

Supporting Information

Enhanced photocatalytic performance of g-C₃N₄ by introducing gradient energy band structure and activated n-π* electronic transition for visible light hydrogen generation

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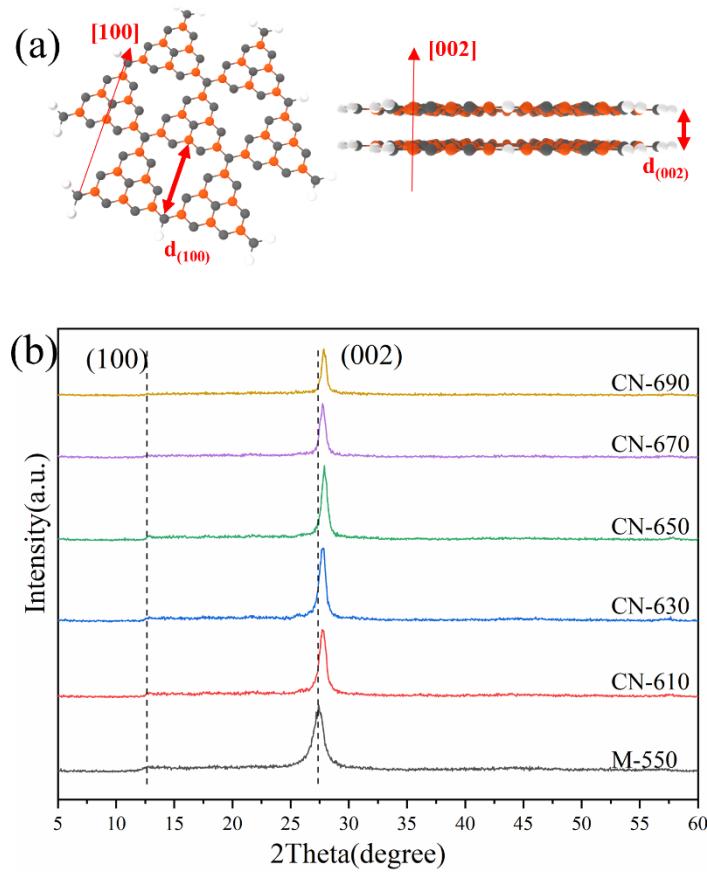


Fig. S1 (a) Ideal $\text{g-C}_3\text{N}_4$ structure constructed by stacking heptazine layers, (b) XRD patterns of M-550 and CN-x samples.

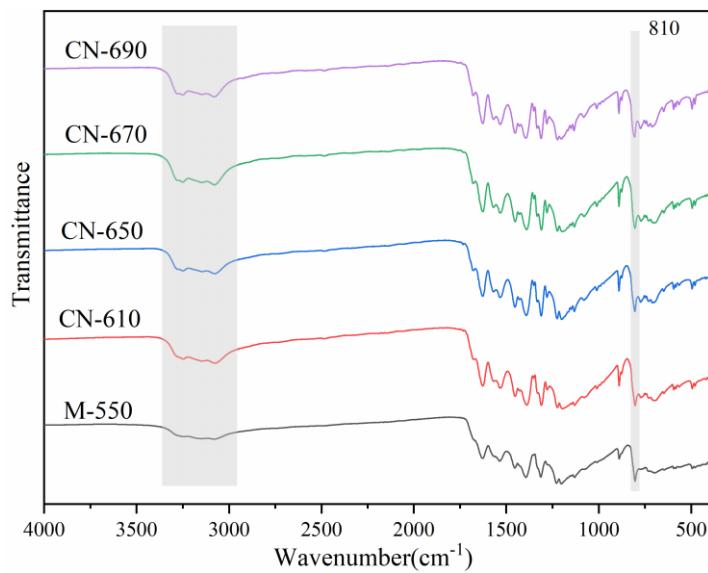


Fig. S2 FT-IR spectra of M-550 and CN-x samples.

Table. S1 Element contents in M-550, CN-670, KCN-30, KCN-75 and KCN105 samples estimated by elemental analysis.

Elemental analysis	C(wt%)	N(wt%)	H(wt%)	C/N(mol ratio)
M-550	35.4	62.4	2.23	0.662
CN-670	35.8	62.1	2.04	0.673
KCN-30	36.3	61.8	1.91	0.685
KCN-75	36.7	61.5	1.80	0.697
KCN-105	37.3	61.1	1.63	0.711

Table. S2 K element contents in CN-670, KCN-30, KCN-75 and KCN-105 samples estimated by ICP analysis.

ICP	CN-670	KCN-30	KCN-75	KCN-105
K(wt%)	Trace	2.26	5.37	7.92

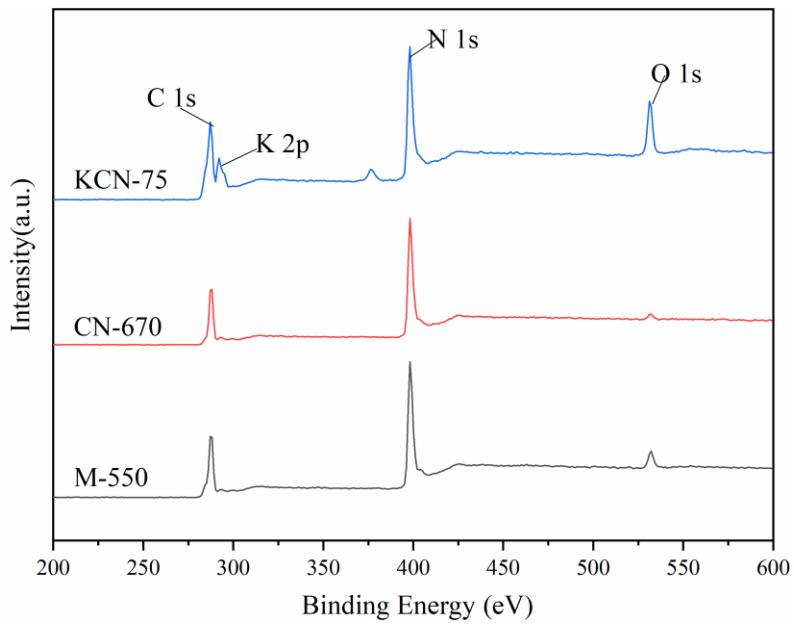


Fig. S3 XPS survey spectra of M-550, CN-670, and KCN-75 samples.

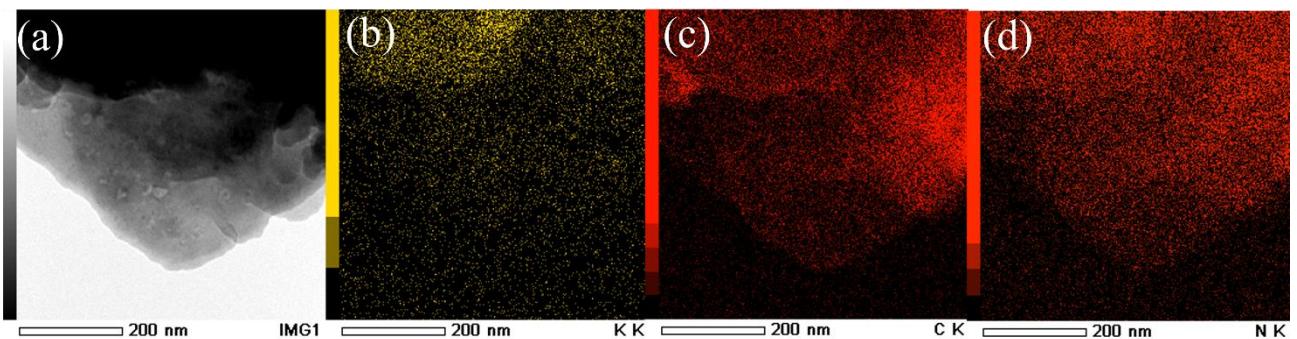


Fig. S4 EDX element mapping of KCN-75 samples.

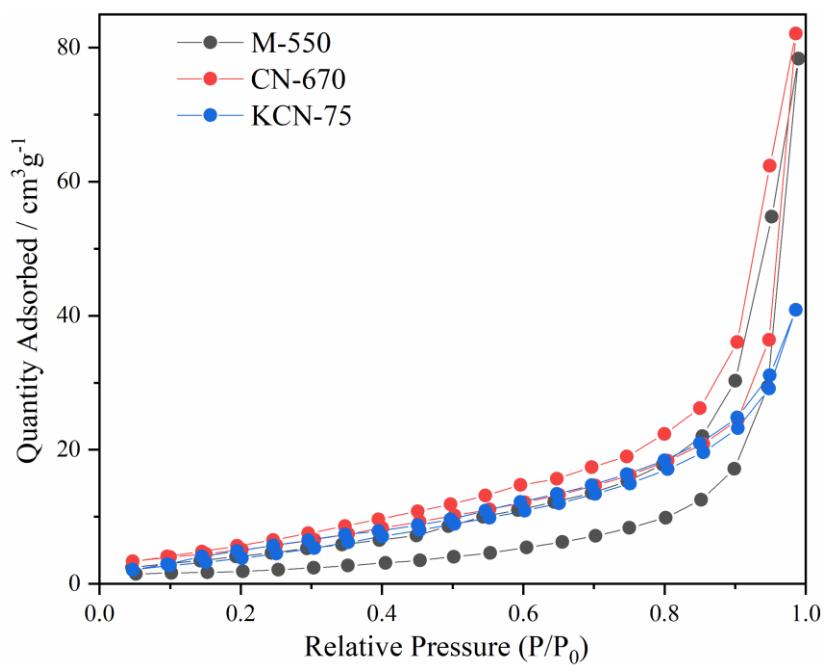
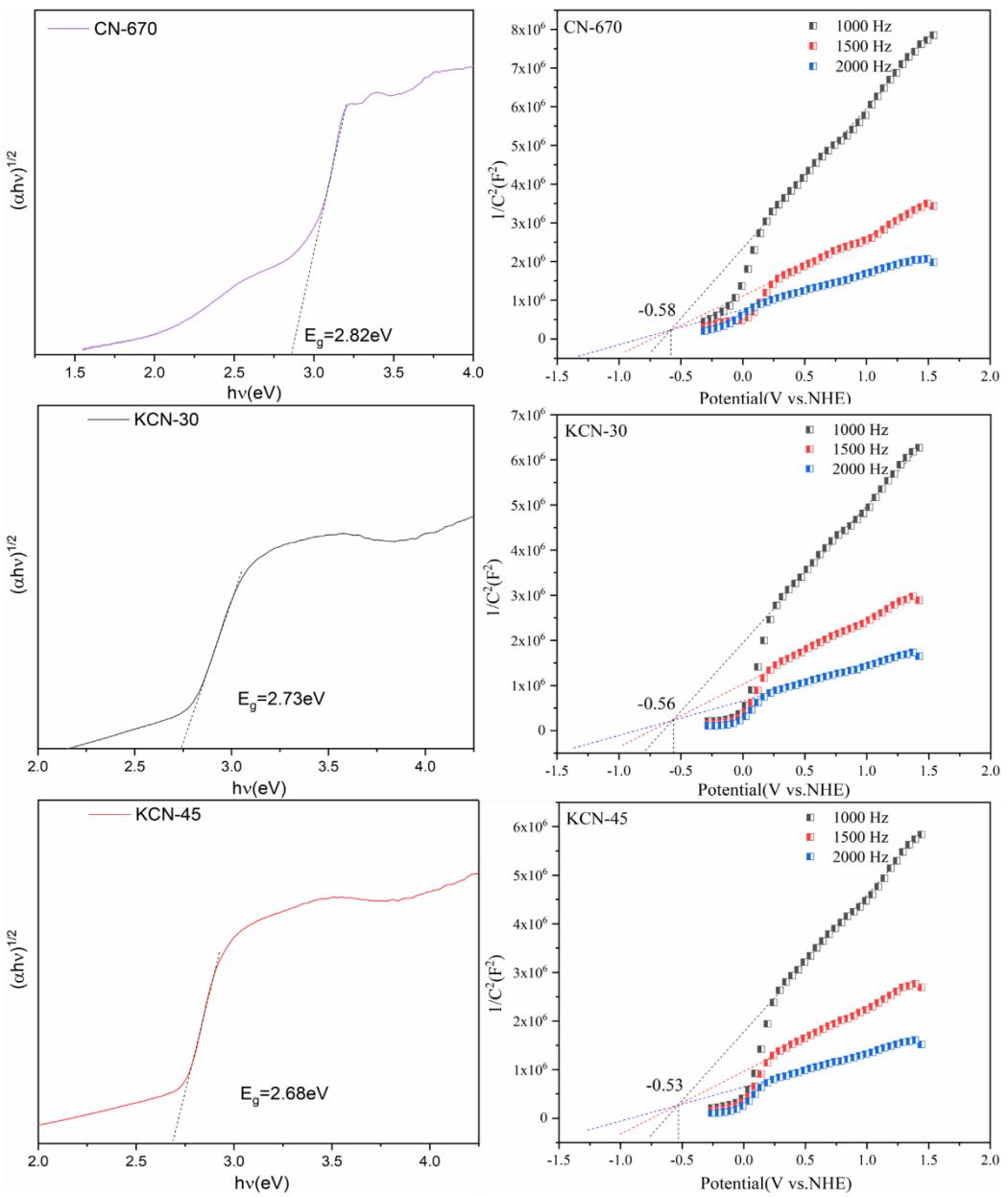


Fig. S5 N₂ adsorption-desorption isotherms of M-550, CN-670, and KCN-75 samples.



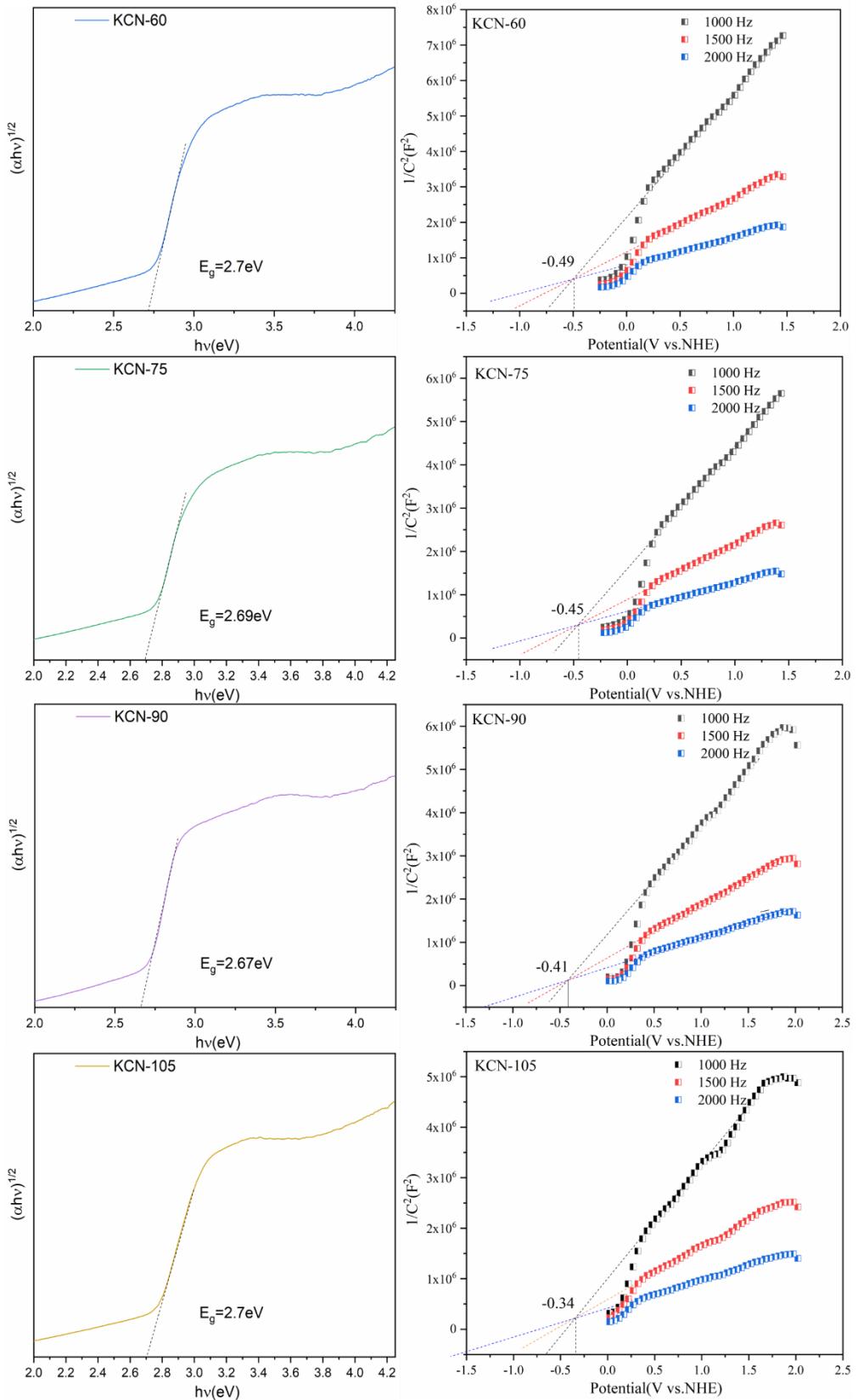


Fig. S6 Plots of transformed Kubelka-Munk function and Mott-Schottky plots for CM-670 and KCN-x samples

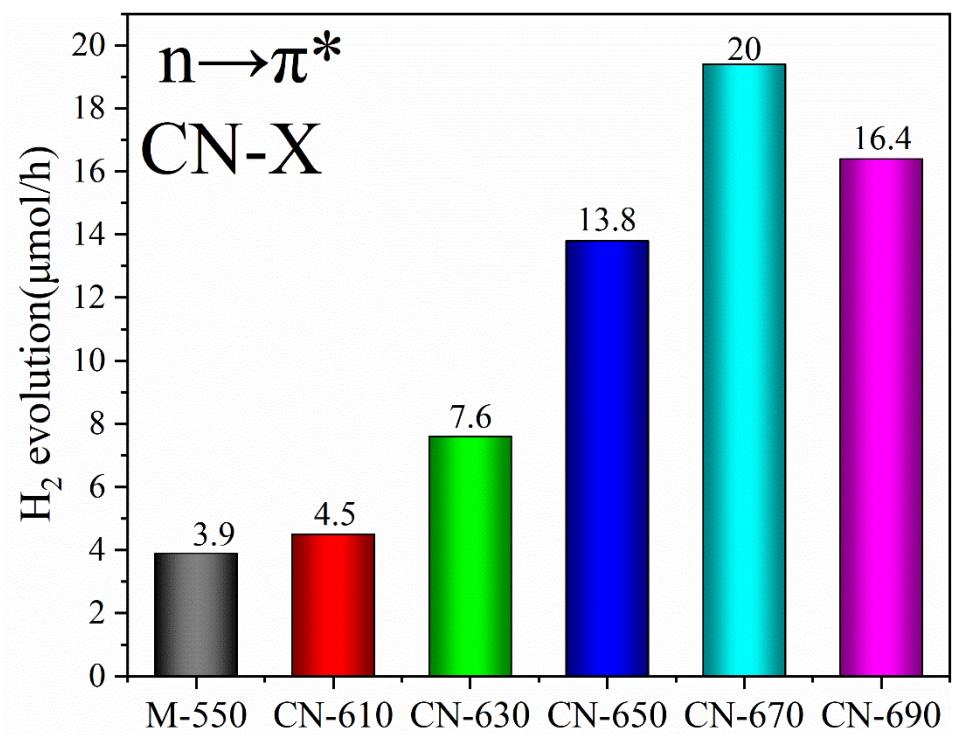


Fig. S7 H₂ production rates of M-550 and CN-x samples.

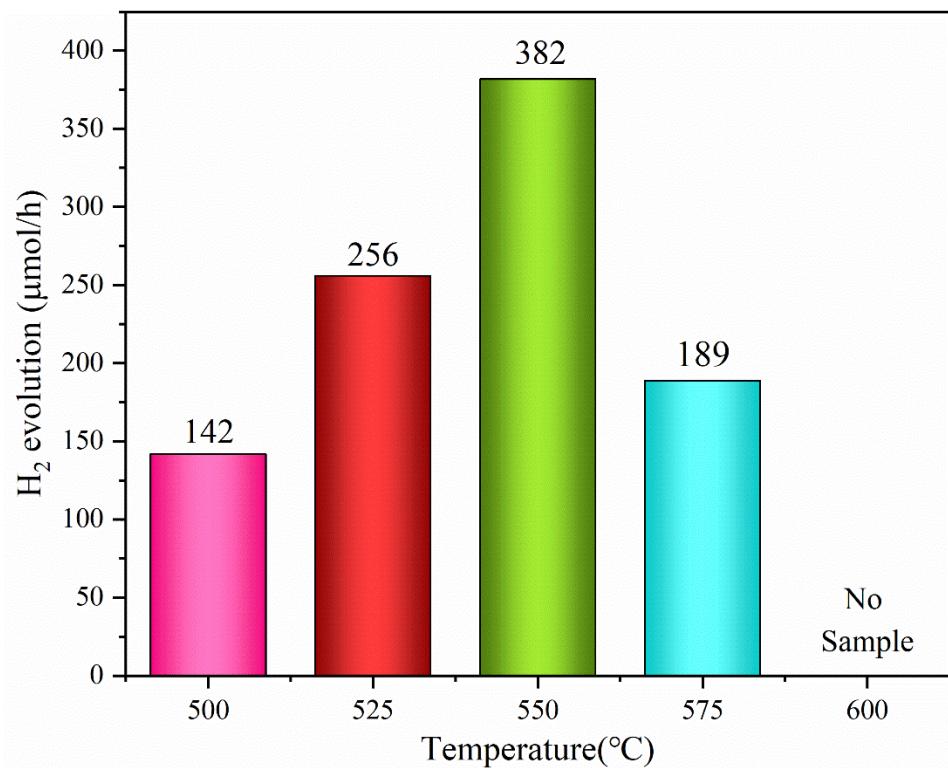


Fig. S8 H₂ production rates of samples with different temperature.

Table S3 Values of elements in equivalent circuit resulted from fitting the EIS data in Fig 7(b).

Sample	$R_s(\Omega \text{ cm}^{-2})$	$R_{ct}(\Omega \text{ cm}^{-2})$	CPE(F cm^{-2})
CN-670	11.5	2.67E5	3.23E-5
KCN-30	11.5	2.06E5	3.15E-5
KCN-45	10.3	1.85E5	2.88E-5
KCN-60	11.5	1.70E5	3.18E-5
KCN-75	9.34	1.43E5	3.04E-5
KCN-90	11.3	1.53E5	2.94E-5
KCN-105	10.4	1.56E5	3.01E-5

Table S4 Bandgap E_g , conduction band position E_c and valence band position E_v (vs. NHE) of CM-670 and KCN-x samples from Fig S6.

Sample	$E_g(\text{eV})$	$E_c(\text{eV})$	$E_v(\text{eV})$
CN-670	2.82	-0.58	2.24
KCN-30	2.73	-0.56	2.17
KCN-45	2.68	-0.53	2.15
KCN-60	2.70	-0.49	2.21
KCN-75	2.69	-0.45	2.24
KCN-90	2.67	-0.41	2.26
KCN-105	2.70	-0.34	2.36

Table. S5 Summary table of H₂-evolution by different g-C₃N₄-based materials.

Catalysts	Light Source	Reaction Conditions	H ₂ evolution Rate (μmol h ⁻¹)	Reference
KCN-75	300 W Xenon lamp (λ>420 nm)	50 mg catalyst, 100 mL, 20 vol% TEOA, 3wt%Pt	382	This work
CN-M-630	300 W Xenon lamp (λ>420 nm)	25 mg catalyst, 100 mL, 10 vol% TEOA, 3wt%Pt	132	[1]
K-CN 0.4	300 W Xenon lamp (λ>420 nm)	50 mg catalyst, 50 mL, 10 vol% TEOA, 3wt%Pt	200	[2]
UCN-670	300 W Xenon lamp (λ>420 nm)	10 mg catalyst, 30 mL, 10 vol% TEOA, 1.1wt%Pt	92	[3]
KCCN 2	300 W Xenon lamp (λ>420 nm)	50 mg catalyst, 50 mL, 20 vol% TEOA, 3wt%Pt	262	[4]
SA-Pt/g-C ₃ N ₄ -8.7	300 W Xenon lamp (λ>420 nm)	10 mg catalyst, 100 mL, 10 vol% TEOA, 8.7wt%Pt	227	[5]
Few-layer C ₃ N ₄	300 W Xenon lamp (λ>420 nm)	20 mg catalyst, 100 mL, 20 vol% TEOA, 1wt%Pt	160	[6]
S-g-C ₃ N ₄ -D	300 W Xenon lamp (λ>420 nm)	20 mg catalyst, 100 mL, 10 vol% TEOA, 3wt%Pt	113	[7]
BPCN1.5	300 W Xenon lamp (λ>420 nm)	5 mg catalyst, 3 mL, 10 vol% TEOA, 1wt%Pt	23	[8]
CNV-mCN-Br	300 W Xenon lamp (λ>420 nm)	10 mg catalyst, 100 mL, 10 vol% TEOA, 1wt%Pt	222	[9]
DA-HM	300 W Xenon lamp (λ>420 nm)	50 mg catalyst, 100 mL, 10 vol% TEOA, 3wt%Pt	207	[10]
HcPCN	300 W Xenon lamp (λ>420 nm)	20 mg catalyst, 100 mL, 10 vol% TEOA, 3wt%Pt	190	[11]
cMel-5	200 W Xe lamp (λ ≥ 420 nm)	50 mg catalyst, 80 mL, 10 vol% TEOA, 3wt%Pt	324	[12]
Cu-Ni(OH) ₂ -C ₃ N ₄	300 W Xe lamp (AM 1.5G)	20 mg in 100 mL 10 vol% TEOA	41	[13]
20-CNCo	300 W Xe lamp (AM 1.5G)	50 mg catalyst, 100 mL, 10 vol% TEOA, 3wt%Pt	7	[14]
K-I-CCN	300 W Xenon lamp (λ>400 nm)	50 mg catalyst, 100 mL, 10 vol% TEOA, 3wt%Pt	256	[15]

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