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## **Supporting Information**

## Theragenerative Injectable Bone-Adhesive Hydrogels for Combined

## **Photothermal Osteosarcoma Therapy and Bone Repair**

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*Chemical and Microstructural Characterizations:* For chemical and microstructure characterization, the precursors were measured after drying, while the hydrogels were tested after freeze-casting in liquid nitrogen and dried by supercritical CO<sub>2</sub> drying method to obtain the corresponding aerogel. For <sup>1</sup>H nuclear magnetic resonance (<sup>1</sup>H NMR) spectra of precursors (SF, SS, SSDopa and dopamine monomer), 0.005 g dry sample was dissolved in deuterium oxide (D<sub>2</sub>O, 0.5 mL) and analyzed on a Bruker Avance II 300 MHz spectrometer. The molecular structure with main functional groups of SF, SS, SFO and final aerogels were characterized by Fourier-transform infrared (FT-IR) spectra using a Spectrum 400 (PerkinElmer) at room temperature in the wavenumber range of 400- 4000 cm<sup>-1</sup>. The successful oxidation of SF to SFO was characterized by the change in absorption peaks of the 15x diluted solutions against DI water in UV-Vis spectra recorded by a PerkinElmer Lambda 950 at room temperature. The porous microstructure of the final aerogels were observed by scanning electron microscopy (SEM) on a Zeiss Neon40 CrossBeam machine. The aerogels were cut horizontally and vertically presenting different morphologies, which were examined after being treated with gold sputtering with 10 nm thickness. Energy dispersive X-ray spectroscopy (EDX) is acquired simultaneously during SEM testing as the machine is equipped with Oxford's INCA system for SEM-EDX images.

**Preparation of Commercial Aqueous SF and SS Solution:** Commercially available SF and SS solutions (Fibrothelium GmbH, Aachen, Germany) were used as a standard for the comparison of the extracted proteins. Specifically, SF and SS aqueous solutions were obtained using PureSilk® technology, enabling medical-grade quality production of silk on an industrial scale for a broad range of concentrations. Briefly, the two proteins were separated by degumming in a hot alkali solution. Fibroin was then dissolved in a proprietary non-toxic solvent system based on Ajisawa's reagent, purified and fully dialyzed against VE water within 8 hours using tailored extraction processing. The obtained fibroin solution was then diluted to 3.5 wt% and stored at 4°C. After the degumming process, sericin was purified, dialyzed against VE water and lyophilized.

	H <sub>2</sub> N COOH	H <sub>2</sub> N COOH		H <sub>2</sub> N COOH	H <sub>2</sub> N COOH	H <sub>2</sub> N COOH	
	Ser	Asp	Gly	Glu	Thr	Tyr	Ala
SF	12%	4%	35%	2%	1%	10%	30%
SS	30%	20%	11%	6%	9%	5%	4%

Table S1. Amino acid composition of SF and SS. Calculated according to references.<sup>1,2</sup>

Sample	SFO (%)	SSDopa (%)	$Cu^{2+}$ (mol/L)	TA (%)
0	2.18	0.00	0.000	0.00
1	2.18	0.33	0.000	0.00
2	2.16	0.32	0.008	0.00
3	2.19	0.33	0.000	0.27
4	2.14	0.32	0.008	0.27
5	2.09	0.31	0.016	0.26
6	2.13	0.32	0.008	0.53
7	2.13	0.32	0.008	0.80
8	1.50	0.64	0.008	0.27
9	2.80	0.00	0.008	0.27

Table S2. Hydrogel adhesives SFO-SSDopa-Cu-TA with different recipes developed in this study.

Some small differences in concentration of the same substance (e.g. SFO, 2.18% and 2.19%) are due to the disparity in the amount of KOH added for pH adjusting, resulting in a difference of some microliters in the volume of solution for the whole system.

Metal Nitrile gloves Glass Plastic									
Composition in the hydrogel system				Adhesive on different objects				mechanical	
SFO (%)	SSDopa (%)	Cu (mol/L)	TA (%)	Metal	Nitrile gloves	Glass	Plastic	propertiy (soft/hard)	
2.14	0.32	0.008	0.27	~ ~	$\checkmark$	~ ~	~ ~	normal	
2.09	0.31	0.016	0.26	~	×	$\checkmark$	$\checkmark$	normal-hard	
2.04	0.31	0.023	0.25	$\checkmark$	×	$\checkmark$	~	hard	
1.99	0.30	0.030	0.25	$\checkmark$	×	×	$\checkmark$	hard	
1.50	0.64	0.008	0.27	$\checkmark \checkmark$	~ ~	$\checkmark$ $\checkmark$	~ ~	soft-normal	
2.14	0.32	0.008	0.27	~ ~	$\checkmark$	$\checkmark$ $\checkmark$	~ ~	normal	
2.79	0.00	0.008	0.27	$\checkmark$	×	$\checkmark$	$\checkmark$	normal-hard	
2.14	0.32	0.008	0.27	~ ~	$\checkmark$	~ ~	~ ~	normal	
2.13	0.32	0.008	0.53	~ ~	×	$\checkmark$	$\checkmark$	normal	
2.13	0.32	0.008	0.80	$\checkmark$	×	×	$\checkmark$	normal-hard	

SFO1-SSDopa0.3-Cu0.008-TA0.27

**Figure S1.** Subjective evaluation of the adhesion of hydrogel with different components to common materials (metal, nitrile gloves, glass and plastics) and their mechanical properties (hard or soft).



**Figure S2.** Preparation of samples for hydrogel adhesion tensile testing with 50 g glass fitting the joints.



**Figure S3.** Indirect assessment of cell viability and proliferation for a) conditional medium after 1, 3, 5, and 7 days of soaking in low concentration (20 mg/ mL) hydrogel, and b) conditional medium after 7 days of soaking in high concentration (100 mg/ mL) hydrogel.



**Figure S4.** a) Live/Dead fluorescence staining assay for the attachment and proliferation after 2D direct cell culture. b) excitation and emission spectra of Calcein AM and PI. Adapted from *Beyotime*.<sup>3</sup> c) 3D culture system of hydrogels and cells.



Figure S5. Schematic diagram of cell migration assay using culture inserts. Adapted from Pijuan.<sup>4</sup>



Figure S6. a) NMR spectra of SF and SFO after 1 day of oxidization and b) IR spectra of solutions SF, SFO and gel SFO-SSDopa.



**Figure S7.** Effect of different parameters (concentration of  $Cu^{2+}$ , volume ratio of SFO and SSDopa, TA%) on the gelation time and the corresponding digital camera images of the obtained products. Higher concentration of  $Cu^{2+}$  and TA lead to phase separation due to high local concentration. When the concentration of TA reaches a certain critical value (2.73%), what appears is no longer coagulation, but an instantaneous formation of gum. The volume ratio of SFO and SSDopa changed the texture and color of hydrogels.



**Figure S8.** a) Sample lifted from a 50 g water bottle, showing a strong adhesive effect. b) Detachment force of hydrogel samples with different formulations compared to the control group (Reference sample with only Gel or Hap&Gel coating on metal slides). \* p < 0.05, \*\* p < 0.01 and \*\*\* p < 0.001 indicate the statistical significance between compared groups. n = 3 replicates.



**Figure S9.** Digital images from a thermal camera showing the surface temperature of the sample under various power densities for assessment of the PTT efficacy of hydrogel samples with different ratios of SFO/SSDopa irradiated by 808 nm laser with 0.45 W•cm<sup>-2</sup> power densities in the dry environment.



**Figure S10.** PTT effect of hydrogel samples with different ratios of SFO/SSDopa at 0.75 W•cm<sup>-2</sup> power densities in PBS.



**Figure S11.** Dynamic monitoring images (0, 6, 12 and 24 h) of migration of MC3T3-E1 cell in the fresh medium as control and hydrogel extracts (20 mg/ mL, collected after 5 days soaking) of different components.

## **Reference:**

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