

Supporting Information

Depolymerization of Lignin Disassembly into Cycloalkanes over Hydrotalcite-derived NiFe Alloy Catalyst

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Table S1. PPE conversion (%) of Ni₉Fe₁/NiAlO₂ with different solvents and with or without H₂.

Reaction conditions: 500 mg catalyst, 500 mg PPE, 70 mL solvent, 100 °C, 3 h, with (0.8 MPa) or without H₂ pressure.

Solvent	H ₂ O	Methanol	Ethanol	Isopropanol	n-Dodecane
With H ₂	8.5	10.1	19.8	33.4	99.8
Without H ₂	-	-	-	-	-

Table S2. EXAFS fitting parameters at the Ni K-edge for various samples.

Samples	Shell	CN	R (Å)	σ^2 (Å ²)	ΔE_0 (eV)	R factor
Ni foil	Ni-Ni	12	2.5	0.006	6.3 (0.6)	0.003
NiO	Ni-O	6.1 (0.3)	2.1	0.009	-2.4 (1.1)	0.006
	Ni-Ni	11.9 (0.4)	3.0	0.008		
Ni ₉ Fe ₁ /NiAlO _z	Ni-O	6	2.1	0.010	2.4 (2.5)	0.01
	Ni-M ₁	12	2.6	0.015		
	Ni-M ₂	12	3.0	0.014		

CN , coordination number; R , distance between absorber and backscatter atoms; σ^2 , the Mean Square Relative Displacement (MSRD); ΔE_0 , energy shift interpreted as the alignment of the energy grids of the data and the theory; R factor indicates the goodness of the fit.

S_0^2 was fitted as 0.75 for Ni foil by fixing CN as the known crystallographic value. Fitting range of Ni foil is $3.0 \leq k$ (Å⁻¹) ≤ 10.5 and $1 \leq R$ (Å) ≤ 3 . S_0^2 was fixed as 0.98 for the fitting of NiO. Fitting range of NiO is $3.0 \leq k$ (Å⁻¹) ≤ 10 and $1 \leq R$ (Å) ≤ 3 . S_0^2 was fixed as 0.80 for the fitting of Ni₉Fe₁/NiAlO_z. Fitting range of Ni₉Fe₁/NiAlO_z is $3.0 \leq k$ (Å⁻¹) ≤ 10.4 and $1 \leq R$ (Å) ≤ 3 .

Table S3. EXAFS fitting parameters at the Fe K-edge for various samples.

Samples	Shell	CN	$R(\text{\AA})$	$\sigma^2(\text{\AA}^{-2})$	$\Delta E_0(\text{eV})$	R factor
Fe foil	Fe-Fe	12	2.5	0.0078	5.8 (2.8)	0.005
Fe_2O_3	Fe-O ₁	4.2 (1.7)	2.0	0.010		
	Fe-O ₂	1.8	2.2			
	Fe-Fe ₁	3	3.0			
	Fe-Fe ₂	1	2.9	0.005	-1.3 (2.4)	0.004
	Fe-Fe ₃	2	3.4			
	Fe-Fe ₄	6	3.7	0.010		
	Fe-O ₃	3	3.9	0.006		
$\text{Ni}_9\text{Fe}_1/\text{NiAlO}_z$	Fe-O ₁	5.4 (0.7)	2.0	0.014		
	Fe-M ₁	6.0 (1.3)	2.6	0.023	1.5 (1.2)	0.007
	Fe-M ₂	5.8 (1.8)	3.0	0.013		
	Fe-O ₂	4	3.3	0.028		

S_0^2 was fitted as 0.75 for Fe foil by fixing CN as the known crystallographic value. Fitting range of Fe foil is $3.0 \leq k$ (\AA^{-1}) ≤ 11.9 and $1 \leq R$ (\AA) ≤ 3 . S_0^2 was fixed as 1.00 for the fitting of Fe_2O_3 . Fitting range of Fe_2O_3 is $3.0 \leq k$ (\AA^{-1}) ≤ 10 and $1 \leq R$ (\AA) ≤ 3.5 . S_0^2 was fixed as 0.85 for the fitting of $\text{Ni}_9\text{Fe}_1/\text{NiAlO}_z$. Fitting range of $\text{Ni}_9\text{Fe}_1/\text{NiAlO}_z$ is $3.0 \leq k$ (\AA^{-1}) ≤ 8 and $1 \leq R$ (\AA) ≤ 3 .

Table S4 The performance comparison of Ni₉Fe₁/NiAlO₂ with previously reported catalysts for model compound and lignin depolymerization.

Catalysts	Substrate	Reaction condition	Major products and selectivity		Ref.			
Mo ₁ Al/MgO		200 °C, 4 h, 1 MPa N ₂		58 %		J. Am. Chem. Soc. 2023, 145, 12884-12893		
Ni-Fe/MCS		250 °C, 4 h, 5 MPa H ₂		49.1%		49.9%	Appl. Catal. B-Environ. 2019, 253, 348-358	
Br-Ru/SiO ₂		120 °C, 6 h, 0.5 MPa H ₂		44.7%		45.6%	Angew. Chem. Int. Ed. 2021, 60, 12513-125233	
Ni/Ni-PS		160 °C, 2 h, 1 MPa H ₂		33.8%		4	7.6%	Green Chem. 2022, 24, 846-857
NiFe(3)/TiO ₂ -HT		250 °C, 1 h, 5 MPa H ₂		71.8%			Chem. Eng. J. 2022, 446 136578	
RuRe/H-Beta		250 °C, 10 h, 0 MPa H ₂		26.1%		43.8%	J. Energy Chem. 2022, 67, 492-49	
Ni ₉ Fe ₁ /NiAlO ₂		100 °C, 3 h, 0.8 MPa H ₂		40.5%		56.8%	This work	
Mo ₁ Al/MgO	Eucalyptus wood	200 °C, 8 h, 1 MPa N ₂		92%			J. Am. Chem. Soc. 2023, 145, 12884- 12893	
RuRe/H-Beta	Alkaline lignin	200 °C, 2 MPa H ₂ , 10 h		54%			J. Energy Chem. 2022, 67, 492-49	
Pd/C	Extracted lignin from birch	200 °C, 15 h, 4 MPa H ₂		90%			Angew. Chem. Int. Ed. 2018, 57, 1356.	
Ru/NbOPO ₄ ^a	Kraft lignin	310 °C, 40 h, 0.5 MPa H ₂		68%			Chem, 2019, 5, 1521- 1536,	
Ni ₉ Fe ₁ /NiAlO ₂ ^a	Enzymatic hydrolysis lignin	300 °C, 6 h, 3 MPa H ₂		100%			This Work	

^a Breaking the limit of lignin monomer production

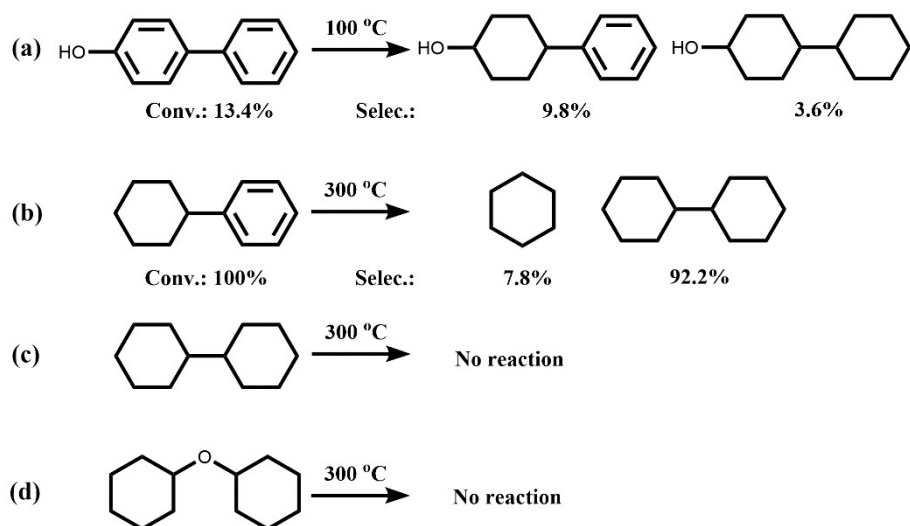


Fig. S1. Product distributions for the conversion of (a) 4-phenylphenol, (b) phenylcyclohexane, (c) bicyclohexyl and (d) cyclohexyloxy-cyclohexane.

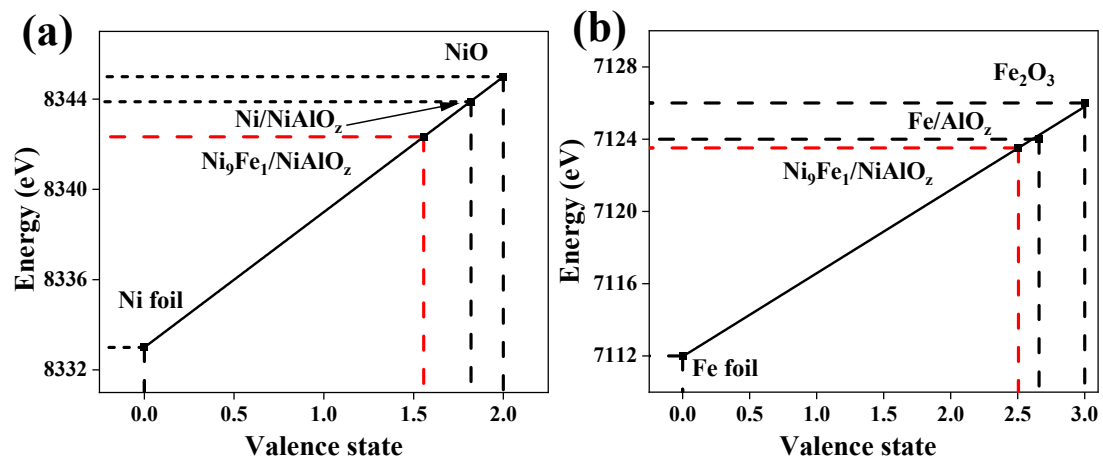


Fig. S2. The valence result of Ni (a) and Fe (b).

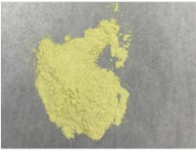

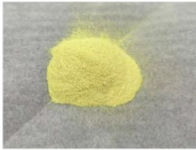
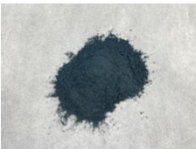


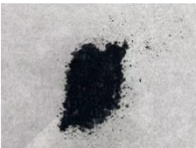


	$\text{Ni}_9\text{Fe}_1/\text{NiAlO}_z$	Ni/NiAlO_z	Fe/AlO_z
0 S			
30 S			
180 S			

Fig. S3. Color change photographs of the mixtures.

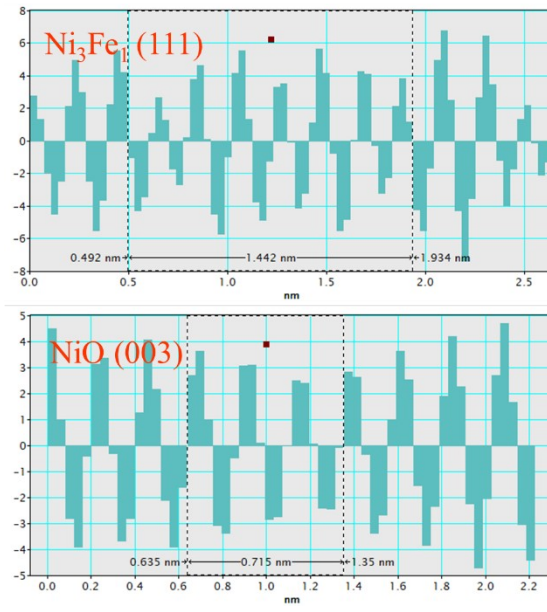


Fig. S4. Lattice stripe calculation for HRTEM.