

**Table S1** Energy storage parameters of BF-, BF-BT-, BNT- and NN-based ceramic systems.

Ceramics	$W_{\text{rec}}$ (J/cm <sup>3</sup> )	$\eta$ (%)	$E_b$ (kV/cm)	$\rho$ (J/kV·cm <sup>2</sup> )	reference
Bi <sub>0.83</sub> Sm <sub>0.17</sub> Fe <sub>0.95</sub> Sc <sub>0.05</sub> O <sub>3</sub>	2.21	76	230	9.6 × 10 <sup>-3</sup>	[41]
□0.67Bi <sub>0.9</sub> Sm <sub>0.1</sub> FeO <sub>3</sub> -0.33BaTiO <sub>3</sub>	2.8	55.8	200	14.0 × 10 <sup>-3</sup>	[41]
□0.56BiFeO <sub>3</sub> -0.3BaTiO <sub>3</sub> -0.14AgNbO <sub>3</sub>	2.11	84	195	10.8 × 10 <sup>-3</sup>	[15]
□0.75BiFeO <sub>3</sub> -□0.25Ba(Zn <sub>1/2</sub> Ta <sub>2/3</sub> )O <sub>3</sub>	2.56	75	160	16.0 × 10 <sup>-3</sup>	[41]
□0.75BiNdFeO <sub>3</sub> -0.12SrBaTiO <sub>3</sub>	1.74	74	150	16.6 × 10 <sup>-3</sup>	[41]
0.62BiFeO <sub>3</sub> -0.3BaTiO <sub>3</sub> -0.08Nd(Zn <sub>0.5</sub> Zr <sub>0.5</sub> )O <sub>3</sub>	2.45	72	240	10.2 × 10 <sup>-3</sup>	[45]
0.61BiFeO <sub>3</sub> -0.33(Ba <sub>0.8</sub> Sr <sub>0.2</sub> )TiO <sub>3</sub> -0.06La(Mg <sub>2/3</sub> Nb <sub>1/3</sub> )O <sub>3</sub>	3.38	59	230	14.6 × 10 <sup>-3</sup>	[42]
0.65BiFeO <sub>3</sub> -0.3BaTiO <sub>3</sub> -0.05Bi(Zn <sub>2/3</sub> Nb <sub>1/3</sub> )O <sub>3</sub>	2.1	55.7	180	11.6 × 10 <sup>-3</sup>	[42]
0.56BiFeO <sub>3</sub> -0.3BaTiO <sub>3</sub> -0.14Ba(Zn <sub>1/3</sub> Nb <sub>2/3</sub> )O <sub>3</sub>	1.61	~65	180	8.9 × 10 <sup>-3</sup>	[42]
0.61BiFeO <sub>3</sub> -0.33BaTiO <sub>3</sub> -0.06Ba(Mg <sub>1/3</sub> Nb <sub>2/3</sub> )O <sub>3</sub>	1.56	75	125	12.4 × 10 <sup>-3</sup>	[42]
0.7(0.67BiFeO <sub>3</sub> -0.33BaTiO <sub>3</sub> )-0.3(Sr <sub>0.7</sub> Bi <sub>0.2</sub> )TiO <sub>3</sub>	2.4	90.4	180	13.3 × 10 <sup>-3</sup>	[42]
0.75(Bi <sub>0.85</sub> Nd <sub>0.15</sub> )FeO <sub>3</sub> -0.25BaTiO <sub>3</sub>	1.8	41.3	170	10.5 × 10 <sup>-3</sup>	[42]
□0.52Bi <sub>0.98</sub> La <sub>0.02</sub> FeO <sub>3</sub> -0.48BaTiO <sub>3</sub>	1.22	58	140	8.71 × 10 <sup>-3</sup>	[42]
0.65Bi <sub>0.5</sub> Na <sub>0.5</sub> TiO <sub>3</sub> -0.35BaTiO <sub>3</sub> -SrZr <sub>0.5</sub> Ti <sub>0.5</sub> O <sub>3</sub>	4.32	93.5	302	14.3 × 10 <sup>-3</sup>	[43]
□□0.8BNT-0.2SrNb <sub>0.5</sub> Al <sub>0.5</sub> O <sub>3</sub>	6.5	89	480	13.5 × 10 <sup>-3</sup>	[44]
□0.16BNT-0.8NaNbO <sub>3</sub> -0.04CaZrO <sub>3</sub>	3.7	82.1	400	9.2 × 10 <sup>-3</sup>	[44]
(Na <sub>0.73</sub> Bi <sub>0.08</sub> Sm <sub>0.01</sub> )(Nb <sub>0.91</sub> Ta <sub>0.09</sub> )O <sub>3</sub>	1.66	83.6	214	7.7 × 10 <sup>-3</sup>	[45]
0.85(0.92NaNbO <sub>3</sub> -0.08Bi(Mg <sub>0.5</sub> Ti <sub>0.5</sub> )O <sub>3</sub> )-0.15SrTiO <sub>3</sub>	6	81	300	20.0 × 10 <sup>-3</sup>	[46]
0.2NaNbO <sub>3</sub> -0.8Sm(Mg <sub>0.5</sub> Zr <sub>0.5</sub> )O <sub>3</sub>	4.3	85.6	560	7.6 × 10 <sup>-3</sup>	[47]
□0.85(0.7BiFeO <sub>3</sub> -0.3BaTiO <sub>3</sub> )-0.15NaNbO <sub>3</sub>	8.2	70	325	25.2 × 10 <sup>-3</sup>	[14]
□□Bi <sub>0.595</sub> Ba <sub>0.255</sub> Na <sub>0.15</sub> Fe <sub>0.595</sub> Ti <sub>0.255</sub> Ta <sub>0.15</sub> O <sub>3</sub>	8.7	65	350 (1Hz)	24.9 × 10 <sup>-3</sup>	This work
Bi <sub>0.595</sub> Ba <sub>0.255</sub> Na <sub>0.15</sub> Fe <sub>0.595</sub> Ti <sub>0.255</sub> Ta <sub>0.15</sub> O <sub>3</sub>	9.6	77	350 (10 Hz)	27.4 × 10 <sup>-3</sup>	This work
□Bi <sub>0.56</sub> Ba <sub>0.24</sub> Na <sub>0.2</sub> Fe <sub>0.56</sub> Ti <sub>0.24</sub> Ta <sub>0.2</sub> O <sub>3</sub>	8.7	66	550 (1 Hz)	15.8 × 10 <sup>-3</sup>	This work
□Bi <sub>0.56</sub> Ba <sub>0.24</sub> Na <sub>0.2</sub> Fe <sub>0.56</sub> Ti <sub>0.24</sub> Ta <sub>0.2</sub> O <sub>3</sub>	10.3	68	550 (10 Hz)	18.7 × 10 <sup>-3</sup>	This work

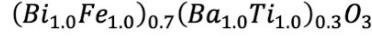
**Table S2** □ Recoverable energy storage intensity ( $\rho$ ). The calculated  $\rho$  values were obtained using the relation  $\rho = W_{\text{rec}}/\Delta E$ , based on Fig. 5(b,c,e,f).

$x$ - NT	$W_{\text{rec}}$ (J/cm <sup>3</sup> )	$\Delta E$ (kV/cm)	$\rho$ (J/kV·cm <sup>2</sup> )	Measuring frequency
0	2	175	$11.4 \times 10^{-3}$	1 Hz
0.05	5.5	275	$20.0 \times 10^{-3}$	1 Hz
0.10	7.4	350	$21.1 \times 10^{-3}$	1 Hz
□ 0.15	8.7	350	$24.9 \times 10^{-3}$	1 Hz
0.15	9.6	350	$27.4 \times 10^{-3}$	10 Hz
0.20	8.7	550	$15.8 \times 10^{-3}$	1 Hz
0.20	10.3	550	$18.7 \times 10^{-3}$	10 Hz
0.30	4.6	425	$10.8 \times 10^{-3}$	1 Hz

$$(1-x)(0.7BiFeO_3 - 0.3BaTiO_3) - xNaTaO_3$$

$$S_{config} = -R \left[ \sum_{(A-site\ cations)} x_i \ln(x_i) + \sum_{(B-site\ cations)} x_j \ln(x_j) + \sum_{(anions)} x_k \ln(x_k) \right]$$

at  $x = 0$

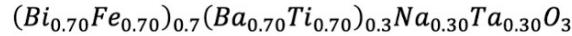


$$S_{config} = -R[2 * 0.7 \ln(0.7) + 2 * 0.3 \ln(0.3) + 1.0 \ln(1.0)]$$

$$S_{config} = -R[2 * 0.7(-0.356) + 2 * 0.3(-1.203) + 0]$$

$$S_{config} = R[0.498 + 0.721] = 1.21R$$

at  $x = 0.30$



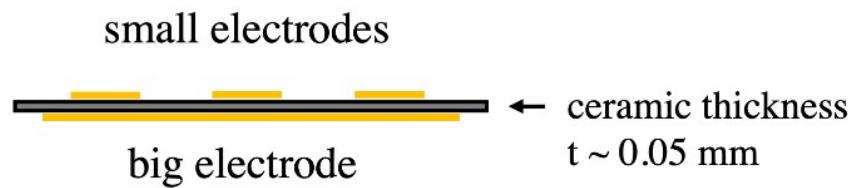
$$S_{config} = -R[2 * 0.49 \ln(0.49) + 2 * 0.21 \ln(0.21) + 2 * 0.30 \ln(0.30) + 1.0 \ln(1.0)]$$

$$S_{config} = -R[2 * 0.49(-0.713) + 2 * 0.21(-1.560) + 2 * 0.30(1.203) + 0]$$

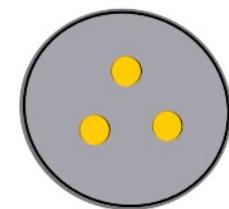
$$S_{config} = R[0.698 + 0.655 + 0.721] = 2.07R$$

**Figure S1** The calculation process of configuration entropy.

## Side view

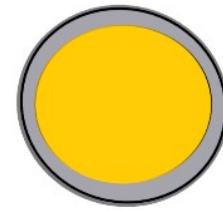


## Top view



top electrodes  
1 mm diameter

## Bottom view

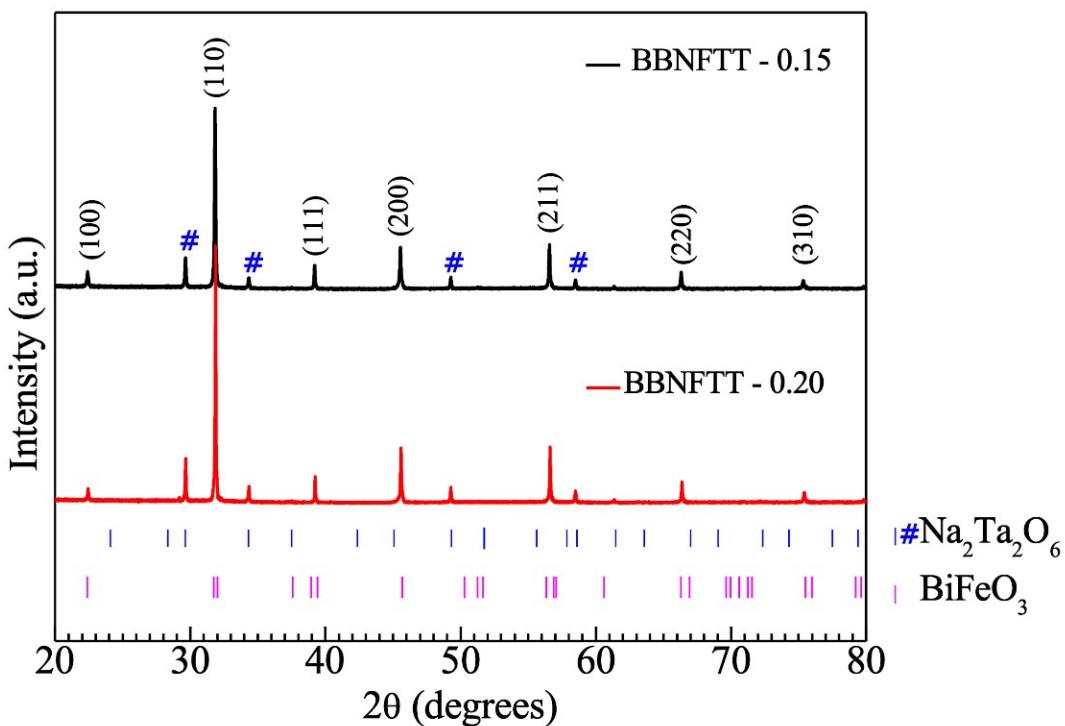


bottom electrode  
5 mm diameter

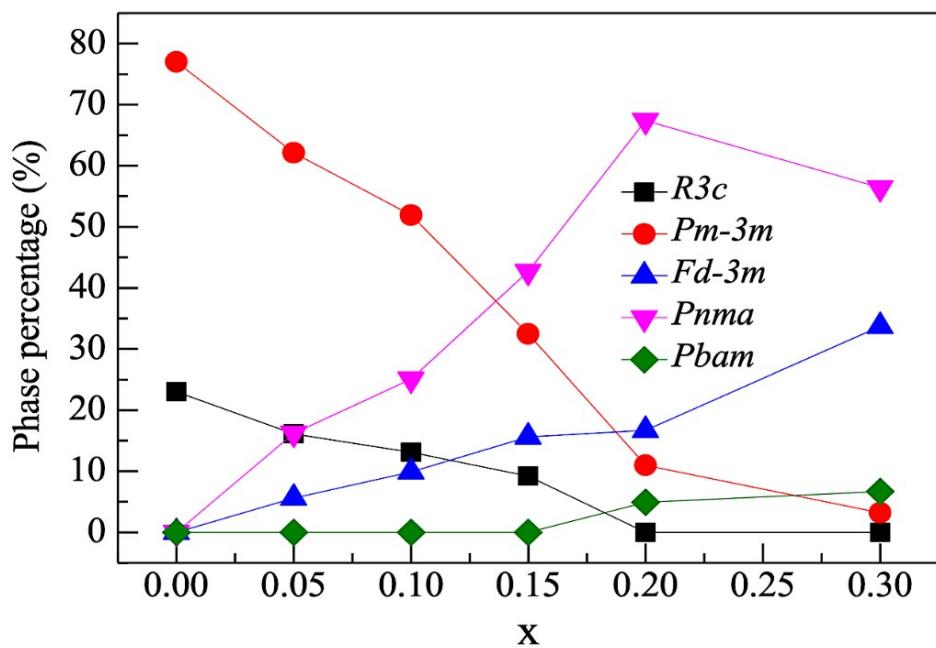
## Actual sample



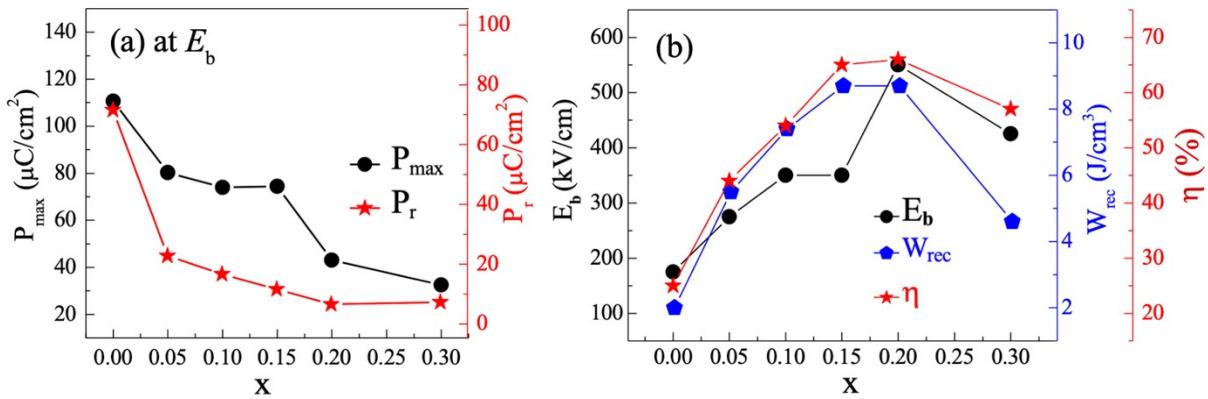
**Figure S2** Electrode's configuration for the P-E loop experiment.



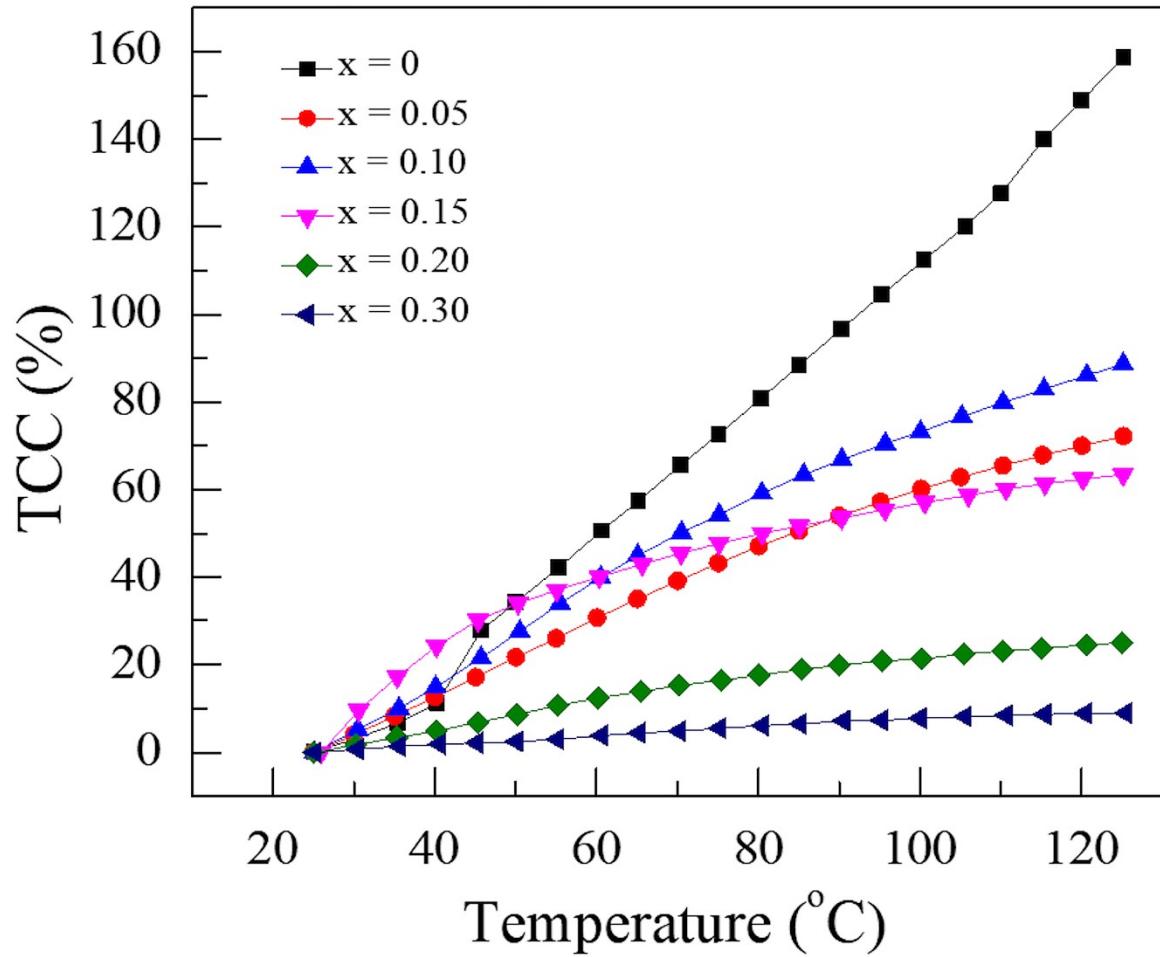
**Figure S3** Identification of the cubic  $Fd\text{-}3m$   $\text{Na}_2\text{Ta}_2\text{O}_6$  phase in the  $\text{NaTaO}_3$ -contained specimens.



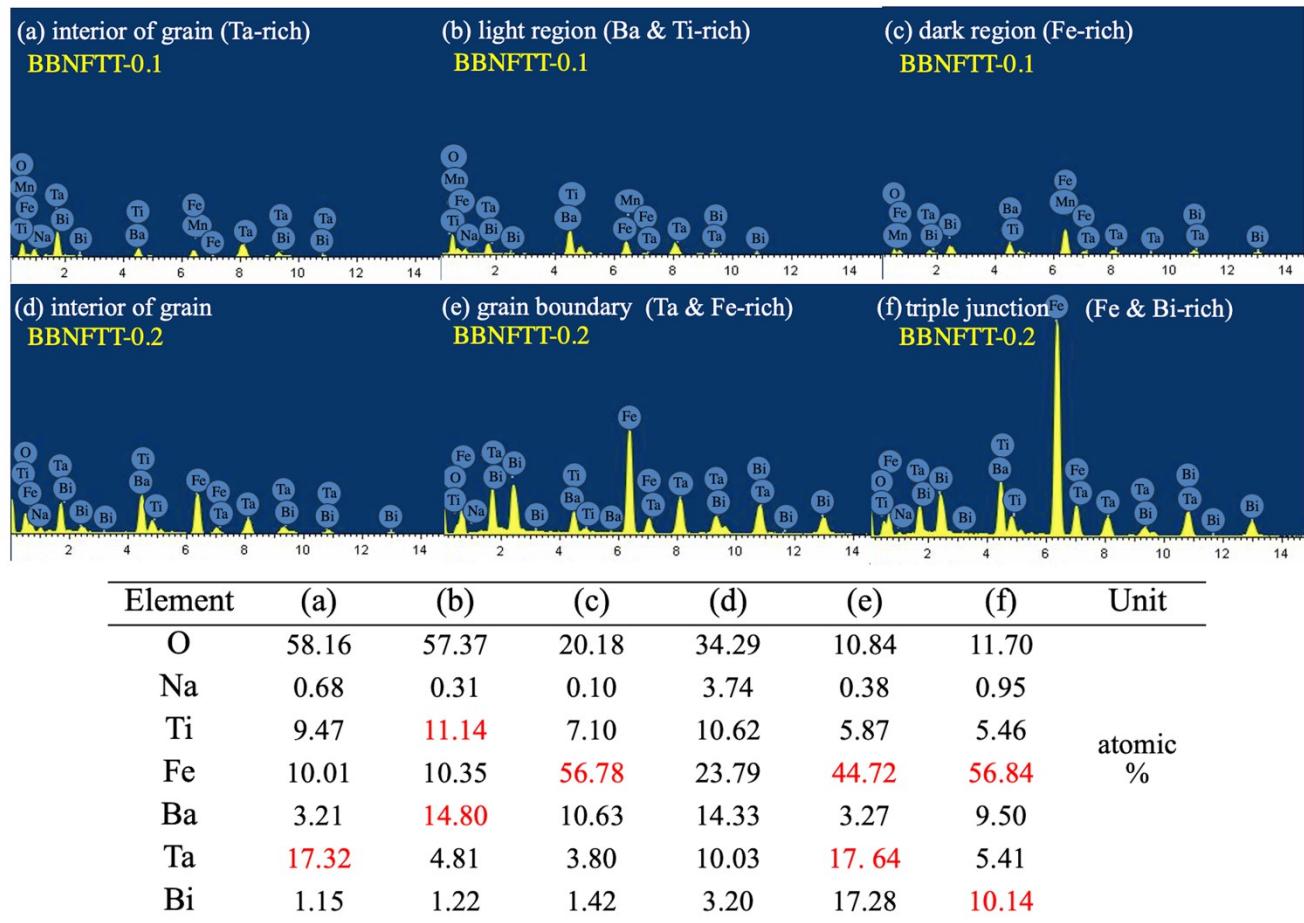
**Figure S4** Variations of different phase percentages (wt.%) of BBNFTT- $x$  ceramics.



**Figure S5** (a)  $P_{max}$  and  $P_r$  measured at  $E_b$  and (b)  $E_b$ ,  $W_{rec}$ , and  $\eta$  vs.  $x$  of BBNFTT- $x$  ceramics.



**Figure S6** Temperature coefficients of capacitance (TCC) for BBNFTT- $x$  from  $x = 0$  to  $x = 0.30$  for measuring frequency of 1 kHz.



**Figure S7** EDX spectra of (a-c) BBNFTT-0.1 and (d-f) BBNFTT-0.2.