Supporting Information

A near-infrared fluorescent probe selectively recognizing cysteine to release H₂S and its applications

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1. Synthesis section

1.1 Synthesis of TMN-ONCS:

TMN-OH (290 mg, 1 mmol) and Br-Ar-NCS (342 mg, 1.5 mmol) were dissolved in anhydrous acetonitrile, K₂CO₃ (17 mg, 0.1 mmol) and KI (3 mg, 0.02 mmol) were added, then heated with stirring under nitrogen at 60 °C for 2 h. The reaction was monitored by TLC until completion, filtered after cooling and the filtrate concentrated under reduced pressure. The crude product was purified by silica gel column chromatography (petroleum ether: dichloromethane = 2:1) to give TMN-ONCS as an orange solid (197 mg, yield 45%). ¹H NMR (600 MHz, DMSO-*d*₆) δ 7.67 (d, *J* = 8.8 Hz, 2H), 7.53 (d, *J* = 8.5 Hz, 2H), 7.47 (d, *J* = 8.5 Hz, 2H), 7.28 (d, *J* = 6.7 Hz, 2H), 7.06 (d, *J* = 8.8 Hz, 2H), 6.84 (s, 1H), 5.19 (s, 2H), 2.61 (s, 2H), 2.55 – 2.52 (m, 2H), 1.02 (s, 6H). ¹³C NMR (150 MHz, DMSO-*d*₆) δ 170.82, 159.90, 156.87, 138.02, 137.24, 134.01, 130.09, 129.96, 129.56, 129.51, 129.49, 127.96, 126.55, 122.39, 115.78, 114.51, 113.69, 75.87, 68.98, 42.78, 40.53, 38.65, 32.15, 27.92. HRMS: *m/z* calcd for C₂₇H₂₃ON₃SNa [M + Na]⁺: 460.1460, found: 460.1460.

2.2 Synthesis of TMN-OH

(3,5,5-Trimethyl-2-cyclohexen-1-ylidene) propanedinitrile (466 mg, 2.5 mmol) and *p*-hydroxybenzaldehyde (370 mg, 3 mmol) were dissolved in 6 mL of anhydrous ethanol. After complete dissolution, 150 µL of piperidine was added. The reaction mixture was then refluxed overnight at 80-85 °C under a nitrogen atmosphere. Upon confirming the completion of the reaction by TLC, after the reaction was confirmed by thin layer chromatography, the solvent was evaporated under reduced pressure and purified by silica gel column chromatography to obtain orange powder (472 mg, yield 65%). ¹H NMR (600 MHz, DMSO-*d*₆) δ 9.98 (s, 1H), 7.56 (d, *J* = 8.8 Hz, 2H), 7.26 – 7.15 (m, 2H), 6.79 (d, *J* = 8.6 Hz, 3H), 2.60 (s, 2H), 2.53 (s, 2H), 1.01 (s, 6H). ¹³C NMR (150 MHz, DMSO-*d*₆) δ 170.80, 159.81, 157.25, 138.80, 130.36, 127.60, 126.74, 121.84, 116.35, 114.62, 113.80, 75.24, 42.80, 40.53, 38.66, 32.15, 27.92. HRMS: *m/z* calcd for C₁₉H₁₉ON₂ [M + H]⁺: 291.1497, found: 291.1500.

2.3 Synthesis of Br-Ar-NCS

N-bromosuccinic acid imide (1.79 g, 10 mmol) and *p*-tolyl isothiocyanate (1.49 g, 10 mmol) were dissolved in 20 mL of anhydrous carbon tetrachloride. The catalytic amount of benzoyl peroxide (24.2 mg, 0.1 mmol) was then added and stirred for 12 h at 80 °C reflux. After the

reaction was confirmed by thin layer chromatography, the reaction mixture was filtered, the filtrate was evaporated under reduced pressure, and the white crystals of **Br-Ar-NCS** (1.6 g, yield 55%). ¹H NMR (600 MHz, DMSO-*d*₆) δ 7.54 – 7.51 (m, 2H), 7.44 – 7.40 (m, 2H), 4.73 (s, 2H). ¹³C NMR (150 MHz, DMSO-*d*₆) δ 138.45, 134.35, 131.29, 130.24, 126.76, 40.53, 33.80. HRMS: *m/z* calcd for C₈H₆BrNSNa [M + Na]⁺: 294.9302, found: 294.9304.

2.4 Synthesis of AzMC¹

The synthesis of AzMC followed the reported procedure, and its structure was confirmed by NMR (Fig. S12 and S13) and HRMS (Fig. S14).

Table S1. The comparison of the reported fluorescent probes for Cys detection.							
Probe	E _x /E _m (nm)	Distinguish objects	Detection limited	H ₂ S Release function	Applications	Ref.	
	461/603	Cys	0.19 μΜ	-	Cells	2	
N N N N O O O	400/518 505/566	Cys, Hcy, GSH	132 nM 105 nM 62 nM	-	Cells, zebrafish	3	
CF3	350/515	Cys	4.8 μΜ	-	Cells	4	
	450, 543/592	Cys, Hcy	2.33 μM, 2.88 μM	-	Cells	5	
	452/510	Cys, Hcy	3.94 μM, 11.01 μM	-	Cells	6	
	468/550	Cys, GSH	250 nm, 370 nm	-	Cells, zebrafish	7	

2. Comparative analysis and additional analytical data

F ₃ C ^N ^S	390/518	Cys	105 nM	Yes	Cells, zebrafish	8
NCS NCS	380/504	Cys	25 nM	Yes	Cells, plants	9
	545/670	Cys	1μΜ	Yes	Cells	This work



Fig. S1. Photostability of TMN-ONCS (25 μ M) after reacting with Cys (250 μ M) for 1 h at the plateau phase. Testing conditions: $\lambda_{ex} = 545$ nm, slit width: 10/10 nm, PBS buffer solution (10 mM, pH = 7.4, 50% DMSO) at 37 °C.



Fig. S2. Calibration curve for H₂S determined by the MB method. Testing conditions: absorbance

measured at 670 nm in PBS buffer solution (10 mM, pH 7.4, 50% DMSO) at 37 °C.

3. NMR spectra and HRMS



Fig. S3. ¹H NMR spectrum of TMN-OH in DMSO-d₆ (600 MHz)



Fig. S4. ¹³C NMR spectrum of TMN-OH in DMSO-*d*₆ (150 MHz)

Elemental Composition Report

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Monoisotopic Mass, Even Electron Ions 366 formula(e) evaluated with 1 results within limits (up to 50 closest results for each mass) Elements Used: C: 19-19 H: 19-19 N: 0-100 O: 0-100 Na: 0-1 14 241127-7-297-4-TMN-OH 8 (0.079)



Fig. S5. HRMS of TMN-OH



Fig. S6. ¹H NMR spectrum of Br-Ar-NCS in DMSO-d₆ (600 MHz)



Fig. S7. ¹³C NMR spectrum of Br-Ar-NCS in DMSO-d₆ (150 MHz)



Fig. S8. HRMS of Br-Ar-NCS



Fig. S9. ¹H NMR spectrum of TMN-ONCS in DMSO-*d*₆ (600 MHz)



Fig. S10. ¹³C NMR spectrum of TMN-ONCS in DMSO-d₆ (150 MHz)

Elemental Composition Report

Single Mass Analysis Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron lons 2057 formula(e) evaluated with 1 results within limits (up to 50 best isotopic matches for each mass) Elements Used: C: 27-27 H: 23-23 N: 0-100 O: 0-100 Na: 0-1 S: 1-3 8-P--N 240927-9-296-2-TMN-ONCS 18 (0.131) 1: TOF MS ES+ 3.05e+005 [M+Na]⁺ 460.1460 100-% 461.1492 455.1899 476.1188 406.3276 415.2155 438.1622 492.1725 508.1460 522.1129 537.5339 545.1290 375.1016 490 500 510 520 530 540 550 0-390 400 410 430 440 470 360 370 380 420 450 460 480 Minimum: Maximum: -1.5 50.0 5.0 10.0 PPM 0.0 DBE 17.5 Conf(%) Formula n/a C27 H23 N3 O Na S Mass 460.1460 Calc. Mass 460.1460 mDa 0.0 i-FIT 665.7 Norm n/a

Fig. S11. HRMS of TMN-ONCS



Fig. S12. ¹H NMR spectrum of AzMC in DMSO-*d*₆ (600 MHz)

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Fig. S13. ¹³C NMR spectrum of AzMC in DMSO-*d*₆ (150 MHz)



Fig. S14. HRMS of AzMC

4. References

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