## **SUPPORTING INFORMATION**

## Green-emitting CsPbI<sub>3</sub> Nanorods Decorated with CsPb<sub>2</sub>I<sub>5</sub> and Cs<sub>4</sub>PbI<sub>6</sub> Nanoclusters

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Figure S1. PL excitation spectra of MP-nanorods and CsPbI<sub>3</sub> QDs mixture measured using  $\lambda_{em}$ =511 nm (dashed black line) and 655 nm (dashed red line)



Figure S2. Tauc plots of direct (A) and indirect (B) band gap of as-synthesized CsPbI<sub>3</sub> QDs



Figure S3. Tauc plots of direct (A) and indirect (B) band gap of MP-nanorods and CsPbI<sub>3</sub> QDs

mixture



**Figure S4.** PL excitation spectra (A), PL emission (solid line), and UV-vis (dashed line) spectra (B) of CsPbI<sub>3</sub> QDs and MP-nanorods mixture assisted by titanium tetra isopropoxide (TTIP)



20 (Degree)

**Figure S5.** The XRD patterns of MP-nanorods and CsPbI<sub>3</sub> QDs mixture prepared using titanium tetra-isopropoxide (TTIP)



Figure S6. FTIR spectrum of dual-emitting colloidal mixture



Figure S7. PL emission (A) and UV-vis spectra (B) of colloidal solution upon the addition of tert-butyl alcohol and isopropanol



**Figure S8.** The XRD patterns (A) and the TEM image (B) of the ZTIB added CsPbI<sub>3</sub> QDs sample synthesized without the presence of excess 1-ODE.



Figure S9. The PL (solid line) and UV-vis (dashed line) spectra of ZTIB added-CsPbI<sub>3</sub> QDs stock solution without the presence of excess 1-ODE; The measurement was conducted in colloidal state solution using *n*-hexane as a solvent. The  $\lambda_{ex}$  = 365 nm was used when measuring the PL emission.



**Figure S10.** <sup>1</sup>H NMR spectra of pure and time-dependent 1-ODE were added to the CsPbI<sub>3</sub> QDs sample without the presence of UV light



**Figure S11.** UV-vis spectra of MP-nanorods and CsPbI<sub>3</sub> QDs colloidal mixture during the 1-ODE treatment over time



Figure S12. EDX spectrum (A) and respectfully selected area (B) of isolated MP-nanorods



Figure S13. Tauc plots for direct (A) and indirect (B) band gap of MP-nanorods



Figure S14. Phonon density of state (PDOS) spectra of  $\alpha$ -CsPbI<sub>3</sub> (A),  $\delta$ -CsPbI<sub>3</sub> (B),  $\gamma$ -CsPbI<sub>3</sub> (C), CsPb<sub>2</sub>I<sub>5</sub> (D), and Cs<sub>4</sub>PbI<sub>6</sub> (E)



Figure S15. The extended linear sweep voltammetry pattern (oxidation and reduction) of MP-CsPbI<sub>3</sub> NRs



**Figure S16.** The XRD pattern of MP-CsPbI<sub>3</sub> NRs/Cs<sub>4</sub>PbI<sub>6</sub> QDs (balck line) and the baseline of solvent mixture (toluene and 1-ODE) (red line)

Phase	Maximum Charge Density	Minimum Charge Density	Band Gap
	(e/bohr <sup>3</sup> )	(e/bohr <sup>3</sup> )	(eV)
α-CsPbI <sub>3</sub>	0.79998	0.00046	2.4553
γ-CsPbI <sub>3</sub>	0.80317	0.00057	1.6363
δ-CsPbI <sub>3</sub>	0.80254	0.00062	2.7282
CsPb <sub>2</sub> I <sub>5</sub>	0.80323	0.00069	1.9032
Cs <sub>4</sub> PbI <sub>6</sub>	0.80175	0.00016	3.4031

Table S1. Simulated band gap and charge density of each crystal phases