

1 **Supplementary Information**

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3 **Minimizing ion/electron pathways through ultrathin conformal** 4 **holey graphene encapsulation in Li- and Mn-rich layered oxide** 5 **cathodes for high-performance lithium-ion batteries**

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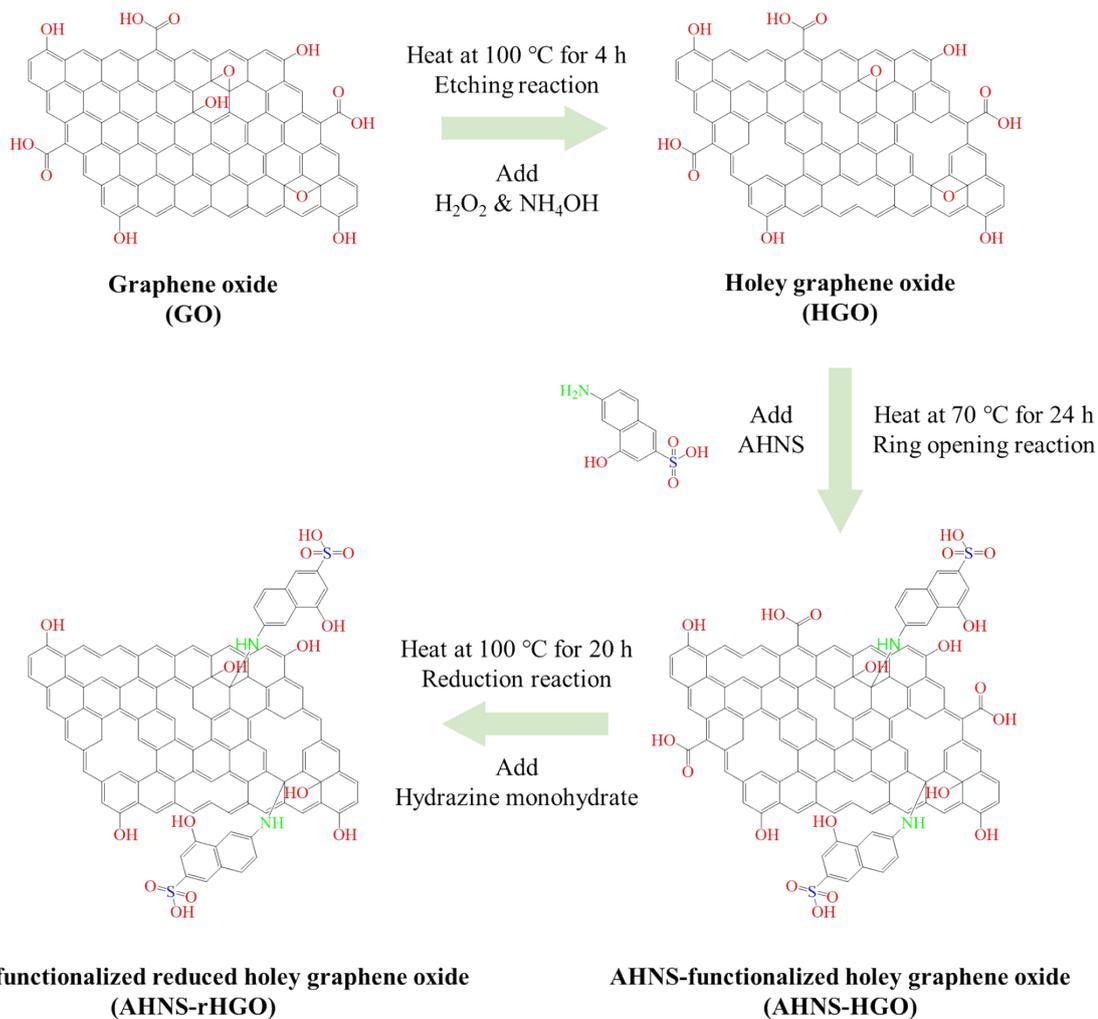
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31 **Fig. S1** Schematic for the synthesis of AHNS-rHGO.

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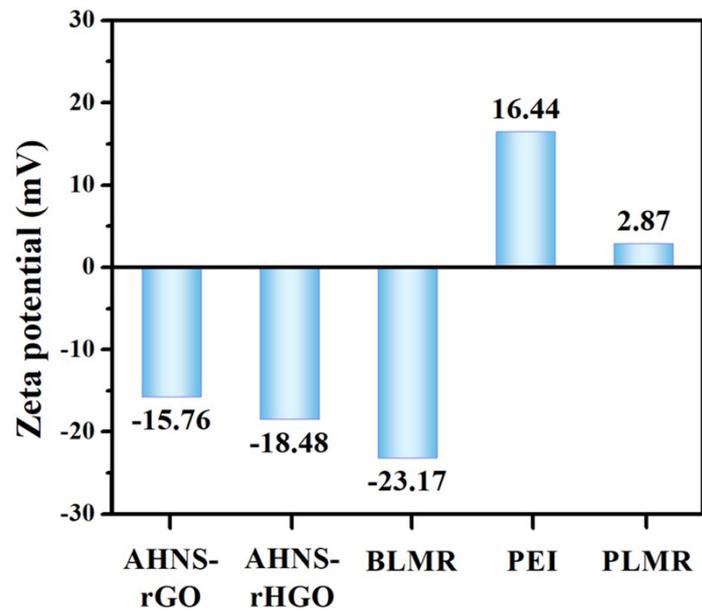
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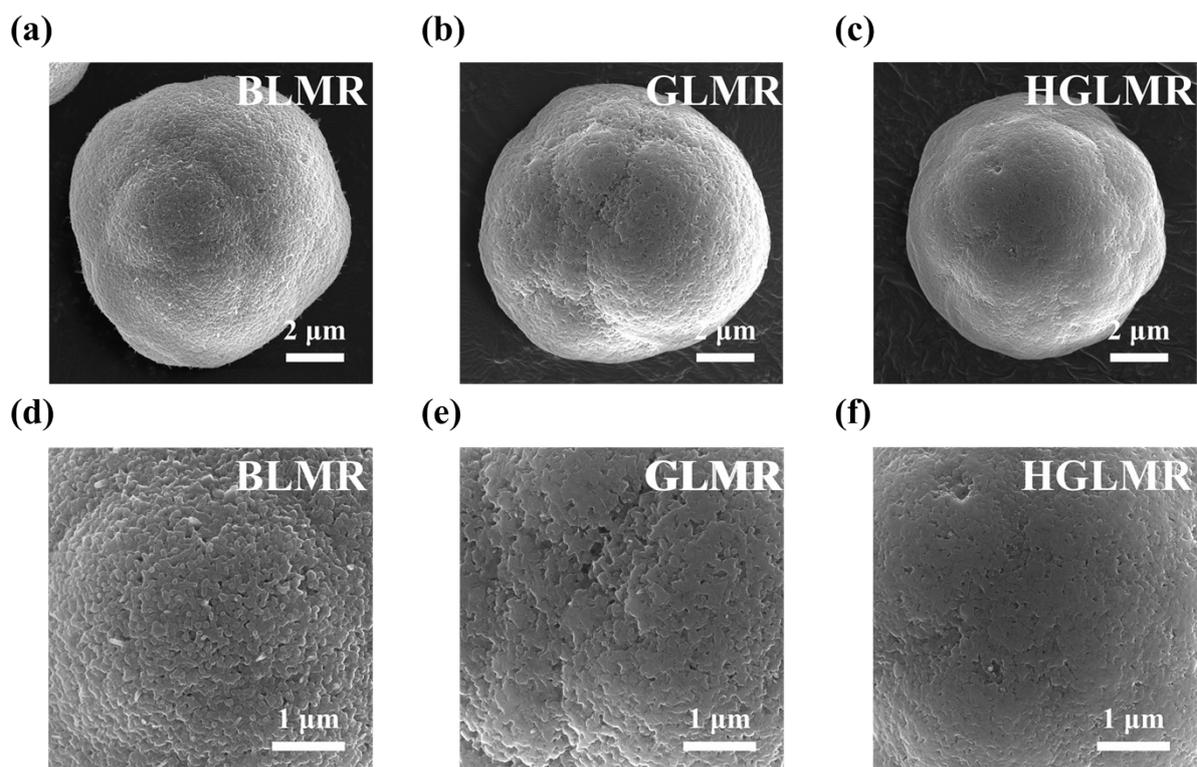
40 **Fig. S2** Zeta potentials of the samples used in the coating process.

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46 **Fig. S3** FE-SEM images of (a, d) BLMR, (b, e) GLMR, and (c, f) HGLMR taken at different
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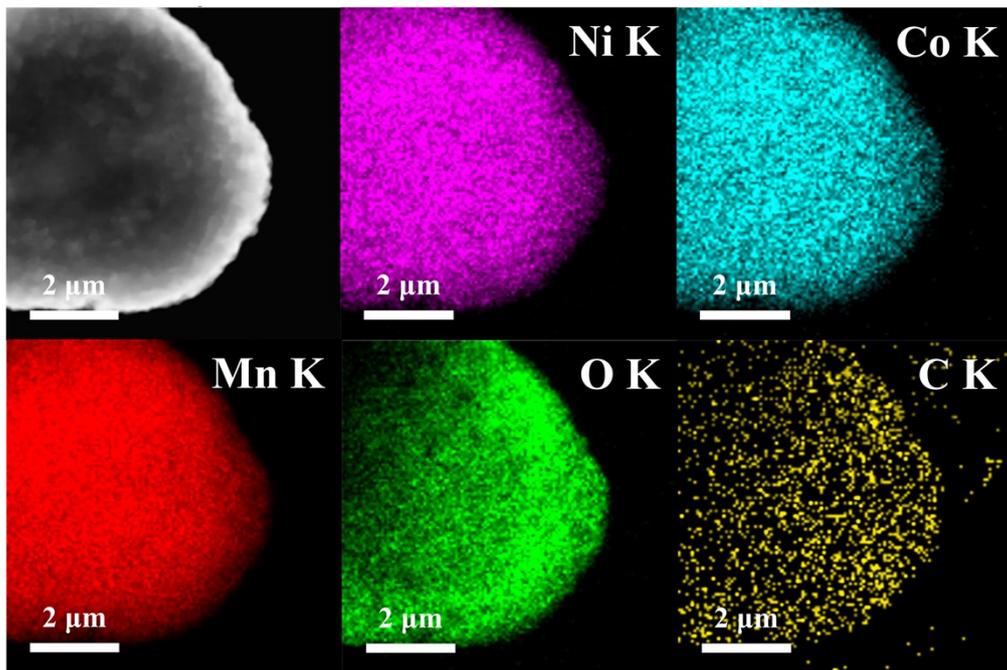
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60 **Fig. S4** TEM-EDS mapping images of HGLMR.

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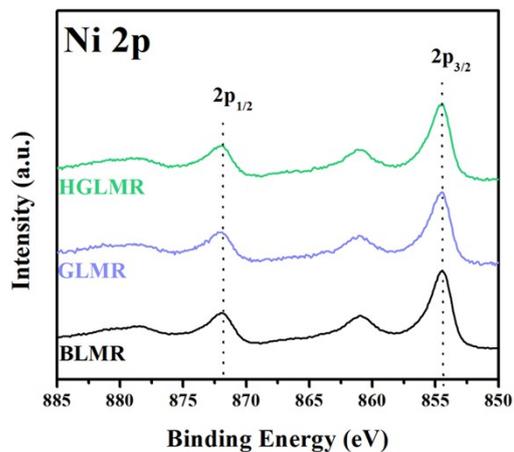
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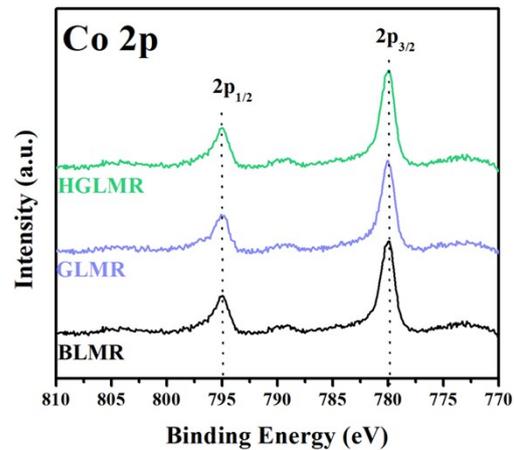
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71 **Fig. S5** (a) Ni 2p and (b) Co 2p XPS spectra of BLMR, GLMR, and HGLMR.

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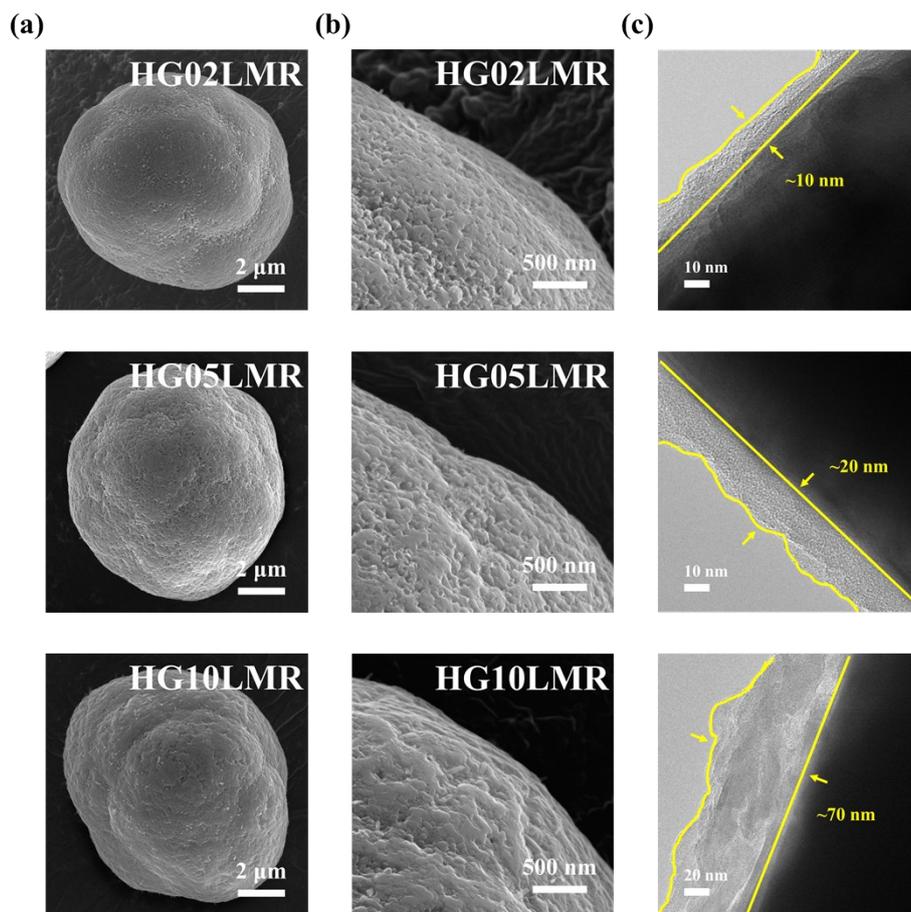
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87 **Fig. S6** (a–c) FE-SEM and HR-TEM images of HG02LMR, HG05LMR, and HG10LMR.

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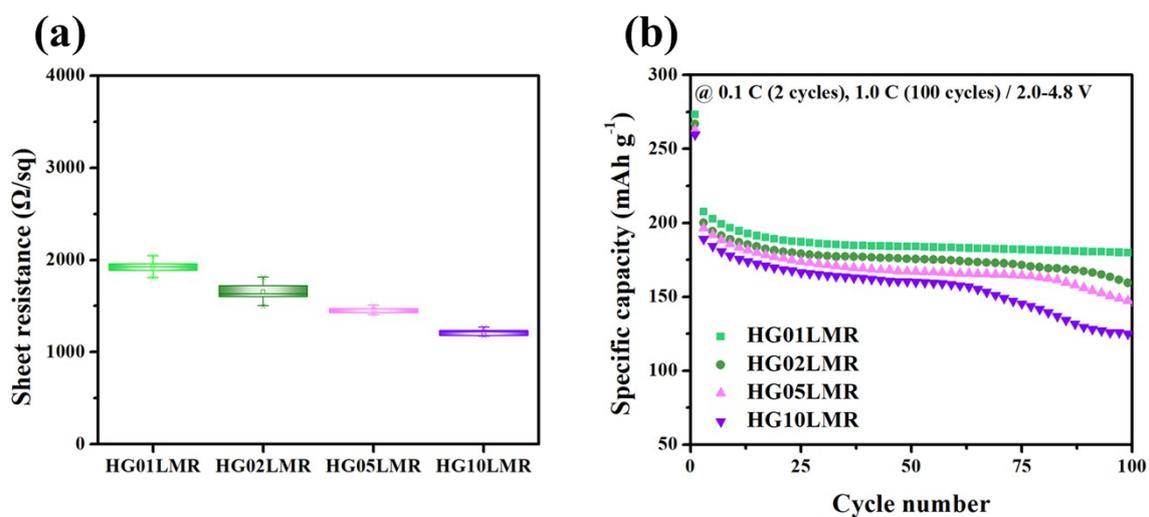
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95 **Fig. S7** (a) Sheet resistance values of the electrodes based on carbon contents. (b) Cycling

96 performance based on carbon contents at 1.0 C.

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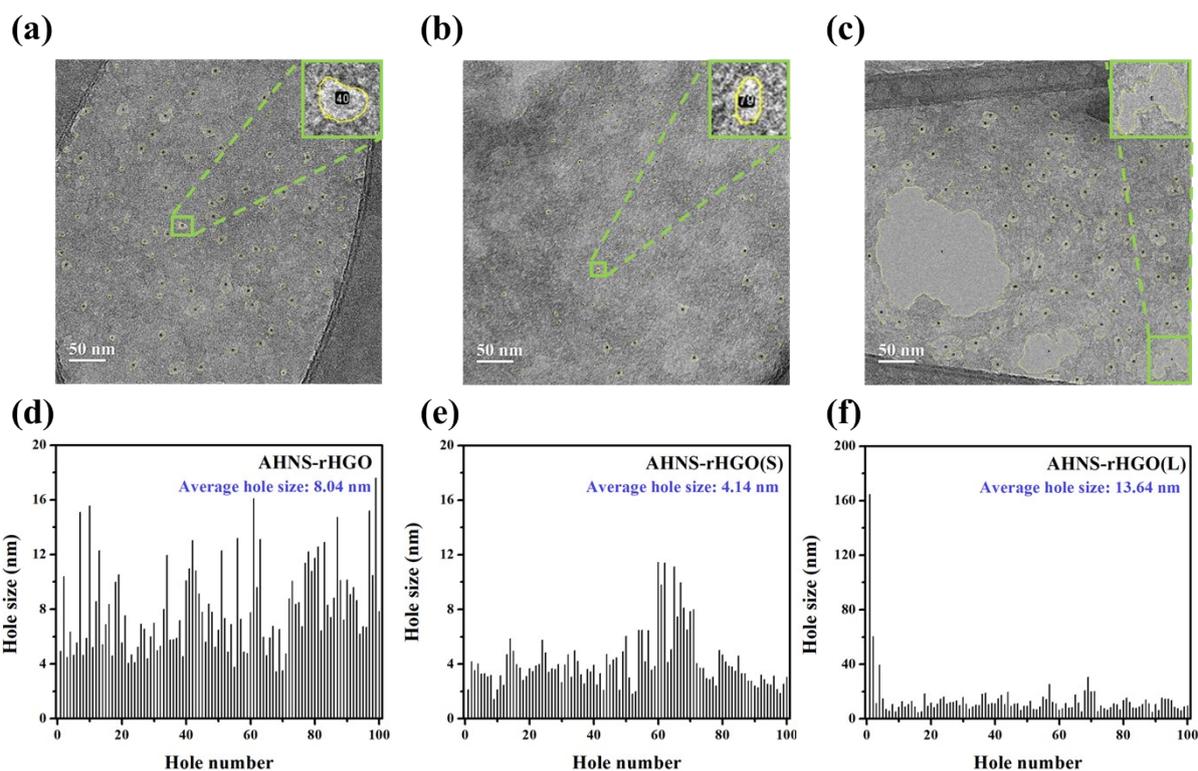
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 111 **Fig. S8** (a–c) HR-TEM images of AHNS-rHGO, AHNS-rHGO(S), and AHNS-rHGO(L). (d–f)
 112 Histograms of the pore size distribution for 100 pores in each HR-TEM image given above.

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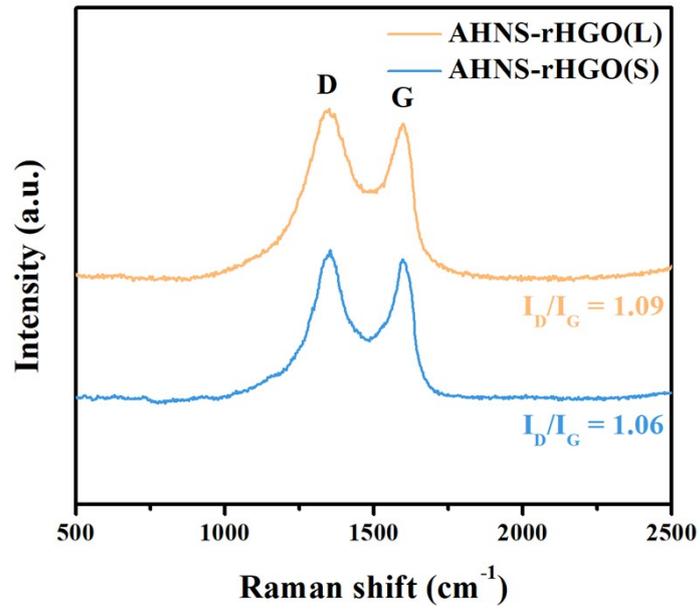
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125 **Fig. S9** Raman spectra of AHNS-rHGO(S) and AHNS-rHGO(L).

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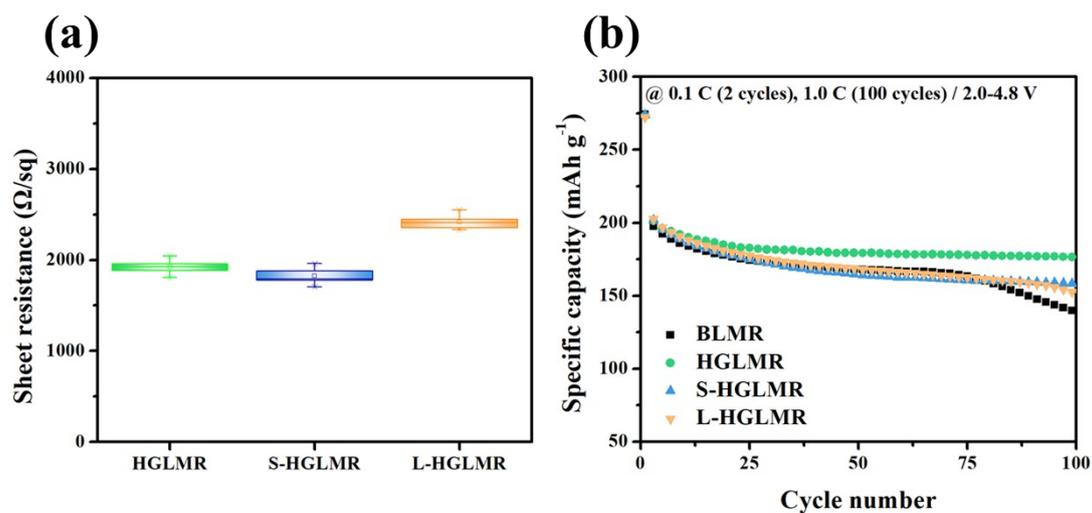
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138 **Fig. S10** (a) Sheet resistances of the HGLMR, S-HGLMR, and L-HGLMR electrodes. (b) cycling

139 performance of the BLMR, HGLMR, S-HGLMR, and L-HGLMR electrodes.

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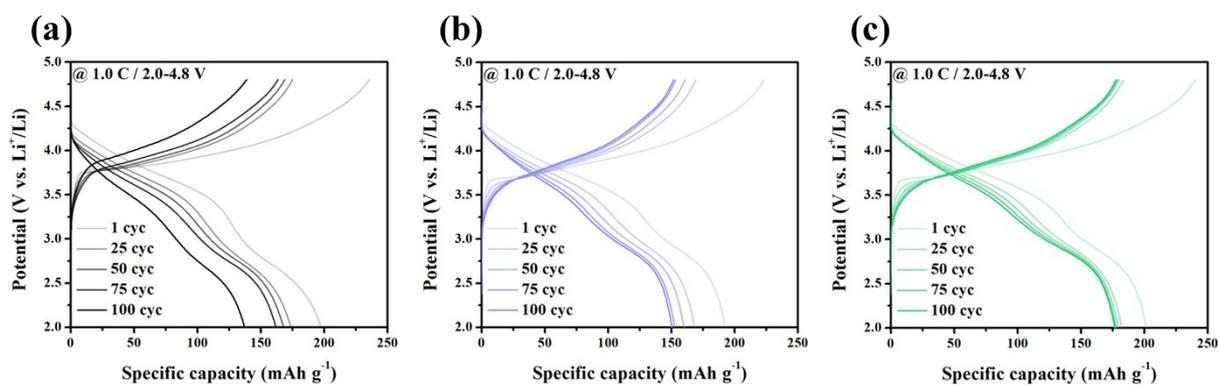
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154 **Fig. S11** (a–c) GCD curves illustrating the specific capacities of BLMR, GLMR, and HGLMR for
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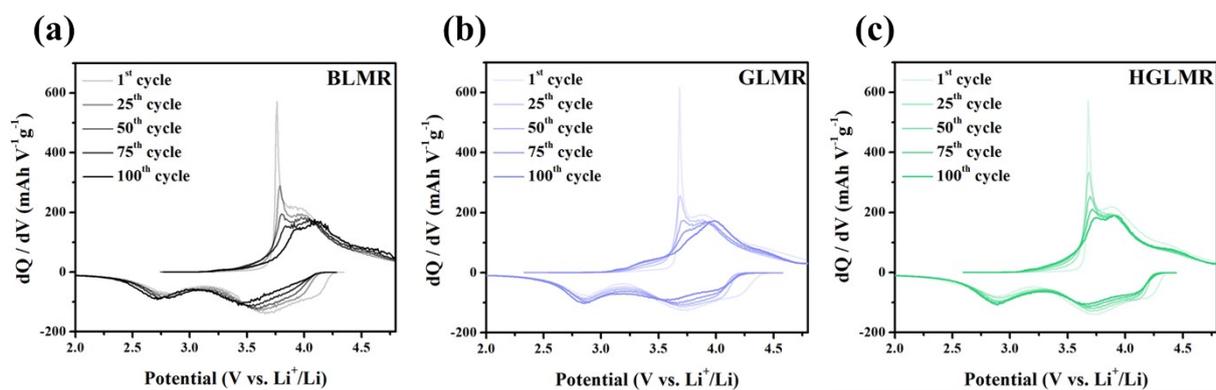
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172 **Fig. S12** (a–c) dQ/dV curves for BLMR, GLMR, and HGLMR for each cycle.

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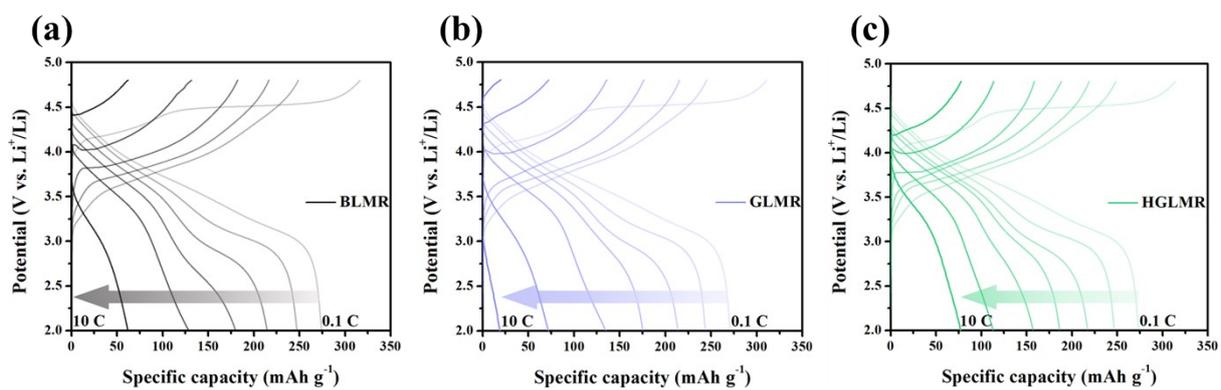
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189 **Fig. S13** (a–c) GCD curves of the BLMR, GLMR, and HGLMR electrodes at various C-rates.

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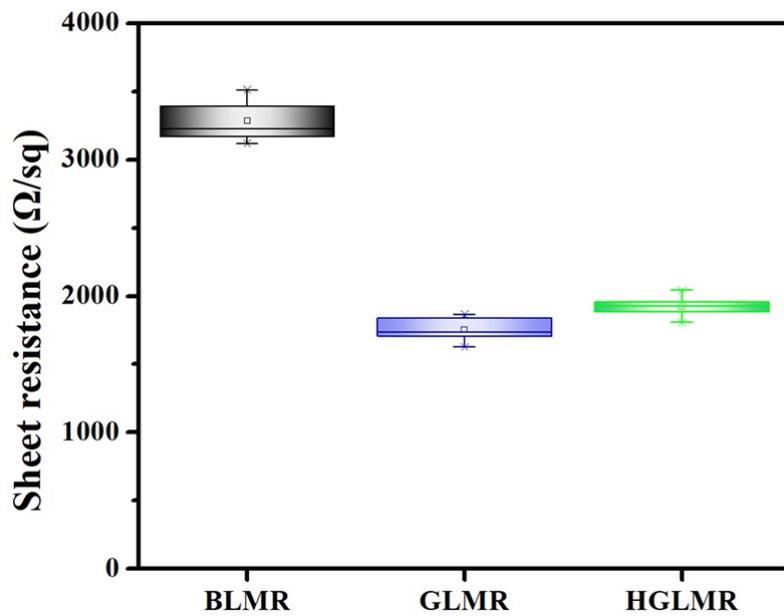
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206 **Fig. S14** Sheet resistance values of the BLMR, GLMR, and HGLMR electrodes.

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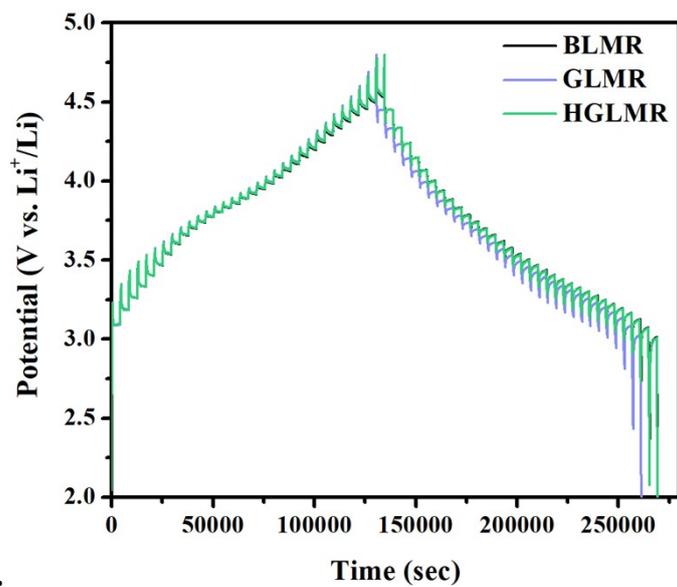
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219 **Fig. S15** GITT curves of the BLMR, GLMR, and HGLMR electrodes.

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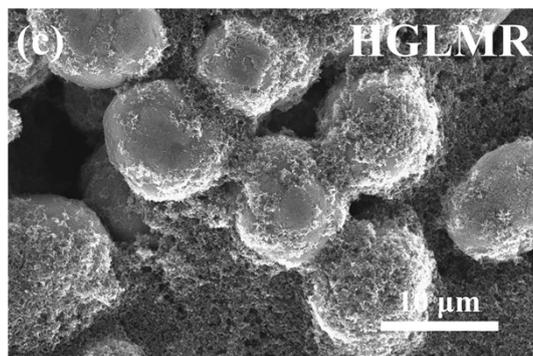
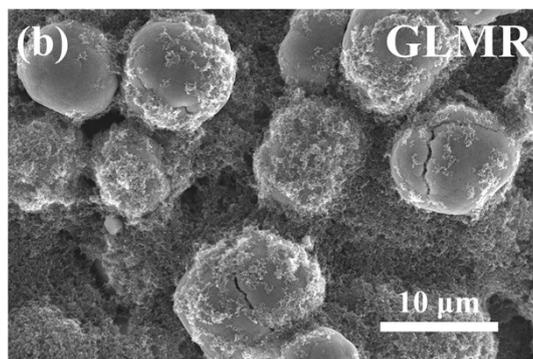
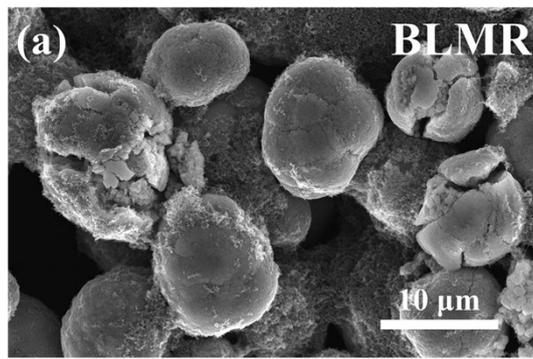
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229 **Fig. S16** Top-view FE-SEM images of the BLMR, GLMR, and HGLMR electrodes after 100 cycles.

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233 **Table S1** ICP-OES analysis of the BLMR, GLMR, and HGLMR particles.

Sample	Analysis results (weight %)			
	Li	Ni	Co	Mn
BLMR	10.05	12.57	6.33	40.07
GLMR	10.04	12.24	6.16	39.06
HGLMR	9.97	12.05	6.06	38.83

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247 **Table S2** BET analysis of the BLMR, GLMR, and HGLMR electrodes.

Sample	Specific surface area (m ² g ⁻¹)	Total pore volume (cm ³ g ⁻¹)	Average pore size (nm)
BLMR	3.947	0.005239	34.137
GLMR	4.2701	0.006089	37.649
HGLMR	4.3092	0.006232	38.679

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261 **Table S3** Tap density analysis results for BLMR, GLMR, and HGLMR.

Sample	Tap density (g cm ⁻³)
BLMR	2.859
GLMR	2.808
HGLMR	2.804

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279 **Table S4** Areal ratio variation in the XPS depth profiles of F 1s based on etching times from 0
 280 to 100 s.

Composition (%)	BLMR			GLMR			HGLMR		
	Etching time (s)								
	0	50	100	0	50	100	0	50	100
LiF	41.3	63.4	65.6	11.3	44.0	49.5	9.8	33.9	40.6
Li _x PO _y F _z	19.1	15.6	17.2	-	1.4	14.8	-	3.8	17.6
CF ₂	39.6	20.8	17.2	88.7	54.6	35.7	90.2	62.3	41.8

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292 **Table S5** Comparison of electrochemical performances of the HGLMR electrode with
 293 previously reported surface-coated LMR electrodes.

Cathode material	Coating material	Mass loading (mg cm ⁻²)	Voltage window (V)	Rate capability	Cycle stability	Ref
Li _{1.2} Ni _{0.13} Co _{0.13} Mn _{0.54} O ₂	LiErO ₂	2.0–3.0	2.0–4.7	162.1 mAh g ⁻¹ at 5.0 C	80% after 100 cycles at 1.0 C	[1]
Li _{1.2} Ni _{0.13} Co _{0.13} Mn _{0.54} O ₂	Li ₂ ZrO ₃	3	2.0–4.8	155.6 mAh g ⁻¹ at 5.0 C	89% after 150 cycles at 1.0 C	[2]
Li _{1.2} Ni _{0.13} Co _{0.13} Mn _{0.54} O ₂	KMnO ₄	2.0	2.0–4.8	91.6 mAh g ⁻¹ at 10.0 C	88% after 170 cycles at 0.2 C	[3]
Li _{1.2} Ni _{0.13} Co _{0.13} Mn _{0.54} O ₂	(NH ₄) ₂ SiF ₆	2.0	2.0–4.8	76.0 mAh g ⁻¹ at 10.0 C	90.4% after 100 cycles at 1.0 C	[4]
Li _{1.2} Ni _{0.13} Co _{0.13} Mn _{0.54} O ₂	In ₂ O ₃	1.3	2.0–4.8	109.1 mAh g ⁻¹ at 5.0 C	80% after 200 cycles at 1.0 C	[5]
Li _{1.2} Ni _{0.13} Co _{0.13} Mn _{0.54} O ₂	PO ₄ ³⁻ -doped layer@spinel @rGO	1.35	2.0–4.8	82.3 mAh g ⁻¹ at 10.0 C	86.5% after 200 cycles at 1.0 C	[6]
Li _{1.2} Ni _{0.13} Co _{0.13} Mn _{0.54} O ₂	Graphene quantum dot	1.2	2.0–4.8	113.2 mAh g ⁻¹ at 5.0 C	86.5% after 150 cycles at 1.0 C	[7]
Li _{1.2} Ni _{0.2} Mn _{0.6} O ₂	N-doped Graphene	3.5	2.0–4.8	≈ 100.0 mAh g ⁻¹ at 8.0 C	86% after 200 cycles at 0.2 C	[8]
Li _{1.2} Ni _{0.13} Co _{0.13} Mn _{0.54} O ₂	Li _{1.4} Y _{0.4} Ti _{1.6} PO ₄	2.65	2.0–4.8	78.3 mAh g ⁻¹ at 10.0 C	83.2% after 100 cycles at 1.0 C	[9]
Li_{1.2}Ni_{0.16}Co_{0.08}Mn_{0.56}O₂	PEI/reduced holey graphene oxide	4	2.0–4.8	78.0 mAh g⁻¹ at 10.0 C	87.8% after 100 cycles at 1.0 C	This work

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