Dynamics of poly(methyl acrylate)/poly(methyl methacrylate)-grafted-Fe₃O₄

nanocomposites

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1. Examine the reproducibility of the measurements



Figure S1. Dielectric loss permittivity $\varepsilon''(\omega)$ for PMA40k/PMMA-g-Fe₃O₄ at 363K (red circles), 353K (olive diamonds), 343K (orange left triangles), 333K (wine hexagons): the lines represent the spectra at first cooling and the symbols represent the spectra at second cooling (symbols). For each comparison, at least two tests have been performed on different specimen.

2. Differential Scanning Calorimetry (DSC)



Figure S2. Temperature derivative of the specific heat capacity, dC_p/dT , of the PMA/PMMA-g-Fe₃O₄ nanocomposites. Only one T_g is observed for all nanocomposites.



Figure S3. (a) Shifted heat capacity, C_p , and (b) the temperature derivative of the specific heat capacity, dC_p/dT , of PMA40k/PMMA-g-Fe₃O₄ composite at cooling rates of 3K/min (blue squares), 5K/min (olive diamonds), 10K/min (pink upper triangles).

3. Dynamics of PMA/PMMA-g-Fe₃O₄ composites



Figure S4. Derivative spectra, ε'_{der} , of PMA40k/PMMA-g-Fe₃O₄ at different temperatures 393K (red circles), 383K (blue squares), 373K (olive diamonds), 363K (pink triangles), 353K (orange left triangles), 343K (wine inverted triangles).



Figure S5. Linear viscoelastic master curve of PMA40k/PMMA-g-Fe₃O₄, G' (open red circles) and G'' (filled red circles), at a reference temperature of 373K. The inset provides the dynamics shift factor, a_T , from the rheological measurements. No vertical shift is needed for the construction of the master cruve. No signs of α relaxation of PMMA can be detected from the rheological measurements



Figure S6. Shape parameters of the α^* process obtained after fittings of the dielectric spectra using the HN function at different temperatures.