Upcycling of low value end-of-life cathode material into next

generation cathode materials

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Slater

Figure S1: Rietveld refinement plot of end-of-life cathode.



Figure S2: Rietveld refinement plot of end-of-life cathode after 20 mins of leaching.



Figure S3: dQ/dV for recovered LMO when cycled at 10 mAg⁻¹ between 3-4.3 V



Figure S4: Galvanostatic charge-discharge profiles for LMO made from pristine starting materials. Cells were cycled at 10 mAg⁻¹ between 3-4.3 V.



Figure S5: XRD pattern of $LiMn_{1.5}Ni_{0.5}O_4$ made from LMO recovered from a Quality Control Rejected cell (Co K α). Tick marks correspond to LMNO (black).



Figure S6: dQ/dV for LMNO when cycled at 10 mAg⁻¹ between 3-4.3 V



Figure S7: Galvanostatic charge-discharge profile for LMNO made from pristine starting materials. Cells were cycled at 10 mAg⁻¹ between 3.5-4.9 V.



Figure S8: Average gravimetric discharge capacity against cycle number for upcycled LMNO made from recovered LMO. Cells were cycled at 10 mAg-1 between 3.5-4.9 V.



Figure S9: Pawley refinement plots of recovered a) $Li_4Mn_2O_5$ and b) $Li_2MnO_{2.25}F$.



Figure S10: dQ/dV for a) $Li_4Mn_2O_5$ and b) $Li_2MnO_{2.25}F$ when cycled at 10 mAg⁻¹ between 3-4.3 V



Figure S11: Galvanostatic charge-discharge profiles for a) $Li_4Mn_2O_5$ and b) $Li_2MnO_{2.25}F$ made from pristine starting materials. Cells were cycled at 10 mAg⁻¹ between 1.5-4.8 V.



Figure S12: Galvanostatic charge-discharge profile for $Li_{1.8}Na_{0.2}MnO_2F$ made from pristine starting materials. Cells were cycled at 10 mAg⁻¹ between 1.5-4.8 V.

	Lattice parameters				
	LiMn ₂ O ₄ phase	Layered oxide phase			
Leaching time/mins	a/Å	a/Å	c/Å		
0	8.208(4)	2.859(5)	14.263(1)		
5	8.177(1)	2.857(6)	14.267(2)		
10	8.201(2)	2.856(7)	14.273(4)		
15	8.204(2)	2.859(4)	14.286(2)		
20	/	2.857(3)	14.272(2)		

Table S1: Pawley refinement results for the lattice parameters of $LiMn_2O_4$ and layered oxide phase at different leaching times

Table S2: ICP-OES results showing the concentration of each element in solution at different leaching times

	Concentration in solution/ppm					
Leaching time/mins	Li	Mn	Ni	Со	AI	
5	436.95	4759.12	61.06	97.61	82.02	
10	672.50	8804.06	90.41	174.25	128.30	
15	852.90	12018.49	156.85	240.55	195.16	
20	891.89	12571.27	287.72	273.18	220.88	
20	891.89	12571.27	287.72	273.18	220.88	