High-Entropy Ag,Pt-Based Catalyst Toward Exceptionally High-Performance and Stable Electrochemical Detection of Nitrite Lan Li, Chao Ding and Wenye Zha

Supporting Information

Experimental Section

1. Materials and Reagents

Mg(NO₃)₂ (99.9%), glucose (99.9%), Na₂CO₃ (99.9%), Na₂SO₄ (99.9%), NaHCO₃ (99.9%), CaCl₂ (99.9%), KCl (99.9%), KNO₂ (99.9%), acetone (99%), ethanol and 5 wt% Nafion solution were supplied by Energy Chemical (Shanghai, China). NaH₂PO₄ (99.9%), Na₂HPO₄ (99.9%), Fe(acac)₃ (97%), Al(acac)₃ (98%), Co(acac)₂ (97%), Ag(acac) (98%), Pt(acac)₂ (97%) were obtained from Sigma-Aldrich (Shanghai, China). 0.1 M phosphate buffer (PB) solution was prepared using NaH₂PO₄ and Na₂HPO₄. All these reagents were used as received without further purification and distilled water was used throughout the experiments.

2. Instruments

The microstructure and morphology of high entropy Ag,Pt-based catalyst (FeAlCoAgPt) was observed by Transmission electron microscopy (TEM, FEI-TALOS-F200X operated at 200 kV and equipped with high angle annular dark field (HAADF)). The crystal structures of the samples was analyzed by X-ray powder diffraction (XRD) patterns using XRD-6100 X-ray diffractometer with Cu K α radiation and diffraction data were collected for 20 angles ranging from 10° to 80° at a scan speed of 2°/min. Nitrite electrochemical detection was performed on CHI660E electrochemical workstation (Shanghai Chenhua Instrument Co., Ltd., China) employing a classical three-electrode electrochemical cell with the catalyst modified glassy carbon electrode (GCE) ($\pi \times 0.15 \times 0.15$ cm²) as working electrodes, a Pt sheet (10 mm × 10 mm × 0.2 mm) as the counter electrode and a standard Ag/AgCl electrode as the reference electrode, respectively.



Fig. S1 The atomic fraction results of EDS mapping.¹ Table S1 The target and measured atomic ratio.



Fig. S2 The amperometric i-t curves of FeAlCoAgPt/GCE for Minghu lake water analysis and milk samples.

Reference

1 Y. Kobayashi, S. Yokoyama, R. Shoji, Metals., 2024, 14, 443-461.