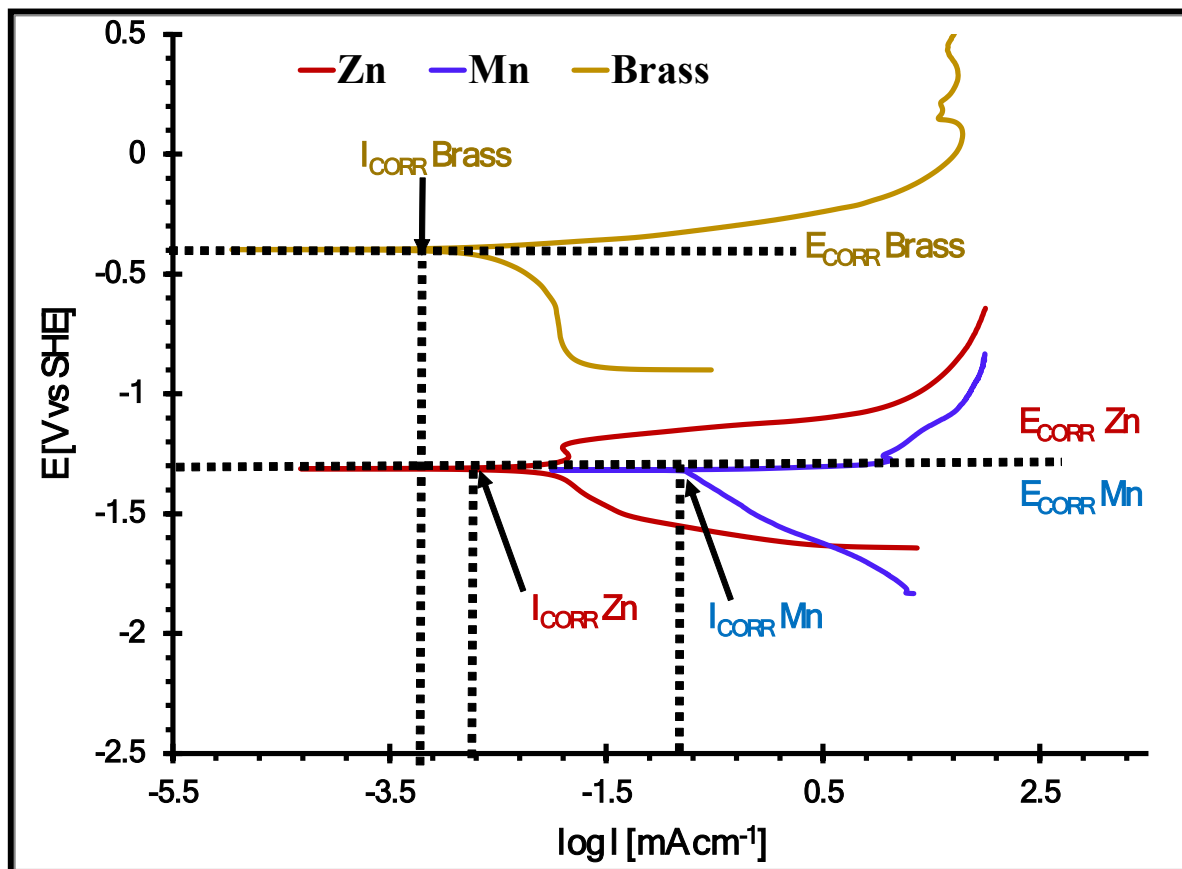


## SUPPLEMENTARY MATERIAL

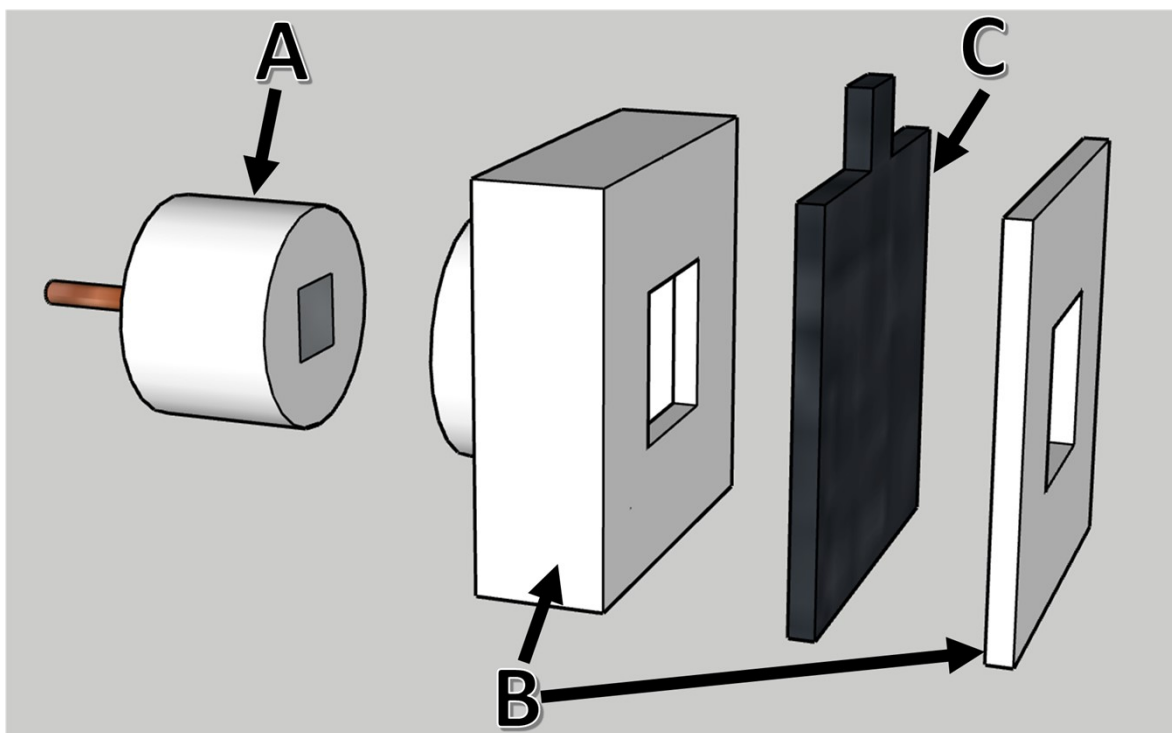
- Brass corrosion Test



**Fig. s1.** Comparison of Tafel diagrams of Zn, Mn and Brass

**Table s1:** ICORR and ECORR of Brass, Manganese and Zinc

| Metal     | $I_{CORR}$<br>[mA·cm <sup>-2</sup> ] | $E_{CORR}$<br>[V] |
|-----------|--------------------------------------|-------------------|
| Brass     | $6.45 \times 10^{-4}$                | 0.401             |
| Manganese | $340 \times 10^{-3}$                 | 1.328             |
| Zinc      | $1.54 \times 10^{-3}$                | 1.300             |



**Fig. s2.** Schematic representation of Metal-Air battery cell used for discharge tests: A) 1 cm<sup>2</sup> anode, B) acrylic cell body, and C) Air-cathode.

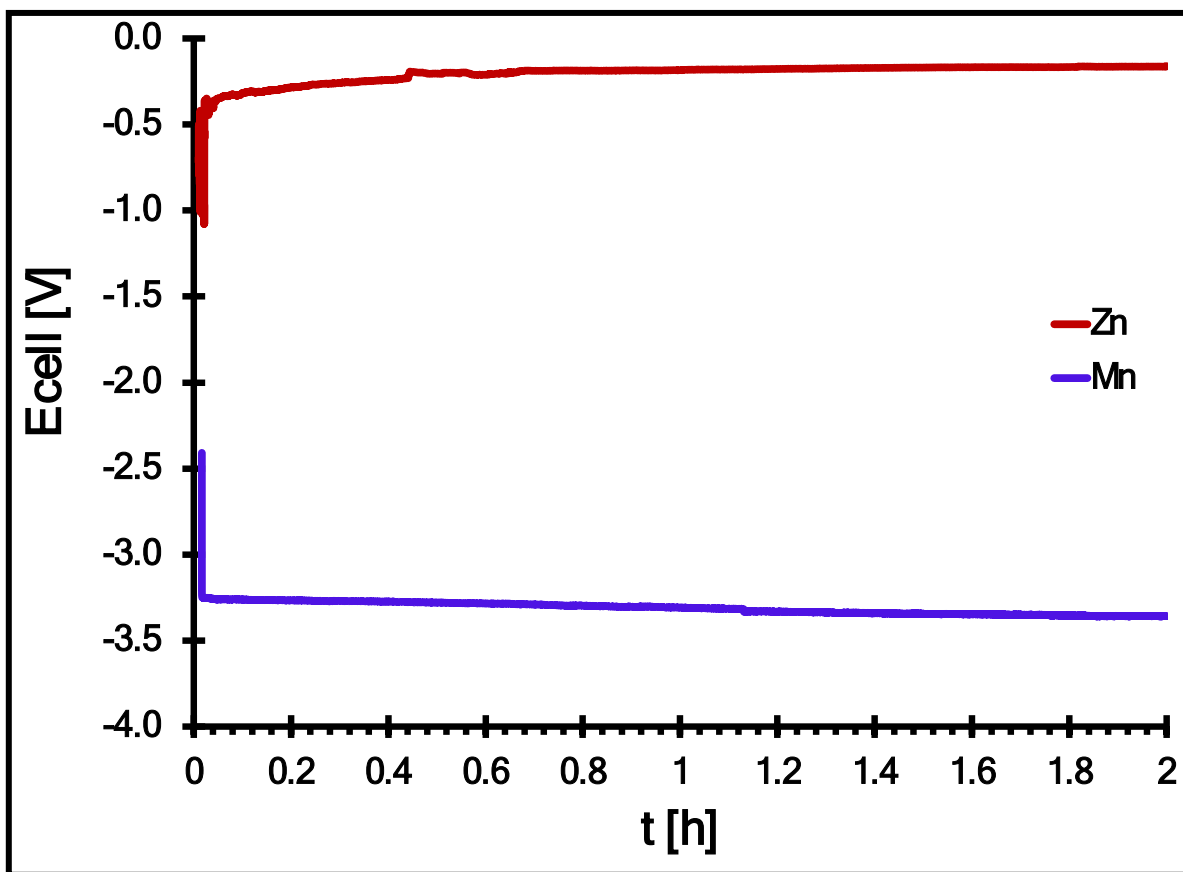
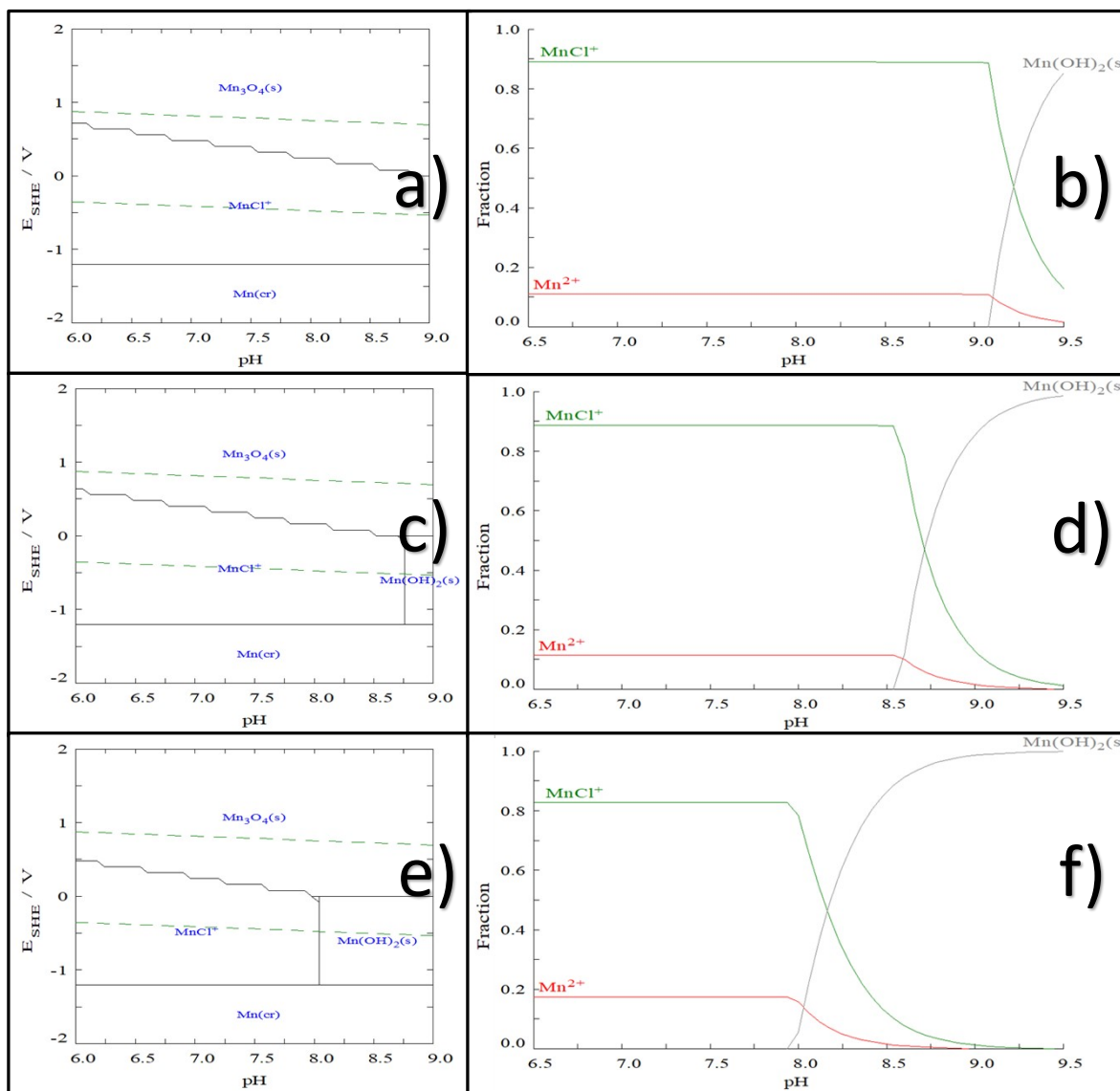
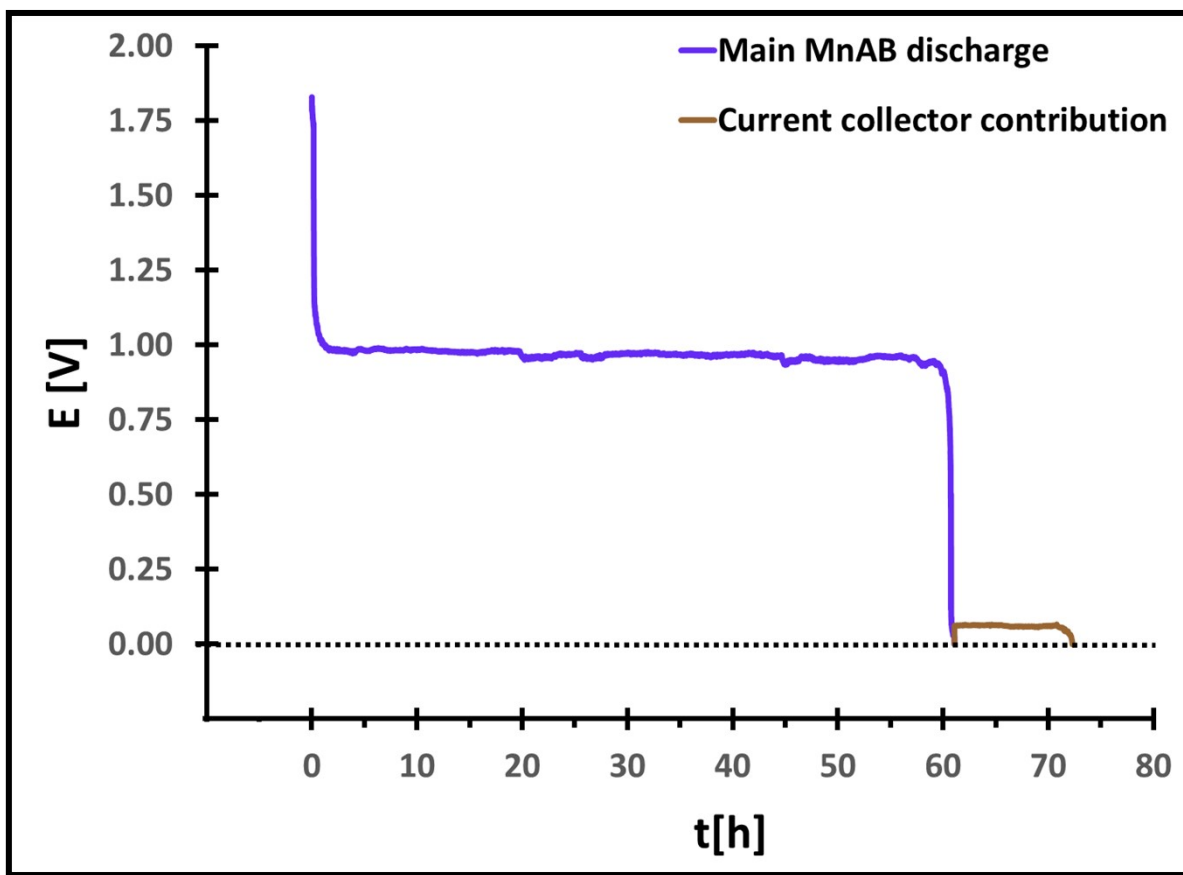


Fig. s3. Galvanostatic transient for the Zn galvanoplastic deposition and Mn galvanostatic electrowinning after 2 hours of electrolysis.

- Equilibrium diagrams of Mn in NaCl 2M



**Fig. s4.** Pourbaix diagrams for 0.01 (a), 0.1 (c), and 1 M (e) of Mn in 2M NaCl and molar fraction equilibrium diagram for 0.01 (a), 0.1 (c), and 1 M (e) of Mn in 2M NaCl.



**Fig. s5.** Raw data discharge curve of manganese-air battery showing the contribution of current collector after main Mn-air battery discharge ends.

|                                                                                                  | Theoretic Ecell [V] | Z (Number of electrons) | Theoretic Specific Capacity [A·h·kg <sup>-1</sup> ] | Theoretic Energy Density [Wh·kg <sup>-1</sup> ] |
|--------------------------------------------------------------------------------------------------|---------------------|-------------------------|-----------------------------------------------------|-------------------------------------------------|
| $3\text{Mn} + 4\text{H}_2\text{O} \rightarrow \text{Mn}_3\text{O}_4 + 8\text{H}^+ + 8\text{e}^-$ | 2.097               | $2\frac{2}{3}$          | 1299.66                                             | 3073.74                                         |
| $\text{Mn} \rightarrow \text{Mn}^{2+} + 2\text{e}^-$                                             | 2.365               | 2                       | 974.75                                              | 2044.01                                         |
| <b>Total energy balance:</b>                                                                     |                     |                         | <b>2274.41</b>                                      | <b>5117.75</b>                                  |

Table s2.- energetic data for the two synergic anodic reactions proposed.

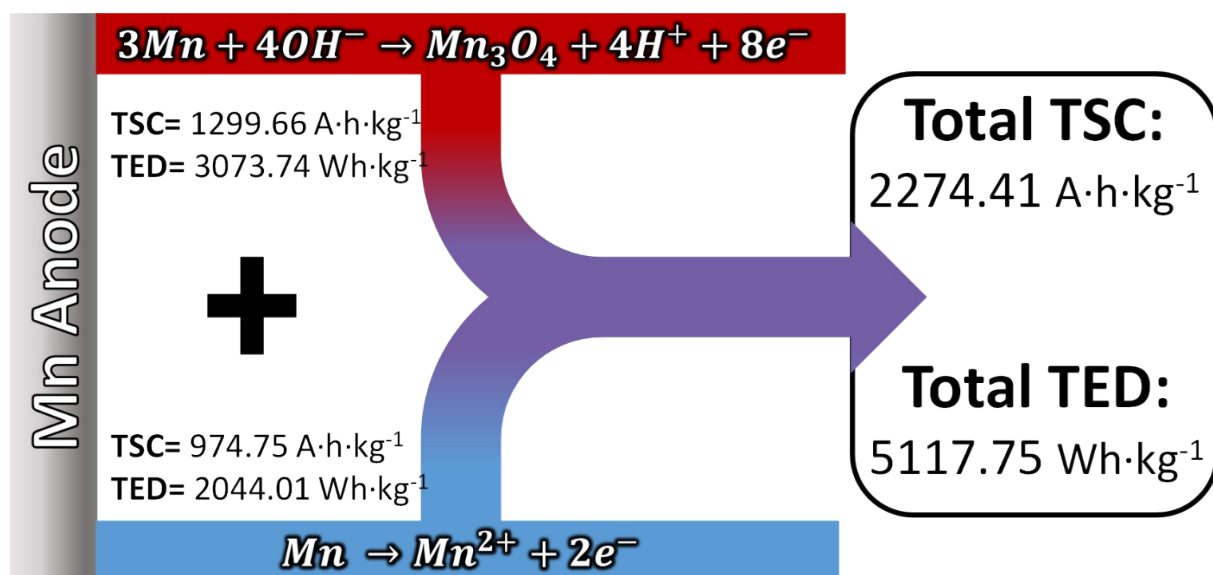


Fig. s6. Graphical representation of discharge mechanism and theoretical specific capacity (TSC) and energy density (TED) determination.

**Table s3:**Comparison of different MABs with MnAB and ZAB

| Ref. | Type             | Electrolyte      | Discharge rate<br>[mA·cm <sup>-2</sup> ] | OCV<br>[V] | Specific Capacity<br>[A·h·kg <sup>-1</sup> ] |
|------|------------------|------------------|------------------------------------------|------------|----------------------------------------------|
| -    | Primary Mn-Air   | NaCl Based       | 0.5                                      | 1.55       | 1930                                         |
| -    | Primary Zn-Air   | NaCl Based       | 0.5                                      | 1.01       | 748                                          |
| 16   | Primary Zn-Air   | Seawater based   | 25                                       | -          | 724                                          |
| 49   | Secondary Zn-Air | KOH based        | 10                                       | 1.405      | 801.2                                        |
| 50   | Secondary Zn-Air | KOH based        | 5                                        | 1.52       | 817                                          |
| 51   | Secondary Zn-Air | KOH based        | 10                                       | 1.45       | 788                                          |
| 17   | Primary Al-Air   | NaCl Based       | 10                                       | 0.85       | 1210                                         |
| 58   | Flexible Al-Air  | KOH based        | 2                                        | 2.20       | 2761                                         |
| 59   | Primary Al-Air   | NaOH Based       | 50                                       | 1.55       | 2777                                         |
| 60   | Primary Al-Air   | KOH based        | 5                                        | 1.60       | 220                                          |
| 61   | Primary Al-Air   | Gelled KOH based | 4.4                                      | 1.31       | 426                                          |