

Supporting Information

Achieving long-lived photogenerated holes in ZnIn_2S_4 loaded with CoO_x clusters for enhanced photocatalytic pure water splitting

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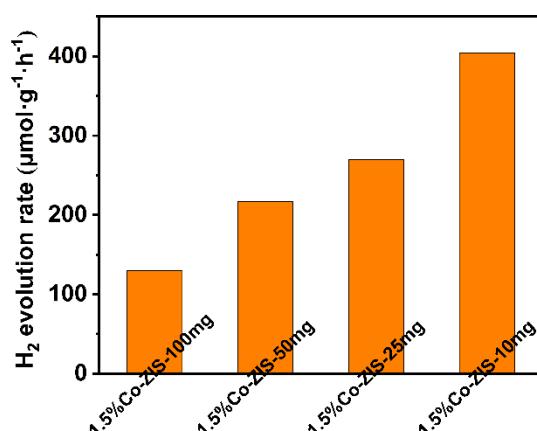


Fig. S1. The H_2 evolution rate of 1.5%Co-ZIS using different usage.

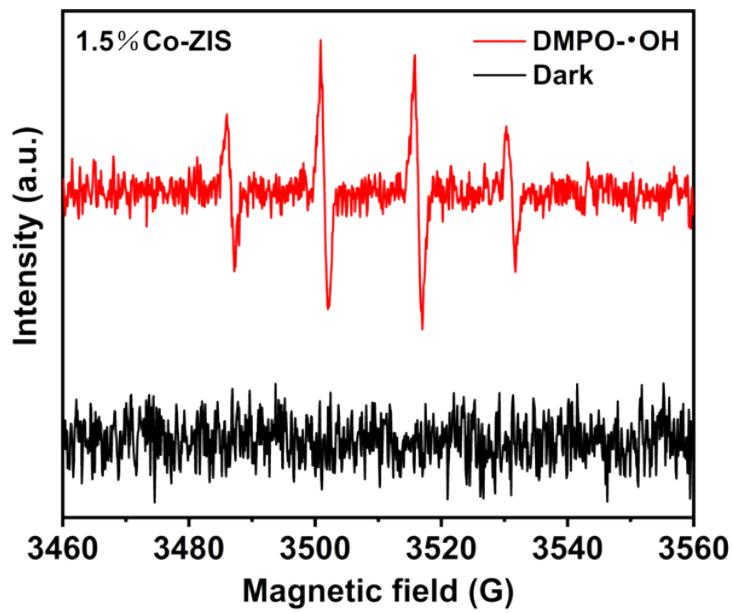


Fig. S2. EPR spectra of 1.5%Co-ZIS.

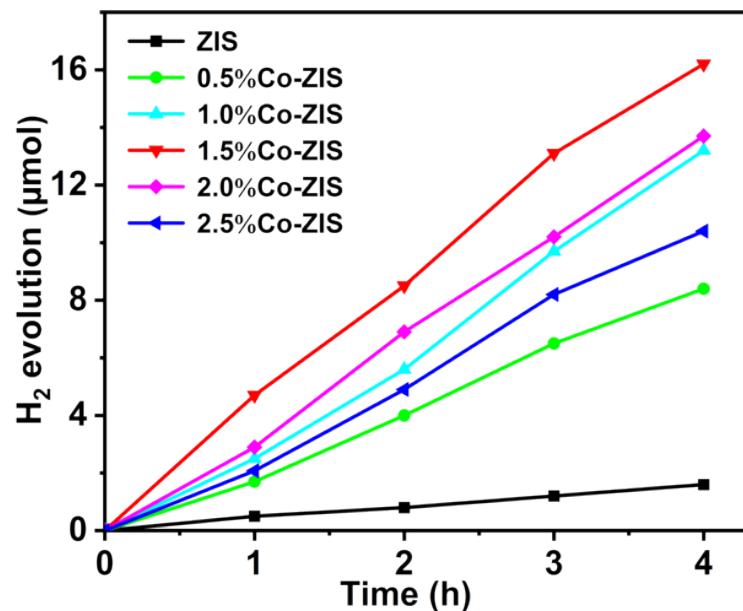


Fig. S3. Time courses of H_2 evolution for ZIS and Co-ZIS composites with different contents of Co.

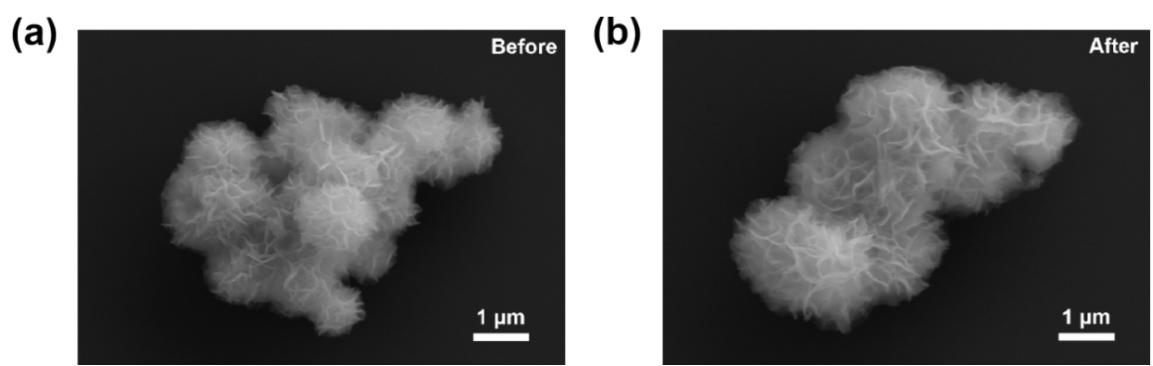


Fig. S4. SEM images of 1.5%Co-ZIS before reaction (a) and after reaction (b).

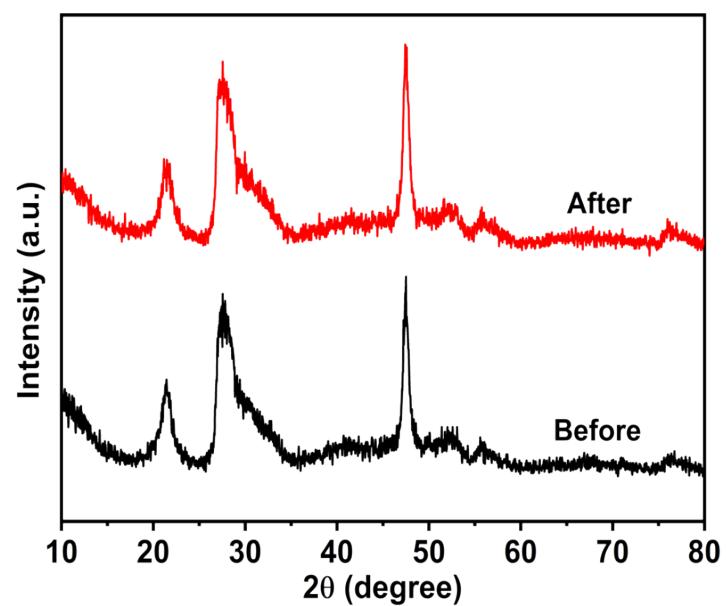


Fig. S5. XRD patterns of 1.5%Co-ZIS before and after the reaction.

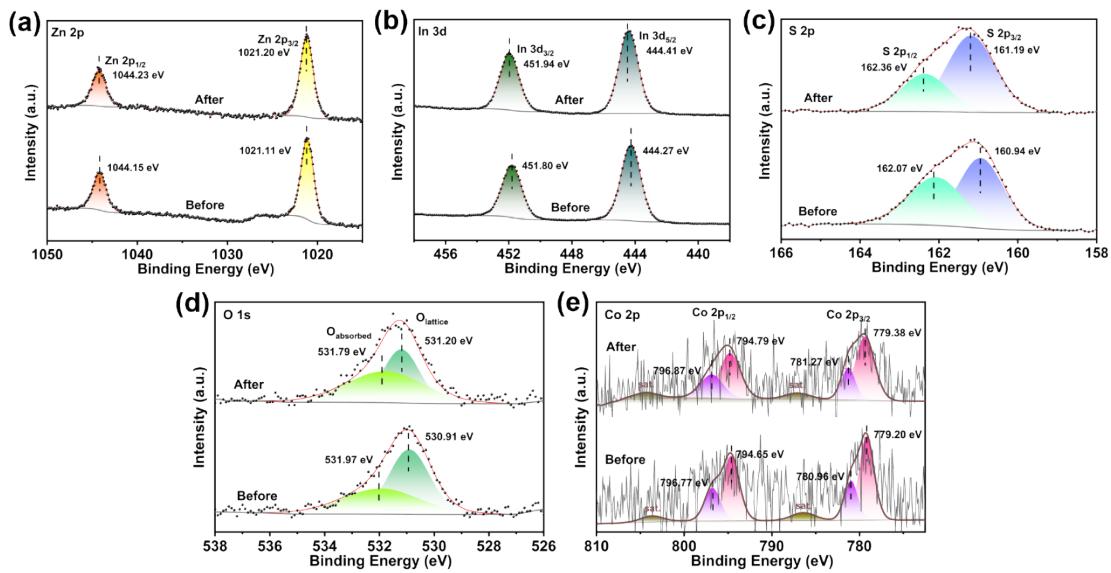


Fig. S6. XPS spectra of Zn 2p (a), In 3d (b), S 2p (c), O 1s (d) and Co 2p (e) for 1.5%Co-ZIS before and after the reaction.

Table S1. Summary of reports on photocatalytic overall water splitting of some common catalysts.

Photocatalyst	Light source	H ₂ (μmol·g ⁻¹ ·h ⁻¹)	O ₂ /H ₂ O ₂ (μmol·g ⁻¹ ·h ⁻¹)	AQE	Refs.
p-Co ₃ O ₄ /n-TiO ₂	300 W Xe-lamp ($\lambda \geq 420$ nm)	8.16	4	/	1
Cu/TiO ₂	300 W Xe-lamp ($\lambda \geq 420$ nm)	35.9	16.8	/	2
P-g-C ₃ N ₄ -Co _x P	300 W Xe-lamp ($\lambda \geq 420$ nm)	305.2	274.4 (H ₂ O ₂)	3.6% (425 nm)	3
S _v -CdS	350 W Xe-lamp ($\lambda \geq 420$ nm)	363.8	181.9	/	4
Pt/CdS@Al ₂ O ₃	300 W Xe-lamp ($\lambda \geq 420$ nm)	62.1	/	0.11% (430 nm)	5
RuO ₂ /CdS/MoS ₂	300 W Xe-lamp ($\lambda \geq 420$ nm)	52	11	0.32% (365 nm)	6
Ag-ZnIn ₂ S ₄	300 W Xe-lamp ($\lambda \geq 420$ nm)	56.6	29.1	0.70% (405 nm)	7
BiVO ₄ @ZnIn ₂ S ₄ /Ti ₃ C ₂	300 W Xe-lamp ($\lambda \geq 400$ nm)	102.7	50.8	2.40% (410 nm)	8
CdS@ZnIn ₂ S	300 W Xe-lamp ($\lambda \geq 400$ nm)	540.3	604.8 (H ₂ O ₂)	1.63% (400 nm)	9
C-N-g-C ₃ N ₄	300 W Xe lamp	98	84 (H ₂ O ₂)	0.86% (420 nm)	10

		(700nm $\geq\lambda\geq$ 420nm)			
PCNNi	300 W Xe-lamp ($\lambda\geq$ 400 nm)	26.2	24 (H ₂ O ₂)	1.12% (420 nm)	¹¹
ZnIn ₂ S ₄ /WO ₃	300 W Xe-lamp (AM 1.5G)	668.6	328.6	3.18% (380 nm)	¹²
1.5%Co-ZIS	300 W Xe-lamp ($\lambda\geq$400 nm)	404.1	371.9 (H₂O₂)	1.74% (400 nm)	This work

Table S2. Kinetic parameters of TA decay for ZIS and 1.5%Co-ZIS

Sample Name	τ_1 (ps)	τ_2 (ps)	A ₁	A ₂	χ^2
ZIS	10.15	558.50	0.49	0.51	0.99
1.5Co%-ZIS	23.13	911.60	0.53	0.47	0.99

The life time was used calculated using the following equation:

$$\tau_{ave} = \frac{B_1 \times (\tau_1)^2 + B_2 \times (\tau_2)^2}{B_1 \times \tau_1 + B_2 \times \tau_2}$$

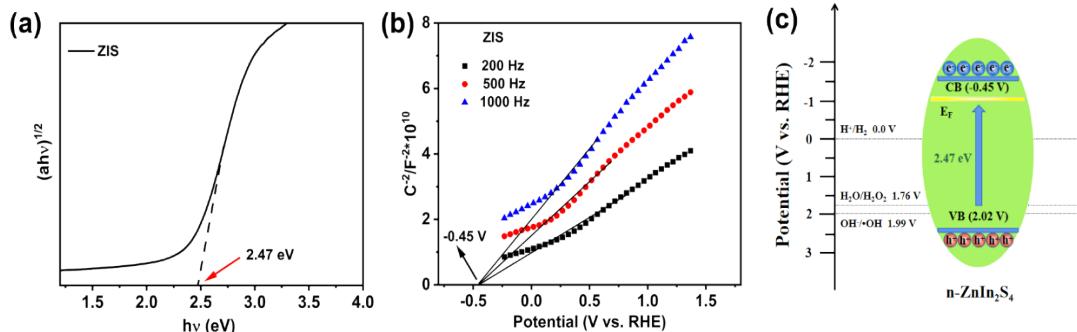


Fig. S7. (a) The plots of $(\alpha h v)^{1/2}$ vs. $h v$ for ZIS. (b) Mott-Schottky curves of ZIS. (c) Schematic diagram of ZIS band structure.

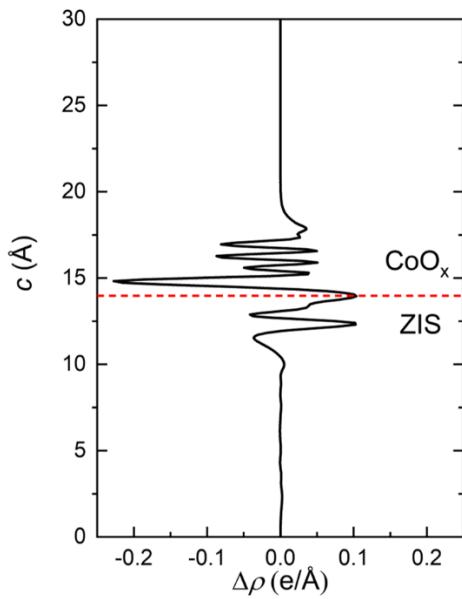


Fig. S8. The planar-averaged electron density difference along the c axis of ZIS/CoO_x interface.

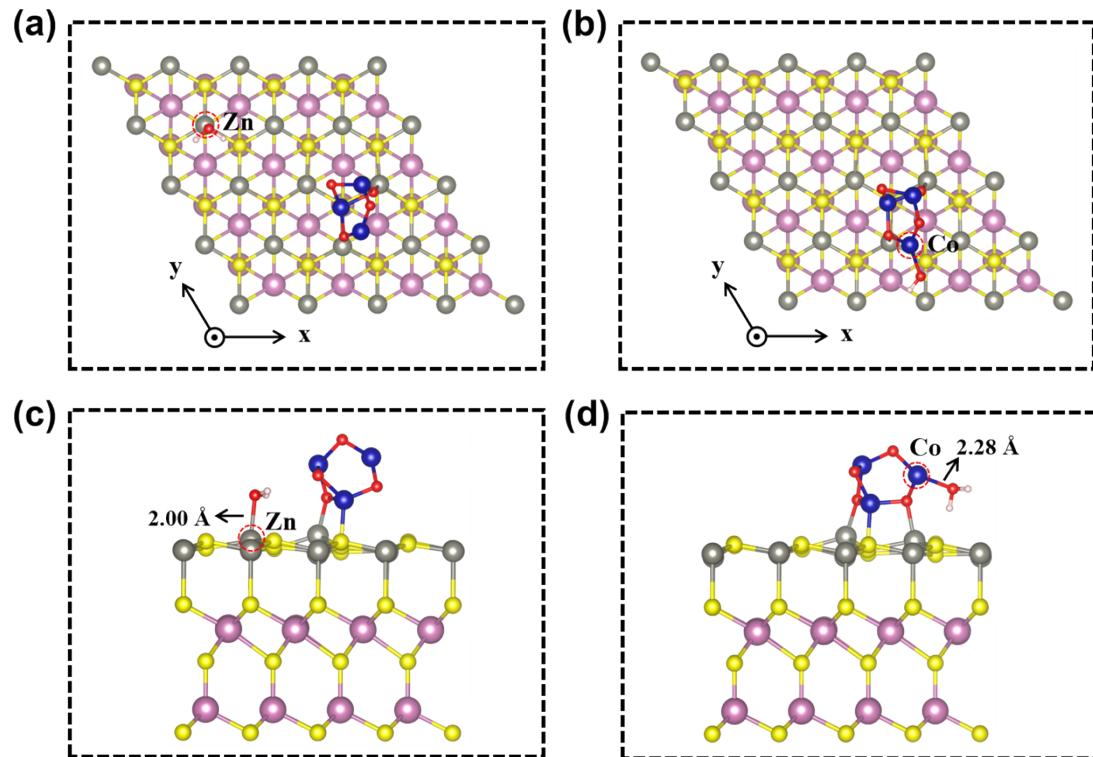


Fig. S9. Top and (b) side view of optimal structure for adsorption H₂O at different sites Zn (a), (c) and Co (b), (d). Gray, purple, yellow, dark blue, red and bright white balls represent Zn, In, S, Co, O and H atoms, respectively.

References

1. Q. Zhang, Z. Hai, A. Jian, H. Xu, C. Xue and S. Sang, *Nanomaterials*, 2016, **6**, 138.
2. D. Wei, Y. Tan, Y. Wang, T. Kong, S. Shen and S. S. Mao, *Sci. Bull.*, 2020, **65**, 1389-1395.
3. F. Xue, Y. Si, C. Cheng, W. Fu, X. Chen, S. Shen, L. Wang and M. Liu, *Nano Energy*, 2022, **103**, 107799.
4. J. He, L. Hu, C. Shao, S. Jiang, C. Sun and S. Song, *ACS Nano*, 2021, **15**, 18006-18013.
5. X. Ning, W. Zhen, Y. Wu and G. Lu, *Appl. Catal. B: Environ.*, 2018, **226**, 373-383.
6. B. Qiu, L. Cai, N. Zhang, X. Tao and Y. Chai, *Adv. Sci.*, 2020, **7**, 1903568.
7. R. Pan, M. Hu, J. Liu, D. Li, X. Wan, H. Wang, Y. Li, X. Zhang, X. Wang, J. Jiang and J. Zhang, *Nano Lett.*, 2021, **21**, 6228-6236.
8. X. Du, T. Zhao, Z. Xiu, Z. Xing, Z. Li, K. Pan, S. Yang and W. Zhou, *Appl. Mater. Today*, 2020, **20**, 100719.
9. E. Zhang, Q. Zhu, J. Huang, J. Liu, G. Tan, C. Sun, T. Li, S. Liu, Y. Li, H. Wang, X. Wan, Z. Wen, F. Fan, J. Zhang and K. Ariga, *Appl. Catal. B: Environ.*, 2021, **293**, 120213.
10. Y. Fu, C. a. Liu, M. Zhang, C. Zhu, H. Li, H. Wang, Y. Song, H. Huang, Y. Liu and Z. Kang, *Adv. Energy Mater.*, 2018, **8**, 1802525.
11. Y. Li, Y. Wang, C. Dong, Y. Huang, J. Chen, Z. Zhang, F. Meng, Q. Zhang, Y. Huangfu, D. Zhao, L. Gu and S. Shen, *Chem. Sci.*, 2021, **12**, 3633-3643.
12. Y. Wang, W. Huang, S. Guo, X. Xin, Y. Zhang, P. Guo, S. Tang and X. Li, *Adv. Energy Mater.*, 2021, **11**, 2102452.