

Supplementary information

Proton transport enhanced by octahedral distortion and built-in electric field in PMN-TiO₂ heterointerface

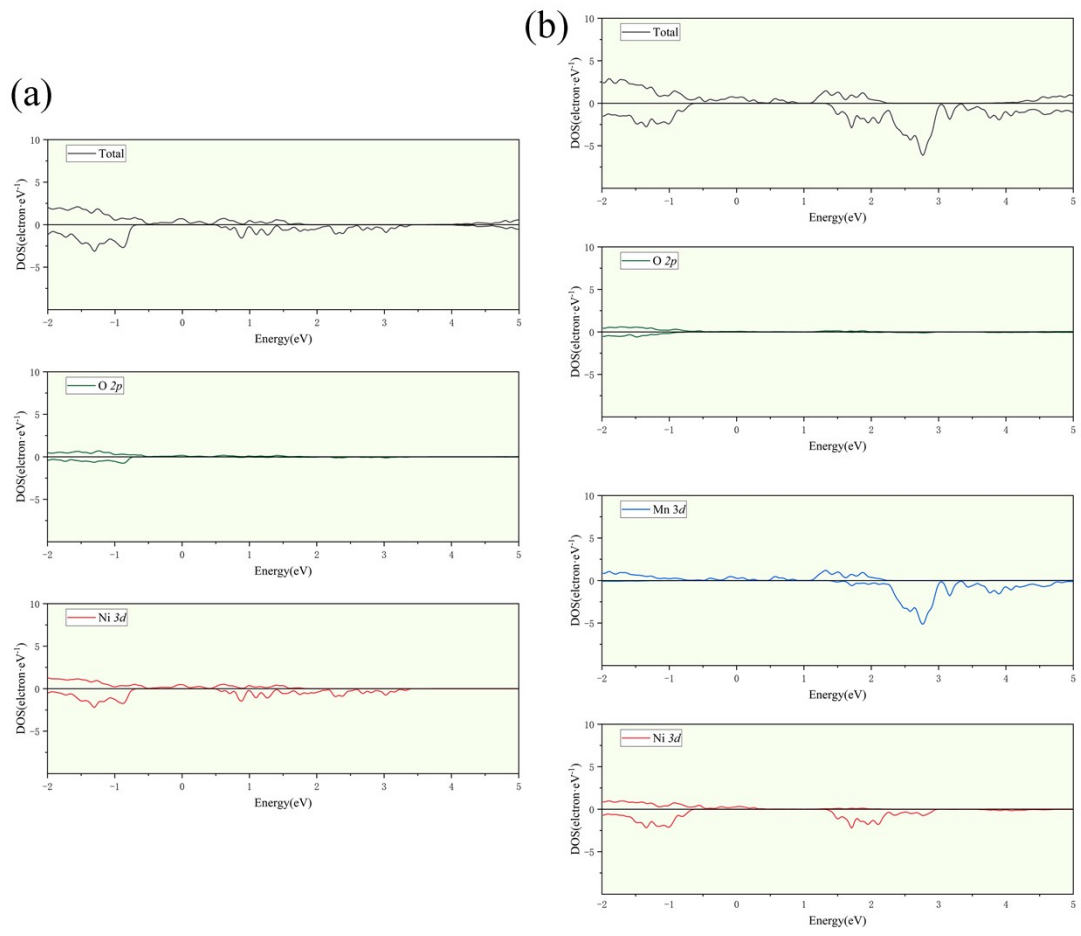
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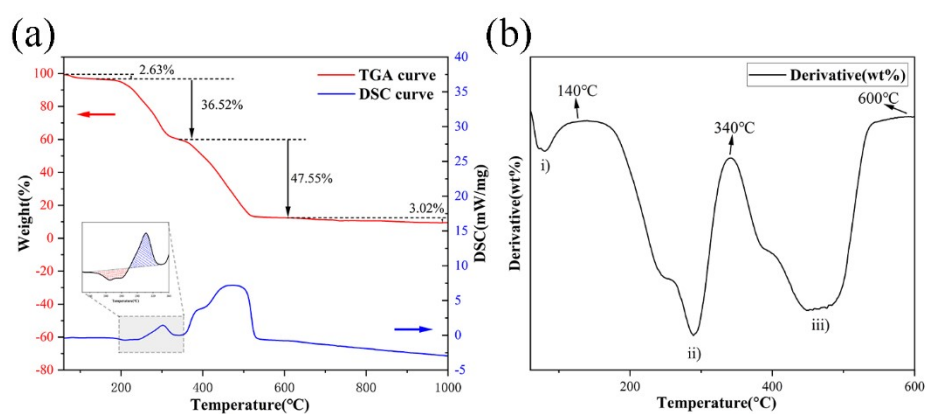
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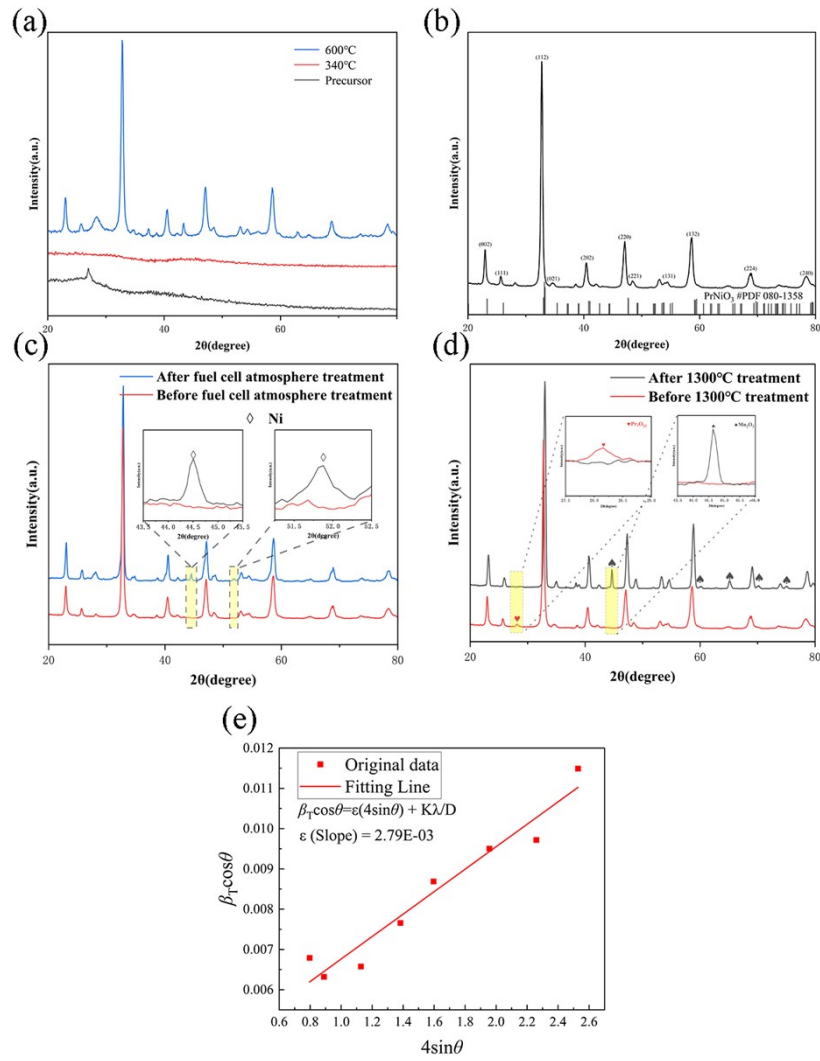
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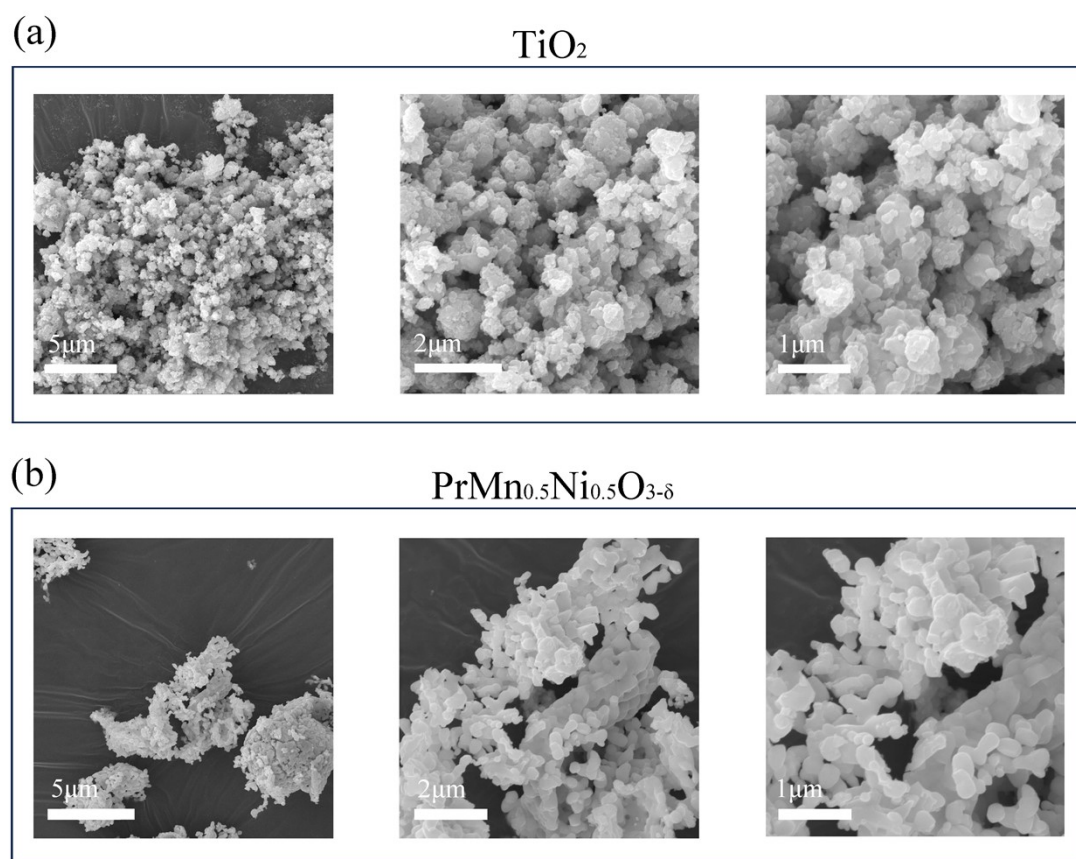
Supplementary Fig. 1 Projected density of states of (a)PNO and (b)PMN



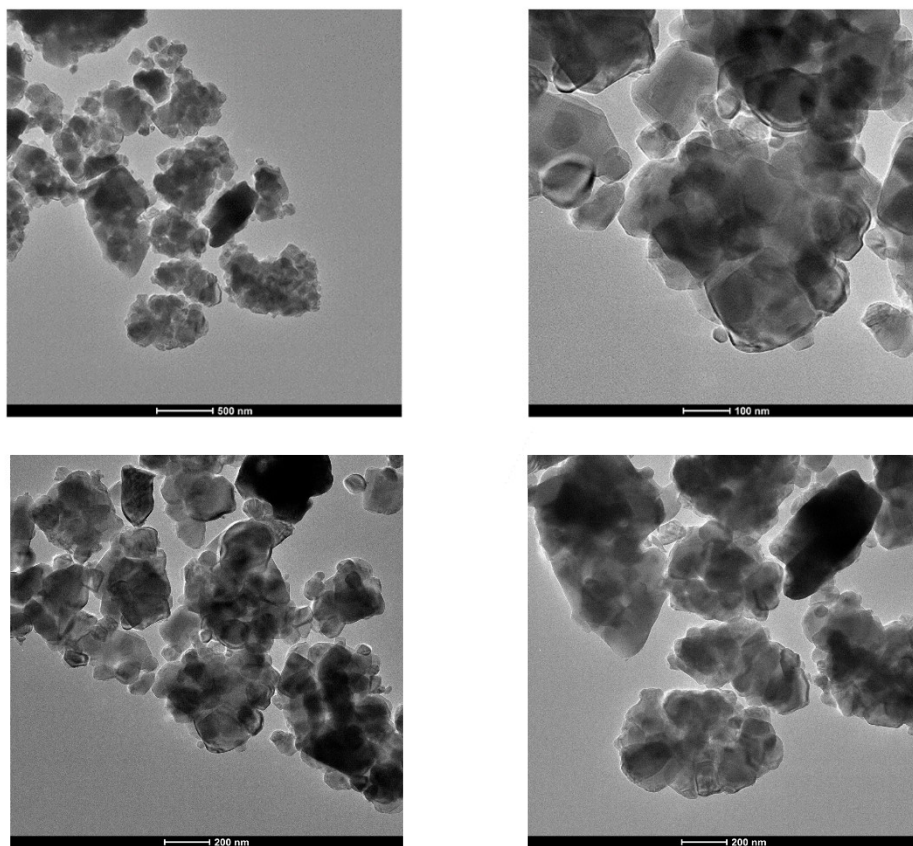
Supplementary Fig. 2 (a) TG analysis (TGA)/DSC curves recorded during the thermal decomposition of PMN precursors. (b) First derivative TGA curves.



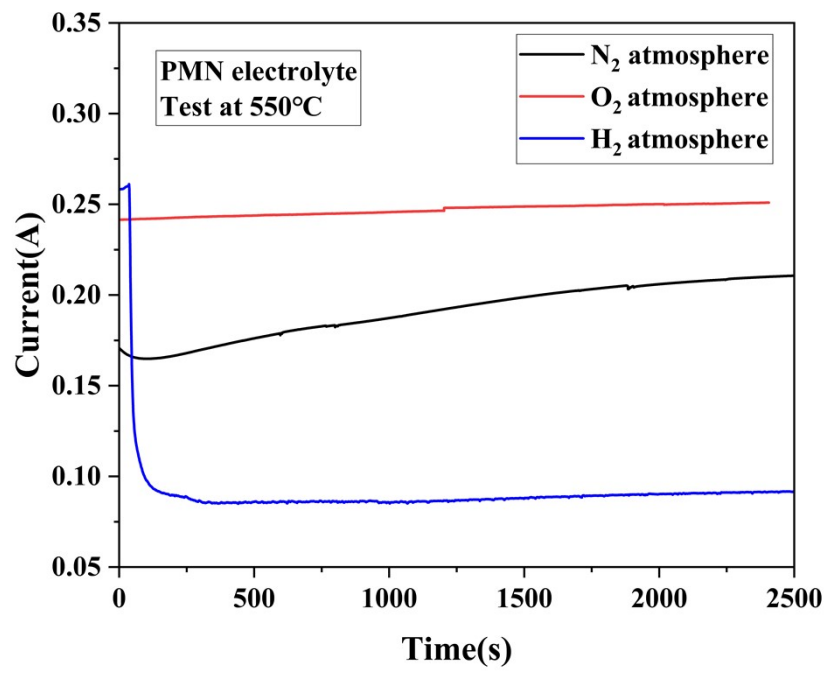
Supplementary Fig. 3 X-ray diffraction patterns (a) precursor, (b) lattice plane of PMN, (c) H_2 treated PMN, (d) high temperature treatment at 1300°C for 4h and (e) Williamson-Hall plot of PMN.



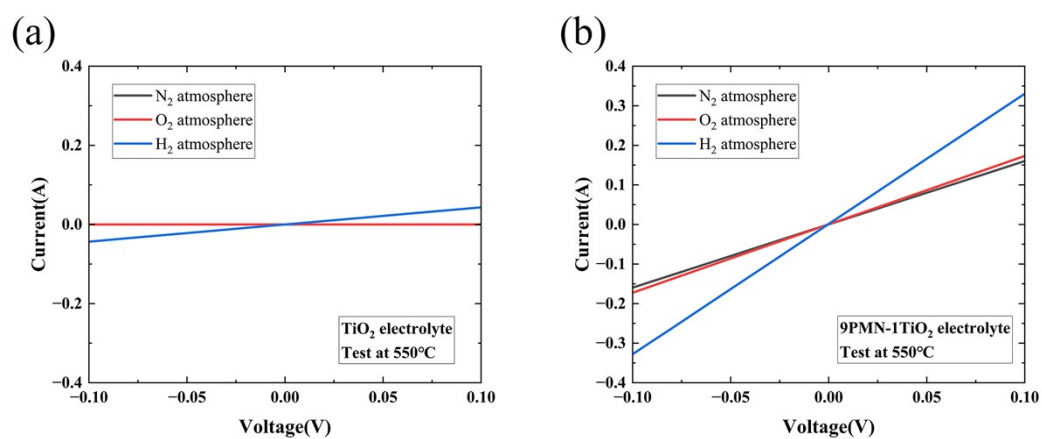
Supplementary Fig. 4 SEM images of (a) $\text{PrMn}_{0.5}\text{Ni}_{0.5}\text{O}_{3-\delta}$ and (b) TiO_2 .



Supplementary Fig. 5 TEM images of PMN-TiO₂ composite, showing the interfaces and interact between particles and grains.



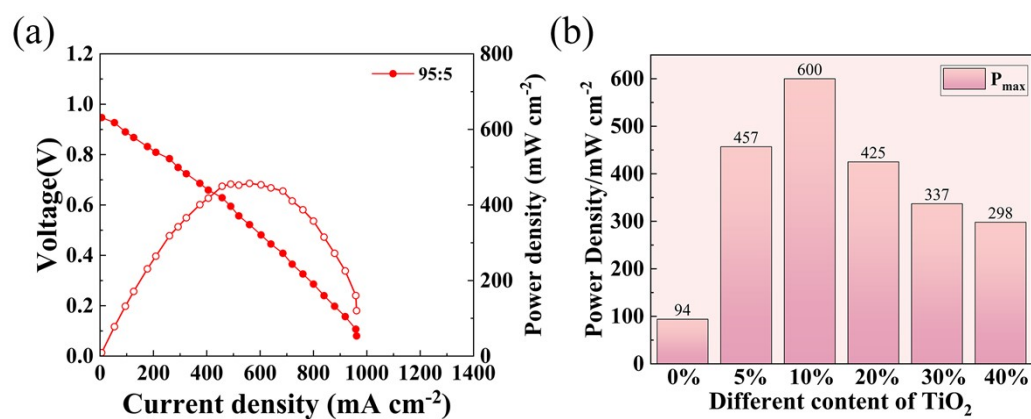
Supplementary Fig. 6 (a) Current-time test of PMN under 0.1V voltage bias.



Supplementary Fig. 7 Voltage sweep test after 30 mins atmosphere (N₂, O₂, H₂) treatment of (a)TiO₂ and (b)9PMN-1TiO₂ composite electrolyte from -0.1 V to 0.1 V. Sample coated with Pt on each side for test in O₂ and H₂ atmosphere and coated with Ag on each side for test in N₂ atmosphere.

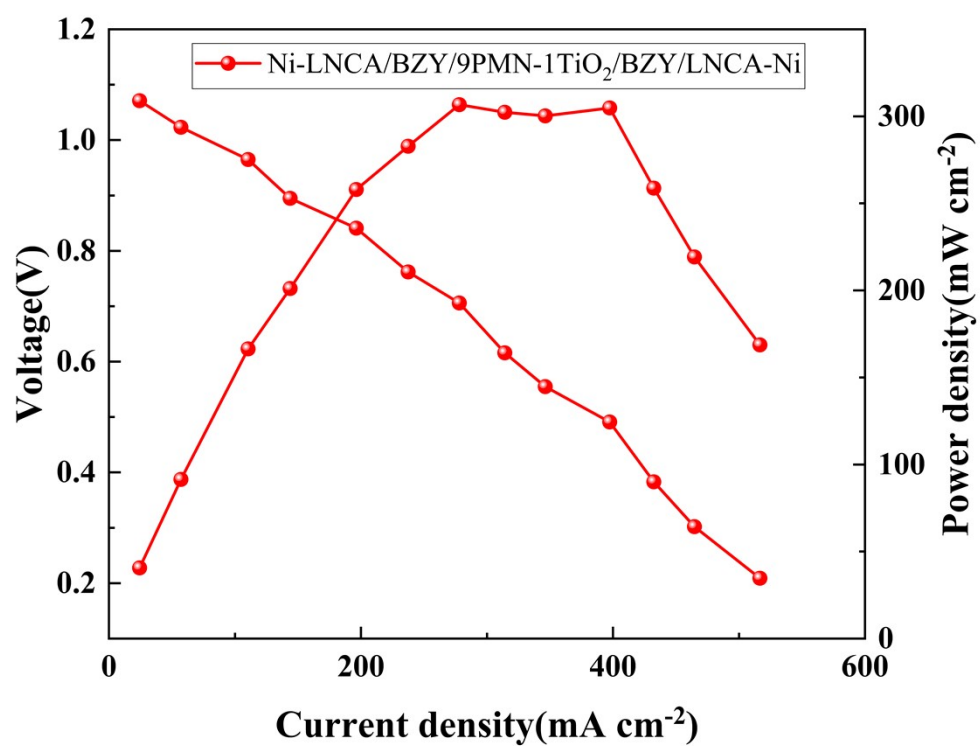


Supplementary Fig. 8 Thickness of PMN, TiO₂ and 9PMN-1TiO₂ electrolyte in DC test.

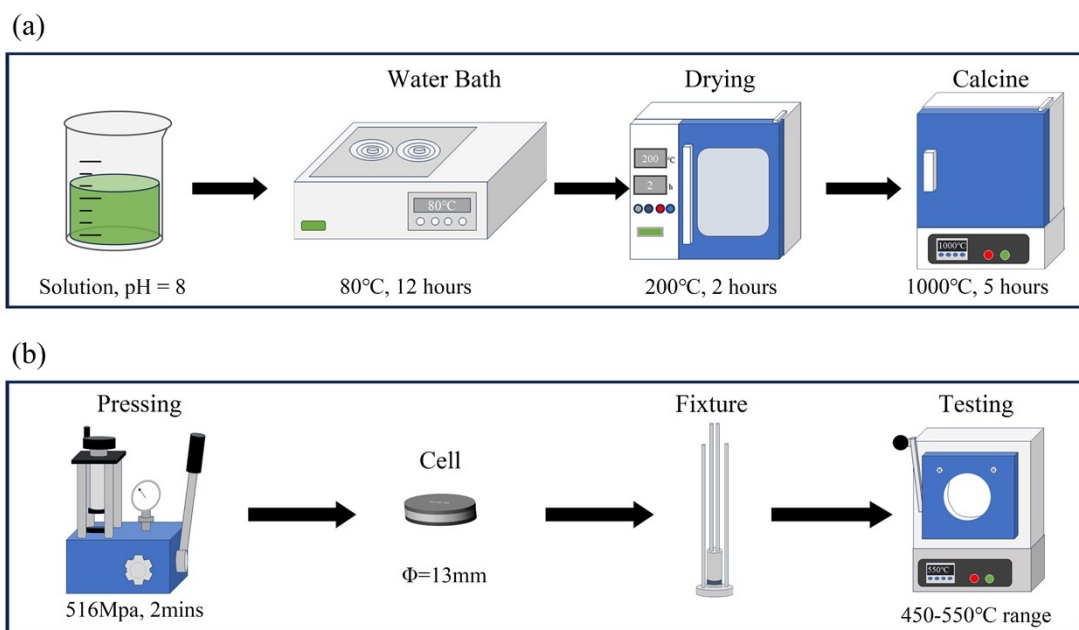


Supplementary Fig. 9 (a) I-V-P result of 95:5 composite materials electrolyte applying in SOFC.

(b) Maximum power density of different content of TiO_2 .



Supplementary Fig. 10 I-V-P result of Ni/LNCA/BZY/9PMN-1TiO₂/BZY/LNCA-Ni fuel cell at 550°C



Supplementary Fig. 11 (a) Sythesis process of PMN powder and (b) fuel cells fabrications, assembling and test.

Supplementary Table 1

The Metal-Oxygen(M-O) bonds distance of PrNiO_3 , $\text{PrMn}_{0.5}\text{Ni}_{0.5}\text{O}_{3-\delta}$ and PMN-TiO₂ heterostructure, where No.1 to No.6 stand for different M-O bond in octahedron.

PrNiO₃

Distance (Å)	No.1	No.2	No.3	No.4	No.5	No.6
Ni-O	1.94371	1.94371	1.95433	1.95433	1.9426	1.9426
Ni-O Octahedron Distortion factor(Δ): 0.0014%						

PrMnO₃

Distance (Å)	No.1	No.2	No.3	No.4	No.5	No.6
Mn-O	1.9927	1.9927	1.9752	1.9752	2.0279	2.0279
Mn-O Octahedron Distortion factor(Δ): 0.0121%						

PrMn_{0.5}Ni_{0.5}O_{3-δ}

Distance (Å)	No.1	No.2	No.3	No.4	No.5	No.6
Ni-O	2.04781	2.04781	2.07347	2.07347	1.98959	1.98959
Ni-O Octahedron Distortion factor(Δ): 0.0605%						

Distance (Å)	No.1	No.2	No.3	No.4	No.5	No.6
Mn-O	1.98843	1.98843	1.94492	1.94492	1.94045	1.94045
Mn-O Octahedron Distortion factor(Δ): 0.0239%						

PMN-TiO₂ Heterostructure

Distance (Å)	No.1	No.2	No.3	No.4	No.5	No.6
Ni-O	2.07151	2.04045	2.04764	2.14841	2.0984	2.12402
Ni-O Octahedron Distortion factor(Δ): 0.0735%						

Distance (Å)	No.1	No.2	No.3	No.4	No.5	No.6
Mn-O	1.88058	1.94257	1.95253	1.96495	2.35328	2.24927
Mn-O Octahedron Distortion factor(Δ): 1.5260%						

Supplementary Table 2Conductivity results by DC test of PMN, TiO₂, 9PMN-1TiO₂ electrolytes.

Electrolytes	Thickness/cm	Conductance value/S			Conductivity/S cm ⁻¹			
		N ₂	O ₂	H ₂	$\sigma_{e^{-}}$	$\sigma_{O^{2-}}$	$\sigma_{H^{+}}$	σ_{tot}
PMN	0.0983	2.105	2.509	9.16	0.122	0.263	0.018	0.281
TiO ₂	0.1276	2.85E-06	2.16E-06	4.34E-01	2.15E-07	2.15E-07	8.64E-02	8.64E-02
9PMN- 1TiO ₂ Composite	0.0815	1.696	1.722	3.249	0.082	0.138	0.332	0.470