

Supplementary Material

Improving Nitrate-to-Ammonia Conversion on Electrodeposited Nickel Phosphide via Surface δ - FeOOH Modification

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Figure S1. Calibration of the indophenol blue method for NH_3 quantification in alkaline (1.0 M KOH) using NH_4Cl standard solution: (a) UV-Vis spectra of the NH_3 standard solutions after incubation registered from 500 to 800 nm in 1.0 M KOH. (b) Calibration curve used to estimate NH_3 concentration in 1.0 M KOH.

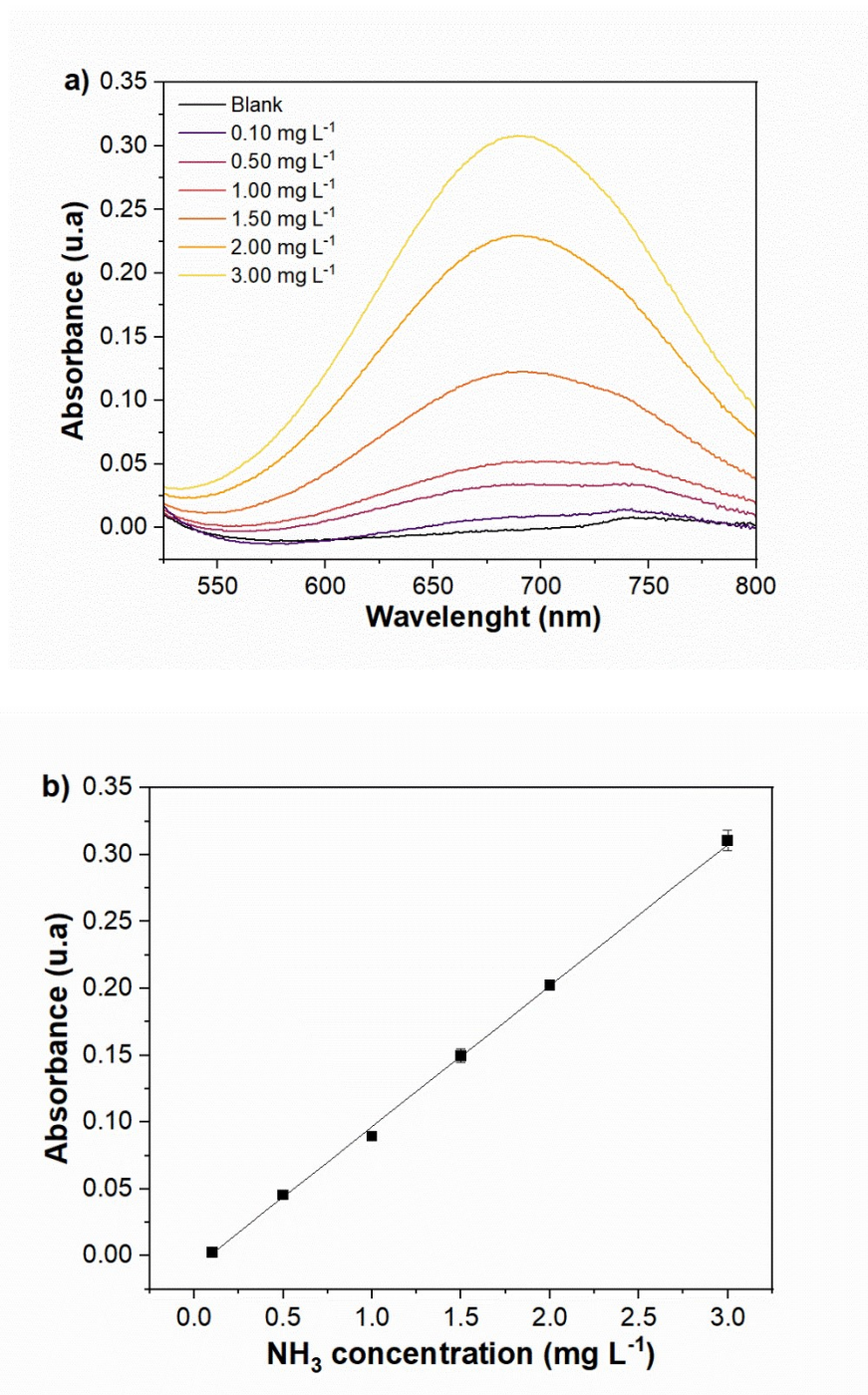


Figure S2. XRD profile of the CFP/ Ni₃P-amorphous and CFP/ Ni₃P-annealed electrodes.

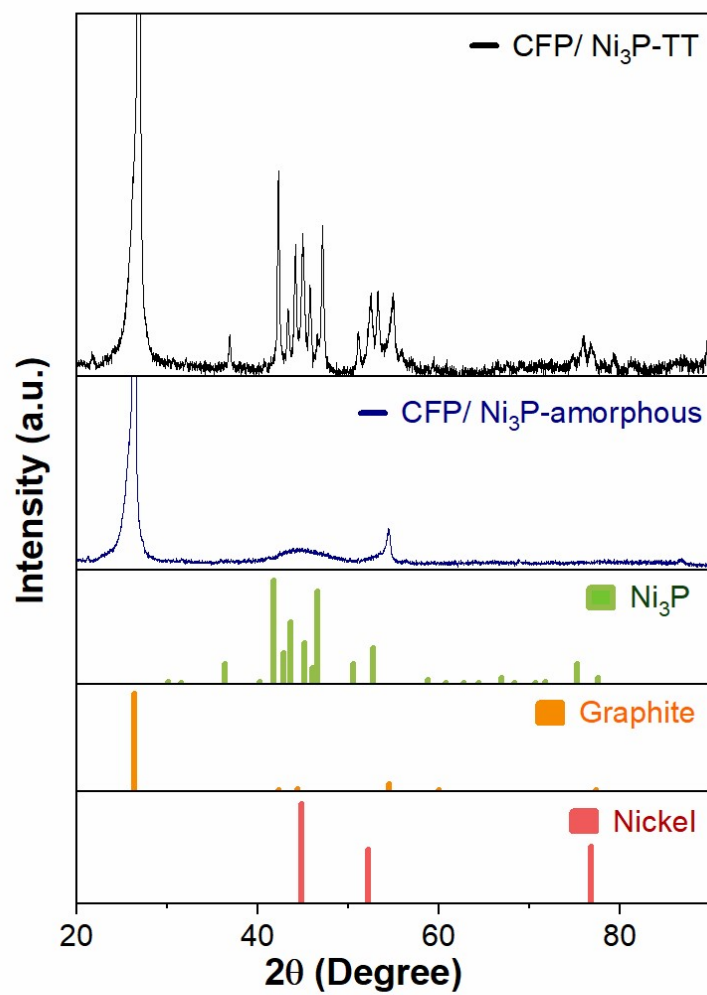


Figure S3. EDS-SEM mapping images for the CFP/Ni₃P/ δ -FeOOH.

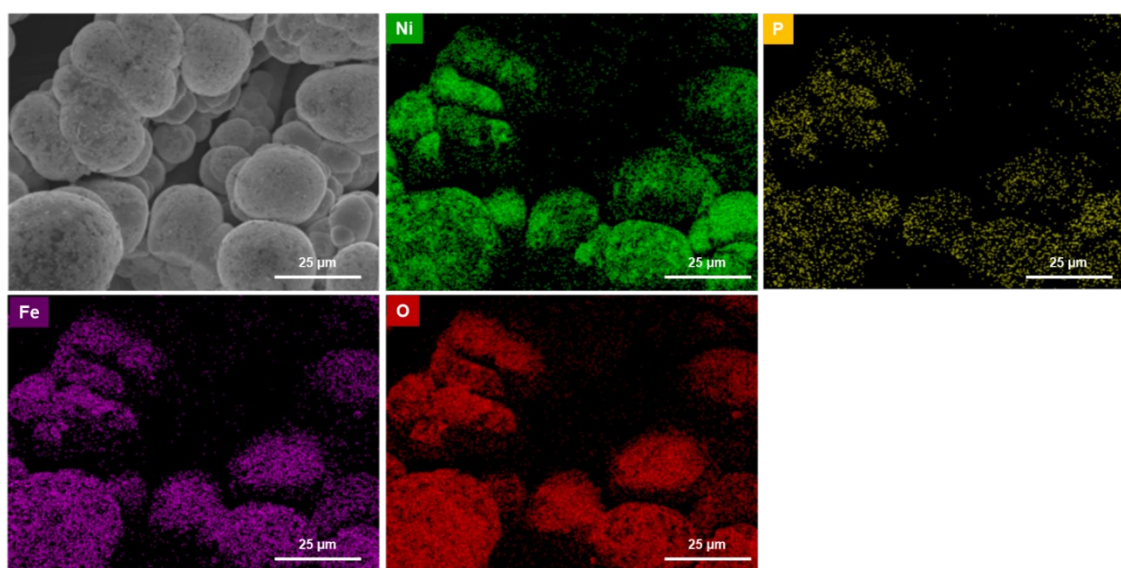


Figure S4. XPS survey spectrum of the CFP/Ni₃P/ δ -FeOOH electrode.

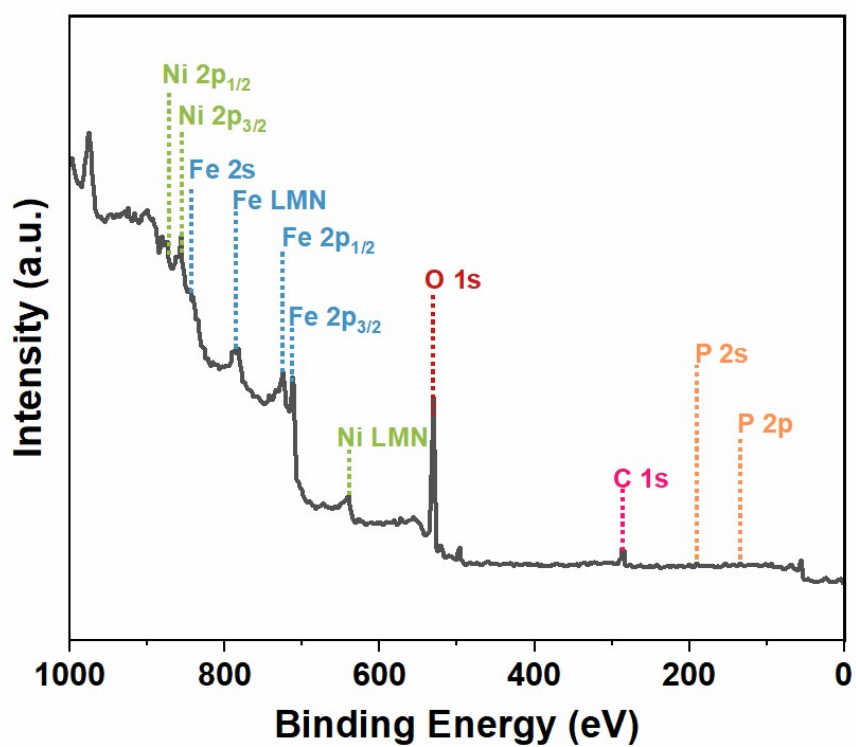


Figure S5. High-resolution Ni 2P and P 2p XPS spectra for the **a)** reference sample (initial Ni₃P surface) and the **b)** Ni₃P surface after 5 min Ar⁺ bombardment.

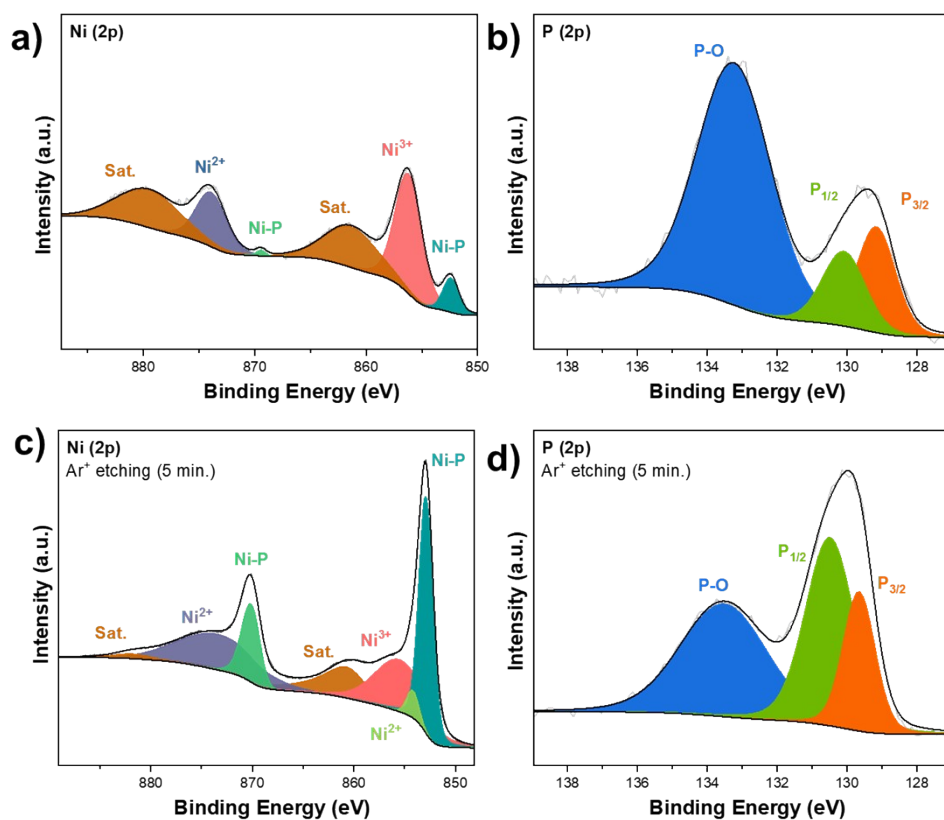


Figure S6. ECSA-normalized Linear sweep voltammetry curves registered in 1.0 M KOH + 0.1 M KNO₃ at 5 mVs⁻¹.

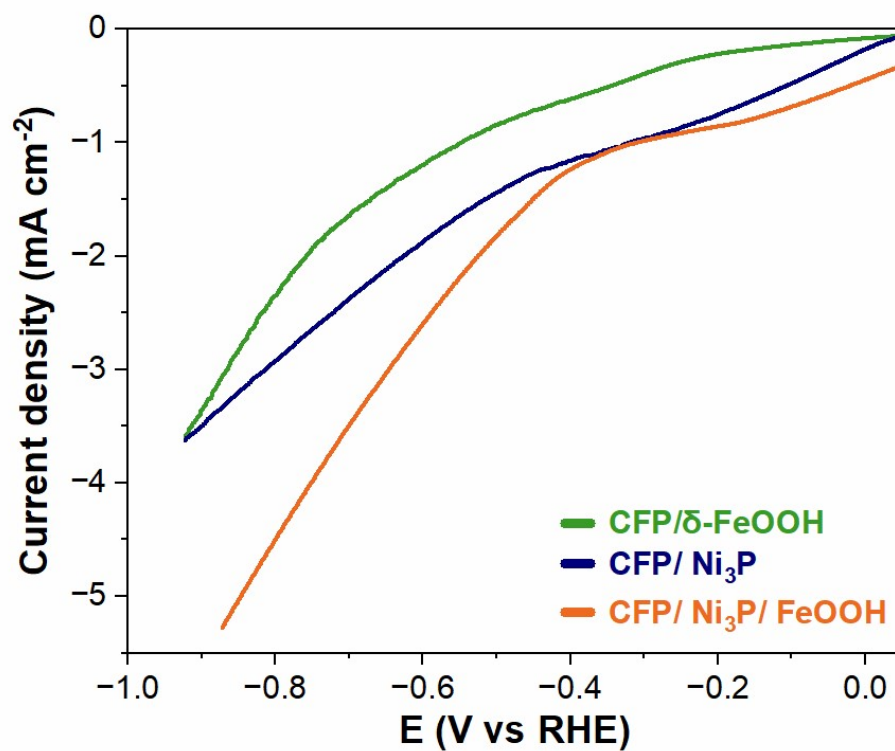


Figure S7. Chronoamperometry curves registered in 1.0 M KOH with 0.10 M KNO_3 at different applied potentials for the a) CFP/ $\text{Ni}_3\text{P}/\delta\text{-FeOOH}$ and b) CFP/ Ni_3P .

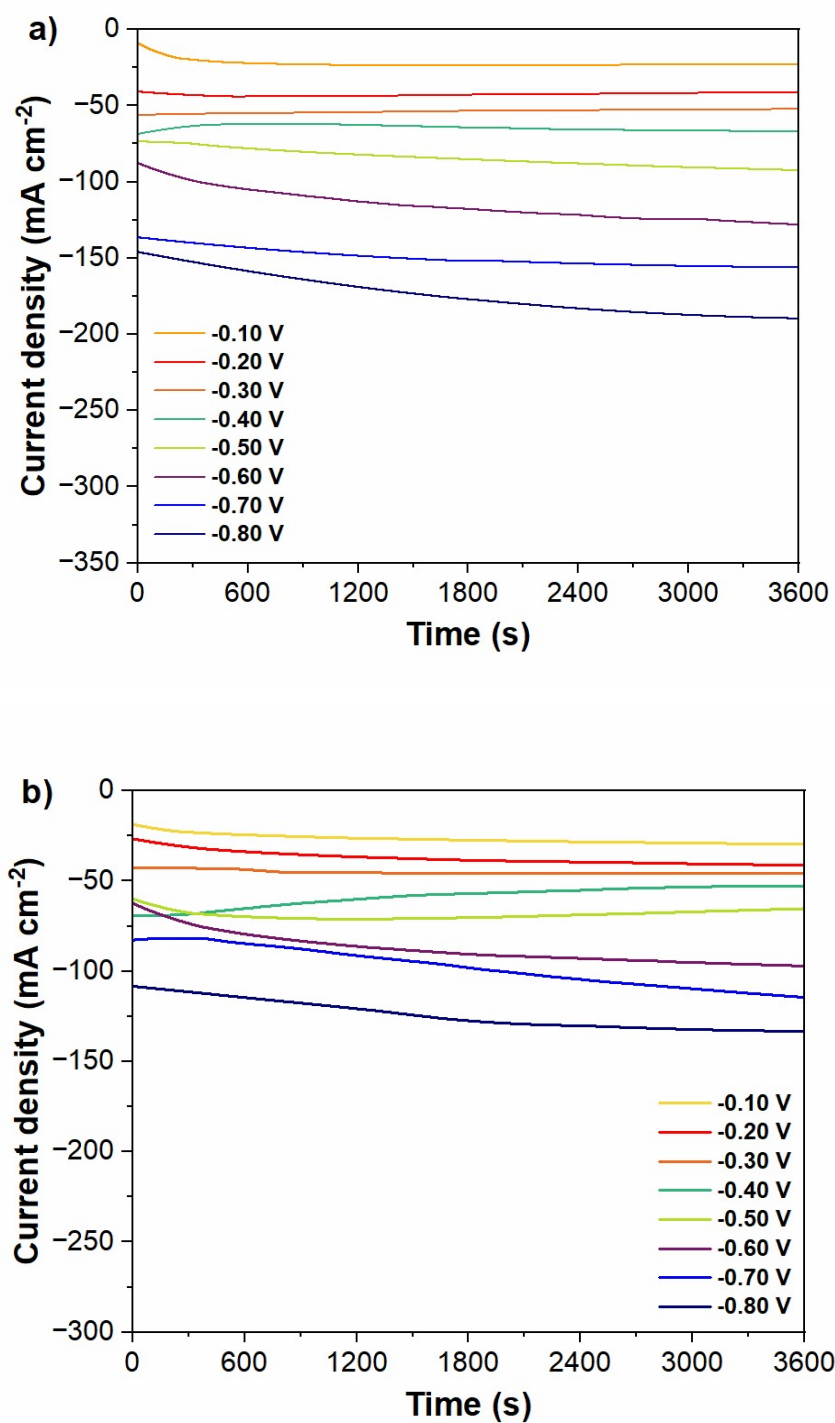


Figure S8. Cyclic voltammograms registered for 1.0 M KOH with different NO₂ concentrations (0.005 to 0.10 M). (b) Standard curve for NO₂- quantification.

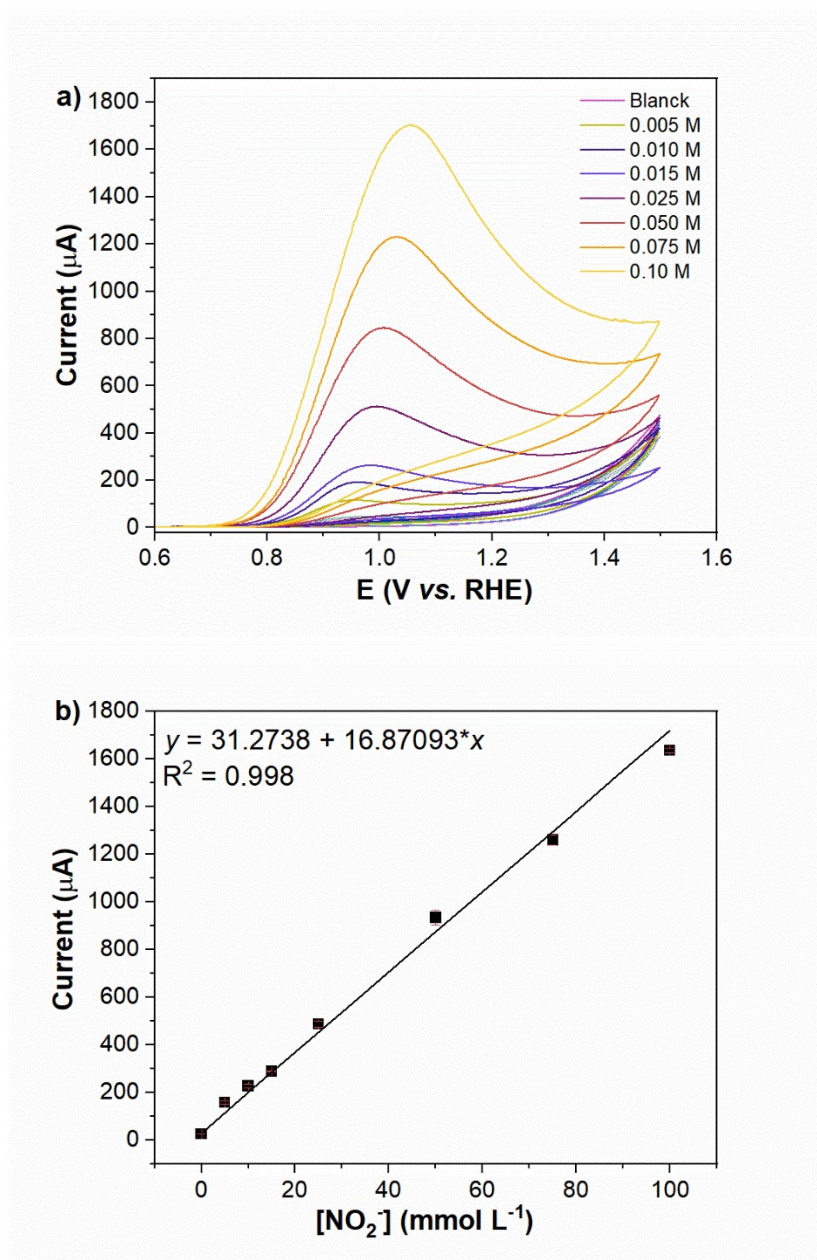


Figure S9. Chronoamperometry curves registered in 1.0 M KOH with 0.10 M KNO_3 at different applied potentials for the a) CFP/ $\text{Ni}_3\text{P}/\delta\text{-FeOOH}$ and b) CFP/ Ni_3P .

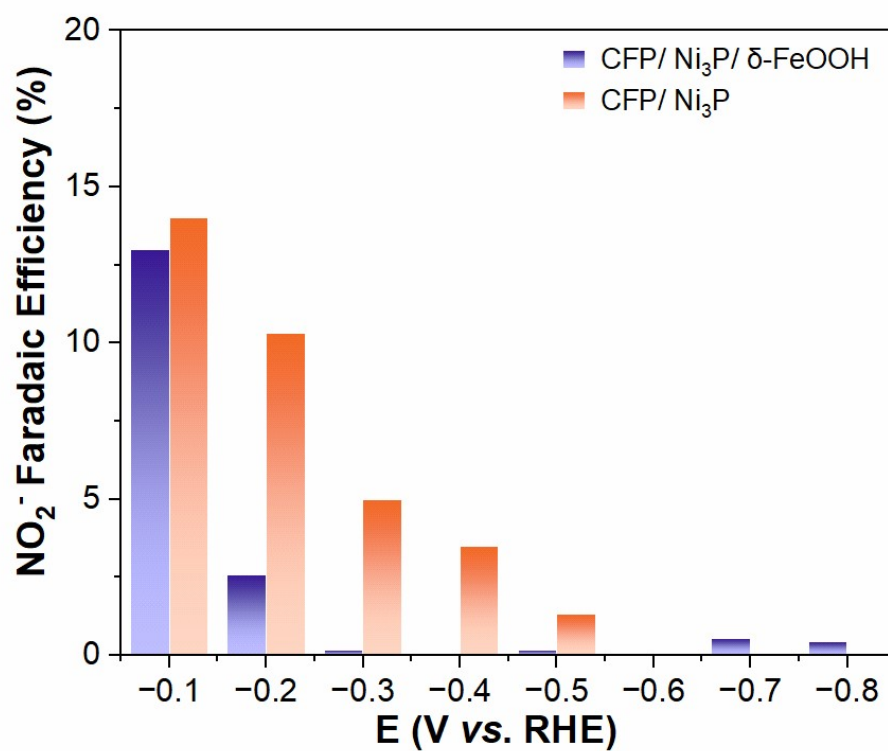


Figure S10. Ammonia yield rate at OCP (CFP/Ni₃P/δ-FeOOH) and at -0.30 V vs RHE of the individual CFP, CFP/Ni₃P, CFP/δ-FeOOH and CFP/Ni₃P/δ-FeOOH electrodes in KOH + 0.1 M KNO₃.

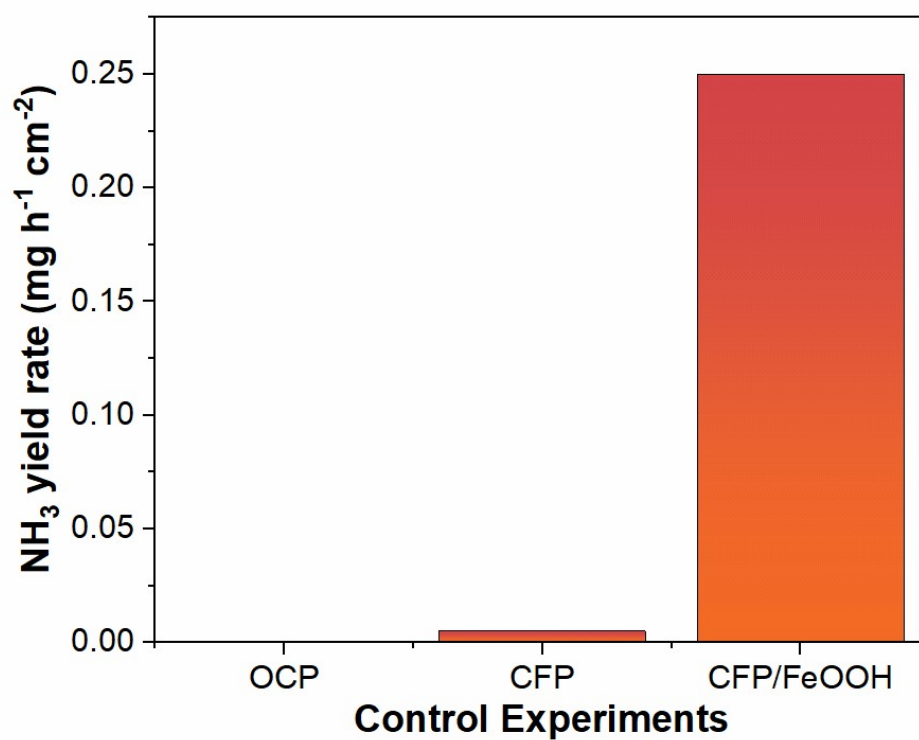


Figure S11. Comparison of NH_3 yield and applied potential for electroreduction of nitrate to ammonia under ambient conditions using alkaline electrolyte (1 M) $^{1-}$

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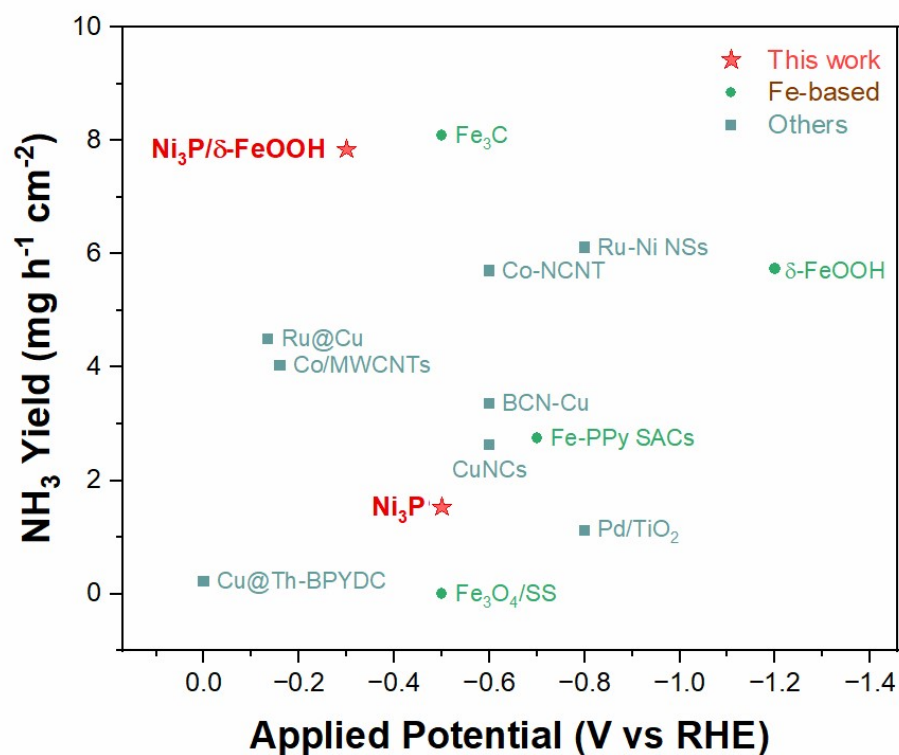


Figure S12. a) Linear sweep voltammetry curves registered with the CFP/Ni₃P/ δ -FeOOH sample at 5 mVs⁻¹ in 1.0 M KOH with different KNO₃ concentrations. b) NH₃ yield rate and Faradaic efficiency for the δ -FeOOH modified Ni₃P electrode in different KNO₃ concentrations at -0.30 V vs. RHE.

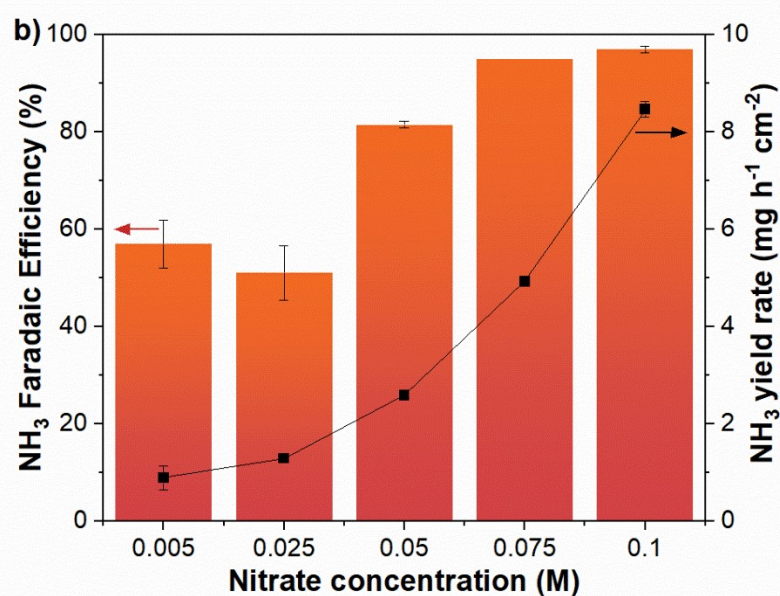
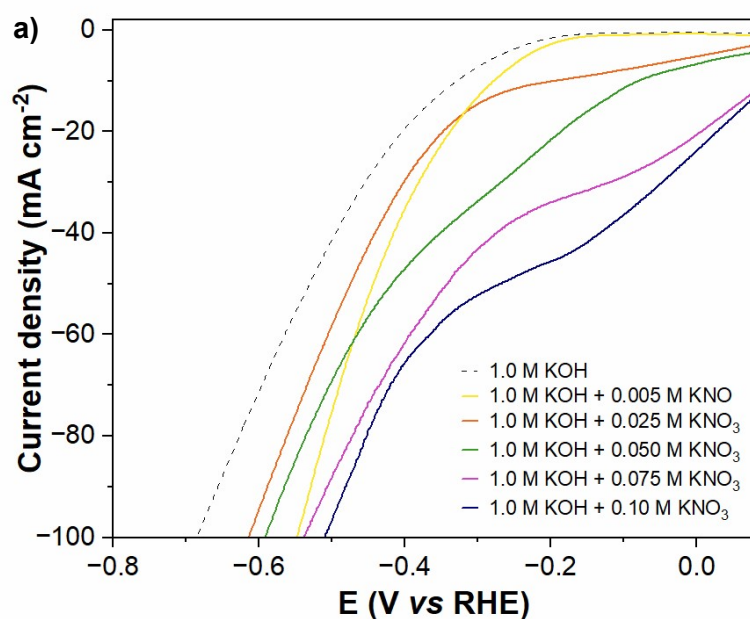
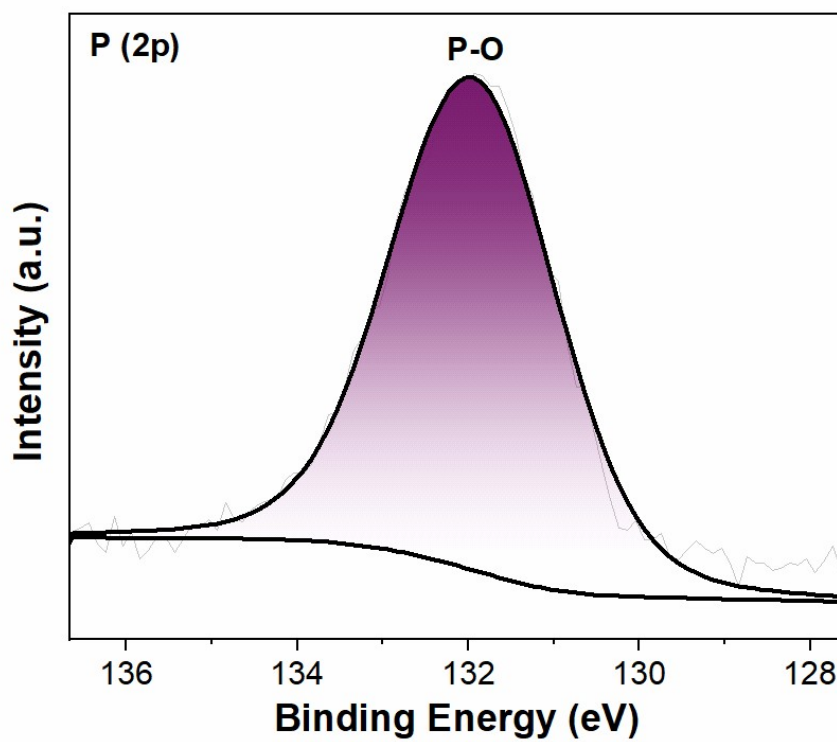


Figure S13. P 2p XPS spectra of the CFP/Ni₃P/ δ -FeOOH sample after consecutive recycling cycles.



References

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