

Supplementary Information for

Insights into the First Multi-Transition-Metal Containing Ruddlesden Popper-Type Cathode for all-solid-state Fluoride Ion Batteries

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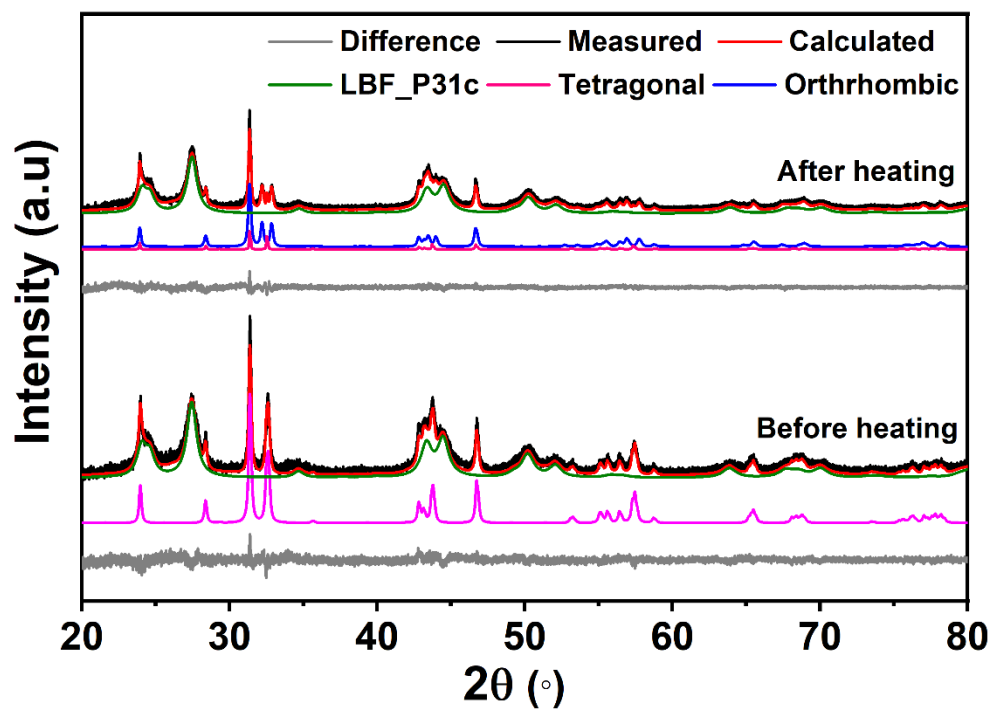


Figure S 1. XRD of cathode material before and after heating at $T= 170^\circ\text{C}$.

Table S 1. Details of the structures of the pristine, cathode material and after heating the cathode material at T= 170°C

		La₂Ni_{0.75}Co_{0.25}O_{4.08}	Cathode composite	Heated cathode composite
Crystal system and space group		Tetragonal I4/mmm	Tetragonal I4/mmm	Tetragonal I4/mmm
Lattice parameter	a	3.88125(4)	3.88081(19)	3.88088(19)
	c	12.57064(18)	12.5688(9)	12.5690(9)
Cell volume		189.365(4) (Å ³)	189.30(2) (Å ³)	189.30(2)

Table S 2. $\text{La}_2\text{Ni}_{0.75}\text{Co}_{0.25}\text{O}_{4.08}$ Weight and Atomic % from the EDAX analysis

Element	Weight %	Atomic %
La L	73.10	33.80
Ni K	11.10	12.10
Co K	04.40	04.80
O K	08.90	35.70

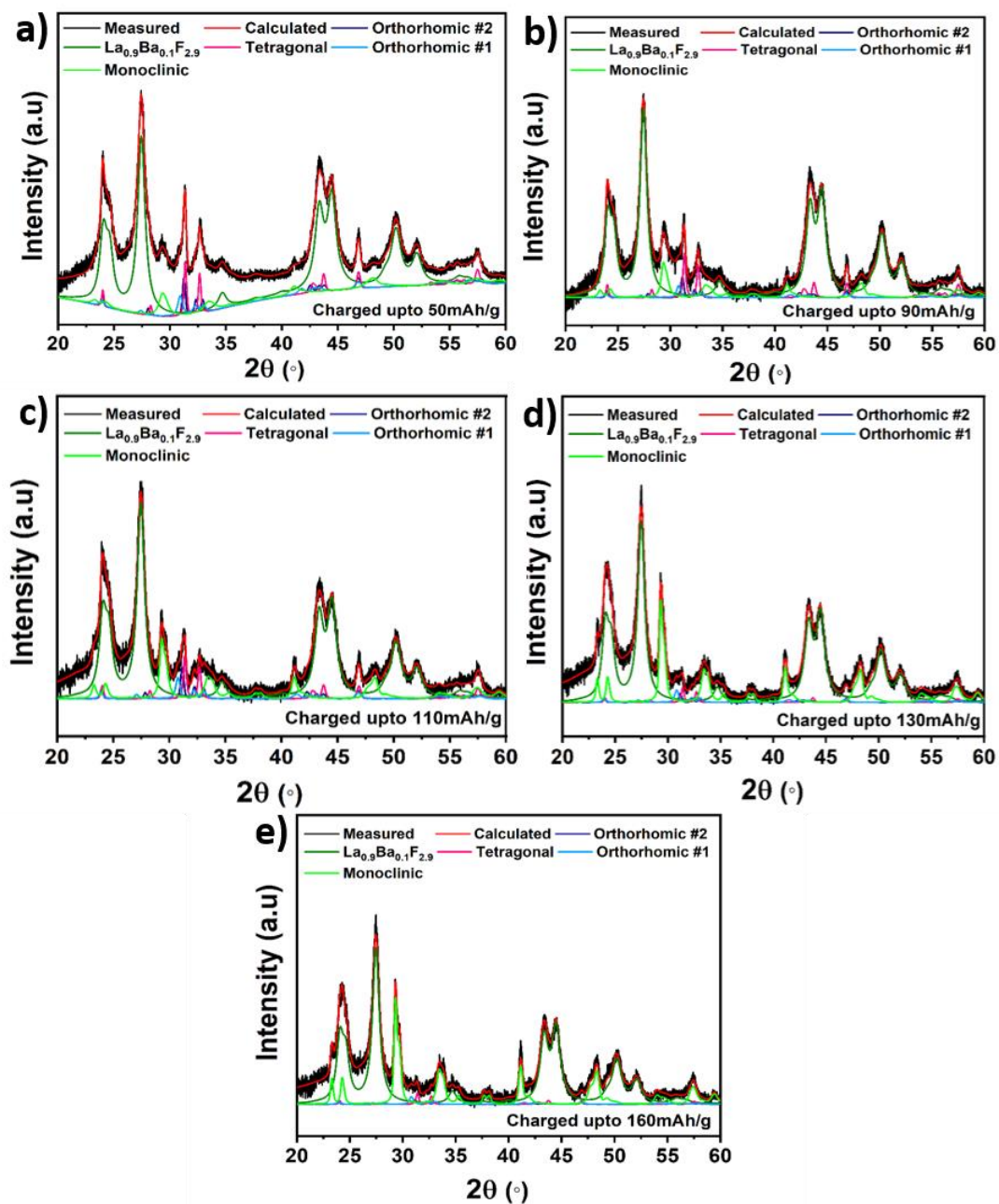


Figure S 2. **Structural changes of the RP-type phase at different cut off capacities.** Retiveld refinement for $\text{La}_2\text{Ni}_{0.75}\text{Co}_{0.25}\text{O}_{4.08} / \text{Pb-PbF}_2$ at $T=170$ dc, $I_{\text{charge}}=+8.0294\mu\text{A}$, $I_{\text{discharge}}=-4.0145\mu\text{A}$, at the charging capacity of (a) 50 mAh g^{-1} (b) 90 mAh g^{-1} (c) 110 mAh g^{-1} (d) 130 mAh g^{-1} (e) 160 mAh g^{-1}

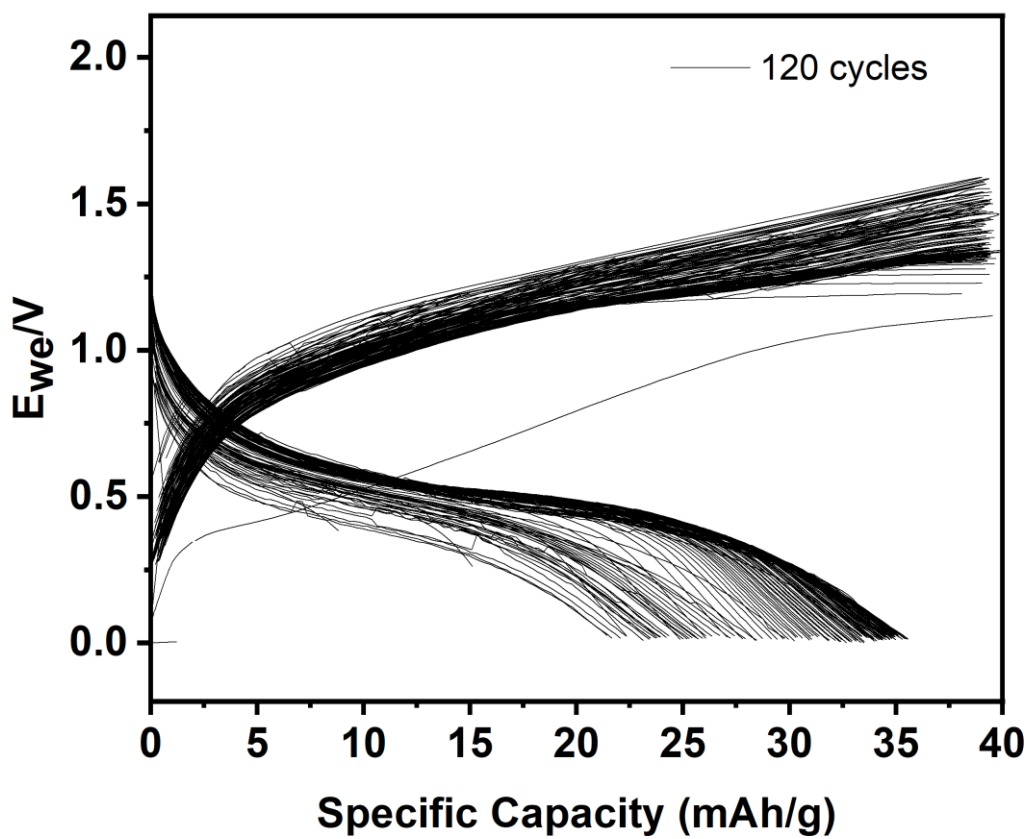


Figure S 3. Capacity-voltage curves for the cell operated for 120 cycles between 0 and 1.5 V

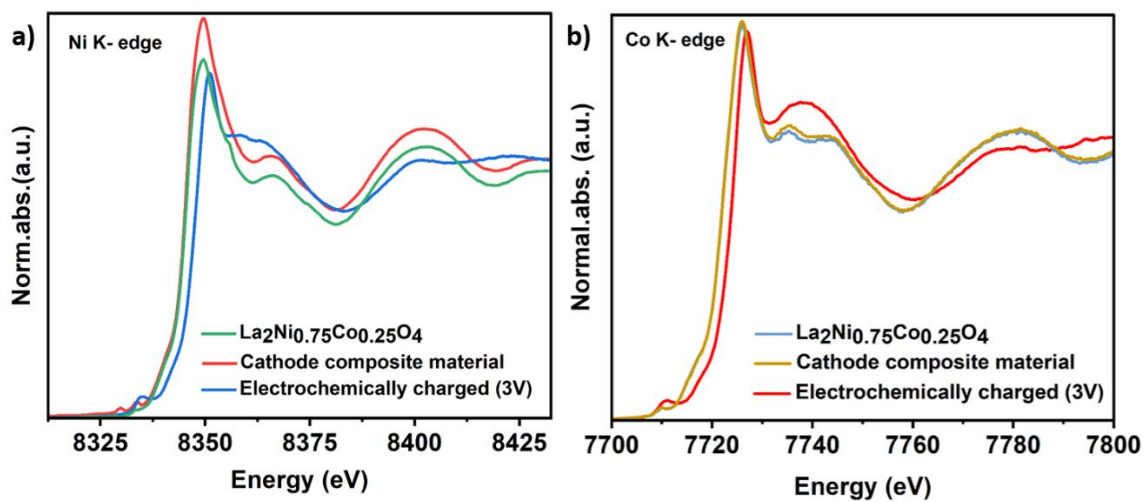


Figure S 4. The absorption edge energies of $\text{La}_2\text{Ni}_{0.75}\text{Co}_{0.25}\text{O}_{4.08}$ compared to cathode composite of $\text{La}_2\text{Ni}_{0.75}\text{Co}_{0.25}\text{O}_{4.08}$ and electrochemically charged upto 3V at Ni, Co K-edge.

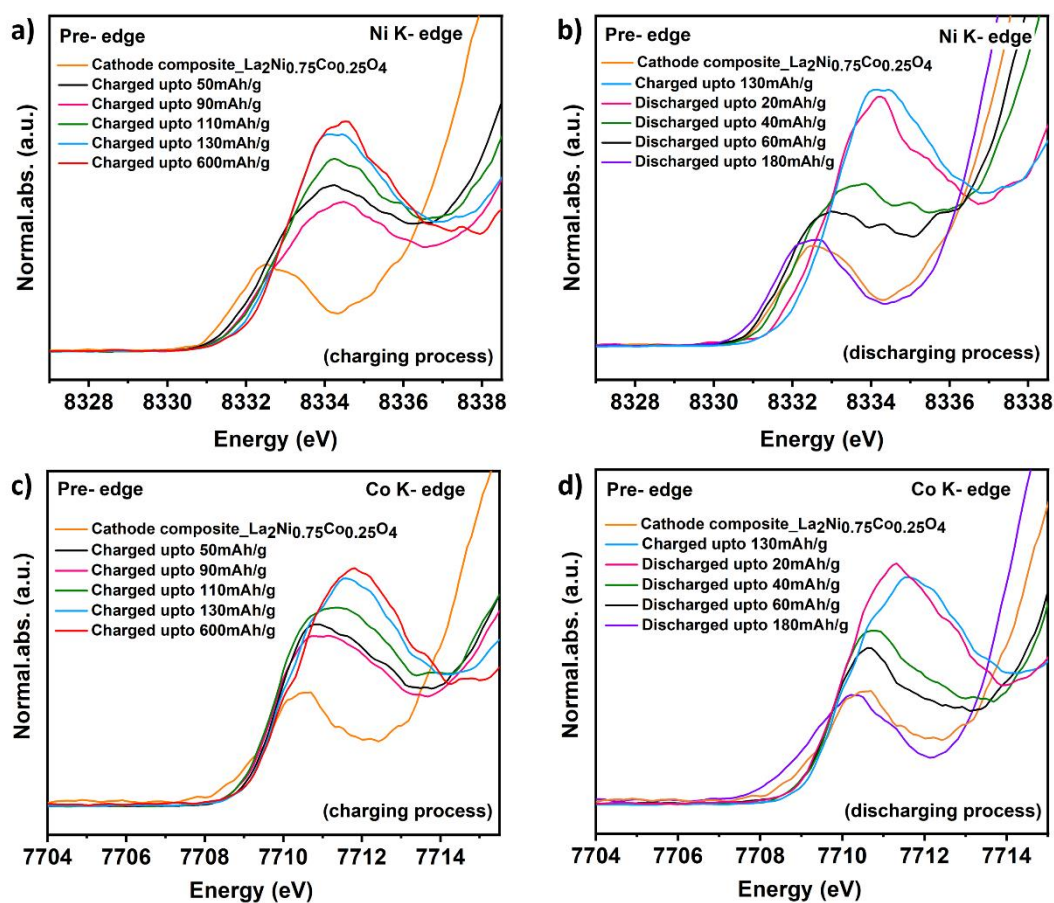


Figure S 5. Pre- edge of HERFD- XAS at Ni K-edge (a) La₂Ni_{0.75}Co_{0.25}O_{4.08} after F insertion (b) La₂Ni_{0.75}Co_{0.25}O_{4.08} after F extraction Pre- edge at Co K-edge (c) La₂Ni_{0.75}Co_{0.25}O_{4.08} after F insertion (d) La₂Ni_{0.75}Co_{0.25}O_{4.08} after F extraction.

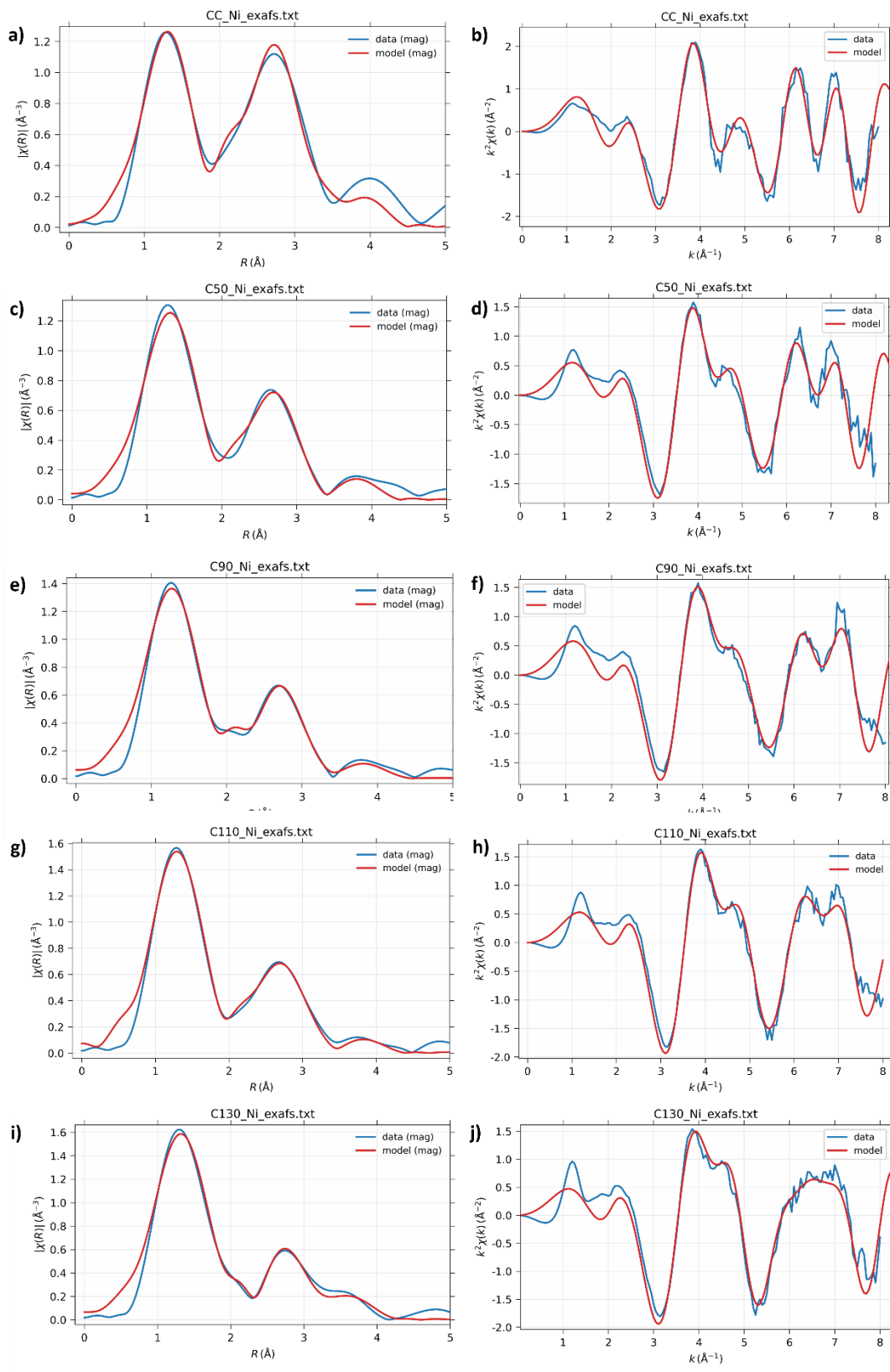


Figure S 6. EXAFS fits in R and k space at Ni K -edge of $\text{La}_2\text{Ni}_{0.75}\text{Co}_{0.25}\text{O}_{4.08}$ after F insertion.

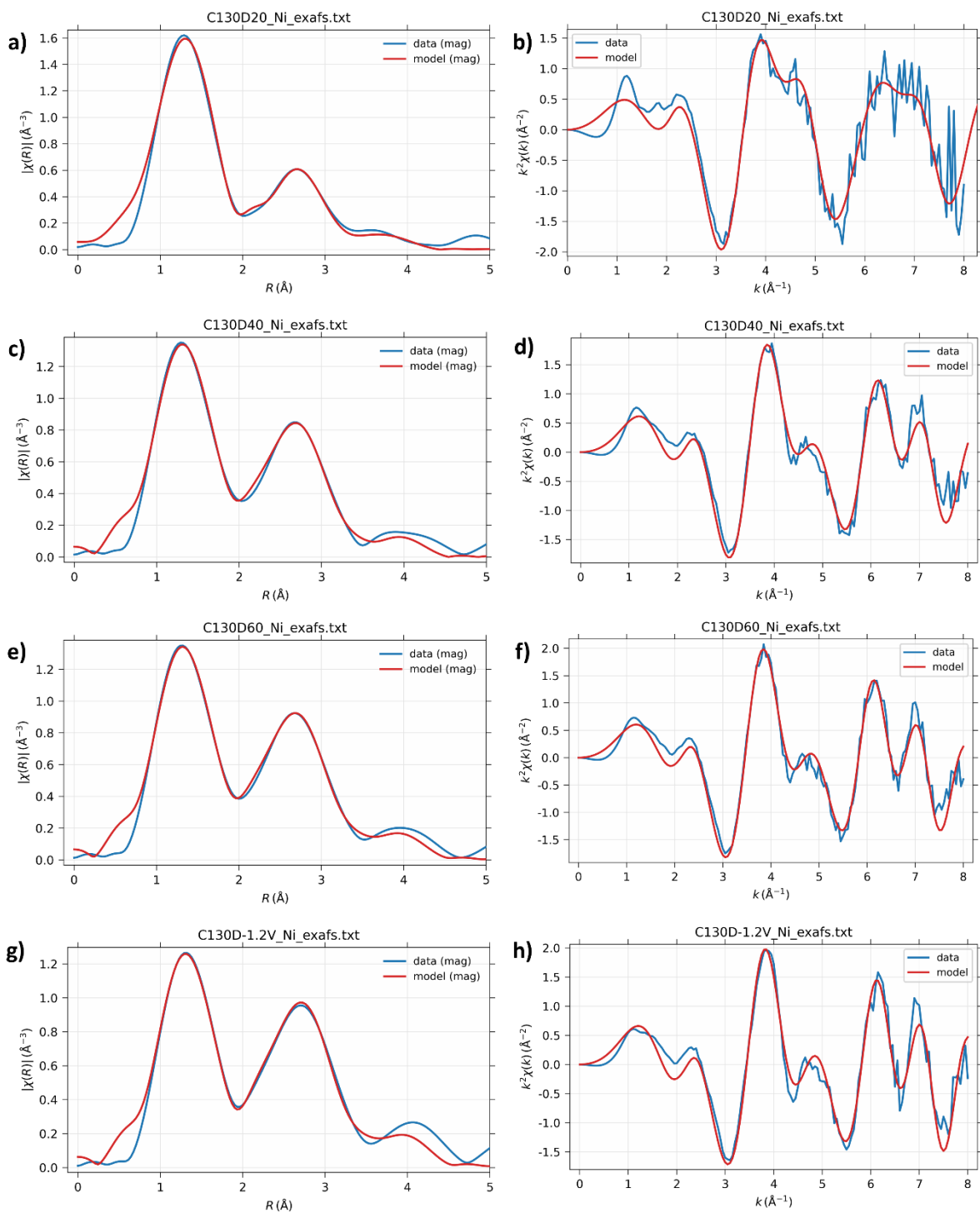


Figure S 7. EXAFS fits in R and k space at Ni K-edge of $\text{La}_2\text{Ni}_{0.75}\text{Co}_{0.25}\text{O}_{4.08}$ after F^- extraction.

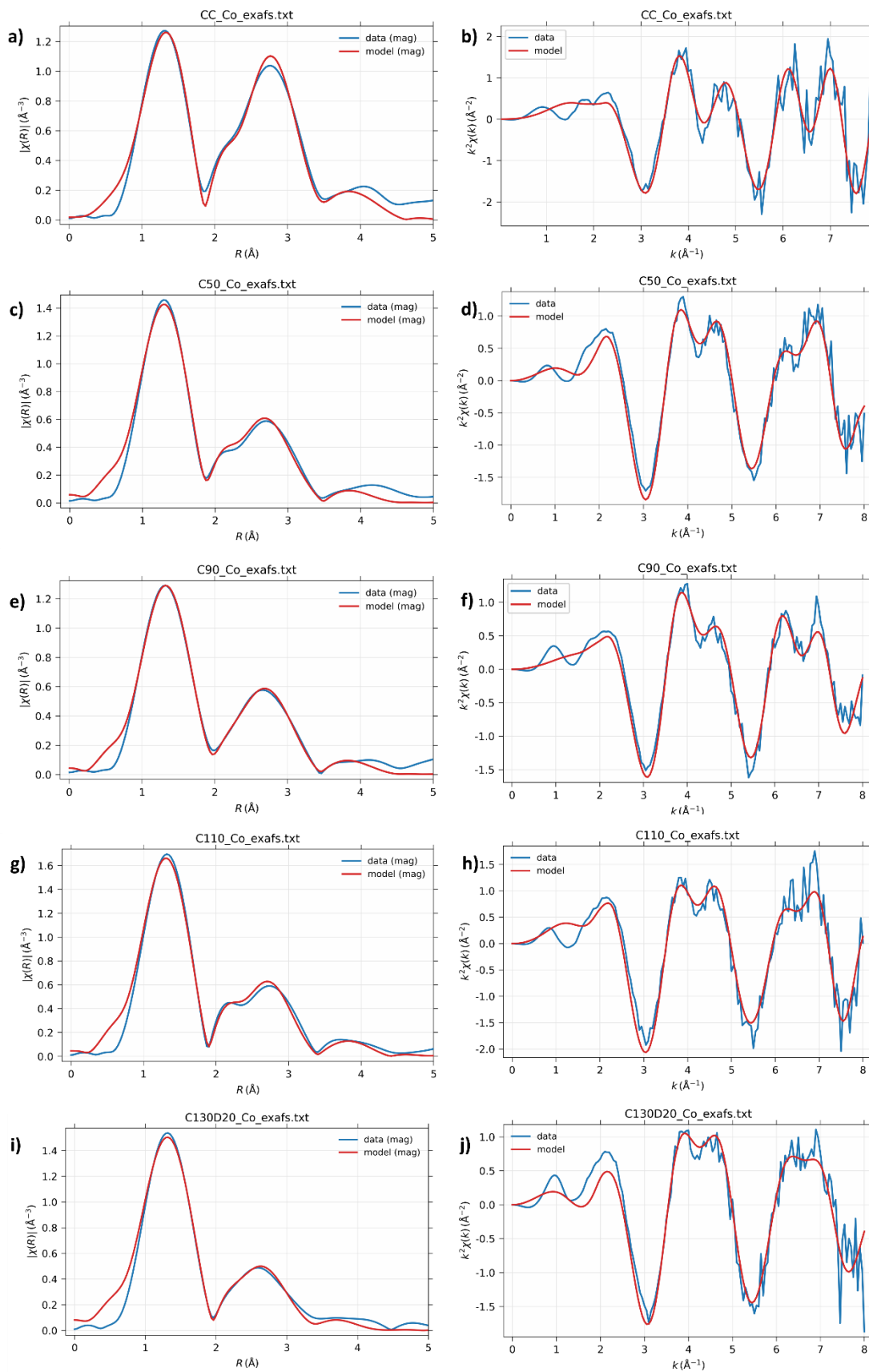


Figure S 8. EXAFS fits in R and k space at Co K -edge of $\text{La}_2\text{Ni}_{0.75}\text{Co}_{0.25}\text{O}_{4.08}$ after F insertion.

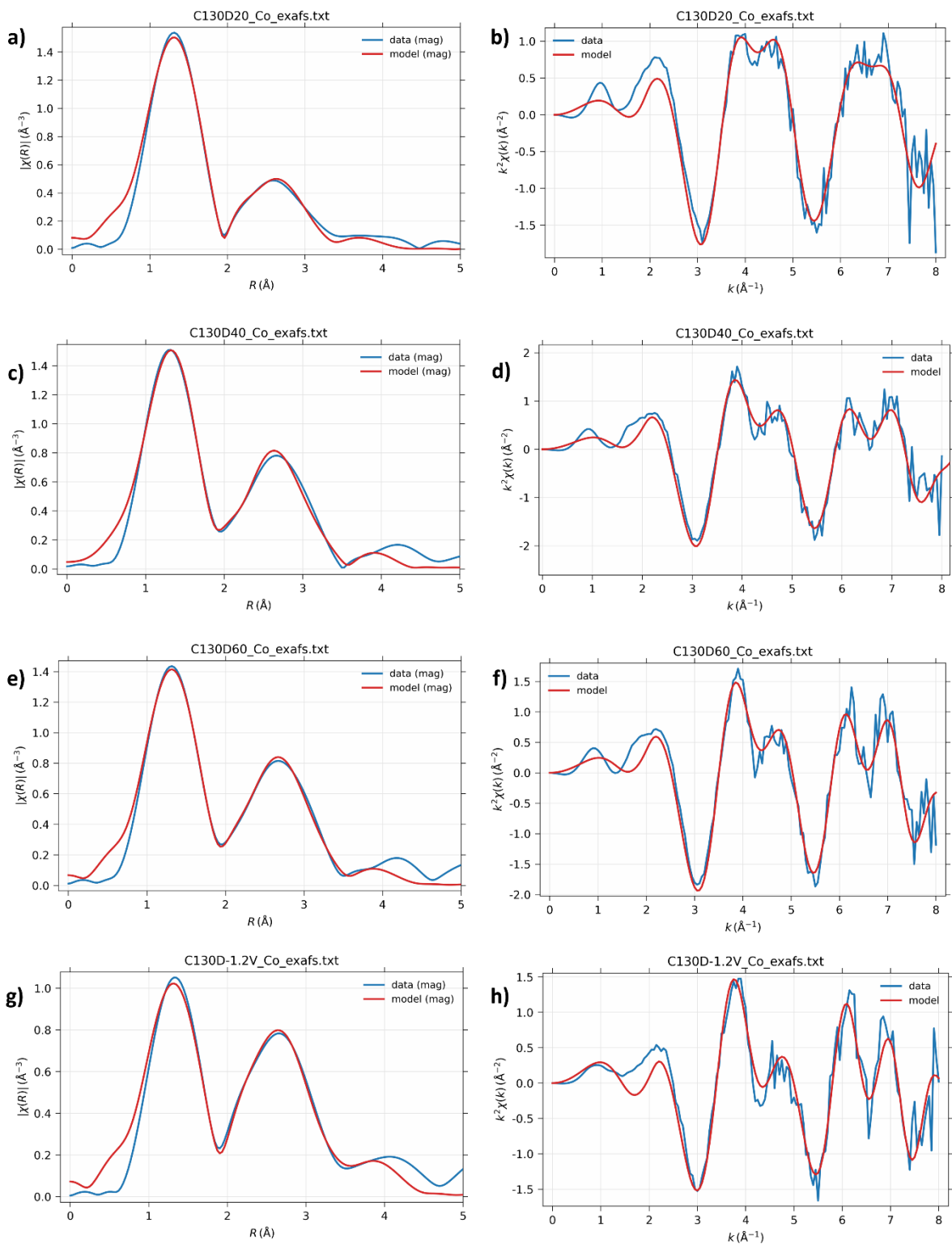


Figure S 9. EXAFS fits in R and k space at Co K-edge of $\text{La}_2\text{Ni}_{0.75}\text{Co}_{0.25}\text{O}_{4.08}$ after F extraction.

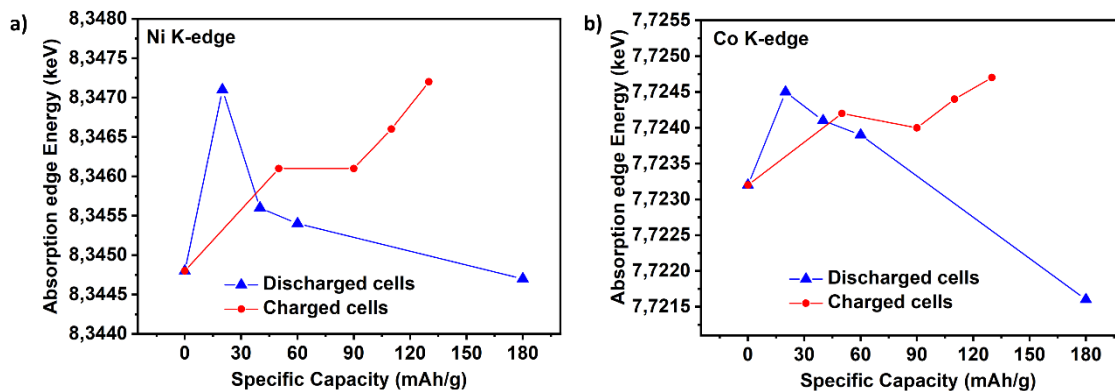


Figure S 10. Adsorption energies shift of $\text{La}_2\text{Ni}_{0.75}\text{Co}_{0.25}\text{O}_{4.08}$ during F insertion and extraction.

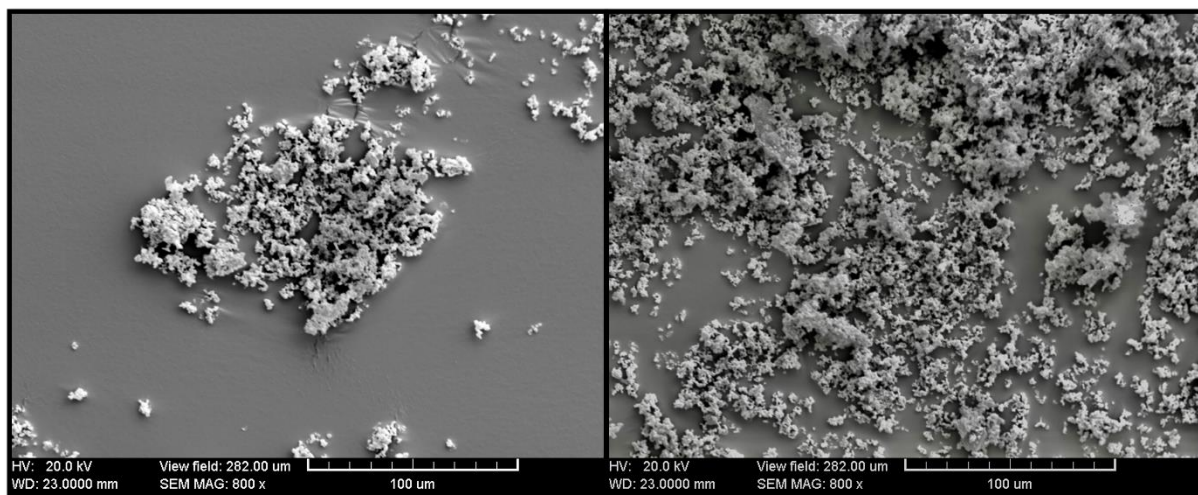


Figure S 11. SEM micrograph of $\text{La}_2\text{Ni}_{0.75}\text{Co}_{0.25}\text{O}_{4.08}$ before sintering at low magnification (800 x).

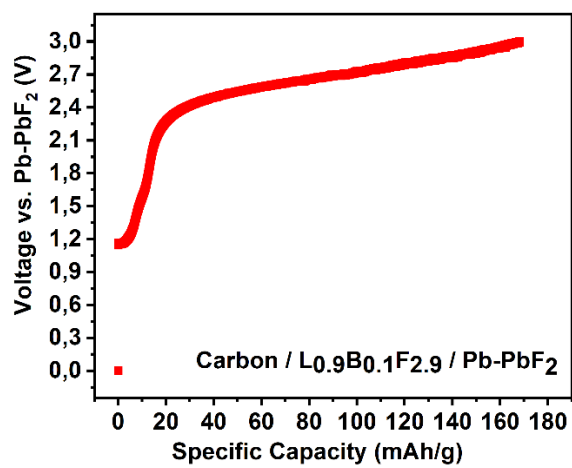


Figure S 12. Capacity-voltage curve for the cell operated only with carbon.