

Supporting Information

Graphene Nanoribbons/Ru as Efficient Cathodic Catalysts for High-Performance Rechargeable Li-CO₂ Batteries

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Table S1. Comparison of GNR Conductivity Before and After Annealing

	Temperature (°C)	Resistivity ($\Omega\cdot\text{cm}$)	Conductivity (S cm^{-1})
Before Annealing	25	0.085	11.764
After Annealing	25	0.043	23

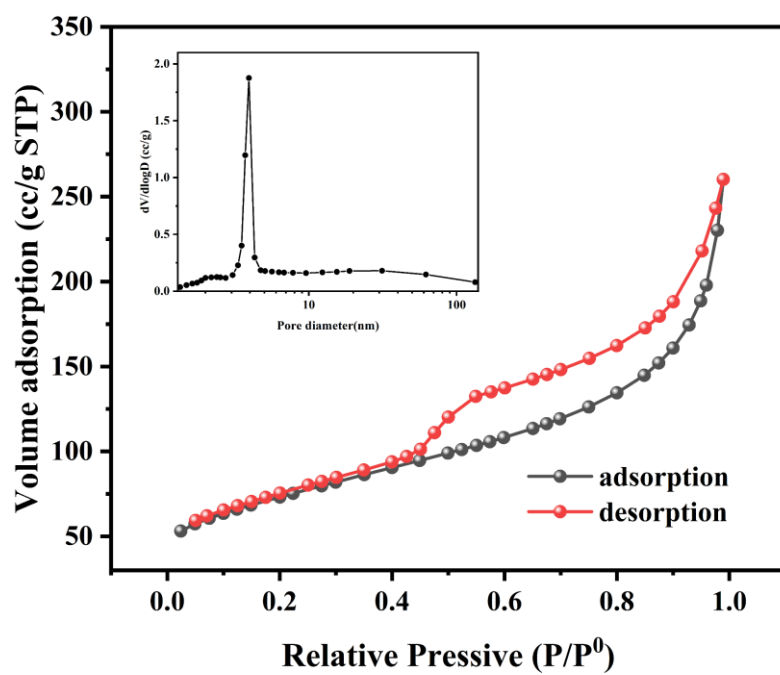


Fig. S1 N₂ adsorption–desorption isotherms of GNR (inset is the pore size distribution).

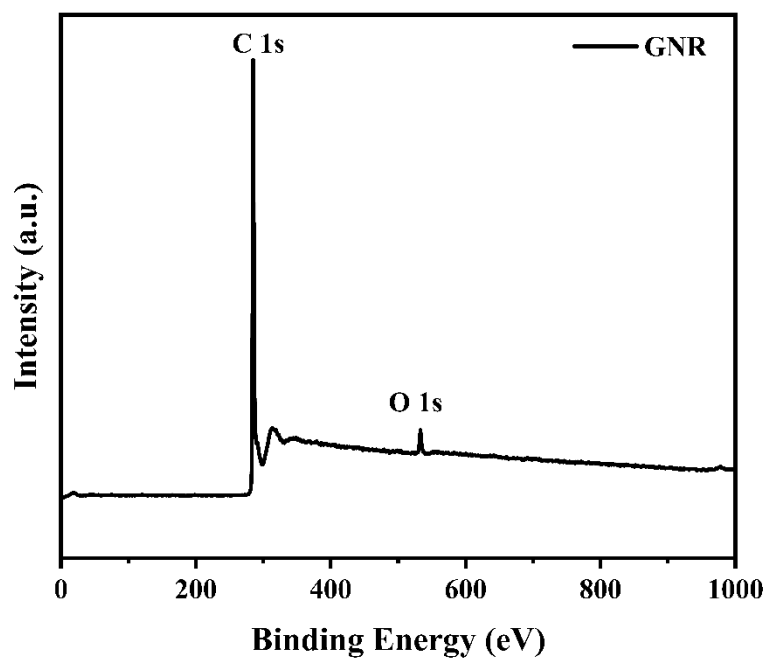


Fig. S2 Full X-ray photoelectron spectroscopy of GNR.

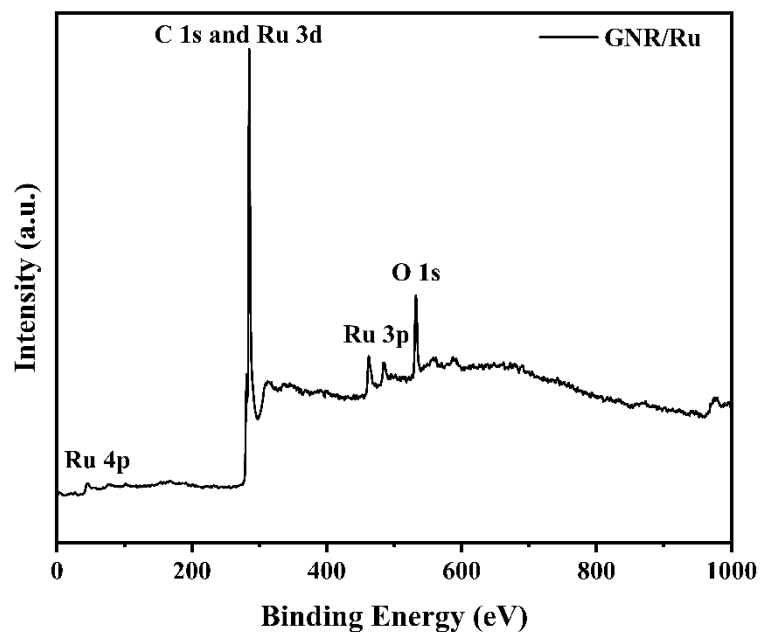


Fig. S3 Full X-ray photoelectron spectroscopy of GNR/Ru.

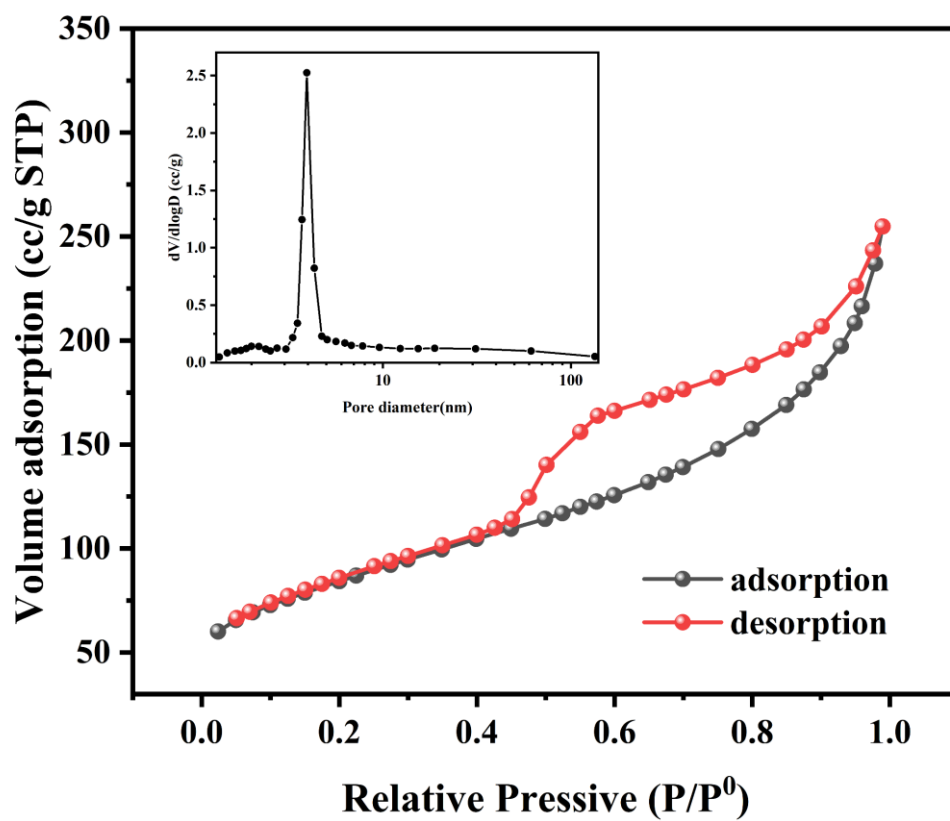


Fig. S4 N₂ adsorption–desorption isotherms of GNR/Ru catalyst.

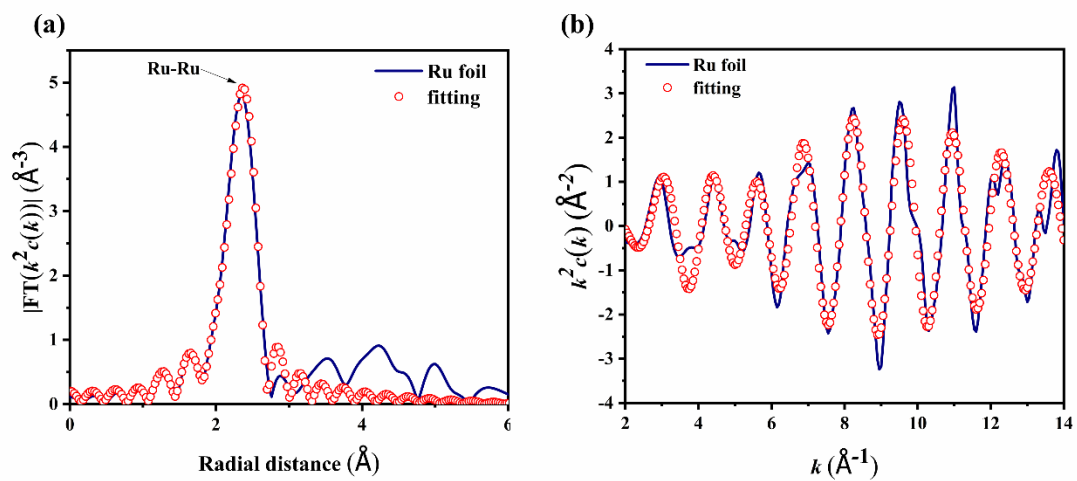


Fig. S5 (a) Fitting of the magnitude of the Fourier transform of the k^2 -weighted EXAFS and (b) the corresponding k^2 -weighted oscillation fitting curves (data-blue and fit-red) for Ru foil.

Table S2. EXAFS data fitting results of Samples.

Sample	Path	CN^a	$R(\text{\AA})^b$	$\sigma^2 (\text{\AA}^2)^c$	$\Delta E_0(\text{eV})^d$	R factor
Ru K-edge ($S_0^2=0.905$)						
Ru foil	Ru-Ru	12*	2.672±0.003	0.0041	-5.0	0.0052
GNR/Ru	Ru-C	3.7±0.8	2.026±0.012	0.0015	-12.25	0.0038
	Ru-Ru	5.1±0.4	2.668±0.003	0.0032	-3.3	

^a CN , coordination number; ^b R , the distance between absorber and backscatter atoms; ^c σ^2 , the Debye Waller factor value; ^d ΔE_0 , inner potential correction to account for the difference in the inner potential between the sample and the reference compound; R factor indicates the goodness of the fit. S_0^2 was fixed to 0.905, according to the experimental EXAFS fit of Ru foil by fixing CN as the known crystallographic value. * This value was fixed during EXAFS fitting, based on the known structure of Ru. Fitting conditions: k range: 3.0 - 14.0; R range: 1.0 - 3.0; fitting space: R space; k -weight = 2. A reasonable range of EXAFS fitting parameters: $0.800 < S_0^2 < 1.000$; $CN > 0$; $\sigma^2 > 0 \text{ \AA}^2$; $|\Delta E_0| < 15 \text{ eV}$; R factor < 0.02 .

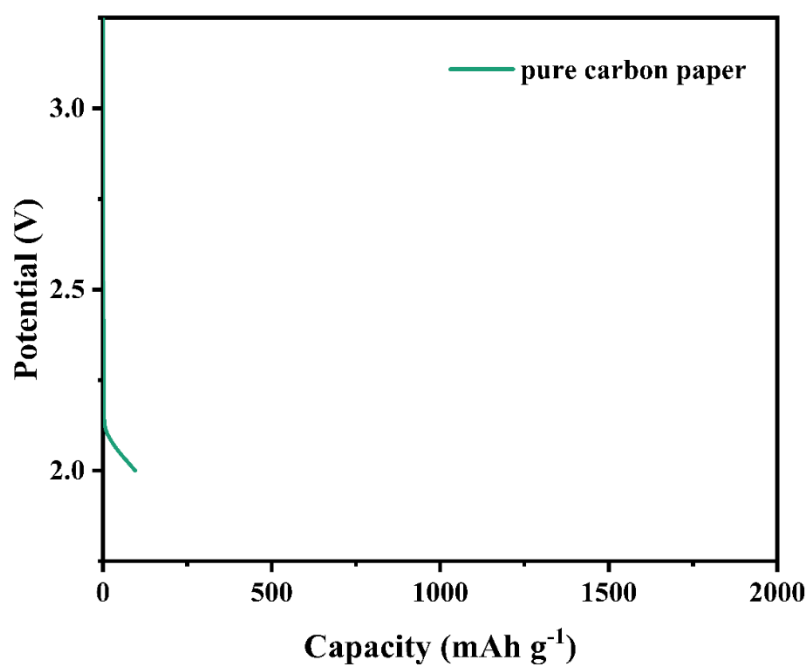


Fig. S6 Full discharge profiles of catalyst-free carbon paper.

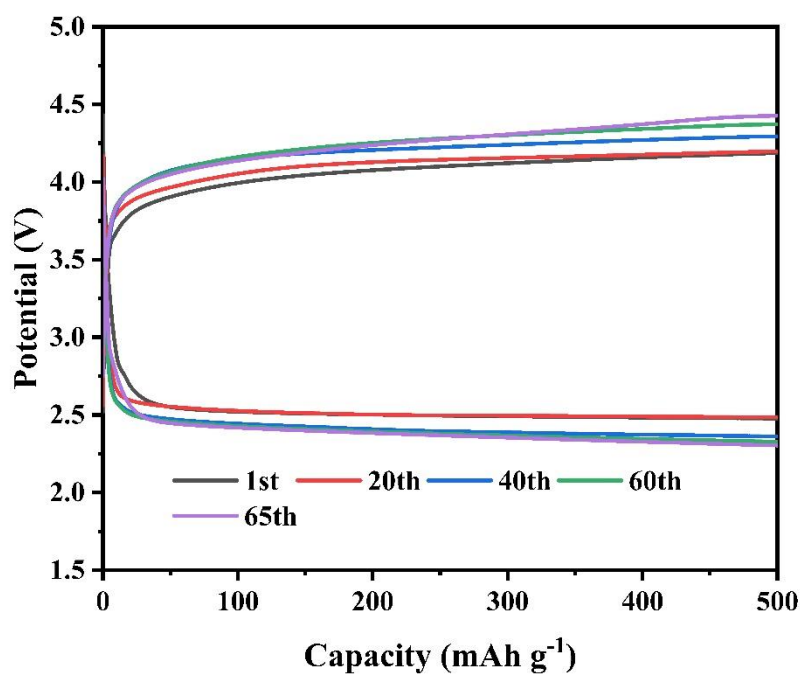


Fig. S7 Selected discharge-charge curves for LCBs with the GNR cathode.

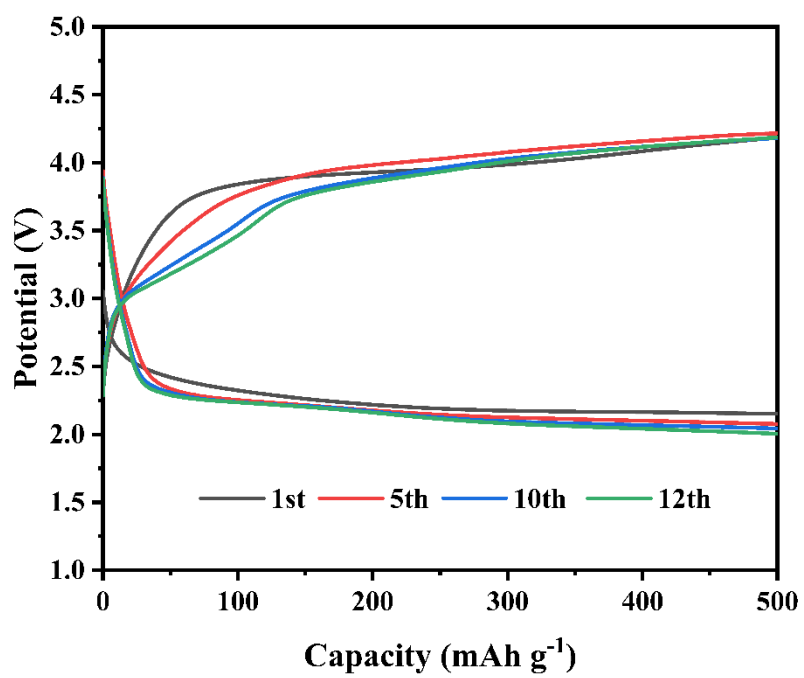


Fig. S8 Selected discharge-charge curves for LCBs with the SWCNT cathodes.

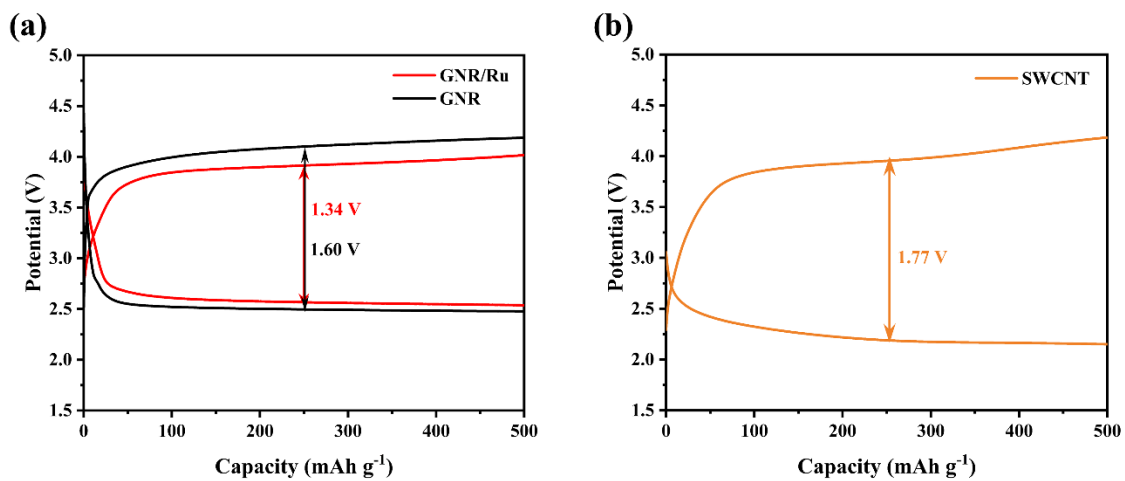


Fig. S9 Cycling overpotentials of (a) GNR/Ru, GNR and (b) SWCNT.

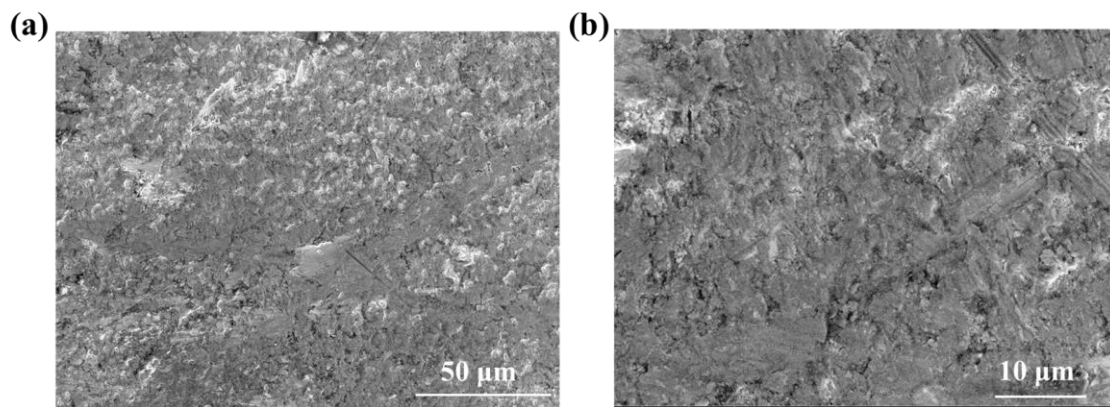


Fig. S10. SEM images of GNR/Ru cathode after long cycles.

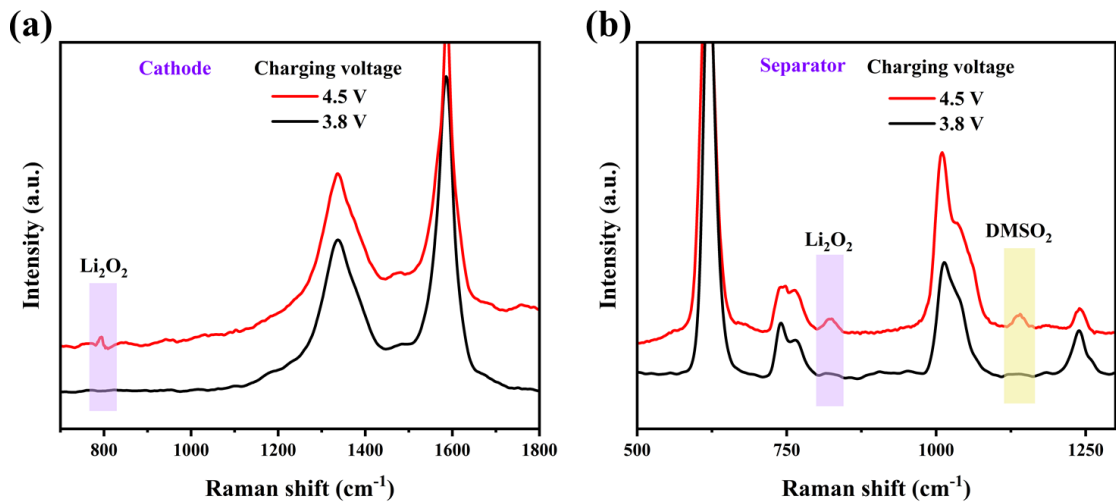


Fig. S11 Raman spectra of (a) Cathode and (b) Separator at different charging voltage.

Table S3. The electrochemical performances comparisons of different LCBs reported in the literature.

Cathodes Material	Discharge capacity (mAh g⁻¹)	Over-potential (V)	Battery operation time (h)	References
GNR/Ru	11470 (100 mA g ⁻¹)	1.34 (100 mA g ⁻¹)	1240 (100 mA g ⁻¹)	This work
CNT/Ru	4541 (100 mA g ⁻¹)	1.24 V (100 mA g ⁻¹)	450 (100 mA g ⁻¹)	[1]
RuCo/CNFs		0.98 (100 mA g ⁻¹)	360 (500 mA g ⁻¹)	[2]
Ru/ACNFs		1.35 (100 mA g ⁻¹)	1000 (100 mA g ⁻¹)	[3]
CNT@RuO ₂	2187 (50 mA g ⁻¹)	~1.4 (100 mA g ⁻¹)	1100 (50 mA g ⁻¹)	[4]
NiPc-CN MDE	18000 (200mA g ⁻¹)	1.4 (50 mA g ⁻¹)	1200 (50 mA g ⁻¹)	[5]
Co _{0.1} Ni _{0.9} O _x /CNT	5871.41 (100 mA g ⁻¹)	1.27 (100 mA g ⁻¹)	500 (100 mA g ⁻¹)	[6]
Au/CNT	6399 (100 mA g ⁻¹)	1.53 (100 mA g ⁻¹)	460 (200 mA g ⁻¹)	[7]
MXene/GO	18326 (100 mA g ⁻¹)	2.4 (100 mA g ⁻¹)	430 (100 mA g ⁻¹)	[8]
Ti ₃ C ₂ T _x		1.38	1000	[9]
MXene/CNT		(200 mA g ⁻¹)	(200 mA g ⁻¹)	
Cu-NG		0.77 (100 mA g ⁻¹)	500 (200 mA g ⁻¹)	[10]
NG/RGO		2.21 (500 mA g ⁻¹)	390 (500 mA g ⁻¹)	[11]
CQD/hG	12300 (50 mA g ⁻¹)	1.02 (100 mA g ⁻¹)	235 (1.0 A g ⁻¹)	[12]
α-MnO ₂ /CNT	7134.1 (50 mA g ⁻¹)	1.4 (100 mA g ⁻¹)	1000 (100 mA g ⁻¹)	[13]

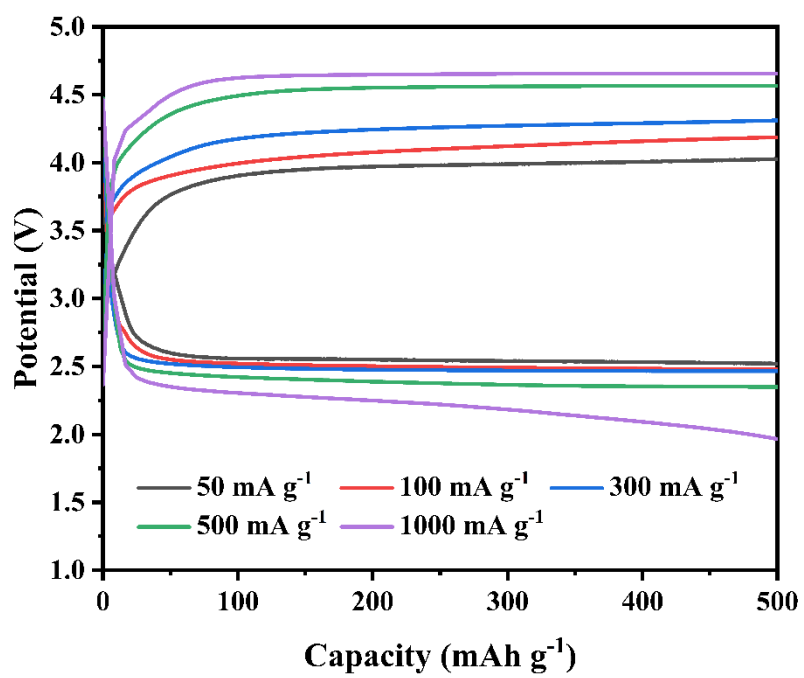


Fig. S12 GNR cathodes at different current densities.

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