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Supporting Information

for

Snowflake-Like Cu₂O-Pt Nanocluster Mediated Fenton Photothermal and

Chemodynamic for Antibiotic Wound Healing Therapy

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Experimental section

Characterization

The morphology and chemical elemental composition of the synthesized HCPNLs were characterized using scanning electron microscopy (SEM) (JEOL JSM-7800F, Japan), transmission electron microscopy (TEM), and high-resolution transmission electron microscopy (HR-TEM) (FEI Tecnai G2 F30, USA). The crystal structure and morphological details of the samples were assessed through X-ray diffractometry (XRD) (D8-Advance, Cu Ka, Germany). The electron binding energy and surface chemistry of the samples were analyzed using X-ray photoelectron spectroscopy (XPS) (PHIX-tool, ULVAC, Japan). Infrared measurements were conducted with a TP5 handheld infrared thermometer (Hikvision, China) to obtain thermograms and assess temperature variations. The concentration of organic dyes in solution was quantified using a UV-Vis spectrophotometer (Nanodrop One, Thermo Fisher Scientific, USA). Bacterial and cellular morphologies were examined via scanning electron microscopy (SEM, Germany) and confocal laser scanning microscopy (CLSM, Germany).

Photothermal conversion efficiency (η) of HCPNLs

According to Li's report(Li et al., 2022), the photothermal conversion efficiency (η) is

calculated by the following equation:

$$\eta = \frac{hA(\Delta T_{max} - \Delta T_{max,H_2O})}{I(1-10^{-A_{\lambda}})}$$

Where h is the heat transfer coefficient, A is the surface area of the vessel, and

 ΔT_{max} is is the temperature change of the HCPNLs solution at the maximum steady state temperature under NIR light irradiation. $\Delta T_{max,H_2O}$ is the temperature change of water under the same conditions. I is the power density of the NIR laser. A_{λ} is the light absorption value of HCPNLs at 808 nm.

$$\tau_{s} = \frac{\sum_{i} m_{i} c_{p,i}}{hA}$$

The equation can be simplified since the mass of HCPNLs is much lower than that of water:

$$\tau_{\rm S} = \frac{m_{\rm H_2O} c_{\rm H_2O}}{hA}$$

 m_{H_2O} is the mass of water, C_{H_2O} is the specific heat capacity of water (4.2 J•g⁻¹•C⁻¹), and τ_s is the time constant, calculated by Eq:

$$\tau_{\rm S} = \frac{t}{-\ln\theta}$$
$$\theta = \frac{T_{\rm surr} - T}{T_{\rm surr} - T_{\rm max}}$$

t is the real time cooling time and T is the real time temperature of t. T_{surr} is the ambient temperature. T_{max} is the maximum stabilisation temperature of the solution.



Fig. S1 (A) HAADF of HCPNLs. (B) EDX analysis of HCPNLs.



Fig. S2 HRTEM of HCPNLs and the lattice fringes of Cu_2O and Pt.



Fig. S3 Photoelectric current measurements of HCu₂O and HCPNLs.



Fig. S4 UV-Vis absorbance spectrum of HCPNLs.



Fig. S5 Heating and cooling curves of HCPNLs under laser irradiation (NIR:808 nm,

2.0 w•cm⁻²).



Fig. S6 Michaelis-Menten kinetic (A) and Lineweaver-Burk plotting (B) for HCPNLs with TMB and H_2O_2 .



Fig. S7 EPR spectra under different reaction conditions.



Fig. S8 GSH consumption caused by different concentrations of HCPNLs (5,

10, 20, 40 and 60 $\mu g/mL).$



Fig. S9 The copper ion release from HCPNLs solution was measured by ICP-OES HCPNLs (60 μ g/mL) treated with different concentrations of GSH for 4 h.



Fig. S10 Representative growth curves of bacteria after treatment with different concentrations of material.



Fig. S11 In vitro antimicrobial assay of PBS, HCu₂O, and HCPNLs (NIR: 2.0 w•cm⁻², H₂O₂: 1 mM, Concentration of catalyst: 60 μ g•mL⁻¹). (mean \pm sem, n = 3, *p < 0.05, **p < 0.01, ***p < 0.001).



Fig. S12 HCPNLs in vitro antimicrobial Combination index.



Fig. S13 Confocal images of ROS levels after different treatments.



Fig. S14 Infrared thermogram of rats irradiated with a near-infrared laser. (NIR:808 nm, 2.0 w•cm⁻²)



Fig. S15 Pictures of healed wounds and corresponding healing marks in the penicillin

grou of bacterial colonisation.



Fig. S16 Changes in body weight of mice in different groups.



100µm

Fig. S17 H&E staining of major organs (heart, liver, spleen, lungs and kidneys) of mice in different treatment groups.



Fig. S18 Blood routine indicators of mice upon different treatment for 7 days.

References

Li, X., S. Luo, Y. Chen, et al., 2022. Facile One-Pot Synthesis of Meteor Hammerlike Au-MnOx Nanozymes with Spiky Surface for NIR-II Light-Enhanced Bacterial Elimination. Chemistry of Materials. 34, 9876-9891. <u>https://doi.org/10.1021/acs.chemmater.2c01775</u>