## Supplementary Information:

## Beyond Molecular Structure: Critically Assessing Machine Learning for Designing Organic Photovoltaic Materials and Devices

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**Figure S1.** Heatmap of model performance for predicting PCE from the molecular structure of the donor and acceptor materials using various models and representations as measured by (a) RMSE, (b) MAE, and (c) R scores.

Target	Score	Donors	Acceptors
НОМО	R <sup>2</sup>	-0.09	0.21
HOMO	MSE	1.24	0.46
LUMO	R <sup>2</sup>	0.01	0.54
	MSE	0.70	0.51
⊏ opt	R <sup>2</sup>	0.43	0.68
	MSE	0.49	0.31
Best loss		0.76	0.42

**Table S1.** Performance metrics of GNN models from which graph embeddings were extracted.

Table S2.	Table	of model	performance	for the F	PUFp r	epresentation.
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Model	R2	RMSE	MAE	R
MLR				0.02±0.04
kNN	0.43±0.03	2.32±0.07	1.76±0.06	0.66±0.02
SVR	0.43±0.03	2.32±0.23	1.74±0.06	0.67±0.02
KRR	0.45±0.03	2.29±0.06	1.76±0.05	0.68±0.02
GP	0.35±0.03	2.47±0.08	1.87±0.06	0.61±0.02
RF	0.48±0.02	2.22±0.07	1.67±0.05	0.70±0.02
HGB	0.40±0.03	2.37±0.06	1.85±0.05	0.65±0.02
XGB	0.47±0.02	2.24±0.07	1.71±0.05	0.70±0.02
NGB	0.42±0.02	2.35±0.07	1.85±0.05	0.66±0.02
MLP	0.37±0.04	2.44±0.09	1.91±0.08	0.65±0.02



**Figure S2.** Heatmap parity plots of HGBR model predictions using 5-fold CV initialized from 7 random seeds. (a) The HGBR algorithm outperforms RF models trained on computed descriptors from the 2023 Greenstein and Hutchison dataset<sup>1</sup> (ca. 1000 unique D:A pairs) using much simpler molecular representations (ECFP and incomplete material properties). (b) The HGBR algorithm performs comparably to RF models trained on mordred descriptors and material properties from the 2021 Miyake and Saeki dataset<sup>2</sup> (ca. 1300 data points) using ECFP and material properties.



**Figure S3.** Heatmaps of performance of multi-output RF, HGB, and ANN models trained on the ECFP molecular representations for predicting FF,  $V_{OC}$ ,  $J_{SC}$ , PCE (directly), and PCE (from Equation 1 in the main text) concurrently, as measure by (a)  $R^2$ , (b) RMSE, (c) MAE scores.

Name	Energy (eV)	Energy level	Measurement technique	Reference
C <sub>60</sub> -bissalt	-3.9	LUMO	CV	3
Са	-2.9	WF		4
EDTA-ZnO	-4.05	WF	UPS	5
FPI	-3.93	LUMO	CV	3
LiF	-3	WF		
AI	-4.3	WF		6
N719:PrC <sub>60</sub> MAI	-3.9	WF	UPS	7
PDIN	-3.6			
PDINO	-3.9			6
PEIE	-4.1	WF		8
PFN	-2.14	LUMO	CV	9
ZnO/PFN	-2.14	LUMO	CV	9
PFN-Br	-2.18	LUMO	CV	10
PFNDI-Br	-4.18	LUMO	CV	11
PNDIT-F3N	-3.9	LUMO	CV	11
PNDIT-F3N-Br	-4.18	LUMO	CV	11
TiO <sub>2</sub> :TOPD	-4.23	WF	КР	12
ZnO	-4.46	WF		13
ZnO/PFN-Br	-4.08	LUMO	UPS + Eg <sup>opt</sup>	10
ZrAcAc	-4.65	WF	UPS	14
CuSCN	-5.3	WF		15
MoOx	-5.3	WF	КР	16
PEDOT:PSS	-5.2	WF	UPS	13
V <sub>2</sub> O <sub>5</sub>	-4.7			17

**Table S3.** Encodings of HTLs and ETLs along with their energy level, measurement method, and reference. CV: cyclic voltammetry; KP: Kelvin probe; UPS: ultraviolet photon spectroscopy; WF: work function.

Abbreviation	Descriptor	Units
dipole	Electric dipole moment	D
dD	Hansen solubility parameter: energy from dispersion forces	MPa <sup>0.5</sup>
dP	Hansen solubility parameter: energy from intermolecular forces	MPa <sup>0.5</sup>
dH	Hansen solubility parameter: energy from hydrogen bonds	MPa <sup>0.5</sup>
dHDon	Hansen solubility parameter: hydrogen bonding donor term	MPa <sup>0.5</sup>
dHAcc	Hansen solubility parameter: hydrogen bonding acceptor term	MPa <sup>0.5</sup>
MW	Molecular weight	
Density	Density at 25 °C	g/cm³
BPt	Boiling point	°C
MPt	Melting point	°C
logKow	Log of octanol-water partition coefficient	
RI	Refractive index	
Trouton	Trouton's rule: (HVap at BPt) / BPt	J/kmol
RER	Relative evaporation rate	
ParachorGA	Parachor for surface tension	
RD	Molecular refractivity	
DCp	Heat capacity at 25 °C	J/kmol
log n	Log of viscosity at 25 °C	mPa•s
SurfTen	Surface tension at 25 °C	mN/m

Table S4. Physicochemical descriptors of solvents and solvent additives from HSPiP.<sup>18</sup>



Figure S4. Plot of dataset features in the various filters.



**Figure S5.** Pearson correlation coefficient heatmaps of the different groups of processing features with the output variables: (a) material properties, (b) processing, (c) device architecture, (d) charge carrier mobilities. Feature groups correspond to those defined in Figure S4. Correlation coefficient values in each cell are rounded to the nearest tenth for readability.



**Figure S6.** Spearman correlation coefficient heatmaps of the different groups of processing features with the output variables: (a) material properties, (b) processing, (c) device architecture, (d) charge carrier mobilities. Feature groups correspond to those defined in Figure S4. Correlation coefficient values in each cell are rounded to the nearest tenth for readability.



**Figure S7.** Pearson correlation coefficient heatmaps of (a) solvent and (b) solvent additive descriptors with the output variables. Correlation coefficient values in each cell are rounded to the nearest tenth for readability.



**Figure S8.** Spearman correlation coefficient heatmaps of (a) solvent and (b) solvent additive descriptors with the output variables. Correlation coefficient values in each cell are rounded to the nearest tenth for readability.



Structural Representations and Device Fabrication Features

**Figure S9.** Heatmaps of model performance as measured by (a) RMSE and (b) MAE scores for predicting PCE from molecular structure and various subsets of OPV device fabrication parameters.



**Figure S10.** Scatter plots of predictions from (a) GP trained on ECFP only, (b) HGB trained on ECFP only, (c) RF trained on ECFP only, (d) RF trained on Mordred descriptors, (e) SVR trained on Mordred descriptors, (f) RF trained on ECFP and material properties.



Figure S11. HGB trained on Mordred descriptors and log mobilities all subset.



**Figure S12.** Heatmaps of model performance as measured by the (a)  $R^2$ , (b) RMSE, (c) MAE scores for predicting PCE from molecular structure and the device architecture subset of features when imputing missing data using different algorithms. Numbers in each cell correspond to the average score of the model over seven independent five-fold cross-validations ± the standard error of the mean.



Figure S13. Distributions of target values in the dataset.



**Figure S14.** D:A heatmap. Most common donors: PTB7-Th, J71, P3HT, PBDB-T. Most common acceptors: ITIC, IT-4F, m-ITIC. X: more positive left, more negative right. Y: more positive top, more negative bottom.





Figure S16. Distributions of solvent descriptor values.



Figure S17. Distributions of solvent additive descriptor values.

Feature	Mean	Std Dev	Median	Mode	Skew	Min	Мах	Missing (%)
Donor PDI	2.15	0.72	2	2	4.6	1.18	8.59	63.8
Donor Mn (kDa)	41.21	32.79	30	23.5	2.22	6.97	180.4	66.85
Donor Mw (kDa)	89.59	73.36	63.6	50.3	2.21	13.4	364.9	81
HOMO_D (eV)	-5.3	0.12	-5.3	-5.3	-0.01	-5.68	-5.01	0
LUMO_D (eV)	-3.4	0.2	-3.43	-3.43	1.11	-3.75	-2.78	0
Ehl_D (eV)	1.9	0.18	1.87	1.87	1.25	1.57	2.79	0
Eg_D (eV)	1.79	0.13	1.77	1.77	-0.17	1.48	2.12	0
HOMO_A (eV)	-5.56	0.16	-5.58	-5.58	-0.65	-6.38	-5.02	0
LUMO_A (eV)	-3.87	0.16	-3.91	-3.92	1.29	-4.23	-2.83	0
Ehl_A (eV)	1.69	0.19	1.66	1.66	0.89	1.1	2.61	0
Eg_A (eV)	1.6	0.18	1.59	1.59	1.21	1.1	2.43	0
D:A ratio (m/m)	0.92	0.26	1	1	2.04	0.33	3.33	2.51
Active layer spin coating speed (rpm)	2158.27	795.79	2000	3000	0.35	600	5000	52.33
total solids conc. (mg/mL)	16.8	5.13	18	20	0.16	6	37.5	20.61
solvent additive conc. (% v/v)	0.44	0.86	0	0	3.18	0	6	1.25
active layer thickness (nm)	103.61	25.17	100	100	3.77	40	321	25.99

**Table S5.** Statistical analysis of feature distributions in the dataset.

Feature	Mean	Std Dev	Median	Mode	Skew	Min	Мах	Missing (%)
Annealing temperatur e (°C)	73.01	51.6	25	25	0.36	25	175	1.79
Annealing time (min)	4.66	7.51	0	0	2.92	0	60	10.75
HTL energy level (eV)	-5.24	0.06	-5.2	-5.2	2.5	-5.3	-4.7	0.72
HTL thickness (nm)	20.63	14.04	10	10	0.32	3	40	35.48
ETL energy level (eV)	-3.73	0.83	-3.9	-4.46	0.85	-4.65	-2.14	1.25
ETL thickness (nm)	18.92	11.6	20	30	0.2	0.7	50	54.12
hole mobility blend (cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> )	0	0	0	0	9.86	0	0.03	18.28
electron mobility blend (cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> )	0	0	0	0	7.59	0	0.01	17.56
Voc (V)	0.89	0.11	0.9	0.9	0.03	0.54	1.34	0
J <sub>SC</sub> (mAcm <sup>-</sup> <sup>2</sup> )	15.07	4.74	15.96	16.8	-0.76	0.05	26.6	0
FF	0.62	0.1	0.63	0.65	-1.13	0.24	0.84	0
PCE (%)	8.44	3.09	9.05	6.4	-0.64	0.01	15.7	0
hole:electro n mobility ratio	33.61	440.17	1.52	2	19.67	0.01	9090.91	20.07
log hole mobility blend (cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> )	-3.79	0.76	-3.7	-3.82	-1.13	-7.36	-1.57	18.28
log electron	-4.02	0.85	-3.87	-3.92	-1.54	-8.96	-2.01	17.56

Feature	Mean	Std Dev	Median	Mode	Skew	Min	Мах	Missing (%)
mobility blend (cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> )								
log hole:electro n mobility ratio	0.23	0.69	0.18	0.3	0.9	-2.04	3.96	20.07
calculated PCE (%)	8.43	3.09	9.02	4.5	-0.64	0.01	15.71	0

**Table S6.** Performance of HGB models trained on ECFP molecular representations with different radii and numbers of bits.

Representation	Radius	Bits	R <sup>2</sup>	RMSE	MAE
ECFP6	3	512	0.56±0.02	2.05±0.07	1.53±0.05
ECFP8	4	1024	0.57±0.03	2.02±0.08	1.50±0.05
ECFP10	5	2048	0.58±0.03	2.00±0.08	1.48±0.05
ECFP12	6	4096	0.58±0.03	1.99±0.08	1.49±0.05

## References

- 1. Greenstein, B. L. & Hutchison, G. R. Screening Efficient Tandem Organic Solar Cells with Machine Learning and Genetic Algorithms. *J. Phys. Chem. C* **127**, 6179–6191 (2023).
- 2. Miyake, Y. & Saeki, A. Machine Learning-Assisted Development of Organic Solar Cell Materials: Issues, Analyses, and Outlooks. *J. Phys. Chem. Lett.* **12**, 12391–12401 (2021).
- 3. Li, C.-Z. *et al.* Effective interfacial layer to enhance efficiency of polymer solar cells via solution-processed fullerene-surfactants. *J. Mater. Chem.* **22**, 8574–8578 (2012).
- 4. Allen, T. G. *et al.* Calcium contacts to n-type crystalline silicon solar cells. *Prog. Photovolt. Res. Appl.* **25**, 636–644 (2017).
- Li, X., Liu, X., Zhang, W., Wang, H.-Q. & Fang, J. Fullerene-Free Organic Solar Cells with Efficiency Over 12% Based on EDTA–ZnO Hybrid Cathode Interlayer. *Chem. Mater.* 29, 4176–4180 (2017).
- Karki, A. *et al.* The role of bulk and interfacial morphology in charge generation, recombination, and extraction in non-fullerene acceptor organic solar cells. *Energy Environ. Sci.* 13, 3679–3692 (2020).
- Li, W. *et al.* A New Function of N719: N719 Based Solution-Processible Binary Cathode Buffer Layer Enables High-Efficiency Single-Junction Polymer Solar Cells. *Sol. RRL* 1, 1700014 (2017).
- Zhang, Y., Chen, L., Hu, X., Zhang, L. & Chen, Y. Low Work-function Poly(3,4ethylenedioxylenethiophene): Poly(styrene sulfonate) as Electron-transport Layer for Highefficient and Stable Polymer Solar Cells. *Sci. Rep.* 5, 12839 (2015).
- 9. Huang, F. et al. High-Efficiency, Environment-Friendly Electroluminescent Polymers with

Stable High Work Function Metal as a Cathode: Green- and Yellow-Emitting Conjugated Polyfluorene Polyelectrolytes and Their Neutral Precursors. *J. Am. Chem. Soc.* **126**, 9845–9853 (2004).

- 10. Yang, T. *et al.* Inverted polymer solar cells with 8.4% efficiency by conjugated polyelectrolyte. *Energy Environ. Sci.* **5**, 8208–8214 (2012).
- 11. Wu, Z. *et al.* n-Type Water/Alcohol-Soluble Naphthalene Diimide-Based Conjugated Polymers for High-Performance Polymer Solar Cells. *J. Am. Chem. Soc.* **138**, 2004–2013 (2016).
- Yan, Y. *et al.* Light-Soaking-Free Inverted Polymer Solar Cells with an Efficiency of 10.5% by Compositional and Surface Modifications to a Low-Temperature-Processed TiO2 Electron-Transport Layer. *Adv. Mater.* 29, 1604044 (2017).
- 13. Schopp, N. *et al.* Understanding Interfacial Recombination Processes in Narrow-Band-Gap Organic Solar Cells. *ACS Energy Lett.* **7**, 1626–1634 (2022).
- 14. Tan, Z. *et al.* High performance polymer solar cells with as-prepared zirconium acetylacetonate film as cathode buffer layer. *Sci. Rep.* **4**, 4691 (2014).
- 15. Montgomery, A. *et al.* Solution-processed copper (I) thiocyanate (CuSCN) for highly efficient CdSe/CdTe thin-film solar cells. *Prog. Photovolt. Res. Appl.* **27**, 665–672 (2019).
- Li, X., Xie, F., Zhang, S., Hou, J. & Choy, W. C. MoOx and V2Ox as hole and electron transport layers through functionalized intercalation in normal and inverted organic optoelectronic devices. *Light Sci. Appl.* 4, e273–e273 (2015).
- 17. Chen, S. *et al.* Efficient Nonfullerene Organic Solar Cells with Small Driving Forces for Both Hole and Electron Transfer. *Adv. Mater.* **30**, 1804215 (2018).
- 18. Abbott, S. HSPiP: Hansen Solubility Parameters in Practice.