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Supplementary Information

Carbon Corrosion in Low-Temperature CO₂ Electrolysis Systems

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Supplementary Figures

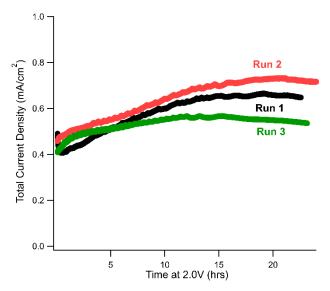


Figure S1: Current density measured during replicate carbon corrosion experiments in phosphate electrolyte solution (5mM, pH = 8). Corrosion voltage held at 2.0V (vs Ag/AgCl) for 24hrs.

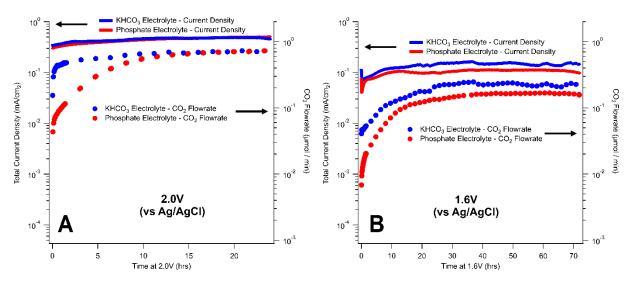


Figure S2: Comparison of carbon corrosion rates in phosphate (red) and bicarbonate (blue) electrolyte solutions. Current density (lines, left axis) and CO₂ evolved (dots, right axis) are plotted against time for experiments held at 2.0V vs Ag/AgCl for 24 hrs (A) or 1.6V vs Ag/AgCl for 72hrs (B).

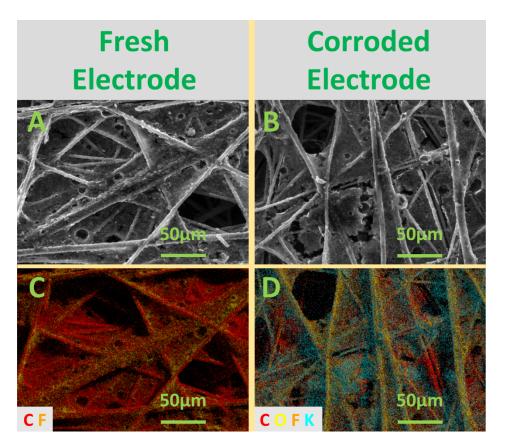


Figure S3: SEM images and EDS mapping of a fresh (A, C) and corroded (B, D) AvCarb GDL on the macroporous carbon fiber side. Corrosion conditions were 48hrs at 2.0V vs Ag/AgCl in bicarbonate electrolyte solution. SEM images of the macroporous fiber layer (A, B) are displayed along with the corresponding EDS map (C, D) for those images.

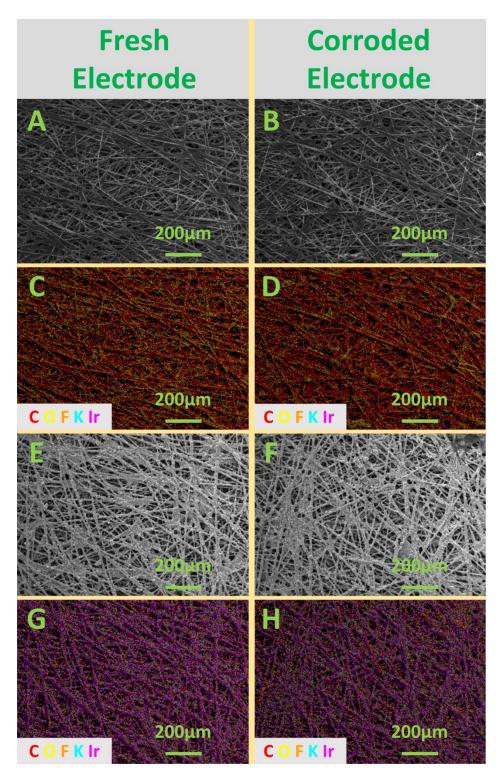


Figure S4: SEM images (A, B, E, F) and EDS mapping (C, D, G, H) of a fresh (left) and corroded (right) IrO₂-coated GDL. Corrosion conditions were 72hrs at 2.0V vs Ag/AgCl in phosphate electrolyte solution. Both the diffusion layer (A, B, C, D) and catalyst layer (E, F, G, H) were imaged and mapped before and after corrosion.

Supplementary Tables

Table S1 Reported properties of GDLs from various manufacturers.

| Anode Material | Reported Thickness (μm) | Area Density (g/cm²) | CFL PTFE Loading (wt. %) | MPL PTFE Loading (wt. %) | Through-Plane Resistivity (mΩ·cm²) |
|-----------------------------------|----------------------------|-------------------------|-----------------------------|-----------------------------|---------------------------------------|
| AvCarb GDS 5130 | 283 | 70 | ? | ? | 11 |
| Sigracet 39BB | 315 | 95 | 5% | 20 – 25% | <13 |
| Toray Paper (TGP-H-120 w/ MPL) | 430 | ~127 | 8-9% | 33 – 35% | ? |
| TiO2 Fiber Felt | 200-300 | - | - | - | - |

Table S2: ICP-OES analysis of fresh and spent electrodes from select corrosion experiments involving IrO_2 -coated and catalyst-free GDLs in phosphate electrolyte.

Electrode ICP Results

| | | | Concentrations (ppm) | | | |
|-------------------------|---------------------------|------|----------------------|----------|--------|--|
| Corrosion Conditions | IrO₂ Coated Electrode? | Cu | lr | K | Р | |
| Fresh Electrode | No | <250 | - | <250 | <250 | |
| 2.0V, 24hrs | No | 109 | 230.62 | 65612.16 | 603.47 | |
| 2.0V ,72hrs | Yes | <20 | 36834.24 | 19896.34 | 68.82 | |
| 1.6V, 168hrs | Yes | 37.4 | 42449.43 | 20194.23 | 72.47 | |

Table S3: ICP-OES analysis of fresh and spent phosphate electrolyte solutions from select corrosion experiments involving IrO_2 -coated and catalyst-free GDLs.

Electrolyte ICP Results

| | | Concentrations (ppm) | | | |
|-------------------------|---|----------------------|-----|---------|--------|
| Corrosion Conditions | Electrolyte from Experiment with IrO ₂ Coated Electrode? | Cu | lr | K | Р |
| Fresh Electrolyte | - | <20 | <20 | 1145.76 | 185.88 |
| 2.0V, 24hrs | No | <20 | <20 | 1023.28 | 187.75 |
| 72hrs, 2.0V | Yes | <20 | <20 | 1169.75 | 184.55 |
| 1.6V, 168hrs | Yes | <20 | <20 | 1390.58 | 177.93 |