Supporting Information

Naphthalimide-based fluorescent 'turn off' probes for palladium ion:

structure-activity relationships

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1. Materials and instruments

Unless otherwise specified, all reagents and chemicals used in this study were purchased from commercial companies. ¹H-NMR and ¹³C-NMR spectra were measured with a Bruker Avance 400 MHz or 600 MHz spectrometer with tetramethylsilane (TMS) as the internal standard substance. Mass spectra were recorded with a Waters Xevo G2XS Tof mass spectrometer. The absorption and emission spectra were respectively measured with an Evolution 220 and a Lumina spectrophotometer from Thermo Fisher Scientific. The confocal fluorescence image was carried out with an A1R laser scanning confocal microscope from Nikon.

2. Cell culture and fluorescence imaging

The resuscitated and passaged L929 cells and Hela cells were incubated with DMEM medium containing 10% fetal bovine serum in the incubator (37 °C, 5% CO₂) for 24 h. Subsequently, the L929 cells and Hela cells were transferred to confocal culture dishes and incubated in the incubator (37 °C, 5% CO₂) for another 24 h. The experiments were divided into three groups: 1) in the control group, the cells were incubated with serum-free medium containing 5 μ M probe for 20 min; 2) in the test group, the cells were pre-incubated with serum-free medium containing $10 \,\mu\text{M}$ PdCl₂ for 30 min, followed by incubated with serum-free medium containing 5 μ M probe for 20 min; 3) in the validation group, to verify the targeting ability of probe NPE and NPF toward cancer cells through receptor-mediated endocytosis, Hela cells were respectively preincubated with 1 mmol/L biotin or 100 µmol/L celecoxib for 30 min to compete with biotin or indomethacin receptors, and then incubated with serum-free medium containing 5 µM probe for 20 min; while L929 cell were pretreated with 1 µg/mL LPS for 12 h to induce inflammation followed by incubated with serumfree medium containing 5 uM probe for 20 min. After washed with PBS for three times, the cells were treated with 1 mL of culture medium for confocal experiments. In the confocal experiments, the fluorescence of blue channel (425-475 nm) and green channel (500-550 nm) were recorded under excited with a 405 nm/488 nm laser, respectively.

3. Synthesis of probes NPA-NPF



Scheme S1 Synthesis routes of NPA, NPB and NPC

NPA-NPC

N-butyl-4-bromo-1,8-naphthalimide (331 mg, 1 mmol) and ethylenediamine (1.2 mL) were dissolved in 20 mL of ethylene glycol monomethyl ether and refluxed for 6 hours under nitrogen protection. The solvent was removed under reduced pressure, and the residue was purified by silica gel column chromatography (DCM: MeOH = 50:4) to give yellow powder **NPA** (192 mg, 62%). ¹H-NMR (400 MHz, CDCl₃) δ (ppm): 8.57 (d, *J* = 7.3 Hz, 1H), 8.44 (d, *J* = 8.4 Hz, 1H), 8.18 (d, *J* = 8.4 Hz, 1H), 7.60 (t, *J* = 7.9 Hz, 1H), 6.69 (d, *J* = 8.4 Hz, 1H), 6.20 (t, *J* = 5.0 Hz, 1H), 4.20-4.12 (m, 2H), 3.42 (q, *J* = 5.3 Hz, 2H), 3.18 (t, *J* = 5.7 Hz, 2H), 1.70 (q, *J* = 8.0 Hz, 2H), 1.61 (s, 2H), 1.44 (h, *J* = 7.4 Hz, 2H), 0.97 (t, *J* = 7.4 Hz, 3H). ¹³C-NMR (151 MHz, CDCl₃) δ (ppm): 164.76, 164.23, 149.67, 134.46, 131.10, 129.82, 126.23, 124.68, 123.13, 120.48, 110.36, 104.41, 44.90, 40.17, 39.99, 30.33, 29.70, 29.32, 20.45, 13.89. HR-MS m/z calcd for C₁₈H₂₁N₃O₂: [M+H]⁺

312.1712; found 312.1701. Compounds **NPB** and **NPC** were obtained by the same procedure with N,N-dimethylethylenediamine or methylpiperazine instead of ethylenediamine. **NPB** (151 mg, 43%): ¹H-NMR (400 MHz, CDCl₃) δ (ppm): 8.60-8.55 (m, 1H), 8.45 (d, J = 8.4 Hz, 1H), 8.18 (d, J = 8.4 Hz, 1H), 7.62 (t, J = 7.8 Hz, 1H), 6.66 (d, J = 8.4 Hz, 1H), 6.33 (t, J = 4.3 Hz, 1H), 4.16 (dd, J = 8.6, 6.5 Hz, 2H), 3.43-3.36 (m, 2H), 2.75 (t, J = 5.8 Hz, 2H), 2.35 (s, 6H), 1.75-1.66 (m, 2H), 1.45 (q, J = 7.5 Hz, 2H), 0.97 (t, J = 7.3 Hz, 3H). ¹³C-NMR (151 MHz, CDCl₃) δ (ppm): 164.79, 164.25, 149.63, 134.52, 131.10, 129.80, 126.46, 124.64, 123.07, 120.43, 110.21, 104.36, 56.94, 45.01, 40.14, 39.98, 30.34, 20.45, 13.90. HR-MS m/z calcd for C₂₀H₂₆N₃O₂: [M+H]⁺ 340.2025; found 340.2014. **NPC** (172 mg, 51%): ¹H-NMR (600 MHz, DMSO- d_6) δ (ppm): 8.46 (dd, J = 7.3, 1.3 Hz, 1H), 8.42 (dd, J = 8.4, 1.2 Hz, 1H), 8.38 (d, J = 8.0 Hz, 1H), 7.80 (dd, J = 8.5, 7.2 Hz, 1H), 7.33 (d, J = 8.1 Hz, 1H), 4.05-3.99 (m, 2H), 3.23 (t, J = 4.9 Hz, 4H), 2.64 (t, J = 4.2 Hz, 4H), 2.31 (s, 3H), 1.63-1.56 (m, 2H), 1.34 (h, J = 7.4 Hz, 2H), 0.92 (t, J = 7.4 Hz, 3H). ¹³C-NMR (151 MHz, DMSO- d_6) δ (ppm): 163.99, 163.46, 156.10, 132.63, 131.07, 130.92, 129.55, 126.45, 125.74, 123.01, 115.95, 115.51, 55.10, 52.98, 46.21, 30.18, 20.27, 14.20. HR-MS m/z calcd for C₂₁H₂₆N₃O₂: [M+H]⁺ 352.2025; found 352.2017.

NPE



Scheme S2 Synthesis routes of NPE

m1: 4-Bromo-1,8-naphthalic anhydride (2.77 g, 10.0 mmol) and ethylenediamine (1 g, 15.1 mmol) in anhydrous ethanol (100 mL) were refluxed for 6 hours under nitrogen protection. After filtering to remove the solid, the resulting solvent was distilled under reduced pressure. The obtained solid was recrystallized with ethanol to give a yellow powder **m1** (870 mg, 27%).

m2: Under room temperature, biotin (241.0 mg, 0.99 mmol) and HATU (294.0 mg, 0.77 mmol) in N,N-dimethylformamide (5 mL) were stirred for 1 hour to activate the carboxyl group. Then, DIPEA (0.56 mL) and compound **m1** (207 mg, 0.65 mmol) were added to the above solution. The resulting solution was stirred for another 16 hours and then added dropwise into water (200 mL). The solid was filtered under vacuum to give **m2** (326.5 mg, 60%).

NPE: Compound **m2** (326 mg, 0.60 mmol) and methylpiperazine (1.2 mL) in ethylene glycol monomethyl ether (20 mL) were refluxed under nitrogen protection for 16 hours. After cooling to room temperature, the solvent was removed by rotary evaporation. The resulting solid was purified by silica gel column chromatography (DCM: MeOH = 50:4) to yield a yellow solid **NPE** (102 mg, 30%). ¹H-NMR (400 MHz, DMSO- d_6) δ (ppm): 8.51-8.37 (m, 3H), 7.89 (t, *J* = 6.1 Hz, 1H), 7.83 (dd, *J* = 8.5, 7.3 Hz, 1H), 7.36 (d, *J* = 8.1 Hz, 1H), 6.40 (s, 1H), 6.36 (s, 1H), 4.32-4.26 (m, 1H), 4.12 (t, *J* = 6.0 Hz, 2H), 4.06 (t, *J* = 5.9 Hz, 1H), 3.25 (s, 4H), 3.03-2.98 (m, 1H), 2.82 (ddd, *J* = 12.4, 9.9, 5.1 Hz, 2H), 2.66 (s, 4H), 2.32 (s, 3H), 1.94 (t, *J* = 7.5 Hz, 2H), 1.51 (dd, *J* = 12.6, 5.9 Hz, 2H), 1.36 (dt, *J* = 15.7, 7.8 Hz, 4H), 1.24 (s, 2H). ¹³C-NMR (151 MHz, DMSO- d_6) δ (ppm): 174.93,

172.69, 164.27, 163.74, 163.17 (d, J = 3.4 Hz), 156.01, 132.54, 130.99, 130.82, 129.75, 126.47, 125.79, 123.30, 116.29, 115.51, 61.40, 59.65, 55.81, 55.09, 52.98, 46.19, 37.07-36.49 (m), 35.69 (d, J = 3.3 Hz), 28.52 (d, J = 3.9 Hz), 28.42 (d, J = 3.6 Hz), 25.63, 25.01. HR-MS m/z calcd for C₂₉H₃₇N₆O₄S: [M+H]⁺ 565.2597; found 565.2600.

NPD and NPF were synthesized according to the literature ^[1].



Fig. S1 ¹H-NMR, ¹³C-NMR and HR-MS of NPA



Fig. S2 ¹H-NMR, ¹³C-NMR and HR-MS of NPB



Fig. S3 ¹H-NMR, ¹³C-NMR and HR-MS of NPC



Fig. S4 ¹H-NMR, ¹³C-NMR and HR-MS of NPE



Fig. S5 ¹H-NMR, ¹³C-NMR and HR-MS of NPF



Fig. S6 The emission spectra of the probes with (red line) and without (black line) Pd^{2+} in PBS (20 mM, pH 7.0). [probe] = 10 μ M, [Pd²⁺] = 200 μ M, λ_{ex} = 425 nm for **NPB** and **NPF**, λ_{ex} = 382 nm for **NPC**, **NPD** and **NPE**.



Fig. S7 The Job's plots of NPA-NPE and Pd^{2+} in PBS (20 mM, pH 7.0), [probe]+[Pd^{2+}] = 15.0 \mu M.



Fig. S8 Fluorescence spectra of NPB after addition of Pd^{2+} and further addition of Na_2S in PBS solution (20 mM, pH 7.0).



Fig. S9 The absorption (a, c, e, g) and emission (b, d, f, h) spectra of NPB/NPC/NPD/NPE under different pH conditions. [probe] = 10 μ M, λ_{ex} = 425 nm for NPB, and λ_{ex} = 382 nm for NPC, NPD and NPE.



Fig. S10 The absorption (a, c, e, g, i) and emission (b, d, f, h, j) spectra of probe-Pd²⁺ mixed system under different pH conditions. [probe] = 10 μ M, [Pd²⁺] = 200 μ M, λ_{ex} = 425 nm for NPA and NPB, and λ_{ex} = 382 nm for NPC, NPD and NPE.



Fig. S11 The plots of fluorescent intensities of **NPB** (a), **NPD** (b) and **NPE** (c) without (black line) and with Pd^{2+} (red line) as a function of pH. [**NPB**] = [**NPD**] = [**NPE**] = 10 μ M, [Pd²⁺] = 200 μ M, $\lambda_{ex} = 425$ nm for **NPB**, $\lambda_{ex} = 382$ nm for **NPD** and **NPE**.



Fig. S12 Effects of metal ions on the emission spectra of **NPB-NPE**. [probe] = 10 μ M, [metal ion] = 200 μ M, λ_{ex} = 425 nm for **NPB**, and λ_{ex} = 382 nm for **NPC**, **NPD** and **NPE**.



Fig. S13 Effects of metal ions on the fluorescence intensities of NPB-NPE in the absence (green) and presence (orange) of Pd^{2+}



Fig. S14 Partial ¹H NMR spectra of NPB and NPB after reaction with $PdCl_2$



Fig. S15 Frontier molecular orbital of probe **NPB/NPC** in the absence and presence of Pd^{2+} . LANL2DZ basis group for Pd(II) and 6-31G(d, p) for other atoms.



Fig. S16 Pd^{2+} effect on the emission spectra of **NPB-NPE**. [probe] = 10 μ M, $[Pd^{2+}] = 0-200 \ \mu$ M, $\lambda_{ex} = 425 \ nm$ for **NPB**, and $\lambda_{ex} = 382 \ nm$ for **NPC**, **NPD** and **NPE**.



Fig. S17 Evaluation of binding constants of Pd^{2+} to NPA-NPE

Probes	Detection limit	Linear range	Ref.
PPC	0.90 µM	0-50 μM	[2]
Cy202	0.52 μM	0-100 µM	[3]
3c	0.45 µM	0-25 μM	[4]
PPIX-L2	0.38 µM	0-4 µM	[5]
IMQU-8	0.0025 µM	0-2 µM	[6]
BTANC	0.0225 μM	0-9 µM	[7]
L_1	0.041 μM	0.25-0.5 μM	[8]
QA	0.485 μM	0-6 µM	[9]
PS-1	0.26 µM	0-5 µM	[10]
1	0.055 μΜ	0-6 µM	[11]
Bis-TPI	0.92 µM	0-1 µM	[12]
QB-3	0.089 µM	0-5 µM	[13]
NPA	3.30 µM	1-20 μM	This work
NPB	0.23 μM	1-20 µM	This work
NPC	9.48 μM	1-20 μM	This work
NPD	0.57 μM	1-20 μM	This work
NPE	4.06 µM	1-20 µM	This work

Table S1 Comparation of the detection limits of Pd²⁺ with some reported probes

Table S2 Comparation analysis of Pd²⁺ in different samples

Samples	Added	Found	Recovery	RSD
	(µM)	(μM)	(%)	(%, n = 3)
River water	3	3.61	120	5.67
	5	6.15	123	5.50
	8	7.93	99	2.61
Tap water	3	2.79	93	5.64
	5	4.47	89	3.94
	8	7.52	94	1.78
Drinking water	3	2.72	91	5.87
	5	4.25	85	5.09
	8	8.09	101	0.61

Before detection, the water samples were filtered through 0.22 μ m membrane filters.

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