

Supporting Information

**Bright Self-Trapped Exciton Emission in Alkali Iodide Nanocrystals
via Sn(II)-Doping**

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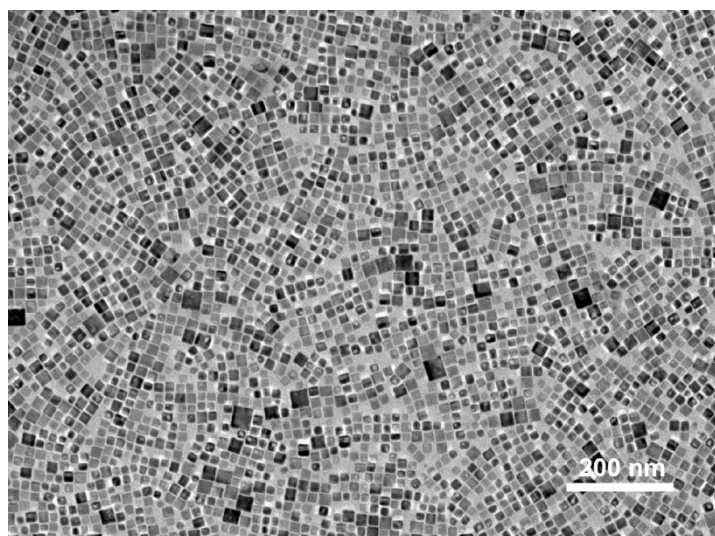


Figure S1 TEM image of RbI NCs

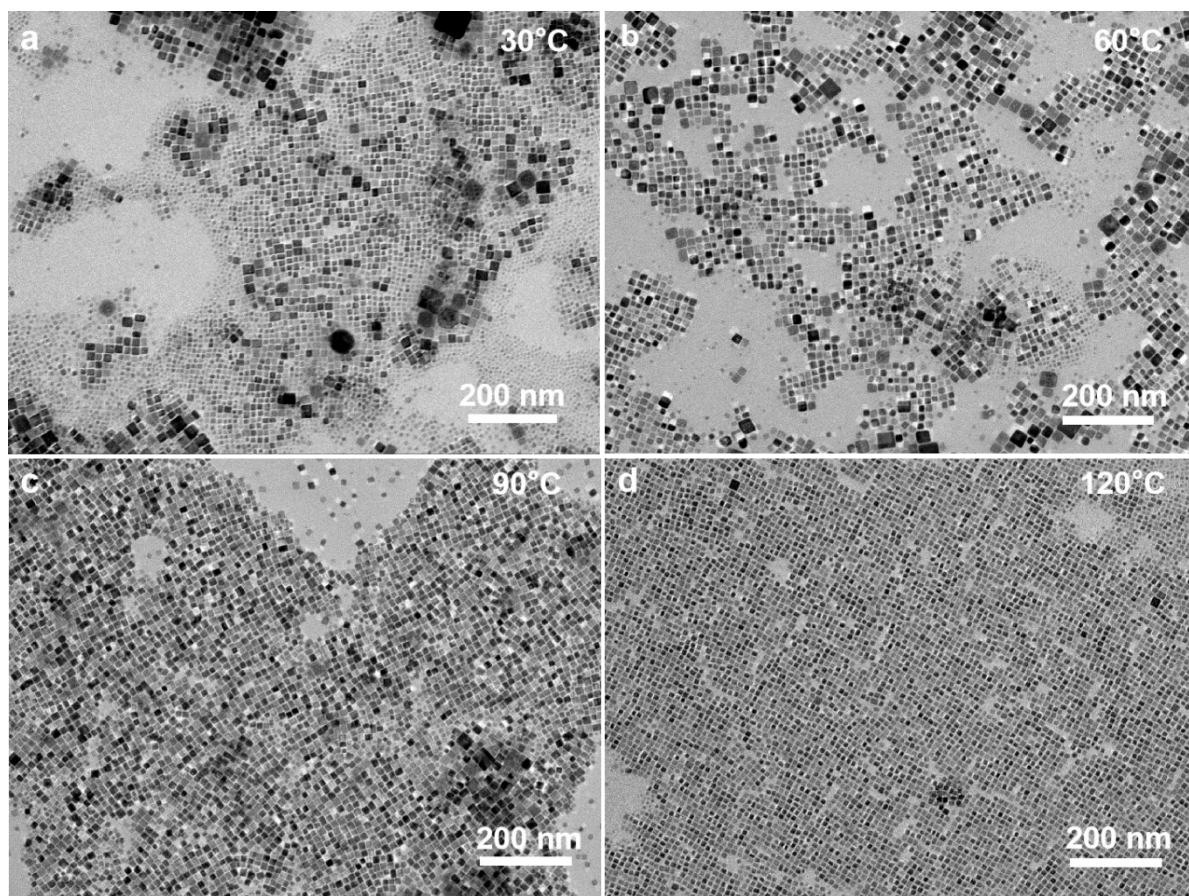


Figure S2 TEM images of Sn^{2+} -doped RbI NCs synthesized at different injection temperatures.

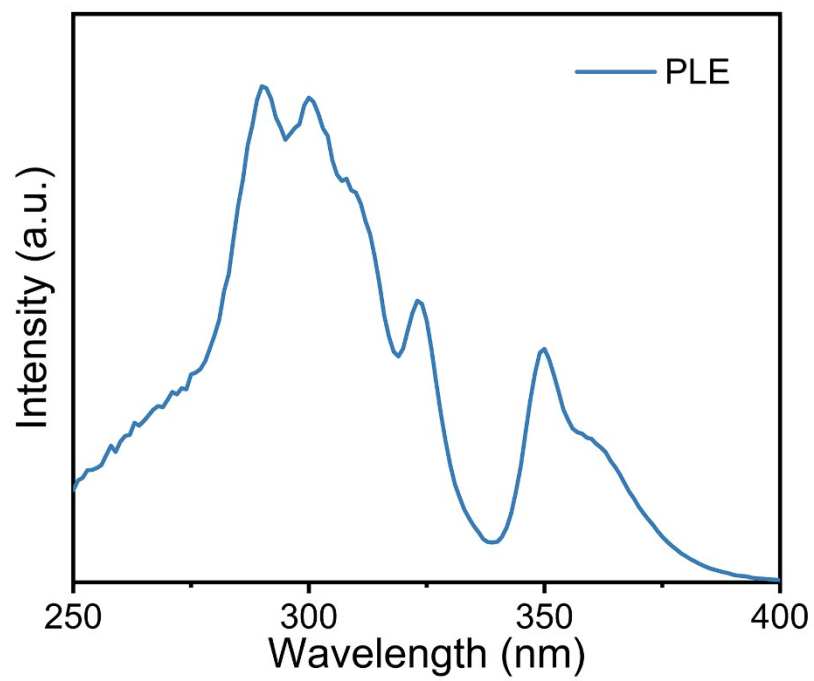


Figure S3 PL excitation spectra of Sn²⁺-doped RbI NCs.

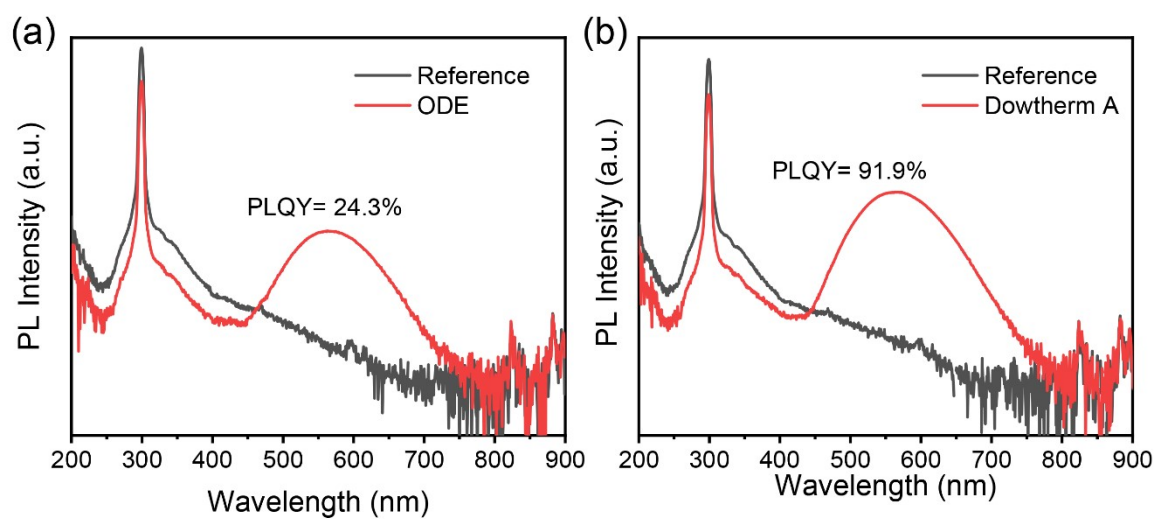


Figure S4 The PLQY of Sn²⁺-doped RbI NCs synthesized by ODE (a) and Dowtherm A (b) as a solvent.

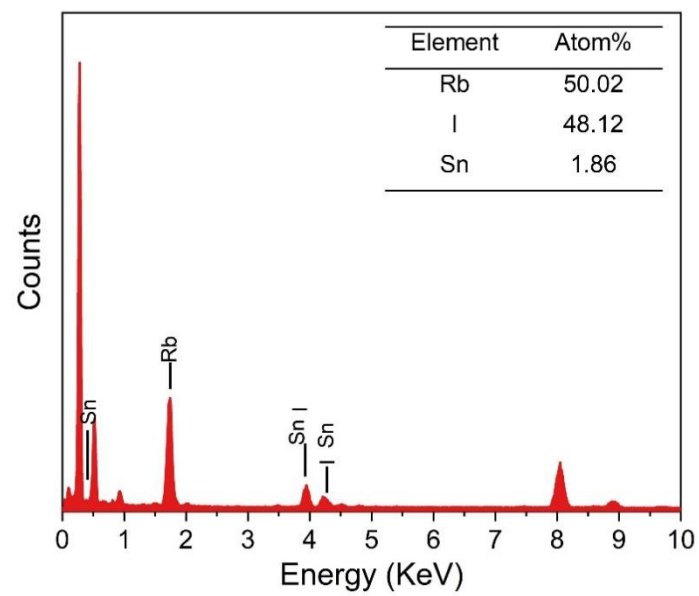


Figure S5 HRTEM EDS of Sn²⁺-doped RbI NCs.

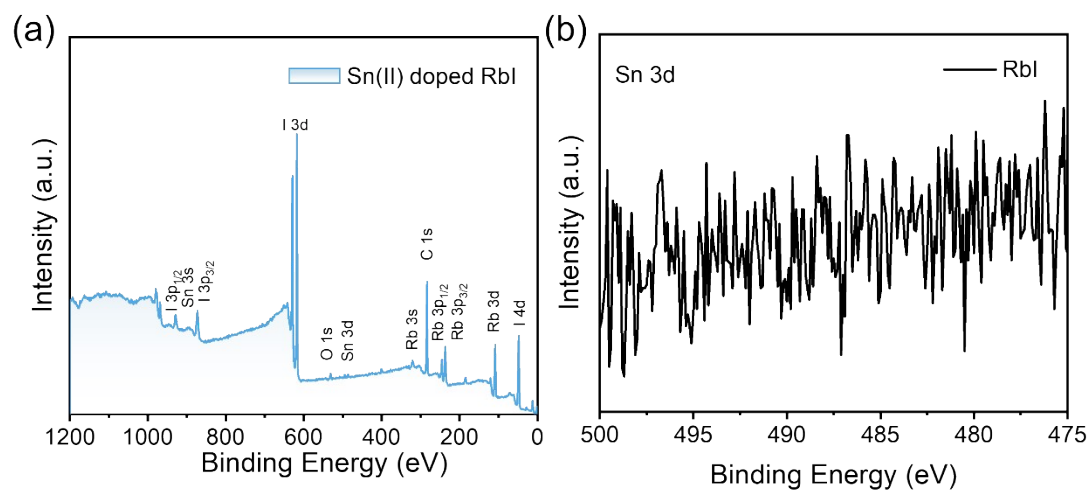


Figure S6 (a) XPS survey spectra of Sn^{2+} -doped RbI NCs. (b) Sn 3d core level spectra for RbI NCs without Sn^{2+} doping.

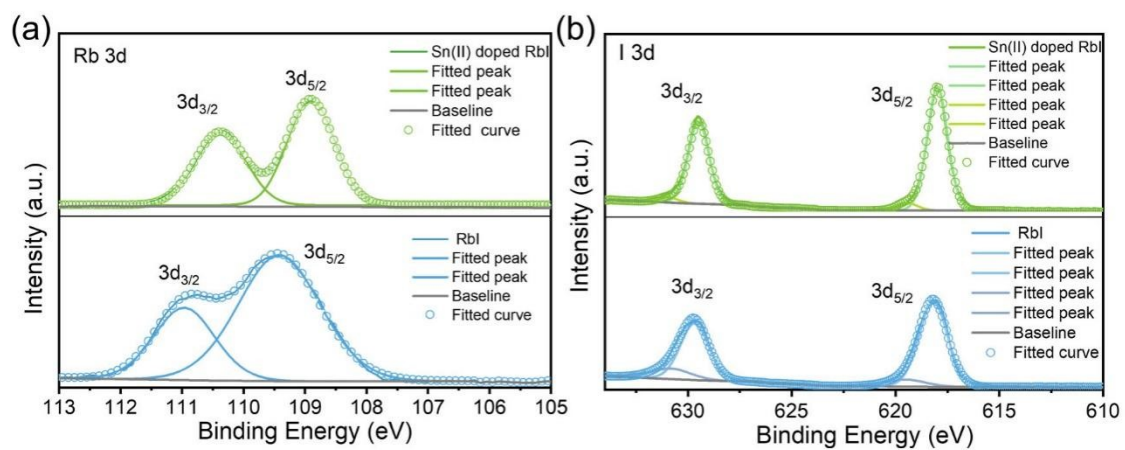


Figure S7 XPS spectra corresponding to Sn^{2+} -doped RbI NCs of Rb 3d (a) and I 3d (b).

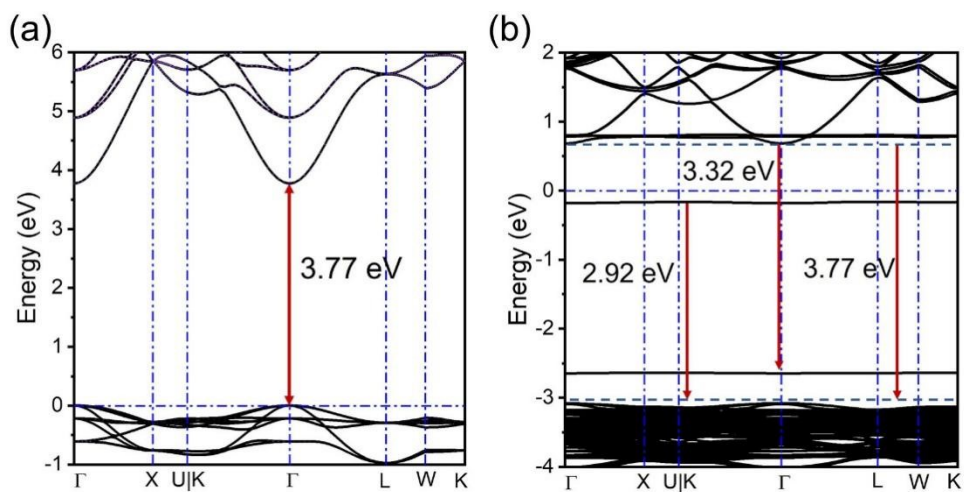


Figure S8 The band structures of RbI (a) and Sn²⁺-doped RbI (b) NCs.

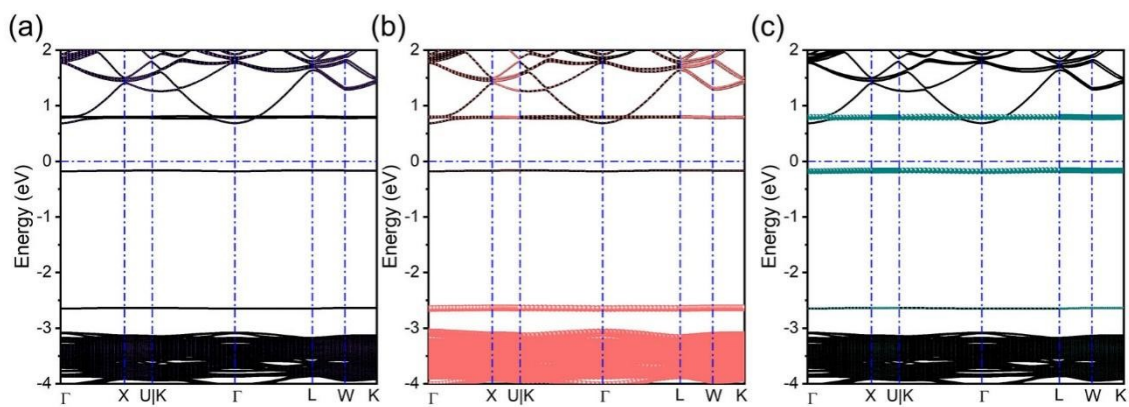


Figure S9 Energy band structure of Rb-pband (a) I-pband (b), Sn-pband (c) in Sn(II)-doped RbI NCs.

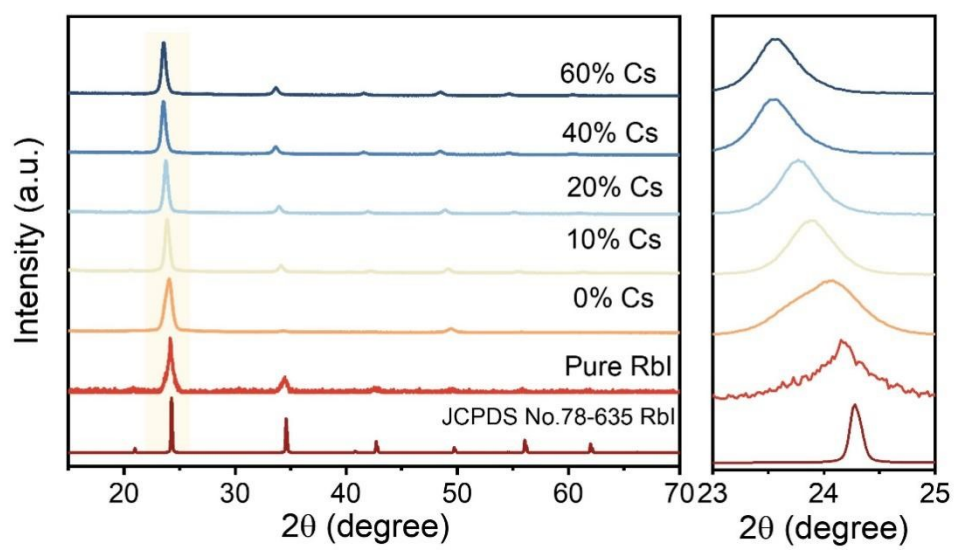


Figure S10 XRD patterns of Sn^{2+} -doped RbI NCs with increased doping of Cs^+ ions.

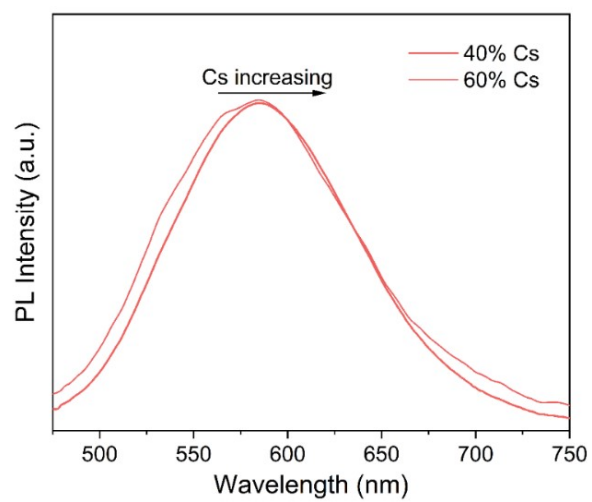


Figure S11 The photoluminescence spectrum of Sn^{2+} doped $\text{Rb}_x\text{Cs}_{1-x}\text{I}$ NCs.

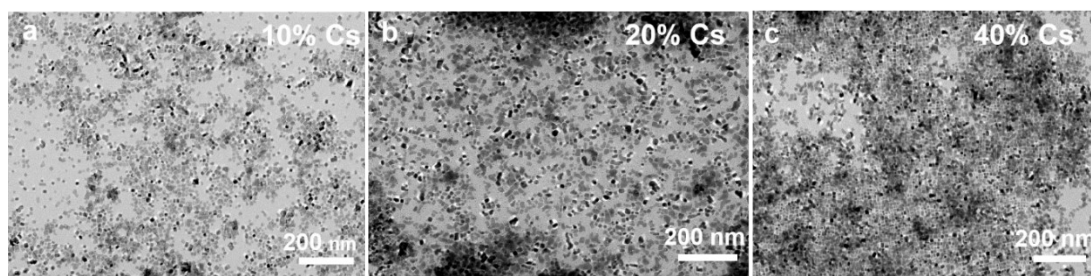


Figure S12 TEM images of Sn^{2+} -doped $\text{Rb}_x\text{Cs}_{1-x}\text{I}$ NCs. (a) $x=0.9$, (b) $x=0.8$, (c) $x=0.6$.

Table S1 The stability of Sn(II) doping and Sn(II)-based perovskites.

Materials	Categories	PL peak (nm)	PLQY (%)	Stability (days)	Ref.
Sn(II)-doped CsI	Microcrystals	750	21.5	Over 3 months	¹
MASnI ₃	Nanocrystals	750	0.06	/	²
FASnI ₃	Nanocrystals	770–830	0.1	/	³
CsSnI ₃	Nanocrystals	849	18.4	/	⁴
CsSnBr ₃	Nanocrystals	680	0.14	/	⁵
CsSn(Br _{0.5} I _{0.5}) ₃	Nanocrystals	720	0.05	/	⁵
Sn²⁺-doped RbI	Nanocrystals	568	91.9	Over 3 week	/

Table S2 Photoluminescence quantum efficiency (PLQY) of Sn²⁺-doped Rb_xCs_{1-x}I NCs.

Compound	PLQY (%)	PL peak (nm)	FWHM (nm)
10% Cs ⁺	32.7	573	106
20% Cs ⁺	55.7	574	109
40% Cs ⁺	16.5	578	113
60% Cs ⁺	5.1	574	104

References

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