

Supplementary Information

Bifunctional conjugated polymer photocatalysts for visible light water oxidation and CO₂ reduction: Function- and site-selective hybridisation of Ru(II) complex catalysts

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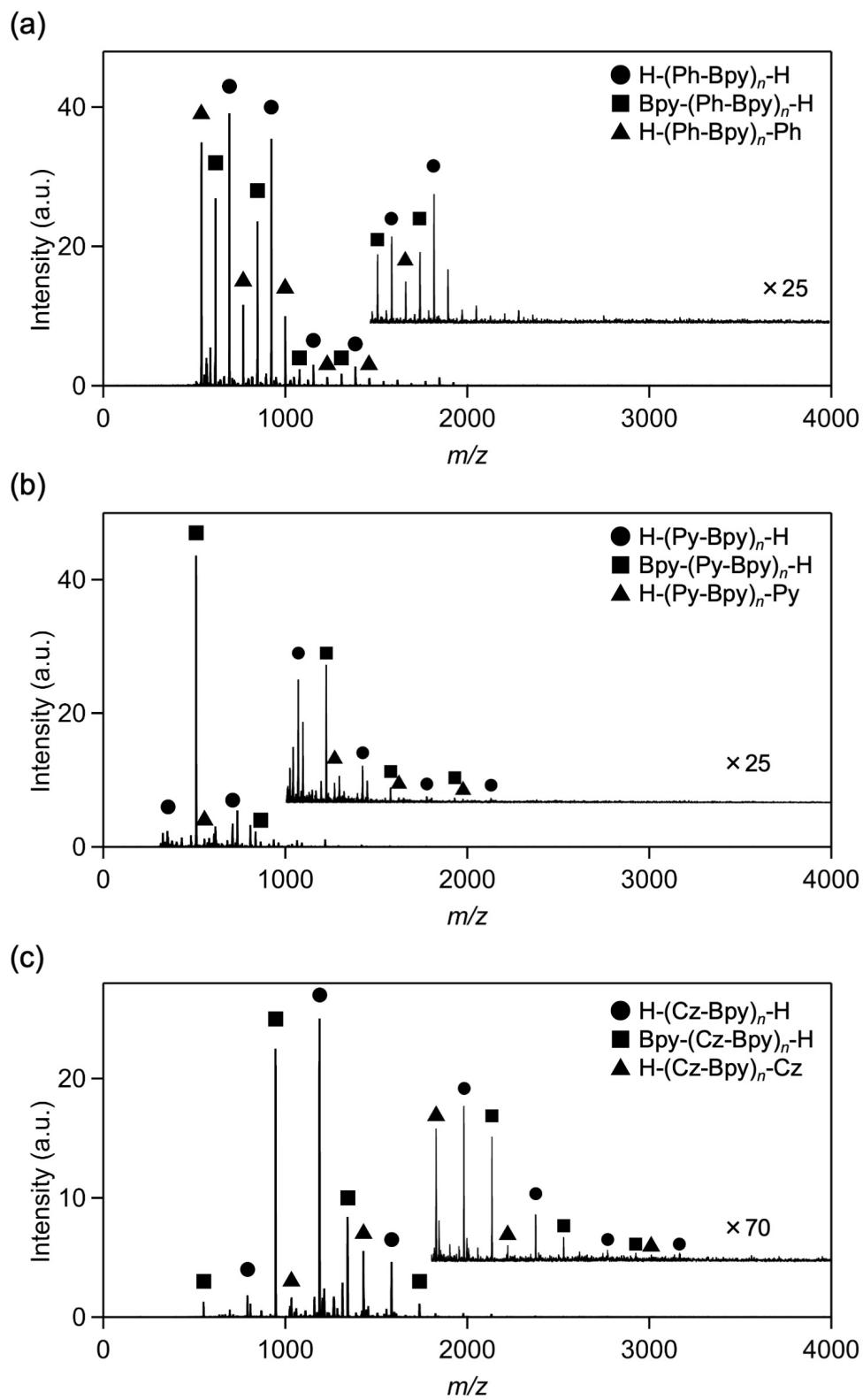


Fig. S1 MALDI-TOF-MS spectra of $[X\text{-bpy}]_n$ ($X =$ (a) Ph, (b) Py, or (c) Cz)

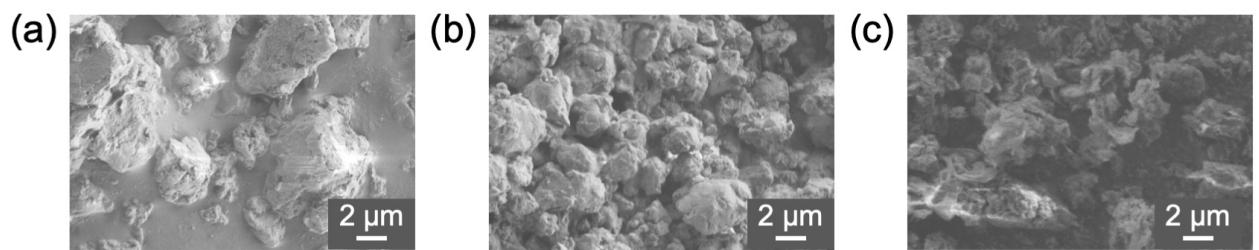


Fig. S2 SEM images of $[X\text{-}\text{bpy}]_n$ ($X = \text{(a) Ph, (b) Py, or (c) Cz}$).

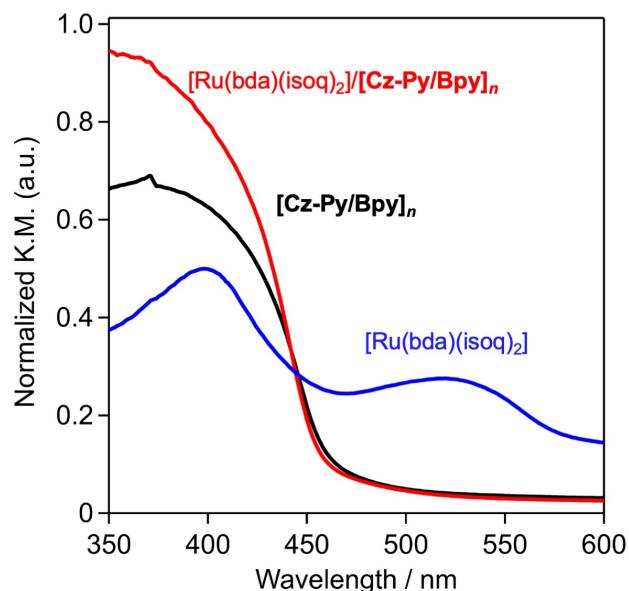


Fig. S3 UV-vis diffuse reflectance spectra of the polymer photocatalysts before (black line), after (red line) adsorption of $[\text{Ru}(\text{bda})(\text{isoq})_2]$, along with that of $[\text{Ru}(\text{bda})(\text{isoq})_2]$ (blue line).

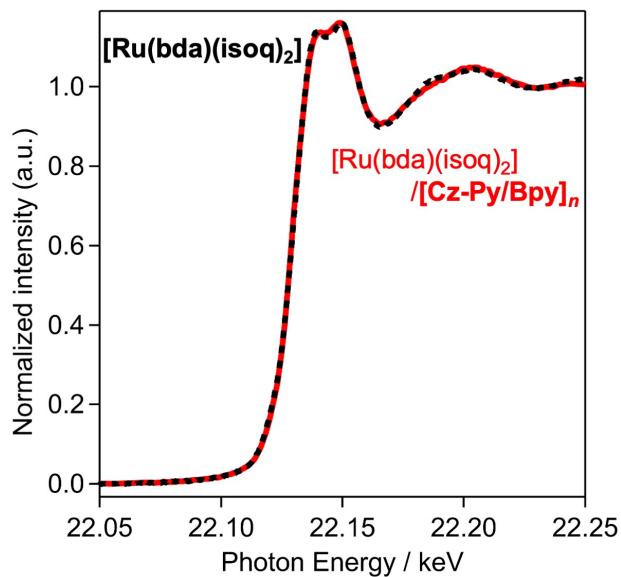


Fig. S4 Ru-K edge XANES spectra of $[\text{Ru}(\text{bda})(\text{isoq})_2]$ (black dotted line) and $[\text{Cz-Py/Bpy}]_n$ after adsorption of $[\text{Ru}(\text{bda})(\text{isoq})_2]$ (red line).

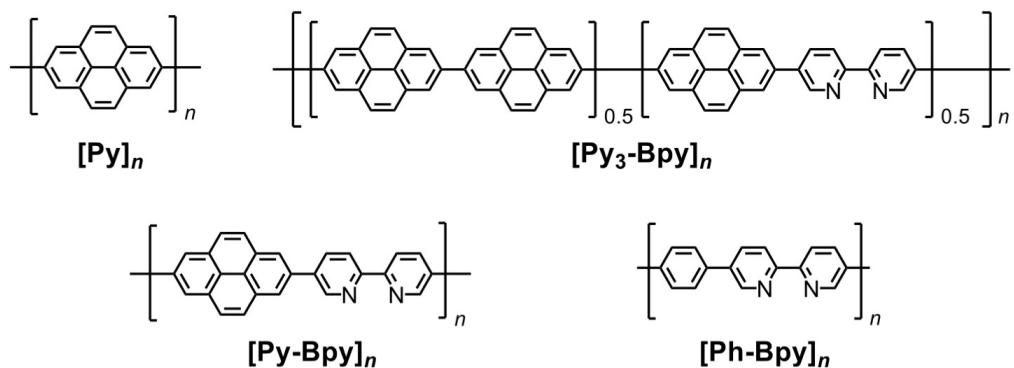


Fig. S5 Molecular structures of pyrene/bipyridine-based polymers with different pyrene contents.

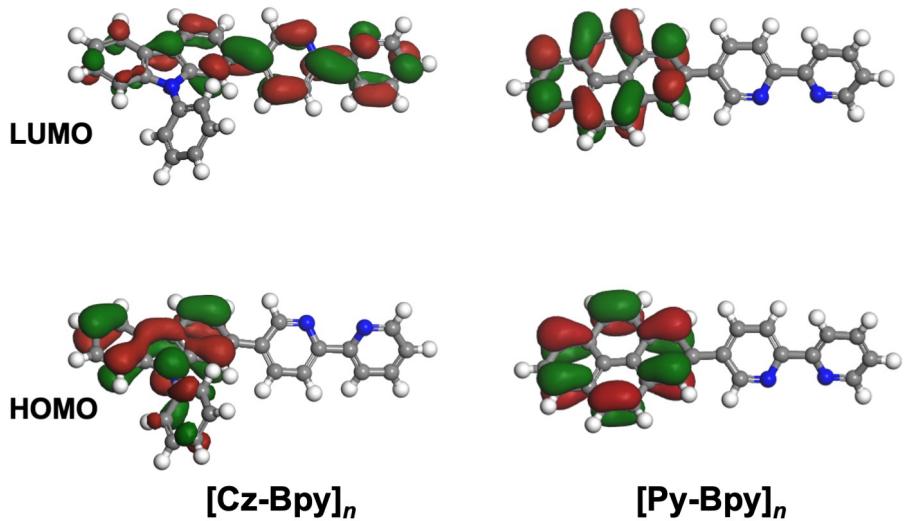


Fig. S6 Calculated HOMO (lower) and LUMO (upper) distributions of $X\text{-}bpy$ ($X = Cz$, and Py).

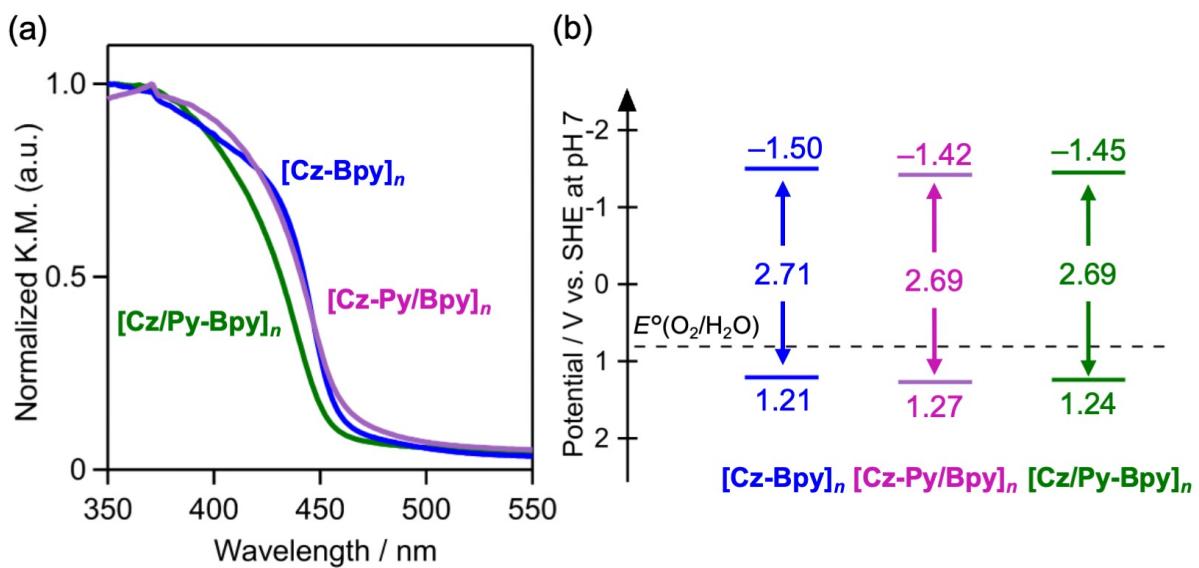


Fig. S7 (a) UV-vis diffuse reflectance spectra of the polymer photocatalysts and (b) HOMO-LUMO potentials of $[Cz\text{-}Bpy]_n$, $[Cz\text{-}Py\text{/}Bpy]_n$, and $[Cz\text{/}Py\text{-}Bpy]_n$.

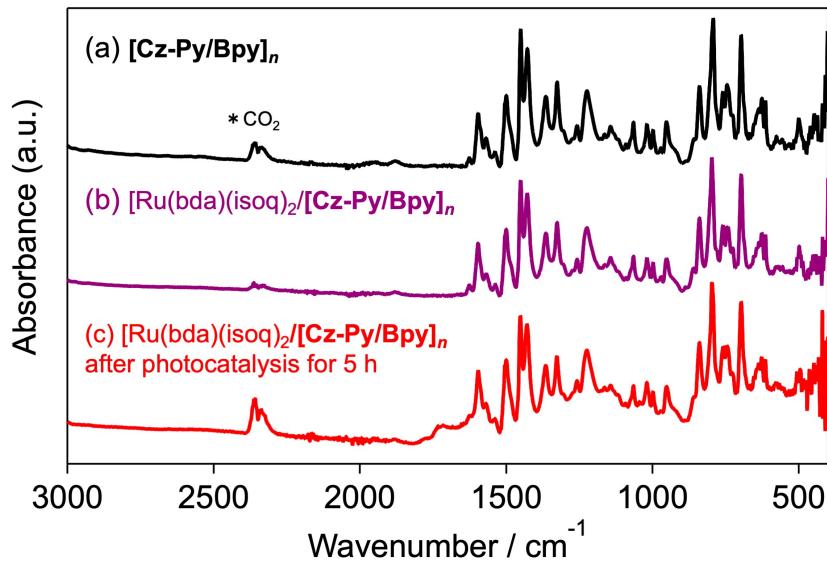


Fig. S8 ATR-IR spectra of (a) $[\text{Cz-Py/Bpy}]_n$ and (b,c) $[\text{Ru(bda)(isoq)}_2]/[\text{Cz-Py/Bpy}]_n$ before and after photocatalytic water oxidation for 5 h.

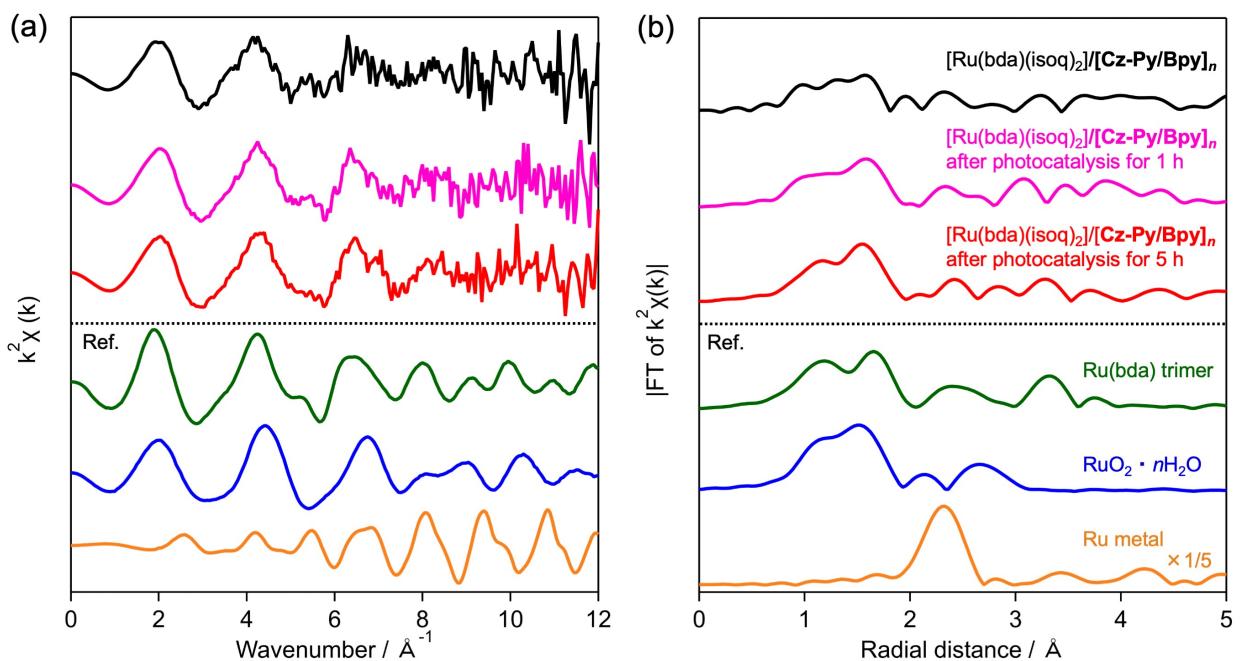


Fig. S9 (a) k^2 -weighted Ru-K edge EXAFS spectra, and (b) their fourier transforms of $[\text{Ru(bda)(isoq)}_2]/[\text{Cz-Py/Bpy}]_n$ before and after photocatalytic water oxidation shown in Fig. 4, along with Ru metal, $\text{RuO}_2 \cdot n\text{H}_2\text{O}$, and Ru(bda) trimer complex as references.

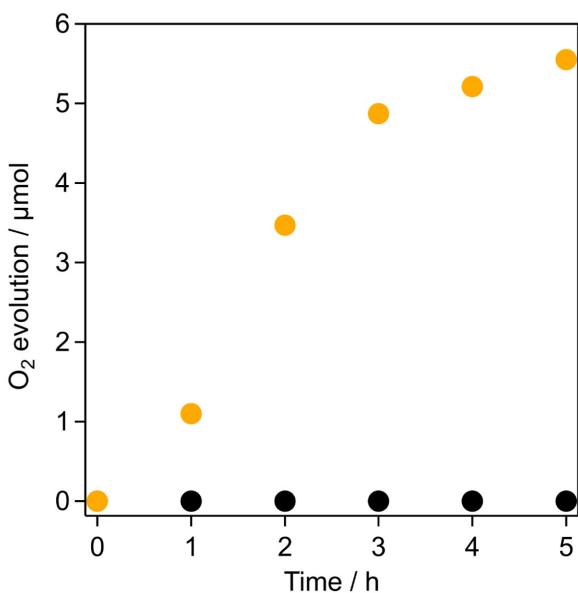


Fig. S10 Time courses of O₂ evolution over [Cz-Py/Bpy]_n (50 mg) in aqueous Na₂S₂O₈ solution (10 mM, 50 mL) containing 10 μM [Ru(bda)(isoq)₂] (orange) or 10 μM RuCl₃ (black).

Table. S1 Photocatalytic CO₂ reduction under various conditions.^a

Entry	Photocatalyst	Control	Product / μmol		
			Formate	CO	H ₂
1	[Cz-Py/BpyRu(CO)] _n	— ^a	2.9	N.D.	0.8
2	[Cz-Py/BpyRu(CO)] _n	Without light	N.D.	N.D.	N.D.
3	[Cz-Py/BpyRu(CO)] _n	Without CO ₂ ^b	N.D.	N.D.	0.1
4	[Cz-Py/BpyRu(CO)] _n	Without sodium ascorbate	Trace	N.D.	0.3
5	[Cz-Py/Bpy] _n	— ^a	N.D.	N.D.	0.1

^a Standard condition: 2 mg of photocatalyst powder in 10 mM sodium ascorbate aqueous solution dispersion was irradiated at $\lambda > 400$ nm for 10 h under CO₂ atmosphere. ^b Under Ar atmosphere.

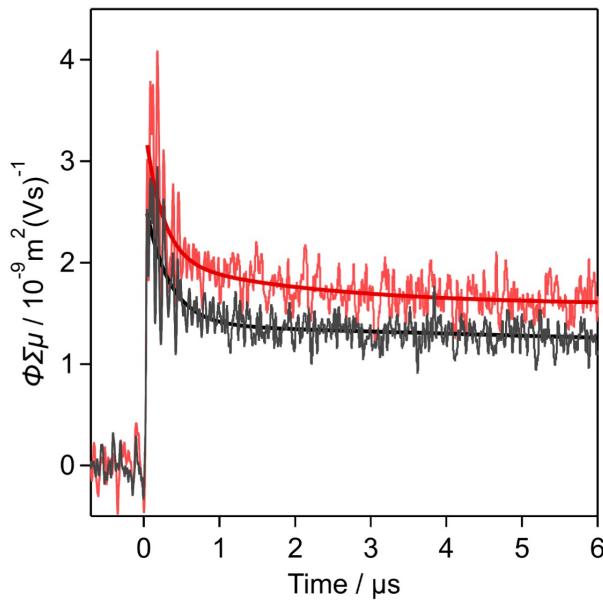


Fig. S11 Transient photoconductivities and their fittings using biexponential function ($A_1\exp(-k_1t) + A_2\exp(-k_2t)$) of $[\text{Cz-Py/Bpy}]_n$ (red) and $[\text{Cz-Bpy}]_n$ (black).

Table. S2 Overview of photocatalytic efficiency for water oxidation and CO₂ reduction using conjugated polymer-based photocatalysts in aqueous solution.

Abbreviation of Photocatalyst	Water oxidation			CO ₂ reduction			Ref
	catalyst	electron acceptor	AQE (%)	catalyst	electron donor	AQE (%)	
$[\text{Py-Cz/Bpy}]_n$	[Ru(bda) (isoq) ₂]	Na ₂ S ₂ O ₈	0.05 (430 nm)	(-bpy-)Ru (CO) ₂ Cl ₂	Sodium ascorbate	0.02 (430 nm)	This work
TAPT-Bpy	Co ²⁺	AgNO ₃	7.6 (420 nm)		Not demonstrated		[S1]
Bpy-CTF	Co ²⁺	AgNO ₃	0.56 (420 nm)		Not demonstrated		[S2]
P10	Co	AgNO ₃	–		Not demonstrated		[S3]
Bp-COF	Co ²⁺	AgNO ₃	0.46 (420 nm)		Not demonstrated		[S4]
CTHP	Co	AgNO ₃	–		Not demonstrated		[S5]
CTF-0-I	–	AgNO ₃	5.2 (420 nm)		Not demonstrated		[S6]
g-C ₄₀ N ₃	Co	AgNO ₃	–		Not demonstrated		[S7]
CTF-1	RuO _x	AgNO ₃	3.8 (420 nm)		Not demonstrated		[S8]
CTP-2	Co	AgNO ₃	–		Not demonstrated		[S9]
aza-CMP	Co	AgNO ₃	–		Not demonstrated		[S10]
g-C ₃ N ₄	Co(OH) ₂	AgNO ₃	–	RuP complex	EDTA·2Na	–	[S11] [S12]

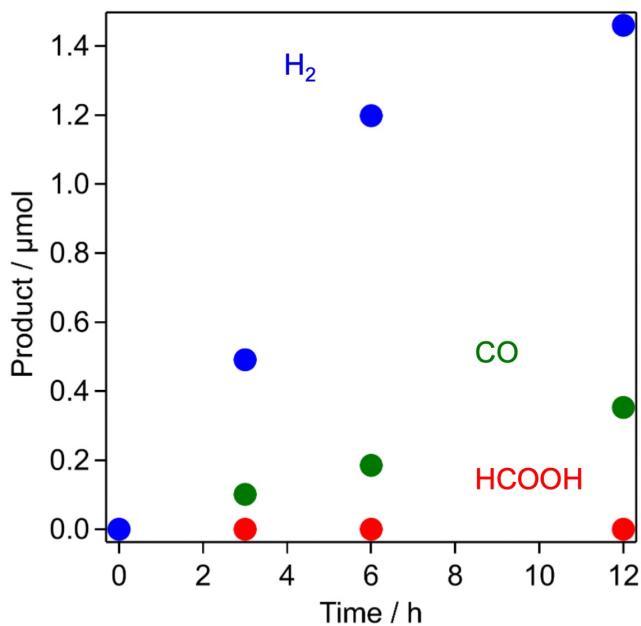


Fig. S12 Time courses of CO_2 reduction over $[\text{Ru}(\text{bda})(\text{isoq})_2]/[\text{Cz-Py/BpyRu}(\text{CO})]_n$ hybrid (2 mg) in water (2 mL) (i.e., without sacrificial reagent) upon visible-light irradiation ($\lambda > 400 \text{ nm}$).

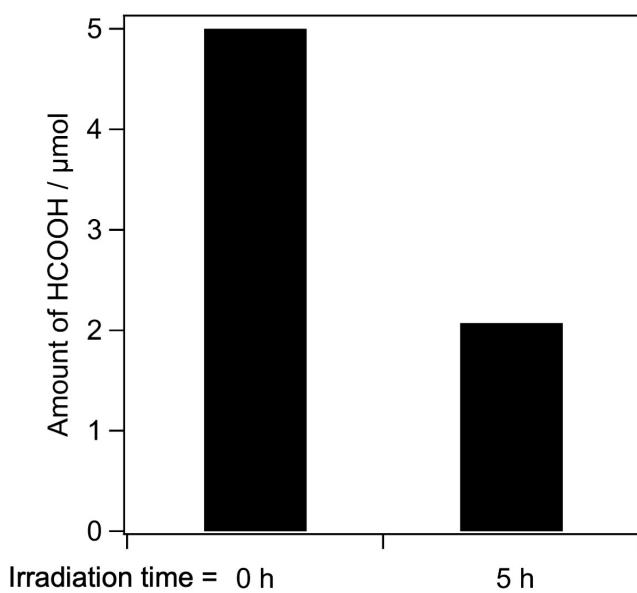


Fig. S13 The amount of HCOOH during water oxidation over $[\text{Ru}(\text{bda})(\text{isoq})_2]/[\text{Cz-Py/Bpy}]_n$ hybrid (5 mg) in 1 mM HCOOH aqueous solution (5 mL) upon visible-light irradiation ($\lambda = 430 \text{ nm}$).

References

- S1 H. Chen, A. M. Gardner, G. Lin, W. Zhao, X. Wang, M. Bahri, N. D. Browning, X. Xu, X. Li, *J. Phys. Chem. C*, **2023**, *127*, 14137-14145.
- S2 H. Chen, A. M. Gardner, G. Lin, W. Zhao, M. Bahri, N. D. Browning, R. S. Sprick, X. Li, X. Xu, A. I. Cooper, *Catal. Sci. Tech.*, **2022**, *12*, 5442-5452.
- S3 R. S. Sprick, Z. Chen, A. J. Cowan, Y. Bai, C. M. Aitchison, Y. Fang, M. A. Zwijnenburg, A. I. Cooper, X. Wang, *Angew. Chem. Int. Ed.*, **2020**, *132*, 18854-18859.
- S4 J. Chen, X. Tao, C. Li, Y. Ma, L. Tao, D. Zheng, J. Zhu, H. Li, R. Li, Q. Yang, *Appl. Catal. B: Environ.*, **2020**, *262*, 118271.
- S5 C. Wu, G. Yu, Y. Yin, Y. Wang, L. Chen, Q. Han, J. Tang, B. Wang, *Small*, **2020**, *16*, 2003162.
- S6 D. Kong, X. Han, J. Xie, Q. Ruan, C. D. Windle, S. Gadielli, K. Shen, Z. Bai, Z. Guo, J. Tang, *ACS Catal.*, **2019**, *9*, 7697-7707.
- S7 S. Bi, C. Yang, W. Zhang, J. Xu, L. Liu, D. Wu, X. Wang, Y. Han, Q. Liang, F. Zhang, *Nat. Commun.*, **2019**, *10*, 2467.
- S8 J. Xie, S. A. Shevlin, Q. Ruan, S. J. A. Moniz, Y. Liu, X. Liu, Y. Li, C. C. Lau, Z. X. Guo, J. Tang, *Energy Environ. Sci.*, **2018**, *11*, 1617-1624.
- S9 Z.-A. Lan, Y. Fang, Y. Zhang, X. Wang, *Angew. Chem. Int. Ed.*, **2018**, *57*, 470-474.
- S10 L. Wang, Y. Wan, Y. Ding, Y. Niu, Y. Xiong, X. Wu, H. Xu, *Nanoscale*, **2017**, *9*, 4090-4096.
- S11 G. Zhang, S. Zang, X. Wang, *ACS Catal.*, **2015**, *9*, 941-947.
- S12 R. Kuriki, H. Matsunaga, T. Nakashima, K. Wada, A. Yamakata, O. Ishitani, K. Maeda, *J. Am. Chem. Soc.*, **2016**, *138*, 5159-5170.