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1	Improved the uniformity of the inkjet-printed polymer film in the
2	bank by Marangoni Flow and Contact Line Sliding
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## 1 Supporting information

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2 Figure S1. The 3d optical profile of the patterned bank.

## 1 Ink Formulation and Drop Behavior

The droplet behavior of the ink has a significant impact on the ability to 2 form uniform and clear patterned films. Different co-solvents with 3 different surface tension are used to observe differences in the drying 4 process. There are three solvent compositions in the studies. Toluene was 5 chosen as the main solvent for its good solubility of F8BT. The addition of 6 different co-solvents substantially increases the viscosity of the ink and 7 enhances the printability of the ink. The priority is to ensure that ink 8 formulations can drop a stable droplet without satellite droplets. Table S1 9 shows the optimized print voltages and pulse lengths. The print voltages of 10 different inks are similar. A stable single printable droplet can be obtained 11 by slightly optimizing the printing voltage and pulse lengths. Figure S2 12 shows the distance between the head droplet and the tail droplet from the 13 printing nozzle. The red line is the distance between the tail and the nozzle, 14 and the black line is the distance between the main droplet and the nozzle 15 over time. The inset shows the entire flight of the droplet with different ink 16 formulations. Unlike the droplet behavior of small molecules, even at high 17 pulses the polymer droplet trailing only briefly separates from the main 18 droplet and forms the satellite droplets. The main droplet and satellite 19 droplets rapidly fuse to form a stable single droplet. Therefore, the F8BT 20 droplet without a long tail and satellite droplets is obtained by optimizing 21 the printing voltage of different inks. The velocity and diameter of the 22

- flying droplets are shown in Table S1. The final droplets with stable flight
   velocity and droplet diameter were obtained. Also, the flight velocity is 1
   m/s < v < 2 m/s, which is suitable for printing requirements.</li>
- 4 **Table S1.** Printed voltage, pulse length, pressure, velocity and diameter of
- 5 the different ink.

Ink formulation	Voltage (V)	Pulse length	Pressur e (mbar)	Droplet velocity (m/s)	Droplet diameter (µm)
TOL/1-CN=5/5	72	35.8	-8	1.64	82
TOL/CHB=5/5	72	35.9	-8	1.45	82
TOL/BB=5/5	76	35.9	-8	1.48	82





8 Figure S2. Inkjetted drop formation of (a) TOL/1-CN (b) TOL/CHB (c)
9 TOL/BB (insets show transient behavior of the drops ).

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## 11 Wettability and the drying process of the droplet on the substrate

12 The wettability and the drying process of the droplet on the substrate are 13 influenced by the characteristics of the substrate and the composition of the 14 solution. Conditions such as the physical properties of the solvent, particle 15 interactions of the solute, additives, and substrate modification all affect the contact line of the droplet on the substrate and evaporation kinetics. As
the droplets dry in the bank, the droplet contacts and interacts with the
photoresist bank and substrate. Therefore, the interaction of different
droplets and the evaporation process on the ITO substrate and photoresist
substrate were studied. The ITO substrate and photoresist substrate were
surface-modified with oxygen plasma and CF<sub>4</sub> plasma, the same process
as the patterned substrate.

The fluorescence microscopy images of the contact angle and the 8 evaporation process of droplets with different ink formulations on the ITO 9 substrate are shown in Figure S3. The contact angles of the drop with 10 TOL/BB, TOL/CHB, and TOL/1-CN ink on the substrate are 18.0°, 16.0 11 °, and 16.9°. The contact angle shows that different inks are fully wetted 12 on the substrate. Also, there is no significant change in the diameter of the 13 droplet during the whole drying process. This means that each droplet is in 14 a constant contact area drying mode. The droplet spreads completely and 15 the contact area does not change, without movement of the contact line. 16 The drying process of different ink droplets is all from the outside to the 17 inside. However, the final drying profile is not the same. There is no 18 obvious solute flow within the droplet with TOL/BB and TOL/CHB and 19 the brightness of the final dried pattern is uniform in every place. However, 20 for the droplet with TOL/1-CN ink, it is obvious that there is more solute 21 in the edge than in the center. This is mainly because the surface tension of 22

1-CN is larger than CHB and BB and the difference-driven Marangoni 1 flow on the surface is stronger, thus forming the coffee ring finally. 2 Meanwhile, the contact angle and drying process of droplets were tested 3 on the CF<sub>4</sub> etched photoresist substrate, as shown in Figure S4. The contact 4 angles of the drop with TOL/BB, TOL/CHB, and TOL/1-CN ink on plasma 5 etched photoresist are 50.9 °, 60.4 °, and 60.1 °. The droplets of all inks did 6 not undergo contact line slide and the contact area did not change. The 7 contact angles of TOL/CHB and TOL/1-CN ink are similar. Meanwhile, 8 the final film shows a slight coffee ring effect. However, the interface 9 between the drying droplet and the photoresist in the bank is vertical, which 10 could be affected by gravity. Therefore, the evaporation of the droplets in 11 12 physically restricted spaces may be different.



14 Figure S3. The contact angle of the (a) TOL/BB, (b) TOL/CHB, (c)
15 TOL/CN drop on the ITO/PEDOT:PSS substrate and the fluorescence

1 images of the drying process.



4 Figure S4. The contact angle of the (a) TOL/BB, (b) TOL/CHB, (c)
5 TOL/CN drop on the plasma etched ITO/photoresist substrate and the
6 fluorescence images of the drying process.



2 Figure S5. Time evolution of the droplet surface profile in the bank with

3 (a) TOL/BB, (b) TOL/CHB, (c) TOL/1-CN.











Solvent	Α	В	С	P (120 °C)
TOL	7.137	1457.287	231.827	131.33
1-CN	7.696	1953.172	184.470	2.54
CHB	7.145	1932.038	193.790	1.29
BB	7.596	2225.320	221.955	1.63

Table S2. The empirical constants number and the calculated saturation
 vapor pressures of TOL, 1-CN, CHB and BB.