sub> distance has been decrease from 4.017 to 2.017 Å by step of 0.100 Å. Energy ( $\Delta E$ ) and distances are given in kcal·mol<sup>-1</sup> and angstroms. All data have been computed at the B3LYP-D3(BJ)/SDD+f(Mo), 6-31G\*\* (other atoms) level of theory. Hydrogen atoms in the diphosphine ligands have been omitted for clarity.



This scan shows that the addition of CO on Mo centre is barrierless.

### III.3. Cartesian Coordinates and E+ZPE and free energies (G) of optimized structures

|                |           |           |           |
|----------------|-----------|-----------|-----------|
| 78             |           |           |           |
| 1 <sup>+</sup> |           |           |           |
| C              | -3.217456 | 1.206178  | 0.857417  |
| H              | -3.338445 | 0.631257  | 1.780648  |
| H              | -3.882919 | 2.071919  | 0.922112  |
| C              | -3.544541 | 0.337127  | -0.358677 |
| H              | -3.499656 | 0.927059  | -1.280209 |
| H              | -4.552647 | -0.081071 | -0.290618 |
| C              | -2.557665 | -1.639238 | -2.239516 |
| H              | -3.629815 | -1.821148 | -2.370923 |
| H              | -2.279944 | -0.813241 | -2.901606 |
| C              | -1.746438 | -2.891329 | -2.579625 |
| H              | -0.676155 | -2.697612 | -2.477425 |
| H              | -1.943207 | -3.201368 | -3.609370 |
| H              | -2.006416 | -3.730757 | -1.927378 |
| C              | -2.927375 | -2.360400 | 0.581168  |
| H              | -2.154790 | -3.135189 | 0.592752  |
| H              | -2.931817 | -1.934391 | 1.589920  |
| C              | -4.290581 | -2.953270 | 0.219435  |
| H              | -5.078831 | -2.195256 | 0.197981  |
| H              | -4.581544 | -3.702450 | 0.961051  |
| H              | -4.272468 | -3.447007 | -0.755735 |
| C              | -1.518853 | 3.214247  | -0.354383 |
| H              | -0.488071 | 3.484304  | -0.599739 |
| H              | -1.946800 | 2.822059  | -1.283421 |
| C              | -2.289656 | 4.443872  | 0.131929  |
| H              | -1.805654 | 4.910703  | 0.993395  |
| H              | -3.317532 | 4.199671  | 0.415079  |
| H              | -2.341023 | 5.193565  | -0.662673 |
| C              | -1.140238 | 2.491575  | 2.457040  |
| H              | -0.188366 | 3.029342  | 2.385094  |
| H              | -1.918853 | 3.238935  | 2.640863  |
| C              | -1.089767 | 1.477474  | 3.601851  |
| H              | -0.895817 | 1.991585  | 4.547223  |
| H              | -0.303648 | 0.738374  | 3.437245  |
| H              | -2.036841 | 0.940327  | 3.707997  |
| C              | 3.509016  | -0.522970 | 0.382185  |
| H              | 3.616177  | 0.110446  | 1.269294  |
| H              | 4.381836  | -1.180820 | 0.340775  |
| C              | 3.400318  | 0.332119  | -0.882639 |
| H              | 3.372388  | -0.304499 | -1.772239 |
| H              | 4.258293  | 1.002513  | -0.987431 |
| C              | 2.253236  | 2.760976  | 0.199109  |
| H              | 1.322402  | 3.292972  | 0.413889  |
| H              | 4.241726  | 3.223234  | -0.595529 |
| H              | 2.954693  | 4.251918  | -1.236696 |
| C              | 2.310631  | -3.036281 | -0.395059 |
| H              | 2.337353  | -2.713985 | -1.440529 |
| H              | 3.316492  | -3.380689 | -0.132438 |
| C              | 1.295754  | -4.166379 | -0.211899 |
| H              | 1.263691  | -4.515824 | 0.824174  |
| H              | 1.563196  | -5.019416 | -0.841352 |
| H              | 0.290496  | -3.844102 | -0.492890 |
| C              | 2.010766  | -2.082937 | 2.360223  |
| H              | 1.910926  | -1.172239 | 2.959415  |
| H              | 1.091610  | -2.653110 | 2.526683  |
| C              | 3.247769  | -2.887457 | 2.768031  |
| H              | 3.197062  | -3.141536 | 3.830392  |
| H              | 4.170946  | -2.322560 | 2.609680  |
| H              | 3.325837  | -3.824712 | 2.210644  |
| Mo             | -0.026324 | -0.174754 | 0.004373  |
| H              | 0.765797  | 0.506106  | 1.357972  |
| H              | -0.638942 | 0.611695  | -1.385860 |
| H              | -0.230567 | -1.842747 | 0.039516  |
| H              | -0.700203 | -0.712078 | 1.489221  |
| H              | 0.499473  | -0.920400 | -1.447675 |
| P              | -1.443201 | 1.751938  | 0.789737  |
| P              | -2.287055 | -1.019411 | -0.522825 |
| P              | 1.955752  | -1.519729 | 0.603655  |

|   |          |          |           |
|---|----------|----------|-----------|
| P | 1.815904 | 1.301253 | -0.864216 |
| C | 1.702674 | 1.999166 | -2.572787 |
| H | 0.919526 | 2.764722 | -2.540233 |
| H | 2.645108 | 2.512506 | -2.789537 |
| C | 1.390715 | 0.969215 | -3.660646 |
| H | 2.166878 | 0.200841 | -3.724438 |
| H | 1.334656 | 1.461636 | -4.635349 |
| H | 0.440629 | 0.468470 | -3.465834 |
| H | 2.569674 | 2.327973 | 1.154341  |
| C | 3.306303 | 3.730628 | -0.342887 |
| H | 3.541063 | 4.489103 | 0.409230  |

Sum of electronic and zero-point Energies= -2227.173063  
Sum of electronic and thermal Free Energies= -2227.242084

|                |           |           |           |
|----------------|-----------|-----------|-----------|
| 79             |           |           |           |
| 3 <sup>+</sup> |           |           |           |
| C              | -0.131195 | -0.169060 | -2.829063 |
| H              | -0.195128 | -0.206088 | -3.924935 |
| C              | 2.476078  | -2.587895 | -0.013681 |
| H              | 2.163895  | -3.154356 | -0.897864 |
| H              | 3.509161  | -2.875021 | 0.200645  |
| C              | 1.556175  | -2.904989 | 1.171507  |
| H              | 1.914781  | -2.405237 | 2.077411  |
| H              | 1.532551  | -3.980303 | 1.376075  |
| C              | -1.286889 | 2.849799  | 1.230662  |
| H              | -1.550677 | 2.370776  | 2.179515  |
| H              | -1.219392 | 3.925404  | 1.416784  |
| C              | -2.335805 | 2.536776  | 0.160677  |
| H              | -2.108480 | 3.086858  | -0.759465 |
| H              | -3.332255 | 2.853788  | 0.481034  |
| C              | 3.131148  | -0.769110 | -2.177953 |
| H              | 4.037403  | -1.383889 | -2.161039 |
| H              | 2.421164  | -1.274640 | -2.840187 |
| C              | 3.436534  | 0.645049  | -2.684933 |
| H              | 2.537977  | 1.266259  | -2.682266 |
| H              | 3.821574  | 0.607313  | -3.707676 |
| H              | 4.196118  | 1.135839  | -2.069351 |
| C              | 3.654188  | 0.035357  | 0.573865  |
| H              | 3.599779  | 1.102370  | 0.334435  |
| H              | 3.296863  | -0.062059 | 1.604425  |
| C              | 5.093497  | -0.467772 | 0.444511  |
| H              | 5.180878  | -1.526922 | 0.701143  |
| H              | 5.748794  | 0.086087  | 1.122830  |
| H              | 5.481427  | -0.334676 | -0.568594 |
| C              | -1.062288 | -2.588689 | 2.418769  |
| H              | -2.126289 | -2.448735 | 2.206134  |
| H              | -0.922050 | -3.646861 | 2.669306  |
| C              | -0.640783 | -1.693865 | 3.584498  |
| H              | 0.420674  | -1.807613 | 3.823394  |
| H              | -1.208928 | -1.950510 | 4.482696  |
| H              | -0.820955 | -0.641508 | 3.355144  |
| C              | -0.825086 | -3.605495 | -0.282068 |
| H              | -0.269463 | -3.530062 | -1.219815 |
| H              | -0.572304 | -4.566194 | 0.181837  |
| C              | -2.325534 | -3.515625 | -0.561140 |
| H              | -2.918862 | -3.633258 | 0.350342  |
| H              | -2.628244 | -4.307099 | -1.252266 |
| H              | -2.577172 | -2.559747 | -1.020370 |
| C              | -3.579575 | -0.037259 | 0.848831  |
| H              | -3.572476 | -1.109068 | 0.628956  |
| H              | -3.140977 | 0.068856  | 1.847340  |
| C              | -5.008278 | 0.509335  | 0.816787  |
| H              | -5.044162 | 1.578362  | 1.043922  |
| H              | -5.624038 | -0.001963 | 1.562260  |
| H              | -5.479063 | 0.357348  | -0.157757 |
| C              | -3.250886 | 0.829056  | -1.934649 |
| H              | -4.113729 | 1.489676  | -1.796282 |
| H              | -2.569853 | 1.341125  | -2.621784 |
| C              | -3.690473 | -0.517703 | -2.515034 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | -2.835767 | -1.173182 | -2.693637 |
| H  | -4.207759 | -0.368091 | -3.466888 |
| H  | -4.382162 | -1.038730 | -1.846615 |
| C  | 1.432348  | 2.509902  | 2.195185  |
| H  | 2.468912  | 2.402730  | 1.861172  |
| H  | 1.289720  | 3.563831  | 2.455714  |
| C  | 1.171518  | 1.622527  | 3.412471  |
| H  | 0.139675  | 1.710425  | 3.765035  |
| H  | 1.827065  | 1.911592  | 4.238389  |
| H  | 1.354573  | 0.570988  | 3.181690  |
| C  | 1.003655  | 3.332652  | -0.557243 |
| H  | 2.025838  | 3.000360  | -0.770514 |
| H  | 0.436279  | 3.135790  | -1.469635 |
| C  | 0.978360  | 4.821357  | -0.199535 |
| H  | -0.041305 | 5.185111  | -0.046371 |
| H  | 1.411617  | 5.405532  | -1.016358 |
| H  | 1.555497  | 5.041799  | 0.702333  |
| O  | -0.302202 | -1.220355 | -2.145050 |
| O  | 0.111749  | 0.928106  | -2.243331 |
| P  | 2.374781  | -0.785979 | -0.488394 |
| P  | -0.167132 | -2.280107 | 0.832636  |
| P  | -2.353529 | 0.727261  | -0.314127 |
| P  | 0.364737  | 2.158725  | 0.726955  |
| Mo | 0.010307  | -0.073202 | -0.195286 |
| H  | 0.905283  | -0.211920 | 1.212548  |
| H  | -0.788369 | 0.107410  | 1.265374  |

Sum of electronic and zero-point Energies=-2414.607735  
Sum of electronic and thermal Free Energies= -2414.677985

78

4<sup>+</sup>

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -0.032575 | 0.242372  | 2.040087  |
| C | 0.206231  | 0.279767  | -1.972385 |
| C | -3.180046 | 1.153404  | -1.066960 |
| H | -4.006943 | 1.868963  | -1.101403 |
| H | -2.904509 | 0.907669  | -2.097460 |
| C | -3.572246 | -0.102426 | -0.291047 |
| H | -4.339180 | -0.678267 | -0.816610 |
| H | -3.992727 | 0.173225  | 0.681217  |
| C | -2.332529 | -2.571442 | -1.176014 |
| H | -1.549338 | -3.314920 | -0.998244 |
| H | -3.284166 | -3.058612 | -0.942948 |
| C | -2.322389 | -2.114603 | -2.637140 |
| H | -3.144891 | -1.423855 | -2.843820 |
| H | -2.446305 | -2.973514 | -3.302390 |
| H | -1.391826 | -1.611518 | -2.904641 |
| C | -2.550837 | -1.963535 | 1.669397  |
| H | -1.744546 | -2.646243 | 1.946044  |
| H | -2.507100 | -1.143654 | 2.394425  |
| C | -1.242785 | 3.313202  | -1.393457 |
| H | -0.371234 | 3.798993  | -0.943048 |
| H | -0.887789 | 2.854487  | -2.322119 |
| C | -2.351180 | 4.331001  | -1.673729 |
| H | -3.218669 | 3.867991  | -2.151604 |
| H | -1.982751 | 5.108200  | -2.348923 |
| H | -2.690753 | 4.825421  | -0.759559 |
| C | -2.400809 | 2.711698  | 1.231239  |
| H | -3.275427 | 3.291300  | 0.916690  |
| H | -2.765239 | 1.902431  | 1.872530  |
| C | -1.419804 | 3.592114  | 2.007437  |
| H | -0.588492 | 3.004346  | 2.399292  |
| H | -1.924105 | 4.061494  | 2.856197  |
| H | -1.010586 | 4.391258  | 1.382484  |
| C | 3.572602  | -0.582175 | -0.278810 |
| H | 3.595376  | -0.734443 | -1.363360 |
| H | 4.593085  | -0.352356 | 0.040246  |
| C | 3.033027  | -1.820753 | 0.432176  |
| H | 3.086680  | -1.689387 | 1.516923  |
| H | 3.610129  | -2.716356 | 0.184339  |
| C | 1.351220  | -2.823483 | -1.706187 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 0.330122  | -2.882161 | -2.090034 |
| H  | 1.847811  | -2.066923 | -2.322800 |
| C  | 2.052470  | -4.177166 | -1.841420 |
| H  | 3.067022  | -4.160802 | -1.433239 |
| H  | 2.132030  | -4.450294 | -2.897403 |
| H  | 1.499544  | -4.974593 | -1.338716 |
| C  | 0.772711  | -3.496378 | 1.081156  |
| H  | -0.242238 | -3.797392 | 0.802344  |
| H  | 1.424460  | -4.339057 | 0.831728  |
| C  | 0.856010  | -3.195112 | 2.580868  |
| H  | 1.888078  | -3.022759 | 2.896743  |
| H  | 0.479802  | -4.045151 | 3.156763  |
| H  | 0.273936  | -2.314896 | 2.858465  |
| C  | 2.948904  | 2.111272  | -1.198707 |
| H  | 2.624730  | 1.713769  | -2.165986 |
| H  | 2.332134  | 2.996484  | -1.014027 |
| C  | 4.436081  | 2.473522  | -1.226915 |
| H  | 4.767823  | 2.906332  | -0.279272 |
| H  | 4.627543  | 3.212048  | -2.010211 |
| H  | 5.062655  | 1.602657  | -1.438333 |
| C  | 3.113104  | 1.508508  | 1.662222  |
| H  | 2.851128  | 0.774366  | 2.430848  |
| H  | 4.205703  | 1.525014  | 1.590392  |
| C  | 2.581094  | 2.891145  | 2.044095  |
| H  | 2.893830  | 3.653347  | 1.324661  |
| H  | 2.961431  | 3.185745  | 3.025600  |
| H  | 1.490977  | 2.898896  | 2.089368  |
| C  | -3.904307 | -2.676424 | 1.742363  |
| H  | -4.096940 | -3.004009 | 2.767781  |
| H  | -3.933576 | -3.564621 | 1.106297  |
| H  | -4.729784 | -2.022784 | 1.447378  |
| Mo | 0.060340  | 0.187512  | 0.025991  |
| O  | -0.081466 | 0.326434  | 3.198658  |
| O  | 0.320518  | 0.388776  | -3.127565 |
| P  | -2.089672 | -1.193037 | 0.044352  |
| P  | -1.682463 | 1.908875  | -0.274731 |
| P  | 1.245229  | -2.071395 | -0.008893 |
| P  | 2.445058  | 0.857405  | 0.062123  |
| H  | 0.480407  | 1.832189  | 0.194658  |

Sum of electronic and zero-point Energies= -2451.553012  
Sum of electronic and thermal Free Energies= -2451.625804

78

5<sup>+</sup>

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 0.305086  | -0.410821 | -2.911757 |
| H | 0.075691  | 0.061906  | -3.884446 |
| C | 1.250784  | -3.265513 | -0.324971 |
| H | 1.144352  | -3.277960 | -1.411294 |
| H | 1.984776  | -4.027065 | -0.046890 |
| C | -0.091577 | -3.522461 | 0.364519  |
| H | 0.038671  | -3.587968 | 1.449849  |
| H | -0.525908 | -4.471961 | 0.036507  |
| C | 0.051610  | 3.609586  | 0.009700  |
| H | -0.140307 | 3.818535  | 1.067220  |
| H | 0.498672  | 4.511117  | -0.420394 |
| C | -1.250973 | 3.243683  | -0.707597 |
| H | -1.071172 | 3.086136  | -1.776269 |
| H | -1.990558 | 4.044699  | -0.620167 |
| C | 3.437022  | -1.460760 | -0.931799 |
| H | 3.916800  | -2.444967 | -0.895312 |
| H | 3.097239  | -1.311058 | -1.958437 |
| C | 4.423930  | -0.375817 | -0.498563 |
| H | 3.977061  | 0.618951  | -0.549180 |
| H | 5.294683  | -0.376362 | -1.159758 |
| H | 4.785231  | -0.529860 | 0.521926  |
| C | 2.518421  | -1.750235 | 1.839832  |
| H | 2.936830  | -0.775132 | 2.109584  |
| H | 1.631248  | -1.868758 | 2.468332  |
| C | 3.531390  | -2.868476 | 2.093259  |
| H | 3.105500  | -3.855879 | 1.895720  |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 3.849503  | -2.854549 | 3.139651  |
| H  | 4.427253  | -2.760265 | 1.475946  |
| C  | -2.647855 | -2.486264 | 1.262817  |
| H  | -3.519188 | -1.900756 | 0.955934  |
| H  | -2.921430 | -3.541962 | 1.161375  |
| C  | -2.277220 | -2.158433 | 2.712377  |
| H  | -1.421432 | -2.747426 | 3.055020  |
| H  | -3.117378 | -2.387979 | 3.373334  |
| H  | -2.016929 | -1.104773 | 2.829826  |
| C  | -1.983570 | -2.519605 | -1.624003 |
| H  | -2.520274 | -1.627400 | -1.957499 |
| H  | -1.113224 | -2.611663 | -2.279300 |
| C  | -2.876296 | -3.760175 | -1.710257 |
| H  | -2.364842 | -4.659069 | -1.353988 |
| H  | -3.161619 | -3.939801 | -2.750387 |
| H  | -3.795576 | -3.642860 | -1.130631 |
| C  | -2.755518 | 2.071184  | 1.544685  |
| H  | -3.230441 | 1.147531  | 1.892387  |
| H  | -1.936920 | 2.259909  | 2.246515  |
| C  | -3.753716 | 3.230860  | 1.525298  |
| H  | -3.272388 | 4.175550  | 1.258218  |
| H  | -4.199093 | 3.360293  | 2.515539  |
| H  | -4.568740 | 3.057327  | 0.817088  |
| C  | -3.328989 | 1.351337  | -1.272372 |
| H  | -3.776817 | 2.321910  | -1.511961 |
| H  | -2.838828 | 0.989590  | -2.181930 |
| C  | -4.405019 | 0.371171  | -0.803351 |
| H  | -3.983228 | -0.610812 | -0.583414 |
| H  | -5.160165 | 0.238147  | -1.582570 |
| H  | -4.916534 | 0.726360  | 0.095204  |
| C  | 2.545782  | 2.644467  | 1.178961  |
| H  | 3.397452  | 1.978693  | 1.011556  |
| H  | 2.886174  | 3.662319  | 0.961289  |
| C  | 2.062148  | 2.530434  | 2.628519  |
| H  | 1.237855  | 3.219564  | 2.834826  |
| H  | 2.876028  | 2.782452  | 3.313585  |
| H  | 1.713887  | 1.520947  | 2.855897  |
| C  | 2.037680  | 2.388421  | -1.733042 |
| H  | 2.557395  | 1.452977  | -1.955831 |
| H  | 1.201592  | 2.436764  | -2.436551 |
| C  | 2.973306  | 3.588168  | -1.898012 |
| H  | 2.480137  | 4.532220  | -1.648183 |
| H  | 3.308464  | 3.659631  | -2.936332 |
| H  | 3.863254  | 3.496903  | -1.269997 |
| O  | 0.966727  | -1.443618 | -2.863698 |
| O  | -0.194442 | 0.226857  | -1.895898 |
| P  | 1.903264  | -1.578162 | 0.093915  |
| P  | -1.294213 | -2.136822 | 0.048535  |
| P  | -1.958211 | 1.645570  | -0.068267 |
| P  | 1.256509  | 2.191278  | -0.066220 |
| Mo | -0.039036 | 0.050802  | 0.259591  |
| O  | -0.058787 | 0.102976  | 1.943430  |

Sum of electronic and zero-point Energies= -2488.709075  
Sum of electronic and thermal Free Energies= -2488.783167

# 78 INT1

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -3.406096 | 0.429692  | 0.790225  |
| H | -3.449314 | -0.170340 | 1.705529  |
| H | -4.266315 | 1.104784  | 0.806778  |
| C | -3.439209 | -0.478951 | -0.441860 |
| H | -3.463988 | 0.119186  | -1.359511 |
| H | -4.331470 | -1.111161 | -0.446526 |
| C | -1.955477 | -2.176275 | -2.252585 |
| H | -2.967716 | -2.552086 | -2.436267 |
| H | -1.806343 | -1.310988 | -2.905964 |
| C | -0.912668 | -3.254920 | -2.549748 |
| H | 0.097569  | -2.872688 | -2.385224 |
| H | -0.991492 | -3.580605 | -3.590425 |
| H | -1.050002 | -4.136451 | -1.916492 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| C  | -2.291994 | -2.983564 | 0.546798  |
| H  | -1.377530 | -3.583675 | 0.590585  |
| H  | -2.439873 | -2.584507 | 1.556622  |
| C  | -3.482814 | -3.846667 | 0.123732  |
| H  | -4.412041 | -3.272014 | 0.078807  |
| H  | -3.634861 | -4.655092 | 0.844399  |
| H  | -3.323019 | -4.305598 | -0.855304 |
| C  | -2.159567 | 2.825698  | -0.264486 |
| H  | -1.226995 | 3.391285  | -0.347491 |
| H  | -2.334395 | 2.389557  | -1.254277 |
| C  | -3.305318 | 3.760351  | 0.131835  |
| H  | -3.101920 | 4.272956  | 1.075250  |
| H  | -4.254321 | 3.227340  | 0.235839  |
| H  | -3.446732 | 4.527785  | -0.634598 |
| C  | -1.807864 | 2.130580  | 2.543891  |
| H  | -1.043198 | 2.915333  | 2.532268  |
| H  | -2.773819 | 2.623018  | 2.695879  |
| C  | -1.527658 | 1.138739  | 3.674449  |
| H  | -1.569287 | 1.647778  | 4.641360  |
| H  | -0.537861 | 0.690714  | 3.563314  |
| H  | -2.266369 | 0.331258  | 3.697937  |
| C  | 3.587676  | 0.236651  | -0.020501 |
| H  | 3.756299  | 0.917281  | 0.820785  |
| H  | 4.538464  | -0.259842 | -0.234186 |
| C  | 3.074494  | 1.004466  | -1.238830 |
| H  | 3.004172  | 0.340093  | -2.105275 |
| H  | 3.746770  | 1.824829  | -1.507342 |
| C  | 1.677394  | 3.151247  | 0.126272  |
| H  | 0.706152  | 3.482494  | 0.504987  |
| H  | 3.388619  | 3.998558  | -0.946352 |
| H  | 1.845159  | 4.768586  | -1.334983 |
| C  | 2.831241  | -2.549347 | -0.362133 |
| H  | 2.707737  | -2.334508 | -1.428285 |
| H  | 3.897743  | -2.720652 | -0.182662 |
| C  | 2.011155  | -3.778204 | 0.033940  |
| H  | 2.146995  | -4.028086 | 1.090641  |
| H  | 2.315904  | -4.648507 | -0.553469 |
| H  | 0.946972  | -3.605079 | -0.141800 |
| C  | 2.787062  | -1.342759 | 2.296760  |
| H  | 2.591905  | -0.404952 | 2.826796  |
| H  | 2.061359  | -2.065970 | 2.683889  |
| C  | 4.218145  | -1.829981 | 2.538401  |
| H  | 4.400769  | -1.942972 | 3.610670  |
| H  | 4.958914  | -1.124037 | 2.152136  |
| H  | 4.401392  | -2.801228 | 2.071303  |
| Mo | 0.039833  | -0.146983 | 0.182230  |
| H  | 0.748100  | 0.746141  | 1.493542  |
| H  | -0.707102 | 0.331316  | -1.270236 |
| H  | 0.002099  | -1.556929 | 1.276740  |
| H  | -0.496561 | -0.911374 | 1.711294  |
| H  | 0.587658  | -1.108336 | -1.119378 |
| P  | -1.804881 | 1.381543  | 0.852094  |
| P  | -1.898389 | -1.520463 | -0.527957 |
| P  | 2.323827  | -1.012901 | 0.533450  |
| P  | 1.358610  | 1.643790  | -0.912419 |
| C  | 0.855375  | 2.323537  | -2.559784 |
| H  | -0.065012 | 2.893116  | -2.395612 |
| H  | 1.622044  | 3.037150  | -2.877549 |
| C  | 0.637433  | 1.261735  | -3.639945 |
| H  | 1.559379  | 0.717936  | -3.864781 |
| H  | 0.299407  | 1.732477  | -4.567177 |
| H  | -0.113012 | 0.531370  | -3.328679 |
| H  | 2.215712  | 2.778153  | 1.003903  |
| C  | 2.427069  | 4.312694  | -0.529986 |
| H  | 2.632484  | 5.092613  | 0.208735  |

Sum of electronic and zero-point Energies= -2227.166127  
Sum of electronic and thermal Free Energies= -2227.235301

76

**INT1'**

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 2.236371  | -2.784650 | -0.200479 |
| H | 1.931110  | -3.333480 | -1.099014 |
| H | 3.255973  | -3.105563 | 0.030723  |
| C | 1.283402  | -3.102661 | 0.958337  |
| H | 1.658829  | -2.667650 | 1.890069  |
| H | 1.196167  | -4.183961 | 1.109874  |
| C | -1.014843 | 3.099773  | 0.865048  |
| H | -1.411344 | 2.775367  | 1.832509  |
| H | -0.794840 | 4.168533  | 0.945844  |
| C | -2.032499 | 2.827853  | -0.246146 |
| H | -1.684198 | 3.259074  | -1.190754 |
| H | -2.997885 | 3.288854  | -0.017915 |
| C | 2.820798  | -1.005359 | -2.405927 |
| H | 3.710147  | -1.643457 | -2.448291 |
| H | 2.044665  | -1.502843 | -2.998943 |
| C | 3.111613  | 0.384460  | -2.976927 |
| H | 2.220276  | 1.015577  | -2.932988 |
| H | 3.429181  | 0.310215  | -4.020441 |
| H | 3.912459  | 0.886219  | -2.425011 |
| C | 3.607420  | -0.236058 | 0.308978  |
| H | 3.571983  | 0.842956  | 0.129425  |
| H | 3.341575  | -0.377126 | 1.362060  |
| C | 5.005861  | -0.782224 | 0.014923  |
| H | 5.069212  | -1.860817 | 0.182949  |
| H | 5.740620  | -0.307413 | 0.671444  |
| H | 5.310255  | -0.584557 | -1.016020 |
| C | -1.360203 | -2.779611 | 2.160055  |
| H | -2.405249 | -2.541769 | 1.940606  |
| H | -1.296065 | -3.863488 | 2.313341  |
| C | -0.903523 | -2.019867 | 3.408061  |
| H | 0.136225  | -2.243833 | 3.663839  |
| H | -1.520368 | -2.298424 | 4.266639  |
| H | -0.990111 | -0.941248 | 3.257632  |
| C | -1.081373 | -3.477288 | -0.673674 |
| H | -0.660510 | -3.150903 | -1.632927 |
| H | -0.693429 | -4.486540 | -0.493211 |
| C | -2.611311 | -3.493473 | -0.745331 |
| H | -3.046208 | -3.968165 | 0.137343  |
| H | -2.946135 | -4.054971 | -1.621650 |
| H | -3.017823 | -2.482561 | -0.822117 |
| C | -3.673857 | 0.504215  | 0.501737  |
| H | -3.730382 | -0.587969 | 0.445559  |
| H | -3.365559 | 0.731383  | 1.528778  |
| C | -5.029065 | 1.136287  | 0.178500  |
| H | -4.988418 | 2.229146  | 0.204212  |
| H | -5.779352 | 0.821157  | 0.909232  |
| H | -5.388532 | 0.838112  | -0.810055 |
| C | -2.890184 | 0.985090  | -2.291070 |
| H | -3.772100 | 1.631961  | -2.343679 |
| H | -2.109850 | 1.446557  | -2.904239 |
| C | -3.213774 | -0.417171 | -2.811150 |
| H | -2.334319 | -1.069529 | -2.774612 |
| H | -3.550938 | -0.375548 | -3.850377 |
| H | -4.006773 | -0.892406 | -2.226542 |
| C | 1.454399  | 2.353792  | 2.186858  |
| H | 2.493130  | 2.058529  | 2.003266  |
| H | 1.463730  | 3.426529  | 2.405533  |
| C | 0.885429  | 1.576300  | 3.374240  |
| H | -0.161418 | 1.832127  | 3.563677  |
| H | 1.450644  | 1.809226  | 4.280965  |
| H | 0.938017  | 0.499705  | 3.203173  |
| C | 1.522971  | 3.182251  | -0.595686 |
| H | 2.383555  | 2.580352  | -0.902590 |
| H | 0.893112  | 3.267334  | -1.486795 |
| C | 1.969611  | 4.559759  | -0.099794 |
| H | 1.127546  | 5.165120  | 0.248303  |
| H | 2.450804  | 5.110407  | -0.913187 |
| H | 2.691265  | 4.485360  | 0.717447  |
| P | 2.200796  | -0.965861 | -0.658892 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| P  | -0.395639 | -2.350899 | 0.644758  |
| P  | -2.233125 | 0.998183  | -0.557625 |
| P  | 0.552548  | 2.129186  | 0.585489  |
| Mo | -0.056304 | -0.073247 | -0.209677 |
| H  | 0.847043  | -0.265052 | 1.196154  |
| H  | -0.877491 | 0.232981  | 1.233658  |
| H  | 0.234378  | 0.925203  | -1.605757 |

Sum of electronic and zero-point Energies= -2225.967024

Sum of electronic and thermal Free Energies= -2226.036119

78

**INT2B**

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 0.210457  | 0.989834  | -2.145953 |
| H | 0.388170  | -1.220893 | -1.707377 |
| C | 2.282915  | -2.763636 | -0.250438 |
| H | 1.986179  | -3.184626 | -1.216846 |
| H | 3.270185  | -3.166435 | -0.011908 |
| C | 1.261509  | -3.122457 | 0.829471  |
| H | 1.575374  | -2.719824 | 1.798268  |
| H | 1.162386  | -4.207046 | 0.942368  |
| C | -1.061019 | 2.588310  | 1.586917  |
| H | -1.168709 | 1.910804  | 2.438868  |
| H | -0.942497 | 3.600253  | 1.985743  |
| C | -2.268652 | 2.480192  | 0.663775  |
| H | -2.200868 | 3.220035  | -0.141190 |
| H | -3.199807 | 2.680318  | 1.200483  |
| C | 3.272901  | -0.733836 | -2.069483 |
| H | 4.111238  | -1.438333 | -2.077870 |
| H | 2.578495  | -1.045259 | -2.855473 |
| C | 3.767832  | 0.692468  | -2.326690 |
| H | 2.947084  | 1.413297  | -2.290183 |
| H | 4.223223  | 0.763433  | -3.317825 |
| H | 4.523014  | 0.996103  | -1.595863 |
| C | 3.619225  | -0.394706 | 0.783259  |
| H | 3.747636  | 0.682419  | 0.648692  |
| H | 3.166566  | -0.528923 | 1.769312  |
| C | 4.985391  | -1.084427 | 0.719548  |
| H | 4.914186  | -2.158240 | 0.909213  |
| H | 5.650266  | -0.664729 | 1.479994  |
| H | 5.468401  | -0.945185 | -0.250838 |
| C | -1.351775 | -2.734414 | 1.990839  |
| H | -2.411366 | -2.594950 | 1.764092  |
| H | -1.215287 | -3.801321 | 2.204142  |
| C | -0.956423 | -1.891829 | 3.202730  |
| H | 0.094891  | -2.032007 | 3.470653  |
| H | -1.557728 | -2.173177 | 4.071753  |
| H | -1.105923 | -0.826686 | 3.010320  |
| C | -1.097251 | -3.669930 | -0.714311 |
| H | -0.453387 | -3.665389 | -1.599278 |
| H | -0.981899 | -4.645399 | -0.227473 |
| C | -2.553115 | -3.447295 | -1.125561 |
| H | -3.228838 | -3.494781 | -0.266200 |
| H | -2.869915 | -4.220284 | -1.830926 |
| H | -2.679741 | -2.478366 | -1.611885 |
| C | -3.542963 | -0.154230 | 0.867417  |
| H | -3.563763 | -1.159464 | 0.437240  |
| H | -3.092306 | -0.253063 | 1.860592  |
| C | -4.963090 | 0.408810  | 0.966039  |
| H | -4.978415 | 1.421220  | 1.379037  |
| H | -5.571308 | -0.219005 | 1.623506  |
| H | -5.455491 | 0.436745  | -0.009026 |
| C | -3.269133 | 1.228637  | -1.686774 |
| H | -4.129269 | 1.835144  | -1.383415 |
| H | -2.610807 | 1.879138  | -2.272227 |
| C | -3.723919 | 0.041460  | -2.535276 |
| H | -2.869235 | -0.530971 | -2.904202 |
| H | -4.289169 | 0.393655  | -3.402404 |
| H | -4.372059 | -0.636787 | -1.972175 |
| C | 1.738809  | 2.210377  | 2.061894  |
| H | 2.727259  | 2.087476  | 1.614884  |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 1.689035  | 3.240530  | 2.428005  |
| C  | 1.539175  | 1.238034  | 3.226328  |
| H  | 0.625859  | 1.463008  | 3.783332  |
| H  | 2.376010  | 1.314466  | 3.925993  |
| H  | 1.467884  | 0.204373  | 2.883148  |
| C  | 0.893122  | 3.523846  | -0.391360 |
| H  | 1.793273  | 3.239961  | -0.947678 |
| H  | 0.090559  | 3.574616  | -1.133837 |
| C  | 1.083469  | 4.886058  | 0.280830  |
| H  | 0.202947  | 5.187732  | 0.854984  |
| H  | 1.252842  | 5.652353  | -0.480612 |
| H  | 1.946166  | 4.896855  | 0.951084  |
| O  | 0.331427  | 1.585354  | -3.130498 |
| P  | 2.352639  | -0.918494 | -0.473027 |
| P  | -0.403519 | -2.389436 | 0.433746  |
| P  | -2.321735 | 0.801640  | -0.151012 |
| P  | 0.488168  | 2.077821  | 0.700989  |
| Mo | -0.034336 | -0.121849 | -0.474577 |
| H  | 0.071569  | -0.224536 | 1.253118  |
| H  | -1.250310 | -0.743983 | -1.465550 |

Sum of electronic and zero-point Energies= -2339.349168  
Sum of electronic and thermal Free Energies= -2339.418681

79

# INT2

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -0.359712 | -0.023570 | 3.444373  |
| H | -0.479267 | -1.527610 | 1.335858  |
| C | -2.021858 | -2.797453 | -0.840524 |
| H | -1.951052 | -3.494021 | 0.001242  |
| H | -2.911163 | -3.077397 | -1.410569 |
| C | -0.760655 | -2.871263 | -1.704190 |
| H | -0.868399 | -2.232837 | -2.586353 |
| H | -0.575722 | -3.892975 | -2.052078 |
| C | 0.748474  | 3.315503  | -0.328588 |
| H | 1.190235  | 3.354665  | -1.329200 |
| H | 0.385084  | 4.321714  | -0.102889 |
| C | 1.782503  | 2.841746  | 0.689240  |
| H | 1.378005  | 2.894551  | 1.706329  |
| H | 2.681838  | 3.463559  | 0.662984  |
| C | -3.376255 | -1.437452 | 1.303130  |
| H | -4.202445 | -2.059845 | 0.944018  |
| H | -2.803873 | -2.043436 | 2.013033  |
| C | -3.909692 | -0.176052 | 1.986154  |
| H | -3.093427 | 0.458611  | 2.342041  |
| H | -4.534269 | -0.435757 | 2.845204  |
| H | -4.524413 | 0.421132  | 1.305228  |
| C | -3.332098 | -0.242259 | -1.340033 |
| H | -3.483249 | 0.774533  | -0.965953 |
| H | -2.744745 | -0.148340 | -2.259441 |
| C | -4.690166 | -0.887460 | -1.629138 |
| H | -4.590474 | -1.905778 | -2.013234 |
| H | -5.228682 | -0.307430 | -2.384319 |
| H | -5.318949 | -0.926107 | -0.736014 |
| C | 2.090334  | -2.335044 | -1.954342 |
| H | 2.985771  | -1.985906 | -1.430552 |
| H | 2.253926  | -3.394737 | -2.182190 |
| C | 1.872162  | -1.544628 | -3.246116 |
| H | 1.074736  | -1.984639 | -3.850937 |
| H | 2.781777  | -1.545531 | -3.853239 |
| H | 1.603081  | -0.507001 | -3.039755 |
| C | 1.035588  | -3.736051 | 0.324233  |
| H | 0.141382  | -3.882720 | 0.937356  |
| H | 1.111744  | -4.594042 | -0.354449 |
| C | 2.267893  | -3.658306 | 1.222120  |
| H | 3.184638  | -3.523553 | 0.640294  |
| H | 2.377563  | -4.582928 | 1.795798  |
| H | 2.181654  | -2.830916 | 1.930248  |
| C | 3.150184  | 1.153132  | -1.221320 |
| H | 3.499396  | 0.137693  | -1.425784 |
| H | 2.422553  | 1.374679  | -2.007471 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| C  | 4.317718  | 2.141467  | -1.262298 |
| H  | 3.981733  | 3.175419  | -1.145420 |
| H  | 4.836023  | 2.074239  | -2.223282 |
| H  | 5.052437  | 1.937931  | -0.478676 |
| C  | 3.486072  | 0.739877  | 1.643408  |
| H  | 4.115205  | 1.632958  | 1.719356  |
| H  | 2.957917  | 0.635051  | 2.596794  |
| C  | 4.343207  | -0.496823 | 1.369369  |
| H  | 3.723836  | -1.386184 | 1.248016  |
| H  | 5.029418  | -0.677636 | 2.201063  |
| H  | 4.947023  | -0.374056 | 0.465734  |
| C  | -1.354840 | 2.491422  | -2.103405 |
| H  | -2.406026 | 2.198484  | -2.131273 |
| H  | -1.327958 | 3.578870  | -2.229087 |
| C  | -0.580222 | 1.805978  | -3.229610 |
| H  | 0.476417  | 2.091267  | -3.226446 |
| H  | -0.993880 | 2.091908  | -4.200586 |
| H  | -0.625472 | 0.718760  | -3.137057 |
| C  | -1.946270 | 2.928841  | 0.727845  |
| H  | -2.774046 | 2.220537  | 0.819918  |
| H  | -1.474043 | 2.967411  | 1.714432  |
| C  | -2.467577 | 4.309697  | 0.319192  |
| H  | -1.662117 | 5.040349  | 0.204581  |
| H  | -3.145303 | 4.692097  | 1.087593  |
| H  | -3.023719 | 4.271444  | -0.620705 |
| O  | -0.245935 | -0.804790 | 4.293727  |
| O  | -0.492056 | 0.809831  | 2.619231  |
| P  | -2.222857 | -1.090876 | -0.108589 |
| P  | 0.698864  | -2.242699 | -0.733079 |
| P  | 2.183394  | 1.051677  | 0.363349  |
| P  | -0.700464 | 2.139446  | -0.402752 |
| Mo | 0.041346  | -0.218237 | 0.372088  |
| H  | 0.201054  | -0.194293 | -1.315068 |
| H  | 1.182852  | -0.877598 | 1.442617  |

Sum of electronic and zero-point Energies= -2414.556736  
Sum of electronic and thermal Free Energies= -2414.630036

78

# INT3B

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 0.527487  | 1.387887  | -1.828442 |
| H | 0.569450  | -0.884835 | -1.896841 |
| C | 3.163888  | -1.025540 | 1.352745  |
| H | 3.638727  | -1.951773 | 1.017603  |
| H | 3.933951  | -0.456039 | 1.880529  |
| C | 1.990636  | -1.345523 | 2.293811  |
| H | 1.752642  | -0.476889 | 2.911801  |
| H | 2.255692  | -2.168059 | 2.966912  |
| C | -2.772251 | 1.620074  | 0.950434  |
| H | -2.848598 | 0.893164  | 1.764380  |
| H | -3.326810 | 2.512416  | 1.254095  |
| C | -3.314866 | 1.033951  | -0.353217 |
| H | -3.326255 | 1.795041  | -1.139881 |
| H | -4.341237 | 0.675578  | -0.239077 |
| C | 3.570531  | -0.989714 | -1.519250 |
| H | 4.615954  | -1.088619 | -1.207732 |
| H | 3.153568  | -2.002619 | -1.551021 |
| C | 3.474093  | -0.331449 | -2.897644 |
| H | 2.439110  | -0.267578 | -3.243175 |
| H | 4.038984  | -0.909318 | -3.633948 |
| H | 3.878711  | 0.684231  | -2.890741 |
| C | 3.545202  | 1.539619  | -0.064789 |
| H | 3.179824  | 2.153049  | -0.894353 |
| H | 3.166476  | 2.007650  | 0.850459  |
| C | 5.073824  | 1.477418  | -0.061635 |
| H | 5.458728  | 0.835067  | 0.736222  |
| H | 5.488818  | 2.476974  | 0.094671  |
| H | 5.466229  | 1.105415  | -1.011342 |
| C | -0.693713 | -2.478912 | 2.611687  |
| H | -1.446527 | -3.048211 | 2.058352  |
| H | -0.120072 | -3.206236 | 3.197972  |



|    |           |           |           |
|----|-----------|-----------|-----------|
| C  | -1.376060 | -1.467856 | 3.528040  |
| H  | -0.656618 | -0.951092 | 4.166500  |
| H  | -2.098207 | -1.968041 | 4.179298  |
| H  | -1.911029 | -0.710634 | 2.948638  |
| C  | 1.032084  | -3.380478 | 0.560983  |
| H  | 1.903689  | -3.141152 | -0.054967 |
| H  | 1.387304  | -4.026020 | 1.372783  |
| C  | -0.000143 | -4.124279 | -0.288101 |
| H  | -0.869428 | -4.430121 | 0.300603  |
| H  | 0.439816  | -5.030988 | -0.712492 |
| H  | -0.354022 | -3.511366 | -1.123101 |
| C  | -3.020761 | -1.845288 | -0.164598 |
| H  | -2.355159 | -2.690672 | -0.353409 |
| H  | -2.982037 | -1.663764 | 0.914871  |
| C  | -4.446948 | -2.180923 | -0.608367 |
| H  | -5.137704 | -1.348568 | -0.447603 |
| H  | -4.827568 | -3.032992 | -0.037783 |
| H  | -4.487969 | -2.450533 | -1.666604 |
| C  | -2.670135 | -0.486091 | -2.731531 |
| H  | -3.762256 | -0.495877 | -2.811462 |
| H  | -2.317535 | 0.438583  | -3.199903 |
| C  | -2.070015 | -1.703411 | -3.440328 |
| H  | -0.977679 | -1.661421 | -3.453056 |
| H  | -2.407990 | -1.741777 | -4.479074 |
| H  | -2.368646 | -2.641456 | -2.962174 |
| C  | -0.414070 | 2.772860  | 2.286053  |
| H  | 0.562879  | 3.215254  | 2.059264  |
| H  | -1.092867 | 3.599237  | 2.516210  |
| C  | -0.308079 | 1.832243  | 3.480846  |
| H  | -1.285736 | 1.444228  | 3.775232  |
| H  | 0.112279  | 2.357904  | 4.342683  |
| H  | 0.338362  | 0.983013  | 3.253229  |
| C  | -1.039611 | 3.513730  | -0.397314 |
| H  | 0.002208  | 3.810618  | -0.554766 |
| H  | -1.395637 | 3.173287  | -1.372895 |
| C  | -1.871559 | 4.704651  | 0.085832  |
| H  | -2.921884 | 4.440250  | 0.237710  |
| H  | -1.843694 | 5.498115  | -0.666408 |
| H  | -1.490519 | 5.128470  | 1.017795  |
| O  | 0.796191  | 2.104699  | -2.713501 |
| P  | 2.653115  | -0.078128 | -0.186920 |
| P  | 0.454812  | -1.775655 | 1.327508  |
| P  | -2.220265 | -0.356308 | -0.939303 |
| P  | -0.975319 | 2.009191  | 0.696159  |
| Mo | 0.203103  | 0.102977  | -0.386385 |
| H  | 1.030510  | 0.915762  | 0.857585  |
| H  | 0.277581  | -1.473596 | -1.368372 |

Sum of electronic and zero-point Energies= -2339.343802  
Sum of electronic and thermal Free Energies= -2339.413615

76

#### INT3B'

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 0.388203  | 1.012635  | -1.951492 |
| C | 3.172245  | -0.967034 | 1.418729  |
| H | 3.750647  | -1.846655 | 1.123530  |
| H | 3.871474  | -0.296962 | 1.926298  |
| C | 2.039643  | -1.379272 | 2.379775  |
| H | 1.820953  | -0.568176 | 3.078791  |
| H | 2.341254  | -2.247368 | 2.975468  |
| C | -2.662867 | 1.763778  | 0.904488  |
| H | -2.742739 | 1.122232  | 1.787660  |
| H | -3.162195 | 2.708467  | 1.139665  |
| C | -3.288452 | 1.086049  | -0.314033 |
| H | -3.289952 | 1.768349  | -1.170133 |
| H | -4.327631 | 0.800180  | -0.129729 |
| C | 3.484300  | -1.149420 | -1.452595 |
| H | 4.544218  | -1.179789 | -1.177989 |
| H | 3.110382  | -2.174370 | -1.364094 |
| C | 3.304034  | -0.637095 | -2.882873 |
| H | 2.255142  | -0.655107 | -3.187014 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 3.871504  | -1.257592 | -3.581823 |
| H  | 3.656099  | 0.392337  | -2.992065 |
| C  | 3.487005  | 1.487062  | -0.209681 |
| H  | 3.089174  | 2.028708  | -1.073768 |
| H  | 3.137359  | 2.027643  | 0.676628  |
| C  | 5.014852  | 1.428649  | -0.252519 |
| H  | 5.427663  | 0.846029  | 0.576727  |
| H  | 5.430159  | 2.437788  | -0.182103 |
| H  | 5.377549  | 0.990554  | -1.185633 |
| C  | -0.645558 | -2.521910 | 2.742194  |
| H  | -1.414109 | -3.060701 | 2.177431  |
| H  | -0.063373 | -3.274325 | 3.286679  |
| C  | -1.297134 | -1.528854 | 3.704097  |
| H  | -0.554252 | -1.011222 | 4.316413  |
| H  | -1.985123 | -2.042548 | 4.380845  |
| H  | -1.861792 | -0.769496 | 3.155069  |
| C  | 0.924415  | -3.191416 | 0.389873  |
| H  | 2.014767  | -3.269813 | 0.379623  |
| H  | 0.549267  | -4.131426 | 0.806332  |
| C  | 0.392784  | -2.976784 | -1.033076 |
| H  | -0.689468 | -3.086490 | -1.080076 |
| H  | 0.828000  | -3.681887 | -1.748119 |
| H  | 0.654363  | -1.975520 | -1.437589 |
| C  | -3.175900 | -1.807884 | 0.034371  |
| H  | -2.550433 | -2.698385 | -0.085019 |
| H  | -3.140483 | -1.559450 | 1.101475  |
| C  | -4.608876 | -2.103779 | -0.414323 |
| H  | -5.254722 | -1.224966 | -0.331794 |
| H  | -5.046766 | -2.889795 | 0.207488  |
| H  | -4.642020 | -2.447560 | -1.451271 |
| C  | -2.737703 | -0.578852 | -2.605199 |
| H  | -3.828468 | -0.606059 | -2.696194 |
| H  | -2.393263 | 0.345978  | -3.079129 |
| C  | -2.102534 | -1.786578 | -3.297622 |
| H  | -1.011030 | -1.722665 | -3.278038 |
| H  | -2.411569 | -1.828636 | -4.345210 |
| H  | -2.400644 | -2.729972 | -2.829389 |
| C  | -0.248313 | 2.803104  | 2.154794  |
| H  | 0.746066  | 3.204429  | 1.927383  |
| H  | -0.893749 | 3.657378  | 2.379862  |
| C  | -0.182301 | 1.859291  | 3.351956  |
| H  | -1.174452 | 1.502086  | 3.638506  |
| H  | 0.248410  | 2.368834  | 4.218376  |
| H  | 0.435165  | 0.988272  | 3.126167  |
| C  | -0.831235 | 3.488655  | -0.572656 |
| H  | 0.224898  | 3.665438  | -0.802161 |
| H  | -1.276182 | 3.138538  | -1.508522 |
| C  | -1.508882 | 4.778452  | -0.102754 |
| H  | -2.563033 | 4.623849  | 0.145202  |
| H  | -1.470232 | 5.526692  | -0.899525 |
| H  | -1.015385 | 5.208822  | 0.771845  |
| O  | 0.560086  | 1.573056  | -2.968555 |
| P  | 2.594450  | -0.134944 | -0.172092 |
| P  | 0.452783  | -1.737926 | 1.469664  |
| P  | -2.294653 | -0.406320 | -0.815006 |
| P  | -0.850613 | 2.035200  | 0.582422  |
| Mo | 0.148301  | 0.012394  | -0.300701 |
| H  | 1.033349  | 0.966434  | 0.767283  |

Sum of electronic and zero-point Energies= -2338.161755  
Sum of electronic and thermal Free Energies= -2338.231654

79

#### INT3

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 2.422683  | -2.656297 | 0.034435  |
| H | 2.133783  | -3.211771 | -0.864430 |
| H | 3.442941  | -2.961594 | 0.282095  |
| C | 1.457176  | -2.965130 | 1.183725  |
| H | 1.791801  | -2.479940 | 2.106059  |
| H | 1.404408  | -4.041648 | 1.376839  |
| C | -1.293147 | 2.858150  | 1.298853  |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | -1.561287 | 2.381215  | 2.247238  |
| H  | -1.227258 | 3.934531  | 1.482673  |
| C  | -2.336979 | 2.544106  | 0.223631  |
| H  | -2.115275 | 3.108153  | -0.689582 |
| H  | -3.335643 | 2.851066  | 0.546383  |
| C  | 3.108818  | -0.850827 | -2.128170 |
| H  | 3.986237  | -1.506235 | -2.122165 |
| H  | 2.366669  | -1.314820 | -2.785362 |
| C  | 3.481169  | 0.545803  | -2.638045 |
| H  | 2.615635  | 1.212047  | -2.667893 |
| H  | 3.873275  | 0.481592  | -3.656071 |
| H  | 4.251115  | 1.011594  | -2.016094 |
| C  | 3.676057  | -0.063843 | 0.629547  |
| H  | 3.639702  | 1.006931  | 0.401632  |
| H  | 3.325722  | -0.165060 | 1.662329  |
| C  | 5.103381  | -0.594810 | 0.479556  |
| H  | 5.171487  | -1.658124 | 0.724641  |
| H  | 5.776712  | -0.061547 | 1.156526  |
| H  | 5.483219  | -0.459505 | -0.536188 |
| C  | -1.216862 | -2.670638 | 2.308877  |
| H  | -2.264423 | -2.473715 | 2.065064  |
| H  | -1.129413 | -3.745666 | 2.505914  |
| C  | -0.802015 | -1.856553 | 3.536263  |
| H  | 0.238151  | -2.044077 | 3.817733  |
| H  | -1.426759 | -2.121781 | 4.393263  |
| H  | -0.911184 | -0.785334 | 3.351765  |
| C  | -0.842155 | -3.543735 | -0.454874 |
| H  | -0.323172 | -3.322298 | -1.391648 |
| C  | -0.511732 | -4.532556 | -0.116055 |
| H  | -2.354535 | -3.528297 | -0.682605 |
| H  | -2.905438 | -3.813341 | 0.217892  |
| H  | -2.622479 | -4.237635 | -1.470123 |
| H  | -2.692610 | -2.541965 | -1.000866 |
| C  | -3.591883 | -0.043759 | 0.841716  |
| H  | -3.579259 | -1.111558 | 0.603727  |
| H  | -3.179634 | 0.046734  | 1.852975  |
| C  | -5.020320 | 0.500944  | 0.778186  |
| H  | -5.065299 | 1.566843  | 1.017933  |
| H  | -5.654303 | -0.021351 | 1.500301  |
| H  | -5.463226 | 0.359607  | -0.210808 |
| C  | -3.185570 | 0.862689  | -1.925749 |
| H  | -4.067932 | 1.499383  | -1.798010 |
| H  | -2.499616 | 1.401713  | -2.587148 |
| C  | -3.568003 | -0.480126 | -2.552026 |
| H  | -2.685111 | -1.098672 | -2.724551 |
| H  | -4.056508 | -0.316259 | -3.516137 |
| H  | -4.267286 | -1.036692 | -1.920747 |
| C  | 1.421616  | 2.477267  | 2.292572  |
| H  | 2.459077  | 2.351094  | 1.966688  |
| H  | 1.297958  | 3.529589  | 2.568126  |
| C  | 1.125128  | 1.575189  | 3.491672  |
| H  | 0.092276  | 1.683944  | 3.835306  |
| H  | 1.779292  | 1.835612  | 4.327947  |
| H  | 1.284794  | 0.523364  | 3.244945  |
| C  | 1.008801  | 3.369948  | -0.456298 |
| H  | 1.963655  | 2.958829  | -0.801209 |
| H  | 0.329746  | 3.311572  | -1.313976 |
| C  | 1.175030  | 4.823482  | -0.003312 |
| H  | 0.232512  | 5.253443  | 0.346431  |
| H  | 1.520396  | 5.433946  | -0.842138 |
| H  | 1.911216  | 4.919696  | 0.798352  |
| P  | 2.373398  | -0.849817 | -0.429035 |
| P  | -0.238607 | -2.307866 | 0.786019  |
| P  | -2.340374 | 0.741659  | -0.279255 |
| P  | 0.363400  | 2.170777  | 0.809736  |
| Mo | 0.033594  | -0.049076 | -0.127336 |
| H  | 0.950538  | -0.205684 | 1.263545  |
| H  | -0.743842 | 0.136733  | 1.349470  |
| H  | 0.344010  | 1.036158  | -1.676227 |
| C  | 0.010409  | 0.286652  | -2.577685 |
| O  | 0.008037  | 0.834532  | -3.653520 |

|   |           |           |           |
|---|-----------|-----------|-----------|
| O | -0.236617 | -0.880527 | -2.140460 |
|---|-----------|-----------|-----------|

Sum of electronic and zero-point Energies= -2414.572249  
Sum of electronic and thermal Free Energies= -2414.643506

77

# INT4-I

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -0.057332 | -0.025538 | -3.061904 |
| H | -0.102308 | -0.131378 | -4.153137 |
| C | 2.670178  | -2.359132 | 0.420973  |
| H | 2.510582  | -3.112118 | -0.357900 |
| H | 3.707996  | -2.456437 | 0.748969  |
| C | 1.690688  | -2.572003 | 1.575393  |
| H | 1.948145  | -1.930699 | 2.425031  |
| H | 1.697414  | -3.608183 | 1.929408  |
| C | -1.306389 | 2.508828  | 1.397040  |
| H | -1.514753 | 1.905186  | 2.286703  |
| H | -1.230593 | 3.547472  | 1.733031  |
| C | -2.416719 | 2.340386  | 0.356894  |
| H | -2.219183 | 2.973228  | -0.514714 |
| H | -3.386031 | 2.643404  | 0.763146  |
| C | 3.204703  | -0.916105 | -2.025718 |
| H | 4.169994  | -1.411056 | -1.877965 |
| H | 2.568529  | -1.600138 | -2.592760 |
| C | 3.380836  | 0.405908  | -2.780834 |
| H | 2.437230  | 0.951759  | -2.867287 |
| H | 3.759789  | 0.225076  | -3.790540 |
| H | 4.095434  | 1.064197  | -2.277816 |
| C | 3.594694  | 0.394904  | 0.516842  |
| H | 3.445348  | 1.408094  | 0.131303  |
| H | 3.278622  | 0.409030  | 1.563031  |
| C | 5.075041  | 0.014746  | 0.419163  |
| H | 5.267225  | -0.993429 | 0.795341  |
| H | 5.677638  | 0.704189  | 1.017765  |
| H | 5.442973  | 0.064276  | -0.608081 |
| C | -1.065941 | -2.334791 | 2.505362  |
| H | -2.097682 | -2.413624 | 2.151646  |
| H | -0.804681 | -3.304823 | 2.943810  |
| C | -0.950403 | -1.222058 | 3.545722  |
| H | 0.056852  | -1.163314 | 3.964944  |
| H | -1.644352 | -1.391181 | 4.373812  |
| H | -1.180273 | -0.247094 | 3.105424  |
| C | -0.480127 | -3.586281 | -0.015168 |
| H | 0.251246  | -3.652730 | -0.826228 |
| H | -0.351359 | -4.472797 | 0.616979  |
| C | -1.893022 | -3.536728 | -0.602120 |
| H | -2.656889 | -3.472909 | 0.178582  |
| H | -2.096227 | -4.440453 | -1.183488 |
| H | -2.012480 | -2.679962 | -1.268513 |
| C | -3.669149 | -0.260471 | 0.871463  |
| H | -3.645839 | -1.318939 | 0.593305  |
| H | -3.228966 | -0.202438 | 1.873186  |
| C | -5.107594 | 0.262322  | 0.877914  |
| H | -5.157656 | 1.324497  | 1.133684  |
| H | -5.703210 | -0.279682 | 1.617857  |
| H | -5.589332 | 0.127611  | -0.093794 |
| C | -3.377456 | 0.746097  | -1.869713 |
| H | -4.305018 | 1.298092  | -1.685802 |
| H | -2.742522 | 1.368445  | -2.505676 |
| C | -3.664933 | -0.597318 | -2.545246 |
| H | -2.741187 | -1.135183 | -2.775741 |
| H | -4.202843 | -0.441479 | -3.484175 |
| H | -4.281901 | -1.246059 | -1.916377 |
| C | 1.452580  | 2.422774  | 2.156484  |
| H | 2.472300  | 2.505066  | 1.774934  |
| H | 1.158680  | 3.426065  | 2.481647  |
| C | 1.390683  | 1.449483  | 3.332716  |
| H | 0.403941  | 1.448787  | 3.802246  |
| H | 2.119302  | 1.727545  | 4.099120  |
| H | 1.603599  | 0.425796  | 3.013940  |
| C | 0.781941  | 3.270092  | -0.532913 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 1.795256  | 3.022464  | -0.868476 |
| H  | 0.139295  | 3.088068  | -1.396057 |
| C  | 0.701428  | 4.727866  | -0.073337 |
| H  | -0.311113 | 5.002998  | 0.235575  |
| H  | 0.978130  | 5.389360  | -0.899181 |
| H  | 1.378882  | 4.940437  | 0.757418  |
| O  | 0.173060  | -1.072769 | -2.346373 |
| O  | -0.219783 | 1.074225  | -2.489119 |
| P  | 2.379967  | -0.693976 | -0.379248 |
| P  | -0.011461 | -2.091787 | 0.997634  |
| P  | -2.470298 | 0.588859  | -0.263026 |
| P  | 0.355911  | 1.971312  | 0.723750  |
| Mo | -0.055196 | -0.174064 | -0.440117 |

Sum of electronic and zero-point Energies= -2413.391949  
Sum of electronic and thermal Free Energies= -2413.463052

77

#### INT4

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 0.429114  | -0.897459 | -2.289858 |
| C | 2.610181  | -2.375531 | -0.224403 |
| H | 2.303712  | -2.739499 | -1.208880 |
| C | 3.650257  | -2.670901 | -0.067873 |
| C | 1.699658  | -2.908931 | 0.880612  |
| H | 2.044241  | -2.573240 | 1.864039  |
| H | 1.676170  | -4.002843 | 0.899915  |
| C | -1.125093 | 2.721895  | 1.419688  |
| H | -0.966748 | 2.197791  | 2.366439  |
| H | -1.091918 | 3.796077  | 1.626986  |
| C | -2.455320 | 2.315366  | 0.792346  |
| H | -2.691416 | 2.973969  | -0.049169 |
| H | -3.273156 | 2.417294  | 1.509907  |
| C | 3.566654  | -0.055778 | -1.657496 |
| H | 4.574657  | -0.405752 | -1.412579 |
| H | 3.214830  | -0.627361 | -2.519488 |
| C | 3.569479  | 1.438909  | -1.977432 |
| H | 2.571505  | 1.771898  | -2.276141 |
| H | 4.247635  | 1.649219  | -2.808543 |
| H | 3.898231  | 2.045587  | -1.127196 |
| C | 3.357524  | -0.017492 | 1.263068  |
| H | 3.513582  | 1.060217  | 1.190434  |
| H | 2.668982  | -0.169655 | 2.098687  |
| C | 4.694780  | -0.713865 | 1.531465  |
| H | 4.565894  | -1.777221 | 1.748104  |
| H | 5.182367  | -0.261380 | 2.399699  |
| H | 5.383014  | -0.623092 | 0.686864  |
| C | -0.845231 | -2.648665 | 2.237072  |
| H | -1.925265 | -2.611218 | 2.080865  |
| H | -0.596372 | -3.692122 | 2.464725  |
| C | -0.440548 | -1.736549 | 3.395977  |
| H | 0.638668  | -1.761026 | 3.574180  |
| H | -0.934179 | -2.056637 | 4.317601  |
| H | -0.716811 | -0.697942 | 3.198958  |
| C | -0.708224 | -3.589340 | -0.506878 |
| H | -0.110350 | -3.551589 | -1.422727 |
| H | -0.498158 | -4.549850 | -0.021548 |
| C | -2.193562 | -3.462692 | -0.838033 |
| H | -2.818283 | -3.544488 | 0.056671  |
| H | -2.495544 | -4.261614 | -1.520426 |
| H | -2.402265 | -2.512070 | -1.326906 |
| C | -3.438861 | -0.403396 | 1.297768  |
| H | -3.431840 | -1.431155 | 0.924080  |
| H | -2.883204 | -0.411692 | 2.241311  |
| C | -4.879002 | 0.068657  | 1.515970  |
| H | -4.925893 | 1.092826  | 1.895619  |
| H | -5.374644 | -0.572353 | 2.250482  |
| H | -5.462799 | 0.025292  | 0.593560  |
| C | -3.516397 | 0.800581  | -1.372749 |
| H | -4.350619 | 1.428841  | -1.041377 |
| H | -2.937262 | 1.390013  | -2.091264 |
| C | -4.036980 | -0.471173 | -2.044835 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | -3.226079 | -1.017149 | -2.525936 |
| H  | -4.767321 | -0.206375 | -2.814571 |
| H  | -4.537914 | -1.136195 | -1.335368 |
| C  | 1.720002  | 3.073192  | 1.135416  |
| H  | 2.620262  | 2.771633  | 0.596458  |
| H  | 1.603390  | 4.149046  | 0.968486  |
| C  | 1.833774  | 2.789265  | 2.637871  |
| H  | 1.040530  | 3.293085  | 3.194936  |
| H  | 2.788571  | 3.159450  | 3.020152  |
| H  | 1.774681  | 1.721479  | 2.859076  |
| C  | 0.038810  | 3.224809  | -1.248064 |
| H  | 0.989725  | 3.172774  | -1.786401 |
| H  | -0.688893 | 2.682551  | -1.862124 |
| C  | -0.395015 | 4.684218  | -1.068393 |
| H  | -1.374849 | 4.765405  | -0.591109 |
| H  | -0.462538 | 5.171062  | -2.044970 |
| H  | 0.320186  | 5.252881  | -0.468407 |
| O  | -0.827230 | -0.556962 | -2.255022 |
| O  | 1.125763  | -1.458540 | -3.107711 |
| P  | 2.441914  | -0.531140 | -0.271812 |
| P  | -0.031684 | -2.276102 | 0.618657  |
| P  | -2.409906 | 0.575445  | 0.103630  |
| P  | 0.286207  | 2.242911  | 0.316719  |
| Mo | -0.003511 | -0.075342 | -0.416591 |
| H  | 0.220001  | -0.018406 | 1.310875  |

Sum of electronic and zero-point Energies= -2413.389960  
Sum of electronic and thermal Free Energies= -2413.459720

79

#### INT4'

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -0.244140 | 1.462737  | -2.077325 |
| H | -0.008133 | -1.229503 | -1.960341 |
| C | 2.224135  | -2.825421 | -0.211070 |
| H | 1.931098  | -3.269453 | -1.168388 |
| H | 3.188179  | -3.260501 | 0.063610  |
| C | 1.168545  | -3.092119 | 0.860384  |
| H | 1.467735  | -2.636417 | 1.809308  |
| H | 1.034356  | -4.163477 | 1.042886  |
| C | -1.045321 | 2.658265  | 1.470045  |
| H | -1.293492 | 2.093383  | 2.374781  |
| H | -0.898002 | 3.699414  | 1.772252  |
| C | -2.160734 | 2.534117  | 0.434504  |
| H | -1.931453 | 3.124717  | -0.456902 |
| H | -3.112019 | 2.896693  | 0.832329  |
| C | 3.317357  | -0.849839 | -2.052112 |
| H | 4.038443  | -1.674178 | -2.082161 |
| H | 2.597419  | -1.011932 | -2.859091 |
| C | 4.027255  | 0.495176  | -2.245210 |
| H | 3.315646  | 1.321340  | -2.220743 |
| H | 4.525124  | 0.511582  | -3.218442 |
| H | 4.793920  | 0.664381  | -1.484221 |
| C | 3.585034  | -0.475211 | 0.822940  |
| H | 3.749937  | 0.593647  | 0.656567  |
| H | 3.084025  | -0.561194 | 1.790936  |
| C | 4.924071  | -1.218735 | 0.843445  |
| H | 4.801808  | -2.272486 | 1.106639  |
| H | 5.587661  | -0.773785 | 1.590343  |
| H | 5.434680  | -1.170014 | -0.121417 |
| C | -1.490871 | -2.777691 | 1.869521  |
| H | -2.533339 | -2.556658 | 1.638685  |
| H | -1.420596 | -3.865701 | 1.985561  |
| C | -1.070960 | -2.079444 | 3.162893  |
| H | -0.063655 | -2.368660 | 3.473721  |
| H | -1.753520 | -2.350407 | 3.972830  |
| H | -1.084511 | -0.993134 | 3.056381  |
| C | -1.074425 | -3.651643 | -0.833523 |
| H | -0.448613 | -3.571660 | -1.727863 |
| H | -0.870092 | -4.631024 | -0.385903 |
| C | -2.552766 | -3.533688 | -1.206920 |
| H | -3.201199 | -3.675088 | -0.337733 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | -2.818999 | -4.298624 | -1.941025 |
| H  | -2.777782 | -2.560597 | -1.643791 |
| C  | -3.393827 | -0.038523 | 1.115045  |
| H  | -3.606223 | -1.037102 | 0.720766  |
| H  | -2.767263 | -0.179962 | 1.999164  |
| C  | -4.704411 | 0.656195  | 1.494744  |
| H  | -4.533305 | 1.644401  | 1.929102  |
| H  | -5.238674 | 0.061186  | 2.241125  |
| H  | -5.368578 | 0.774862  | 0.635453  |
| C  | -3.510916 | 1.032168  | -1.577422 |
| H  | -4.370039 | 1.581686  | -1.179141 |
| H  | -2.993857 | 1.693212  | -2.273882 |
| C  | -3.967397 | -0.240572 | -2.286211 |
| H  | -3.122254 | -0.753613 | -2.754558 |
| H  | -4.675980 | 0.007121  | -3.081138 |
| H  | -4.466259 | -0.941018 | -1.609180 |
| C  | 1.589956  | 1.927061  | 2.355513  |
| H  | 2.631914  | 1.821110  | 2.043822  |
| H  | 1.503101  | 2.915892  | 2.816788  |
| C  | 1.218646  | 0.846885  | 3.370460  |
| H  | 0.198248  | 0.975791  | 3.742273  |
| H  | 1.888841  | 0.893148  | 4.233405  |
| H  | 1.285168  | -0.153584 | 2.938609  |
| C  | 1.303894  | 3.368965  | -0.147178 |
| H  | 2.171700  | 2.950117  | -0.666244 |
| H  | 0.584275  | 3.621016  | -0.930927 |
| C  | 1.692892  | 4.620246  | 0.645518  |
| H  | 0.848619  | 5.033902  | 1.204994  |
| H  | 2.036282  | 5.397368  | -0.042711 |
| H  | 2.504071  | 4.425566  | 1.351284  |
| O  | -0.922530 | 2.307057  | -2.620137 |
| O  | 0.951485  | 1.058376  | -2.145053 |
| P  | 2.360334  | -0.995735 | -0.471794 |
| P  | -0.476143 | -2.370137 | 0.372099  |
| P  | -2.335087 | 0.785632  | -0.168835 |
| P  | 0.554099  | 1.972634  | 0.818721  |
| Mo | 0.006435  | -0.056862 | -0.548487 |
| H  | -0.108421 | -0.190630 | 1.146582  |
| H  | -0.802006 | -0.885894 | -1.967571 |

Sum of electronic and zero-point Energies= -2414.567755  
Sum of electronic and thermal Free Energies= -2414.636727

80

#### INT5-I

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 0.121724  | -0.089972 | 2.801932  |
| H | 0.167393  | -0.099440 | 3.897976  |
| C | -2.577378 | -2.565744 | 0.496020  |
| H | -2.215482 | -2.966031 | 1.449579  |
| H | -3.621827 | -2.873646 | 0.395230  |
| C | -1.731998 | -3.091175 | -0.669463 |
| H | -2.098870 | -2.685774 | -1.616292 |
| H | -1.782828 | -4.182749 | -0.736937 |
| C | 1.384961  | 2.961646  | -0.959879 |
| H | 1.602593  | 2.529249  | -1.937985 |
| H | 1.410701  | 4.049442  | -1.063793 |
| C | 2.416165  | 2.487589  | 0.066023  |
| H | 2.238029  | 2.966990  | 1.035251  |
| H | 3.426529  | 2.768771  | -0.244079 |
| C | -3.218860 | -0.366349 | 2.257614  |
| H | -4.153458 | -0.934117 | 2.318321  |
| H | -2.550672 | -0.791471 | 3.014306  |
| C | -3.462853 | 1.120957  | 2.525233  |
| H | -2.525526 | 1.679700  | 2.498448  |
| H | -3.914206 | 1.259858  | 3.511314  |
| H | -4.146684 | 1.554066  | 1.789276  |
| C | -3.659850 | -0.069704 | -0.627886 |
| H | -3.636127 | 1.019781  | -0.522012 |
| H | -3.230677 | -0.286424 | -1.610050 |
| C | -5.095407 | -0.589066 | -0.519573 |
| H | -5.151990 | -1.669321 | -0.676886 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | -5.721062 | -0.117255 | -1.282383 |
| H  | -5.539402 | -0.365322 | 0.454357  |
| C  | 0.854216  | -3.117306 | -2.069904 |
| H  | 1.804768  | -2.580611 | -2.136164 |
| H  | 1.092053  | -4.177519 | -1.927636 |
| C  | 0.055147  | -2.937437 | -3.365570 |
| H  | -0.860906 | -3.533804 | -3.357734 |
| H  | 0.660150  | -3.279061 | -4.209627 |
| H  | -0.209403 | -1.895790 | -3.546597 |
| C  | 0.719289  | -3.746149 | 0.746247  |
| H  | 0.290518  | -3.492650 | 1.717190  |
| H  | 0.357560  | -4.741984 | 0.464895  |
| C  | 2.248035  | -3.728293 | 0.819097  |
| H  | 2.703222  | -4.058897 | -0.118306 |
| H  | 2.598757  | -4.396434 | 1.610194  |
| H  | 2.616659  | -2.726759 | 1.042226  |
| C  | 3.613747  | -0.129485 | -0.718551 |
| H  | 3.575790  | -1.201369 | -0.495539 |
| H  | 3.218746  | -0.011527 | -1.729958 |
| C  | 5.046658  | 0.397149  | -0.609821 |
| H  | 5.109283  | 1.462103  | -0.850336 |
| H  | 5.687995  | -0.130877 | -1.321016 |
| H  | 5.470628  | 0.250872  | 0.387141  |
| C  | 3.173126  | 0.638111  | 2.065903  |
| H  | 4.043177  | 1.298711  | 1.983603  |
| H  | 2.481419  | 1.132688  | 2.754226  |
| C  | 3.597672  | -0.728347 | 2.606552  |
| H  | 2.740424  | -1.388309 | 2.747700  |
| H  | 4.097091  | -0.609479 | 3.572339  |
| H  | 4.303088  | -1.225812 | 1.935372  |
| C  | -1.424992 | 2.766753  | -1.919849 |
| H  | -2.108309 | 1.919195  | -1.999825 |
| H  | -2.017330 | 3.643132  | -1.638012 |
| C  | -0.736529 | 3.001768  | -3.268991 |
| H  | -0.079000 | 3.874569  | -3.241372 |
| H  | -1.494878 | 3.184414  | -4.035366 |
| H  | -0.148265 | 2.136188  | -3.580296 |
| C  | -0.867413 | 3.552362  | 0.843671  |
| H  | -1.932132 | 3.339267  | 0.988209  |
| H  | -0.371256 | 3.259128  | 1.770327  |
| C  | -0.646358 | 5.037643  | 0.542272  |
| H  | 0.416597  | 5.291530  | 0.526213  |
| H  | -1.115840 | 5.648108  | 1.318755  |
| H  | -1.078447 | 5.336757  | -0.417239 |
| O  | 0.294749  | -1.150051 | 2.134328  |
| O  | -0.101569 | 0.984347  | 2.170061  |
| P  | -2.449355 | -0.713368 | 0.612660  |
| P  | 0.039602  | -2.555623 | -0.493988 |
| P  | 2.359902  | 0.647656  | 0.394461  |
| P  | -0.313127 | 2.381754  | -0.479501 |
| Mo | -0.054549 | -0.075731 | 0.176656  |
| C  | 0.490493  | 0.023074  | -1.841800 |
| O  | -0.780607 | -0.204216 | -1.806521 |
| O  | 1.278619  | 0.197946  | -2.748707 |

Sum of electronic and zero-point Energies= -2601.995038  
Sum of electronic and thermal Free Energies= -2602.067630

77

#### INT5

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 0.429898  | 1.201062  | -2.075318 |
| C | 2.095307  | -2.878275 | 0.084440  |
| H | 1.758120  | -3.405454 | -0.814247 |
| H | 3.063016  | -3.305248 | 0.362401  |
| C | 1.070768  | -3.030252 | 1.211579  |
| H | 1.403906  | -2.491645 | 2.105123  |
| H | 0.953226  | -4.082324 | 1.492068  |
| C | -0.985500 | 2.714076  | 1.459367  |
| H | -1.192733 | 2.047140  | 2.301060  |
| H | -0.878691 | 3.725141  | 1.863472  |
| C | -2.117269 | 2.646565  | 0.432468  |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | -1.912579 | 3.314619  | -0.411001 |
| H  | -3.066054 | 2.969961  | 0.869966  |
| C  | 3.154507  | -1.231596 | -2.046129 |
| H  | 3.907276  | -2.024592 | -1.975853 |
| H  | 2.386312  | -1.567901 | -2.749316 |
| C  | 3.791319  | 0.076071  | -2.524281 |
| H  | 3.057964  | 0.882820  | -2.588221 |
| H  | 4.217593  | -0.058413 | -3.521726 |
| H  | 4.599920  | 0.400073  | -1.862805 |
| C  | 3.582747  | -0.426065 | 0.735490  |
| H  | 3.737779  | 0.614123  | 0.433055  |
| H  | 3.134716  | -0.396324 | 1.731838  |
| C  | 4.918453  | -1.175175 | 0.765070  |
| H  | 4.802271  | -2.201542 | 1.121922  |
| H  | 5.613038  | -0.671446 | 1.443057  |
| H  | 5.388090  | -1.211826 | -0.220942 |
| C  | -1.556484 | -2.511888 | 2.271173  |
| H  | -2.609141 | -2.371287 | 2.011751  |
| H  | -1.453604 | -3.551745 | 2.603153  |
| C  | -1.159703 | -1.550590 | 3.394448  |
| H  | -0.123707 | -1.701435 | 3.710644  |
| H  | -1.795811 | -1.705089 | 4.270535  |
| H  | -1.254119 | -0.508739 | 3.078889  |
| C  | -1.251928 | -3.667029 | -0.376654 |
| H  | -0.618418 | -3.683501 | -1.267462 |
| H  | -1.139092 | -4.626110 | 0.141652  |
| C  | -2.704561 | -3.432375 | -0.798635 |
| H  | -3.385185 | -3.423629 | 0.058596  |
| H  | -3.039126 | -4.227950 | -1.469970 |
| H  | -2.796601 | -2.485987 | -1.333830 |
| C  | -3.542482 | 0.103911  | 0.806437  |
| H  | -3.591217 | -0.931283 | 0.457942  |
| H  | -3.108078 | 0.070152  | 1.810890  |
| C  | -4.941238 | 0.725868  | 0.830669  |
| H  | -4.924066 | 1.769760  | 1.156297  |
| H  | -5.582338 | 0.178309  | 1.527135  |
| H  | -5.416011 | 0.686953  | -0.152809 |
| C  | -3.147288 | 1.232034  | -1.876311 |
| H  | -4.035448 | 1.841896  | -1.678161 |
| H  | -2.461979 | 1.841640  | -2.473945 |
| C  | -3.510704 | -0.049514 | -2.629054 |
| H  | -2.618901 | -0.649651 | -2.826774 |
| H  | -3.977275 | 0.199068  | -3.585890 |
| H  | -4.223284 | -0.658277 | -2.064228 |
| C  | 1.767277  | 2.205996  | 2.177196  |
| H  | 2.781926  | 2.059986  | 1.796600  |
| H  | 1.728471  | 3.225324  | 2.573160  |
| C  | 1.457122  | 1.208070  | 3.293778  |
| H  | 0.487688  | 1.411116  | 3.756302  |
| H  | 2.215830  | 1.279336  | 4.078357  |
| H  | 1.430564  | 0.181223  | 2.925569  |
| C  | 1.175580  | 3.645515  | -0.271930 |
| H  | 2.085314  | 3.340506  | -0.801680 |
| H  | 0.419356  | 3.790700  | -1.050061 |
| C  | 1.413602  | 4.955346  | 0.485244  |
| H  | 0.532722  | 5.269851  | 1.052325  |
| H  | 1.645884  | 5.755326  | -0.223291 |
| H  | 2.254117  | 4.880748  | 1.179065  |
| O  | -0.404713 | -1.171752 | -1.934339 |
| O  | 0.628292  | 1.881246  | -2.984490 |
| P  | 2.298288  | -1.101188 | -0.412201 |
| P  | -0.577534 | -2.326091 | 0.704854  |
| P  | -2.285429 | 0.936530  | -0.266848 |
| P  | 0.625056  | 2.162350  | 0.713799  |
| Mo | -0.019697 | -0.120695 | -0.568459 |
| H  | 0.094732  | -0.045672 | 1.196966  |

Sum of electronic and zero-point Energies= -2413.391515  
Sum of electronic and thermal Free Energies= -2413.463310

80  
**INT6-I**

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -0.339334 | 1.384737  | -1.991945 |
| C | 3.385353  | -1.242619 | -0.779734 |
| H | 3.162679  | -1.621092 | -1.782786 |
| H | 4.470867  | -1.122403 | -0.713006 |
| C | 2.871740  | -2.206172 | 0.297540  |
| H | 3.094356  | -1.810779 | 1.292261  |
| H | 3.377932  | -3.173347 | 0.215004  |
| C | -2.217285 | 1.766313  | 1.705472  |
| H | -1.759477 | 1.235903  | 2.535918  |
| H | -2.749385 | 2.629173  | 2.116896  |
| C | -3.196983 | 0.833613  | 0.981858  |
| H | -3.853141 | 1.409561  | 0.321501  |
| H | -3.837271 | 0.338108  | 1.716118  |
| C | 3.187154  | 1.347406  | -2.108865 |
| H | 4.259805  | 1.153865  | -2.216362 |
| H | 2.691895  | 0.916491  | -2.985197 |
| C | 2.917054  | 2.853078  | -2.016849 |
| H | 1.858809  | 3.063145  | -1.839092 |
| H | 3.197511  | 3.345583  | -2.951119 |
| H | 3.494655  | 3.315558  | -1.211844 |
| C | 3.313739  | 1.216008  | 0.826434  |
| H | 2.857506  | 2.207577  | 0.899110  |
| H | 2.948247  | 0.657008  | 1.690351  |
| C | 4.841013  | 1.318504  | 0.817208  |
| H | 5.314709  | 0.333334  | 0.813063  |
| H | 5.184583  | 1.841298  | 1.714101  |
| H | 5.210573  | 1.873740  | -0.049237 |
| C | 0.583797  | -3.487959 | 1.661567  |
| H | -0.417577 | -3.181204 | 1.966558  |
| H | 0.520253  | -4.525597 | 1.318161  |
| C | 1.538698  | -3.401225 | 2.856342  |
| H | 2.528887  | -3.795850 | 2.614970  |
| H | 1.141172  | -3.989516 | 3.687609  |
| H | 1.663189  | -2.376561 | 3.215213  |
| C | 0.927770  | -3.666242 | -1.219957 |
| H | 1.325361  | -3.168650 | -2.108045 |
| H | 1.593892  | -4.501605 | -0.974625 |
| C | -0.489847 | -4.166574 | -1.504043 |
| H | -0.937561 | -4.644960 | -0.627106 |
| H | -0.477802 | -4.906748 | -2.308780 |
| H | -1.132781 | -3.343396 | -1.820079 |
| C | -2.968074 | -2.108954 | 0.609423  |
| H | -2.459517 | -2.880853 | 0.025604  |
| H | -2.554353 | -2.141081 | 1.617728  |
| C | -4.481389 | -2.341939 | 0.611198  |
| H | -5.016874 | -1.557571 | 1.154235  |
| H | -4.706895 | -3.290099 | 1.107222  |
| H | -4.892679 | -2.393892 | -0.400061 |
| C | -3.375476 | -0.309415 | -1.653459 |
| H | -4.438502 | -0.324037 | -1.388953 |
| H | -3.157307 | 0.692186  | -2.038389 |
| C | -3.067319 | -1.366612 | -2.714869 |
| H | -2.010614 | -1.355047 | -2.992166 |
| H | -3.657992 | -1.177731 | -3.615356 |
| H | -3.318959 | -2.371069 | -2.362138 |
| C | 0.185493  | 3.496494  | 1.588832  |
| H | 1.021556  | 3.782277  | 0.940288  |
| H | -0.398379 | 4.406439  | 1.759380  |
| C | 0.697576  | 2.953014  | 2.924696  |
| H | -0.128328 | 2.683782  | 3.588772  |
| H | 1.287662  | 3.721454  | 3.431775  |
| H | 1.321647  | 2.072016  | 2.784724  |
| C | -1.779465 | 3.560903  | -0.526357 |
| H | -1.013389 | 4.081671  | -1.110897 |
| H | -2.353926 | 2.971631  | -1.246852 |
| C | -2.704915 | 4.569168  | 0.161835  |
| H | -3.554730 | 4.078122  | 0.642836  |
| H | -3.107795 | 5.263007  | -0.581150 |
| H | -2.188945 | 5.166931  | 0.917677  |

|    |           |           |           |
|----|-----------|-----------|-----------|
| O  | 0.173358  | -0.935051 | -2.054714 |
| O  | -0.483090 | 2.075234  | -2.901978 |
| P  | 2.557773  | 0.409025  | -0.649691 |
| P  | 1.025831  | -2.449819 | 0.172262  |
| P  | -2.405868 | -0.489043 | -0.080285 |
| P  | -0.876265 | 2.356578  | 0.575271  |
| Mo | 0.070517  | -0.062675 | -0.579182 |
| C  | 0.017784  | -0.388619 | 2.640216  |
| O  | -1.073817 | -0.937643 | 2.750228  |
| O  | 0.556028  | 0.071094  | 1.552385  |
| H  | 0.650308  | -0.222053 | 3.531805  |

Sum of electronic and zero-point Energies= -2601.988271  
Sum of electronic and thermal Free Energies= -2602.063346

## 75 INT6

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 2.509512  | -2.584164 | -0.285257 |
| H | 2.240594  | -2.976659 | -1.271852 |
| H | 3.522636  | -2.931090 | -0.065313 |
| C | 1.513567  | -3.060215 | 0.775446  |
| H | 1.799242  | -2.692371 | 1.765675  |
| H | 1.475512  | -4.153052 | 0.826927  |
| C | -1.060546 | 2.593542  | 1.554619  |
| H | -0.847714 | 1.991930  | 2.442273  |
| H | -1.033585 | 3.646403  | 1.851489  |
| C | -2.423251 | 2.213398  | 0.972605  |
| H | -2.760966 | 2.973150  | 0.261005  |
| H | -3.178105 | 2.159522  | 1.761380  |
| C | 3.337608  | -0.357410 | -1.979679 |
| H | 4.294651  | -0.888765 | -1.991651 |
| H | 2.710800  | -0.771409 | -2.774957 |
| C | 3.533653  | 1.145705  | -2.195237 |
| H | 2.567995  | 1.658205  | -2.218451 |
| H | 4.034456  | 1.334268  | -3.148327 |
| H | 4.146457  | 1.594409  | -1.406575 |
| C | 3.595276  | -0.200302 | 0.947050  |
| H | 3.553202  | 0.887967  | 0.995961  |
| C | 3.144939  | -0.559964 | 1.877343  |
| H | 5.049721  | -0.662066 | 0.819804  |
| H | 5.138394  | -1.750787 | 0.779608  |
| H | 5.625059  | -0.321137 | 1.685166  |
| H | 5.527143  | -0.252459 | -0.074160 |
| C | -1.159395 | -2.880292 | 1.890025  |
| H | -2.215987 | -2.815048 | 1.620548  |
| H | -0.946769 | -3.941087 | 2.069553  |
| C | -0.876853 | -2.053760 | 3.145539  |
| H | 0.171921  | -2.122263 | 3.446946  |
| H | -1.487520 | -2.415021 | 3.977830  |
| H | -1.093919 | -0.995938 | 2.981891  |
| C | -0.752427 | -3.619787 | -0.885402 |
| H | -0.107170 | -3.471884 | -1.757113 |
| H | -0.552667 | -4.624649 | -0.494871 |
| C | -2.218557 | -3.480017 | -1.294584 |
| H | -2.894337 | -3.672146 | -0.455735 |
| H | -2.463355 | -4.199597 | -2.080491 |
| H | -2.421527 | -2.481598 | -1.683258 |
| C | -3.613602 | -0.482590 | 0.894175  |
| H | -3.576189 | -1.452475 | 0.390071  |
| H | -3.241248 | -0.644167 | 1.911386  |
| C | -5.048676 | 0.051405  | 0.915617  |
| H | -5.113642 | 1.041454  | 1.375714  |
| H | -5.689621 | -0.620171 | 1.493476  |
| H | -5.467510 | 0.119477  | -0.091410 |
| C | -3.150981 | 1.050410  | -1.564275 |
| H | -4.100205 | 1.540946  | -1.323457 |
| H | -2.499760 | 1.810845  | -2.005145 |
| C | -3.358049 | -0.095602 | -2.552095 |
| H | -2.401500 | -0.544112 | -2.830044 |
| H | -3.831948 | 0.279790  | -3.463247 |
| H | -4.009515 | -0.871942 | -2.140233 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| C  | 1.794934  | 2.896128  | 1.243497  |
| H  | 2.661528  | 2.657383  | 0.620090  |
| H  | 1.707148  | 3.987573  | 1.246192  |
| C  | 1.967749  | 2.377672  | 2.675008  |
| H  | 1.222498  | 2.816571  | 3.343055  |
| H  | 2.953803  | 2.648571  | 3.061943  |
| H  | 1.863479  | 1.291599  | 2.723121  |
| C  | 0.095675  | 3.409315  | -1.049517 |
| H  | 1.047072  | 3.393346  | -1.592978 |
| H  | -0.638885 | 2.967975  | -1.728491 |
| C  | -0.296801 | 4.842842  | -0.681713 |
| H  | -1.282110 | 4.884079  | -0.208895 |
| H  | -0.338573 | 5.461812  | -1.582168 |
| H  | 0.421585  | 5.306392  | -0.000280 |
| O  | 0.101206  | -0.230814 | -2.153045 |
| P  | 2.450290  | -0.730098 | -0.406240 |
| P  | -0.183659 | -2.399934 | 0.393564  |
| P  | -2.354606 | 0.583604  | 0.049037  |
| P  | 0.314440  | 2.219527  | 0.364112  |
| Mo | 0.024550  | -0.094841 | -0.436903 |
| H  | 0.495156  | -0.175795 | 1.373863  |

Sum of electronic and zero-point Energies= -2300.077126  
Sum of electronic and thermal Free Energies= -2300.146722

## 78 TS1

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 2.619179  | 2.003493  | -1.136065 |
| H | 2.768801  | 1.420105  | -2.049602 |
| H | 3.004646  | 3.010135  | -1.323155 |
| C | 3.332052  | 1.339820  | 0.041849  |
| H | 3.260839  | 1.964078  | 0.938736  |
| H | 4.395727  | 1.191925  | -0.165119 |
| C | 3.142208  | -0.609274 | 2.162751  |
| H | 4.226365  | -0.453353 | 2.161340  |
| H | 2.700476  | 0.167640  | 2.794268  |
| C | 2.802998  | -1.998806 | 2.707894  |
| H | 1.722759  | -2.142314 | 2.791372  |
| H | 3.233279  | -2.128519 | 3.704537  |
| H | 3.201638  | -2.794166 | 2.071062  |
| C | 3.419910  | -1.516334 | -0.605702 |
| H | 2.874331  | -2.458594 | -0.495585 |
| H | 3.243599  | -1.193772 | -1.637384 |
| C | 4.912715  | -1.706765 | -0.329795 |
| H | 5.470907  | -0.770103 | -0.418197 |
| H | 5.340561  | -2.408492 | -1.051288 |
| H | 5.094183  | -2.111975 | 0.668925  |
| C | 0.606138  | 3.511948  | 0.313701  |
| H | -0.424965 | 3.499765  | 0.678782  |
| H | 1.223190  | 3.267682  | 1.184836  |
| C | 0.962581  | 4.889745  | -0.248727 |
| H | 0.278101  | 5.192741  | -1.045058 |
| H | 1.979726  | 4.919380  | -0.650022 |
| H | 0.902823  | 5.644583  | 0.540433  |
| C | 0.101802  | 2.643253  | -2.437792 |
| H | -0.950716 | 2.886504  | -2.259184 |
| H | 0.607733  | 3.578109  | -2.699100 |
| C | 0.226322  | 1.631947  | -3.579487 |
| H | -0.238611 | 2.029421  | -4.485820 |
| H | -0.261011 | 0.688297  | -3.324068 |
| H | 1.272032  | 1.414604  | -3.815156 |
| C | -3.152336 | -1.511022 | -0.417473 |
| H | -3.414019 | -0.923184 | -1.304327 |
| H | -3.796706 | -2.394919 | -0.410017 |
| C | -3.345106 | -0.678311 | 0.854027  |
| H | -3.155874 | -1.291646 | 1.741149  |
| H | -4.370824 | -0.306573 | 0.930753  |
| C | -2.940097 | 2.016925  | -0.143227 |
| H | -2.229850 | 2.844289  | -0.231118 |
| H | -5.037740 | 1.739213  | 0.423459  |
| H | -4.234906 | 3.067522  | 1.270318  |

|    |           |           |           |
|----|-----------|-----------|-----------|
| C  | -1.286479 | -3.586882 | 0.364792  |
| H  | -1.474289 | -3.321017 | 1.410432  |
| H  | -2.114886 | -4.226453 | 0.043149  |
| C  | 0.046415  | -4.327666 | 0.231591  |
| H  | 0.237272  | -4.626055 | -0.803361 |
| H  | 0.041297  | -5.233905 | 0.843090  |
| H  | 0.881877  | -3.703612 | 0.560382  |
| C  | -1.226984 | -2.539643 | -2.367229 |
| H  | -1.367541 | -1.628057 | -2.956914 |
| H  | -0.183365 | -2.833741 | -2.514121 |
| C  | -2.185536 | -3.645063 | -2.816200 |
| H  | -2.042640 | -3.856348 | -3.879568 |
| H  | -3.231949 | -3.359317 | -2.676350 |
| H  | -2.014516 | -4.577917 | -2.272160 |
| Mo | 0.081129  | -0.144855 | 0.135730  |
| H  | -0.822821 | 0.159340  | -1.278051 |
| H  | 0.512245  | 0.826900  | 1.507803  |
| H  | 0.552278  | -1.565656 | 0.996971  |
| H  | 0.857881  | -0.819373 | -1.233895 |
| H  | -0.210789 | -1.074568 | 1.575411  |
| P  | 0.787555  | 2.053556  | -0.822562 |
| P  | 2.524806  | -0.285242 | 0.449215  |
| P  | -1.367983 | -2.009333 | -0.604496 |
| P  | -2.121089 | 0.723984  | 0.911739  |
| C  | -2.293938 | 1.385068  | 2.630577  |
| H  | -1.797928 | 2.362140  | 2.633135  |
| H  | -3.357515 | 1.562601  | 2.820382  |
| C  | -1.697681 | 0.488327  | 3.717399  |
| H  | -2.171683 | -0.498005 | 3.733750  |
| H  | -1.848470 | 0.940282  | 4.701662  |
| H  | -0.626355 | 0.346336  | 3.561306  |
| H  | -3.006206 | 1.573706  | -1.142997 |
| C  | -4.303249 | 2.541912  | 0.314682  |
| H  | -4.700611 | 3.247747  | -0.420305 |

Sum of electronic and zero-point Energies= -2227.167198  
Sum of electronic and thermal Free Energies= -2227.236354

78

#### TS2B

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 2.319131  | -2.710076 | -0.073972 |
| H | 1.975582  | -3.271257 | -0.950872 |
| H | 3.342181  | -3.039532 | 0.128421  |
| C | 1.404339  | -2.989155 | 1.123323  |
| H | 1.785411  | -2.485540 | 2.017692  |
| H | 1.360589  | -4.060696 | 1.344810  |
| C | -0.876655 | 3.162792  | 0.768368  |
| H | -1.226881 | 2.895272  | 1.770912  |
| H | -0.626226 | 4.227807  | 0.786187  |
| C | -1.955186 | 2.863399  | -0.269710 |
| H | -1.628868 | 3.191305  | -1.262946 |
| H | -2.882467 | 3.396822  | -0.043038 |
| C | 3.052711  | -0.975771 | -2.253524 |
| H | 3.990179  | -1.536902 | -2.184502 |
| H | 2.372878  | -1.561384 | -2.878216 |
| C | 3.281806  | 0.403408  | -2.874335 |
| H | 2.341614  | 0.955530  | -2.952828 |
| H | 3.705441  | 0.306268  | -3.877568 |
| H | 3.980977  | 1.001947  | -2.281365 |
| C | 3.618939  | -0.144804 | 0.478206  |
| H | 3.604962  | 0.925781  | 0.251679  |
| H | 3.274776  | -0.238979 | 1.513730  |
| C | 5.038260  | -0.694303 | 0.321260  |
| H | 5.089969  | -1.766439 | 0.529421  |
| H | 5.714542  | -0.194714 | 1.020850  |
| H | 5.427960  | -0.528163 | -0.686222 |
| C | -1.160659 | -2.538698 | 2.440469  |
| H | -2.222033 | -2.343806 | 2.258414  |
| H | -1.070265 | -3.595038 | 2.721388  |
| C | -0.653177 | -1.633402 | 3.562021  |
| H | 0.408715  | -1.795007 | 3.769772  |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | -1.201919 | -1.834721 | 4.486187  |
| H  | -0.789954 | -0.581559 | 3.304646  |
| C  | -1.073624 | -3.704694 | -0.180861 |
| H  | -0.571421 | -3.730626 | -1.152331 |
| H  | -0.826649 | -4.643680 | 0.327945  |
| C  | -2.589256 | -3.579283 | -0.360855 |
| H  | -3.115014 | -3.706962 | 0.588822  |
| H  | -2.956311 | -4.348675 | -1.045411 |
| H  | -2.865322 | -2.605474 | -0.770418 |
| C  | -3.522932 | 0.672208  | 0.903753  |
| H  | -3.784448 | -0.384668 | 0.778941  |
| H  | -2.976416 | 0.732982  | 1.850309  |
| C  | -4.778690 | 1.546142  | 0.945120  |
| H  | -4.538729 | 2.586755  | 1.179236  |
| H  | -5.461905 | 1.188516  | 1.720820  |
| H  | -5.323148 | 1.534263  | -0.002520 |
| C  | -3.283968 | 1.019365  | -1.988795 |
| H  | -3.868107 | 1.946428  | -2.001013 |
| H  | -2.569258 | 1.086475  | -2.815298 |
| C  | -4.214802 | -0.182195 | -2.173996 |
| H  | -3.670809 | -1.127027 | -2.167684 |
| H  | -4.732127 | -0.106970 | -3.134557 |
| H  | -4.978815 | -0.226616 | -1.393981 |
| C  | 1.604407  | 2.450462  | 2.025498  |
| H  | 2.653285  | 2.214970  | 1.822667  |
| H  | 1.556884  | 3.527516  | 2.216293  |
| C  | 1.116551  | 1.675676  | 3.247996  |
| H  | 0.055134  | 1.852687  | 3.446368  |
| H  | 1.674585  | 1.983100  | 4.136825  |
| H  | 1.251256  | 0.601049  | 3.114757  |
| C  | 1.606912  | 3.166159  | -0.776118 |
| H  | 2.449429  | 2.545340  | -1.093532 |
| H  | 0.952256  | 3.240413  | -1.650366 |
| C  | 2.089845  | 4.547979  | -0.329444 |
| H  | 1.268887  | 5.177251  | 0.026454  |
| H  | 2.555439  | 5.068572  | -1.171126 |
| H  | 2.834332  | 4.483275  | 0.467842  |
| P  | 2.287133  | -0.907488 | -0.568552 |
| P  | -0.311059 | -2.326980 | 0.809403  |
| P  | -2.275184 | 1.034654  | -0.422824 |
| P  | 0.658337  | 2.153742  | 0.456971  |
| Mo | -0.045844 | -0.067311 | -0.235820 |
| H  | 0.847341  | -0.161274 | 1.176069  |
| H  | -0.873836 | 0.174767  | 1.197780  |
| H  | 0.119465  | 1.032530  | -1.571517 |
| C  | -0.175289 | -1.203997 | -2.411984 |
| O  | -1.298507 | -1.423882 | -2.477873 |

Sum of electronic and zero-point Energies= -2339.284475  
Sum of electronic and thermal Free Energies= -2339.354456

79

#### TS2

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -0.322875 | 0.975229  | -3.099679 |
| H | 0.450345  | -1.429498 | -1.553190 |
| C | 2.064386  | -2.934107 | 0.376509  |
| H | 1.960548  | -3.512555 | -0.547265 |
| H | 2.964640  | -3.296078 | 0.879016  |
| C | 0.827046  | -3.104932 | 1.259420  |
| H | 0.964405  | -2.576674 | 2.207767  |
| H | 0.643283  | -4.159793 | 1.488999  |
| C | -0.698616 | 3.244003  | 0.617788  |
| H | -1.066628 | 3.226391  | 1.648069  |
| H | -0.345219 | 4.260648  | 0.424969  |
| C | -1.813450 | 2.837860  | -0.344140 |
| H | -1.512558 | 3.000558  | -1.385291 |
| H | -2.715707 | 3.433124  | -0.177331 |
| C | 3.401634  | -1.320217 | -1.591040 |
| H | 4.197044  | -2.027379 | -1.332351 |
| H | 2.801103  | -1.784170 | -2.379774 |
| C | 3.995547  | 0.001354  | -2.084521 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 3.211813  | 0.696427  | -2.393190 |
| H  | 4.645612  | -0.172012 | -2.946528 |
| H  | 4.602456  | 0.486125  | -1.313191 |
| C  | 3.394443  | -0.465899 | 1.184971  |
| H  | 3.568324  | 0.580870  | 0.917828  |
| H  | 2.808887  | -0.454583 | 2.110191  |
| C  | 4.736778  | -1.169744 | 1.401798  |
| H  | 4.611265  | -2.213142 | 1.701637  |
| H  | 5.299023  | -0.669940 | 2.195995  |
| H  | 5.354937  | -1.151297 | 0.500680  |
| C  | -2.006534 | -2.596941 | 1.662522  |
| H  | -2.904802 | -2.138535 | 1.238538  |
| H  | -2.205215 | -3.672944 | 1.726467  |
| C  | -1.708035 | -2.033372 | 3.054083  |
| H  | -0.929749 | -2.612531 | 3.557561  |
| H  | -2.602498 | -2.072675 | 3.682109  |
| H  | -1.375672 | -0.994368 | 3.004804  |
| C  | -1.035783 | -3.707704 | -0.809517 |
| H  | -0.168884 | -3.765953 | -1.474266 |
| H  | -1.080986 | -4.647505 | -0.246258 |
| C  | -2.307452 | -3.522128 | -1.634501 |
| H  | -3.200213 | -3.514522 | -1.002475 |
| H  | -2.419441 | -4.344346 | -2.346794 |
| H  | -2.274975 | -2.587912 | -2.200621 |
| C  | -2.918521 | 0.972801  | 1.554800  |
| H  | -3.256399 | -0.055032 | 1.704828  |
| H  | -2.100428 | 1.116456  | 2.265216  |
| C  | -4.059810 | 1.954785  | 1.828053  |
| H  | -3.726277 | 2.994857  | 1.776395  |
| H  | -4.464301 | 1.792547  | 2.831433  |
| H  | -4.883265 | 1.832536  | 1.119380  |
| C  | -3.614082 | 0.774874  | -1.257124 |
| H  | -4.248977 | 1.663659  | -1.179091 |
| H  | -3.226787 | 0.743977  | -2.278141 |
| C  | -4.424962 | -0.486685 | -0.954683 |
| H  | -3.793818 | -1.375807 | -0.976496 |
| H  | -5.211344 | -0.619741 | -1.702477 |
| H  | -4.908687 | -0.432379 | 0.024609  |
| C  | 1.473600  | 2.240401  | 2.222398  |
| H  | 2.519274  | 1.930258  | 2.173594  |
| H  | 1.473871  | 3.310524  | 2.454010  |
| C  | 0.739240  | 1.461171  | 3.313480  |
| H  | -0.300580 | 1.786785  | 3.413364  |
| H  | 1.224309  | 1.619875  | 4.280559  |
| H  | 0.730358  | 0.390029  | 3.099543  |
| C  | 1.965294  | 2.969440  | -0.556384 |
| H  | 2.771388  | 2.264188  | -0.775479 |
| H  | 1.455416  | 3.143029  | -1.508632 |
| C  | 2.531238  | 4.279300  | -0.000317 |
| H  | 1.745135  | 4.995238  | 0.256070  |
| H  | 3.170659  | 4.753485  | -0.750071 |
| H  | 3.138810  | 4.112789  | 0.892389  |
| O  | -1.275278 | 1.100749  | -3.757749 |
| O  | 0.722803  | 0.951732  | -2.544382 |
| P  | 2.271402  | -1.151540 | -0.130875 |
| P  | -0.655296 | -2.360890 | 0.415016  |
| P  | -2.154666 | 1.015047  | -0.140980 |
| P  | 0.751249  | 2.064470  | 0.521638  |
| Mo | -0.004106 | -0.212274 | -0.456289 |
| H  | -0.137701 | -0.359656 | 1.228406  |
| H  | -1.164404 | -0.720905 | -1.589816 |

Sum of electronic and zero-point Energies= -2414.555902  
Sum of electronic and thermal Free Energies= -2414.625505

79

#### TS2-outer-sphere

|   |           |          |           |
|---|-----------|----------|-----------|
| C | -3.395933 | 0.440045 | -1.369843 |
| H | -3.153661 | 0.607892 | -2.425208 |
| H | -4.415041 | 0.045837 | -1.334752 |
| C | -3.266687 | 1.738143 | -0.578885 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | -3.662900 | 1.620259  | 0.435061  |
| H  | -3.814251 | 2.561584  | -1.048738 |
| C  | 2.360876  | -1.009041 | 1.926054  |
| H  | 2.433186  | 0.033606  | 2.249198  |
| H  | 2.691886  | -1.646650 | 2.751068  |
| C  | 3.207864  | -1.248987 | 0.671561  |
| H  | 3.375355  | -2.318495 | 0.510381  |
| H  | 4.190873  | -0.780724 | 0.767052  |
| C  | -2.348324 | -2.116964 | -2.057590 |
| H  | -3.417935 | -2.283465 | -2.221494 |
| H  | -1.948760 | -1.668082 | -2.973531 |
| C  | -1.636812 | -3.436473 | -1.762843 |
| H  | -0.564525 | -3.279722 | -1.623358 |
| H  | -1.771490 | -4.134822 | -2.593100 |
| H  | -2.030109 | -3.916140 | -0.861532 |
| C  | -3.053957 | -1.535108 | 0.725786  |
| H  | -2.366584 | -2.254281 | 1.178719  |
| H  | -3.134524 | -0.710716 | 1.443074  |
| C  | -4.415936 | -2.188235 | 0.481495  |
| H  | -5.136957 | -1.489072 | 0.049707  |
| H  | -4.836457 | -2.544225 | 1.426405  |
| H  | -4.339332 | -3.048898 | -0.187909 |
| C  | -1.518350 | 3.691158  | 0.660034  |
| H  | -0.583494 | 4.226586  | 0.491590  |
| H  | -2.341692 | 4.312111  | 0.286722  |
| C  | -1.693502 | 3.374088  | 2.147320  |
| H  | -2.539873 | 2.704473  | 2.332849  |
| H  | -1.878706 | 4.296387  | 2.703814  |
| H  | -0.794568 | 2.910609  | 2.556588  |
| C  | -0.965415 | 2.760723  | -2.116214 |
| H  | -1.756913 | 2.496706  | -2.823912 |
| H  | -0.835120 | 3.845697  | -2.152848 |
| C  | 0.341153  | 2.028565  | -2.454014 |
| H  | 1.161328  | 2.395280  | -1.838540 |
| H  | 0.618467  | 2.134260  | -3.507317 |
| H  | 0.199027  | 0.921849  | -2.341426 |
| C  | 3.622277  | 0.552203  | -1.644090 |
| H  | 4.459533  | -0.103810 | -1.912511 |
| H  | 3.190221  | 0.905940  | -2.585538 |
| C  | 4.117257  | 1.730391  | -0.806609 |
| H  | 3.343524  | 2.481916  | -0.660106 |
| H  | 4.955260  | 2.210201  | -1.319746 |
| H  | 4.476619  | 1.413977  | 0.176935  |
| C  | 2.465941  | -2.014737 | -2.040525 |
| H  | 3.513295  | -2.333068 | -2.100300 |
| H  | 1.902426  | -2.835138 | -1.587041 |
| C  | 1.911113  | -1.707684 | -3.433132 |
| H  | 0.862170  | -1.401061 | -3.375308 |
| H  | 1.966474  | -2.596278 | -4.067487 |
| H  | 2.474923  | -0.914366 | -3.931117 |
| C  | -0.229638 | -1.211968 | 3.217902  |
| H  | -1.196428 | -1.720810 | 3.139711  |
| H  | 0.391506  | -1.817347 | 3.885146  |
| C  | -0.406612 | 0.197935  | 3.784581  |
| H  | 0.499082  | 0.799972  | 3.684288  |
| H  | -0.668248 | 0.141675  | 4.844864  |
| H  | -1.210557 | 0.726206  | 3.268908  |
| C  | 0.500132  | -3.124903 | 1.187504  |
| H  | -0.550868 | -3.365081 | 1.004888  |
| H  | 1.000784  | -3.266503 | 0.226321  |
| C  | 1.090163  | -4.064913 | 2.242544  |
| H  | 2.142782  | -3.846189 | 2.441833  |
| H  | 1.031874  | -5.100436 | 1.895105  |
| H  | 0.548996  | -4.006636 | 3.189727  |
| P  | -2.167800 | -0.821307 | -0.747394 |
| P  | -1.478249 | 2.200066  | -0.422990 |
| P  | 2.374600  | -0.583849 | -0.863866 |
| P  | 0.567271  | -1.301479 | 1.551390  |
| Mo | 0.004133  | 0.210198  | -0.279827 |
| H  | -0.841621 | 0.380164  | 1.162995  |
| H  | 1.135240  | 1.367582  | 0.574160  |



|   |           |           |           |
|---|-----------|-----------|-----------|
| H | -0.019193 | -1.118429 | -1.324725 |
| C | 1.611278  | 2.540361  | 1.177994  |
| O | 1.949129  | 2.224767  | 2.292249  |
| O | 1.532460  | 3.445522  | 0.377988  |

Sum of electronic and zero-point Energies= -2414.521606  
Sum of electronic and thermal Free Energies= -2414.591660

78

**TS3B**

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -0.320424 | -1.939176 | -1.492960 |
| H | -0.581742 | -0.141634 | -2.019807 |
| C | -3.199024 | 0.612189  | 1.361849  |
| H | -3.784902 | 1.478593  | 1.043970  |
| H | -3.880929 | -0.034476 | 1.920741  |
| C | -2.033408 | 1.057906  | 2.263809  |
| H | -1.681753 | 0.215325  | 2.863545  |
| H | -2.365020 | 1.844222  | 2.947798  |
| C | 3.179102  | -0.962022 | 0.817165  |
| H | 3.265487  | -0.052556 | 1.421312  |
| H | 3.920444  | -1.673327 | 1.192384  |
| C | 3.394519  | -0.648273 | -0.665251 |
| H | 3.373990  | -1.567296 | -1.259904 |
| H | 4.365321  | -0.176456 | -0.837966 |
| C | -3.618122 | 0.529811  | -1.526428 |
| H | -4.666050 | 0.583559  | -1.212210 |
| H | -3.245361 | 1.557106  | -1.591205 |
| C | -3.493178 | -0.164177 | -2.885041 |
| H | -2.452791 | -0.195617 | -3.219950 |
| H | -4.073502 | 0.372246  | -3.640429 |
| H | -3.863877 | -1.192449 | -2.850812 |
| C | -3.436228 | -1.968071 | -0.036880 |
| H | -3.051161 | -2.569335 | -0.866274 |
| H | -3.016258 | -2.405976 | 0.874923  |
| C | -4.966015 | -1.994582 | -0.009499 |
| H | -5.375106 | -1.371503 | 0.791651  |
| H | -5.319577 | -3.015731 | 0.157849  |
| H | -5.394973 | -1.650649 | -0.954012 |
| C | 0.643728  | 2.412778  | 2.362728  |
| H | 1.407372  | 1.666498  | 2.583550  |
| H | 1.136545  | 3.181871  | 1.758078  |
| C | 0.096498  | 3.024058  | 3.655030  |
| H | -0.684586 | 3.764049  | 3.458476  |
| H | 0.896287  | 3.530772  | 4.202419  |
| H | -0.323089 | 2.259821  | 4.314889  |
| C | -1.385065 | 3.156561  | 0.480032  |
| H | -2.314492 | 2.829449  | 0.008102  |
| H | -1.668419 | 3.810756  | 1.312465  |
| C | -0.547674 | 3.924169  | -0.540137 |
| H | 0.383816  | 4.300555  | -0.107708 |
| H | -1.105133 | 4.788914  | -0.910898 |
| H | -0.302540 | 3.291734  | -1.397953 |
| C | 2.637856  | 2.140403  | -0.832982 |
| H | 1.837058  | 2.834145  | -1.093978 |
| H | 2.700231  | 2.149197  | 0.260616  |
| C | 3.960324  | 2.593619  | -1.457120 |
| H | 4.787803  | 1.923466  | -1.208986 |
| H | 4.227474  | 3.588547  | -1.089386 |
| H | 3.892624  | 2.654697  | -2.546116 |
| C | 2.235710  | 0.369647  | -3.124630 |
| H | 3.296509  | 0.505799  | -3.360355 |
| H | 1.965328  | -0.650624 | -3.415762 |
| C | 1.378370  | 1.386312  | -3.883959 |
| H | 0.314994  | 1.240697  | -3.677177 |
| H | 1.529708  | 1.279190  | -4.961167 |
| H | 1.637560  | 2.414744  | -3.614760 |
| C | 1.276288  | -1.937185 | 2.845043  |
| H | 0.366364  | -2.539749 | 2.942807  |
| H | 2.113658  | -2.575182 | 3.142824  |
| C | 1.206425  | -0.719980 | 3.762656  |
| H | 2.105681  | -0.101891 | 3.687361  |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 1.121183  | -1.042224 | 4.804112  |
| H  | 0.341001  | -0.096193 | 3.538178  |
| C  | 1.596930  | -3.367261 | 0.390608  |
| H  | 0.585627  | -3.783779 | 0.434720  |
| H  | 1.835274  | -3.283405 | -0.672837 |
| C  | 2.593760  | -4.297111 | 1.087559  |
| H  | 3.607585  | -3.886509 | 1.096080  |
| H  | 2.635603  | -5.250947 | 0.554286  |
| H  | 2.306253  | -4.514249 | 2.118669  |
| O  | -0.479284 | -2.863912 | -2.182800 |
| P  | -2.645176 | -0.300243 | -0.185415 |
| P  | -0.602015 | 1.633698  | 1.215753  |
| P  | 2.025117  | 0.447738  | -1.288769 |
| P  | 1.460139  | -1.625616 | 1.028077  |
| Mo | -0.196458 | -0.306424 | -0.381972 |
| H  | -0.922529 | -1.191004 | 0.911292  |
| H  | -0.570714 | 1.086591  | -1.302458 |

Sum of electronic and zero-point Energies= -2339.344278  
Sum of electronic and thermal Free Energies= -2339.413876

79

**TS3**

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -2.281708 | -2.754118 | -0.085673 |
| H | -1.971286 | -3.319295 | 0.800735  |
| H | -3.294827 | -3.085329 | -0.330541 |
| C | -1.316700 | -3.023534 | -1.246296 |
| H | -1.676949 | -2.540130 | -2.160098 |
| H | -1.239195 | -4.096702 | -1.450109 |
| C | 1.010905  | 3.073825  | -1.087530 |
| H | 1.328694  | 2.724631  | -2.075506 |
| H | 0.828166  | 4.149654  | -1.166809 |
| C | 2.086806  | 2.773038  | -0.043566 |
| H | 1.810032  | 3.218063  | 0.918939  |
| H | 3.048131  | 3.208469  | -0.330872 |
| C | -3.002617 | -1.039726 | 2.125037  |
| H | -3.885620 | -1.686890 | 2.097066  |
| H | -2.267113 | -1.551129 | 2.755852  |
| C | -3.353799 | 0.333224  | 2.706434  |
| H | -2.486716 | 0.998484  | 2.712717  |
| H | -3.710212 | 0.234213  | 3.735053  |
| H | -4.146088 | 0.819113  | 2.129383  |
| C | -3.638744 | -0.193552 | -0.588219 |
| H | -3.630934 | 0.873885  | -0.344804 |
| H | -3.311845 | -0.268571 | -1.630691 |
| C | -5.044200 | -0.771584 | -0.411484 |
| H | -5.082356 | -1.837328 | -0.652691 |
| H | -5.746694 | -0.262459 | -1.077454 |
| H | -5.411105 | -0.645237 | 0.610290  |
| C | 1.305831  | -2.587384 | -2.444431 |
| H | 2.355524  | -2.376679 | -2.218644 |
| H | 1.236662  | -3.653734 | -2.690688 |
| C | 0.844224  | -1.725906 | -3.620771 |
| H | -0.201646 | -1.915483 | -3.879526 |
| H | 1.446004  | -1.942884 | -4.507513 |
| H | 0.944555  | -0.663391 | -3.389144 |
| C | 1.074358  | -3.603624 | 0.256094  |
| H | 0.527524  | -3.537574 | 1.201448  |
| H | 0.849825  | -4.584413 | -0.178649 |
| C | 2.576985  | -3.454645 | 0.505439  |
| H | 3.155710  | -3.627263 | -0.406023 |
| H | 2.914652  | -4.180378 | 1.250347  |
| H | 2.817685  | -2.457299 | 0.877056  |
| C | 3.566094  | 0.376457  | -0.914465 |
| H | 3.663274  | -0.703106 | -0.757415 |
| H | 3.116257  | 0.493607  | -1.906568 |
| C | 4.934126  | 1.058209  | -0.847874 |
| H | 4.862910  | 2.136264  | -1.016844 |
| H | 5.596384  | 0.652132  | -1.617863 |
| H | 5.420972  | 0.901974  | 0.118291  |
| C | 3.164497  | 1.013596  | 1.911248  |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 3.960527  | 1.759473  | 1.811613  |
| H  | 2.455184  | 1.418365  | 2.641162  |
| C  | 3.744691  | -0.314783 | 2.402087  |
| H  | 2.966783  | -1.064856 | 2.560383  |
| H  | 4.268477  | -0.171699 | 3.351173  |
| H  | 4.464683  | -0.727619 | 1.689783  |
| C  | -1.556934 | 2.424320  | -2.242070 |
| H  | -2.600342 | 2.208998  | -1.990127 |
| H  | -1.503995 | 3.491414  | -2.481419 |
| C  | -1.120639 | 1.591140  | -3.446812 |
| H  | -0.071907 | 1.766973  | -3.704514 |
| H  | -1.722770 | 1.850997  | -4.321869 |
| H  | -1.238024 | 0.523199  | -3.253853 |
| C  | -1.432674 | 3.272829  | 0.525402  |
| H  | -2.306977 | 2.710719  | 0.869371  |
| H  | -0.762127 | 3.331286  | 1.388880  |
| C  | -1.835472 | 4.669232  | 0.046051  |
| H  | -0.980829 | 5.235783  | -0.335022 |
| H  | -2.258653 | 5.240356  | 0.877273  |
| H  | -2.591448 | 4.628782  | -0.741953 |
| P  | -2.278158 | -0.952433 | 0.419105  |
| P  | 0.367806  | -2.312122 | -0.874112 |
| P  | 2.266314  | 0.942132  | 0.283933  |
| P  | -0.570346 | 2.170094  | -0.693833 |
| Mo | 0.003426  | -0.051375 | 0.093948  |
| H  | -0.904670 | -0.240165 | -1.301238 |
| H  | 0.749685  | 0.272079  | -1.375141 |
| H  | -0.304707 | 1.021223  | 1.449928  |
| C  | 0.205278  | -0.230790 | 2.904046  |
| O  | 0.067758  | 0.436949  | 3.849853  |
| O  | 0.433586  | -1.137597 | 2.157492  |

Sum of electronic and zero-point Energies= -2414.561989  
Sum of electronic and thermal Free Energies= -2414.632992

80

#### TS4-I

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -0.138691 | 0.135087  | -2.958630 |
| H | -0.212191 | 0.201743  | -4.050559 |
| C | 2.693160  | -2.314665 | -0.872633 |
| H | 2.330598  | -2.559722 | -1.876943 |
| H | 3.751115  | -2.588747 | -0.834415 |
| C | 1.891397  | -3.070067 | 0.192338  |
| H | 2.292752  | -2.849551 | 1.185981  |
| H | 1.950518  | -4.154262 | 0.052015  |
| C | -1.360215 | 2.622253  | 1.546630  |
| H | -1.252542 | 2.064327  | 2.480287  |
| H | -1.444459 | 3.675654  | 1.830709  |
| C | -2.592646 | 2.152755  | 0.768804  |
| H | -2.835539 | 2.887314  | -0.005324 |
| H | -3.463889 | 2.092061  | 1.425610  |
| C | 3.356072  | 0.225443  | -2.095083 |
| H | 4.328737  | -0.268472 | -2.188315 |
| H | 2.775494  | -0.044438 | -2.983191 |
| C | 3.514406  | 1.745548  | -2.012265 |
| H | 2.536174  | 2.224707  | -1.934208 |
| H | 4.009934  | 2.132291  | -2.906983 |
| H | 4.118543  | 2.041271  | -1.148795 |
| C | 3.583093  | -0.106109 | 0.792002  |
| H | 3.498206  | 0.970513  | 0.966348  |
| H | 3.135957  | -0.584293 | 1.670262  |
| C | 5.053277  | -0.507496 | 0.651584  |
| H | 5.172008  | -1.586261 | 0.519538  |
| H | 5.611386  | -0.226347 | 1.549630  |
| H | 5.528663  | -0.011488 | -0.198712 |
| C | -0.652367 | -3.283571 | 1.680424  |
| H | -1.505735 | -2.660444 | 1.962732  |
| H | -1.074502 | -4.245722 | 1.373340  |
| C | 0.282497  | -3.509002 | 2.875212  |
| H | 1.035999  | -4.267272 | 2.650755  |
| H | -0.292244 | -3.861444 | 3.735887  |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 0.816868  | -2.604893 | 3.179784  |
| C  | -0.565787 | -3.610204 | -1.191412 |
| H  | -0.092198 | -3.299523 | -2.124213 |
| H  | -0.226528 | -4.626611 | -0.959340 |
| C  | -2.085895 | -3.560590 | -1.333982 |
| H  | -2.594365 | -3.875299 | -0.417273 |
| H  | -2.411242 | -4.231178 | -2.134166 |
| H  | -2.413659 | -2.553384 | -1.586410 |
| C  | -3.496106 | -0.639031 | 0.801818  |
| H  | -3.380872 | -1.619449 | 0.329893  |
| H  | -3.093625 | -0.740554 | 1.814998  |
| C  | -4.975040 | -0.247595 | 0.864840  |
| H  | -5.122946 | 0.725200  | 1.342018  |
| H  | -5.535963 | -0.983235 | 1.448918  |
| H  | -5.427053 | -0.202776 | -0.128721 |
| C  | -3.299721 | 0.876537  | -1.669571 |
| H  | -4.197841 | 1.431517  | -1.377655 |
| H  | -2.674468 | 1.570842  | -2.242362 |
| C  | -3.679834 | -0.327377 | -2.533816 |
| H  | -2.796685 | -0.869087 | -2.874330 |
| H  | -4.238209 | 0.002611  | -3.415004 |
| H  | -4.318148 | -1.029330 | -1.989768 |
| C  | 1.530267  | 3.119686  | 1.489940  |
| H  | 2.426281  | 3.048093  | 0.864098  |
| H  | 1.299788  | 4.185275  | 1.588594  |
| C  | 1.790887  | 2.523620  | 2.876080  |
| H  | 0.899799  | 2.573201  | 3.509971  |
| H  | 2.576782  | 3.089837  | 3.383204  |
| H  | 2.116120  | 1.484104  | 2.817570  |
| C  | -0.065696 | 3.591060  | -0.867468 |
| H  | 0.886366  | 3.615982  | -1.404807 |
| H  | -0.779914 | 3.156955  | -1.568240 |
| C  | -0.497033 | 4.999173  | -0.450611 |
| H  | -1.492168 | 5.005538  | 0.002337  |
| H  | -0.538682 | 5.642446  | -1.334147 |
| H  | 0.197615  | 5.463058  | 0.254434  |
| O  | -0.426774 | -0.932003 | -2.343039 |
| O  | 0.238680  | 1.122260  | -2.264261 |
| P  | 2.450572  | -0.478283 | -0.640420 |
| P  | 0.107171  | -2.521096 | 0.147154  |
| P  | -2.333377 | 0.498781  | -0.115818 |
| P  | 0.155177  | 2.355350  | 0.502094  |
| Mo | 0.068223  | -0.017048 | -0.351587 |
| O  | 0.552188  | -0.270849 | 2.109075  |
| C  | -0.414472 | -0.282376 | 2.784763  |
| O  | -1.322578 | -0.299970 | 3.508563  |

Sum of electronic and zero-point Energies= -2601.956285  
Sum of electronic and thermal Free Energies= -2602.029393

79

#### TS4

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -0.222092 | 1.428404  | -2.119643 |
| H | 0.186351  | -1.284516 | -1.765463 |
| C | 2.152496  | -2.883624 | -0.281576 |
| H | 1.847922  | -3.267129 | -1.260762 |
| H | 3.100603  | -3.362339 | -0.024349 |
| C | 1.085127  | -3.166145 | 0.773778  |
| H | 1.394810  | -2.759430 | 1.741362  |
| H | 0.926443  | -4.240849 | 0.910619  |
| C | -0.953625 | 2.723989  | 1.456795  |
| H | -1.190719 | 2.171595  | 2.371683  |
| H | -0.788804 | 3.766575  | 1.745021  |
| C | -2.089072 | 2.601611  | 0.444751  |
| H | -1.867049 | 3.173057  | -0.461084 |
| H | -3.026804 | 2.987653  | 0.853020  |
| C | 3.328684  | -0.858412 | -2.016442 |
| H | 4.043007  | -1.687413 | -2.067945 |
| H | 2.614882  | -0.983205 | -2.834838 |
| C | 4.050926  | 0.486943  | -2.146594 |
| H | 3.344269  | 1.316833  | -2.099418 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 4.560478  | 0.539446  | -3.112343 |
| H  | 4.808918  | 0.621001  | -1.369878 |
| C  | 3.561537  | -0.615159 | 0.879895  |
| H  | 3.763410  | 0.452954  | 0.757696  |
| H  | 3.040676  | -0.721484 | 1.835363  |
| C  | 4.877569  | -1.399290 | 0.892950  |
| H  | 4.719155  | -2.459394 | 1.106227  |
| H  | 5.538317  | -1.007298 | 1.671286  |
| H  | 5.408463  | -1.322793 | -0.059180 |
| C  | -1.522950 | -2.766990 | 1.862536  |
| H  | -2.566481 | -2.517721 | 1.667051  |
| H  | -1.480552 | -3.856717 | 1.976886  |
| C  | -1.042743 | -2.081214 | 3.141640  |
| H  | -0.036303 | -2.401985 | 3.423408  |
| H  | -1.708598 | -2.329469 | 3.972566  |
| H  | -1.024454 | -0.994875 | 3.033589  |
| C  | -1.225647 | -3.639440 | -0.859946 |
| H  | -0.641876 | -3.546012 | -1.780417 |
| H  | -1.016739 | -4.629194 | -0.437827 |
| C  | -2.717744 | -3.493889 | -1.160435 |
| H  | -3.327980 | -3.644926 | -0.265417 |
| H  | -3.027902 | -4.240191 | -1.896518 |
| H  | -2.945374 | -2.509915 | -1.569236 |
| C  | -3.352702 | 0.057520  | 1.188943  |
| H  | -3.573684 | -0.947288 | 0.816795  |
| H  | -2.713299 | -0.070370 | 2.066212  |
| C  | -4.654886 | 0.767202  | 1.569662  |
| H  | -4.474579 | 1.767550  | 1.971533  |
| H  | -5.177808 | 0.195990  | 2.342102  |
| H  | -5.332458 | 0.861650  | 0.717893  |
| C  | -3.504807 | 1.105213  | -1.514829 |
| H  | -4.329667 | 1.698251  | -1.106139 |
| H  | -2.979279 | 1.729921  | -2.237612 |
| C  | -4.031094 | -0.160589 | -2.185375 |
| H  | -3.218846 | -0.714820 | -2.663086 |
| H  | -4.749500 | 0.103691  | -2.966045 |
| H  | -4.541513 | -0.825803 | -1.481972 |
| C  | 1.685500  | 1.969351  | 2.303222  |
| H  | 2.717714  | 1.817439  | 1.978355  |
| H  | 1.643568  | 2.969758  | 2.745546  |
| C  | 1.287573  | 0.922814  | 3.344489  |
| H  | 0.285116  | 1.108244  | 3.740543  |
| H  | 1.981533  | 0.948455  | 4.189212  |
| H  | 1.292016  | -0.085414 | 2.926396  |
| C  | 1.366876  | 3.384874  | -0.222407 |
| H  | 2.223347  | 2.949498  | -0.746815 |
| H  | 0.635325  | 3.623062  | -0.999864 |
| C  | 1.777189  | 4.650493  | 0.535868  |
| H  | 0.944164  | 5.083536  | 1.097493  |
| H  | 2.117931  | 5.409197  | -0.173819 |
| H  | 2.596219  | 4.463725  | 1.234720  |
| O  | -0.911658 | 2.264980  | -2.654185 |
| O  | 0.958357  | 1.006173  | -2.182593 |
| P  | 2.352511  | -1.052833 | -0.457269 |
| P  | -0.548063 | -2.387897 | 0.329805  |
| P  | -2.310914 | 0.849245  | -0.125537 |
| P  | 0.624809  | 2.015215  | 0.783486  |
| Mo | -0.003944 | -0.097330 | -0.558994 |
| H  | -0.094219 | -0.245747 | 1.136043  |
| H  | -1.019917 | -0.742048 | -1.757595 |

Sum of electronic and zero-point Energies= -2414.566195  
Sum of electronic and thermal Free Energies= -2414.635363

80

# TS5-I

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -0.185120 | -0.069312 | -2.689532 |
| H | -0.265586 | -0.022818 | -3.784521 |
| C | 3.156227  | -1.651146 | -0.891499 |
| H | 2.793972  | -2.032509 | -1.852329 |
| H | 4.249026  | -1.642275 | -0.939913 |

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 2.663956  | -2.531882 | 0.258825  |
| H | 3.017937  | -2.135323 | 1.214263  |
| H | 3.045960  | -3.553878 | 0.165868  |
| C | -2.163426 | 2.507029  | 1.018196  |
| H | -2.291981 | 1.986871  | 1.971300  |
| H | -2.489755 | 3.540111  | 1.165816  |
| C | -2.994194 | 1.830933  | -0.078620 |
| H | -2.862802 | 2.353471  | -1.032224 |
| H | -4.061037 | 1.870066  | 0.160325  |
| C | 3.049364  | 0.788770  | -2.375920 |
| H | 4.106104  | 0.531024  | -2.507892 |
| H | 2.501291  | 0.243629  | -3.151705 |
| C | 2.833991  | 2.295231  | -2.524789 |
| H | 1.773644  | 2.541998  | -2.460586 |
| H | 3.205708  | 2.636185  | -3.494935 |
| H | 3.370895  | 2.857470  | -1.754800 |
| C | 3.564414  | 0.919249  | 0.513028  |
| H | 3.291498  | 1.979458  | 0.463942  |
| H | 3.201797  | 0.561011  | 1.480483  |
| C | 5.077914  | 0.745366  | 0.371845  |
| H | 5.378908  | -0.298322 | 0.495882  |
| H | 5.595042  | 1.325641  | 1.141142  |
| H | 5.443201  | 1.089402  | -0.600192 |
| C | 0.450786  | -3.392289 | 1.964026  |
| H | -0.623812 | -3.279616 | 2.137023  |
| H | 0.637482  | -4.463292 | 1.822126  |
| C | 1.234657  | -2.876276 | 3.176576  |
| H | 2.298221  | -3.115790 | 3.094296  |
| H | 0.861645  | -3.358362 | 4.084193  |
| H | 1.134682  | -1.797378 | 3.296281  |
| C | 0.320080  | -3.863657 | -0.876376 |
| H | 0.548890  | -3.480987 | -1.872401 |
| H | 0.946362  | -4.744996 | -0.695715 |
| C | -1.164796 | -4.225486 | -0.771218 |
| H | -1.398740 | -4.698310 | 0.186514  |
| H | -1.446437 | -4.924888 | -1.562867 |
| H | -1.791760 | -3.337144 | -0.874166 |
| C | -3.599799 | -0.953942 | 0.721714  |
| H | -3.262093 | -1.990277 | 0.614032  |
| H | -3.354247 | -0.664620 | 1.745288  |
| C | -5.107792 | -0.843166 | 0.478827  |
| H | -5.463545 | 0.186886  | 0.571387  |
| H | -5.641740 | -1.437242 | 1.225445  |
| H | -5.398972 | -1.213896 | -0.506458 |
| C | -3.255559 | -0.199672 | -2.074121 |
| H | -4.268675 | 0.214068  | -2.022370 |
| H | -2.699665 | 0.440035  | -2.764178 |
| C | -3.290261 | -1.647833 | -2.568014 |
| H | -2.284292 | -2.057006 | -2.667076 |
| H | -3.777806 | -1.695393 | -3.545715 |
| H | -3.855983 | -2.293049 | -1.890492 |
| C | 0.543200  | 3.118518  | 2.043293  |
| H | 1.600987  | 2.909722  | 1.853487  |
| H | 0.419301  | 4.206227  | 2.014032  |
| C | 0.141747  | 2.573478  | 3.416827  |
| H | -0.906085 | 2.784297  | 3.646784  |
| H | 0.746795  | 3.055278  | 4.189978  |
| H | 0.303211  | 1.497912  | 3.481939  |
| C | -0.130208 | 3.729007  | -0.708978 |
| H | 0.952383  | 3.887302  | -0.763742 |
| H | -0.421555 | 3.286326  | -1.662831 |
| C | -0.856932 | 5.056131  | -0.471793 |
| H | -1.941837 | 4.940009  | -0.542460 |
| H | -0.557965 | 5.784713  | -1.230663 |
| H | -0.628013 | 5.489861  | 0.505857  |
| O | -0.051491 | -1.172805 | -2.101552 |
| O | -0.235202 | 0.996313  | -2.006789 |
| P | 2.509105  | 0.084956  | -0.750634 |
| P | 0.809311  | -2.562491 | 0.340252  |
| P | -2.516071 | 0.055079  | -0.392275 |
| P | -0.360727 | 2.421921  | 0.585493  |

|    |           |           |          |
|----|-----------|-----------|----------|
| Mo | 0.041353  | -0.083500 | 0.056711 |
| C  | -0.804397 | -0.382193 | 1.832423 |
| O  | 0.932064  | -0.046269 | 1.676358 |
| O  | -1.269718 | -0.528205 | 2.899463 |

Sum of electronic and zero-point Energies= -2601.961127  
Sum of electronic and thermal Free Energies= -2602.035144

77

# TS5

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 0.796850  | -0.412170 | -2.312391 |
| C | 1.888520  | -3.020667 | 0.021348  |
| H | 1.829384  | -3.217266 | -1.054358 |
| H | 2.748455  | -3.575372 | 0.405779  |
| C | 0.596169  | -3.427637 | 0.725483  |
| H | 0.689516  | -3.285009 | 1.805598  |
| H | 0.351417  | -4.480700 | 0.554875  |
| C | -0.553028 | 3.307030  | 0.457581  |
| H | -0.707390 | 3.264015  | 1.539206  |
| H | -0.274028 | 4.336350  | 0.212925  |
| C | -1.824214 | 2.879491  | -0.273646 |
| H | -1.689209 | 2.927710  | -1.359366 |
| H | -2.664788 | 3.530973  | -0.019178 |
| C | 3.666399  | -0.947756 | -0.817262 |
| H | 4.335303  | -1.786534 | -0.596826 |
| H | 3.349973  | -1.053219 | -1.858329 |
| C | 4.402338  | 0.376691  | -0.612499 |
| H | 3.774632  | 1.230957  | -0.872400 |
| H | 5.286511  | 0.415361  | -1.254054 |
| H | 4.741708  | 0.500656  | 0.420279  |
| C | 2.721060  | -1.032420 | 1.966880  |
| H | 2.954843  | 0.022438  | 2.122497  |
| H | 1.844063  | -1.234367 | 2.590041  |
| C | 3.909026  | -1.904424 | 2.381182  |
| H | 3.678266  | -2.970784 | 2.315939  |
| H | 4.181913  | -1.694080 | 3.419104  |
| H | 4.791336  | -1.712380 | 1.764814  |
| C | -2.147684 | -2.672347 | 1.340354  |
| H | -3.020146 | -2.112655 | 0.994823  |
| H | -2.410347 | -3.735553 | 1.290383  |
| C | -1.785445 | -2.267660 | 2.773111  |
| H | -1.029459 | -2.932219 | 3.199617  |
| H | -2.667068 | -2.324578 | 3.417130  |
| H | -1.398205 | -1.246410 | 2.809613  |
| C | -1.309003 | -3.280299 | -1.418483 |
| H | -0.411156 | -3.324279 | -2.043476 |
| H | -1.540322 | -4.306280 | -1.107682 |
| C | -2.468380 | -2.675716 | -2.209909 |
| H | -3.401965 | -2.713803 | -1.643833 |
| H | -2.620984 | -3.244983 | -3.130636 |
| H | -2.261445 | -1.638794 | -2.484730 |
| C | -2.871095 | 1.156610  | 1.842564  |
| H | -3.255345 | 0.153620  | 2.052612  |
| H | -1.996120 | 1.279759  | 2.488797  |
| C | -3.933408 | 2.216329  | 2.146864  |
| H | -3.530944 | 3.228785  | 2.056173  |
| H | -4.301277 | 2.101979  | 3.170454  |
| H | -4.793389 | 2.137939  | 1.476139  |
| C | -3.704331 | 0.849324  | -0.977889 |
| H | -4.218685 | 1.811659  | -1.072218 |
| H | -3.291525 | 0.608419  | -1.962394 |
| C | -4.688397 | -0.225020 | -0.512171 |
| H | -4.206265 | -1.198613 | -0.409880 |
| H | -5.493113 | -0.338657 | -1.243540 |
| H | -5.148071 | 0.027428  | 0.447040  |
| C | 2.079442  | 2.543108  | 1.378354  |
| H | 2.984183  | 1.971287  | 1.162009  |
| H | 2.349773  | 3.600262  | 1.294413  |
| C | 1.556328  | 2.240563  | 2.788038  |
| H | 0.800722  | 2.967026  | 3.096489  |
| H | 2.372948  | 2.296884  | 3.512926  |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 1.109874  | 1.245252  | 2.848443  |
| C  | 1.541170  | 2.921402  | -1.519925 |
| H  | 2.234303  | 2.193531  | -1.949120 |
| H  | 0.688242  | 2.949753  | -2.207149 |
| C  | 2.201428  | 4.296423  | -1.394809 |
| H  | 1.540883  | 5.034026  | -0.929776 |
| H  | 2.461085  | 4.672685  | -2.388014 |
| H  | 3.123651  | 4.251227  | -0.809502 |
| O  | -0.790121 | 0.286806  | -2.177415 |
| O  | 1.285626  | -0.601219 | -3.363095 |
| P  | 2.145506  | -1.191624 | 0.211496  |
| P  | -0.799605 | -2.364716 | 0.114604  |
| P  | -2.225474 | 1.106769  | 0.107525  |
| P  | 0.862043  | 2.172732  | 0.037809  |
| Mo | -0.045030 | -0.083104 | -0.493466 |
| H  | -0.144025 | -0.000313 | 1.269854  |

Sum of electronic and zero-point Energies= -2413.374377  
Sum of electronic and thermal Free Energies= -2413.444812

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# TS6-II

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -1.352828 | -0.598399 | -2.126206 |
| C | 0.777712  | 3.365111  | -0.862005 |
| H | 1.140078  | 3.159646  | -1.874777 |
| H | 0.533302  | 4.430482  | -0.814644 |
| C | 1.837458  | 2.997651  | 0.183989  |
| H | 1.464286  | 3.183162  | 1.194875  |
| H | 2.730762  | 3.612714  | 0.041419  |
| C | -1.612721 | -2.345229 | 1.702024  |
| H | -1.055112 | -1.870379 | 2.507728  |
| H | -2.447871 | -2.895967 | 2.144260  |
| C | -0.696332 | -3.285909 | 0.915753  |
| H | -1.269741 | -3.860940 | 0.181339  |
| H | -0.225682 | -4.009683 | 1.586544  |
| C | -1.799401 | 2.834431  | -2.116420 |
| H | -1.744772 | 3.919998  | -2.250659 |
| H | -1.321571 | 2.375806  | -2.987795 |
| C | -3.256647 | 2.379435  | -1.990232 |
| H | -3.327203 | 1.308705  | -1.777480 |
| H | -3.794833 | 2.565062  | -2.922940 |
| H | -3.777545 | 2.917958  | -1.193809 |
| C | -1.632221 | 3.104370  | 0.797257  |
| H | -2.568653 | 2.553592  | 0.912772  |
| H | -1.021239 | 2.867402  | 1.669481  |
| C | -1.905627 | 4.608660  | 0.702006  |
| H | -0.980461 | 5.188224  | 0.653052  |
| H | -2.447609 | 4.938861  | 1.592334  |
| H | -2.514211 | 4.866017  | -0.169029 |
| C | 3.372626  | 0.952363  | 1.586227  |
| H | 2.755163  | 1.160574  | 2.461400  |
| H | 3.626985  | -0.110638 | 1.629233  |
| C | 4.641215  | 1.811661  | 1.637490  |
| H | 5.325869  | 1.586031  | 0.816587  |
| H | 5.175566  | 1.624905  | 2.573312  |
| H | 4.408973  | 2.879467  | 1.602170  |
| C | 3.496328  | 1.175942  | -1.315233 |
| H | 2.919466  | 1.374883  | -2.220675 |
| H | 4.195939  | 2.005175  | -1.172024 |
| C | 4.243336  | -0.151184 | -1.460890 |
| H | 4.832911  | -0.390864 | -0.570631 |
| H | 4.932208  | -0.109509 | -2.309030 |
| H | 3.545219  | -0.969077 | -1.644941 |
| C | 2.194549  | -2.619964 | 0.924187  |
| H | 2.959409  | -2.056350 | 0.382050  |
| H | 2.029800  | -2.100992 | 1.872133  |
| C | 2.647577  | -4.065886 | 1.143919  |
| H | 1.895343  | -4.657614 | 1.673059  |
| H | 3.555909  | -4.080118 | 1.752423  |
| H | 2.874272  | -4.570753 | 0.201235  |
| C | 0.807784  | -3.476191 | -1.531366 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 0.935594  | -4.505412 | -1.177975 |
| H  | -0.157892 | -3.430212 | -2.045846 |
| C  | 1.936867  | -3.079446 | -2.482371 |
| H  | 1.813557  | -2.052135 | -2.834198 |
| H  | 1.940391  | -3.740017 | -3.353404 |
| H  | 2.914668  | -3.167281 | -2.000201 |
| C  | -3.429414 | -0.032892 | 1.654747  |
| H  | -3.789031 | 0.795688  | 1.035418  |
| H  | -4.294235 | -0.677831 | 1.836281  |
| C  | -2.880849 | 0.472836  | 2.990200  |
| H  | -2.535108 | -0.350671 | 3.619887  |
| H  | -3.673297 | 0.990875  | 3.537432  |
| H  | -2.056304 | 1.172177  | 2.863109  |
| C  | -3.448663 | -1.939022 | -0.507588 |
| H  | -3.984472 | -1.183450 | -1.092493 |
| H  | -2.849700 | -2.504226 | -1.226119 |
| C  | -4.438612 | -2.883805 | 0.180757  |
| H  | -3.929633 | -3.707074 | 0.688592  |
| H  | -5.101742 | -3.324021 | -0.569048 |
| H  | -5.070642 | -2.374755 | 0.912315  |
| O  | 0.901175  | 0.164919  | -2.153817 |
| O  | -1.984389 | -0.832139 | -3.058943 |
| P  | -0.769224 | 2.366545  | -0.655598 |
| P  | 2.293458  | 1.193363  | 0.090620  |
| P  | 0.634069  | -2.394700 | -0.037722 |
| P  | -2.260192 | -1.011848 | 0.594567  |
| Mo | 0.001647  | -0.032923 | -0.680696 |
| H  | 0.002151  | 0.214470  | 1.194331  |
| C  | 0.409529  | 0.424520  | 2.660660  |
| O  | 0.633501  | -0.670187 | 3.093855  |
| O  | 0.377201  | 1.611923  | 2.826751  |

Sum of electronic and zero-point Energies= -2601.956164  
Sum of electronic and thermal Free Energies= -2602.030577

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#### TS6-I

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -0.287942 | 1.597041  | 2.384205  |
| C | -2.728055 | -2.307596 | 0.803346  |
| H | -2.419532 | -2.518467 | 1.832714  |
| H | -3.778299 | -2.599225 | 0.712889  |
| C | -1.855213 | -3.070546 | -0.199907 |
| H | -2.173769 | -2.848193 | -1.221389 |
| H | -1.951441 | -4.151451 | -0.057769 |
| C | 1.380543  | 2.414404  | -1.776825 |
| H | 1.122332  | 1.798745  | -2.634480 |
| H | 1.534085  | 3.434333  | -2.140577 |
| C | 2.663582  | 1.871633  | -1.133246 |
| H | 3.143543  | 2.646061  | -0.526674 |
| H | 3.370831  | 1.586324  | -1.915365 |
| C | -3.489905 | 0.267029  | 1.943923  |
| H | -4.434574 | -0.275909 | 2.054431  |
| H | -2.906190 | 0.083347  | 2.851610  |
| C | -3.748990 | 1.765451  | 1.757004  |
| H | -2.822535 | 2.312757  | 1.558052  |
| H | -4.197625 | 2.190603  | 2.658210  |
| H | -4.433710 | 1.952758  | 0.925280  |
| C | -3.512295 | -0.110373 | -0.968814 |
| H | -3.457335 | 0.972010  | -1.111487 |
| H | -2.950493 | -0.547328 | -1.797118 |
| C | -4.969201 | -0.579407 | -0.955732 |
| H | -5.047343 | -1.667550 | -0.887159 |
| H | -5.465899 | -0.275491 | -1.881353 |
| H | -5.532652 | -0.148007 | -0.123694 |
| C | 0.802021  | -3.371550 | -1.482760 |
| H | 1.552916  | -2.661353 | -1.828122 |
| H | 1.332724  | -4.243242 | -1.087145 |
| C | -0.087801 | -3.798677 | -2.654476 |
| H | -0.800654 | -4.574963 | -2.365152 |
| H | 0.535221  | -4.202859 | -3.456658 |
| H | -0.652339 | -2.960633 | -3.070663 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| C  | 0.461748  | -3.624519 | 1.399806  |
| H  | -0.119562 | -3.292482 | 2.263701  |
| H  | 0.168472  | -4.655874 | 1.171193  |
| C  | 1.955302  | -3.542704 | 1.717927  |
| H  | 2.571153  | -3.862469 | 0.871533  |
| H  | 2.198353  | -4.193669 | 2.562063  |
| H  | 2.236315  | -2.524814 | 1.989504  |
| C  | 3.600907  | -0.897996 | -0.613822 |
| H  | 3.435237  | -1.782857 | 0.007441  |
| H  | 3.263989  | -1.138934 | -1.622681 |
| C  | 5.079392  | -0.498805 | -0.601328 |
| H  | 5.261590  | 0.411992  | -1.179194 |
| H  | 5.680195  | -1.293959 | -1.051560 |
| H  | 5.454941  | -0.337174 | 0.412258  |
| C  | 3.204486  | 1.007390  | 1.555544  |
| H  | 4.189710  | 1.404670  | 1.288032  |
| H  | 2.601123  | 1.858211  | 1.888195  |
| C  | 3.327835  | -0.025962 | 2.674948  |
| H  | 2.348876  | -0.411924 | 2.968511  |
| H  | 3.791984  | 0.427223  | 3.555331  |
| H  | 3.954765  | -0.869674 | 2.372206  |
| C  | -1.474111 | 3.072199  | -1.564843 |
| H  | -2.353910 | 2.965814  | -0.920798 |
| H  | -1.300434 | 4.147203  | -1.675848 |
| C  | -1.726293 | 2.446295  | -2.939398 |
| H  | -0.881082 | 2.613749  | -3.612256 |
| H  | -2.605937 | 2.904726  | -3.399933 |
| H  | -1.893009 | 1.372473  | -2.866363 |
| C  | 0.335634  | 3.781572  | 0.562531  |
| H  | -0.579171 | 3.961492  | 1.138904  |
| H  | 1.078661  | 3.414586  | 1.276149  |
| C  | 0.830878  | 5.086440  | -0.068542 |
| H  | 1.810344  | 4.961487  | -0.537124 |
| H  | 0.933326  | 5.854207  | 0.703299  |
| H  | 0.142773  | 5.474250  | -0.824306 |
| O  | 0.152814  | -0.747104 | 2.058097  |
| O  | -0.536235 | 1.820511  | 3.474592  |
| P  | -2.533606 | -0.480986 | 0.551741  |
| P  | -0.063607 | -2.581209 | -0.033531 |
| P  | 2.415957  | 0.389556  | -0.013502 |
| P  | -0.046318 | 2.376482  | -0.598852 |
| Mo | -0.034767 | -0.099176 | 0.490108  |
| C  | 0.215345  | -0.459769 | -2.731041 |
| O  | 1.441232  | -0.491759 | -2.798929 |
| O  | -0.502117 | -0.289456 | -1.664149 |
| H  | -0.394905 | -0.569630 | -3.646892 |

Sum of electronic and zero-point Energies= -2601.980620  
Sum of electronic and thermal Free Energies= -2602.054717

77

#### TS6

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 0.239504  | 1.589675  | -2.697628 |
| C | 2.522054  | -2.579991 | -0.040176 |
| H | 2.259613  | -3.067155 | -0.985203 |
| H | 3.536477  | -2.896997 | 0.214376  |
| C | 1.521376  | -2.957443 | 1.054308  |
| H | 1.791708  | -2.485664 | 2.004443  |
| H | 1.497955  | -4.039778 | 1.216898  |
| C | -1.255079 | 2.340487  | 1.797953  |
| H | -1.218168 | 1.583033  | 2.585197  |
| H | -1.258543 | 3.320491  | 2.284205  |
| C | -2.493137 | 2.144607  | 0.923817  |
| H | -2.577087 | 2.953605  | 0.191019  |
| H | -3.407383 | 2.162503  | 1.523089  |
| C | 3.283077  | -0.571852 | -2.003086 |
| H | 4.150221  | -1.239812 | -2.041959 |
| H | 2.552088  | -0.934564 | -2.732015 |
| C | 3.695389  | 0.869380  | -2.316463 |
| H | 2.858670  | 1.561823  | -2.184259 |
| H | 4.034324  | 0.952717  | -3.352160 |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 4.513668  | 1.204548  | -1.672875 |
| C  | 3.671195  | -0.038888 | 0.844436  |
| H  | 3.713779  | 1.031931  | 0.629439  |
| H  | 3.229958  | -0.136847 | 1.839475  |
| C  | 5.082322  | -0.633512 | 0.811279  |
| H  | 5.090216  | -1.687503 | 1.099303  |
| H  | 5.727347  | -0.099799 | 1.514889  |
| H  | 5.536830  | -0.551141 | -0.179488 |
| C  | -1.169114 | -2.757632 | 2.099340  |
| H  | -2.221297 | -2.761032 | 1.805351  |
| H  | -0.918037 | -3.790890 | 2.367430  |
| C  | -0.952076 | -1.827429 | 3.293401  |
| H  | 0.094285  | -1.812470 | 3.610871  |
| H  | -1.554192 | -2.159031 | 4.144140  |
| H  | -1.229008 | -0.799374 | 3.050040  |
| C  | -0.703691 | -3.682437 | -0.605598 |
| H  | -0.022006 | -3.605189 | -1.457020 |
| H  | -0.535531 | -4.653754 | -0.125914 |
| C  | -2.145659 | -3.557523 | -1.098250 |
| H  | -2.867253 | -3.668526 | -0.282869 |
| H  | -2.363978 | -4.338417 | -1.831824 |
| H  | -2.302596 | -2.592772 | -1.582208 |
| C  | -3.633630 | -0.550230 | 0.803575  |
| H  | -3.533780 | -1.534152 | 0.338673  |
| H  | -3.293747 | -0.658507 | 1.838980  |
| C  | -5.091377 | -0.084357 | 0.753538  |
| H  | -5.220747 | 0.918352  | 1.170943  |
| H  | -5.721473 | -0.763105 | 1.335072  |
| H  | -5.475419 | -0.075199 | -0.269403 |
| C  | -3.174069 | 0.989621  | -1.645437 |
| H  | -4.141301 | 1.455773  | -1.429075 |
| H  | -2.534517 | 1.764373  | -2.079999 |
| C  | -3.330535 | -0.179401 | -2.618280 |
| H  | -2.361797 | -0.631153 | -2.846046 |
| H  | -3.773871 | 0.169593  | -3.554748 |
| H  | -3.988536 | -0.953310 | -2.211796 |
| C  | 1.608512  | 2.518499  | 2.024319  |
| H  | 2.568078  | 2.440288  | 1.508298  |
| H  | 1.492742  | 3.568778  | 2.307769  |
| C  | 1.585492  | 1.639461  | 3.278361  |
| H  | 0.723647  | 1.873047  | 3.908971  |
| H  | 2.484758  | 1.810847  | 3.876284  |
| H  | 1.529324  | 0.578336  | 3.025145  |
| C  | 0.292637  | 3.640562  | -0.307158 |
| H  | 1.247887  | 3.622594  | -0.843822 |
| H  | -0.473417 | 3.471011  | -1.068474 |
| C  | 0.074498  | 4.997299  | 0.367971  |
| H  | -0.899348 | 5.054443  | 0.862113  |
| H  | 0.105938  | 5.792898  | -0.381835 |
| H  | 0.843430  | 5.222019  | 1.111077  |
| O  | -0.089845 | -0.904949 | -1.961188 |
| O  | 0.214840  | 1.552556  | -3.834480 |
| P  | 2.458350  | -0.750224 | -0.357861 |
| P  | -0.179412 | -2.363955 | 0.586067  |
| P  | -2.386905 | 0.544177  | -0.025971 |
| P  | 0.286442  | 2.131760  | 0.781753  |
| Mo | 0.012689  | -0.161664 | -0.389586 |
| H  | 0.333951  | -0.247575 | 1.396311  |

Sum of electronic and zero-point Energies= -2413.383185  
Sum of electronic and thermal Free Energies= -2413.456044

78  
**TS7**

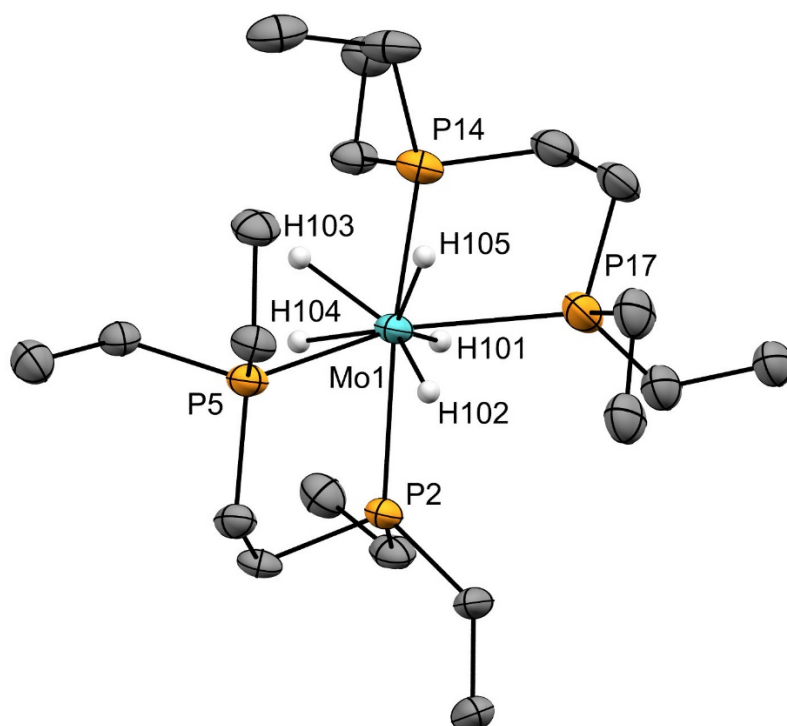
|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -2.229495 | -2.675864 | 0.419132  |
| H | -2.098290 | -2.626698 | 1.504768  |
| H | -3.185188 | -3.171822 | 0.229347  |
| C | -1.070390 | -3.419737 | -0.241654 |
| H | -1.216199 | -3.482612 | -1.325338 |
| H | -0.983783 | -4.442782 | 0.139114  |
| C | 0.986906  | 3.330307  | 0.144179  |

|    |           |           |           |
|----|-----------|-----------|-----------|
| H  | 1.288645  | 3.622861  | -0.865719 |
| H  | 0.761772  | 4.248708  | 0.694151  |
| C  | 2.098181  | 2.532341  | 0.825957  |
| H  | 1.845920  | 2.313309  | 1.867869  |
| H  | 3.043990  | 3.081319  | 0.819995  |
| C  | -3.590327 | -0.223237 | 0.951655  |
| H  | -4.302599 | -1.033433 | 1.142304  |
| H  | -3.090793 | -0.027689 | 1.904553  |
| C  | -4.339435 | 1.006588  | 0.434005  |
| H  | -3.668165 | 1.835491  | 0.205335  |
| H  | -5.046663 | 1.358494  | 1.189935  |
| H  | -4.909295 | 0.781698  | -0.470723 |
| C  | -3.113124 | -1.018826 | -1.841429 |
| H  | -3.263236 | 0.019141  | -2.157279 |
| H  | -2.359322 | -1.421631 | -2.523632 |
| C  | -4.418921 | -1.812949 | -1.916206 |
| H  | -4.258744 | -2.875082 | -1.712502 |
| H  | -4.845917 | -1.735975 | -2.920072 |
| H  | -5.169178 | -1.445955 | -1.210305 |
| C  | 1.718911  | -3.445693 | -1.001882 |
| H  | 2.718023  | -3.291194 | -0.585414 |
| H  | 1.489326  | -4.510556 | -0.875791 |
| C  | 1.683095  | -3.047353 | -2.478665 |
| H  | 0.693111  | -3.204133 | -2.915955 |
| H  | 2.398137  | -3.646750 | -3.048386 |
| H  | 1.928617  | -1.991473 | -2.612416 |
| C  | 0.905799  | -3.098311 | 1.774734  |
| H  | 0.037195  | -2.844013 | 2.387007  |
| H  | 0.950286  | -4.191994 | 1.710570  |
| C  | 2.175370  | -2.546132 | 2.419052  |
| H  | 3.064787  | -2.752819 | 1.818661  |
| H  | 2.325180  | -3.009409 | 3.397947  |
| H  | 2.099648  | -1.469407 | 2.572786  |
| C  | 2.941762  | 1.318403  | -1.712903 |
| H  | 3.178351  | 0.364736  | -2.194608 |
| H  | 2.092116  | 1.722970  | -2.269081 |
| C  | 4.137341  | 2.272344  | -1.759856 |
| H  | 3.890974  | 3.253929  | -1.345575 |
| H  | 4.451090  | 2.424264  | -2.796376 |
| H  | 4.997811  | 1.882304  | -1.209855 |
| C  | 3.802173  | 0.221447  | 0.860894  |
| H  | 4.482301  | 1.063851  | 1.025746  |
| H  | 3.468023  | -0.107596 | 1.846738  |
| C  | 4.520643  | -0.901288 | 0.111675  |
| H  | 3.832420  | -1.699489 | -0.171656 |
| H  | 5.301529  | -1.338955 | 0.738882  |
| H  | 4.996114  | -0.535179 | -0.801777 |
| C  | -1.515147 | 3.103797  | -1.341734 |
| H  | -2.518649 | 2.669749  | -1.327180 |
| H  | -1.616865 | 4.162613  | -1.082012 |
| C  | -0.893565 | 2.942755  | -2.732604 |
| H  | 0.096156  | 3.406159  | -2.788752 |
| H  | -1.523211 | 3.435655  | -3.478129 |
| H  | -0.791882 | 1.890465  | -3.006499 |
| C  | -1.402329 | 2.680263  | 1.606642  |
| H  | -2.135911 | 1.897360  | 1.796633  |
| H  | -0.629391 | 2.561455  | 2.371285  |
| C  | -2.048279 | 4.065251  | 1.691372  |
| H  | -1.338307 | 4.867244  | 1.468543  |
| H  | -2.424635 | 4.235486  | 2.703653  |
| H  | -2.892724 | 4.162524  | 1.004142  |
| P  | -2.299640 | -0.921233 | -0.178371 |
| P  | 0.524608  | -2.511304 | 0.061400  |
| P  | 2.316038  | 0.891251  | -0.013577 |
| P  | -0.548103 | 2.285321  | 0.009569  |
| Mo | 0.049558  | -0.101768 | -0.483366 |
| H  | -0.232965 | -0.199631 | 1.394990  |
| C  | -0.209470 | -0.091330 | 3.101417  |
| O  | 0.717287  | 0.624779  | 3.308371  |
| O  | -1.137443 | -0.782071 | 3.372022  |
| O  | -0.047692 | -0.248762 | -2.181211 |

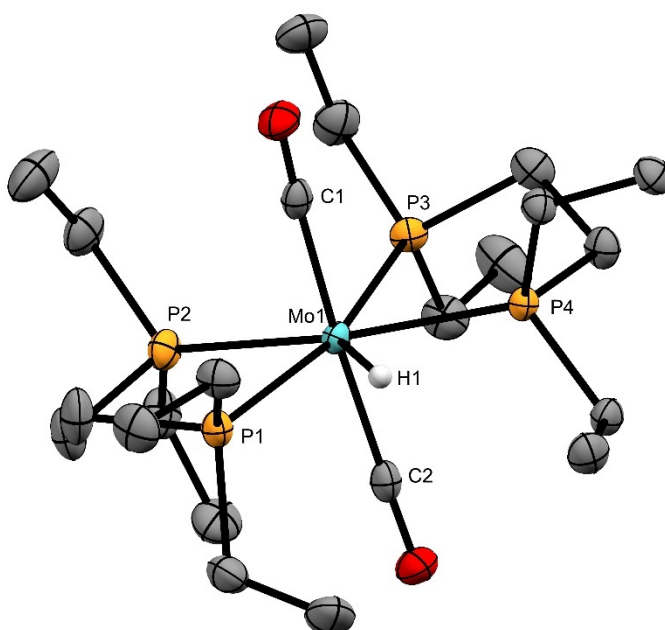
Sum of electronic and zero-point Energies= -2488.654229  
Sum of electronic and thermal Free Energies= -2488.727678

#### IV. Crystallographic data

##### IV.1. $[\text{Mo}(\text{depe})_2\text{H}_5][\text{B}(\text{C}_6\text{H}_5)_4]$ (1.BPh<sub>4</sub>) – Molecular structure



##### IV.2. $[\text{Mo}(\text{depe})_2(\text{CO})_2\text{H}][\text{B}(\text{C}_6\text{H}_5)_4]$ (4.BPh<sub>4</sub>) – Molecular structure





### IV.3. [Mo(depe)<sub>2</sub>H<sub>5</sub>][B(C<sub>6</sub>H<sub>5</sub>)<sub>4</sub>] (1.BPh<sub>4</sub>) – Tables

| Crystal Data                                                                            |                                                                                                                                                                                    |
|-----------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Chemical Formula                                                                        | C <sub>20</sub> H <sub>53</sub> MoP <sub>4</sub> ·C <sub>24</sub> H <sub>20</sub> B                                                                                                |
| M <sub>r</sub>                                                                          | 828.79                                                                                                                                                                             |
| Crystal system, space group                                                             | monoclinic, <i>P</i> 2 <sub>1</sub> / <i>c</i>                                                                                                                                     |
| Temperature (K)                                                                         | 100                                                                                                                                                                                |
| a, b, c (Å)                                                                             | 13.9330 (3), 16.7804 (4), 19.2576 (5)                                                                                                                                              |
| α, β, γ (°)                                                                             | 90, 95.774 (2), 90                                                                                                                                                                 |
| V (Å <sup>3</sup> )                                                                     | 4479.61 (19)                                                                                                                                                                       |
| Z                                                                                       | 4                                                                                                                                                                                  |
| Radiation type                                                                          | Cu Kα radiation, λ = 1.54184 Å                                                                                                                                                     |
| μ (mm <sup>-1</sup> )                                                                   | 3.96                                                                                                                                                                               |
| Crystal size (mm)                                                                       | 0.10 × 0.07 × 0.02                                                                                                                                                                 |
| Data Collection                                                                         |                                                                                                                                                                                    |
| Diffractometer                                                                          | XtaLAB Synergy, Dualflex, HyPix                                                                                                                                                    |
| Absorption Correction                                                                   | multi-scan CrysAlisPro 1.171.41.120a (Rigaku Oxford Diffraction, 2021) Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm. |
| T <sub>min</sub> , T <sub>max</sub>                                                     | 0.774, 1.000                                                                                                                                                                       |
| No. of measured, independent and observed [ <i>I</i> > 2σ( <i>I</i> )] reflections      | 30591, 7622, 5143                                                                                                                                                                  |
| R <sub>int</sub>                                                                        | 0.150                                                                                                                                                                              |
| θ <sub>max</sub> (°)                                                                    | 65.1                                                                                                                                                                               |
| Refinement                                                                              |                                                                                                                                                                                    |
| R[ <i>F</i> <sup>2</sup> > 2σ( <i>F</i> <sup>2</sup> )], wR( <i>F</i> <sup>2</sup> ), S | <b>0.078</b> , 0.280, 1.16                                                                                                                                                         |
| No. of reflections                                                                      | 7622                                                                                                                                                                               |
| No. of parameters                                                                       | 479                                                                                                                                                                                |
| H-atom treatment                                                                        | H atoms treated by a mixture of independent and constrained refinement                                                                                                             |
| Δρ <sub>max</sub> , Δρ <sub>min</sub> (e Å <sup>-3</sup> )                              | 1.32, -1.94                                                                                                                                                                        |

#### IV.4. [Mo(depe)<sub>2</sub>(HCOO)H<sub>2</sub>][B(C<sub>6</sub>H<sub>5</sub>)<sub>4</sub>] (3.BPh<sub>4</sub>) – Tables

| Crystal Data                                                      |                                                                                                                                          |
|-------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------|
| Chemical Formula                                                  | 2(C <sub>21</sub> H <sub>49</sub> MoO <sub>2</sub> P <sub>4</sub> )·2(C <sub>24</sub> H <sub>20</sub> B)·C <sub>4</sub> H <sub>8</sub> O |
| M <sub>r</sub>                                                    | 1817.37                                                                                                                                  |
| Crystal system, space group                                       | triclinic, P -1                                                                                                                          |
| Temperature (K)                                                   | 100                                                                                                                                      |
| a, b, c (Å)                                                       | 15.332 (3), 15.378 (3), 23.433 (5)                                                                                                       |
| α, β, γ (°)                                                       | 101.532 (6), 99.455 (5), 108.140 (5)                                                                                                     |
| V (Å <sup>3</sup> )                                               | 4989.8 (16)                                                                                                                              |
| Z                                                                 | 2                                                                                                                                        |
| Radiation type                                                    | Mo Kα radiation, λ = 0.71073 Å                                                                                                           |
| μ (mm <sup>-1</sup> )                                             | 0.43                                                                                                                                     |
| Crystal size (mm)                                                 | 0.10 × 0.08 × 0.02                                                                                                                       |
| Data Collection                                                   |                                                                                                                                          |
| Diffractometer                                                    | Bruker Kappa Apex2 diffractometer                                                                                                        |
| Absorption Correction                                             | multi-scan SADABS (Siemens, 1996)                                                                                                        |
| T <sub>min</sub> , T <sub>max</sub>                               | 0.676, 0.740                                                                                                                             |
| No. of measured, independent and observed [I > 2σ(I)] reflections | 105966, 16987, 11913                                                                                                                     |
| R <sub>int</sub>                                                  | 0.124                                                                                                                                    |
| θ <sub>max</sub> (°)                                              | 24.7                                                                                                                                     |
| Refinement                                                        |                                                                                                                                          |
| R[F <sup>2</sup> > 2σ(F <sup>2</sup> )], wR(F <sup>2</sup> ), S   | 0.099, 0.225, 1.19                                                                                                                       |
| No. of reflections                                                | 16987                                                                                                                                    |
| No. of parameters                                                 | 1016                                                                                                                                     |
| H-atom treatment                                                  | H atoms parameters constrained                                                                                                           |
| Δρ <sub>max</sub> , Δρ <sub>min</sub> (e Å <sup>-3</sup> )        | 1.21, -1.73                                                                                                                              |

#### IV.5. [Mo(depe)<sub>2</sub>(CO)<sub>2</sub>H][B(C<sub>6</sub>H<sub>5</sub>)<sub>4</sub>] (4.BPh<sub>4</sub>) – Tables

| Crystal Data                                                                            |                                                                                                                                                                                    |
|-----------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Chemical Formula                                                                        | C <sub>22</sub> H <sub>49</sub> MoO <sub>2</sub> P <sub>4</sub> ·C <sub>24</sub> H <sub>20</sub> B                                                                                 |
| M <sub>r</sub>                                                                          | 884.64                                                                                                                                                                             |
| Crystal system, space group                                                             | monoclinic, <i>P</i> 2 <sub>1</sub> / <i>c</i>                                                                                                                                     |
| Temperature (K)                                                                         | 100                                                                                                                                                                                |
| a, b, c (Å)                                                                             | 21.5133 (2), 18.8557 (2), 22.6970 (2)                                                                                                                                              |
| α, β, γ (°)                                                                             | 90, 99.354 (1), 90                                                                                                                                                                 |
| V (Å <sup>3</sup> )                                                                     | 9084.57 (15)                                                                                                                                                                       |
| Z                                                                                       | 8                                                                                                                                                                                  |
| Radiation type                                                                          | Cu Kα radiation, λ = 1.54184 Å                                                                                                                                                     |
| μ (mm <sup>-1</sup> )                                                                   | 3.96                                                                                                                                                                               |
| Crystal size (mm)                                                                       | 0.20 × 0.15 × 0.04                                                                                                                                                                 |
| Data Collection                                                                         |                                                                                                                                                                                    |
| Diffractometer                                                                          | XtaLAB Synergy, Dualflex, HyPix                                                                                                                                                    |
| Absorption Correction                                                                   | multi-scan CrysAlisPro 1.171.41.120a (Rigaku Oxford Diffraction, 2021) Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm. |
| T <sub>min</sub> , T <sub>max</sub>                                                     | 0.432, 1.000                                                                                                                                                                       |
| No. of measured, independent and observed [ <i>I</i> > 2σ( <i>I</i> )] reflections      | 162235, 17141, 15649                                                                                                                                                               |
| R <sub>int</sub>                                                                        | 0.051                                                                                                                                                                              |
| θ <sub>max</sub> (°)                                                                    | 70.1                                                                                                                                                                               |
| Refinement                                                                              |                                                                                                                                                                                    |
| R[ <i>F</i> <sup>2</sup> > 2σ( <i>F</i> <sup>2</sup> )], wR( <i>F</i> <sup>2</sup> ), S | 0.030, 0.075, 1.02                                                                                                                                                                 |
| No. of reflections                                                                      | 17141                                                                                                                                                                              |
| No. of parameters                                                                       | 1000                                                                                                                                                                               |
| H-atom treatment                                                                        | H atoms treated by a mixture of independent and constrained refinement                                                                                                             |
| Δρ <sub>max</sub> , Δρ <sub>min</sub> (e Å <sup>-3</sup> )                              | 0.93, -0.79                                                                                                                                                                        |

#### IV.6. [Mo(depe)<sub>2</sub>(<sup>i</sup>PrNCHN<sup>i</sup>Pr)H<sub>2</sub>] (7.BPh<sub>4</sub>) – Tables

| Crystal Data                                                      |                                                                                                                                                                                           |
|-------------------------------------------------------------------|-------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Chemical Formula                                                  | C <sub>27</sub> H <sub>63</sub> MoN <sub>2</sub> P <sub>4</sub> ·C <sub>24</sub> H <sub>20</sub> B                                                                                        |
| M <sub>r</sub>                                                    | 954.82                                                                                                                                                                                    |
| Crystal system, space group                                       | monoclinic, <i>P</i> 2 <sub>1</sub>                                                                                                                                                       |
| Temperature (K)                                                   | 100                                                                                                                                                                                       |
| a, b, c (Å)                                                       | 11.2886 (1), 25.1547 (2), 20.9232 (2)                                                                                                                                                     |
| α, β, γ (°)                                                       | 90, 90.436 (1), 90                                                                                                                                                                        |
| V (Å <sup>3</sup> )                                               | 5941.21 (9)                                                                                                                                                                               |
| Z                                                                 | 4                                                                                                                                                                                         |
| Radiation type                                                    | Cu Kα radiation, λ = 1.54184 Å                                                                                                                                                            |
| μ (mm <sup>-1</sup> )                                             | 3.04                                                                                                                                                                                      |
| Crystal size (mm)                                                 | 0.15 × 0.09 × 0.02                                                                                                                                                                        |
| Data Collection                                                   |                                                                                                                                                                                           |
| Diffractometer                                                    | XtaLAB Synergy, Dualflex, HyPix                                                                                                                                                           |
| Absorption Correction                                             | multi-scan <i>CrysAlis PRO</i> 1.171.41.93a (Rigaku Oxford Diffraction, 2020) Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm. |
| T <sub>min</sub> , T <sub>max</sub>                               | 0.801, 1.000                                                                                                                                                                              |
| No. of measured, independent and observed [I > 2σ(I)] reflections | 112023, 19997, 19255                                                                                                                                                                      |
| R <sub>int</sub>                                                  | 0.055                                                                                                                                                                                     |
| θ <sub>max</sub> (°)                                              | 65.1                                                                                                                                                                                      |
| Refinement                                                        |                                                                                                                                                                                           |
| R[F <sup>2</sup> > 2σ(F <sup>2</sup> )], wR(F <sup>2</sup> ), S   | 0.055, 0.136, 1.05                                                                                                                                                                        |
| No. of reflections                                                | 19997                                                                                                                                                                                     |
| No. of parameters                                                 | 1035                                                                                                                                                                                      |
| H-atom treatment                                                  | H atoms treated by a mixture of independent and constrained refinement                                                                                                                    |
| Δρ <sub>max</sub> , Δρ <sub>min</sub> (e Å <sup>-3</sup> )        | 1.06, -0.82                                                                                                                                                                               |

## V. Supplementary references

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