

Supporting Information

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1. Computational details

We have extensively explored the structures of $\text{SrB}_x\text{C}_{10-x}$ ($x=1-9$) at pressures of 0-20 GPa. The formation enthalpy is defined as $\Delta H/11 = H(\text{SrB}_x\text{C}_{10-x}) - H(\text{Sr}) - xH(\text{B}) - (10-x)H(\text{C})$ in unit of eV/atom. $H(\text{SrB}_x\text{C}_{10-x})$ is the enthalpy of $\text{SrB}_x\text{C}_{10-x}$. $H(\text{Sr})$, $H(\text{B})$ and $H(\text{C})$ are the enthalpies of elemental Sr, B and C solids, respectively. The $Fm-3m$ and $Im-3m$ phases of elemental Sr (phase transition pressure is 5.7 GPa), the $R-3m$ phase of elemental B and the $Fd-3m$ structure of elemental C are used in the calculation. In addition, the $C2/m$ and $P6_3/mmc$ SrB (phase transition pressure is 7.3 GPa), $Immm$ SrB_2 , $R-3m$ SrB_4 , $Pm-3m$ SrB_6 , $I4/mmm$ and $C2/m$ SrC_2 (phase transition pressure is 6.3 GPa) are used to calculate the synthesis pathways. All DFT-GGA calculations were performed for SrB_5C_5 , SrB_7C_3 , SrB_8C_2 , and SrB_9C compounds with atoms numbers of 11, 22, 44, and 44, respectively, which are primitive cells.

2. The COHP curves and ELF distributions.

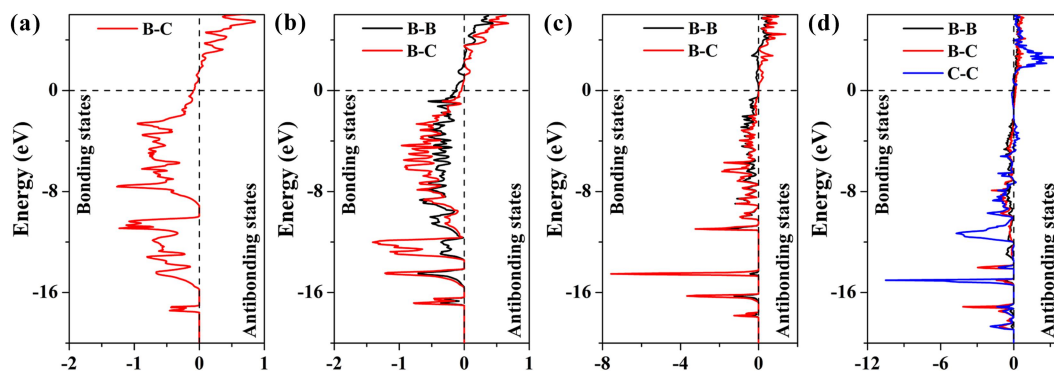


Fig. S1 The COHP curves at 0 GPa for (a) SrB_5C_5 , (b) SrB_7C_3 and (d) SrB_9C using PBE functional, and (c) SrB_8C_2 by HSE06 functional.

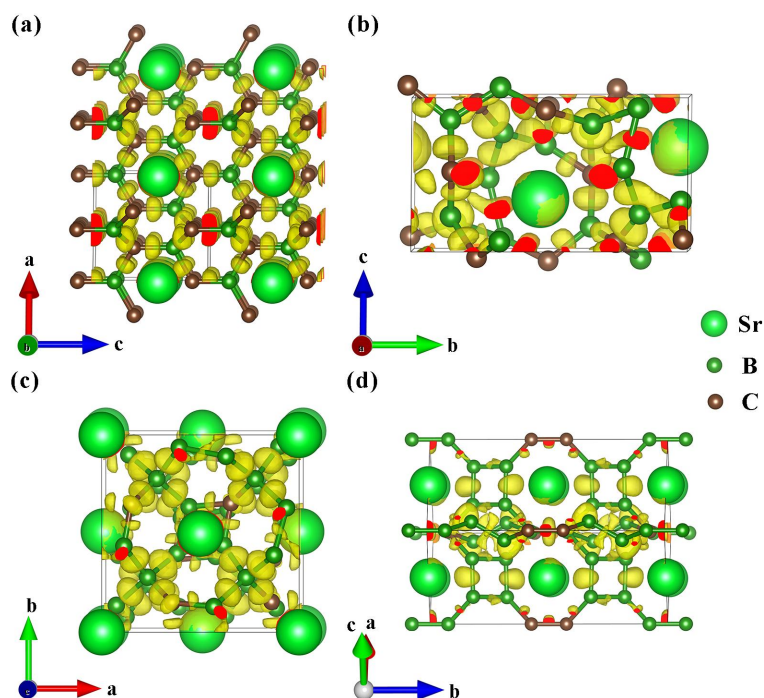


Fig. S2 The ELF distributions for (a) SrB_5C_5 , (b) SrB_7C_3 , (c) SrB_8C_2 and (d) SrB_9C at ambient pressure (The yellow isosurface represents an ELF value of 0.80).

3. The phonon dispersion curves.

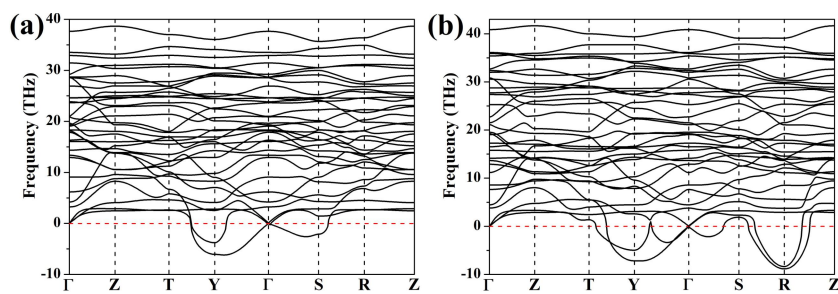


Fig. S3 The phonon dispersion curves for SrB₅C₅ at (a) 0 GPa and (b) 20 GPa.

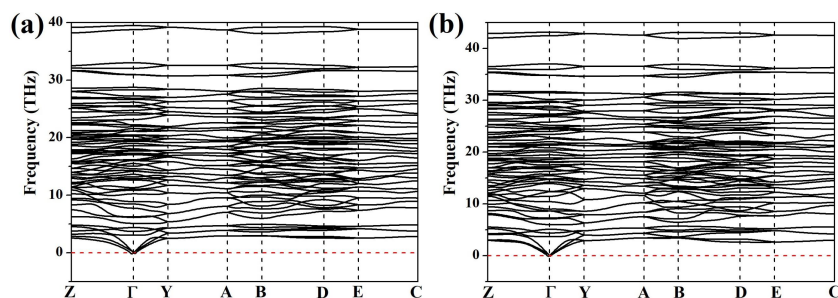


Fig. S4 The phonon dispersion curves for SrB₇C₃ at (a) 0 GPa and (b) 20 GPa.

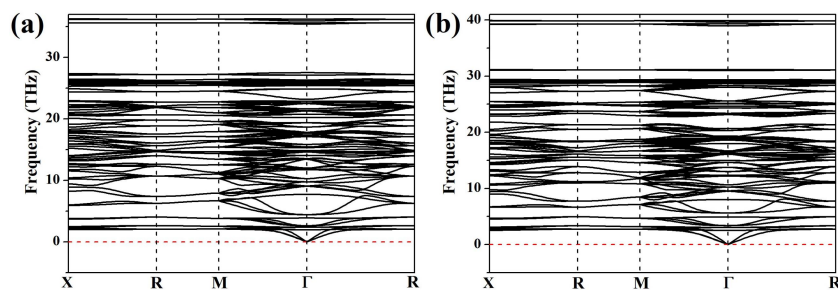


Fig. S5 The phonon dispersion curves for SrB₈C₂ at (a) 0 GPa and (b) 20 GPa.

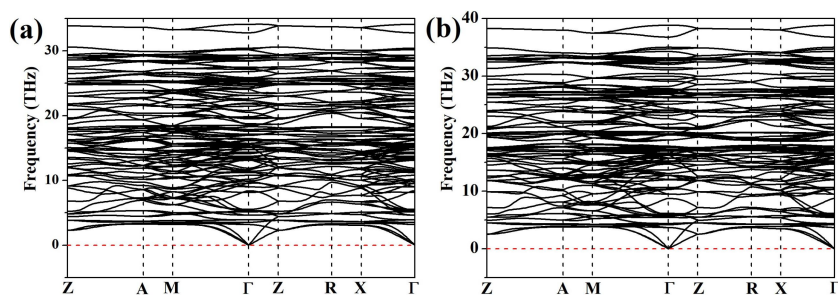


Fig. S6 The phonon dispersion curves for SrB₉C at (a) 0 GPa and (b) 20 GPa.

4. The electronic band structure, transport properties and ZT values for SrB_8C_2 at 20 GPa.

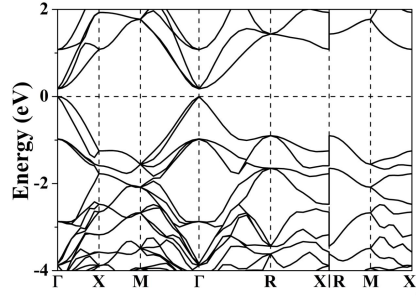


Fig. S7 The electronic band structure of SrB_8C_2 calculated by HSE06 functional at 20 GPa (the bandgap is 0.17 eV, and this value is used to calculate the transport properties).

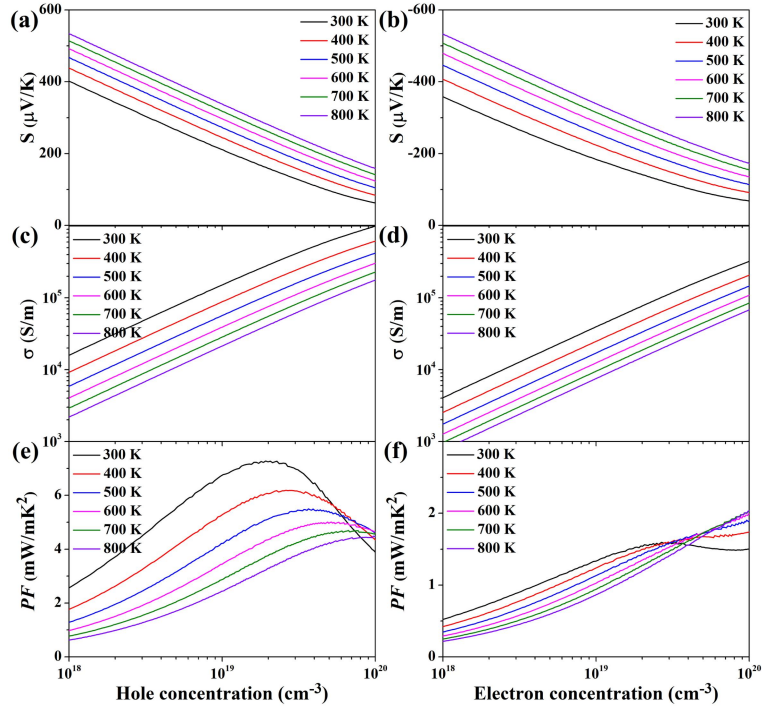


Fig. S8 The transport properties of SrB_8C_2 as a function of carrier concentration between 300 and 800 K at 20 GPa. (a, b) Seebeck coefficients, (c, d) electrical conductivity and (e, f) power factor.

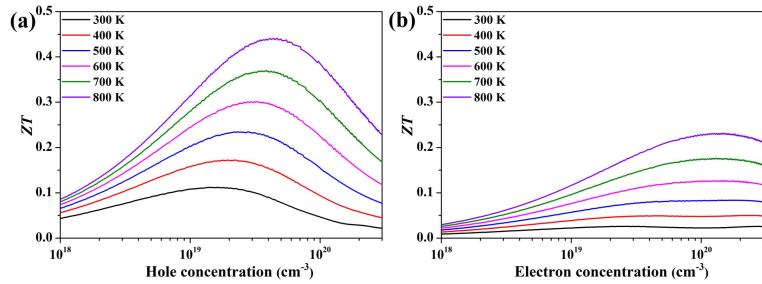


Fig. S9 The (a) p -type and (b) n -type ZT values for SrB_8C_2 between 300 and 800 K at 20 GPa.

5. The detailed structure information and the possible synthesis pathways for SrB₅C₅, SrB₇C₃, SrB₈C₂ and SrB₉C.

Table S1 Detailed crystal structure information for SrB₅C₅, SrB₇C₃, SrB₈C₂ and SrB₉C at ambient pressure.

	Space group	Volume per unit	Lattice parameters	Atomic positions
SrB ₅ C ₅	<i>Cmm2</i>	100.0 Å ³	$a = 8.109 \text{ \AA}$, $b = 5.173 \text{ \AA}$ $c = 4.766 \text{ \AA}$ $\alpha = \beta = \gamma = 90.0^\circ$	Sr 2 <i>a</i> (0,0,0.58356)
				B 4 <i>c</i> (0.25,0.25,0.19784)
				B 2 <i>b</i> (0,0.5,0.20458)
				B 4 <i>d</i> (0.67644,0,0.71311)
				C 4 <i>c</i> (0.25,0.25,0.8757)
				C 2 <i>b</i> (0,0.5,0.8801)
				C 4 <i>d</i> (0.67653,0,0.36699)
SrB ₇ C ₃	<i>P2₁</i>	95.3 Å ³	$a = 5.114 \text{ \AA}$ $b = 8.216 \text{ \AA}$ $c = 4.898 \text{ \AA}$ $\alpha = \gamma = 90.0^\circ$ $\beta = 112.12^\circ$	Sr 2 <i>a</i> (0.59367,0.46204,0.33407)
				B 2 <i>a</i> (0.79982,0.6514,0.82655)
				B 2 <i>a</i> (0.96201,0.31936,0.06623)
				B 2 <i>a</i> (0.88196,0.13853,0.85647)
				B 2 <i>a</i> (0.57058,0.13715,0.21969)
				B 2 <i>a</i> (0.61191,0.82162,0.23853)
				B 2 <i>a</i> (0.84074,0.97401,0.32316)
				B 2 <i>a</i> (0.91586,0.76771,0.54251)
				C 2 <i>a</i> (0.86218,0.14404,0.51885)
				C 2 <i>a</i> (0.35779,0.73361,0.04503)
C 2 <i>a</i> (0.99374,0.98445,0.08073)				
SrB ₈ C ₂	<i>Pa-3</i>	98.2 Å ³	$a = b = c = 7.325 \text{ \AA}$ $\alpha = \beta = \gamma = 90.0^\circ$	Sr 4 <i>a</i> (0,0,0)
				B 24 <i>d</i> (0.62818,0.60976,0.43442)
				B 8 <i>c</i> (0.77309,0.77309,0.77309)
				C 8 <i>c</i> (0.64678,0.64678,0.64678)
SrB ₉ C	<i>P4/nnc</i>	97.6 Å ³	$a = b = 5.981 \text{ \AA}$ $c = 10.912 \text{ \AA}$ $\alpha = \beta = \gamma = 90.0^\circ$	Sr 4 <i>c</i> (0,0.5,0)
				B 16 <i>k</i> (0.90284,0.80307,0.18249)
				B 16 <i>k</i> (0.40453,0.30215,0.82196)
				B 4 <i>e</i> (0,0,0.92363)
				C 4 <i>e</i> (0,0,0.57366)

Table S2 Formation enthalpies of SrB₅C₅, SrB₇C₃, SrB₈C₂ and SrB₉C within different synthesis pathways at 20 GPa.

	Synthesis pathways	ΔH with respect to diamond carbon (meV/atom)
SrB ₅ C ₅	Sr+5B+5C	-16
	SrB+4B+5C	-52
	SrB ₂ +3B+5C	-129
	SrB ₄ +B+5C	-197
	SrC ₂ +5B+3C	-74
	(SrB+SrC ₂ +9B+8C)/2	-127
	(SrB ₂ +SrC ₂ +8B+8C)/2	-203
	(SrB ₄ +SrC ₂ +6B+8C)/2	-271
(SrB ₆ +SrC ₂ +8B+8C)/2	-444	
SrB ₇ C ₃	Sr+7B+3C	-111
	SrB+6B+3C	43
	SrB ₂ +5B+3C	-34
	SrB ₄ +3B+3C	-102
	SrB ₆ +B+3C	-275
	SrC ₂ +7B+C	21
	(SrB+SrC ₂ +13B+4C)/2	64
	(SrB ₂ +SrC ₂ +12B+4C)/2	-13
(SrB ₄ +SrC ₂ +10B+4C)/2	-80	
(SrB ₆ +SrC ₂ +8B+4C)/2	-254	
SrB ₈ C ₂	Sr+8B+2C	-106
	SrB+7B+2C	37
	SrB ₂ +6B+2C	-39
	SrB ₄ +4B+2C	-107
	SrB ₆ +2B+2C	-28
	SrC ₂ +8B	15
	(SrB+SrC ₂ +15B+2C)/2	52

	$(\text{SrB}_2+\text{SrC}_2+14\text{B}+2\text{C})/2$	-24
	$(\text{SrB}_4+\text{SrC}_2+12\text{B}+2\text{C})/2$	-92
	$(\text{SrB}_6+\text{SrC}_2+10\text{B}+2\text{C})/2$	-265
	$\text{Sr}+9\text{B}+\text{C}$	-95
	$\text{SrB}+8\text{B}+\text{C}$	26
	$\text{SrB}_2+7\text{B}+\text{C}$	-50
	$\text{SrB}_4+5\text{B}+\text{C}$	-118
SrB_9C	$\text{SrB}_6+3\text{B}+\text{C}$	-291
	$(\text{SrB}+\text{SrC}_2+17\text{B})/2$	31
	$(\text{SrB}_2+\text{SrC}_2+16\text{B})/2$	46
	$(\text{SrB}_4+\text{SrC}_2+14\text{B})/2$	-114
	$(\text{SrB}_6+\text{SrC}_2+12\text{B})/2$	-286

6. The calculated elastic constants, moduli and Bader charges for SrB₇C₃, SrB₈C₂ and SrB₉C.

Table S3 The calculated elastic constants (C_{ij}), bulk modulus (B), shear modulus (G), Young's modulus (Y), Pugh ratio (G/B) and simulated hardness values (H_v) by employing three different empirical model for SrB₇C₃, SrB₈C₂ and SrB₉C compounds at 20 GPa. The units of C_{ij} , B , G , Y , and H_v are GPa.

	C_{11}	C_{22}	C_{33}	C_{44}	C_{55}	C_{66}	C_{12}	C_{13}	C_{23}	C_{15}
SrB ₇ C ₃	505.5	576.0	549.0	236.8	212.6	231.9	138.4	161.0	139.6	51.6
	C_{25}	C_{35}	C_{46}	B	G	Y	G/B	H_v^{Chen}	H_v^{Tian}	H_v^{Teter}
	8.9	-11.3	30.1	277.7	212.7	508.4	0.77	30.7	30.2	32.1
SrB ₈ C ₂	C_{11}	C_{44}	C_{12}	B	G	Y	G/B	H_v^{Chen}	H_v^{Tian}	H_v^{Teter}
	497.2	206.6	140.0	259.1	194.9	467.4	0.75	28.3	27.8	29.4
SrB ₉ C	C_{11}	C_{33}	C_{44}	C_{66}	C_{12}	C_{13}				
	466.7	655.6	216.2	219.2	165.8	106.9				
				B	G	Y	G/B	H_v^{Chen}	H_v^{Tian}	H_v^{Teter}
			259.7	207.7	491.9	0.80	31.9	31.2	31.4	

Table S4 The Bader charges for SrB₇C₃, SrB₈C₂ and SrB₉C compounds at ambient pressure.

	Atom	Atom numbers	Charge value (e)	δ (e)
SrB ₇ C ₃	Sr	2	8.57	+1.43
	B	14	2.25	+0.75
	C	6	6.22	-2.22
SrB ₈ C ₂	Sr	4	8.45	+1.55
	B	32	2.57	+0.43
	C	8	6.48	-2.47
SrB ₉ C	Sr	4	8.60	+1.40
	B	36	2.98	+0.02
	C	4	5.59	-1.59

7. The effective mass, elastic constant, DP constant and relaxation time for SrB₈C₂.

Table S5 The corresponding effective mass m^* , elastic constant C and DP constant E at 0 GPa are listed.

Carrier type	$ m^* /m_0$	C (eV/Å ³)	$ E $ (eV per strain)
Electron	0.76	10.6	23.94
Hole	1.66	10.6	25.55

Table S6 The carrier relaxation time for the SrB₈C₂ at 0 GPa.

	300 K	400 K	500 K	600 K	700 K	800 K
τ (10 ⁻¹⁴ s)						
CBM	2.97	1.93	1.38	1.05	0.83	0.68
VBM	8.41	5.46	3.91	2.98	2.36	1.93

Table S7 The corresponding effective mass m^* , elastic constant C and DP constant E at 20 GPa are listed.

Carrier type	$ m^* /m_0$	C (eV/Å ³)	$ E $ (eV per strain)
Electron	0.79	14.6	25.38
Hole	1.52	14.6	27.70

Table S8 The carrier relaxation time for the SrB₈C₂ at 20 GPa.

	300 K	400 K	500 K	600 K	700 K	800 K
τ (10 ⁻¹⁴ s)						
CBM	4.17	2.71	1.94	1.47	1.17	0.96
VBM	9.36	6.08	4.35	3.31	2.63	2.15