

## A stable Zn-coordination polymer for the quantitative and selective detection of biomarker phenylglyoxylic acid in urine

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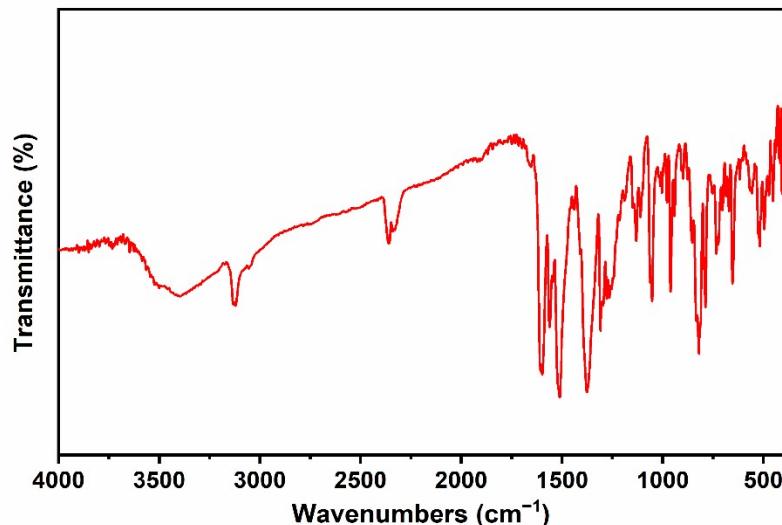


Figure S1 FT-IR curves of **1**.

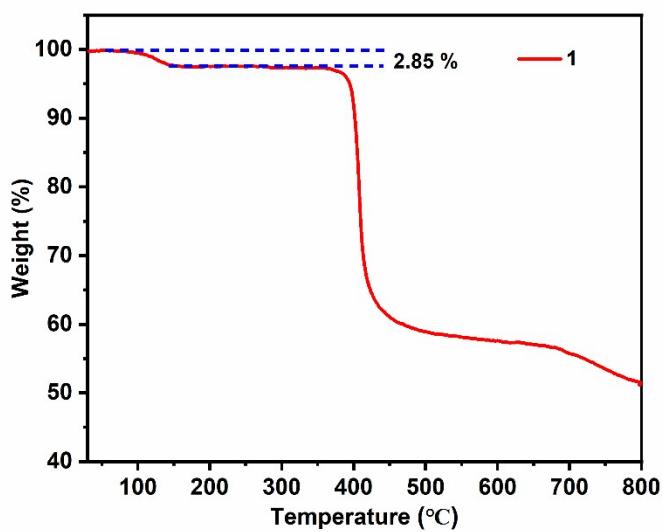


Figure S2 TGA curves of **1**.

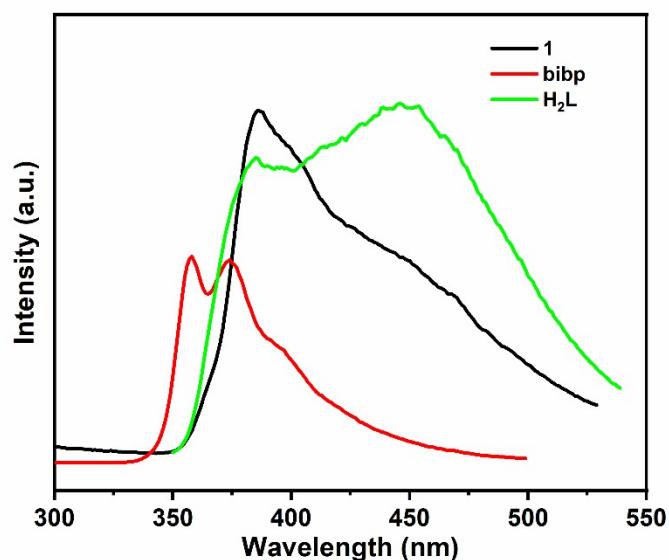


Figure S3 Solid-state fluorescence emissions of **1**,  $\text{H}_2\text{L}$  and bibp.

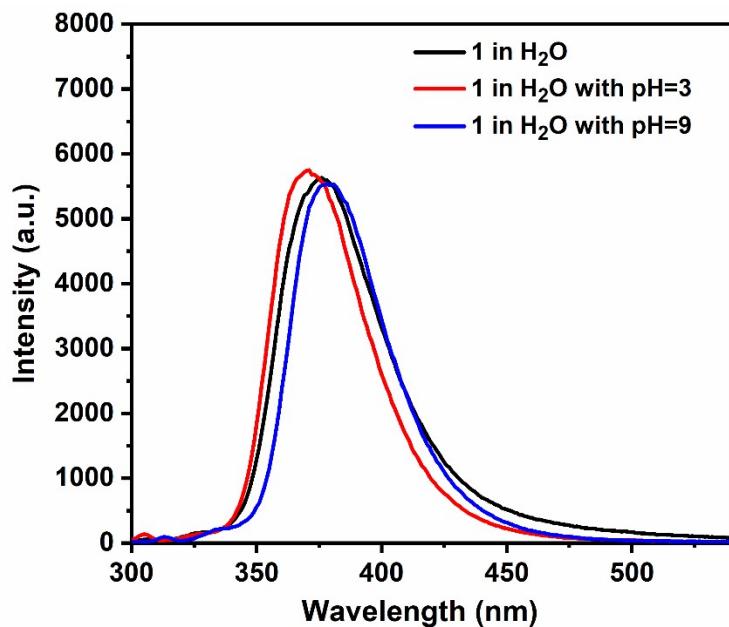


Figure S4 Fluorescence emissions of **1** in aqueous solutions with different pH values.

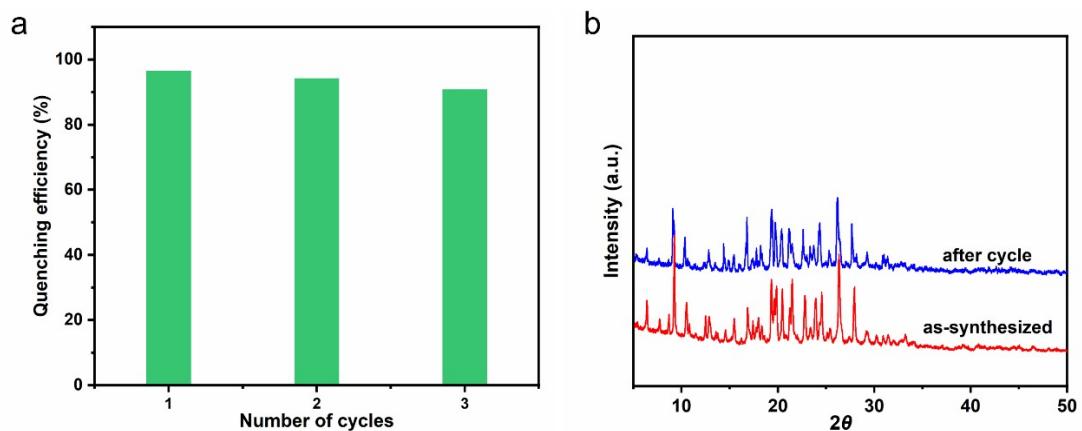


Figure S5 (a) Cyclic stability of **1** for the detection of PGA. (b) PXRD patterns of **1** before and after soaking in the solution of PGA.

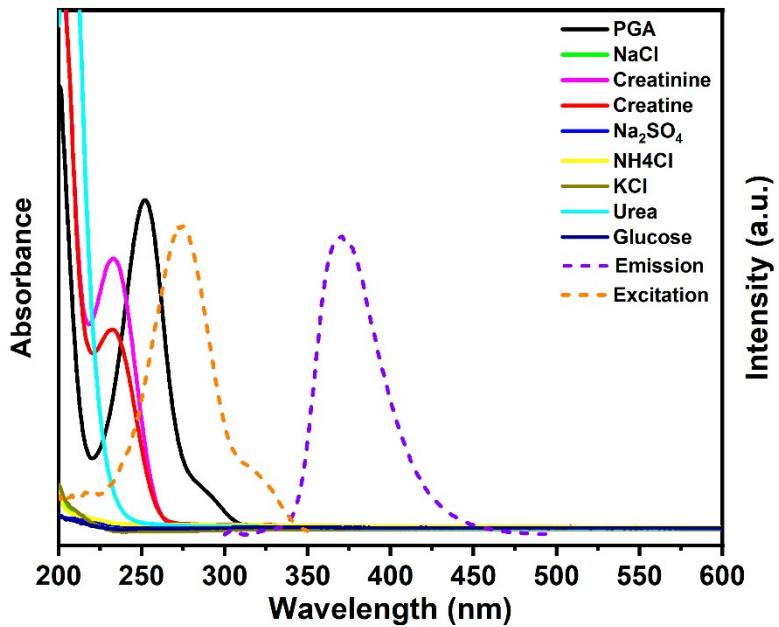


Figure S6 UV–Vis absorption spectra of different urine components and the fluorescence emission/excitation bands of **1** in water.

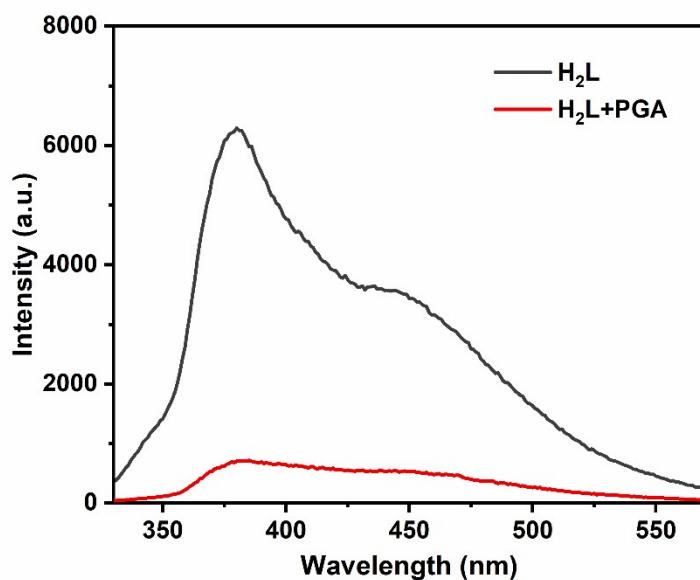


Figure S7 The behaviour with  $\text{H}_2\text{L}$  alone to detect PGA.

Table S1 Selected bond lengths (Å) and angles (°) for **1**

Selected bond lengths		Selected bond lengths angle	
Zn1—O2	1.955(2)	O2—Zn1—O3 <sup>i</sup>	108.67(10)
Zn1—O3 <sup>i</sup>	1.959(2)	O2—Zn1—N1	121.58(11)
Zn1—N1	2.008(3)	O2—Zn1—N4 <sup>ii</sup>	109.40(11)
Zn1—N4 <sup>ii</sup>	2.036(3)	O3 <sup>i</sup> —Zn1—N1	112.45(11)
		O3 <sup>i</sup> —Zn1—N4 <sup>ii</sup>	95.92(11)
		N1—Zn1—N4 <sup>ii</sup>	105.67(11)

Symmetry code: (i)  $x-1/2, -y+1/2, z+1/2$ ; (ii)  $-x+7/2, y+1/2, -z+3/2$ .

Table S2 Non-bonding interactions geometry (Å, °)

Complex	D—H...A	d(D—H)	d(H...A)	d(D...A)	∠DHA
<b>1</b>	O <sub>5</sub> —H <sub>5A</sub> ...N <sub>9</sub>	0.767	2.286	2.937	143.27
	C <sub>20</sub> —H <sub>20</sub> ...O <sub>5</sub>	0.940	2.364	3.116	136.78
	C <sub>62</sub> —H <sub>62</sub> ...O <sub>1</sub>	0.940	2.612	3.205	121.55
	C <sub>14</sub> —H <sub>14</sub> ...O <sub>5</sub>	0.940	2.691	3.605	164.35

Table S3 Comparison of various methods used for sensing PGA

Method	Material	Sample	LOD (µM)	Ref
Luminescence	LCP 1	Water	3.4	
	LCP 2	Water	3.8	S1
	LCP 3	Water	5.8	
HPLC-MS	/	Urine	6.67	S2
Capillary electrophoresis (CE)	/	Urine	20	S3
Voltammetry	/	Urine	33	S4
Potentiometry	Salen-Co(III)	Water	1.96	S5
Luminescence	<b>1</b>	Urine	2.67	This work

Table S4 **1** was used to detect the concentration of PGA in human urine.

Complex	Add ( $\mu$ M)	Found ( $\mu$ M)	Recovery (%)	RSD (%)
<b>1</b>	5.00	4.73	94.60	3.57
	10.00	11.54	115.40	3.48
	15.00	14.92	99.47%	2.31

## References

- [S1] S. Wu , M. Zhu , Y. Zhang , M. Kosinova , V. P. Fedin and E. Gao , Appl. Organomet. Chem., 2020, 35.
- [S2] J. Z. Wang , X. J. Wang , Y. H. Tang , S. J. Shen , Y. X. Jin and S. Zeng , J. Chromatogr. B: Anal. Technol. Biomed. Life Sci., 2006, 840, 50—55.
- [S3] C. Y. Wang , C. T. Huang and Y. Z. Hsieh , J. Sep. Sci., 2003, 26, 69—74.
- [S4] T. Navrátil , Z. Šenholdová , K. Shanmugam and J. Barek , Electroanalysis, 2006, 18 , 201—206.
- [S5] B. Li , J. Hou , Y. Y. ang , S. Chen and L. Xu , Chem. Sens., 2011, 01.