## **Supplementary Information**

## Flavin-based Probe for Real-Time Monitoring of Hypochlorous Acid Dynamics in Live Cells

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## 1. Photophysical studies



**Figure S1**: UV-visible spectra of Propyl flavin (PFl), TPA, and **TPA-vinylene-Flavin** (**TVF**) at 19 μM concentration in methanol.

$\lambda_{abs}^{a}$ a	$\lambda_{em}^{a}$	Stokes shift <sup>b</sup>	ε <sup>c</sup>	${\pmb \Phi}_{f}{}^{d}$	$\tau^{e}$
506	656	4519	6990	0.09	0.96
493	636	4561	7440	0.014	1.00
525	-	-	7490	-	-
509	-	-	8100	-	-
499	-	-	6560	-	-
531	-	-	4550	-	-
	λ <sub>abs</sub> <sup>a</sup> 506 493 525 509 499 531	$\lambda_{abs}^{a}$ $\lambda_{em}^{a}$ 506         656           493         636           525         -           509         -           499         -           531         -	λ <sub>abs</sub> <sup>a</sup> λ <sub>em</sub> <sup>a</sup> Stokes shift <sup>b</sup> 506         656         4519           493         636         4561           525         -         -           509         -         -           499         -         -           531         -         -	$\lambda_{abs}^{a}$ $\lambda_{em}^{a}$ Stokes shiftb $\epsilon^{c}$ 50665645196990493636456174405257490509-8100499-6560531-4550	$\lambda_{abs}^{a}$ $\lambda_{em}^{a}$ Stokes shiftb $\epsilon^{c}$ $\Phi_{f}^{d}$ 506656451969900.09493636456174400.0145257490-509-8100-499-6560-531-4550-

Table S1. Summary of photophysical properties of TVF in different polarities of solvent.

<sup>a</sup>nm <sup>b</sup>cm<sup>-1</sup> <sup>c</sup>Molar extinction coefficient (M<sup>-1</sup>cm<sup>-1</sup>) <sup>d</sup>Quantum yield (reference standard: Ru(bpy)<sup>3+</sup> in Acetonitrile  $\lambda_{exc}$  = 450 nm and  $\Phi$  = 0.094) <sup>e</sup>Average lifetime (ns).



**Figure S2**: Calibration curve for measurement of the extinction coefficient of **TVF** in toluene, dioxane, DCM, Methanol, DMSO, and Water at each solvent absorption maxima.



**Figure S3**: Fluorescence lifetime spectra of **TVF** (Bi-exponential decay) in (a) Toluene and (b) Dioxane under air.

Table S2: Fluorescence lifetime measurement parameter of TVF.

Solvent	T1 (ns)	T2 (ns)	B1	B2	Chi.sq.
Toluene	0.32	2.6	0.17	0.009	1.07
Dioxane	0.04	2.6	2.16	0.02	1.12



Figure S4: Fluorescence spectra of TVF (62  $\mu$ M) with the addition of different concentrations of HOCl (0–6 eq) in methanol.  $\lambda_{ex} = 511$  nm.



**Figure S5**: (a) UV-Visible and (b),(c) Fluorescence spectra of TPA-Flavin (62  $\mu$ M) and TPA-Flavin incubated with 10 equivalents of HOCl in methanol. Fluorescence spectra were recorded with excitation at (b) 435 nm and (c) 491 nm.



Figure S6: Mass spectrum peaks of the product of TVF and ClO<sup>-</sup>. (a) and (b) is the different zoomed region of the mass spectrum



**Figure S7:** (a) UV-visible and (b) Fluorescence spectra of TVF recorded in different viscosities solvents such as MeOH, PEG400, and addition of HOCl (10 eq) in PEG400.

# 2. Table for the comparison of TVF with published probes for HOCl sensing

Probe	Emission wavelengt h	Target Localization	Response type	<b>LOD</b> (nM)	Reference s
	520 nm	Non-specific	Ratiometri c	500	1
	774 nm	Non-specific	ON-OFF	700	2
	515 nm	Non-specific	Ratiometri c	53	3
	435 nM	Mitochondri a	Ratiometri C	27	4
	511 nm	Lysosomes	Ratiometri c	10.6	5
	455 nm	Mitochondri a	Ratiometri c	182	6

Table S3: The comparison of TVF with published probes for HOCl sensing.

	530 nm	Non-specific	OFF-ON	68	7
о о о он , , , , , , , , , , , , , , , , , , ,	435 nm and 525 nm	Mitochondri a	OFF-ON	12	8
Et <sub>2</sub> N C C N	490 nm	Non-specific	OFF-ON	190	9
	520 nm	Non-specific	OFF-ON	8.3	10
	535 nm	Non-specific	Ratiometri c	738	11
	507 nm	Mitochondri a	OFF-ON	360	This work

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## 3. List of NMR



Figure S8. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) of compound 3.



Figure S9. <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) of compound 3.



Figure S10. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) of compound TVF.



Figure S11. <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ ) of compound TVF.

## 4. List of HRMS



Figure S12. HRMS spectra of compound 3.



Figure S13. HRMS spectra of compound TVF.