# Pressure and Temperature Phase Diagram of $\mathrm{CsCaCl}_{3}$ 

## SUPPLEMENTARY INFORMATION

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Table S1: Derived structural parameters for a perovskite in both cubic ( $P \overline{3} \mathrm{~m}$ ) and tetragonal $(I 4 / \mathrm{mcm})$ symmetry, where $(\tan (\phi)=4 u)$ and $u$ is the dispalcement of the x and y co-ordinate of the Cl atom in the tetragonal phase (after K. S. Knight et al, Physics and Chemistry of Minerals 41, 461 (2014))
Structural $\operatorname{Pm} \overline{3} m \quad I 4 / \mathrm{mcm}$

Parameter

| $\mathrm{A}-\mathrm{X}(1)$ | $\frac{a}{\sqrt{2}}$ | $\frac{a}{2}$ |
| :---: | :---: | :---: |
| $\mathrm{~A}-\mathrm{X}(2)$ | - | $\frac{1}{4}\left[2 a^{2}\left(1-2 \phi+\phi^{2}-\frac{2}{3} \phi^{3}\right)+c^{2}\right]^{0.5}$ |
| $\mathrm{~A}-\mathrm{X}(2)$ | - | $\frac{1}{4}\left[2 a^{2}\left(1+2 \phi+\phi^{2}+\frac{2}{3} \phi^{3}\right)+c^{2}\right]^{0.5}$ |
| $\mathrm{~B}-\mathrm{X}(1)$ | $\frac{a}{2}$ | $\frac{c}{4}$ |
| $\mathrm{~B}-\mathrm{X}(2)$ | - | $\frac{a}{2 \sqrt{2}}\left[1+\frac{1}{2} \phi^{2}+\frac{5}{24} \phi^{4}\right]$ |
| $\mathrm{AX}_{n}$ volume | $\frac{5 a^{3}}{6}$ | $\frac{a^{2} c}{24}\left[3-\phi-\frac{1}{3} \phi^{2}-\frac{2}{15} \phi^{5}-\frac{17}{315} \phi^{7}\right]$ |
| $\mathrm{BX}_{6}$ volume | $\frac{a^{3}}{6}$ | $\frac{a^{2} c}{24}\left[1+\phi^{2}+\frac{2}{3} \phi^{4}+\frac{17}{45} \phi^{6}+\frac{62}{315} \phi^{8}\right]$ |

$$
l_{p c}(T)=l_{p c 0}+\frac{A_{p c}}{\exp \left(\frac{B_{p c}}{T}\right)-1}
$$

Einstein internal energy model fitted to the pseudo-cubic lattice parameter $l_{p c}$ with temperature $(T)$ using an Einstein temperature $\left(B_{p c}\right)$. For these models we find $a_{p c 0}=5.3650 \AA, A a_{p c}=0.0466 \AA, B a_{p c}=129.5 \mathrm{~K}$ and $c_{p c 0}=5.3981 \AA, A c_{p c}=-0.0431 \AA, B c_{p c}=140.7 \mathrm{~K}$.

$$
\begin{aligned}
e_{a} & =\left(2 e_{1}+e_{3}\right) \\
e_{t z} & =\frac{2}{\sqrt{3}}\left(e_{3}-e_{1}\right) \\
e_{1} & =\frac{a / \sqrt{2}-a_{0}}{a_{0}} \\
e_{3} & =\frac{c / 2-a_{0}}{a_{0}}
\end{aligned}
$$

Equations for the symmetry-adapted tetragonal strain, $e_{t z}$, volumetric strain, $e_{a}$, and natural linear strain components $e_{1}$ and $e_{3} . a_{0}$ is calculated from the pseudo cubic unit-cell volume of the tetragonal phase $\left(\sqrt{[3]} a_{p c}^{2} c_{p c}\right)$ at each temperature. As $e_{1}<0, e_{3}>0$ and $e_{3} \simeq 2\left|e_{1}\right|$, by definition $e_{a}=0$ and $e_{t z}>0$.

$$
\begin{gathered}
\cos \left(\alpha_{p c}\right)-\cos \left(\alpha_{p c}\right)_{c a l c}=\left(\frac{2}{3}\right) \zeta\left[1-\left(\frac{2}{3}\right) \sin ^{2} \omega\right]^{-1} \\
\cos \left(\alpha_{p c}\right)_{c a l c}=\frac{\sin ^{2} \omega}{3-2 \sin ^{2} \omega}
\end{gathered}
$$

Formulation for the octahedral strain $\zeta$ (equal to 0 in the undistorted, cubic phase). $\omega$ is the octahedral tilt angle in the rhombohedral phase.

$$
\begin{gathered}
V(T)=V_{0}+\frac{9 N \gamma_{1} k_{B} z T}{B_{0}}\left(\frac{T}{\Theta_{D 1}}\right)^{3} \int_{0}^{\frac{\Theta_{D 1}}{T}} \frac{x^{3} d x}{e^{x}-1}+\frac{9 N \gamma_{2} k_{B}(1-z) T}{B_{0}}\left(\frac{T}{\Theta_{D 2}}\right)^{3} \int_{0}^{\frac{\Theta_{D 2}}{T}} \frac{x^{3} d x}{e^{x}-1} \\
c_{V}(T)=9 N k_{B} z\left(\frac{T}{\Theta_{D 1}}\right)^{3} \int_{0}^{\frac{\Theta_{D 1}}{T}} \frac{x^{4} e^{x} d x}{\left(e^{x}-1\right)^{2}}+9 N k_{B}(1-z)\left(\frac{T}{\Theta_{D 2}}\right)^{3} \int_{0}^{\frac{\Theta_{D 2}}{T}} \frac{x^{4} e^{x} d x}{\left(e^{x}-1\right)^{2}}
\end{gathered}
$$

Formulation of unit-cell volume and isochoric heat capacity as a function of temperature according to the two-term Debye approximation. $V_{0}$ is the unit-cell volume at zero temperature, N is the number of atoms, $\gamma_{n}$ are the Grüneisen parameters, $k_{B}$ is the Boltzmann constant, $B_{0}$ is the isothermal bulk modulus, $z$ is the mixing coefficient for the two-term model, and $\theta_{D n}$ are the Debye temperatures. In the present study the fitting was performed using the GlobalAnalysis package in IgorPro (©WaveMetrics), allowing for the global minimisation of both integrals against two physical datasets with shared fitting parameters.

$$
M_{i}=\left(\frac{R_{i} N_{i}}{B}\right) \exp \left[\left(\frac{R_{0}-R_{i}}{B}\right)\right]
$$

Formulation of $M_{i}$, the total estimated variation of bond valence in a polyhedral site due to the change of average bond distance. $R_{i}$ is the average measured bond distance in the polyhedra, $N_{i}$ is the coordination number in the polyhedra, $R_{0}$ is a constant for a particular atom pair and $B$ is a universal constant (0.37), the values of which are given and defined in the work of Brown and Altermatt (Acta Cryst. B41, 244-247 (1985)) The relative values of $M$ for the A site $\left(M_{A}\right)$ and $\left(M_{B}\right)$ give an indication of the relative compressibilities (full derivation and description is given in Zhao et al Acta Cryst B 60, 3, 263-272 (2004). For $M_{A} / M_{B} \geq 1$ the material will become more symmetric upon compression, $M_{A} / M_{B} \approx 1$ the distortion does not change with pressure and $M_{A} / M_{B} \leq 1$ indicates that the distortion will increase with increasing pressure.

Table S2: Determined structural parameters of $\mathrm{CsCaCl}_{3}$ with temperature. For the tetragonal structure the space-group is $I 4 / m c m$, $\mathrm{Cs} 4 b, \mathrm{Ca} 4 c, \mathrm{Cl}(1) 4 a, \mathrm{Cl}(2) 8 h \frac{1}{4}+u, \frac{3}{4}+u$, 0 . For $\operatorname{Pm} \overline{3} m$; Wyckoff positions: Cs $1 b, \mathrm{Ca} 1 a, \mathrm{Cl} 3 d$ For further details see main text

| Temperature (K) | Phase | $\mathrm{a}(\AA)$ | $\mathrm{c}(\AA \AA)$ | $\mathrm{V}\left(\AA^{3}\right)$ | u |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 5 | Tetragonal | $7.7880(3)$ | $10.7952(5)$ | $621.57(7)$ | 0.0233 |
| 25 | Tetragonal | $7.5864(3)$ | $10.7968(5)$ | $621.40(7)$ | 0.02387 |
| 45 | Tetragonal | $7.5920(3)$ | $10.7923(5)$ | $622.06(7)$ | 0.02197 |
| 65 | Tetragonal | $7.5970(3)$ | $10.7846(6)$ | $622.42(7)$ | 0.0193 |
| 70 | Tetragonal | $7.6998(3)$ | $10.7825(6)$ | $622.77(6)$ | 0.01777 |
| 80 | Tetragonal | $7.6033(4)$ | $10.7787(8)$ | $623.12(6)$ | 0.01595 |
| 90 | Tetragonal | $7.6079(6)$ | $10.7732(10)$ | $623.56(6)$ | 0.01323 |
| 100 | Cubic | $5.3836(2)$ | - | $156.030(14)$ | - |
| 120 | Cubic | $5.3848(2)$ | - | $156.138(15)$ | - |
| 140 | Cubic | $5.3869(2)$ | - | $156.323(17)$ | - |
| 160 | Cubic | $5.3888(2)$ | - | $156.49(2)$ | - |
| 180 | Cubic | $5.3902(2)$ | - | $156.61(2)$ | - |
| 200 | Cubic | $5.3923(3)$ | - | $156.79) 2)$ | - |
| 220 | Cubic | $5.3941(3)$ | - | $156.95(3)$ | - |
| 240 | Cubic | $5.3962(3)$ | - | $157.13(3)$ | - |
| 260 | Cubic | $5.3991(3)$ | - | $157.38(3)$ | - |
| 280 | Cubic | $5.4004(3)$ | - | $157.50(3)$ | - |



Figure S1: Isobaric heat capacity as a function of temperature, colelcted on warming from 4 K . There is a clear discontinuity at the expected transition temperature between 90 and 100 K .

