

The Mechanism of Spin-Phonon Relaxation in Endohedral Metallofullerene Single Molecule Magnets

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Table S1: Summary of all the Endohedral metallofullerene based SMMs along with blocking temperature and the U_{eff} .

Sr. No.	Molecule	$T_{B,100s}$ (K)	$T_{B,hys}$ (K) _(Field sweep mTs-1)is given in bracket)	$T_{B,ZFC}$ (K) _{(temperature sweep (K min-1) used is given in bracket)}	U_{eff} (K)	Ref
1	DySc ₂ N@D ₃ (6140)-C ₆₈	2.3	5 (2.9)	3.8 (5)	23.6	1
2	DySc ₂ N@D _{5h} (6)-C ₈₀	3.6	7 (2.9)	5.9 (5)	17.7	1
3	DySc ₂ N@I _h (7)-C ₈₀	4.6	7 (2.9)	6.9 (5)	23.6	2
4	DyY ₂ N@I _h (7)-C ₈₀	~6	8 (2.9)	8.4 (5)	929	3
5	DyLu ₂ N@I _h (7)-C ₈₀	~6.5	9 (2.9)	9.5 (5)	24.2	3, 4
6	HoSc ₂ N@I _h (7)-C ₈₀	—	—	—	16.5	5
7	Dy ₂ ScN@I _h (7)-C ₈₀	5	7 (2.9)	8 (5)	10.7/1735	6
8	Dy ₂ YN@I _h (7)-C ₈₀	~3.5	5 (2.9)	4.7 (5)	43.8/680	3
9	Dy ₂ LaN@I _h (7)-C ₈₀	~2	4 (2.9)	3.3 (5)	—	3
10	Dy ₂ LuN@I _h (7)-C ₈₀	5.2	8 (2.9)	8 (5)	4.3	3, 4
11	Dy ₂ GdN@I _h (7)-C ₈₀	~1.5	~1.8 (5.3)	—	15.1	7
12	DyErScN@I _h (7)-C ₈₀	~4.5	9 (33)	~8 (3)	12.5	8
13	Tb ₂ ScN@I _h (7)-C ₈₀	0.4	~0.4 (3.3)	—	1/10.5/56.4	9
14	Dy ₂ ScN@D _{5h} (6)-C ₈₀	2.6	7 (2.9)	5.3 (5)	8.4	1
15	Dy ₂ ScN@D _s (51365)-C ₈₄	~1.8	5 (2.9)	3.3 (5)	—	1
16	Dy ₃ N@I _h (7)-C ₈₀	—	~2 (0.8)	—	—	10
17	TbCN@C _{2v} (19138)-C _{76b}	—	—	—	12	11
18	TbCN@C ₂ (5)-C _{82b}	—	—	—	10–20	12
19	TbCN@Cs(6)-C _{82b}	—	—	—	10–20	12
20	TbCN@C _{2v} (9)-C _{82b}	—	—	—	10–20	12
21	Dy ₂ O@C _s (10528)-C ₇₂	3.4	7 (2.9)	8 (5)	—	13
22	Dy ₂ O@C ₂ (13333)-C ₇₄	5.0	14 (2.9)	14 (5)	—	13
23	Dy ₂ O@C _{2v} (5)-C ₈₀	3.2	6 (2.9)	11 (5)	25.9	14
24	Dy ₂ O@C _s (6)-C ₈₂	2.8	6 (2.9)	10 (5)	10.8	15
25	Dy ₂ O@C _{3v} (8)-C ₈₂	5.9	7 (2.9)	9 (5)	7.8	15
26	Dy ₂ O@C _{2v} (9)-C ₈₂	3.7	7 (2.9)	8 (5)	18.6	15
27	Dy ₂ O@C ₁ (26)-C ₈₈	6	8 (2.9)	10.5 (5)	20.4	16
28	Dy ₂ O@C _s (32)-C ₈₈	4.6	8 (2.9)	8.5 (5)	—	16
29	Dy ₂ O@D ₂ (35)-C ₈₈	3.9	8 (2.9)	8.5 (5)	—	16
30	Dy ₂ S@C _s (10528)-C ₇₂	—	3.0 (8.33)	—	—	17
31	Dy ₂ S@C _s (6)-C ₈₂	—	3.0 (8.33)	—	17.8	17, 18

32	Dy ₂ S@C _{3v} (8)-C ₈₂	2	5 (8.33)	4.0 (5)	6	17, 18
33	DyScS@C _s (6)-C ₈₂	~4	9 (10)	7.3 (5)	15.2	19
34	DyScS@C _{3v} (8)-C ₈₂	~2	9 (10)	7.3 (5)	6.5	19
35	DyYTIC@I _h (7)-C ₈₀	~5	7 (2.9)	8 (5)	14.9	20
36	Dy ₂ TiC@I _h (7)-C ₈₀	1.7	3 (5)	—	9.5	21
37	Dy ₂ TiC ₂ @I _h (7)-C ₈₀	—	1.8 (5)	—	—	21
38	Dy ₂ TiC@D _{5h} (6)-C ₈₀	—	1.8 (5)	—	—	21
39	Dy ₂ C ₂ @C _s (6)-C ₈₂	—	3.0 (8.33)	—	17.4	17
40	Dy ₂ C ₂ @C _s (32)-C ₈₈	—	2.1 (2.9)	—	—	16
41	Dy ₂ C ₂ @D ₂ (35)-C ₈₈	—	2.1 (2.9)	—	—	16
42	Dy@C _s (6)-C ₈₁ N	45	39 (3.5)/60 (10)	69 (1)	—	22
43	Dy ₂ @C ₈₀ (CH ₂ Ph)	18	22 (2.9)	21.9 (5)	613	23
44	Dy ₂ @C ₇₉ N	12	24 (20)	21 (3)	669	24
45	Tb ₂ @C ₇₉ N	24	26 (2.9)	28 (5)	757	25
46	Tb ₂ @C ₈₀ (CH ₂ Ph)	25.2	27 (9.5)	28.9 (5)	799	26
47	Tb ₂ @C ₈₀ (CF ₃)	25	26 (2.9)	28.5 (5)	801	27
48	Ho ₂ @C ₈₀ (CH ₂ Ph)	—	—	—	334	26
49	Er ₂ @C ₈₀ (CH ₂ Ph)a	—	—	—	—	26
50	TbGd@C ₈₀ (CH ₂ Ph)	—	—	14.4 (5)	—	26
51	TbY@C ₈₀ (CH ₂ Ph)	—	5 (2.9)	5 (5)	—	26
52	Nd ₂ @C ₈₀ (CF ₃)a	—	—	—	—	28
53	Gd ₂ @C ₇₉ Na	—	—	—	6.5	29, 30
54	DyEr@C _{3v} (8)-C ₈₂	—	3 (33)	5 (3)	—	31

Table S2: |V(r)/G(r)| ratio of DySSc@C₈₂ and Fragment DyScS(C₈H₆)₂. (Here V(r) is the Virial Field function, G(r) is the electronic kinetic energy density, |V(r)/G(r)| is the ratio of Virial field function to Electronic kinetic energy.

Bond	V(r)	G(r)	(∇ ² ρ(r))	H(r)	V(r)/G(r)
DySSc@C ₈₂					
Dy-S	−0.071	0.056	0.159	−0.015	1.285
Dy-C	−0.074	0.062	0.202	−0.008	1.189
Dy-S	−0.081	0.043	0.137	−0.011	1.143
Dy-C	−0.076	0.062	0.194	−0.014	1.219
Note: V(r)/G(r) < 1 – ionic interaction, V(r)/G(r) > 2 – covalent interaction, 1 < V(r)/G(r) < 2 – intermediate interaction.					

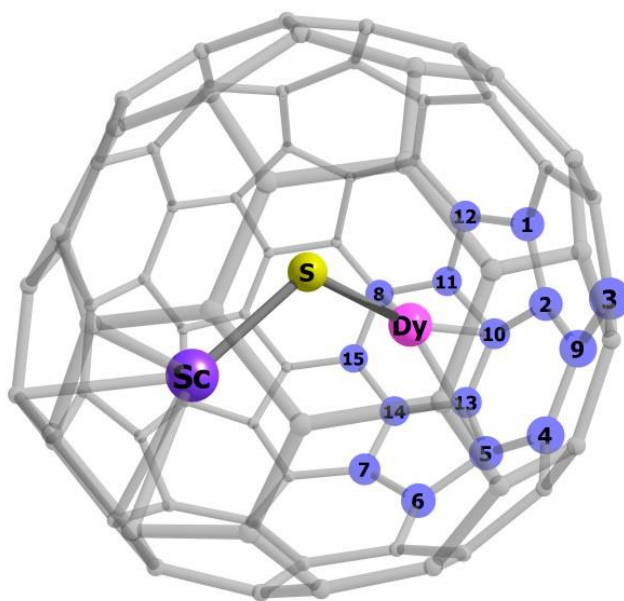


Figure S1: DyScS@C₈₂ structure showing the marked C atoms as the one having considerable delocalization indices.

Table S3: Delocalisation index $\delta(\text{Dy}, \text{C}_{\text{cage}})$ between Dy ^{III} and the cage carbon atoms.	
Carbon number	$\delta(\text{Dy}, \text{C}_{\text{cage}})$
1	0.236
2	0.315
3	0.101
4	0.208
5	0.276
6	0.163
7	0.163
8	0.237
9	0.237
10	0.345
11	0.315
12	0.236
13	0.328
14	0.276
15	0.208

Table S4: Energies, g-tensor, angle of the ground excited g_{zz} with the ground g_{zz} in $\text{DyScS}@C_{82}$ molecule and the $\text{DyScS}(\text{C}_8\text{H}_6)_2$.

KD	Energies (cm^{-1})	g_{xx}	g_{yy}	g_{zz}	Angle ($^\circ$)
<u>$\text{DyScS}@C_{82}$</u>					
1	0.0	0.002	0.003	19.925	
2	239.3	0.009	0.012	17.078	2.5
3	471.0	0.190	0.235	14.133	5.9
4	659.7	0.249	0.567	11.697	20.1
5	800.4	0.641	1.288	8.531	4.4
6	889.7	6.186	5.695	3.801	45.7
7	941.6	0.996	3.247	14.668	96.1
8	1011.6	0.182	1.136	18.890	89.8
<u>$\text{DyScS}(\text{C}_8\text{H}_6)_2$</u>					
1	0.0	0.001	0.001	20.032	
2	271.9	0.012	0.018	17.726	27.8
3	389.9	0.058	0.079	14.218	21.5
4	520.3	0.875	1.029	12.453	16.6
5	596.8	1.725	3.809	10.140	36.5
6	671.5	3.383	5.859	9.881	86.8
7	764.8	0.909	1.938	16.382	87.9
8	988.9	0.043	0.075	20.059	93.4

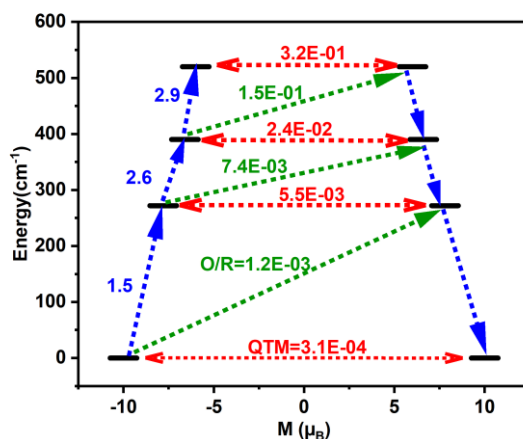


Figure S2: Relaxation mechanism of spin reversal in fragment $\text{DyScS}(\text{C}_8\text{H}_6)_2$

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ELEMENT S

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ELEMENT Sc

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References

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